### **Recommendations for Successful Crystal Growth**

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#### Introduction

The successful growth of high-quality crystals is both an art and a science. This report compiles experiential and practical advice to guide researchers, particularly students and early-career scientists, in the practice of crystal growth for X-ray diffraction and other analytical techniques. While there is no single formula for success, adherence to the following principles will greatly increase the likelihood of obtaining diffraction-quality crystals.

#### **General Best Practices**

## 1. Purity of Material

- Ensure that the compound is as pure as possible; impurities inhibit crystal formation.
- A messy NMR spectrum does not preclude crystal growth, but higher purity typically yields better results.

## 2. Solution Preparation

- Always filter your solution before setting it aside for crystallization.
- Perform small-scale solubility tests before committing large quantities of solvent or material.

### 3. Glassware and Equipment

- o Use clean, residue-free glassware to prevent contamination.
- Homemade crystal growth vessels can be prepared by sealing cut glass tubes (e.g., bulk glass stock) with a flame.

# **Techniques and Conditions**

### 4. Slow Cooling

 For high-temperature crystal growths, gradual cooling is critical. Rapid thermal changes can damage samples and compromise safety. Heating solutions and cooling them slowly enhances nucleation and growth.

## 5. Evaporation and Vapor Diffusion

- o For slow evaporation, draw a mark at the solvent level to monitor evaporation rates.
- o Avoid letting solutions dry out completely.
- Vapor diffusion setups should be left undisturbed; layering and polarity adjustment may enhance success.

#### 6. Environment

- o Crystallizations should be placed in still, vibration-free environments.
- Maintain a cool, temperature-controlled setting, particularly during the summer months.
- Homemade insulated setups (e.g., vermiculite in a vessel) can stabilize temperature.

### **Behavioural Advice**

### 7. Timing and Patience

- o Set up crystallizations on Fridays and allow the weekend to work its magic.
- The best crystals often form over extended periods (weeks to months); benign neglect is a key strategy.
- o Be patient; good crystals form slowly. Avoid checking too frequently.

### 8. Mindset

- o Do not be discouraged by initial failures. Crystallization is inherently unpredictable.
- o Try many different conditions and solvent systems; persistence is essential.

### **Glovebox-Specific Guidance**

## 9. Crystallization in Gloveboxes

- o Avoid growing crystals in the glovebox unless absolutely necessary.
- For glovebox vapor diffusion setups, seal and leave them undisturbed for extended periods.
- Never bring the entire batch of an air-sensitive sample to the diffractometer; only a small, representative portion.

### Safety and Professionalism

# 10. Laboratory Conduct

- Do not allow others, including supervisors, to disturb or "pick at" your crystal growths prematurely.
- Avoid rapid heating and cooling of solids; this poses fire hazards and risks damaging the material.
- Keep your crystallographer happy by delivering well-formed, colourful crystals in appropriate containers.

### **Cultural Notes**

# 11. Respect the Crystals

- Change the polarity of the solvent when needed.
- Name and respect your crystals; avoid making eye contact (metaphorically speaking).
- Recognize that treating the process with care, humour, and patience can contribute to success.

#### Conclusion

Successful crystal growth relies on a combination of chemical understanding, experimental rigour, and disciplined patience. By following these recommendations and learning from both success and failure, researchers can improve their ability to grow high-quality crystals suitable for structural analysis.

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