Implementation of an Alternate Chemistry Library into OpenFOAMTM

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Implementation of an Alternate Chemistry Library into OpenFOAMTM

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Acknowledgements



The chemistryModel-class

- Part of the OPENFOAMTM-distribution
 - Solves a set of chemical reactions and returns a reaction-rate
 - Used in the solvers dieselFoam and reactingFoam
- Chemical reactions can be specified in two formats:
 - A generic OPENFOAMTM-format (allows the inclusion of non-standard reaction rates)
 - The CHEMKINTM-format which is very "FORTRAN"
- Both formats are converted to the same internal representation
- ... and are solved using a number of different solvers





CANTERA

CANTERA [1] is a open-source chemical kinetics package

- Very flexible and easy to handle
- Important functions for chemical species, (heterogeneous) kinetics, equilibrium, transport, diffusion, heat conduction are available and efficiently implemented
- Kernel written in C++; accessible via PYTHON, MATLAB. FORTRAN and of course C/C++
- Good CHEMKINTM-Lexer
- A set of ideal reactors is available.
- Uses the CVODE-Solver of the Sundials Suite [2] for integration of the ODEs.





Goals of this work

As mentioned in the Milan presentation [3], Cantera would be benefitial for the development of solvers for reacting flows:

- Flexible exchange of chemistry engines without the necessity to change the solver
- Easy access to thermochemical functions
- Better stability (JANAF-error)
- Stationary chemistry solver





The Alternate Chemistry Library

- A Library that provides a framework for adding further chemistry solvers
- Should require only minimal changes to existing solvers
 - Solvers should be implementation-agnostic (by using the RunTime-selection mechanism)
- This is achieved by replacing the chemistryModel object (usually called chemistry) in the solvers with a autoPtr<alternateChemistryModel>
 - Direct subclassing of chemistryModel is not possible because necessary properties are private
 - Usually every call chemistry.xxxx() is replaced by a chemistry().xxxx()





Interface

 Only a small part of the original chemistryModel-interface is needed by the solvers:

solve Solve the reaction system

RR Access the reaction rates

tc Access the chemical time-scale

- Only these have to be implemented by a new chemical solver class
- In addition to these the methods

calcDQ for the calculation of dQ

tf Access the flow time-scale (described later)

were added to the interface



chemistryModelProxy

- Provides access to the functionality of the original chemistryModel-class
- Has a private chemistryModel-object
- Calls are passed through to the chemistryModel-object
 - solve() and tc()
 - calcDQ() implemented after example from original solvers
- With this chemistry model implementation the solver is equivalent to the original reactingFoam





The CANTERA-model

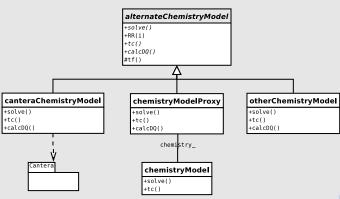
- Implements interface to the CANTERA-package
- Requires the thermoType to be a hMixtureThermo<CanteraMixture>
 - Instead of the usual hMixtureThermo<reactingMixture>
 - Currently required to have consistent calculations of the mixture and the chemistry
 - Calculation of thermophysical properties is done in CanteraThermo a wrapper to the CANTERA-class IdealGasMix
- Reactions are read from a cti-file (CANTERA Input)
 - Also the thermophysical data of the species. Thus the requirement on the thermoType
- Calculations in solve() are done via the Reactor/ReactorNet-classes of CANTERA
 - Solution of the ODEs is done via CVODE



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Class overwiew







alternateReactingFoam

Solver almost unchanged. Basis: reactingFoam. Remember the Chalmers **PaSR-Model** for turbulent combustion:

$$\kappa = \frac{dT + tc}{dT + tc + tk} \tag{1}$$

Where *tc* is the **chemical time scale** and *tk* is the **turbulent** time scale.

- Implementation of tc()-method for the CANTERA calculation necessary.
- For the CANTERA solver no ODE solver from OF necessary (choice of the ODE solver in chemistryProperties doesn't have any effect).



alternate Steady Reacting Foam

Steady solvers are a must when simulating large-scale chemical reactors or burners.

Necessary changes from the transient case:

- Usage of the SIMPLE-algorithm for p-U-coupling
- Coupling of chemistry to the flow-time
- **Stabilization** of the solution by reduction of κ if $|Y_i^{old} Y_i^{new}|$ is too large (only at start-up!)





Flow time scale

In **unsteady** calculations Co < 1 and integration time equals dT. For **steady-state** solvers Co >> 1. So we limit the integration time by the residence time in the cell i:

$$tf_i = \frac{D_i}{U_i} \tag{2}$$

and hence we substitute dT in (1) by df:

$$\kappa = \frac{tf + tc}{tf + tc + tk} \tag{3}$$



We define the **flow time scale** tf.



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CVODE

An OF library that makes use of the $\mathrm{CVODE} ext{-}\mathsf{Solver}$ of the Sundials Suite [2] was written.





Plug flow reactor (PFR) test case

Simple test case, which is easily comparable to ideal PFR models as available in CHEMKINTM or CANTERA:

- Ignitable premixed composition at inlet: $Y_{H2} = 0.009$, $Y_{O2} = 0.026$, $Y_{Ar} = 0.965$, T = 1600 K, p = 1 bar
- Tube idealized as 2D-domain $x_{max}=1$ m, $y_{max}=0.1$ m, 10x100 cells, inlet left patch, outlet right patch, upper and lower patch symmetry
- $H_2 O_2$ mechanism with 9 species and 27 reactions





PFR alternateReactingFoam

For the transient simulations we were interested in

- Behaviour of the Cantera chemistry library
- Choice of ODE solvers and their performance

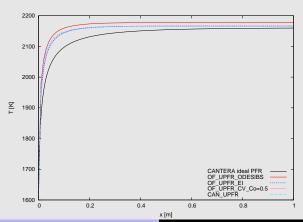
All plots were done along the **x direction**. The ideal Solution was obtained by calculating the time into a position with U and ρ .





PFR alternateReactingFoam ODE-solvers (T)

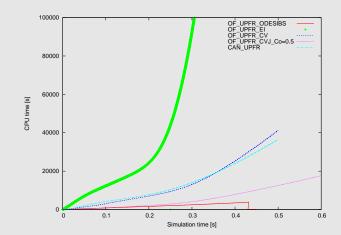
ODESIBS: ODE SIBS; EI: Euler Implicit; CV: CVODE







PFR alternateReactingFoam ODE CPUTime







PFR alternateReactingFoam Summary

Transient PFR results:

- With implementation of CANTERA chemistry and tc() results for both chemistry engines give identical results for PFR.
- ODE SIBS results deviate sligtly from the others (EI, CV) (due to semi-impliciteness?)
- CVODE more stable than SIBS, EI but sometimes slower
- CVODE benefits from larger integration steps (internal time-stepping). → So use higher Co, if results aren't affected.
- Compared to ideal reactor solvers are 'faster' and there is a temperature difference of 5 K. Possibly induced by discretisation errors or idealisation.



PFR alternateSteadyReactingFoam

The following aspects for the runs of the **steady-state simulations** are of particular interest:

- A Performance and stability of the steady-state implementation in relation to transient case and the ideal case.
- **B** Influence of the implementation of κ
- C Influence of Cmix

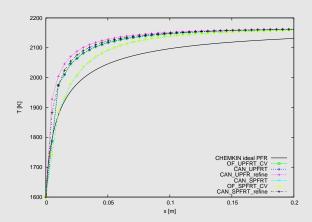
Plots are similar to the transient ones but scaling was changed.





PFR alternateSteadyReactingFoam (T)

A Transient vs. steady

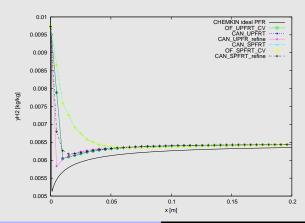






PFR alternateSteadyReactingFoam (H2)

A Transient vs. steady

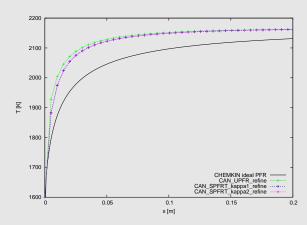






PFR alternateSteadyReactingFoam (T)

B Influence of κ

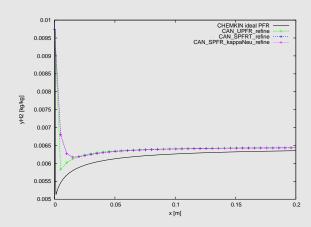






PFR alternateSteadyReactingFoam (H2)

B Influence of κ

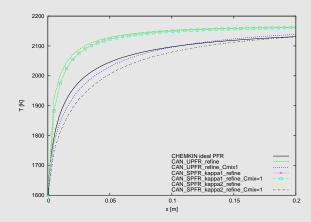






PFR alternateSteadyReactingFoam (T)

C Influence of Cmix

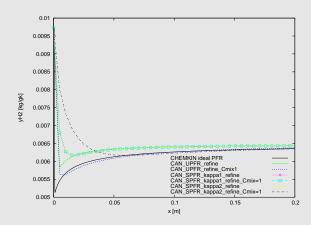






PFR alternateSteadyReactingFoam (H2)

C Influence of Cmix

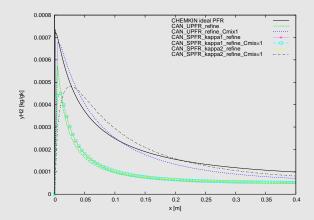






PFR alternateSteadyReactingFoam (OH)

C Influence of Cmix







PFR Steady Summary

A:

 Transient/Steady solutions do not agree totally but refinement reduces deviation

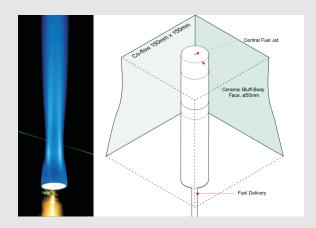
B+C:

- Definition of new κ according to (3) brings no changes for small values of Cmix. But for Cmix=1 the solution seems closer to ideal solution.
- However, no clear conclusion could be drawn.





Sydney Bluff-body Flame (HM1)

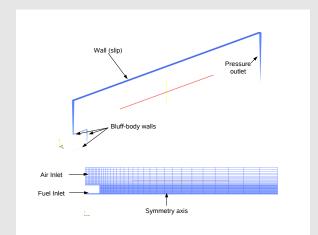




Data from Sandia bluff-body flame HM1 conducted at Sydney university [4].



HM1 case set-up







HM1 set-up

Boundary Conditions:

- Diameters: $D_{jet} = 3.6$ mm; $D_{channel} = 150$ mm (300mm), $D_{bluffbody} = 50$ mm
- Turbulence: 8.5 % (2.5%) turbulence intensity, 0.135 mm (5.625 mm) mixing length for jet (co-flow)
- Mesh: from blockMesh with 1095

Inlet conditions:

• HM1: $U_{iet} = 118 \text{m/s} \ U_{coflow} = 40 \text{m/s}$

Other models:

- $k \epsilon$ turbulence model, ATRMech (reduced GRI 3.0)
- OF used KRR4 ODE solver for chemistry



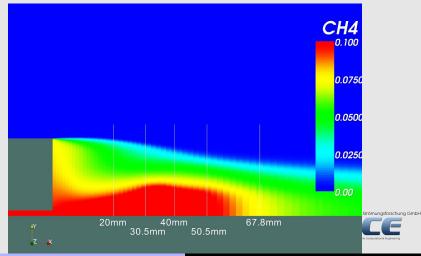
HM1 Plots & Points of Measurement

- Plots of radius against measurement values (Exp HM1) at different x-positions are shown together with simulation results.
- EDC referes to the EDCSimpleFoam solver [3].
- CAN and OF mark CANTERA- and OPENFOAM TM-solution
- **K1** and **K2** stands for κ_1 and κ_2





HM1 Plots & Points of Measurement



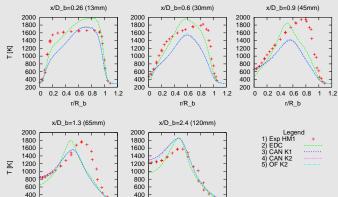


HM1 results T

200

0.2 0.4 0.6 0.8

r/R_b



0.2 0.4 0.6 0.8

r/R b

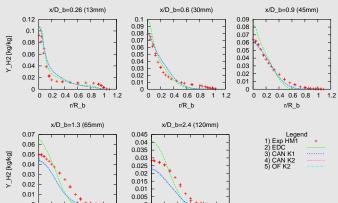




200

1

HM1 results H2



0.2 0.4 0.6 0.8 1

r/R b





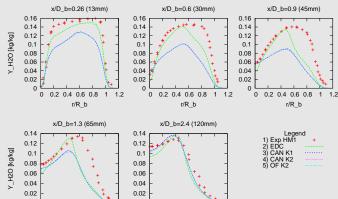
0.2 0.4 0.6 0.8

r/R b

1 1.2

1.2

HM1 results H2O



0.2 0.4 0.6 0.8

r/R b





1 1.2

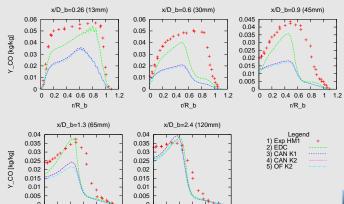
0.2 0.4 0.6 0.8

r/R b

0

1

HM1 results CO



0.2 0.4 0.6 0.8

r/R b





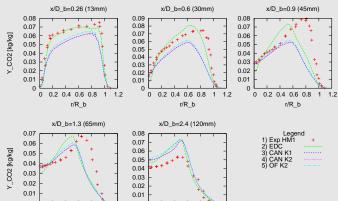
1 1.2

0.2 0.4 0.6 0.8

r/R b

1.2

HM1 results CO2



0.2 0.4 0.6 0.8

r/R b





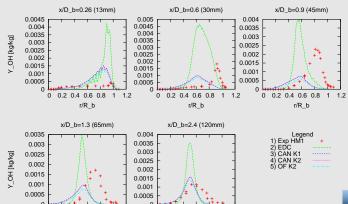
1 1.2

0.2 0.4 0.6 0.8

r/R b

- 1

HM1 results OH



0.2 0.4 0.6 0.8

r/R b



1 1.2

0.2 0.4 0.6 0.8

r/R b

1 1.2

Discussion HM1

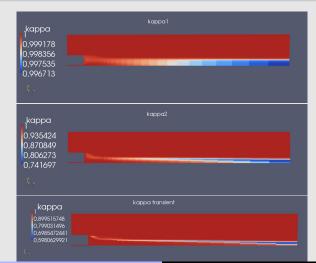
Why are all the resuls **so similar**? (No influence of κ -implementation or Cmix (not shown))

- In case Cmix=0.005 no difference in κ notable too fast mixing as in PFR case.
- Furthermore, in the domain close to the inlets $\kappa \approx 1$ in every case and influence of Cmix can only be seen further downstream where no measurements are available.
- Comparison with **transient simulation** of HM1 indicates that κ_2 seems more appropriate (see next slide)
- Still, further testing is needed





HM1 comparison κ_1, κ_2 for Cmix=1







Conclusion HM1

- Flame was rendered with acceptable quality by the new solvers
- Steady solvers are very robust if the right ODE solvers are used (CVODE for CANTERA, KRR4 for OF)
- Checking of dY_i provides further stability and and prevents composition from being driven to unphysical states.
- Solutions for a finer grid (20000 cells) are still not available flame blows off very easily.





Summary

The implementation of the Cantera chemistry into OpenFOAMTM can be considered successful and provides:

- A flexible interface to access a wide range of thermophysical data, and chemistry functions
- Robust solvers for transient and steady-state calculations
- Valuable test-cases for validation and further investigation
- Collaborators are welcome to test the code and contribute own developments.





How to get the library

 \bullet The libraries and solvers presented here will be made available on the openfoam-extend-SVN on sourceforge.net

Libraries Two independent libraries:

alternateChemistryModel The general

framework an the $OPENFOAM^{TM}$ -interface

canteraChemistryModel The interface to

CANTERA

Solvers The steady and the transient solver with the test-cases PFR and HM1

URL will be given on the WIKI



 Interfaces to other chemistry libraries, more solvers, critique, and improvements are encouraged



Future Work

There is always room for improvement:

- Both chemistry solvers read the same input files
 - Converter between CANTERA and OPENFOAMTM-format
- Make the CANTERACHEMISTRYMODEL independent of canteraMixture
- Implementation of reduction/tabulation techniques for more time-efficient solution.
- More combustion models.





Acknowledgements

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