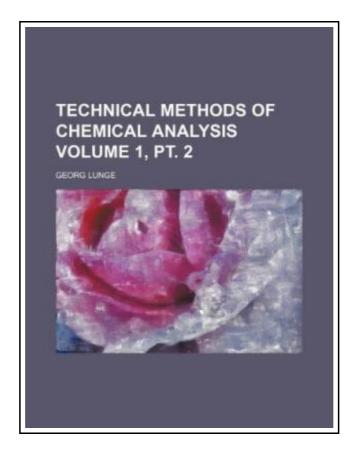
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1908 Excerpt: .acquires a faint yellow colour; if it is large in amount, the residue becomes brown or black. If the organic matter is mainly of vegetable origin, only a faint, non-characteristic smell is noticeable on decomposition, whereas nitrogenous organic matter, if present in considerable quantity, imparts a smell of singed horn to the fumes. Alkalinity. Recently collected spring or well water almost always shows a weakly acid reaction, due to the presence of free carbonic acid. Such water remains colourless on addition of a few drops of alcoholic phenolphthalein solution and a drop of lime water. If methyl orange is used as an indicator, most natural waters are apparently alkaline, for it is necessary to add a certain amount of hydrochloric acid in order to obtain a definite acid reaction; when this point is reached, all bicarbonates present are just converted into chlorides. If an indicator is used which is sensitive to free carbonic acid, such as phenolphthalein, alizarin, or litmus, this point can only be obtained by vigorously boiling the water during the addition of the hydrochloric acid, and thus driving out the liberated carbon dioxide. The alkalinity of natural waters is to be regarded from this consideration. 1 Mitteilungen d. k. Gesundheitamtes, I., 1881. According to Lungel the alkalinity is more easily determined by using methyl orange as indicator, as it is then unnecessary to boil, and the use of glass vessels is permissible. Two to three drops of aqueous methyl orange solution (1 to...



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