Table 1. ¹H NMR Data

Table 1. ¹ H NMR Data											
	proton	mult	CDCl ₃	(CD ₃) ₂ CO	(CD ₃) ₂ SO	C ₆ D ₆	CD ₃ CN	CD ₃ OD	D ₂ O		
solvent residual peak			7.26	2.05	2.50	7.16	1.94	3.31	4.79		
H_2O		S	1.56	2.84^{a}	3.33^{a}	0.40	2.13	4.87			
acetic acid	CH_3	S	2.10	1.96	1.91	1.55	1.96	1.99	2.08		
acetone	CH_3	S	2.17	2.09	2.09	1.55	2.08	2.15	2.22		
acetonitrile	CH_3	S	2.10	2.05	2.07	1.55	1.96	2.03	2.06		
benzene	CH	S	7.36	7.36	7.37	7.15	7.37	7.33			
tert-butyl alcohol	CH_3	S	1.28	1.18	1.11	1.05	1.16	1.40	1.24		
	OH^c	S			4.19	1.55	2.18				
tert-butyl methyl ether	CCH_3	S	1.19	1.13	1.11	1.07	1.14	1.15	1.21		
	OCH_3	S	3.22	3.13	3.08	3.04	3.13	3.20	3.22		
BHT^b	ArH	S	6.98	6.96	6.87	7.05	6.97	6.92			
	OH ^c	S	5.01	2.22	6.65	4.79	5.20	2.21			
	ArCH ₃	S	2.27	2.22	2.18	2.24	2.22	2.21			
chloroform	ArC(CH ₃) ₃ CH	S	1.43	1.41 8.02	1.36 8.32	1.38 6.15	1.39 7.58	1.40 7.90			
cyclohexane	CH CH ₂	s s	7.26 1.43	1.43	1.40	1.40	7.36 1.44	1.45			
1,2-dichloroethane	CH_2 CH_2	S	3.73	3.87	3.90	2.90	3.81	3.78			
dichloromethane	CH_2 CH_2	S	5.73	5.63	5.76	4.27	5.44	5.49			
diethyl ether	CH ₃	t, 7	1.21	1.11	1.09	1.11	1.12	1.18	1.17		
diethyl ethel	CH ₂	q, 7	3.48	3.41	3.38	3.26	3.42	3.49	3.56		
diglyme	CH ₂	m, /	3.65	3.56	3.51	3.46	3.53	3.61	3.67		
digiyine	CH ₂	m	3.57	3.47	3.38	3.34	3.45	3.58	3.61		
	OCH ₃	S	3.39	3.28	3.24	3.11	3.29	3.35	3.37		
1,2-dimethoxyethane	CH ₃	s	3.40	3.28	3.24	3.12	3.28	3.35	3.37		
-,	CH ₂	S	3.55	3.46	3.43	3.33	3.45	3.52	3.60		
dimethylacetamide	CH ₃ CO	S	2.09	1.97	1.96	1.60	1.97	2.07	2.08		
,	NCH ₃	s	3.02	3.00	2.94	2.57	2.96	3.31	3.06		
	NCH_3	s	2.94	2.83	2.78	2.05	2.83	2.92	2.90		
dimethylformamide	CH	S	8.02	7.96	7.95	7.63	7.92	7.97	7.92		
	CH_3	S	2.96	2.94	2.89	2.36	2.89	2.99	3.01		
	CH_3	S	2.88	2.78	2.73	1.86	2.77	2.86	2.85		
dimethyl sulfoxide	CH_3	S	2.62	2.52	2.54	1.68	2.50	2.65	2.71		
dioxane	CH_2	S	3.71	3.59	3.57	3.35	3.60	3.66	3.75		
ethanol	CH_3	t, 7	1.25	1.12	1.06	0.96	1.12	1.19	1.17		
	CH_2	$q, 7^d$	3.72	3.57	3.44	3.34	3.54	3.60	3.65		
	OH	$\mathbf{s}^{c,d}$	1.32	3.39	4.63	4.65	2.47	201	205		
ethyl acetate	CH ₃ CO	s	2.05	1.97	1.99	1.65	1.97	2.01	2.07		
	CH_2CH_3	q,7	4.12	4.05	4.03	3.89	4.06	4.09	4.14		
-4h14h1 l4	CH_2CH_3	t, 7	1.26	1.20	1.17	0.92	1.20	1.24	1.24		
ethyl methyl ketone	CH₃CO	s 7	2.14 2.46	2.07 2.45	2.07 2.43	1.58 1.81	2.06 2.43	2.12 2.50	2.19 3.18		
	CH_2CH_3 CH_2CH_3	q,7 t,7	1.06	0.96	0.91	0.85	0.96	1.01	1.26		
ethylene glycol	CH CH	s ^e	3.76	3.28	3.34	3.41	3.51	3.59	3.65		
"grease" f	CH ₃	m	0.86	0.87	3.34	0.92	0.86	0.88	3.03		
grease	CH ₂	br s	1.26	1.29		1.36	1.27	1.29			
n-hexane	CH ₃	t	0.88	0.88	0.86	0.89	0.89	0.90			
	CH_2	m	1.26	1.28	1.25	1.24	1.28	1.29			
$HMPA^g$	CH ₃	d, 9.5	2.65	2.59	2.53	2.40	2.57	2.64	2.61		
methanol	CH ₃	S^h	3.49	3.31	3.16	3.07	3.28	3.34	3.34		
	OH	$s^{c,h}$	1.09	3.12	4.01		2.16				
nitromethane	CH_3	S	4.33	4.43	4.42	2.94	4.31	4.34	4.40		
n-pentane	CH_3	t, 7	0.88	0.88	0.86	0.87	0.89	0.90			
	CH_2	m	1.27	1.27	1.27	1.23	1.29	1.29			
2-propanol	CH_3	d, 6	1.22	1.10	1.04	0.95	1.09	1.50	1.17		
	CH	sep, 6	4.04	3.90	3.78	3.67	3.87	3.92	4.02		
pyridine	CH(2)	m	8.62	8.58	8.58	8.53	8.57	8.53	8.52		
	CH(3)	m	7.29	7.35	7.39	6.66	7.33	7.44	7.45		
	CH(4)	m	7.68	7.76	7.79	6.98	7.73	7.85	7.87		
silicone greasei	CH ₃	S	0.07	0.13	4.76	0.29	0.08	0.10	4.00		
tetrahydrofuran	CH ₂	m	1.85	1.79	1.76	1.40	1.80	1.87	1.88		
4 - 1	CH ₂ O	m	3.76	3.63	3.60	3.57	3.64	3.71	3.74		
toluene	CH ₃	S	2.36	2.32	2.30	2.11	2.33	2.32			
	CH(o/p)	m	7.17	7.1 - 7.2	7.18	7.02	7.1 - 7.3	7.16			
triethylamine	CH(m) CH_3	m t ,7	7.25 1.03	7.1 - 7.2 0.96	7.25 0.93	7.13 0.96	7.1-7.3 0.96	7.16 1.05	0.99		
ti ictii y iaiii ille	CH ₃ CH ₂	ι, <i>τ</i> q, 7	2.53	2.45	2.43	2.40	2.45	2.58	2.57		
	C112	q, /	4.55	4.73	4.→3	2.40	4.73	2.50	4.51		

^a In these solvents the intermolecular rate of exchange is slow enough that a peak due to HDO is usually also observed; it appears at 2.81 and 3.30 ppm in acetone and DMSO, respectively. In the former solvent, it is often seen as a 1:1:1 triplet, with ${}^2J_{\rm H,D}=1$ Hz. ^b 2,6-Dimethyl-4-tert-butylphenol. ^c The signals from exchangeable protons were not always identified. ^d In some cases (see note a), the coupling interaction between the CH₂ and the OH protons may be observed $(J=5\,{\rm Hz})$. ^e In CD₃CN, the OH proton was seen as a multiplet at δ 2.69, and extra coupling was also apparent on the methylene peak. ^f Long-chain, linear aliphatic hydrocarbons. Their solubility in DMSO was too low to give visible peaks. ^g Hexamethylphosphoramide. ^h In some cases (see notes a, d), the coupling interaction between the CH₃ and the OH protons may be observed $(J=5.5\,{\rm Hz})$. ⁱ Poly(dimethylsiloxane). Its solubility in DMSO was too low to give visible peaks.

show their degree of variability. Occasionally, in order to distinguish between peaks whose assignment was

ambiguous, a further $1-2 \mu L$ of a specific substrate were added and the spectra run again.

Table 2. ¹³C NMR Data^a

Table 2. ¹³ C NMR Data ^a									
		CDCl ₃	(CD ₃) ₂ CO	(CD ₃) ₂ SO	C ₆ D ₆	CD ₃ CN	CD ₃ OD	D ₂ O	
solvent signals		77.16 ± 0.06	29.84 ± 0.01 206.26 ± 0.13	39.52 ± 0.06	128.06 ± 0.02	$1.32 \pm 0.02 \\ 118.26 \pm 0.02$	49.00±0.01		
acetic acid	CO CH ₃	175.99 20.81	172.31 20.51	171.93 20.95	175.82 20.37	173.21 20.73	175.11 20.56	177.21 21.03	
acetone	CO	207.07	205.87	206.31	204.43	207.43	209.67	215.94	
	CH_3	30.92	30.60	30.56	30.14	30.91	30.67	30.89	
acetonitrile	CN CH_3	116.43 1.89	117.60 1.12	117.91 1.03	116.02 0.20	118.26 1.79	118.06 0.85	119.68 1.47	
benzene	CH ₃	128.37	129.15	128.30	128.62	129.32	129.34	1.4/	
tert-butyl alcohol	C	69.15	68.13	66.88	68.19	68.74	69.40	70.36	
tout butul mother other	CH ₃	31.25	30.72	30.38	30.47	30.68	30.91	30.29	
tert-butyl methyl ether	OCH ₃ C	49.45 72.87	49.35 72.81	48.70 72.04	49.19 72.40	49.52 73.17	49.66 74.32	49.37 75.62	
	CCH_3	26.99	27.24	26.79	27.09	27.28	27.22	26.60	
BHT	C(1)	151.55	152.51	151.47	152.05	152.42	152.85		
	C(2) CH(3)	135.87 125.55	138.19 129.05	139.12 127.97	136.08 128.52	138.13 129.61	139.09 129.49		
	C(4)	128.27	126.03	124.85	125.83	126.38	126.11		
	CH ₃ Ar	21.20	21.31	20.97	21.40	21.23	21.38		
	<i>C</i> H₃C C	30.33 34.25	31.61 35.00	31.25 34.33	31.34 34.35	31.50 35.05	31.15 35.36		
chloroform	СН	77.36	79.19	79.16	77.79	79.17	79.44		
cyclohexane	CH_2	26.94	27.51	26.33	27.23	27.63	27.96		
1,2-dichloroethane dichloromethane	CH ₂	43.50 53.52	45.25 54.95	45.02 54.84	43.59 53.46	45.54 55.32	45.11 54.78		
diethyl ether	CH ₂ CH ₃	15.20	15.78	15.12	15.46	15.63	15.46	14.77	
,	CH_2	65.91	66.12	62.05	65.94	66.32	66.88	66.42	
diglyme	CH ₃	59.01	58.77	57.98	58.66	58.90	59.06	58.67	
	CH ₂ CH ₂	70.51 71.90	71.03 72.63	69.54 71.25	70.87 72.35	70.99 72.63	71.33 72.92	70.05 71.63	
1,2-dimethoxyethane	CH ₃	59.08	58.45	58.01	58.68	58.89	59.06	58.67	
	CH_2	71.84	72.47	17.07	72.21	72.47	72.72	71.49	
dimethylacetamide	CH ₃ CO	21.53 171.07	21.51 170.61	21.29 169.54	21.16 169.95	21.76 171.31	21.32 173.32	21.09 174.57	
	NCH ₃	35.28	34.89	37.38	34.67	35.17	35.50	35.03	
	NCH_3	38.13	37.92	34.42	37.03	38.26	38.43	38.76	
dimethylformamide	CH	162.62 36.50	162.79 36.15	162.29	162.13 35.25	163.31 36.57	164.73 36.89	165.53 37.54	
	CH_3 CH_3	31.45	31.03	35.73 30.73	30.72	31.32	31.61	32.03	
dimethyl sulfoxide	CH_3	40.76	41.23	40.45	40.03	41.31	40.45	39.39	
dioxane	CH ₂	67.14	67.60	66.36	67.16	67.72	68.11	67.19	
ethanol	CH ₃ CH ₂	18.41 58.28	18.89 57.72	18.51 56.07	18.72 57.86	18.80 57.96	18.40 58.26	17.47 58.05	
ethyl acetate	CH_3CO	21.04	20.83	20.68	20.56	21.16	20.88	21.15	
	CO	171.36	170.96	170.31	170.44	171.68	172.89	175.26	
	CH ₂ CH ₃	60.49 14.19	60.56 14.50	59.74 14.40	60.21 14.19	60.98 14.54	61.50 14.49	62.32 13.92	
ethyl methyl ketone	CH_3CO	29.49	29.30	29.26	28.56	29.60	29.39	29.49	
	CO	209.56	208.30	208.72	206.55	209.88	212.16	218.43	
	CH ₂ CH ₃ CH ₂ CH ₃	36.89 7.86	36.75 8.03	35.83 7.61	36.36 7.91	37.09 8.14	37.34 8.09	37.27 7.87	
ethylene glycol	CH ₂	63.79	64.26	62.76	64.34	64.22	64.30	63.17	
"grease"	CH ₂	29.76	30.73	29.20	30.21	30.86	31.29		
n-hexane	CH_3 $CH_2(2)$	14.14 22.70	14.34 23.28	13.88 22.05	14.32 23.04	14.43 23.40	14.45 23.68		
	$CH_2(3)$	31.64	32.30	30.95	31.96	32.36	32.73		
$HMPA^b$	CH ₃	36.87	37.04	36.42	36.88	37.10	37.00	36.46	
methanol nitromethane	CH ₃ CH ₃	50.41 62.50	49.77 63.21	48.59 63.28	49.97 61.16	49.90 63.66	49.86 63.08	49.50° 63.22	
n-pentane	CH ₃	14.08	14.29	13.28	14.25	14.37	14.39	03.22	
•	$CH_2(2)$	22.38	22.98	21.70	22.72	23.08	23.38		
2	$CH_2(3)$	34.16 25.14	34.83 25.67	33.48 25.43	34.45 25.18	34.89 25.55	35.30 25.27	24.38	
2-propanol	CH ₃ CH	64.50	63.85	64.92	64.23	64.30	64.71	64.88	
pyridine	CH(2)	149.90	150.67	149.58	150.27	150.76	150.07	149.18	
	CH(3)	123.75 135.96	124.57	123.84	123.58 135.28	127.76	125.53 138.35	125.12	
silicone grease	CH(4) CH ₃	1.04	136.56 1.40	136.05	1.38	136.89	2.10	138.27	
tetrahydrofuran	CH_2	25.62	26.15	25.14	25.72	26.27	26.48	25.67	
4-1	CH ₂ O	67.97	68.07	67.03	67.80	68.33	68.83	68.68	
toluene	CH_3 C(i)	21.46 137.89	21.46 138.48	20.99 137.35	21.10 137.91	21.50 138.90	21.50 138.85		
	CH(o)	129.07	129.76	128.88	129.33	129.94	129.91		
	CH(m)	128.26	129.03	128.18	128.56	129.23	129.20		
triethylamine	CH(p) CH_3	125.33 11.61	126.12 12.49	125.29 11.74	125.68 12.35	126.28 12.38	126.29 11.09	9.07	
	CH ₂	46.25	47.07	45.74	46.77	47.10	46.96	47.19	

 $[^]a$ See footnotes for Table 1. b $^2J_{PC} = 3$ Hz. c Reference material; see text.