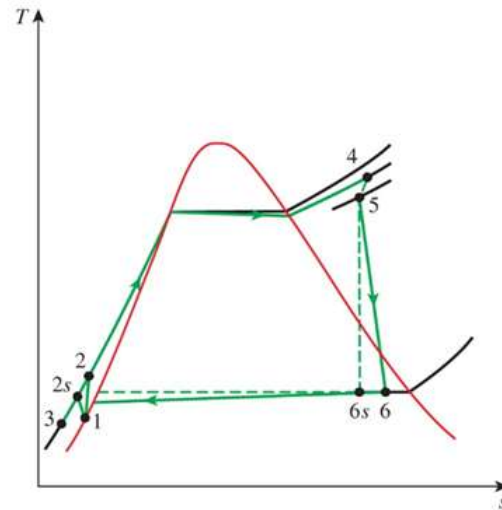
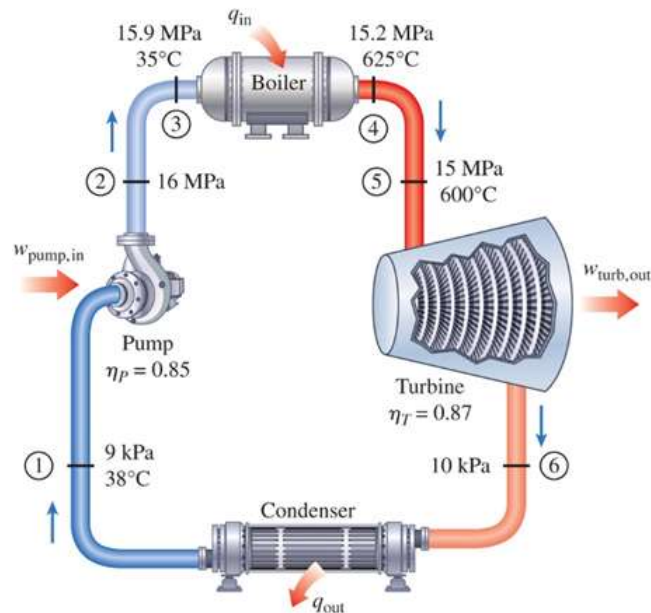


Given:

A steam power plant operates on the cycle shown below.

$$\begin{array}{llll}
 P_1 := 9 \text{ kPa} & T_1 := 38 \text{ }^\circ\text{C} & P_4 := 15.2 \text{ MPa} & T_4 := 625 \text{ }^\circ\text{C} & \eta_p := 0.85 \\
 P_2 := 16 \text{ MPa} & & P_5 := 15 \text{ MPa} & T_5 := 600 \text{ }^\circ\text{C} & \eta_t := 0.87 \\
 P_3 := 15.9 \text{ MPa} & T_3 := 35 \text{ }^\circ\text{C} & P_6 := 10 \text{ kPa} & &
 \end{array}$$

**Required:**

If the isentropic efficiency of the turbine is 87% and the isentropic efficiency of the pump is 85%, determine the thermal efficiency of the cycle and the net power output for a mass flow rate of 15 kg/s.

Solution:

The mass flow rate of the cycle is defined as

$$m' := 15 \frac{\text{kg}}{\text{s}}$$

Going to Table A-4 @ $T_1 = 38 \text{ }^\circ\text{C}$ and $P_1 = 9 \text{ kPa}$ shows that the state is compressed liquid and will be approximated as a saturated liquid.

$$\begin{array}{ll}
 T_a := 35 \text{ }^\circ\text{C} & T_b := 40 \text{ }^\circ\text{C} \\
 v_a := 0.001006 \frac{\text{m}^3}{\text{kg}} & v_b := 0.001008 \frac{\text{m}^3}{\text{kg}} \\
 v_1 := \frac{T_1 - T_a}{T_b - T_a} \cdot (v_b - v_a) + v_a = 0.001007 \frac{\text{m}^3}{\text{kg}}
 \end{array}$$

The specific isentropic work of the pump is given by

$$w_{ps} := v_1 \cdot (P_2 - P_1) = 16.11 \frac{\text{kJ}}{\text{kg}} \quad (\text{since the fluid is incompressible})$$

Solution (contd.):

The actual work of the pump is then found by the definition of the isentropic efficiency of the pump. This is shown below.

$$\eta_p = \frac{w_{ps}}{w_{pa}} \quad \text{or} \quad w_{pa} := \frac{w_{ps}}{\eta_p} = 18.95 \frac{\text{kJ}}{\text{kg}}$$

Going to Table A-5 @ $T_5 = 600.0^\circ\text{C}$ and $P_5 = 15000 \text{ kPa}$ shows that the state is superheated.

Going to Table A-6 @ $T_5 = 600.0^\circ\text{C}$ and $P_5 = 15.00 \text{ MPa}$ shows

$$h_5 := 3583.1 \frac{\text{kJ}}{\text{kg}} \quad s_5 := 6.6796 \frac{\text{kJ}}{\text{kg K}}$$

For the ideal cycle, the specific entropy at state 5 and state 6 are the same.

$$s_{6s} := s_5 = 6.680 \frac{\text{kJ}}{\text{kg K}}$$

Going to Table A-5 @ $P_6 = 10.00 \text{ kPa}$ and $s_{6s} = 6.680 \frac{\text{kJ}}{\text{kg K}}$ shows that the state is in the two phase region.

$$s_f := 0.6492 \frac{\text{kJ}}{\text{kg K}} \quad s_g := 8.1488 \frac{\text{kJ}}{\text{kg K}} \quad h_f := 191.81 \frac{\text{kJ}}{\text{kg}} \quad h_g := 2583.9 \frac{\text{kJ}}{\text{kg}}$$

$$x_{6s} := \frac{s_{6s} - s_f}{s_g - s_f} = 0.8041 \quad h_{6s} := h_f + x_{6s} \cdot (h_g - h_f) = 2115 \frac{\text{kJ}}{\text{kg}}$$

The specific work of the turbine for the ideal case may then be found by

$$w_{ts} := h_5 - h_{6s} = 1468 \frac{\text{kJ}}{\text{kg}}$$

The specific work of the turbine for the actual case may then be found by using the definition of isentropic efficiency. This is shown below.

$$\eta_t = \frac{w_{ta}}{w_{ts}} \quad \text{or} \quad w_{ta} := \eta_t \cdot w_{ts} = 1277 \frac{\text{kJ}}{\text{kg}}$$

Going to Table A-5 @ $T_3 = 35.00^\circ\text{C}$ and $P_3 = 15900 \text{ kPa}$ shows the state is a compressed liquid but in this case, we can actually use the compressed liquid tables.

Going to Table A-7 @ $T_3 = 35.00^\circ\text{C}$ and $P_3 = 15.90 \text{ MPa}$ shows that double interpolation is needed.

$$P_a := 15 \text{ MPa}$$

$$P_b := 20 \text{ MPa}$$

$$T_a := 20^\circ\text{C} \quad h_{aa} := 97.93 \frac{\text{kJ}}{\text{kg}} \quad h_{ab} := 102.57 \frac{\text{kJ}}{\text{kg}}$$

$$T_b := 40^\circ\text{C} \quad h_{ba} := 180.77 \frac{\text{kJ}}{\text{kg}} \quad h_{bb} := 185.16 \frac{\text{kJ}}{\text{kg}}$$

$$h_{a3} := \frac{P_3 - P_a}{P_b - P_a} \cdot (h_{ab} - h_{aa}) + h_{aa} = 98.77 \frac{\text{kJ}}{\text{kg}}$$

$$h_{b3} := \frac{P_3 - P_a}{P_b - P_a} \cdot (h_{bb} - h_{ba}) + h_{ba} = 181.6 \frac{\text{kJ}}{\text{kg}}$$

$$h_3 := \frac{T_3 - T_a}{T_b - T_a} \cdot (h_{b3} - h_{a3}) + h_{a3} = 160.9 \frac{\text{kJ}}{\text{kg}}$$

Solution (contd.):

Going to Table A-5 @ $T_4 = 625.0^\circ\text{C}$ and $P_4 = 15200\text{ kPa}$ shows the state is superheated.

Going to Table A-6 @ $T_4 = 625.0^\circ\text{C}$ and $P_4 = 15.20\text{ MPa}$ shows double interpolation is needed.

$$P_a := 15\text{ MPa}$$

$$P_b := 17.5\text{ MPa}$$

$$T_a := 600^\circ\text{C} \quad h_{aa} := 3583.1 \frac{\text{kJ}}{\text{kg}} \quad h_{ab} := 3561.3 \frac{\text{kJ}}{\text{kg}}$$

$$T_b := 650^\circ\text{C} \quad h_{ba} := 3712.1 \frac{\text{kJ}}{\text{kg}} \quad h_{bb} := 3693.8 \frac{\text{kJ}}{\text{kg}}$$

$$h_{a4} := \frac{P_4 - P_a}{P_b - P_a} \cdot (h_{ab} - h_{aa}) + h_{aa} = 3581 \frac{\text{kJ}}{\text{kg}}$$

$$h_{b4} := \frac{P_4 - P_a}{P_b - P_a} \cdot (h_{bb} - h_{ba}) + h_{ba} = 3711 \frac{\text{kJ}}{\text{kg}}$$

$$h_4 := \frac{T_4 - T_a}{T_b - T_a} \cdot (h_{b4} - h_{a4}) + h_{a4} = 3646 \frac{\text{kJ}}{\text{kg}}$$

The specific heat added to the cycle is given by

$$q_{in} := h_4 - h_3 = 3485 \frac{\text{kJ}}{\text{kg}}$$

The specific net work of the cycle is given by

$$w_{net} := w_{ta} - w_{pa} = 1258 \frac{\text{kJ}}{\text{kg}}$$

The thermal efficiency is then found by

$$\eta_{th} := \frac{w_{net}}{q_{in}} = 36.10\%$$

The power produced by the power plant is then found by

$$\dot{W}'_{net} := \dot{m} \cdot w_{net} = 18.87\text{ MW}$$