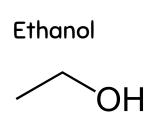
# Hand sanitizer & detergent

How do E.coli react to ethanol and SDS shock?

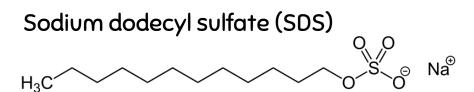


Pauline GASQUET - Clara HAAS - Adèle NEVOT - Alexandra PERRON

# Ethanol and SDS are found everywhere









# Why are they not used in the same products?

#### Different chemical structures and therefore different characteristics

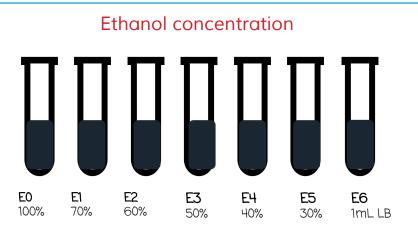




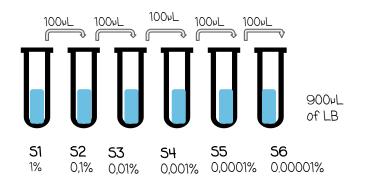
Ethanol found naturally in nature (in alcohol we drink!)



#### Determining detergent concentration needed



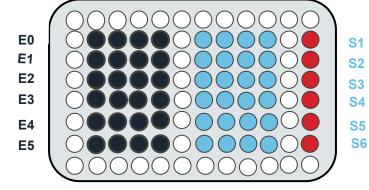
#### SDS concentration









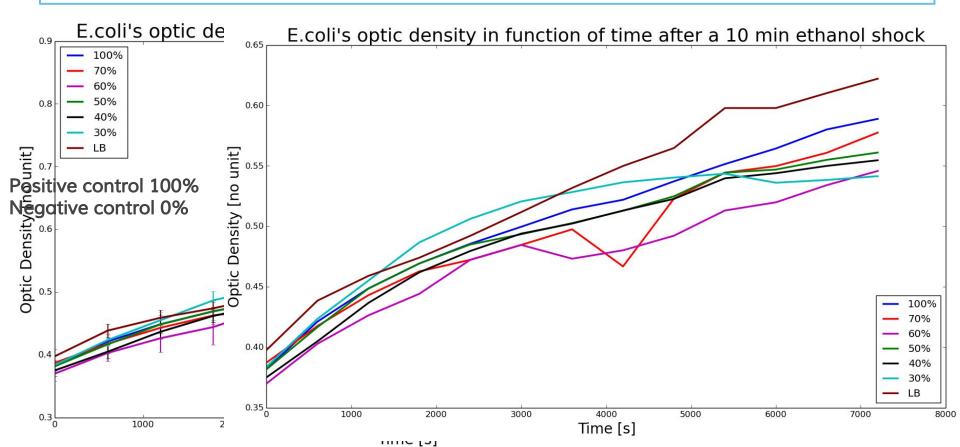




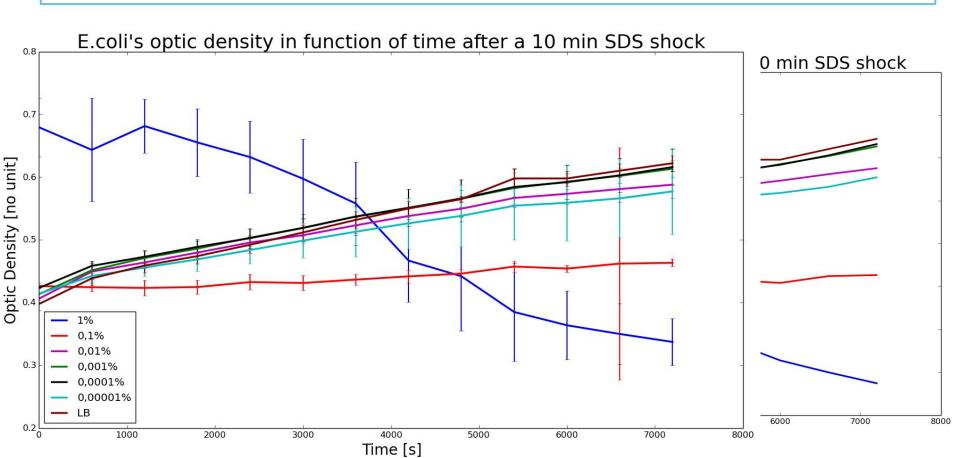




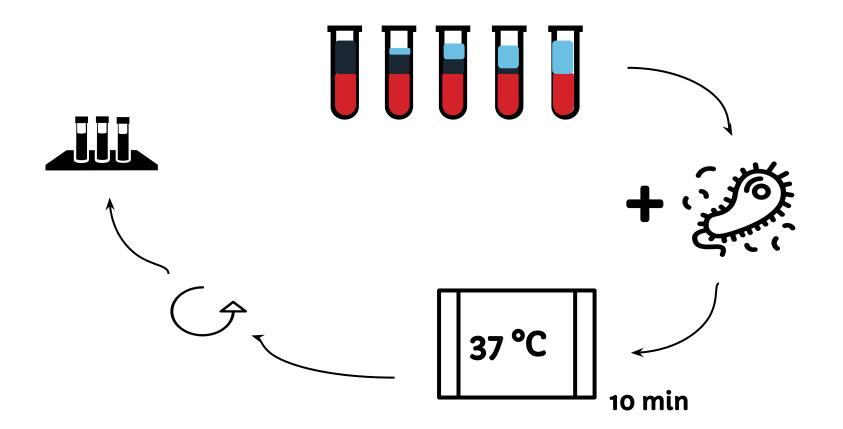
# No reaction to shock by ethanol or SDS



# No reaction to shock by ethanol or SDS

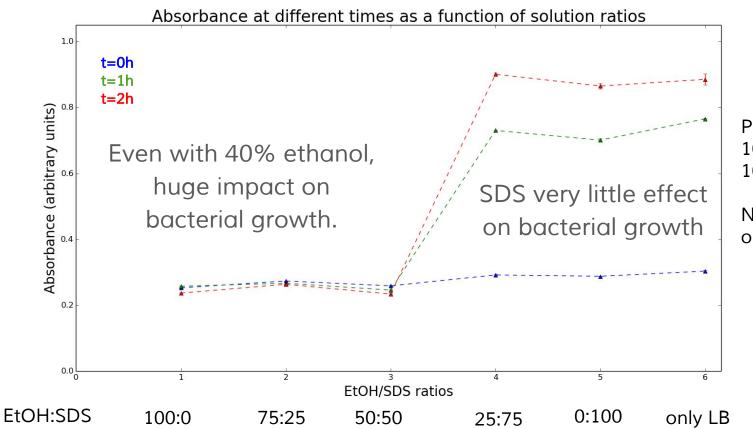


### Reactions to different detergent ratios





#### Ethanol more effective than SDS



Positive controls: 100% of SDS and 100% EtOH

Negative control: only LB

#### 50:50 EtOH and SDS mix most efficient.

A mixed solution of **SDS and EtOH** with a **50:50 ratio** is **more efficient** than 100:0 or 0:100.

**Ethanol** is more effective than SDS at slowing down bacterial growth after 10 minutes of exposition.





#### Biases and sources of error



- Manipulation done by different people
- No biological replicate
- Few technical replicates
- No exponential phase (might be due to the moment we took the cells)
- Final concentration choice (influenced by experimental mistakes)



#### What to remember

**Ethanol** is much more effective than **SDS** at stopping bacterial growth.



A **50:50 ratio** might be **optimal** in experimental conditions but perhaps not applicable.

We are right to **sterilize our hands** in the lab with ethanol!

## Acknowledgement

- Tamara
- Ivan & Aïmen
- Nicolas for the help with the sensors, the code and GitHub
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- Our mentor Nicholas Garcia (Microchem laboratory in Texas)

#### **Biosensors Week 4**

#### twitter @BacteriaCleaner











#### Resources

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