In an experiment with photoelectric effect, the stopping potential,

- is $\left(rac{1}{e}
 ight)$ times the maximum kinetic energy of the emitted photoelectrons
- increases with increase in the intensity of the incident light
- decreases with increase in the intensity of the incident light
- increases with increase in the wavelength of the incident light

Q 2 UV light of $4.13~\rm eV$ is incident on a photosensitive metal surface having work function $3.13~\rm eV$. The maximum kinetic energy of ejected photoelectrons will be:

- A 4.13 eV
- B 1 eV
- C 7.26 eV
- D 3.13 eV

A proton, an electron and an alpha particle have the same energies. Their de-Broglie wavelengths will be compared as:

$$oldsymbol{\mathbb{C}}$$
 $\lambda_{
m e} > \lambda_{lpha} > \lambda_{
m p}$

$$oldsymbol{ ext{D}} \lambda_{ ext{p}} < \lambda_{ ext{e}} < \lambda_{lpha}$$

In photoelectric experiment energy of $2.48~\rm eV$ irradiates a photo sensitive material. The stopping potential was measured to be $0.5~\rm V$. Work function of the photo sensitive material is :

- A 1.98 eV
- 📵 1.68 eV
- 2.48 eV
- D 0.5 eV

Which of the following statement is not true about stopping potential (V_0) ?

- It depends upon frequency of the incident light.
- It is 1/e times the maximum kinetic energy of electrons emitted.
- It increases with increase in intensity of the incident light.
- It depends on the nature of emitter material.

Given below are two statements: one is labelled as Assertion $\bf A$ and the other is labelled as Reason R.

Assertion A: Number of photons increases with increase in frequency of light.

Reason R: Maximum kinetic energy of emitted electrons increases with the frequency of incident radiation.

In the light of the above statements, choose the most appropriate answer from the options given below:

- $oldsymbol{A}$ is not correct but $oldsymbol{R}$ is correct.
- f A is correct but ${f R}$ is not correct.
- Both ${f A}$ and ${f R}$ are correct and ${f R}$ is the correct explanation of ${f A}$.
- Both ${f A}$ and ${f R}$ are correct and ${f R}$ is NOT the correct explanation of ${f A}$.

The de Broglie wavelengths of a proton and an α particle are λ and 2λ respectively. The ratio of the velocities of proton and α particle will be :

A 8:1

B 1:2

C 1:8

 \bigcirc 4:1

When a metal surface is illuminated by light of wavelength λ , the stopping potential is 8~V. When the same surface is illuminated by light of wavelength 3λ , stopping potential is 2~V. The threshold wavelength for this surface is:

- \bigcirc 3 λ
- **B** 9λ
- \bigcirc 5 λ
- \bigcirc 4.5 λ

When UV light of wavelength $300~\mathrm{nm}$ is incident on the metal surface having work function $2.13~\mathrm{eV}$, electron emission takes place. The stopping potential is :

(Given $hc = 1240 \, eV \, nm$)

- A 4 V
- B 2 V
- **O** 4.1 V
- D 1.5 V

The work function of a substance is $3.0\,\mathrm{eV}$. The longest wavelength of light that can cause the emission of photoelectrons from this substance is approximately;

- A 215 nm
- 400 nm
- **6** 414 nm
- D 200 nm

The work functions of Aluminium and Gold are $4.1~{\rm eV}$ and and $5.1~{\rm eV}$ respectively. The ratio of the slope of the stopping potential versus frequency plot for Gold to that of Aluminium is

- A 1.5
- B 1.24
- **C** 1
- **D** 2

The ratio of de-Broglie wavelength of an α particle and a proton accelerated from rest by the same potential is $\frac{1}{\sqrt{m}}$, the value of m is -

- A 2
- **B** 16
- **C** 8
- **D** 4

Q 13
The ratio of wavelengths of proton and deuteron accelerated by potential V_p and V_d is 1 : $\sqrt{2}.$ Then the ratio of V_p to V_d will be :

 \bigcirc $\sqrt{2}:1$

The light of two different frequencies whose photons have energies 3.8 eV and 1.4 eV respectively, illuminate a metallic surface whose work function is 0.6 eV successively. The ratio of maximum speeds of emitted electrons for the two frequencies respectively will be:

- A 1:1
- B 2:1
- **C** 4:1
- D 1:4

A moving proton and electron have the same de-Broglie wavelength. If K and P denote the K.E. and momentum respectively. Then choose the correct option:

- A $K_p < K_e$ and $P_p = P_e$
- \mathbb{B} $K_p = K_e$ and $P_p = P_e$
- \bigcirc $K_p < K_e$ an $P_p < P_e$
- \mathbb{D} K_p > K_e and P_p = P_e

In a photoelectric experiment, increasing the intensity of incident light:

- increases the number of photons incident and also increases the K.E. of the ejected electrons
- increases the frequency of photons incident and increases the K.E. of the ejected electrons
- increases the frequency of photons incident and the K.E. of the ejected electrons remains unchanged
- increases the number of photons incident and the K.E. of the ejected electrons remains unchanged

An electron moving with speed v and a photon moving with speed c, have same D-Broglie wavelength. The ratio of kinetic energy of electron to that of photon is:

$$\frac{3c}{v}$$

$$\frac{v}{3c}$$

$$\frac{v}{2c}$$

An α particle and a proton are accelerated from rest by a potential difference of 200V. After this, their de Broglie wavelengths are λ_{α} and $\lambda_{\rm p}$ respectively. The ratio $\frac{\lambda_p}{\lambda_{\alpha}}$ is :

- A 8
- B 2.8
- 7.8
- D 3.8

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Given below are two statements:

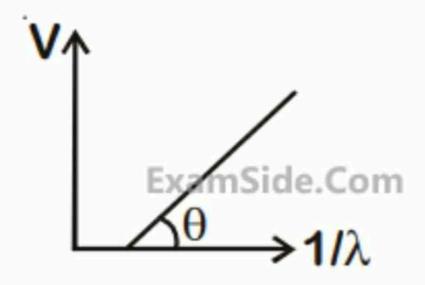
Statement I: Two photons having equal linear momenta have equal wavelengths.

Statement II: If the wavelength of photon is decreased, then the momentum and energy of a photon will also decrease.

In the light of the above statements, choose the correct answer from the options given below.

- A Statement I is false but Statement II is true
- Both Statement I and Statement II are false
- Both Statement I and Statement II are true
- Statement I is true but Statement II is false

In a photoelectric effect experiment, the graph of stopping potential V versus reciprocal of wavelength obtained is shown in the figure. As the intensity of incident radiation is increased:



- A Slope of the straight line get more steep
- B Graph does not change
- Straight line shifts to left
- Straight line shifts to right

A particle moving with kinetic energy E has de Broglie wavelength λ . If energy Δ E is added to its energy, the wavelength become λ /2. Value of Δ E, is :

- A E
- B 3E
- © 2E
- D 4E

Q 22 If a source of power 4kW produces 10^{20} photons/second, the radiation belongs to a part of the spectrum called

lacksquare X -rays

B ultraviolet rays

microwaves

 \bigcirc γ - rays

A Laser light of wavelength 660 nm is used to weld Retina detachment. If a Laser pulse of width 60 ms and power 0.5 kW is used the approximate number of photons in the pulse are:

[Take Planck's constant h = 6.62×10^{-34} Js]

- \triangle 10²⁰
- **B** 10¹⁸
- C 10²²
- D 10¹⁹

A particle A of mass m and initial velocity v collides with a particle B of mass m/2 which is at rest. The collision is head on, and elastic. The ratio of the de-Broglie wavelengths λ_A to λ_B after the collision is:

$$\frac{\lambda_A}{\lambda_B} = \frac{1}{3}$$

$$rac{\lambda_A}{\lambda_B}=2$$

$$\frac{\lambda_A}{\lambda_B} = \frac{2}{3}$$

$$rac{\lambda_A}{\lambda_B} = rac{1}{2}$$

An X-ray tube is operated at 1.24 million volt. The shortest wavelength of the produced photon will be:

m A 10 $^{-2}$ nm

 $^{\,}$ 10 $^{-1}$ nm

 $^{\circ}$ 10 $^{-3}$ nm

 $^{-4}\,{
m nm}$

A particle of mass 4M at rest disintegrates into two particles of mass M and 3M respectively having non zero velocities. The ratio of de-Broglie wavelength of particle of mass M to that of mass 3M will be:

- A 1:3
- B 3:1
- 1: $\sqrt{3}$
- 1:1

A particle is travelling 4 time as fast as an electron. Assuming the ratio of de-Broglie wavelength of a particle to that of electron is 2 : 1, the mass of the particle is :

- $\frac{1}{16}$ times the mass of e⁻
- B 8 times the mass of e
- 16 times the mass of e
- $\frac{1}{8}$ times the mass of e⁻

Two identical photocathodes receive the light of frequencies f_1 and f_2 respectively. If the velocities of the photo-electrons coming out are v_1 and v_2 respectively, then

$$igwedge v_1-v_2=\left[rac{2h}{m}(f_1-f_2)
ight]^{rac{1}{2}}$$

$$v_1^2 + v_2^2 = rac{2h}{m}[f_1 + f_2]$$

$$\bigcirc$$
 $v_1+v_2=\left[rac{2h}{m}(f_1+f_2)
ight]^{rac{1}{2}}$

$$v_1^2 - v_2^2 = rac{2h}{m}[f_1 - f_2]$$

The de-Broglie wavelength of a proton and α -particle are equal. The ratio of their velocities is :

A 4:2

B 4:3

C 4:1

D 1:4

A photoelectric surface is illuminated successively by monochromatic light of wavelengths λ and $\frac{\lambda}{2}$. If the maximum kinetic energy of the emitted photoelectrons in the second case is 3 times that in the first case, the work function of the surface is .

$$\frac{hc}{3\lambda}$$

$$\frac{hc}{2\lambda}$$

$$\frac{hc}{\lambda}$$

$$\frac{3 hc}{\lambda}$$