

NHC-Catalyzed Reaction of Carboxylic Acids Using Allene Ketones as Substrates and Activating Reagents

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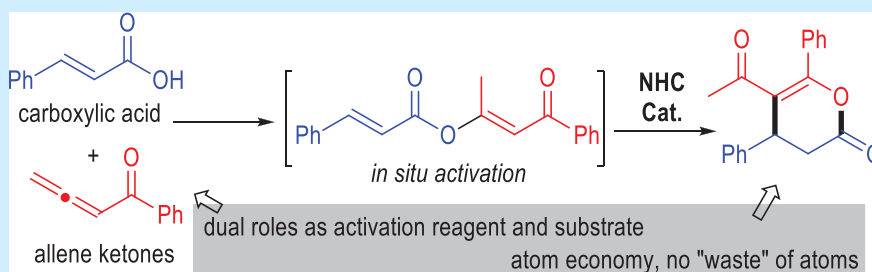
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ABSTRACT: We present a new reaction between carboxylic acids and allene ketones mediated by *N*-heterocyclic carbene (NHC) catalysts, which exhibit, in principle, nearly perfect atom economy. In this new approach, allene ketones act as both an activating reagent and a reactant. All atoms in the substrates end up in the product without the need for coupling reagents. The present study aims to encourage further explorations of NHC catalytic reactions with alternative activation strategies and better atom economy.

Dihydropyranones are important structural motifs found in natural products, pharmaceutical compounds, and synthetic intermediates.¹ The dihydropyranone ring unit, particularly as a functional subunit of isocoumarins, has attracted significant attention because of its notable bioactivity, including antifungal, anti-inflammatory, and antimicrobial properties.² The dihydropyranone skeleton serves as a versatile building block for synthesizing biologically significant molecules, like nepetalactone,³ neocucurbitacin D,⁴ and (–)-7-deoxyloganin⁵ (Figure 1a). Given the attractive biological activity and inherent structural importance, considerable efforts have been devoted to effectively access dihydropyranones. Notable examples include the Michael addition reaction catalyzed by isothiourea⁶ or *N*-heterocyclic carbene (NHC)⁷ catalysts, hetero-Diels–Alder reactions,⁸ and a range of other versatile methods.⁹ Further development of a mild and facile method for the synthesis of dihydropyranones starting from easily available materials remains highly desirable.

Carboxylic acid substrates are highly valuable in organic synthesis, medicinal chemistry, and biochemistry because of their widespread presence.¹⁰ The diverse range of carboxylic acid substrates makes them integral to research and applications in organic chemistry by offering versatility and significance in various fields. However, catalytic activation of carboxylic acids often depends on additional activation reagents.¹¹ Existing methods often rely on additional reagents to overcome the inertness of the carboxylic acids. Reagents, such as PivCl, CDI, HATU, HOBt, DCC, and others, are commonly employed to activate carboxylic acids by forming highly reactive intermediates, facilitating further functionaliza-

tion (Figure 1b). In 2009, Lupton and colleagues¹² demonstrated the use of NHC for the activation of enol-activated carboxylic acids in Michael addition reactions. In 2011, Smith et al.⁶ developed an *in situ* activation method using PivCl to form mixed anhydrides, which enabled asymmetric Michael addition reactions catalyzed by isothioureas. In 2014, Scheidt and co-workers¹³ reported that carboxylic acids could be transformed into NHC-bounded enolates via the *in situ* activation strategy in the presence of CDI. In addition, Ye and co-workers¹⁴ demonstrated an elegant access to α,β -unsaturated acyl azolium through mixed anhydride generated *in situ* from an α,β -unsaturated acid. In 2015,¹⁵ we reported direct β -carbon activation of propionic acid by carbene organocatalysis. Moreover, Yao and colleagues^{11b,16} used HATU to activate the carboxylic acid, thereby initiating a sequence of reactions catalyzed by NHC. Developing efficient carboxylic acid activation methods remains an active area of research in the field of organic synthesis.

Herein, we develop an NHC-catalyzed¹⁷ reaction of carboxylic acids using allene ketones as substrates and activating reagents (Figure 1c). This approach eliminates the need for additional activation reagents and attains a 100% atom economy. The NHC-catalyzed Michael addition

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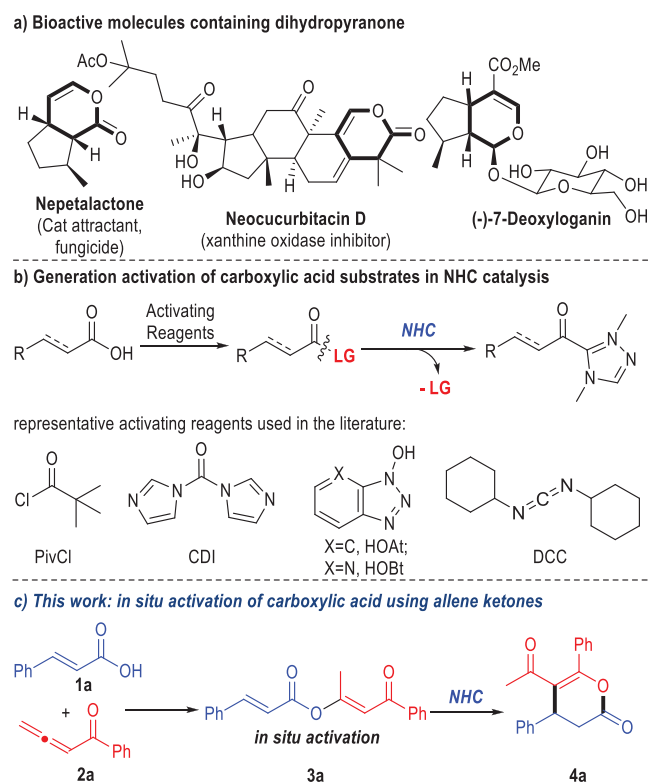
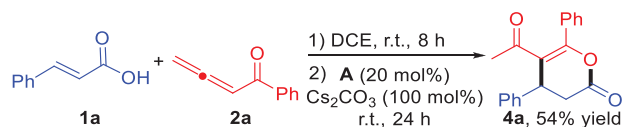


Figure 1. NHC catalytic reaction of carboxylic acids to prepare dihydropyranones.

reactions between α,β -unsaturated carboxylic acids and allene ketones, we have achieved the successful synthesis of diverse functional molecules featuring dihydropyranone frameworks, thereby attaining moderate to good yields.

Initially, α,β -unsaturated carboxylic acid **1a** and ketone **2a** were selected for the condition optimization. Surprisingly, dihydropyranone product **4a** was successfully formed. We have extensively optimized the conditions, but the reaction yields and enantioselectivity remain low (see Scheme 1 and the

Scheme 1. NHC-Catalyzed Reaction of Carboxylic Acid with Allene Ketone



Supporting Information). To make the reaction more efficient, we selected α -carbonyl vinyl ester **3a**, which is produced through the *in situ* reaction of carboxylic acid with allene ketones, as the model substrate for optimizing the reaction conditions in the Michael addition reaction (Table 1). First, various NHCs (A–D) were investigated using DBU as a base in THF (Table 1, entries 1–4). The expected substituted dihydropyranones **4a** were smoothly formed with yields of 28–39%. We also explored the morpholin-3-one-derived NHC E, but it did not enhance the yield of **4a**. The choice of base had a significant impact on the product yields. The utilization of organic bases or weak inorganic bases resulted in reduced product yields in the presence of NHC A (entries 6–9). However, the use of strong inorganic base Cs_2CO_3 led to further improvement in the product yield of **4a**. The reaction

Table 1. Optimization of Reaction Conditions^a

3a 4a

A: Ar=Mes; B: Ar=Ph; C: Ar=C₆F₅; D: Ar=PMP; E: BF₄

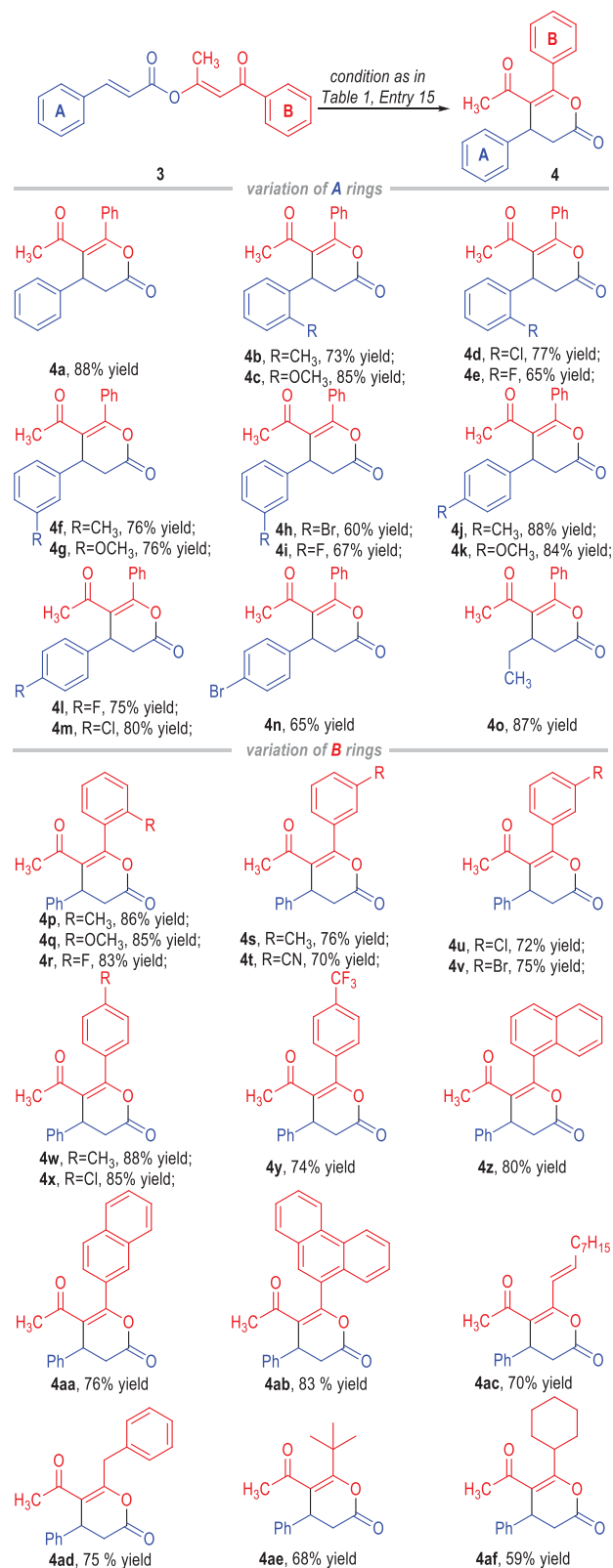
entry	NHC	base	solvent	yield (%) ^b
1	A	DBU	THF	39
2	B	DBU	THF	37
3	C	DBU	THF	28
4	D	DBU	THF	31
5	E	DBU	THF	29
6	A	NEt ₃	THF	trace
7	A	DABCO	THF	18
8	A	Na ₂ CO ₃	THF	15
9	A	K ₂ CO ₃	THF	12
10	A	Cs ₂ CO ₃	THF	45
11	A	Cs ₂ CO ₃	EA	25
12	A	Cs ₂ CO ₃	CH ₃ CN	16
13	A	Cs ₂ CO ₃	PhCl	23
14	A	Cs ₂ CO ₃	CH ₂ Cl ₂	78
15	A	Cs ₂ CO ₃	DCE	93 (88) ^c

^aUnless otherwise specified, the reactions were carried out using **3a** (0.10 mmol), NHC (0.02 mmol), base (0.10 mmol), 4 Å MS (50 mg), and solvent (1.0 mL) at rt for 24 h. ^bEstimated via ¹H NMR analysis of crude reaction mixture with internal standard. ^cIsolated yield of **4a**.

was also solvent-sensitive (entries 11–15). When highly polar organic solvents other than THF were used, the yields of the target product **4a** decreased greatly (entries 11–13). Conversely, employing nonpolar organic solvents (entries 14 and 15) significantly increased the product yield. By using dichloroethane (DCE) as the solvent, compound **4a** could be isolated with a yield of 88%.

With the optimal reaction conditions in hand (Table 1, entry 15), we then examined the substrate scope of this NHC-catalyzed Michael addition reaction using substrates α -carbonyl vinyl ester **3** bearing different substitution patterns (Scheme 2). Both electron-donating and electron-withdrawing substituents could be installed on the benzene ring A to give various substituted dihydropyranones in good yields. Both bearing electron-donating and -withdrawing substituents on the *o*-position of the phenyl ring proceeded smoothly to give the desired cycloadducts in good to high yields under the optimized conditions (**4b–4e**). Ring A with *m*- and *p*-substituents on the phenyl ring also gave the corresponding cyclization products in moderate to good yields (**4f–4n**). Notably, when the phenyl ring was alkyl-substituted, the protocol could still work well (**4o**).

Furthermore, the scope of phenyl ring B with various substituents was then examined briefly. As can be seen, both electron-donating and -withdrawing substituents on the *o*-position of phenyl ring B proceeded smoothly to give the desired cycloadducts in high yields (**4p–4r**). The yield of all products was slightly reduced when the substituents were installed at the *m*-position (**4s–4v**). *p*-Substituents on the phenyl ring B with a strong electron-drawing group $-\text{CF}_3$

Scheme 2. Scope of the α -Carbonyl Vinyl Ester 3^a

^aReaction conditions as stated in Table 1, entry 15. Yields are isolated yields.

resulting in a slight reduction in yield (**4y**). Moreover, the aromatic fused rings instead of phenyl rings, like 1-naphthalene and 2-naphthalene, also worked well in this NHC-catalyzed

Michael addition reaction (**4z–4ab**). The corresponding products could be given in good yield when replacing phenyl ring B with the alkenyl group (**4ac**). The aromatic phenyl ring B in the substrate could also be switched into an aliphatic benzyl group and *t*-butyl group without much erosion on the reaction yields (**4ad**, **4ae**). However, when the cyclohexyl group was substituted for phenyl ring B, the yield of the target product (**4af**) could only be achieved at a moderate level.

The NHC-catalyzed Michael addition reaction of α -carbonyl vinyl ester **3a** can be carried out at a 1.0 mmol scale without much erosion on the product yield (Figure 2a). We have also

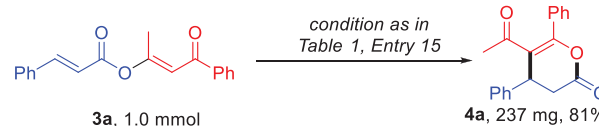
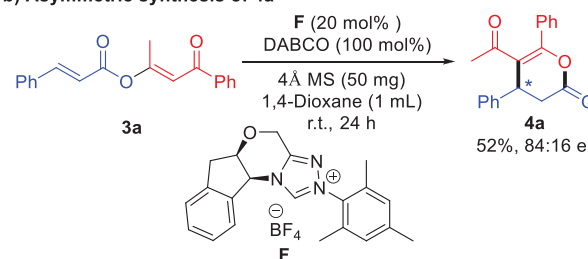
a) Synthesis of **4a** at a 1.0 mmol scaleb) Asymmetric synthesis of **4a**

Figure 2. Synthesis of **4a** at a 1.0 mmol scale and asymmetric synthesis of **4a**.

explored the enantioselective reaction using various chiral NHC catalysts (Supporting Information, Table 4). The dihydropyranone **4a** could be obtained in 52% yield with an 84:16 er value by using NHC-F and DABCO in 1,4-dioxane (Figure 2b).

In summary, we have developed an NHC-catalyzed Michael addition reaction for quick and efficient synthesis of dihydropyranone molecules. Substituents with various electronic and steric effects are well tolerated, with the target products afforded in good to excellent yields. It is noteworthy to mention that in situ activation of α,β -unsaturated carboxylic acid can be effectively accomplished using allene ketone. Additionally, the allene ketone serves a dual role as both an activating reagent and a substrate, which greatly enhances the atom economy in the reaction.

■ ASSOCIATED CONTENT

Data Availability Statement

The data underlying this study are available in the published article and its online Supporting Information.

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.3c03623>.

Experimental procedures and spectral data for all new compounds (PDF)

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Author Contributions

C.M. contributed to designs and most of the experiments. Z.J. conceptualized and directed the project and drafted the manuscript with assistance from all coauthors. All authors contributed to part of the experiments and/or discussions.

Notes

The authors declare no competing financial interest.

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