

RESEARCH ARTICLE

[View Article Online](#)
[View Journal](#)


Cite this: DOI: 10.1039/c8qo00008e

Synthesis of *trans*-disubstituted-2,3-dihydro-benzofurans by a formal [4 + 1] annulation between *para*-quinone methides and sulfonium salts†

 Ying Zhi,^a Kun Zhao,^a Carolina von Essen,^b Kari Rissanen^b and Dieter Enders^a

 Received 4th January 2018,
 Accepted 26th February 2018

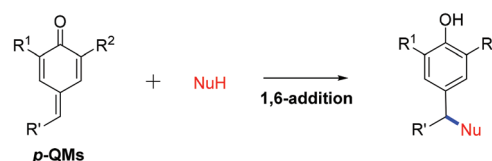
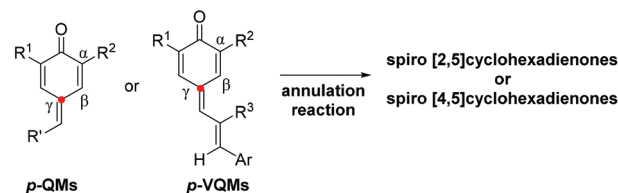
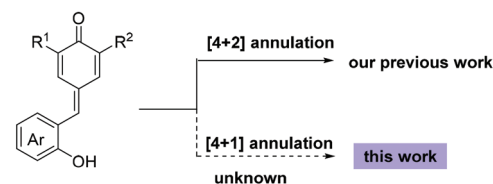
DOI: 10.1039/c8qo00008e

rsc.li/frontiers-organic

An efficient protocol for the synthesis of *trans*-disubstituted-2,3-dihydrobenzofurans through [4 + 1] annulation of *para*-quinone methides with sulfonium salts has been developed. Under very mild conditions this unprecedented reaction occurs in good to excellent yields (up to 99%), offering a straightforward access to a variety of 2,3-dihydrobenzofurans.

The 2,3-dihydrobenzofuran skeleton is an important structural unit that frequently appears not only in many synthetic bio-active molecules but also numerous natural products.¹ Thus, the development of methods to synthesize this kind of skeleton has attracted considerable attention in the synthetic community, and therefore numerous synthetic strategies have been reported to assemble 2,3-dihydrobenzofurans.² Among those strategies, the [4 + 1] annulation reaction has been shown to be very reliable for the construction of 2,3-dihydrobenzofuran frameworks.³ However, the majority of the published [4 + 1] annulation reactions are based on the use of *o*-quinone methide intermediates.⁴ Clearly, it is still very desirable to develop alternative methods to access this important skeleton.

On the other hand, due to their intrinsic reactivity *para*-quinone methides (*p*-QMs)⁵ have been extensively explored during the past several years, and can react with a large number of nucleophiles in racemic⁶ or enantioselective fashions (Scheme 1a).⁷ Moreover, applications of *para*-quinone methides in annulation reactions have also been realized. The examples of use of *p*-QMs or vinyl *p*-QMs for annulation reactions to access spirocyclohexadienones were reported by the research groups of Yao, Fan, Zhao and Waser (Scheme 1b).⁸ Among those reports the γ -carbon could serve as a potential nucleophilic site. Very recently, our group demonstrated that hydroxy-substituted *para*-quinone methides could be used as

a) 1,6-conjugate addition to *p*-QMs:b) annulation reactions based on *p*-QMs or vinyl *p*-QMs:c) annulation reactions based on hydroxy-substituted *p*-QMs:Scheme 1 Reported reactions based on *p*-QMs and our design.

bifunctional substrates in an organocatalytic oxa-Michael/1,6-addition domino reaction, allowing the preparation of a series of 4-phenyl-substituted chromans in excellent stereoselectivities.⁹ In 2017, the use of hydroxy-substituted *p*-QMs in the synthesis of spiro[chromane-2,1'-isochromene] derivatives and dihydrocounmarins have also been reported by the Tu/Jiang group¹⁰ and Mei group,¹¹ respectively.

^aInstitute of Organic Chemistry, RWTH Aachen University, Landoltweg 1, 52074 Aachen, Germany. E-mail: kun.zhao@rwth-aachen.de, enders@rwth-aachen.de

^bDepartment of Chemistry, University of Jyväskylä, 40014 Jyväskylä, Finland

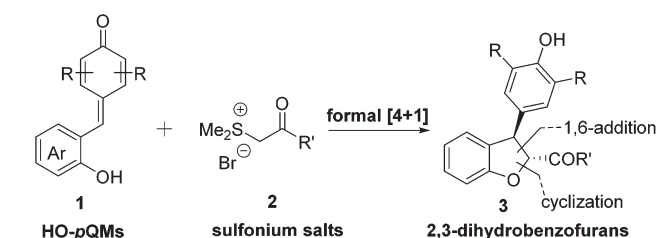
†Electronic supplementary information (ESI) available. CCDC 1814040. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c8qo00008e

Despite those reports mentioned above, to the best of our knowledge the reaction of hydroxy-substituted *p*-QMs with other reaction partners, such as sulfonium salts, in a [4 + 1] annulation fashion is still unknown at the outset of this work (Scheme 1c).¹² In this context, we envisioned that a formal [4 + 1] reaction between hydroxy-substituted *p*-QMs **1** and sulfonium salts **2** might occur in the presence of an appropriate base. This expected annulation reaction could provide a general and straightforward method to access 2,3-dihydrobenzofurans **3**. Herein, we wish to present our preliminary results (Scheme 2).

To test the feasibility of our reaction design, the *para*-quinone methide **1a** and ethyl dimethylsulfonium acetate bromide **2a** were selected as the model substrates to optimize the reaction conditions (Table 1). The initial experiment was conducted in which **1a** and **2a** were exposed to triethylamine (1.2 equiv.) in dichloromethane at room temperature. We were

pleased to obtain the expected product **3a** in an acceptable yield of 50% (Table 1, entry 1). We found that the use of a suitable base is very critical for the success of this annulation reaction and thus an extensive screening of base was performed. The commonly used organic bases DABCO and DIPEA resulted in worse results, affording the desired product **3a** in 12% and 37% yield, respectively (entries 2 and 3). Further optimization revealed that inorganic bases had a positive influence on this transformation (entries 4–9). When cesium carbonate was used, the yield was improved to 90% (entry 9). Striving for higher efficiency, the effects of solvents in this model reaction were then examined (entries 10–16) and we found that acetonitrile as the reaction solvent delivered the best result (96% yield, entry 12).

With the optimum reaction conditions in hand, we sought to investigate the generality of this [4 + 1] annulation process. First, the sulfonium salt component was evaluated. As shown in Table 2, a variety of sulfonium salts **2** underwent the formal [4 + 1] annulation reaction to furnish **3a–3i** in 72–99% yield. In details, sulfonium salts containing ester groups ($R^2 = \text{EtO}$, MeO and $t\text{-BuO}$) readily reacted with **1a** to afford the desired annulation products in high yields. Besides the alkyl ester sulfonium salts, the ketone ones ($R^2 = \text{aryl group}$) proved to be



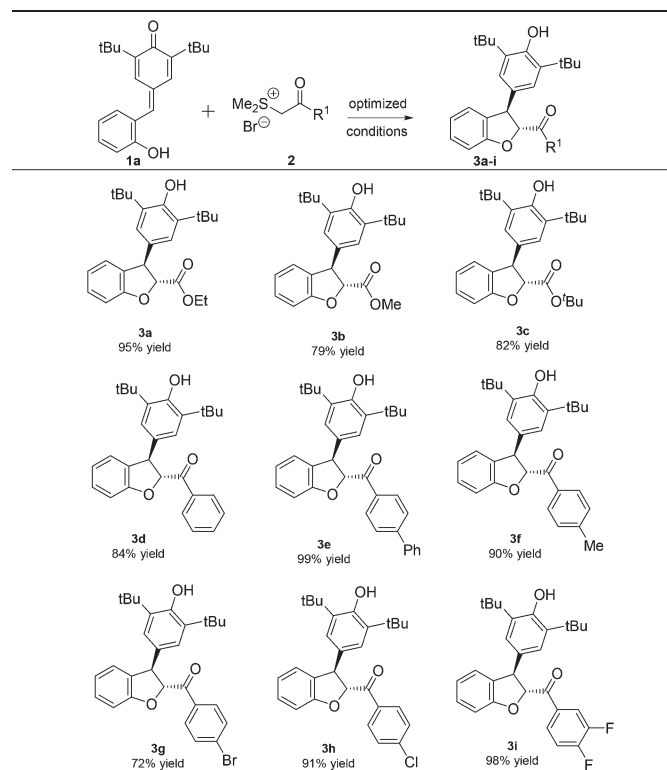
Scheme 2 Reaction design.

Table 1 Reaction condition optimization studies^a

Entry	Base	Solvent	<i>t</i> (h)	Yield ^{b,c} (%)
1	Et ₃ N	CH ₂ Cl ₂	48	50
2	DABCO	CH ₂ Cl ₂	48	12
3	DIPEA	CH ₂ Cl ₂	48	37
4	Na ₂ CO ₃	CH ₂ Cl ₂	48	43
5	K ₂ CO ₃	CH ₂ Cl ₂	4	87
6	K ₃ PO ₄	CH ₂ Cl ₂	4	84
7	NaOH	CH ₂ Cl ₂	48	42
8	KOH	CH ₂ Cl ₂	48	67
9	Cs ₂ CO ₃	CH ₂ Cl ₂	4	90
10	Cs ₂ CO ₃	Toluene	2	93
11	Cs ₂ CO ₃	THF	2	81
12	Cs ₂ CO ₃	CH ₃ CN	0.5	96
13	Cs ₂ CO ₃	EtOAc	4	85
14	Cs ₂ CO ₃	CHCl ₃	4	75
15	Cs ₂ CO ₃	CCl ₄	4	88
16	Cs ₂ CO ₃	Et ₂ O	4	82

^a All reactions were conducted with 0.2 mmol of **1a** (1.0 equiv.), 0.24 mmol of **2a** (1.2 equiv.), and 120 mol% of base in 2 mL of solvent at room temperature. ^b Yield of isolated compound **3a** after chromatography. ^c All diastereomeric ratios were higher than 20 : 1.

Table 2 Substrate scope of sulfonium bromides^a



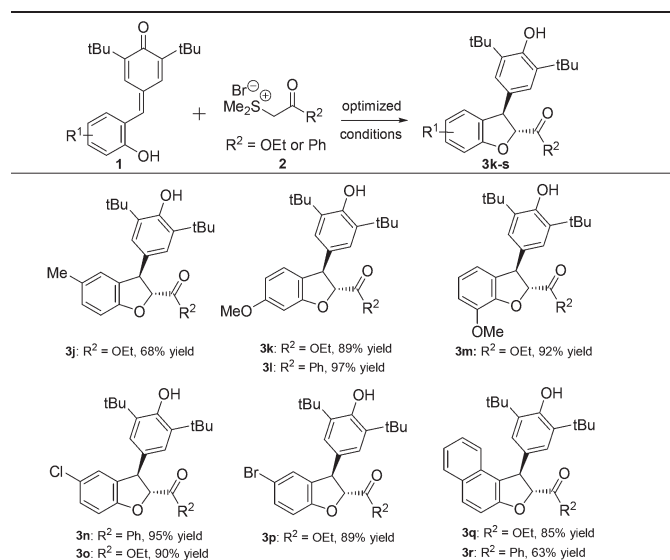
^a Unless otherwise noted, the reactions were conducted with 0.40 mmol of **1** (1.0 equiv.), 0.48 mmol of **2** (1.2 equiv.) and 120 mol% of base in acetonitrile (4.0 mL) at room temperature. Yields are those of the isolated products **3a–3i** after column chromatography. The diastereomeric ratio was determined by ¹H NMR and all of them were higher than 20 : 1.

suitable substrates as well. Both electron-donating substituents (Ph, Me) and electron-withdrawing groups (Br, Cl) in the *para* position of the aryl ring of compounds **2** were well tolerated and the corresponding dihydrobenzofurans **3d–3h** could be obtained in very good to excellent yields (up to 99% yield). It is worth to mention that the 3,4-difluoro-substituted sulfonium salt also worked very well, delivering product **3i** in 98% yield.

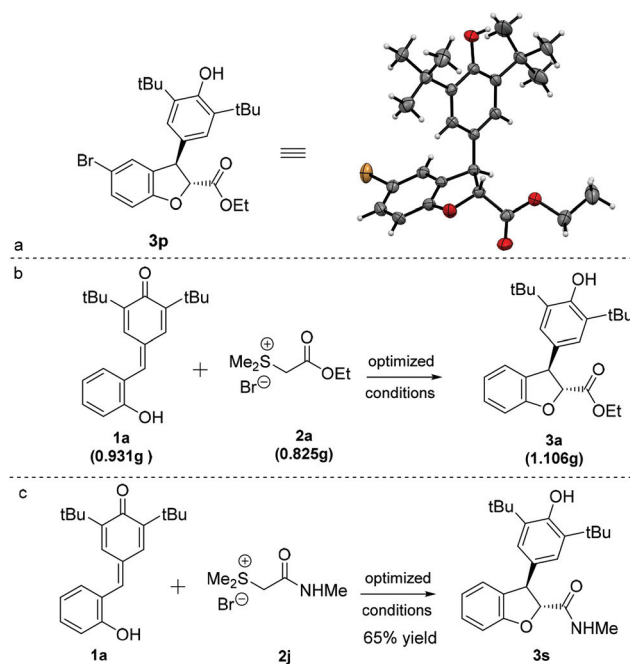
Next, the substrate scope of this reaction was examined further by varying the reaction partner **1** (Table 3). We found that a wide range of hydroxy-substituted *p*-QMs were viable substrates and gave products **3j–3r** in 63–97% yield. Substrates bearing electron-donating (R^1 = Me, OMe), or electron-withdrawing groups (R^1 = Cl, Br) at the C4, C5, or C6 position of the phenyl ring of *p*-QMs **1** readily underwent this annulation process to furnish the corresponding 2,3-dihydrobenzofuran products in good to excellent efficiencies (**3j–3p**). Moreover, a naphthyl-substituted *p*-QMs smoothly reacted with the sulfonium salts, and the products **3q** and **3r** were obtained in 85% and 63% yield, respectively. Notably, the relative configuration of **3p** was unambiguously determined by X-ray crystallographic analysis (Scheme 3a).¹³

To show the general applicability and robustness of this novel [4 + 1] annulation reaction, we performed a gram scale reaction using **1a** and **2a** under the optimal conditions; the expected product **3a** could be isolated in 93% yield without erosion of the efficiency of this process (Scheme 3b). Furthermore, our protocol was also suitable for the amide-substituted sulfonium salt, affording the expected product **3s** in 65% yield (Scheme 3c).

Table 3 Substrate scope of the *p*-QMs^a



^a Unless otherwise noted, the reactions were conducted with 0.40 mmol of **1** (1.0 equiv.), 0.48 mmol of **2** (1.2 equiv.) and 120 mol% of base in acetonitrile (4.0 mL) at room temperature. Yields are those of the isolated products **3j–3r** after column chromatography. The diastereomeric ratio was determined by ¹H NMR and all of them were higher than 20 : 1.



Scheme 3 X-ray crystal structure of **3p**, gram-scale synthesis of **3a** and usage of the amide-substituted sulfonium salt.

In conclusion, we have developed a novel [4 + 1] annulation reaction between hydroxyphenyl-substituted *p*-QMs and sulfonium salts. This new protocol offers a facile and direct entry into a series of 2,3-dihydrobenzofurans. The procedure works under relatively mild and very simple experimental conditions and this is easily scalable to gram amounts.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- (a) D. C. Chauret, C. B. Bernard, J. T. Arnason, T. Durst, H. G. Krishnamurty, P. Sanchez-Vindas, N. Moreno, L. San Roman and L. Poveda, *J. Nat. Prod.*, 1996, **59**, 152; (b) K.-i. Kataoka, T. Shiota, T. Takeyasu, T. Minoshima, K. Watanabe, H. Tanaka, T. Mochizuki, K. Taneda, M. Ota, H. Tanabe and H. Yamaguchi, *J. Med. Chem.*, 1996, **39**, 1262; (c) C.-H. Chen, C.-Y. Shaw, C.-C. Chen and Y.-C. Tsai, *J. Nat. Prod.*, 2002, **65**, 740; (d) N. Beldjoudi, L. Mambu, M. Labaïed, P. Grellier, D. Ramanitrahasimbola, P. Rasoanaivo, M. T. Martin and F. Frappier, *J. Nat. Prod.*, 2003, **66**, 1447; (e) G. Q. Shi, J. F. Dropinski, Y. Zhang, C. Santini, S. P. Sahoo, J. P. Berger, K. L. MacNaul, G. Zhou, A. Agrawal, R. Alvaro, T.-q. Cai, M. Hernandez, S. D. Wright, D. E. Moller, J. V. Heck and P. T. Meinke, *J. Med. Chem.*, 2005, **48**, 5589; (f) R. J. Nevagi, S. N. Dighe and S. N. Dighe, *Eur. J. Med. Chem.*, 2015, **97**, 561;

- (g) A. Radadiya and A. Shah, *Eur. J. Med. Chem.*, 2015, **97**, 356.
- 2 For a leading review, see: (a) T. D. Sheppard, *J. Chem. Res.*, 2011, **35**, 377. For selected examples, see: (b) B. M. Trost, O. R. Thiel and H.-C. Tsui, *J. Am. Chem. Soc.*, 2003, **125**, 13155; (c) E. D. Coy B, L. Jovanovic and M. Sefkow, *Org. Lett.*, 2010, **12**, 1976; (d) J. Meng, T. Jiang, H. A. Bhatti, B. S. Siddiqui, S. Dixon and J. D. Kilburn, *Org. Biomol. Chem.*, 2010, **8**, 107; (e) D.-H. Wang and J.-Q. Yu, *J. Am. Chem. Soc.*, 2011, **133**, 5767; (f) X.-X. Sun, H.-H. Zhang, G.-H. Li, L. Meng and F. Shi, *Chem. Commun.*, 2016, **52**, 2968.
- 3 For selected reviews, see: (a) C. Zhu, Y. Ding and L.-W. Ye, *Org. Biomol. Chem.*, 2015, **13**, 2530; (b) J.-R. Chen, X.-Q. Hu, L.-Q. Lu and W.-J. Xiao, *Chem. Rev.*, 2015, **115**, 5301.
- 4 For an excellent review on *o*-QMs, see: (a) Z. Wang and J. Sun, *Synthesis*, 2015, 3629. For selected examples, see: (b) M.-W. Chen, L.-L. Cao, Z.-S. Ye, G.-F. Jiang and Y.-G. Zhou, *Chem. Commun.*, 2013, **49**, 1660; (c) B. Wu, M.-W. Chen, Z.-S. Ye, C.-B. Yu and Y.-G. Zhou, *Adv. Synth. Catal.*, 2014, **356**, 383; (d) N. Meisinger, L. Roiser, U. Monkowius, M. Himmelsbach, R. Robiette and M. Waser, *Chem. – Eur. J.*, 2017, **23**, 5137; (e) Q.-Q. Yang and W.-J. Xiao, *Eur. J. Org. Chem.*, 2017, 233.
- 5 (a) M. G. Peter, *Angew. Chem., Int. Ed. Engl.*, 1989, **28**, 555; (b) M. M. Toteva and J. P. Richard, *Adv. Phys. Org. Chem.*, 2011, **45**, 39; (c) A. Parra and M. Tortosa, *ChemCatChem*, 2015, **7**, 1524; (d) P. Chauhan, U. Kaya and D. Enders, *Adv. Synth. Catal.*, 2017, **359**, 888.
- 6 For selected examples of non-enantioselective reactions, see: (a) K. Gai, X. Fang, X. Li, J. Xu, X. Wu, A. Lin and H. Yao, *Chem. Commun.*, 2015, **51**, 15831; (b) A. López, A. Parra, C. Jarava-Barrera and M. Tortosa, *Chem. Commun.*, 2015, **51**, 17684; (c) B. T. Ramanjaneyulu, S. Mahesh and R. V. Anand, *Org. Lett.*, 2015, **17**, 3952; (d) Z. Yuan, W. Wei, A. Lin and H. Yao, *Org. Lett.*, 2016, **18**, 3370; (e) K.-X. Xie, Z.-P. Zhang and X. Li, *Org. Lett.*, 2017, **19**, 6708; (f) Z.-P. Zhang, N. Dong and X. Li, *Chem. Commun.*, 2017, **53**, 1301; (g) K. Zhao, Y. Zhi, A. Wang and D. Enders, *Synthesis*, 2018, **50**, 872; (h) N. Molleti and J. Y. Kang, *Org. Lett.*, 2017, **19**, 958.
- 7 (a) W.-D. Chu, L.-F. Zhang, X. Bao, X.-H. Zhao, C. Zeng, J.-Y. Du, G.-B. Zhang, F.-X. Wang, X.-Y. Ma and C.-A. Fan, *Angew. Chem., Int. Ed.*, 2013, **52**, 9229; (b) L. Caruana, F. Kniep, T. K. Johansen, P. H. Poulsen and K. A. Jørgensen, *J. Am. Chem. Soc.*, 2014, **136**, 15929; (c) C. Jarava-Barrera, A. Parra, A. López, F. Cruz-Acosta, D. Collado-Sanz, D. J. Cárdenas and M. Tortosa, *ACS Catal.*, 2015, **6**, 442; (d) Y. Lou, P. Cao, T. Jia, Y. Zhang, M. Wang and J. Liao, *Angew. Chem., Int. Ed.*, 2015, **54**, 12134; (e) Z. Wang, Y. F. Wong and J. Sun, *Angew. Chem., Int. Ed.*, 2015, **54**, 13711; (f) Y.-H. Deng, X.-Z. Zhang, K.-Y. Yu, X. Yan, J.-Y. Du, H. Huang and C.-A. Fan, *Chem. Commun.*, 2016, **52**, 4183; (g) N. Dong, Z.-P. Zhang, X.-S. Xue, X. Li and J.-P. Cheng, *Angew. Chem., Int. Ed.*, 2016, **55**, 1460; (h) F.-S. He, J.-H. Jin, Z.-T. Yang, X. Yu, J. S. Fossey and W.-P. Deng, *ACS Catal.*, 2016, **6**, 652; (i) X. Li, X. Xu, W. Wei, A. Lin and H. Yao, *Org. Lett.*, 2016, **18**, 428; (j) Y. F. Wong, Z. Wang and J. Sun, *Org. Biomol. Chem.*, 2016, **14**, 5751; (k) X.-Z. Zhang, Y.-H. Deng, X. Yan, K.-Y. Yu, F.-X. Wang, X.-Y. Ma and C.-A. Fan, *J. Org. Chem.*, 2016, **81**, 5655; (l) K. Zhao, Y. Zhi, A. Wang and D. Enders, *ACS Catal.*, 2016, **6**, 657.
- 8 (a) Z. Yuan, X. Fang, X. Li, J. Wu, H. Yao and A. Lin, *J. Org. Chem.*, 2015, **80**, 11123; (b) C. Ma, Y. Huang and Y. Zhao, *ACS Catal.*, 2016, **6**, 6408; (c) X.-Z. Zhang, J.-Y. Du, Y.-H. Deng, W.-D. Chu, X. Yan, K.-Y. Yu and C.-A. Fan, *J. Org. Chem.*, 2016, **81**, 2598; (d) L. Roiser and M. Waser, *Org. Lett.*, 2017, **19**, 2338; (e) Z. Yuan, K. Gai, Y. Wu, J. Wu, A. Lin and H. Yao, *Chem. Commun.*, 2017, **53**, 3485; (f) Z. Yuan, L. Liu, R. Pan, H. Yao and A. Lin, *J. Org. Chem.*, 2017, **82**, 8743; (g) X.-Z. Zhang, Y.-H. Deng, K.-J. Gan, X. Yan, K.-Y. Yu, F.-X. Wang and C.-A. Fan, *Org. Lett.*, 2017, **19**, 1752.
- 9 K. Zhao, Y. Zhi, T. Shu, A. Valkonen, K. Rissanen and D. Enders, *Angew. Chem., Int. Ed.*, 2016, **55**, 12104.
- 10 S. Liu, X.-C. Lan, K. Chen, W.-J. Hao, G. Li, S.-J. Tu and B. Jiang, *Org. Lett.*, 2017, **19**, 3831.
- 11 Z. Cao, G.-X. Zhou, C. Ma, K. Jiang and G.-J. Mei, *Synthesis*, 2017, **49**, DOI: 10.1055/s-0036-1591724.
- 12 During the preparation of this manuscript, two closely related papers appeared: (a) L. Liu, Z. Yuan, R. Pan, Y. Zeng, A. Lin, H. Yao and Y. Huang, *Org. Chem. Front.*, 2018, **5**, 623. In this excellent report the sulfonium salt containing an alkyl ester group gave the desired compound in low yield (37%). Our protocol is suitable for alkyl ester, amide and ketone sulfonium salts; (b) X.-M. Chen, K.-X. Xie, D.-F. Yue, X.-M. Zhang, X.-Y. Xu and W.-C. Yuan, *Tetrahedron*, 2018, **74**, 600.
- 13 CCDC 1814040.†