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B. Pharm. (First Semester) EXAMINATION, June, 2008 PHARMACEUTICAL CHEMISTRY—I

(Physical Chemistry)

(PY-104)

Time: Three Hours

Maximum Marks: 70

Note: Attempt any five questions. All questions carry equal marks.

- 1. (a) What is an ideal gas? Write the ideal gas equation and give the units for each term in the equation.
 - (b) Explain the deviation of ideal behaviour of gases.
- 2. (a) Describe the intermolecular forces involved in liquids.
 - (b) How will you measure viscosity of a compound? Explain.
- 3. (a) Write exhaustive note on application of partition coefficient in pharmacy.
 - (b) What is pH? Explain.
- 4. (a) Write the equation relating to the half life of a first order reaction to the rate constant. How does it differ from that of a second order reaction?

- 5. (a) Define the following:
 - (i) Enthalpy
 - (ii) Entropy
 - (iii) Free energy
 - (b) Define Hess's Law with an example, the usefulness of Hess's law in thermochemistry.
- 6. Explain Langmuir theory of adsorption.
- 7. (a) Are enzyme catalysed reactions example of homogeneous or heterogeneous catalysis? Explain.
- (b) What is quantum mechanics? Explain.
- 8. Write notes on the following:
 - (a) Surface tension
 - (b) Jablonski diagram

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