## e-folding

## **Definition**

In physics and other sciences, the e-folding time is the time it takes for an exponentially growing quantity to increase by a factor of e, which is approximately 2.71828. It is the base-e analog doubling time.

For example, if a population of bacteria is growing exponentially, the e-folding time would be the time it taks for the population to double. E-folding is also used to describe the time it takes for a radioactive substance to decay by half.

## **Formula**

The formua for the e-folding time is:

$$au = 1/k$$

## where:

- $\tau$  is the e-folding time
- k is the rate constant

The rate constant is a measure of how quickly a quantity changes over time.

For example, the rate constant for radioactive decay is the probability that a radioactive atom will decay in a given time interval.

e-folding 1

The e-folding time is a useful concept for understanding exponential growth and decay. For example, if a population of bacteria is growing exponentially, the e-folding time would be the time it takes for the population to double. E-folding time is also used to describe the time it takes for a radioactive substance to decay by half.

Here are some examples of how the e-folding time can be used:

- Radioactive decay: The e-folding time for radioactive decay is the time it takes for half of the radioactive atoms in a sample to decay.

  For example, the e-folding time for carbon-14 is 5,730 years. This means that half of the carbon-14 atoms in a sample will decay in 5,730 years, and half of the remaining atoms will decay in the next 5,730 years, and so on.
- Population growth: The e-folding time for population growth is the time it takes for a population to double.
   For example, the e-folding time for the human population is about 35 years. This means that the human population doubles every 35 years.
- Chemical reactions: The e-folding time for a chemical reaction is the time it takes for the concentration of reactants to decrease by a factor of e.

  For example, the e-folding time for the reaction between hydrogen and oxygen to form water is about 10 seconds. This means that the concentration of hydrogen and oxygen in a mixture will decrease by a factor of e every 10 seconds.

e-folding 2