

# X\_ray

February 19, 2023

## 1 X-Ray Diffraction

```
[ ]: import numpy as np
import matplotlib.pyplot as plt
import scipy as sp
import scipy.optimize as opt
import scipy.constants as const

title_size = 20
axis_size = 16
```

### 1.1 Pre-lab Questions

1.a) Read labscrip and relevant course material. Elements used from mathematical techniques are: scipy curve\_fit and chi squared test.

1.b) From the equation  $m\lambda = 2d \sin \theta$  from this for a higher wavelength the angle of diffraction will be greater than for a lower wavelength. As the energy of the K-alpha is lower as it comes from the  $n = 2$  to the  $n = 1$  shell, the wavelength will be higher and the angle of diffraction will be greater then that of the K-beta which comes from the  $n = 3$  to the  $n = 1$  shell.

1.c) Calculate the Interatomic spacing of a LiF cubic crystal using  $d = \frac{a}{\sqrt{3}}$  where a is the lattice constant. First calculate the lattice constant from the mass and the density of the crystal.

$$\rho = 2.64g/cm^3$$

$$m_{tot} = 25.94u$$

$$m_{tot} = \frac{25.94}{6.022 \times 10^{23}} = 4.308 \times 10^{-23}g$$

$$Volume = \frac{m_{tot}}{\rho} = \frac{4.308 \times 10^{-23}}{2.64} = 1.632 \times 10^{-23}cm^3$$

$$a = \sqrt[3]{Volume} = \sqrt[3]{1.632 \times 10^{-23}} = 2.536 \times 10^{-8}cm$$

$$d = \frac{a}{\sqrt{3}} = \frac{2.536 \times 10^{-8}}{\sqrt{3}} = 1.46 \times 10^{-8}cm$$

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