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A room-temperature synthesis of 2,2'-bisoxazoles through palladium-catalyzed oxidative coupling of α -isocyanoacetamides†

Jian Wang, Shuang Luo, Jing Li and Qiang Zhu*

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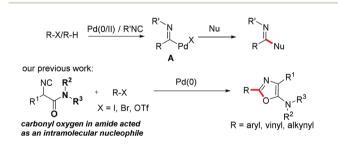
A palladium-catalyzed synthesis of symmetric and unsymmetric 2,2'-bisoxazoles starting from readily available α -isocyanoacetamides was developed. The reaction was performed at room temperature in air which acted as the sole oxidant of Pd⁽⁰⁾. Mechanistic studies suggested that double isocyanide insertion into the Pd^(II)-O bond was involved.

Acting as an isoelectronic equivalent of carbon monoxide, isocyanide has shown its great potential in palladium-catalyzed isocyanide insertion reactions. Imidoyl palladium(II) complex **A** was considered as a general intermediate in reaction with various nucleophiles followed by reductive elimination, generating amidines, amides, ketimines, imidates, thioimidates and aldehydes correspondingly (Scheme 1). Functionalized heterocycles could be generated by linking a nucleophile to substrates R-X/R-H ready for imidoyl palladium(II) intermediate formation upon oxidative addition or C-H bond activation and isocyanide insertion. Another strategy for heterocycle construction involving isocyanide insertion as a key step employs bisnucleophiles and isocyanides under oxidative con-

ditions.8 For instance, Orru and co-workers reported an efficient synthesis of cyclic guanidine derivatives and related heterocycles via palladium-catalyzed isocyanide insertion with diamines or amino alcohols.8a Recently, our group developed a different strategy aimed at construction of heterocycles by linking a nucleophile to an isocyanide substrate. α-Isocyanoacetamides, in which the carbonyl oxygen in the amide moiety acted as an intramolecular nucleophile, reacted with aryl, vinyl, or alkynyl halides under palladium catalysis to provide C2-diversified oxazoles. During the optimization of reaction conditions, a symmetric 2,2'-bisoxazole byproduct was identified, albeit in very low yield (Scheme 1). In this novel process, two oxazole rings are formed in one pot through multiple bond formation including two C-O bonds and one C-C bond starting from acyclic substrates. This unprecedented and unexpected transformation intrigued us to investigate it in detail.

 $C(sp^2)$ - $C(sp^2)$ direct linked bisheterocycles are of vital importance in pharmaceuticals, natural products, and functional materials. 10 Traditional approaches to these compounds are mostly based on heteroaryl (pseudo)halides and organometallic reagents.11 In recent years, more step-efficient and atom-economic strategies employing oxidative coupling of existing heterocyclic skeletons through C-H bond activation were developed. 12 For instance, methods directed towards 2,2'bisoxazoles were successfully developed by transition metal catalyzed coupling of dual C-H bonds.13 However, limitations of these methods, including high reaction temperatures, using a stoichiometric or excess amount of Cu/Ag-based oxidant, still exist. Herein, we report a novel palladium-catalyzed synthesis of symmetric and unsymmetric 2,2'-bisoxazoles by oxidative homo- and cross-coupling of readily available α-isocyanoacetamides.14 This reaction occurs smoothly at room temperature and uses air as the sole oxidant.

The reaction conditions were screened with 2-isocyano-2-phenyl-1-(piperidin-1-yl)ethanone ${\bf 1a}$ as a test substrate catalyzed by Pd(OAc)₂ (10 mol%) in air at room temperature



this work:
$$\begin{array}{c} NC \\ NR^{1}R^{2} \end{array} \xrightarrow{Pd(II)} \begin{array}{c} R^{2}RN \\ NR^{1}R^{2} \end{array}$$

Scheme 1 Palladium-catalyzed isocyanide insertion reactions.

State Key Laboratory of Respiratory Disease, Guangzhou Institutes of Biomedicine and Health, Chinese Academy of Sciences, 190 Kaiyuan Avenue, Guangzhou 510530, China. E-mail: zhu_qiang@gibh.ac.cn; Fax: (+86) 20-3201-5299

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Table 1 Optimization of reaction conditions^a

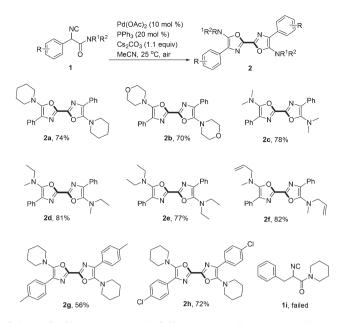
Entry	Solvent	Base	Ligand	Atmosphere	Yield ^b
1	DMF	Cs ₂ CO ₃	PPh ₃	Air	26%
2	DCM	Cs_2CO_3	PPh_3	Air	66%
3	Dioxane	Cs_2CO_3	PPh_3	Air	44%
4	MeCN	Cs_2CO_3	PPh_3	Air	70%
5	MeCN	Cs_2CO_3	PPh_3	O_2	44%
6 ^c	MeCN	Cs_2CO_3	_	Air	51%
7	MeCN	LiOtBu	PPh_3	Air	50%
8^d	MeCN	Cs_2CO_3	PPh_3	Air	74%

^a Reaction conditions: 1a (0.2 mmol), Pd(OAc)₂ (10 mol%), base (0.22 mmol, 1.1 equiv.), PPh $_3$ (20 mol%), solvent (2 mL), in air, 25 °C, 0.5 h. b Isolated yield. c 2.0 h. d A solution of 1a in MeCN (1 mL) was added to the reaction mixture via a syringe pump within 0.5 h.

(Table 1). Among various solvents tested, the reaction performed best in MeCN in the presence of Cs₂CO₃ (1.1 equiv.) and PPh3 (20 mol%), delivering the desired symmetric 2,2'bisoxazole 2a in 70% yield (entries 1-4). Further investigations including changing the reaction atmosphere from air to pure O₂ or replacing the base from Cs₂CO₃ to LiOtBu gave lower yields of 2a (entries 5 and 7). In the absence of PPh₃, the transformation was much less efficient (51% yield, entry 6). When a solution of 1a in MeCN (1.0 mL) was added slowly via a syringe pump over 0.5 h to a mixture containing Pd(OAc)₂, PPh₃, Cs₂CO₃ and 1 mL of MeCN, the yield of 2a increased slightly to 74% (entry 8).

With the optimized reaction conditions in hand, the scope of α -isocyanoacetamides was then screened (Scheme 2). Besides piperidinyl amide, a cyclic morpholino analogue of 1a also generated the corresponding product 2b smoothly in 70% yield. Other α-isocyanoacetamides derived from acyclic secondary amines including N,N-dimethylamine (1c), N-methyl-Nethylamine (1d), N,N-diethylamine (1e) and N-methylallylamine (1f) all homo-coupled efficiently to produce the corresponding symmetric 2,2'-bisoxazoles (2c-2f) in good yields. It is noteworthy that the terminal alkene in 2f survived the reaction well. Methyl and chloro substituted 2,2'-bisoxazoles 2g and 2h were obtained in 56% and 72% yields respectively. Unfortunately, isocyanoacetamide bearing a benzyl group rather than an aryl one at the α -position was not a suitable substrate in this transformation (1i).

When two different α-isocyanoacetamides were present, an unsymmetric 2,2'-bisoxazole product derived from crosscoupling together with two homo-coupling products was obtained (Scheme 3). For example, addition of a solution of 1a (0.2 mmol, 1 equiv.) and 1b (3 equiv.) in 4 mL of MeCN to an open reaction tube containing the catalyst, ligand, base and CH₃CN (1 mL) via a syringe pump in 1 h generated an unsymmetric 2,2'-bisoxazole product 3a in synthetically useful yield



Scheme 2 Scope of symmetric 2,2'-bisoxazoles. Reaction conditions: a solution of 1a (0.20 mmol) in MeCN (1 mL) was added to the reaction mixture containing Pd(OAc)₂ (0.02 mmol, 10 mol%), PPh₂ (0.04 mmol, 20 mol%), Cs₂CO₃ (0.22 mmol, 1.1 equiv.) and MeCN (1 mL) via a syringe pump within 0.5 h at 25 °C in air.

Scheme 3 Scope of unsymmetric 2,2'-bisoxazoles. Reaction conditions: a solution of 1 (0.20 mmol) and 1b (0.6 mmol) in MeCN (4 mL) was added to the reaction mixture containing Pd(OAc)2 (0.02 mmol, 10 mol%), PPh₃ (0.04 mmol, 20 mol%), Cs₂CO₃ (0.22 mmol, 1.1 equiv.) and MeCN (1 mL) via a syringe pump within 1 h at 25 °C in air. Isolated yields of 2b are based on 1b. Other yields are based on another reactant 1.

(51%) after careful chromatographic isolation. Symmetric 2,2'bisoxazoles 2a and 2b generated from homo-coupling were also obtained in 13% and 69% yields, respectively. The selectivity for cross-coupling was better in a reaction of 1c and 1b, generating unsymmetric product 3b in 63% yield. Unsymmetric 2,2'-bisoxazole 3c containing an aromatic chloride func-

Scheme 4 Large scale synthesis and further transformations.

tionality was also isolated in 57% yield. The current strategy provides an efficient approach to both symmetric and unsymmetric 2,2′-bisoxazoles in one step starting from simple acyclic α -isocyanoacetamides. It is notable that two heterocyclic rings are constructed simultaneously at ambient temperature in open air. Three chemical bonds including two C–O bonds and one C–C bond are formed with 100% atom-economy during this process.

This reaction was scalable, as exemplified by sub-gram preparation of **2a** with equal efficiency (a, Scheme 4). Further diversification of the obtained oxazole product **2h** was also performed. Transforming the chloride moiety to boronic acid ester through palladium catalysis was realized in 89% yield. The product **4** containing two aromatic boronic acid ester moieties is expected to be a useful precursor for more complicated symmetric **2**,2'-bisoxazole synthesis (b).¹⁵ Suzuki coupling of **2h** with phenyl boronic acid also performed smoothly, giving highly conjugated product **5** in high yield (c).¹⁶

To verify the reaction pathway, C2 unsubstituted oxazole 6 was treated under the standard aerobic conditions. Most of the starting material 6 was recovered with no homo-coupling product 2a being detected, which suggested that 6 was an unlikely reaction intermediate. Although the role of triphenyl phosphine was not fully understood, it may facilitate the process of reductive elimination and stabilize the Pd(0) species before being oxidized to $Pd^{(II)}$ by O_2 in air.

A plausible reaction mechanism was proposed in Scheme 5. Coordination of the carbonyl oxygen in α -isocyanoacetamide 1

$$\begin{array}{c} \text{CN} \\ \text{R}^1 \\ \text{NR}^2 \text{R}^3 \\ \text{Pd}(\text{OAc})_2 \\ \text{Pd}(\text$$

Scheme 5 Proposed mechanism.

with $Pd(OAc)_2$ affords intermediate **I.** Deprotonation and the subsequent isocyanide insertion to the Pd–O bond forms the first oxazole ring in intermediate **III.** Repeating the same process furnishes the key bisoxazole ligated palladium(II) intermediate **VI.** Reductive elimination releases the homo-coupling product **2** and the $Pd^{(0)}$ species which is reoxidized to $Pd^{(II)}$ by O_2 in air. It is also possible that isocyanide insertion to the Pd–O bond in $Pd(OAc)_2$ takes place before its coordination with the carbonyl oxygen.

In summary, we have developed a novel palladium-catalyzed synthesis of symmetric and unsymmetric 2,2'-bisoxazoles starting from readily available acyclic α -isocyanoacetamides. Double isocyanide insertion was believed as a key step in this transformation. The reaction was performed at room temperature in air which acted as the sole oxidant of Pd(0). The resulting symmetric or unsymmetric products were highly π -conjugated, showing their great potential in functional material synthesis.

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