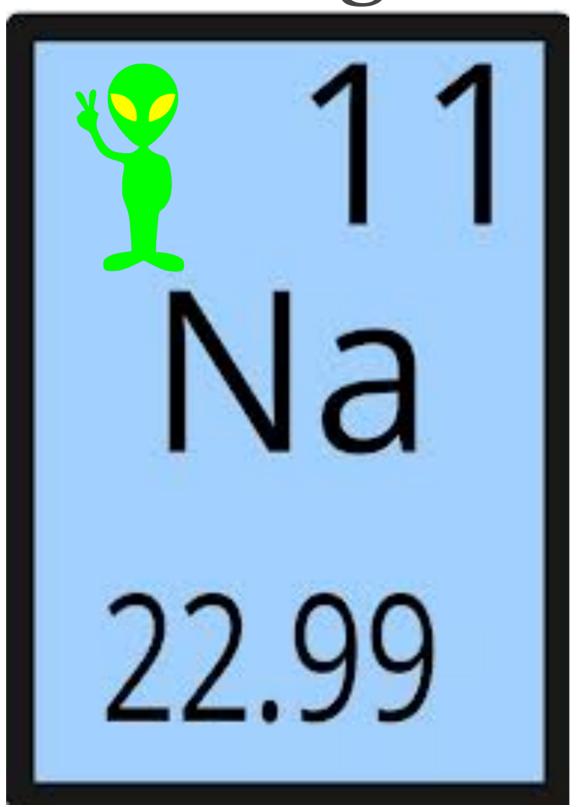
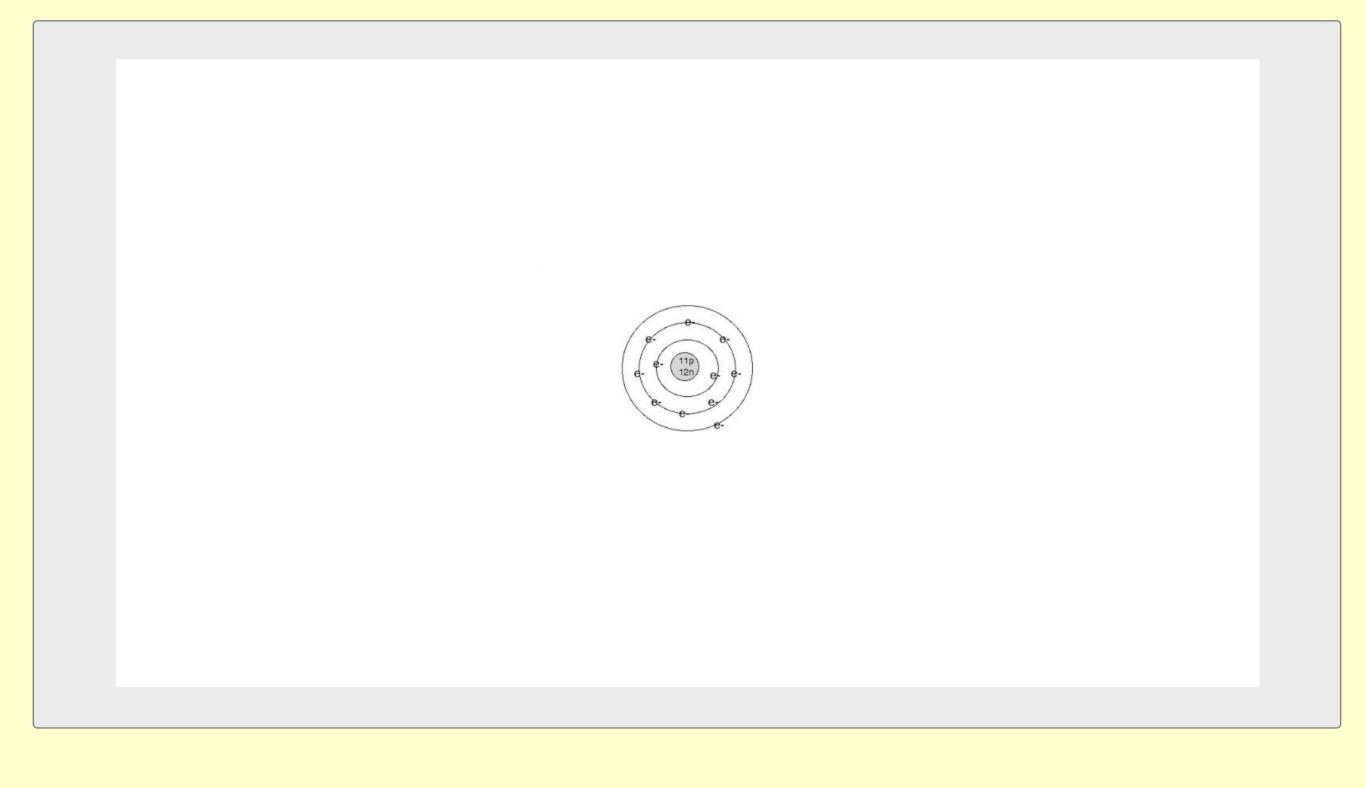
$\underline{Natronlauge}(Sodium)$



The Bohr Diagram!



Sodium

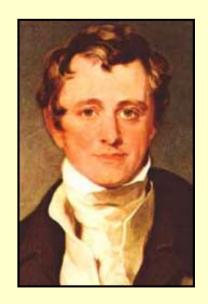
Sodium's symbol is Na. This comes from the German word for caustic soda: natronlauge. Sir Humphry Davy was given the credit of discovering Sodium in 1807 at the Royal Institution, London. He named it sodium because he used caustic soda as the source of the element.

Sodium's atomic number is eleven, and the atomic mass is twenty-three. Sodium has eleven protons, eleven electrons, and twelve neutrons. Sodium is a metal. It is in period 3, and belongs to family 1a, alkali metals.

The Periodic Table of the Elements

1																	2
H																	He
Hydrogen 1.00794																	Helium 4.003
3	4											5	6	7	8	9	10
Li	Be											В	C	N	O	F	Ne
Lithium	Beryllium											Boron	Carbon	Nitrogen	Oxygen	Fluorine	Neon
6.941	9.012182											10.811	12.0107	14.00674	15.9994	18.9984032	20.1797
Na	Mg											Al	Si	P	S	Cl	Ar
Sodium	Magnesium											Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon
22.989770	24.3050	21	22	23	24	25	26	27	28	29	30	26.981538	28.0855	30.973761	32.066	35.4527 35	39.948
		l			l .					I							
K Potassium	Ca	Sc Scandium	Ti	V Vanadium	Cr	Mn Manganese	Fe	Co	Ni Nickel	Cu	Zn Zinc	Ga Gallium	Germanium	As Arsenic	Se Selenium	Br Bromine	Kr
39.0983	40.078	44.955910	47.867	50.9415	51.9961	54.938049	55.845	58.933200	58.6934	63.546	65.39	69.723	72.61	74.92160	78.96	79.904	83.80
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Rubidium 85.4678	Strontium 87.62	Yttrium 88.90585	Zirconium 91.224	Niobium 92.90638	Molybdenum 95.94	Technetium (98)	Ruthenium 101.07	Rhodium 102.90550	Palladium 106.42	Silver 107.8682	Cadmium 112.411	Indium 114.818	Tin 118.710	Antimony 121.760	Tellurium 127.60	Iodine 126.90447	Xenon 131.29
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Cesium 132.90545	Barium 137,327	Lanthanum 138,9055	Hafnium 178.49	Tantalum 180.9479	Tungsten 183.84	Rhenium 186.207	Osmium 190.23	Iridium 192.217	Platinum 195.078	Gold 196.96655	Mercury 200.59	Thallium 204.3833	Lead 207.2	Bismuth 208,98038	Polonium (209)	Astatine (210)	Radon (222)
87	88	89	104	105	106	107	108	109	110	111	112	113	114		(=0,5)	(=14)	(===)
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt									
Francium (223)	Radium (226)	Actinium (227)	Rutherfordium	Dubnium (262)	Seaborgium (263)	Bohrium (262)	Hassium (265)	Meitnerium	(269)	(272)	(277)						
(223)	(220)	(227)	(261)	(202)	(203)	(202)	(203)	(266)	(209)	(2/2)	(277)						
				58	59	60	61	62	63	64	65	66	67	68	69	70	71
				Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
				Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium
				140.116 90	140.90765 91	144.24 92	93	150.36 94	151.964 95	157.25 96	158.92534 97	162.50 98	164.93032 99	167.26	168.93421	173.04	174.967
															101	102	103
				Th Thorium	Pa Protactinium	Uranium	Np Neptunium	Pu Plutonium	Am Americium	Cm	Bk Berkelium	Cf Californium	Es Einsteinium	Fermium	Md Mendelevium	No Nobelium	Lr
				232.0381	231.03588	238.0289	(237)	(244)	(243)	(247)	(247)	(251)	(252)	(257)	(258)	(259)	(262)





Properties

Sodium has a silver luster, and it is smooth. It is a solid at room temperature. The melting temperature of sodium is 97.72 degrees Celsius. The boiling temperature is 883 degrees Celsius! Wow!

Sodium is found in the compounds NaCl, table salt, NaOh, caustic soda, and also NaHCO₃, which is baking soda. It is also used for glass, soap, paper, and to purify molten metals. Sodium is obtained by being chemically separated from a compound.











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Isotopes

Sodium has 18-22 isotopes, because opinions differ. There are many isotopes, but only one of them is stable. Isotope-24 is used to study the electrolytes in the body.

Sodium-18: 1.3*10⁻²¹

Sodium-19: < 40 nanoseconds

Sodium-20: 447.9 milliseconds

Sodium-21: 22.49 seconds

Sodium-22 : 2.6027 years

Sodium-23: stable

Sodium-24: 14.997 hours

Sodium-25: 59.1 seconds

Sodium-26 : 1.07128 seconds

Sodium-27: 301 milliseconds

Sodium-28: 30.5 milliseconds

Sodium-29: 44.9 milliseconds

Sodium-30: 48 milliseconds

Sodium-31: 17 milliseconds

Sodium-32: 13.2 milliseconds

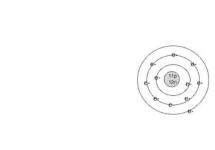
Sodium-33: 8 milliseconds

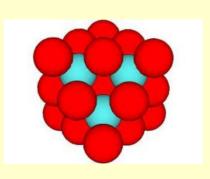
Sodium-34: 5.5 milliseconds

Sodium-35: 1.5 milliseconds

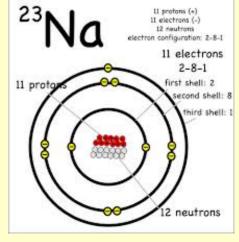
Sodium-36: <180nanoseconds

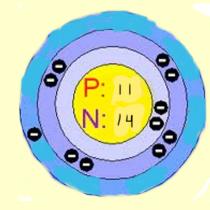
Sodium-37 : >60 nanoseconds





Sodium 23





Fun Facts

Sodium is the sixth most abundant element on Earth.

If you ate sodium, it would kill you. But, if the two dangerous elements, sodium and chlorine combined, it would create NaCl, or sodium chloride. This is also called table salt.

Neurons in the brain are like sodium batteries. Without sodium ions, you wouldn't be able to think. You would die without them.

Sodium reacts with both water and air. If it comes in contact with oxygen, a crust of sodium oxide. Sodium reacts explosively with water. Fire forms, and they give of the gas hydrogen.

Yellow sodium oxide crust!





