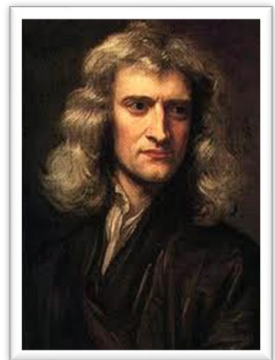




Chapter Two

Properties of fluids

Newton
1642-1727



What is fluid?

- Fluid is a substance which can flow and deform under shear stress indefinitely without returning to its original position.
- Fluid can be either gas or liquid



Flowing water



Tornado

Fluid Properties

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- Every fluid has certain characteristics by which its physical conditions may be described
- The properties outlines below are general properties of fluids which are of interest in engineering
 - Mass Density/Specific Weight
 - Bulk Modulus
 - Vapor Pressure
 - Surface tension
 - Viscosity



Mass Density/Specific Weight

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- Density is defined as an objects mass per unit volume. (Kg/m^3 SI Unit)

$$\rho = \frac{m}{V} \quad (\text{mass density of water at } 4^{\circ}\text{C } 1000 \text{ kg/m}^3)$$

- Specific weight is defined as weight per unit volume (N/m^3 SI Unit)

$$\gamma = \rho g$$

- Relative Density is the ratio of the substance to the density of water at 4°C

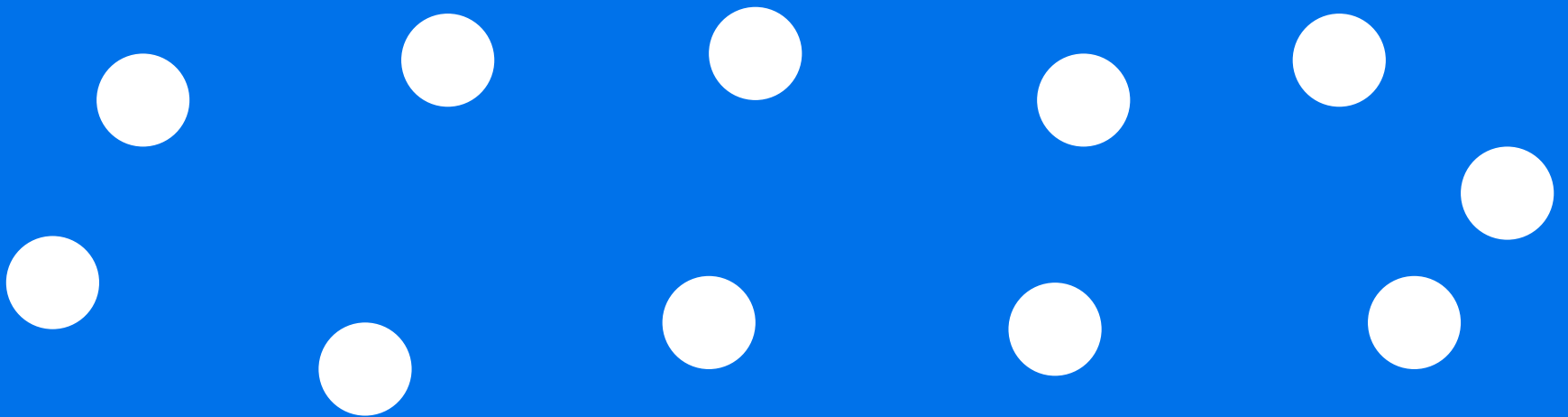
*** m denotes mass, g denotes acceleration of gravity



Bulk Modulus

- Bulk modulus is defined as the ratio of the infinitesimal pressure increase to the resulting relative decrease of the volume. (Pa SI Unit)

$$K = -V \frac{dP}{dV}$$



Vapor Pressure

- Vapor pressure: at or below which a liquid will boil
- Vapor pressure decreases with the decrease of liquid temperature

Temperature(°C)	Pressure(atm)
-10	0.0027
0	0.0060
20	0.0230
40	0.0727
60	0.1965
80	0.4672
100	1.0000
110	1.4144

Cavitation occurs



- Viscosity is a measure of the **resistance** of a fluid which is being deformed by either shear stress or tensile stress
- Viscosity describes a fluid's internal resistance to flow and may be thought of as a measure of fluid friction
- When fluid is still, the viscosity isn't reflected, but is its substantial property.

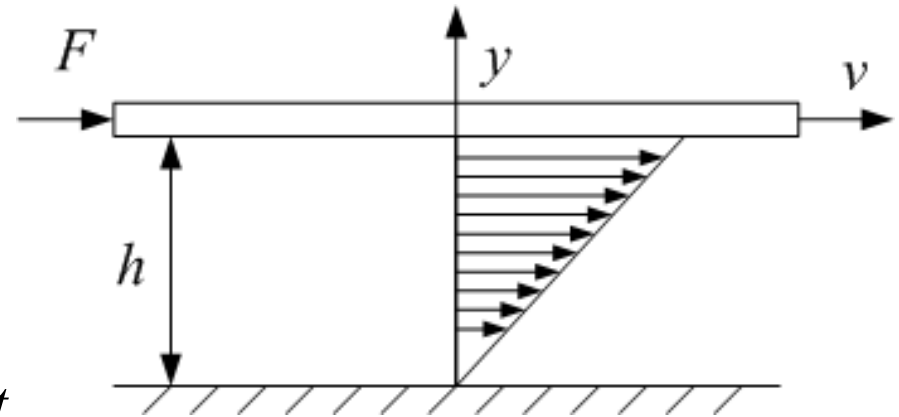


Newton's law of viscosity

- Newton found that

$$F \propto A \frac{v}{h}$$

A is the plate area, h is the height, and v is the plate velocity



Viscous force experiment

Differential
Form:

$$\tau = \pm \mu \frac{dv}{dy}$$

τ - shear stress (an applied force per unit area needed to produce deformation in a fluid) [Pa]

μ - dynamic viscosity [$\text{N}\cdot\text{s}/\text{m}^2$]

v - fluid velocity

y - distance from solid surface

Newton's law of viscosity



- Dynamic viscosity
 - Symbol μ
 - SI Unit Pa•s
- Kinematic viscosity
 - Symbol ν
 - SI Unit m^2/s
 - $\nu = \mu / \rho$



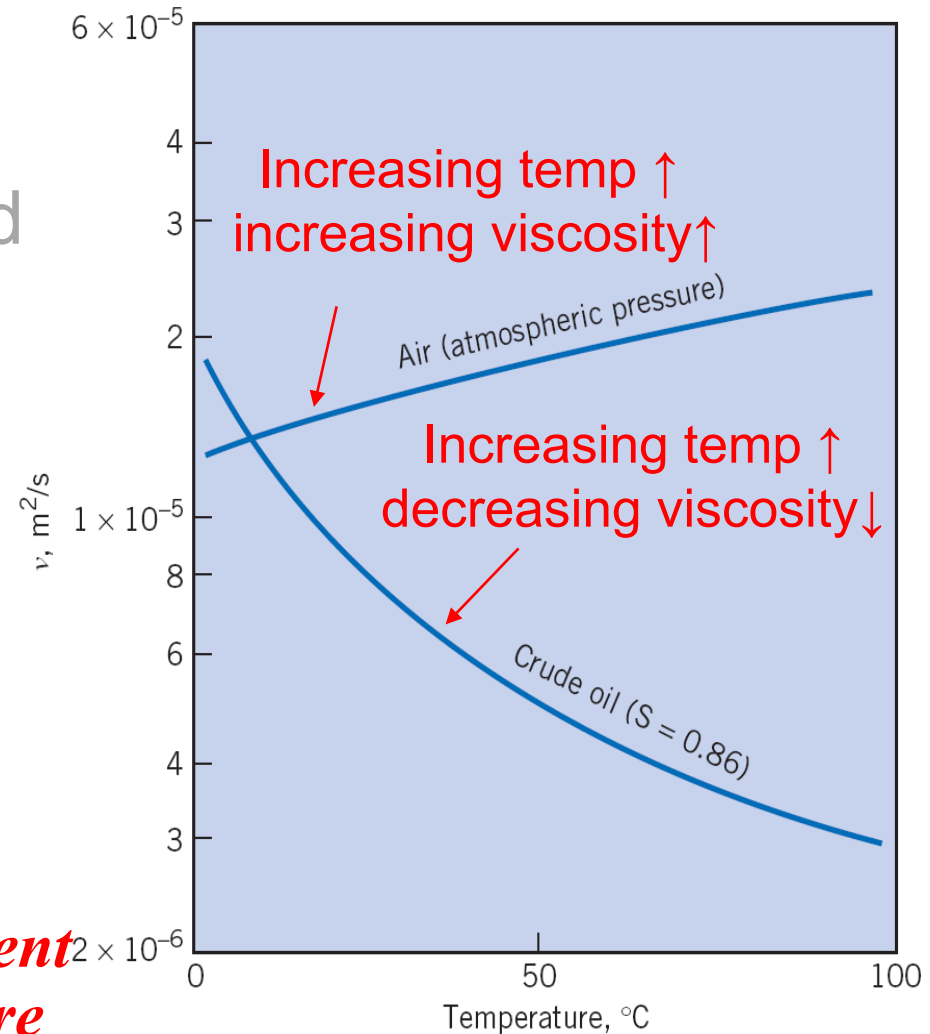
Pitch has a viscosity approximately 230 billion (2.3×10^{11} Pa•s) times that of water (Year1927- now)

****Note: 46# hydraulic oil means its kinematic viscosity is $46 \times 10^{-6} m^2/s$ at $40^\circ C$***

Why the fluid is viscous?

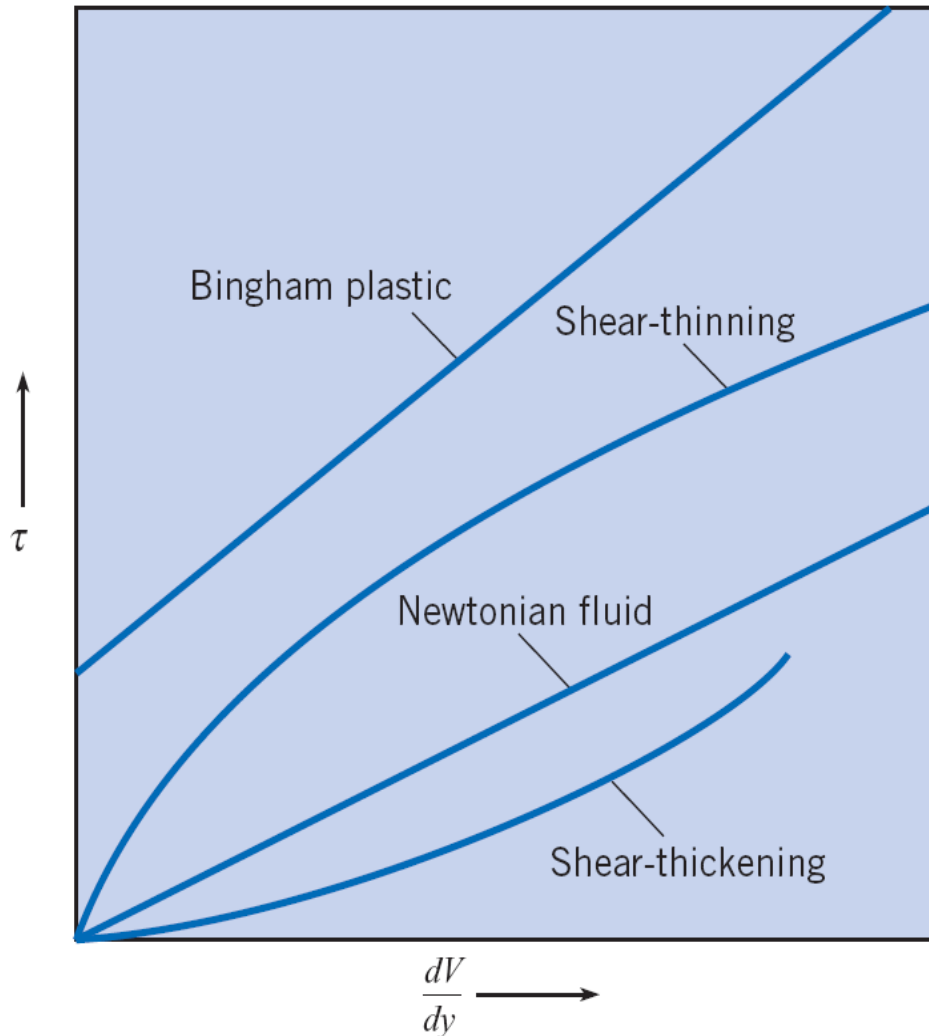
- In gases, molecular collisions transfer momentum between fluid layers
- In liquids, friction between fluid layers comes from the interattractive force of molecules sliding past each other

Note: Gas and liquid show different trends for increasing temperature



Newtonian vs. Non-Newtonian Fluids

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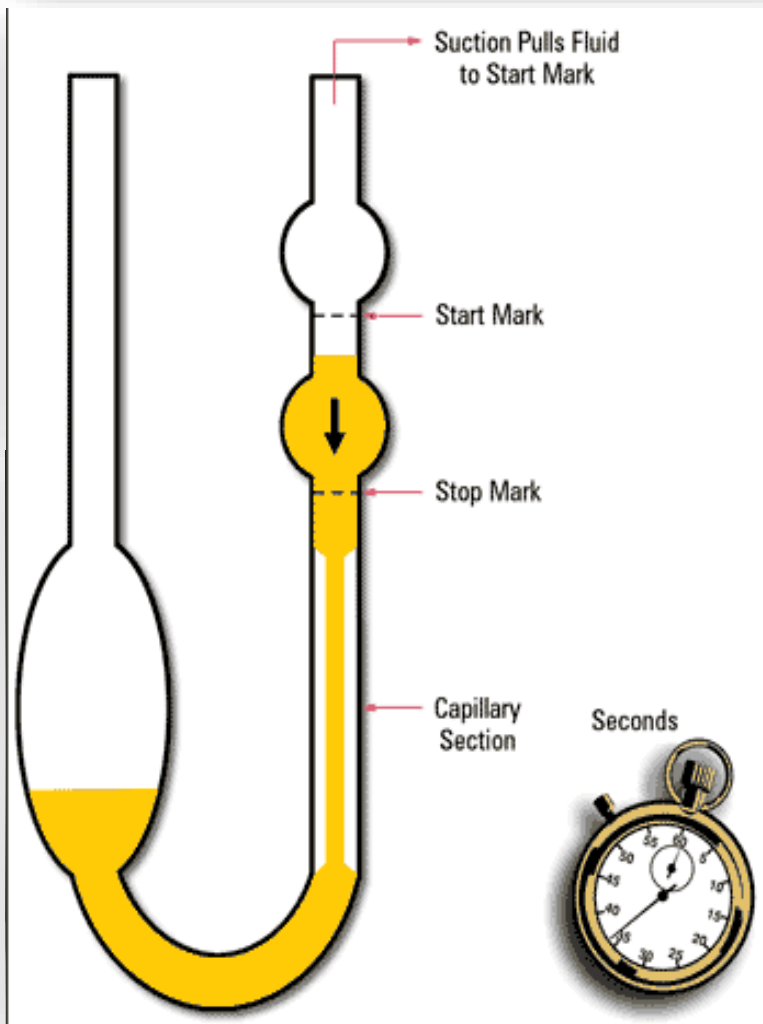


Slope of line is dynamic viscosity

- **Newtonian fluid:**
Shear stress is proportional to shear strain
- **Non-Newtonian fluid:**
Shear stress is not proportional to shear strain (*not linear to shear strain or not through the zero point*), such as toothpaste, catsup, paint, gypsum-water mixtures



How to measure viscosity?



- **Capillary viscometer** :*(for kinematic viscosity measurement)*
is used for those fluids which has a single value of viscosity and not sensitive to temperature
The time taken for the level of the liquid to pass between these marks is proportional to the kinematic viscosity.
- **Characteristics**
Simple, Low price

Compressible or not?

- All real fluids are compressible - even water
- If the density change of fluids in pressure is small, it can be regarded as incompressible, otherwise compressible
- In most steady conditions liquids are treated as incompressible, on the contrary gases are treated as compressible, but not always
- Incompressible fluid means ρ is constant



Viscous or not?

- All real fluids are viscous
- Ideal fluid is inviscid

Terminology

Compressible fluid	• ρ is not constant
Incompressible fluid	• ρ is constant
Ideal fluid	• μ is zero
Real fluid	• μ is not zero



The Continuum Assumption

- Fluids are composed of many finite-size molecules with finite distance between them. These molecules are in constant random motion and collisions
- This motion is described by statistical mechanics (Kinetic Theory)
- Within the continuum assumption there are no molecules.
The fluid is continuous
- Fluid properties as density, velocity etc. are continuous and differentiable in space & time
- A fluid particle is a volume large enough to contain a sufficient number of molecules of the fluid to give an average value for any property that is continuous in space, independent of the number of molecules



The Continuum Assumption

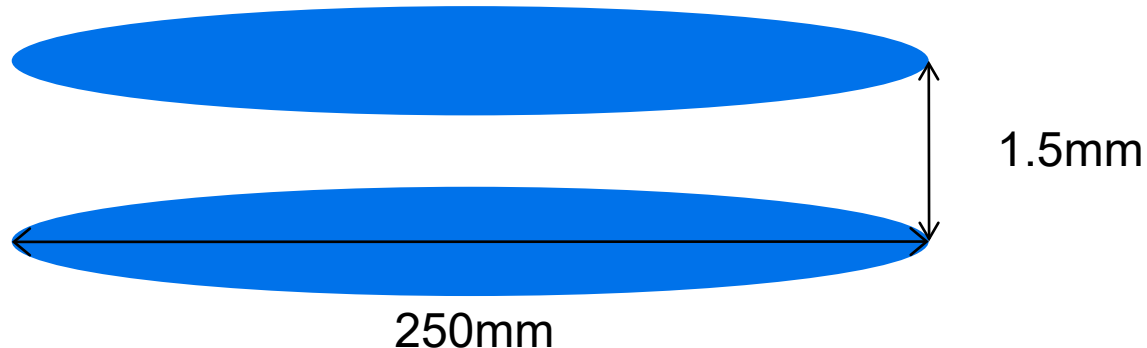
- Characteristic scales for standard atmosphere:
 - atomic diameter $\sim 10^{-10}$ m
 - distance between molecules $\sim 10^{-8}$ m
 - mean free path, λ (sea level) $\sim 10^{-7}$ m
- Knudsen number: $Kn = \lambda/L$
 - λ - mean free path
 - L - characteristic length
- L is the size of investigated object, in most cases, $\sim 10^{-3}$

Continuum assumption: $Kn \ll 1$



Example

- Two discs of 250 mm diameter are placed 1.5 mm apart and the gap is filled with an oil. A power of 500 W is required to rotate the upper disc at 500 rpm while keeping the lower one stationary. Determine the viscosity of the oil.



[0.71 kg/ms]

Example

- Newton fluid is linear to du/dy , if shear stress is linear to du/dy , is it Newton fluid?

