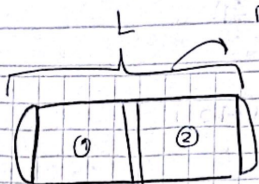


piston en equilibrio a

$$L = \frac{1}{3}$$



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gas monoatomica

- 1 mol gas ideal monoatomico

$$T = ?$$

$$- 400 K$$

inductividad de

$$K = 389,6$$

$$A = 0,01 m^2$$

$$L = 0,30 m$$

$$a) T_0^2 = 200 K. =$$

Como es un gas ideal y como las ecuaciones

$$\begin{cases} P_1 V_1 = n R T_1 \\ P_2 V_2 = n R T_2 \end{cases}$$

$$P_1 = \frac{n R T_1}{V_1}$$

$$P_2 = \frac{n R T_2}{V_2}$$

En equilibrio cuando P_1 es igual a P_2

$$\text{Entonces } \frac{n R T_1}{V_1} = \frac{n R T_2}{V_2}$$

despejamos para T_2 Reemplazando

$$T_2 = T_1 \frac{V_2}{V_1} \quad T_2 = 400 K \frac{V_2}{V_1}$$



Como sabemos que es un cilindro entonces calculamos su volumen como

$$V = A b \times h = \pi r^2 \times h$$

$$V_1 = \left(L - \frac{L}{3} \right) \cdot \pi r^2$$

$$= \frac{\pi r^2 2L}{3}$$

$$h \begin{cases} ① L - L/3 \\ ② L/3 \end{cases}$$

$$V_2 = \left(\frac{L}{3} \right) \cdot \pi r^2 = \frac{L \pi r^2}{3}$$

$$T_2 = 400 K \frac{\frac{\pi r^2 2L}{3}}{\frac{L \pi r^2}{3}} = 400 K \cdot 2 = 800 K$$

$$T_2 = \frac{400 K}{2}$$

$$T_2 = 200 K$$

$$C = \frac{KA}{nCVL}$$

$$C_V = \frac{3}{2}R$$

b.)

$$nC_V \frac{dT_1}{dt} = -\frac{KA}{L} (T_1 - T_2)$$

$$nC_V \frac{dT_2}{dt} = \frac{KA}{L} (T_1 - T_2)$$

$$\left. \frac{dT_1}{dt} \right|_{t=0} = -C (T_1^0 - T_2^0)$$

$$\left. \frac{dT_2}{dt} \right|_{t=0} = C (T_1^0 - T_2^0)$$

la primera ley dice $\Delta U = nC_V \Delta T$
 $= \Delta Q - \Delta W$ $\rightarrow \Delta W = 0$

Entonces $dU = nC_V dT = dQ$

usando la ley de la transferencia de calor

$$nC_V \frac{dT_1}{dt} = -\frac{KA}{L} (T_1 - T_2)$$

como $C = \frac{KA}{nCVL}$ Entonces

$$\frac{dT_1}{dt} = -C (T_1 - T_2)$$

$$\frac{dT_2}{dt} = -\frac{KA}{L} (T_1 - T_2)$$

$$\frac{dT_2}{dt} = C (T_1 - T_2)$$

Enonces

$$\left. \frac{dT_1}{dt} \right|_{t=0} = -C (400 - 200) = -C (200)$$

$$\left. \frac{dT_1}{dt} \right|_{t=0} = C (400 - 200) = C (200)$$

c)

$$\begin{pmatrix} \frac{dT_1}{dt} \\ \frac{dT_2}{dt} \end{pmatrix} = C \begin{pmatrix} -(T_1 - T_2) \\ (T_1 - T_2) \end{pmatrix}$$

$$\begin{pmatrix} \frac{dT_1}{dt} \\ \frac{dT_2}{dt} \end{pmatrix} = -C \underbrace{\begin{pmatrix} 1 & -1 \\ -1 & 1 \end{pmatrix}}_A \begin{pmatrix} T_1 \\ T_2 \end{pmatrix}$$

$\det(A - rI)$

$$\det(A) = \begin{pmatrix} r+C & -C \\ -C & r+C \end{pmatrix} \quad F_2 \rightarrow F_2 + F_1 \quad \begin{pmatrix} -C \\ 0 \end{pmatrix}$$

$$\det(A) = r^2 + 2c + c^2 - c^2 = 0$$

$$r(r+2c) \quad \begin{cases} r = 0 \\ r = -2c \end{cases}$$

$$r = 0$$

$$\begin{pmatrix} c & -c \\ -c & c \end{pmatrix} \begin{pmatrix} a_1 \\ a_2 \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \end{pmatrix}$$

$$\begin{pmatrix} a_1 \\ a_2 \end{pmatrix} = \begin{pmatrix} 1 \\ 1 \end{pmatrix}$$

$$r = -2c$$

$$\begin{pmatrix} c & -c \\ -c & c \end{pmatrix} \begin{pmatrix} a_1 \\ a_2 \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \end{pmatrix}$$

$$\begin{pmatrix} a_1 \\ a_2 \end{pmatrix} = \begin{pmatrix} 1 \\ -1 \end{pmatrix}$$

$$T_1 = c_1 + c_2 e^{-2ct}$$

$$T_2 = c_1 - c_2 e^{-2ct}$$