# FIRST PRINCIPLE CALCULATIONS OF DEFECT STRUCTURES IN ZINC OXIDE

Ву

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An undergraduate thesis submitted in partial fulfillment of the requirements for the degree of

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# FIRST PRINCIPLE CALCULATIONS OF DEFECT STRUCTURES IN ZINC OXIDE

#### Abstract

by Christian Loer T. Llemit, BS University of the Philippines - Diliman March 2020

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# TABLE OF CONTENTS

				Page
ABS	ΓRA	.CT .		. ii
LIST	OF	TABL	LES	. viii
LIST	<b>OF</b>	FIGU	RES	. ix
СНА	PTE	$^{2}$ R		
1	INT	ГROD	UCTION	. 1
	1.1	Purpo	ose and Motivation	. 1
	1.2	Objec	etives	. 1
	1.3	Outlin	ne	. 1
2	Rev	view o	f Related Literature	. 2
	2.1	Semic	conductors	. 2
		2.1.1	Properties	. 2
		2.1.2	Applications of Semiconductors	. 2
		2.1.3	Defects in Semiconductors	. 2
	2.2	Zinc (	Oxide	. 2
		2.2.1	Crystal Structure	. 2
		2.2.2	Crystallographic Directions and Planes	. 3
		2.2.3	Brillouin Zone Symmetry	. 3
		2.2.4	Photoluminescence Properties	. 3
		2.2.5	Defects	. 3
3	$\mathbf{TH}$	EORE	ETICAL FRAMEWORK	. 4
	3.1	Electr	conic Structure	. 4
		3.1.1	Electronic Band structure	. 5
		3.1.	1.1 Band structure of free electron	. 6

		3.1.2	Density of States 6
		3.1.3	Projected Density of States
	3.2	Many	-body Quantum Mechanics
		3.2.1	Time Independent Schrödinger Equation
		3.2.2	Simplifying Assumptions
		3.2.3	Use of Atomic Units
		3.2.4	Hamiltonian Operator
		3.2.5	Indistinguishability of electrons
	3.3	Early	First Principle Calculations
		3.3.1	n-electron problem
		3.3.2	Hartree Method
		3.3.3	Hartree-Fock Method
	3.4	Densi	ty Functional Theory
		3.4.1	Electron Density
		3.4.2	Hohenberg-Kohn (HK) Formalism
		3.4.	2.1 First HK Theorem
		3.4.	2.2 Second HK Theorem
		3.4.3	Kohn Sham (KS) Formalism
		3.4.	3.1 KS Equation
		3.4.	3.2 Energy Terms
		3.4.4	Self Consistent Field Calculation
	3.5	Excha	ange-correlation Functional
		3.5.1	Local Density Approximation (LDA)
		3.5.2	Generalized Gradient Approximation (GGA) 9
	3.6	Corre	ctions to DFT
		3.6.1	GW Method
		3.6.2	Hybrid Functionals
		3.6.3	Hubbard U Correction
4	DF	T Cal	culation of Solids
	4.1	Basis	Sets
		4.1.1	Plane Wave
		4.1.2	Gaussian Orbital
		4.1.3	Slater type orbitals

4.2	Pseudopotential Approach	10 11 11
	4.2.3 Common Pseudopotentials	11
	4.2.3.1 Norm-Conserving PP	11
	4.2.3.2 Ultrasoft PP	11
4.3	4.2.3.3 Projector Augmented Wave	11 11 11
	4.3.1.1 Periodic Boundary Conditions (PBC)	11 11
	4.3.2.1 Reciprocal Lattice	11
	4.3.2.2 First Brillouin Zone	11
	4.3.2.3 Irreducible Brillouin Zone	11 11
	4.3.3.1 Monkhorst-Pack method	11
4.4	4.3.3.2 Gamma Point Sampling	11 13
	4.4.1 Electrons in solid	13 13 13
	4.4.3.1 Fourier Expansions	13
	4.4.3.2 Fast Fourier Transformation (FFT)	13
4.5	4.4.3.3 Kohn-Sham Matrix Representations	13 13 13
	4.5.1.1 Local Basis Set	13
	4.5.1.2 Plane Wave Basis Set	13 13
	4.5.2.1 Charge Density	13

		4.5.2.2 Kinetic Energy	13
		4.5.2.3 Effective Potential	13
	4.6	Electronic Structure	13
		4.6.1 Band Structure of free electrons	13
		4.6.2 Band Structure of electrons in solids	13
		4.6.3 Electronic Density of States	13
	4.7	Practical Aspects	13
		4.7.1 Energy Cutoffs	13
		4.7.1.1 Cutoff for Wavefunction	13
		4.7.1.2 Cutoff for Charge Density	13
		4.7.2 Smearing	13
		4.7.2.1 Gaussian Smearing	13
		4.7.2.2 Fermi Smearing	13
		4.7.2.3 Methfessel–Paxton Smearing	13
5	Sof	tware Implementation	14
	5.1	QUANTUM ESPRESSO	14
		5.1.1 MKL Libraries	14
		5.1.2 PWSCF routines	14
	5.2	Intel Compilers	15
	5.3		15
	5.4	Computational Details	15
			15
		5.4.2 Hubbard correction parameters	15
		5.4.3 Supercell creation	15
		5.4.4 Slab Model	15
			15
			15
			15
		5.4.8 dos calculation	15
APP	END	IX	
<b>A</b>			1 7

D	PPE	סי	T	N	C	Tr (	Q																					1.	6
	$\mathbf{E}$								•									•										2	2
	D					•		•			•							•								•		2	1
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	В							•			•							•										1	9

# LIST OF TABLES

# LIST OF FIGURES

A.1	Cost per raw megabase of DNA sequence from 2001 to 2015	 17
A.2	Cost per raw megabase of DNA sequence from 2001 to 2015	 18
B.1	Cost per raw megabase of DNA sequence from 2001 to 2015	 19
C.1	Cost per raw megabase of DNA sequence from 2001 to 2015	 20
D.1	Cost per raw megabase of DNA sequence from 2001 to 2015	 21
E.1	Cost per raw megabase of DNA sequence from 2001 to 2015	 22

#### Dedication

This dissertation/thesis is dedicated to my mother and father who provided both emotional and financial support

# Chapter One

# **INTRODUCTION**

### 1.1 Purpose and Motivation

Describe the importance of defects in ZnO

### 1.2 Objectives

Study the mechanisms of different defects in ZnO

### 1.3 Outline

This is an example of how to cite [Prades2009]

# Chapter Two

## Review of Related Literature

- 2.1 Semiconductors
- 2.1.1 Properties
- 2.1.2 Applications of Semiconductors
- 2.1.3 Defects in Semiconductors

### 2.2 Zinc Oxide

describe ZnO in broad perspective

### 2.2.1 Crystal Structure

Consider different phases

- 2.2.2 Crystallographic Directions and Planes
- 2.2.3 Brillouin Zone Symmetry
- 2.2.4 Photoluminescence Properties
- 2.2.5 Defects

### Chapter Three

### THEORETICAL FRAMEWORK

#### 3.1 Electronic Structure

The problem of electronic structure methods begins with the attempt to solve the general non-relativistic time-independent Schrödinger equation given as

$$\hat{\mathcal{H}}\Psi = E\Psi \tag{3.1}$$

where  $\hat{\mathcal{H}}$  is the Hamiltonian operator for a system of electrons,  $\Psi$  is the electronic wavefunction and E is the energy of the system. Consider a single electron in three dimensional system, the Schrödinger equation can be expressed as

$$\hat{\mathcal{H}}\Psi_n = -\frac{\hbar^2}{2m} \left( \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) \Psi_n + V \Psi_n = \epsilon_n \Psi_n \tag{3.2}$$

where m is the mass of electron, V is the effective potential energy and  $\epsilon_n$  is the energy of electron in the orbital. The term orbital denotes the solution of the Schrödinger equation for a system of only one electron. This will be useful in later sections because this will allow to distinguish between the exact quantum state of a system of N interacting electrons from the approximate quantum state of N electrons in N orbitals, where each orbital is a solution to one-electron wavefunction in (3.2). If V is zero for the case of free electrons (i.e. non-interacting), then the orbital model is exact.

Since electrons are restricted by the potential inside the atom, the simplest way of solving (3.2) is by considering an infinite potential well. The electrons are confined inside a cube of length L where the potential V inside is zero and infinite at outside must satisfy the boundary condition

$$\Psi_n(L_x, L_y, L_z) = 0 \tag{3.3}$$

where  $L_x, L_y, L_z$  can be either 0 or L. The solution will have a sine dependence

$$\Psi_n(x,y,z) = \sqrt{\left(\frac{2}{L}\right)^3} \sin\left(\frac{n_x \pi}{L}x\right) \sin\left(\frac{n_y \pi}{L}y\right) \sin\left(\frac{n_z \pi}{L}z\right)$$
(3.4)

where  $n_x, n_y, n_z$  are integer quantum states. Provided that  $k_i = n_i \pi / L$  where i = x, y, or z; then the energy dispersion relation can be expressed as

$$\epsilon_k = \frac{\hbar^2}{2m} (k_x^2 + k_y^2 + k_z^2) = \frac{\hbar^2}{2m} k^2 \propto k^2$$
 (3.5)

Note that energy levels are discretized by the quantum states which arises from imposing the boundary conditions.

#### 3.1.1 Electronic Band structure

Inside the crystal lattice, the periodic arrangement of atoms or ions causes the potential to be periodic which eventually gives rise to the formation of energy bands. The wavefunction  $\Psi$  will become periodic in space with a period L and must obey the Born-von Karman boundary condition

$$\Psi_k(x, y, z) = \Psi_k(x + L, y, z) \tag{3.6}$$

and similarly for the y and z coordinates. It can be shown that wavefunctions satisfying (3.2) and (3.6) are the Bloch form of a travelling plane wave

$$\Psi_k(\vec{\mathbf{r}}) = u_k(\vec{\mathbf{r}}) \exp(i\vec{\mathbf{k}} \cdot \vec{\mathbf{r}})$$
(3.7)

where  $u_k(\vec{\mathbf{r}})$  has the period of the crystal lattice with  $u_k(\vec{\mathbf{r}}) = u_k(\vec{\mathbf{r}} + \vec{\mathbf{R}})$ . Here  $\vec{\mathbf{R}}$  is the translation vector which can be simply thought as the periodicity expressed as vector. The Bloch expression can be written as

$$\Psi_{k}(\vec{\mathbf{r}} + \vec{\mathbf{R}}) = u_{k}(\vec{\mathbf{r}} + \vec{\mathbf{R}}) \exp\left(i\vec{\mathbf{k}} \cdot (\vec{\mathbf{r}} + \vec{\mathbf{R}})\right) 
\Psi_{k}(\vec{\mathbf{r}} + \vec{\mathbf{R}}) = u_{k}(\vec{\mathbf{r}}) \exp\left(i\vec{\mathbf{k}} \cdot \vec{\mathbf{r}}\right) \exp\left(i\vec{\mathbf{k}} \cdot \vec{\mathbf{R}}\right) 
\Psi_{k}(\vec{\mathbf{r}} + \vec{\mathbf{R}}) = \Psi_{k}(\vec{\mathbf{r}}) \exp\left(i\vec{\mathbf{k}} \cdot \vec{\mathbf{R}}\right)$$
(3.8)

Notice that the wavefunction differs from the plane wave of free electrons only by a periodic modulation given by the new phase factor. This means that the electrons in the crystal lattice are treated as perturbed weakly by the periodic potential of the ion cores.

#### 3.1.1.1 Band structure of free electron

A special case of periodicity is where the potential is set to zero, which is applicable for the free electrons. The wavefunction will be a plane wave

$$\Psi_k(\vec{\mathbf{r}}) = \exp\left(i\vec{\mathbf{k}}\cdot\vec{\mathbf{r}}\right) \tag{3.9}$$

that represents travelling wave with a momentum  $\vec{\mathbf{p}} = \hbar \vec{\mathbf{k}}$ . The energy dispersion relation is still given by (3.5) but this time the allowed energy values are distributed essentially from zero to infinity. Figure 3.1 shows the parabolic dependence of energy with the wavevector k. Figure 3.1a shows the extended scheme insert the symmetry points in IBZ.

#### 3.1.2 Density of States

explains fermi dirac distribution

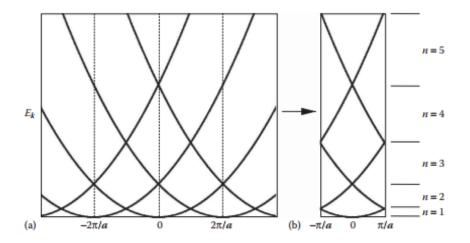


Figure 3.1 Free electron band structure

### 3.1.3 Projected Density of States

### 3.2 Many-body Quantum Mechanics

insert text here

- Time Independent Schrödinger Equation 3.2.1 Simplifying Assumptions 3.2.2 3.2.3 Use of Atomic Units Hamiltonian Operator 3.2.4 3.2.5 Indistinguishability of electrons Early First Principle Calculations 3.3 n-electron problem 3.3.1 Hartree Method 3.3.2 3.3.3 Hartree-Fock Method 3.4 **Density Functional Theory Electron Density** 3.4.13.4.2 Hohenberg-Kohn (HK) Formalism First HK Theorem 3.4.2.1 3.4.2.2 Second HK Theorem 3.4.3 Kohn Sham (KS) Formalism 3.4.3.1 KS Equation
- 3.4.3.2 Energy Terms
- 3.4.4 Self Consistent Field Calculation
- 3.5 Exchange-correlation Functional

# Chapter Four

# **DFT Calculation of Solids**

- 4.1 Basis Sets
- 4.1.1 Plane Wave
- 4.1.2 Gaussian Orbital
- 4.1.3 Slater type orbitals

### 4.2 Pseudopotential Approach

This is sample text

4.2.1 Free	ezing the core electrons
4.2.2 Pseu	udizing the valence electrons
4.2.3 Con	nmon Pseudopotentials
4.2.3.1 Nor	m-Conserving PP
4.2.3.2 Ultr	easoft PP
4.2.3.3 Pro	jector Augmented Wave
4.3 Cho	osing the appropriate Calculation Size
4.3.1 Use	of Supercell
4.3.1.1 Peri	odic Boundary Conditions (PBC)
4.3.2 Use	of Reciprocal Space
4.3.2.1 Reci	iprocal Lattice
4.3.2.2 Firs	t Brillouin Zone
4.3.2.3 Irrec	ducible Brillouin Zone
4.3.3 k-pc	oint sampling
4.3.3.1 Mor	nkhorst-Pack method
4.3.3.2 Gan	ama Point Sampling

Example of double quotes "word". Lore

### 4.4 Bloch Representations

- 4.4.1 Electrons in solid
- 4.4.2 Bloch Theorem in periodic systems
- 4.4.3 Fourier Expansion of Bloch representations
- 4.4.3.1 Fourier Expansions
- 4.4.3.2 Fast Fourier Transformation (FFT)
- 4.4.3.3 Kohn-Sham Matrix Representations

### 4.5 Plane Wave (PW) Expansion

- 4.5.1 Basis Set
- 4.5.1.1 Local Basis Set
- 4.5.1.2 Plane Wave Basis Set
- 4.5.2 Plane Wave Expansion for KS quantities
- 4.5.2.1 Charge Density
- 4.5.2.2 Kinetic Energy
- 4.5.2.3 Effective Potential

#### 4.6 Electronic Structure

- 4.6.1 Band Structure of free electrons
- 4.6.2 Band Structure of electrons in solids
- 4.6.3 Electronic Density of States

### 4.7 Practical Aspects

# Chapter Five

# Software Implementation

## 5.1 QUANTUM ESPRESSO

- 5.1.1 MKL Libraries
- 5.1.2 PWSCF routines

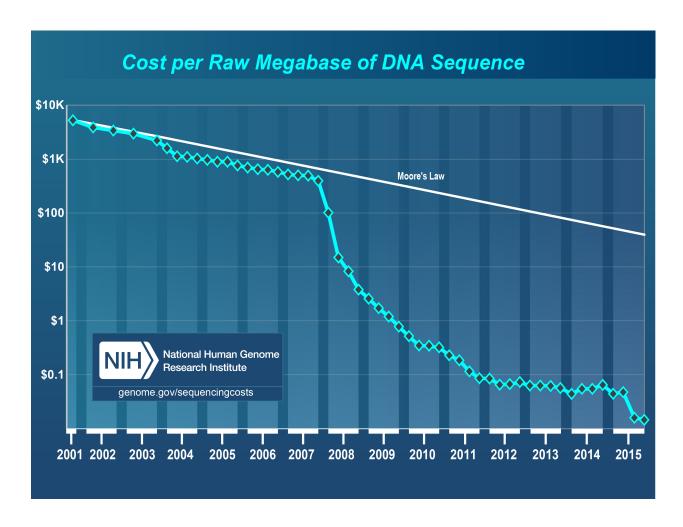
cbands, cegterg, cdiaghg

- 5.2 Intel Compilers
- 5.3 Executables
- 5.4 Computational Details
- 5.4.1 Convergence Testing
- 5.4.2 Hubbard correction parameters
- 5.4.3 Supercell creation
- 5.4.4 Slab Model
- 5.4.5 Structural relaxation
- 5.4.6 scf calculation
- 5.4.7 bandstructure calculation
- 5.4.8 dos calculation

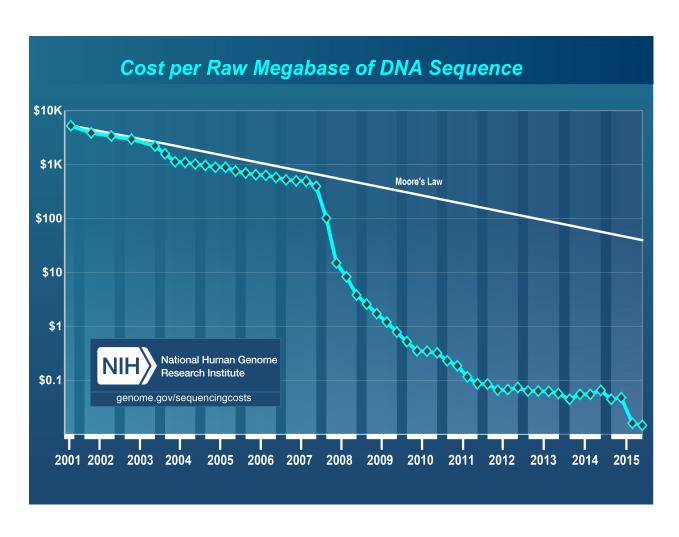
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# Appendix A

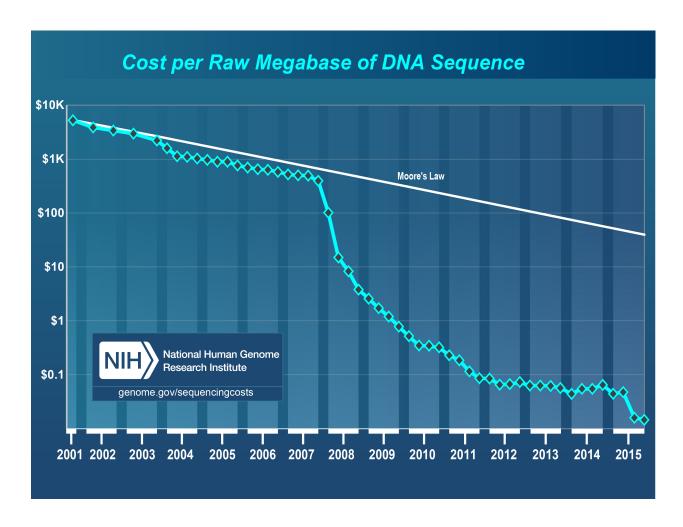


**Figure A.1** Cost per raw megabase of DNA sequence from 2001 to 2015. Straight line - Moore's Law, blue curve - cost in US dollars, Y-axis scale is logarithmic. Graph reproduced from [wetterstrand2016]



**Figure A.2** Cost per raw megabase of DNA sequence from 2001 to 2015. Straight line - Moore's Law, blue curve - cost in US dollars, Y-axis scale is logarithmic. Graph reproduced from [wetterstrand2016]

# Appendix B



**Figure B.1** Cost per raw megabase of DNA sequence from 2001 to 2015. Straight line - Moore's Law, blue curve - cost in US dollars, Y-axis scale is logarithmic. Graph reproduced from [wetterstrand2016]

# Appendix C

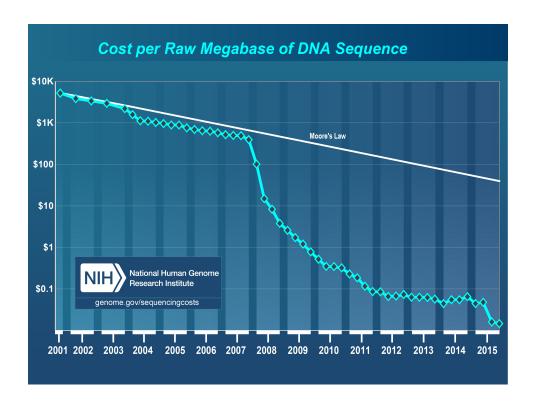
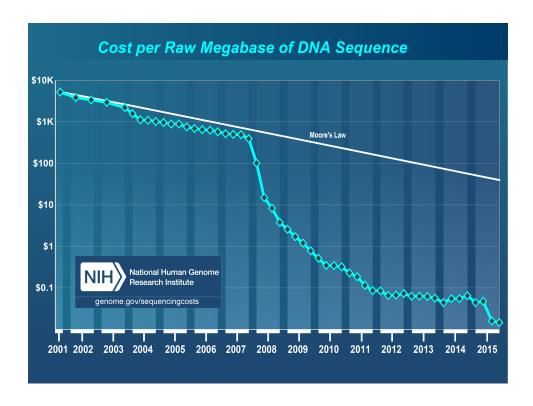


Figure C.1 Cost per raw megabase of DNA sequence from 2001 to 2015. Straight line - Moore's Law, blue curve - cost in US dollars, Y-axis scale is logarithmic. Graph reproduced from [wetterstrand2016]

# Appendix D



**Figure D.1** Cost per raw megabase of DNA sequence from 2001 to 2015. Straight line - Moore's Law, blue curve - cost in US dollars, Y-axis scale is logarithmic. Graph reproduced from [wetterstrand2016]

# Appendix E



Figure E.1 Cost per raw megabase of DNA sequence from 2001 to 2015. Straight line - Moore's Law, blue curve - cost in US dollars, Y-axis scale is logarithmic. Graph reproduced from [wetterstrand2016]