

# Crystal Systems and Crystallographic Point Groups

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# Point Symmetry & Crystallography

- Symmetry is fundamental in crystallography
- Determines what properties a material can have:
  - Piezoelectricity
  - Non-linear optical properties
  - etc.
- Determines how we refine crystal structures and what conclusions we can draw from our structure determinations
  - Centrosymmetric or non-centrosymmetric
  - Polar or non-polar group, enantiomeric group
  - Absolute structure and absolute configuration

# Hermann-Mauguin Symmetry Notation

- Spectroscopists use Schoenflies notation to describe symmetry (e.g.  $C_{2v}$ ,  $D_{4h}$ )
- Crystallographers use Hermann-Mauguin notation (International notation)
- Was introduced by Carl Hermann in 1928, modified by Charles-Victor Mauguin in 1931
- Adopted for the 1935 edition of the *International Tables for Crystallography*
- Hermann-Mauguin notation is preferred for crystallography
  - Easier to add translational symmetry elements
  - Directions of symmetry axes are specified

# Notes on Hermann-Mauguin Notation

- Interpretation of Hermann-Mauguin symbols depends on the crystal system
- “n/m” notation means mirror plane perpendicular to n-fold axis
- Hermann-Mauguin symbols have both “long” and “short” forms
- Not all symmetry elements present are symbolized, some are left implicit

# Crystallographic Point Symmetries

- Point symmetries are symmetries which all pass through a given point and this point does not change with the application of a symmetry operation
- The symmetry elements which constitute the crystallographic point groups are:
  - Proper rotation axes ( $n$ )
  - Mirror planes ( $m$ )
  - Inversion centre ( $\bar{1}$ , or no explicit symbol)
  - Rotary inversion axes ( $\bar{n}$ )
- Only  $n$ -fold axes where  $n = 1, 2, 3, 4, 6$  are allowed for space filling 3 dimensional objects
- 32 unique crystallographic point groups are obtained from combining the various allowed rotation axes, mirror planes, and inversions
- 11 of the 32 crystallographic point groups are **centrosymmetric**

# Lattices and Symmetry

- Lattices are a regular array of points
- We use **basis vectors** to describe the lattice
- The choice of basis vectors is not unique
- **We choose the set of basis vectors which reflects the symmetry present in the lattice**
- Transforming from one set of basis vectors does not change the lattice only our description of it

# 3-fold Rotation in Different Coordinate Systems

- The coordinate system makes a difference how symmetry operations are expressed
- Using cartesian systems are more complicated in crystallography
- Involves calculation of transcendental functions

*Cartesian Coordinate System*

$$\begin{pmatrix} \cos \theta & -\sin \theta & 0 \\ \sin \theta & \cos \theta & 0 \\ 0 & 0 & 1 \end{pmatrix}$$

*Trigonal Coordinate System*

$$\begin{pmatrix} 0 & \bar{1} & 0 \\ 1 & \bar{1} & 0 \\ 0 & 0 & 1 \end{pmatrix}$$

# Crystal Systems

There are 7 crystal systems and they are named:  
**Triclinic, Monoclinic, Orthorhombic,**  
**Tetragonal, Trigonal, Hexagonal, and Cubic.**

What differentiates one crystal system from another?

**The order of its principal or characteristic symmetry**



# Crystal Systems & Their Symmetries

Crystal System	Lattice & point symmetries	Metric Constraints NOTE: “ $\neq$ ” means “not constrained to be equal to” rather than “not equal to”
Triclinic	$\bar{1}$ , 1	$a \neq b \neq c$ ; $\alpha \neq \beta \neq \gamma$
Monoclinic	$2/m$ , 2, m	$a \neq b \neq c$ ; $\alpha = \gamma = 90^\circ$ , $\beta \neq 90^\circ$
Orthorhombic	$mmm$ , $mm2$ , 222	$a \neq b \neq c$ ; $\alpha = \beta = \gamma = 90^\circ$
Tetragonal	$4/mmm$ , $\bar{4}2m$ , $4mm$ , 422, $4/m$ , $\bar{4}$ , 4	$a = b \neq c$ ; $\alpha = \beta = \gamma = 90^\circ$
Trigonal rhombohedral setting hexagonal setting	$\bar{3}m$ , 3m, 32, $\bar{3}$ , 3	$a = b = c$ ; $\alpha = \beta = \gamma \neq 90^\circ$ $a = b \neq c$ ; $\alpha = \beta = 90^\circ$ , $\gamma = 120^\circ$
Hexagonal	$6/mmm$ , $\bar{6}m\bar{2}$ , $6mm$ , 622, $6/m$ , $\bar{6}$ , 6	$a = b \neq c$ ; $\alpha = \beta = 90^\circ$ , $\gamma = 120^\circ$
Cubic	$m\bar{3}m$ , $\bar{4}3m$ , 432, $m\bar{3}$ , 23	$a = b = c$ ; $\alpha = \beta = \gamma = 90^\circ$

# Categories of Crystallographic Point Groups

- Centrosymmetric (11 of the 32 point groups)
- Non-centrosymmetric (non-exclusive categories)

**Enantiomorphic:** Point groups which contain only proper rotation axes

- Enantiopure compounds can only crystallize in crystals which have these point symmetries
- SHELXL hint: Flack parameter should be refined, absolute configuration determination is possible

**Polar:** Point groups which have a “polar” sense to them

- Polar groups are non-centrosymmetric, but may have symmetry elements of the second kind
- Samples which crystallize in these point group may be racemic
- SHELXL hint: Flack parameter should be refined, **absolute structure determination is possible but absolute configuration cannot be determined if point group contains symmetry elements of the second kind**

# Laue Groups and Holohedries

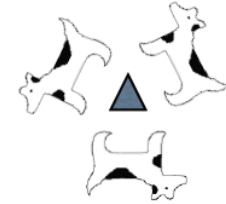
- **Laue groups:** the 11 centrosymmetric groups
  - Symmetry of the diffraction pattern as determined from the observed intensities
  - Matches the space group without any translations and adding a centre of symmetry
  - A crystal system can have more than one Laue group
- **Holohedry:** When the point group of a crystal is identical to the point group of its lattice
  - There are 7 holohedral point groups which correspond to the 7 crystal systems
  - Holohedries are always centrosymmetric
- All holohedries are Laue groups, but not all Laue groups are holohedries

# Proper Rotation Axes

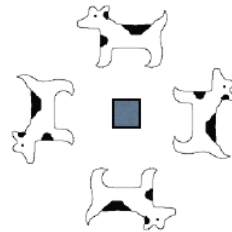
- Rotation about an axis by  $360^\circ/n$ .
- Symmetry operation of the first kind
- Doesn't change handedness of object



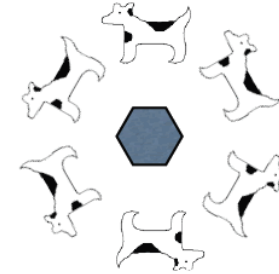
2 fold  
'Diad'



3 fold  
'Triad'



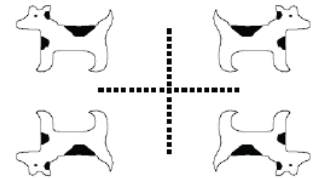
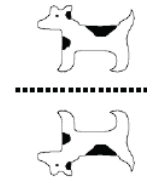
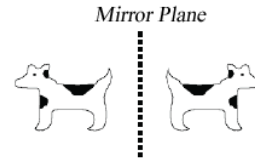
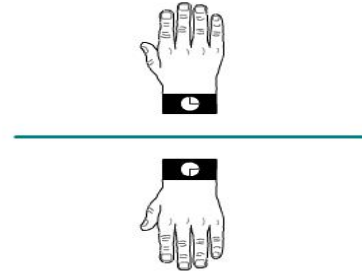
4 fold  
'Tetrad'



6 fold  
'Hexad'

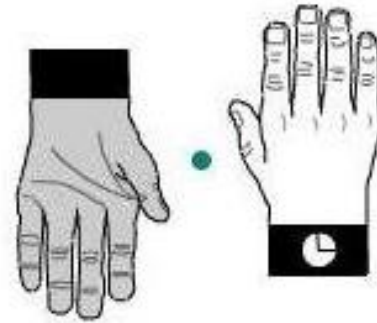
# Mirror plane

- Creates a reflected object
- Symmetry element of the second kind
- Changes handedness of object



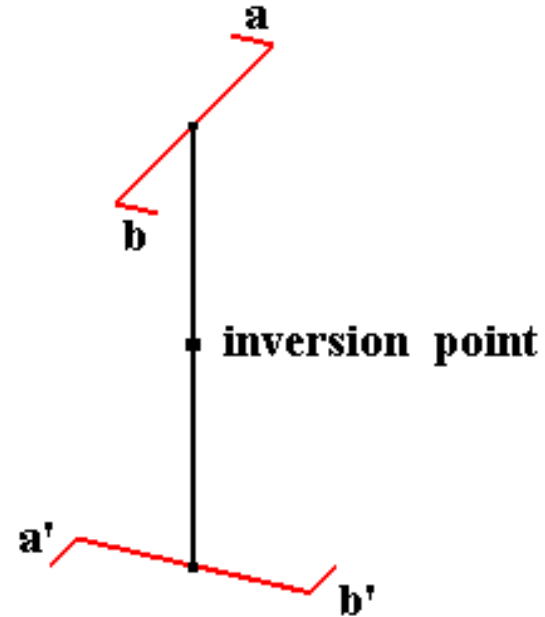
# Inversion Centre

- Transforms  $x, y, z$  into  $\bar{x}, \bar{y}, \bar{z}$
- Symmetry element of the second kind
- Changes handedness of object



# Rotary Inversion Axis

- Rotation of  $360^\circ/n$  followed by inversion
- Symmetry element of the second kind
- Changes handedness of object
- $\bar{1}$  is equivalent to an inversion centre
- $\bar{2}$  is equivalent to a mirror plane



# Characterizing Point Symmetry Rotation Matrices

- Rotation matrices can be characterized using some elementary matrix operations
  - Determinant
  - Trace
- Calculation of rotation angle
- Eigenvalues and Eigenvectors are used to give the orientation of the rotation axis



# Determinant and Trace of a Matrix

- Determinant
  - Characterizes some property of a matrix
  - For crystallographic symmetry rotation matrices the values the determinant are +1 or -1
  - +1 means the symmetry operation maintains the handedness of the object (symmetry operation of the first kind)
  - -1 means the symmetry operation changes the handedness of the object (symmetry operation of the second kind)
- Trace
  - The sum of the diagonal values of a matrix
  - The order of the rotation axis (n) can be derived from the trace

# Using Determinant and Trace to Characterize Symmetry Matrices

Symmetry Element	Determinant	Trace
1	1	3
2	1	-1
3	1	0
4	1	1
6	1	2
$\bar{1}$	-1	-3
$\bar{2}$	-1	1
$\bar{3}$	-1	0
$\bar{4}$	-1	-1
$\bar{6}$	-1	-2

- The combination of the determinant and the trace determines the order of the rotation and whether it is a proper or improper rotation

# Calculation of Rotation Angle from Determinant and Trace

$$\phi = \cos^{-1} \left( 0.5 \left( \frac{t}{d} - 1.0 \right) \right)$$

# Eigenvalues and Eigenvectors of Crystallographic Symmetry Rotation Matrices

- One of the eigenvectors gives the orientation of the rotation axis
- A 3x3 matrix can have 3 eigenvalues and 3 eigenvectors (can be real or complex quantities). How to choose the correct eigenvalue/eigenvector pair?
- Criteria:
  - Choose the **real** eigenvalue which has the **same sign** as the determinant
  - Choose the eigenvector which corresponds to the chosen eigenvalue

# Worked Example of Determining Orientation of Rotation Axis

*Trigonal Coordinate System*

$$\begin{pmatrix} 0 & \bar{1} & 0 \\ 1 & \bar{1} & 0 \\ 0 & 0 & 1 \end{pmatrix}$$

```
julia> three_fold_axis
3×3 Matrix{Int64}:
 0  -1  0
 1  -1  0
 0   0  1

julia> det(three_fold_axis)
1.0

julia> tr(three_fold_axis)
0

julia> eigvals(three_fold_axis)
3-element Vector{ComplexF64}:
 -0.4999999999999994 - 0.8660254037844386im
 -0.4999999999999994 + 0.8660254037844386im
 1.0 + 0.0im

julia> eigvecs(three_fold_axis)
3×3 Matrix{ComplexF64}:
 -0.707107-0.0im      -0.707107+0.0im      0.0+0.0im
 -0.353553-0.612372im -0.353553+0.612372im  0.0+0.0im
 0.0-0.0im           0.0+0.0im           1.0+0.0im

julia> 
```

# Understanding Hermann-Mauguin Notation for Point Groups

- See next slide for your Hermann-Mauguin Secret Decoder Ring!

Crystal System	1 <sup>st</sup> Position	2 <sup>nd</sup> Position	3 <sup>rd</sup> Position	Point Groups
Triclinic	Only one position, denoting all directions in crystal			$\bar{1}$ , 1
Monoclinic	Only 1 symbol: 2 or $\bar{2}$    to Y ( <i>b</i> is principal axis)			<b>2/m</b> , 2, m
Orthorhombic	2 and/or $\bar{2}$    to X	2 and/or $\bar{2}$    to Y	2 and/or $\bar{2}$    to Z	<b>mmm</b> , mm2, 222
Tetragonal	4 and/or $\bar{4}$    to Z	2 and/or $\bar{2}$    to X and Y	2 and/or $\bar{2}$    to [110]	<b>4/mmm</b> , $\bar{4}2m$ , 4mm, 422, 4/m, $\bar{4}$ , 4
Trigonal	3 and/or $\bar{3}$    to Z	2 and/or $\bar{2}$    to X, Y, U		<b><math>\bar{3}m</math></b> , 3m, 32, $\bar{3}$ , 3
Hexagonal	6 and/or $\bar{6}$    to Z	2 and/or $\bar{2}$    to X, Y, U	2 and/or $\bar{2}$ along [110]	<b>6/mmm</b> , $\bar{6}m\bar{2}$ , 6mm, 622, 6/m, $\bar{6}$ , 6
Cubic	2 and/or $\bar{2}$    to X, Y, Z	3 and/or $\bar{3}$    to [111]		$m\bar{3}$ , 23
	4 and/or $\bar{4}$    to X, Y, Z		2 and/or $\bar{2}$ along face diagonals	<b><math>m\bar{3}m</math></b> , $\bar{4}3m$ , 432

# Choosing the Correct Crystal System

- Do not assume the metric relations indicate the correct point group and crystal system!!!
- Correctly identify the Laue group symmetry of the diffraction pattern (equivalent intensities,  $R_{\text{sym}}$ )
- The Laue symmetry indicates the crystal system of your sample
- Correct Laue group assignment narrows space group choices



# Space Groups

- Space groups vs Point groups
  - Point groups describe symmetry of isolated objects
  - Space groups describe symmetry of infinitely repeating space filling objects
- Space groups include point symmetry elements