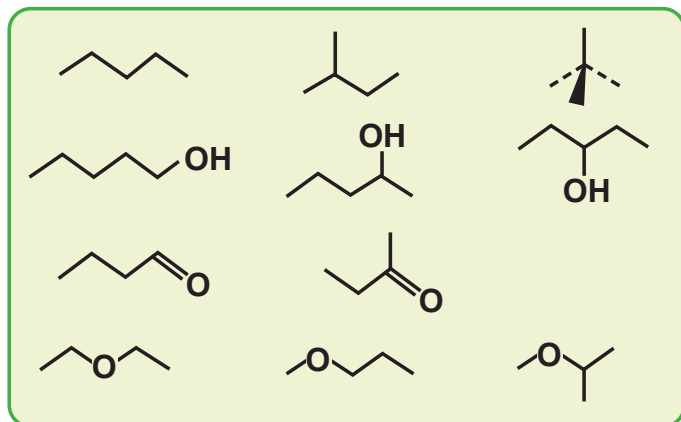


## CONSTITUTIONAL ISOMERISM



### Pair of functional isomers:

- Alcohols and ethers ( $C_nH_{2n+2}O$ )
- Aldehydes and ketones ( $C_nH_{2n}O$ )
- Carboxylic acids and esters ( $C_nH_{2n}O_2$ )
- $1^\circ$ ,  $2^\circ$  &  $3^\circ$  amines ( $C_nH_{2n+3}N$ )

$$DBE = \frac{\sum n(v-2)}{2} + 1 = \text{sum of no. of } \pi \text{ bonds} + \text{rings}$$

in the molecule

( $n$  is no of atoms of particular element &  $v$  is corresponding valency in given molecule).

## CONSTITUTIONAL ISOMERISM [STRUCTURAL]

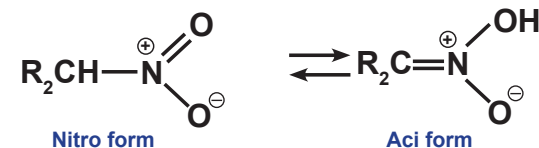
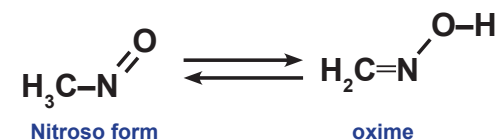
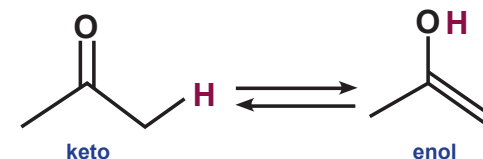
CHAIN [SKELETAL]  
ISOMERISM

POSITION ISOMERISM  
[REGIOISOMERISM]

FUNCTIONAL ISOMERISM

METAMERISM

PROTROPIC TAUTOMERISM  
[PROTROPY]



Q. The number of structural isomers possible from the molecular formula is

- (a) 4    (b) 5    (c) 2    (d) 3

Q. Identify the compound that exhibits tautomerism

- (a) 2-Pentanone    (b) Phenol  
(c) 2-Butene    (d) Lactic acid

Q. The number of structural isomers for  $C_6H_{14}$  is

- (a) 3    (b) 4    (c) 5    (d) 6

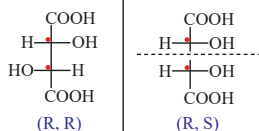
# STEREISOMERISM

## STEREISOMERISM

### CONFIGURATIONAL ISOMERISM

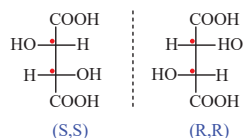
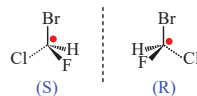
#### OPTICAL ISOMERISM [(+)/d dextrorotatory (-)/l levorotatory]

#### DIASTEREOMER

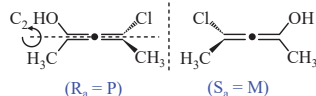


#### ENANTIOMER

chirality centre  
(asymmetric atom):



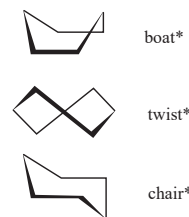
axial chirality:



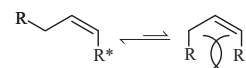
### CONFORMATIONAL ISOMERISM

#### CONFORMER

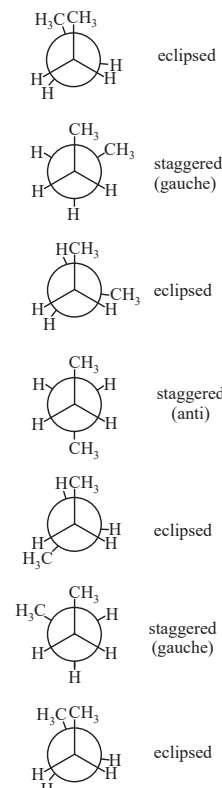
#### RING CONFORMATIONS



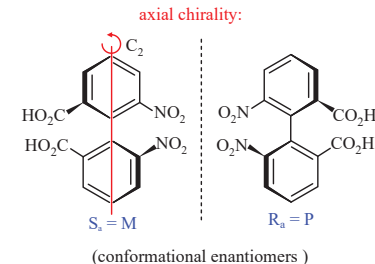
allylic strain:



#### ROTAMER

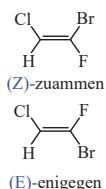


#### ATROPISOMER

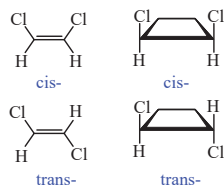


#### GEOMETRIC ISOMERISM

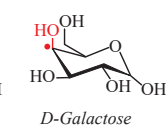
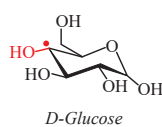
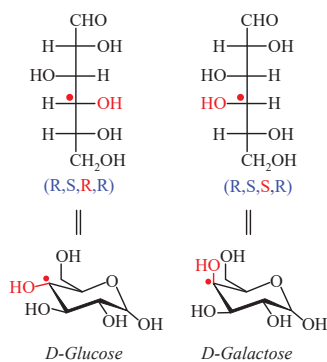
##### ABSOLUTE stereochemistry E/Z-ISOMER



##### RELATIVE stereochemistry cis/trans-ISOMER

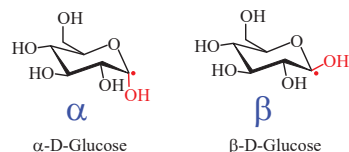


#### EPIMER



#### ANOMER

glycosides, hemiacetals, sugars



Q. Which of the following does not show geometrical isomerism?

- (1) 1, 2-dichloro-1-pentene
- (2) 1, 3-dichloro-2-pentene
- (3) 1, 1-dichloro-1-pentene
- (4) 1, 4-dichloro-2-pentene

Q. With respect to the conformers of ethane, which of the following statements is true?

- (1) Both bond angles and bond length remains same
- (2) Bond angle remains same but bond length changes
- (3) Bond angle changes but bond length remains
- (4) Both bond angle and bond length change

Q. Out of the following, the alkene that exhibits optical isomerism is:

- (1) 2-methyl-2-pentene
- (2) 3-methyl-2-pentene
- (3) 4-methyl-1-pentene
- (4) 3-methyl-1-pentene