

Structural, mechanical, thermodynamic, and electronic properties of thorium hydrides from first principles

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We perform first-principles calculations of the structural, electronic, mechanical, and thermodynamic properties of thorium hydrides (ThH₂ and Th₄H₁₅) based on the density functional theory with generalized gradient approximation. The equilibrium geometries, the total and partial densities of states, charge density, elastic constants, elastic moduli, Poisson's ratio, and phonon dispersion curves for these materials are systematically investigated and analyzed in comparison with experiments and previous calculations. These results show that our calculated equilibrium structural parameters are well consistent with experiments. The Th–H bonds in all thorium hydrides exhibit weak covalent character, but the ionic properties for ThH₂ and Th₄H₁₅ are different due to their different hydrogen concentration. It is found that while in ThH₂ about 1.5 electrons transfer from each Th atom to H, in Th₄H₁₅ the charge transfer from each Th atom is around 2.1 electrons. Our calculated phonon spectrum for the stable body-centered tetragonal phase of ThH₂ accords well with experiments. In addition we show that ThH₂ in the fluorite phase is mechanically and dynamically unstable.

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I. INTRODUCTION

Thorium is one kind of important nuclear materials and together with its compounds has been widely investigated both experimentally and theoretically. Among thorium compounds, thorium hydrides (ThH₂ and Th₄H₁₅) are metallic solids and have potential use for advanced nuclear fuels. In addition, Th₄H₁₅ has been reported to have superconductivity with transition temperatures $T_c \sim 8$ K¹ and has been considered as promising candidates for hydrogen storage since its large hydrogen-to-metal ratio². ThH₂ is not superconducting above 1 K³.

Despite the abundant research on thorium hydrides, relatively little is known regarding their chemical bonding, mechanical properties, and phonon dispersion. Only the optical phonon density of states of ThH₂ and Th₄H₁₅ was measured through inelastic neutron scattering⁴ and the bulk modulus of ThH₂ was calculated by linear muffin-tin orbital (LMTO) method⁵. Until now the elastic properties, which relate to various fundamental solid-state properties such as interatomic potentials, equation of state, phonon spectra, and thermodynamical properties, are unknown for ThH₂ and Th₄H₁₅. In addition, although the electronic properties as well as the chemical bonding in thorium hydrides have been calculated recently by Shein *et al.*² through the full-potential LAPW (FLAPW) method, the study of the bonding nature of Th–H bond involving its mixed ionic/covalent character is still lacking. These facts, as a consequence, inhibit deep understanding of thorium hydrides. Motivated by these observations, in this paper, we present a first-principles study by calculating the structural, electronic, mechanical, and thermodynamic properties of thorium

hydrides. Also, the stability of the metastable phase of ThH₂ (fluorite structure with space group $Fm\bar{3}m$) is analyzed and our calculated results show that the fluorite-type ThH₂ is mechanically unstable. We perform the Bader analysis^{6,7} of thorium hydrides and find that about 1.5 (2.1) electrons transfer from each Th atom to H for ThH₂ (Th₄H₁₅).

II. COMPUTATIONAL METHODS

Our total energy calculations are carried out by employing the plane-wave basis pseudopotential method as implemented in Vienna *ab initio* simulation package (VASP)⁸. The exchange and correlation effects are described by the density functional theory (DFT) within generalized gradient approximation (GGA)⁹. The projected augmented wave (PAW) method of Blöchl¹⁰ is implemented in VASP with the frozen-core approximation. The thorium 6s²7s²6p⁶6d¹5f¹ and the hydrogen 1s¹ electrons are treated as valence electrons. Note that although the 5f states are empty in elemental Th, this level turns to evolve into a hybridization with the hydrogen orbitals both in the valence band and the conduction band, as well as to prominently contribute to the conduction band (see Fig. 3 below). 9×9×9 and 5×5×5 Monkhorst-Pack¹¹ k point-meshes in the full wedge of the Brillouin zone are used for ThH₂ and Th₄H₁₅, respectively. Electron wave function is expanded in plane waves up to a cutoff energy of 450 eV, and all atoms are fully relaxed until the Hellmann-Feynman forces become less than 0.02 eV/Å.

In present work, the theoretical equilibrium volume,