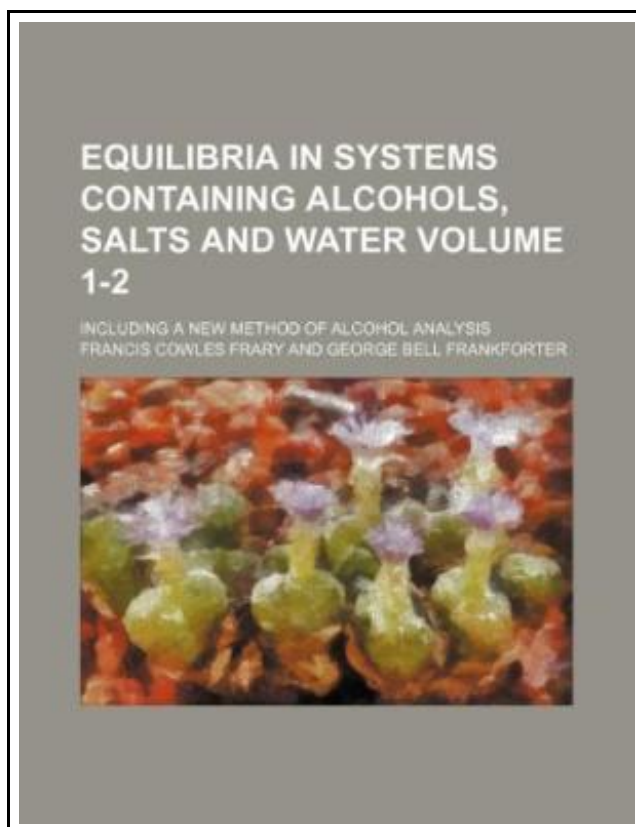


## Equilibria in Systems Containing Alcohols, Salts and Water Volume 1-2; Including a New Method of Alcohol Analysis



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*This publication is definitely not simple to begin on studying but really exciting to read. It is actually rally fascinating through reading time. Your life span will be enhance when you complete looking at this publication.*

*(Laurence Littel)*

## EQUILIBRIA IN SYSTEMS CONTAINING ALCOHOLS, SALTS AND WATER VOLUME 1-2; INCLUDING A NEW METHOD OF ALCOHOL ANALYSIS



Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1912 Excerpt: .be at a point where the two critical temperatures coincide. As the above table contains four variables, the relations it contains cannot be plotted as they stand. One variable is easily eliminated, for purposes of plotting, by expressing the table as above, and then plotting the composition of the solvent as one variable. If the percentage of alcohol be plotted as increasing from left to right, the percentage of water increases in a corresponding way from right to left, and both are represented by a single point. This simplification would make it possible to plot the effect of a change of temperature upon the solubility of the alcohol in a given salt solution as in figure 17, if we could reduce all the values in the table to the same salt G. C<sub>3</sub>H<sub>7</sub>OH/tr /Off Figure 19. Upper and Lower Critical Solution Temperatures in the System: Sodium Chloride. Normal Propyl Alcohol. Water, at a Constant Salt Content. content. All that we need to know for this purpose is the slope of the binodal curve at this point, (determination 7, table 12) but unfortunately this is not easy to determine accurately, as it is so steep that a small error will make a large difference in the results. By comparison with the next lower value (salt 3.686, alcohol 19.69 gm. per 100 gm. solvent) it would appear that over this range 0.406 gm. salt per 100 gm. solvent make a change of 33.06 per cent in the alcohol,...



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