

Roll No .....

**BE - 101****B.E. I & II Semester Examination, December 2013****Engineering Chemistry***Time : Three Hours*

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**Maximum Marks : 70**

**Note:** Total number of question 10. Attempt one question (including all parts) from each unit. Assume missing data if any suitably.

**Unit - I**

1. a) Define softening of water. Describe the Zeolite method of softening of hard water with the help of neat and labelled diagram. 6
- b) What is boiler corrosion. Discuss its causes? 4
- c) Write short note on any two- 4
  - i) Caustic embrittlement
  - ii) Chlorination
  - iii)  $\text{CaCO}_3$  equivalent

OR

2. a) Define hardness of water. Discuss the internal treatment method of water softening. 6
- b) What is Boiler Trouble? Explain the causes of scale formation. 4
- c) A sample of water on analysis was found to contain the following  $\text{Ca}(\text{HCO}_3)_2 = 4 \text{ mg/l}$ ,  $\text{Mg}(\text{HCO}_3)_2 = 6 \text{ mg/l}$ ,  $\text{CaSO}_4 = 8 \text{ mg/l}$ ,  $\text{MgSO}_4 = 10 \text{ mg/l}$ . Calculate the temporary, permanent and total hardness of water in ppm,  $^\circ\text{Cl}$  and  $^\circ\text{Fr}$ . 4

**Unit - II**

3. a) Define calorific value. How it is determine by bomb calorimeter? 6
- b) Discuss the parameters of ultimate analysis of coal? 4
- c) Write short note on any two. 4
  - i) Anti knock compound
  - ii) Octane No
  - iii) Carbanization

OR

4. a) Describe the Otto-Hoffmann's method for the manufacture of coke. Write the recovery of by products. 7
- b) On burning 0.83 g of a solid fuel in a bomb calorimeter, the temperature of 3500g of water increases from  $26.5^\circ \text{C}$  to  $29.2^\circ \text{C}$ . Water equivalent of calorimeter and latent heat of steam are 385g and 587 cal/g respectively. If the fuel contains 0.7% hydrogen, calculate its gross and net calorific value? 7

**Unit - III**

5. a) Define lubricant. Discuss the classification of lubricants. 6

- b) Write short note on any two: 4
- i) Saponification value
  - ii) Thermal spalling
  - iii) Blended oil
- c) Explain cement manufacture by wet process. 4

OR

6. a) Define lubrication. Describe hydrodynamic mechanism of lubrication. 6
- b) What are refractories. Describe the properties of refractories. 4
- c) An oil sample under test has a Saybolt universal viscosity same as that of Standard Gulf oil (low viscosity standard) and Pennsylvanian oil (high viscosity index standard) at 100°F are 61758 and 420 respectively. Calculate viscosity index of the sample oil. 4

**Unit - IV**

7. a) Define polymerisation. Explain free radicals mechanism of polymerisation. 6
- b) Write preparation, properties and uses of buna-N. 4
- c) Explain condensation polymerisation with two examples. 4

OR

8. a) Discuss the classification of polymers on the basis of molecular forces. 6
- b) Explain polyvinyl chloride is stronger and tougher than polyethylene. 4
- c) Write short note on any two- 4
- i) Vulcanisation
  - ii) Phenolic Resin
  - iii) Nylon-6,6

**Unit - V**

9. a) Discuss in brief the type of molecular vibrations. 6
- b) Explain Gas Liquid Chromatography. 4
- c) Write short note on any two- 4
- i) Lambert Beer law
  - ii) Chemical shift
  - iii) Partition chromatography

OR

10. a) Define spectroscopy discuss the principle and working of NMR spectroscopy. 6
- b) Write short note on application of UV spectroscopy. 4
- c) Write short note on any two- 4
- i) Stretching and Bending vibrations
  - ii) Chromophores
  - iii) Significance of IR spectroscopy.

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