

Predicting protein-protein dissociation rates from spontaneous association trajectories

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PRESENTATION OUTLINE

1

Data collection

- a** setting up the simulation box
- b** running molecular dynamics
- c** assessing the quality of the simulations

2

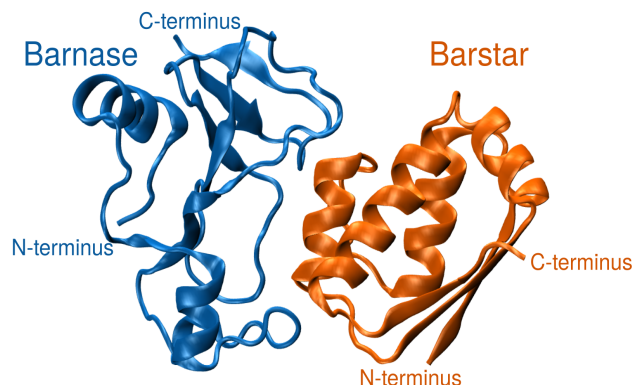
Data analysis

- a** finding a reaction coordinate
- b** building a stochastic model
- c** getting physico-chemical properties

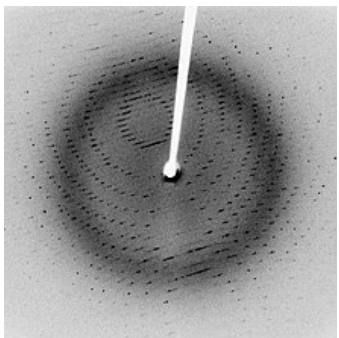
SETTING UP THE SIMULATION BOX

Our system

Ribonuclease **Barnase** and its inhibitor **Barstar**

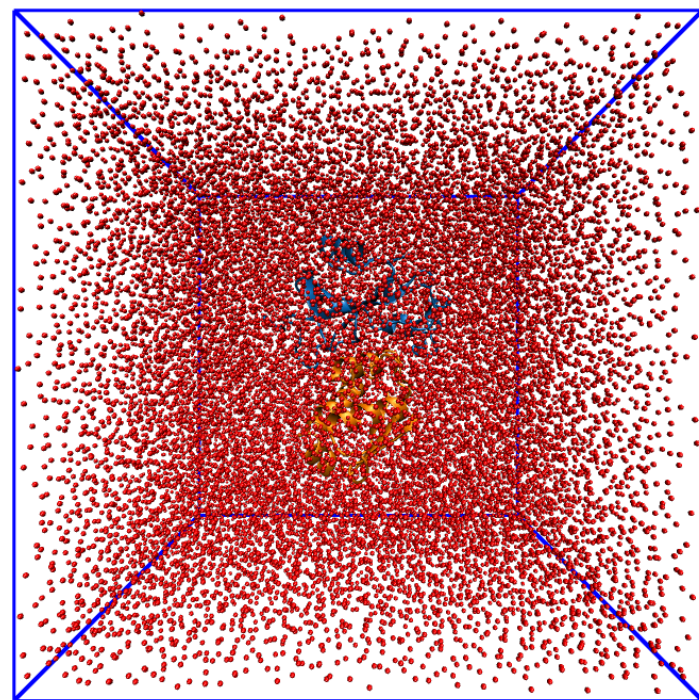
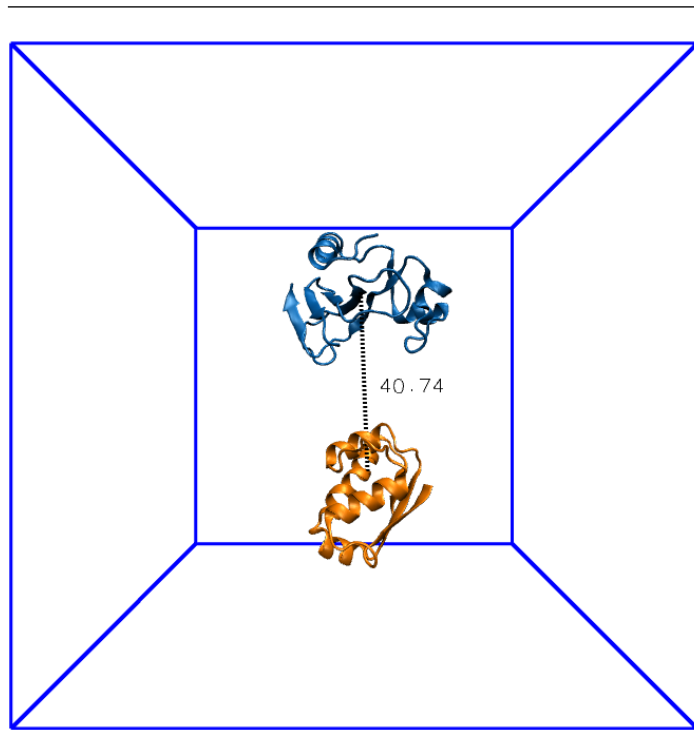


1 Get initial structures from experimental data



RCSB **PDB**
PROTEIN DATA BANK

2 Place the proteins far apart and solvate
9 nm

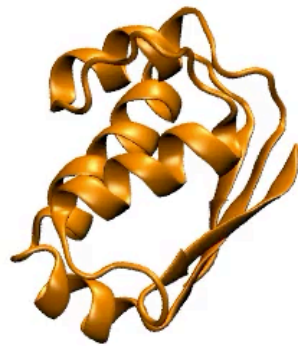
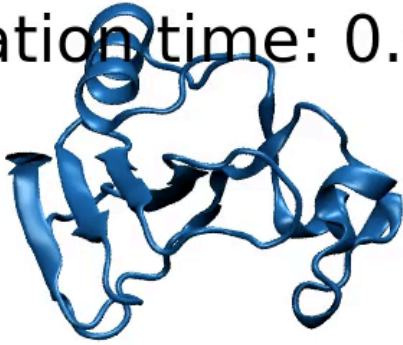


3 Equilibrate with thermostat & barostat

4 Run molecular dynamics

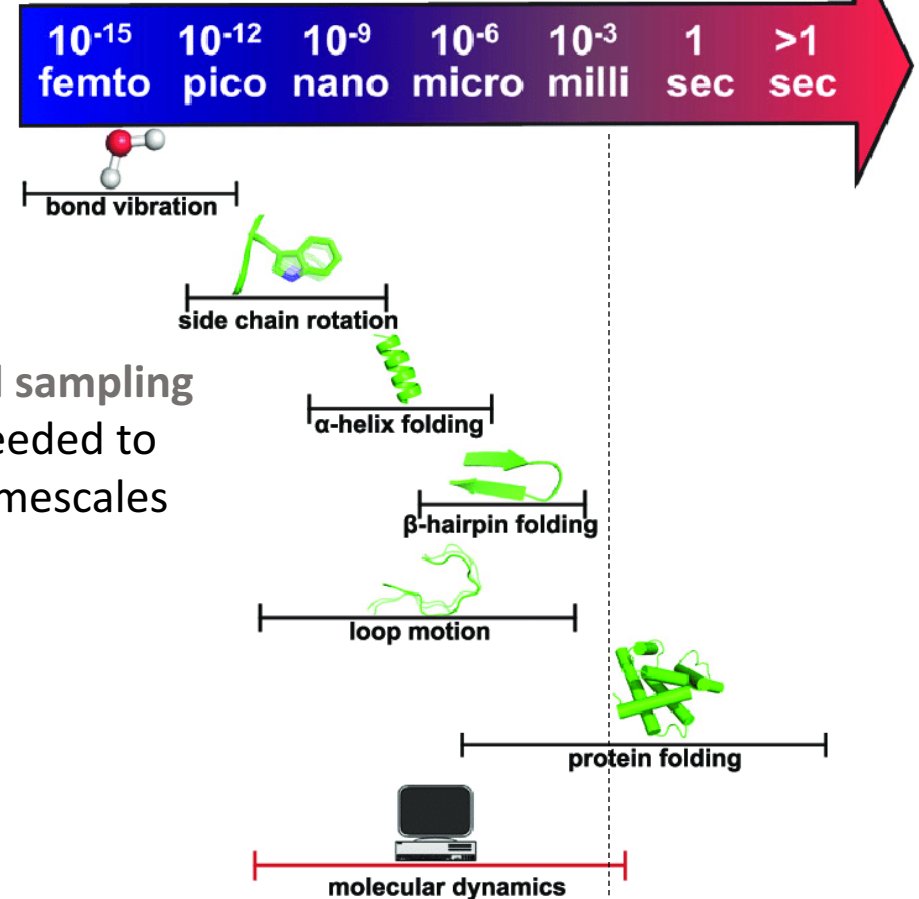
RUNNING MOLECULAR DYNAMICS

Simulation time: 0.0 ns



Computer time: 0 hours

Timescale barrier



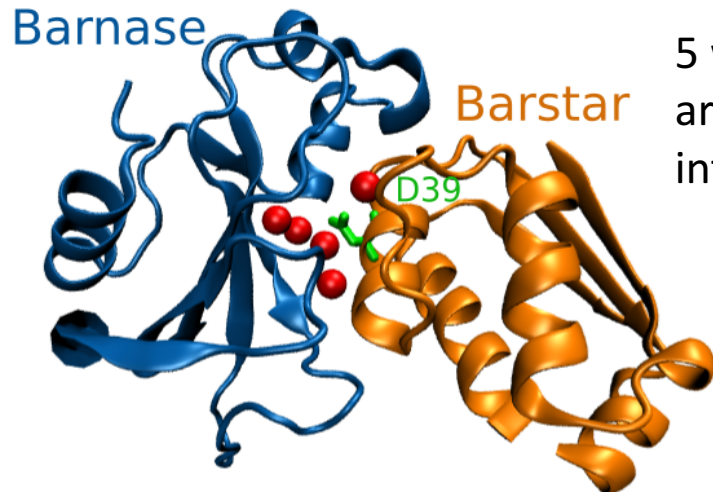
Usually **enhanced sampling** techniques are needed to access relevant timescales

ASSESSING THE QUALITY OF THE SIMULATIONS

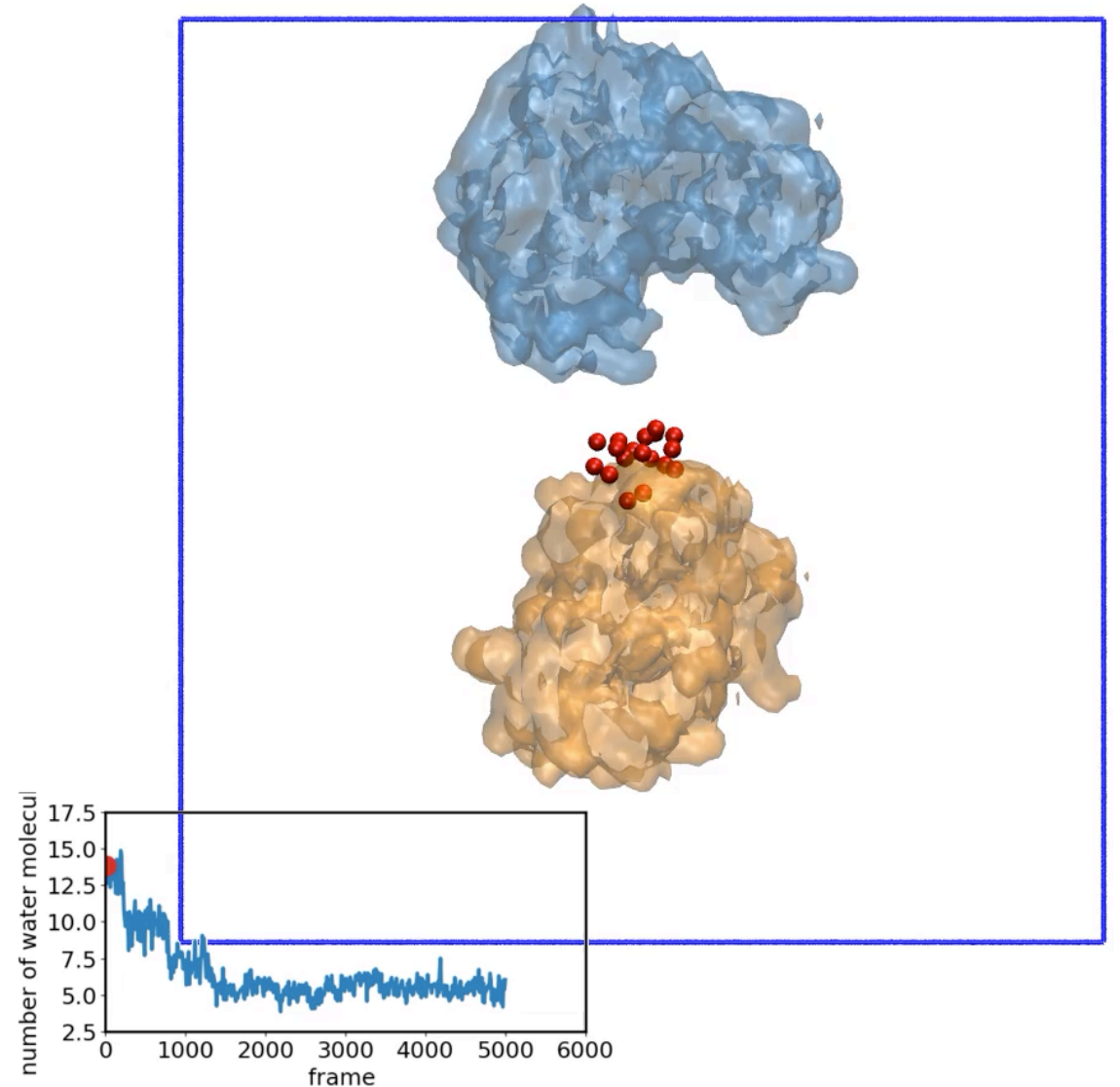
The quality of the model depends on the quality of the data

Is the transition captured by the simulations?

Is the end structure the experimentally determined bound state?



5 water molecules
are buried inside the
interface



THE REACTION COORDINATE

Atomic coordinates of 100 000 atoms in 3 dimensions

→ $3 \times 100\,000$ degrees of freedom

Is there a way to reduce the **high dimensionality** all while keeping the system's **accuracy**?

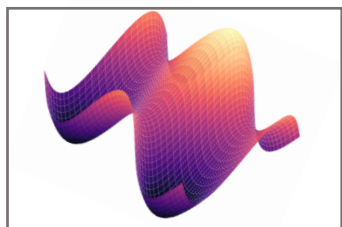
Use **reaction coordinates (RC)**

Order parameter tracking
progress of a reaction

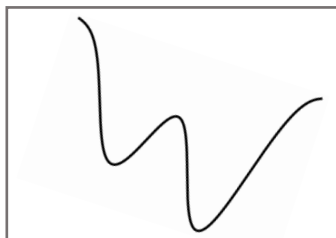
Generic function of
atomic coordinates

$3N$ dimensions

1 dimension



RC



How can we find such reaction coordinates?

Open problem

No general recipe

Our approach

A variational principle: the optimal coordinate q yields the minimal kinetic rate

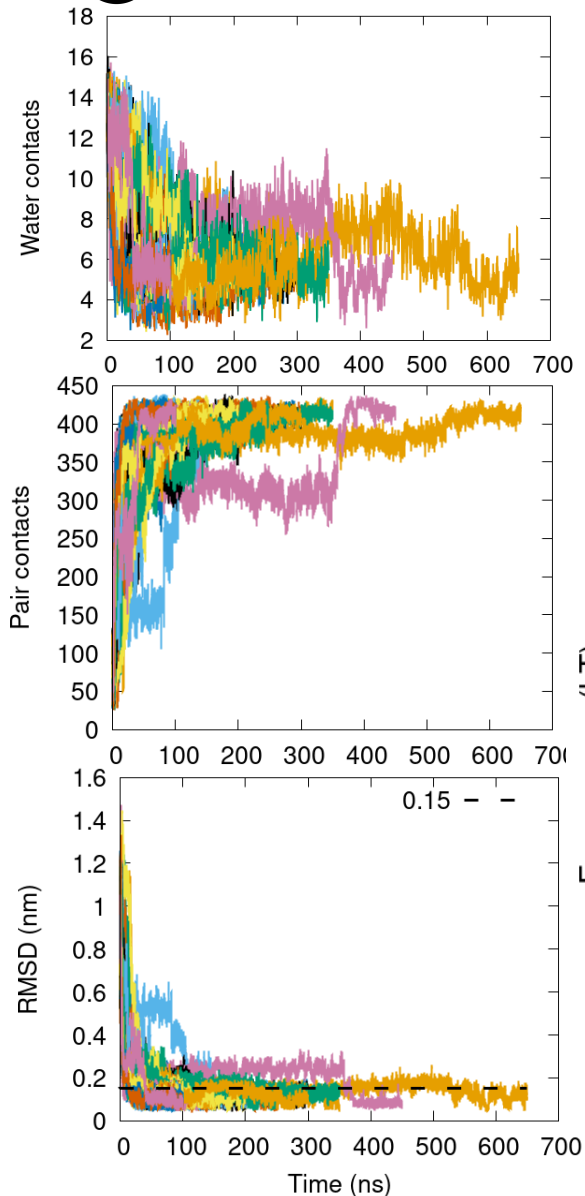
$$k_{A \rightarrow B} \leq k_{A \rightarrow B}^{model}$$

Zhang, Hartmann, and Schütte. *Faraday discussions* (2016)

Mouaffac, Palacio-Rodriguez & Pietrucci , *JCTC* (2023)

GETTING PHYSICO-CHEMICAL PROPERTIES

1 Reaction coordinates



2 Overdamped Langevin equation

$$\dot{q} = -\beta D(q) \frac{\partial F(q)}{\partial q} + \frac{\partial D(q)}{\partial q} + \sqrt{2D(q)} \eta(t)$$

Noise (random fluctuations)

It only depends on **one** variable q : the **reaction coordinate**

Free energy (thermodynamics)

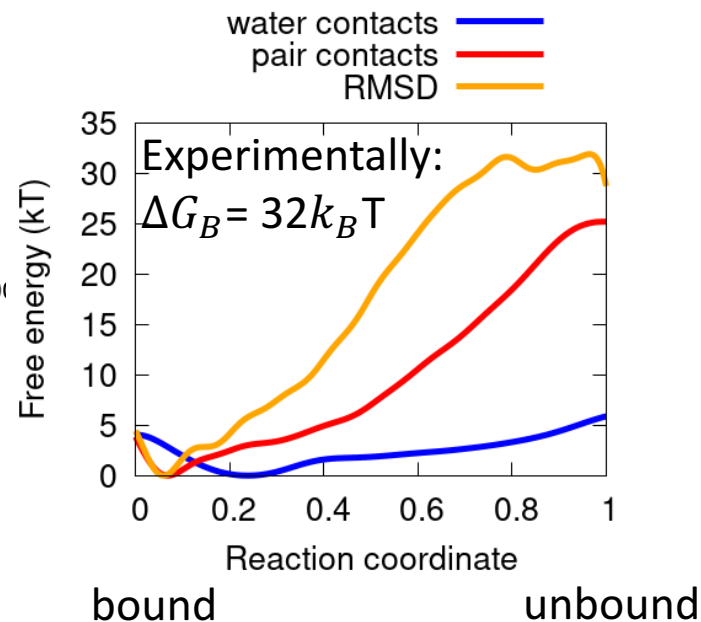
Diffusion profile (kinetics)

3 Dissociation rate

$$\frac{1}{k_{A \rightarrow B}^{model}} = \int_{q_0}^b dy \frac{e^{\beta F(y)}}{D(y)} \int_a^y dz e^{-\beta F(z)}$$

Method	$k_{A-B}^{model} (s^{-1})$
Experimental	8×10^{-6}
Interface RMSD	5.81×10^{-6}
Pair contacts	2.73×10^{-3}
Water contacts	1.60×10^6

Rate is minimized



CONCLUSIONS

We need ~ 1 day of simulation time to calculate dissociation rates

This is impossible to compute using classical molecular dynamics

Using our framework we are able to predict dissociation rates from classical molecular dynamics