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TABLE A-1

Molar mass, gas constant, and critical-point properties

			Gas	Critical-p	S	
Substance	Formula	Molar mass, <i>M</i> kg/kmol	constant, R kJ/kg·K*	Temperature, K	Pressure, MPa	Volume, m³/kmol
Air	_	28.97	0.2870	132.5	3.77	0.0883
Ammonia	NH_3	17.03	0.4882	405.5	11.28	0.0724
Argon	Ar	39.948	0.2081	151	4.86	0.0749
Benzene	C_6H_6	78.115	0.1064	562	4.92	0.2603
Bromine	Br ₂	159.808	0.0520	584	10.34	0.1355
<i>n</i> -Butane	C_4H_{10}	58.124	0.1430	425.2	3.80	0.2547
Carbon dioxide	CO_2	44.01	0.1889	304.2	7.39	0.0943
Carbon monoxide	CO	28.011	0.2968	133	3.50	0.0930
Carbon tetrachloride	CCI ₄	153.82	0.05405	556.4	4.56	0.2759
Chlorine	Cl ₂	70.906	0.1173	417	7.71	0.1242
Chloroform	CHCI ₃	119.38	0.06964	536.6	5.47	0.2403
Dichlorodifluoromethane $(R-12)$	CCI ₂ F ₂	120.91	0.06876	384.7	4.01	0.2179
Dichlorofluoromethane (R-21)	CHČI ₂ F	102.92	0.08078	451.7	5.17	0.1973
Ethane	C ₂ H ₆	30.070	0.2765	305.5	4.48	0.1480
Ethyl alcohol	C ₂ H ₅ OH	46.07	0.1805	516	6.38	0.1673
Ethylene	C_2H_4	28.054	0.2964	282.4	5.12	0.1242
Helium	He	4.003	2.0769	5.3	0.23	0.0578
<i>n</i> -Hexane	C_6H_{14}	86.179	0.09647	507.9	3.03	0.3677
Hydrogen (normal)	H ₂	2.016	4.1240	33.3	1.30	0.0649
Krypton	Kr	83.80	0.09921	209.4	5.50	0.0924
Methane	CH ₄	16.043	0.5182	191.1	4.64	0.0993
Methyl alcohol	CH₃OH	32.042	0.2595	513.2	7.95	0.1180
Methyl chloride	CH ₃ CI	50.488	0.1647	416.3	6.68	0.1430
Neon	Ne	20.183	0.4119	44.5	2.73	0.0417
Nitrogen	N_2	28.013	0.2968	126.2	3.39	0.0899
Nitrous oxide	N_2O	44.013	0.1889	309.7	7.27	0.0961
Oxygen	02	31.999	0.2598	154.8	5.08	0.0780
Propane	C ₃ H ₈	44.097	0.1885	370	4.26	0.1998
Propylene	C_3H_6	42.081	0.1976	365	4.62	0.1810
Sulfur dioxide	SO ₂	64.063	0.1298	430.7	7.88	0.1217
Tetrafluoroethane (R-134a)	CF ₃ CH ₂ F	102.03	0.08149	374.2	4.059	0.1993
Trichlorofluoromethane $(R-11)$	CCI ₃ F	137.37	0.06052	471.2	4.38	0.2478
Water	H_2O	18.015	0.4615	647.1	22.06	0.0560
Xenon	Xe	131.30	0.06332	289.8	5.88	0.1186

^{*}The unit kJ/kg·K is equivalent to kPa·m³/kg·K. The gas constant is calculated from $R = R_u/M$, where $R_u = 8.31447$ kJ/kmol·K and M is the molar mass.

Source of Data: K. A. Kobe and R. E. Lynn, Jr., Chemical Review 52 (1953), pp. 117–236; and ASHRAE, Handbook of Fundamentals (Atlanta, GA: American Society of Heating, Refrigerating and Air—Conditioning Engineers, Inc., 1993), pp. 16.4 and 36.1.

TABLE A-2

Ideal-gas specific heats of various common gases

(a) At 300 K

Gas	Formula	Gas constant, <i>R</i> kJ/kg∙K	$c_{_{ ho}}$ kJ/kg \cdot K	<i>c</i> √ kJ/kg∙K	k
Air	_	0.2870	1.005	0.718	1.400
Argon	Ar	0.2081	0.5203	0.3122	1.667
Butane	C_4H_{10}	0.1433	1.7164	1.5734	1.091
Carbon dioxide	CO_2	0.1889	0.846	0.657	1.289
Carbon monoxide	CO	0.2968	1.040	0.744	1.400
Ethane	C_2H_6	0.2765	1.7662	1.4897	1.186
Ethylene	C_2H_4	0.2964	1.5482	1.2518	1.237
Helium	He	2.0769	5.1926	3.1156	1.667
Hydrogen	H_2	4.1240	14.307	10.183	1.405
Methane	CH₄	0.5182	2.2537	1.7354	1.299
Neon	Ne	0.4119	1.0299	0.6179	1.667
Nitrogen	N_2	0.2968	1.039	0.743	1.400
Octane	C ₈ H ₁₈	0.0729	1.7113	1.6385	1.044
Oxygen	02	0.2598	0.918	0.658	1.395
Propane	C ₃ H ₈	0.1885	1.6794	1.4909	1.126
Steam	H ₂ 0	0.4615	1.8723	1.4108	1.327

Note: The unit kJ/kg·K is equivalent to kJ/kg·°C.

Source of Data: B. G. Kyle, Chemical and Process Thermodynamics, 3rd ed. (Upper Saddle River, NJ: Prentice Hall, 2000).

TABLE A-2

Ideal-gas specific heats of various common gases (Continued)

(b) At various temperatures

Temperature,	<i>c_p</i> kJ/kg⋅K	<i>c</i> √ kJ/kg∙K	k	<i>c_p</i> kJ/kg⋅K	<i>c</i> _v kJ/kg⋅K	k	<i>c_p</i> kJ/kg⋅K	<i>c</i> , kJ/kg∙K	k
K		Air		Car	bon dioxide,	CO ₂	Carb	on monoxide	, CO
250	1.003	0.716	1.401	0.791	0.602	1.314	1.039	0.743	1.400
300	1.005	0.718	1.400	0.846	0.657	1.288	1.040	0.744	1.399
350	1.008	0.721	1.398	0.895	0.706	1.268	1.043	0.746	1.398
400	1.013	0.726	1.395	0.939	0.750	1.252	1.047	0.751	1.395
450	1.020	0.733	1.391	0.978	0.790	1.239	1.054	0.757	1.392
500	1.029	0.742	1.387	1.014	0.825	1.229	1.063	0.767	1.387
550	1.040	0.753	1.381	1.046	0.857	1.220	1.075	0.778	1.382
600	1.051	0.764	1.376	1.075	0.886	1.213	1.087	0.790	1.376
650	1.063	0.776	1.370	1.102	0.913	1.207	1.100	0.803	1.370
700	1.075	0.788	1.364	1.126	0.937	1.202	1.113	0.816	1.364
750	1.087	0.800	1.359	1.148	0.959	1.197	1.126	0.829	1.358
800	1.099	0.812	1.354	1.169	0.980	1.193	1.139	0.842	1.353
900	1.121	0.834	1.344	1.204	1.015	1.186	1.163	0.866	1.343
1000	1.142	0.855	1.336	1.234	1.045	1.181	1.185	0.888	1.335
		Hydrogen, H ₂			Nitrogen, N ₂	?		Oxygen, O ₂	
250	14.051	9.927	1.416	1.039	0.742	1.400	0.913	0.653	1.398
300	14.307	10.183	1.405	1.039	0.743	1.400	0.918	0.658	1.395
350	14.427	10.302	1.400	1.041	0.744	1.399	0.928	0.668	1.389
400	14.476	10.352	1.398	1.044	0.747	1.397	0.941	0.681	1.382
450	14.501	10.377	1.398	1.049	0.752	1.395	0.956	0.696	1.373
500	14.513	10.389	1.397	1.056	0.759	1.391	0.972	0.712	1.365
550	14.530	10.405	1.396	1.065	0.768	1.387	0.988	0.728	1.358
600	14.546	10.422	1.396	1.075	0.778	1.382	1.003	0.743	1.350
650	14.571	10.447	1.395	1.086	0.789	1.376	1.017	0.758	1.343
700	14.604	10.480	1.394	1.098	0.801	1.371	1.031	0.771	1.337
750	14.645	10.521	1.392	1.110	0.813	1.365	1.043	0.783	1.332
800	14.695	10.570	1.390	1.121	0.825	1.360	1.054	0.794	1.327
900	14.822	10.698	1.385	1.145	0.849	1.349	1.074	0.814	1.319
1000	14.983	10.859	1.380	1.167	0.870	1.341	1.090	0.830	1.313

Source of Data: Kenneth Wark, Thermodynamics, 4th ed. (New York: McGraw-Hill, 1983), p. 783, Table A-4M. Originally published in Tables of Thermal Properties of Gases, NBS Circular 564, 1955.

TABLE A-2

Ideal-gas specific heats of various common gases (Concluded)

(c) As a function of temperature

$$\overline{c}_p = a + bT + cT^2 + dT^3$$

(T in K, c_p in kJ/kmol·K)

						Temperature	% e	rror
Substance	Formula	а	b	С	d	range, K	Max.	Avg.
Nitrogen	N_2	28.90	-0.1571×10^{-2}	0.8081×10^{-5}	-2.873×10^{-9}	273-1800	0.59	0.34
Oxygen	02	25.48	1.520×10^{-2}	-0.7155×10^{-5}	1.312×10^{-9}	273-1800	1.19	0.28
Air	_	28.11	0.1967×10^{-2}	0.4802×10^{-5}	-1.966×10^{-9}	273-1800	0.72	0.33
Hydrogen	H_2	29.11	-0.1916×10^{-2}	0.4003×10^{-5}	-0.8704×10^{-9}	273-1800	1.01	0.26
Carbon	_							
monoxide	CO	28.16	0.1675×10^{-2}	0.5372×10^{-5}	-2.222×10^{-9}	273-1800	0.89	0.37
Carbon								
dioxide	CO ₂	22.26	5.981×10^{-2}	-3.501×10^{-5}	7.469×10^{-9}	273-1800	0.67	0.22
Water vapor	H ₂ Ō	32.24	0.1923×10^{-2}	1.055×10^{-5}	-3.595×10^{-9}	273-1800	0.53	0.24
Nitric oxide	NO	29.34	-0.09395×10^{-2}	0.9747×10^{-5}	-4.187×10^{-9}	273-1500	0.97	0.36
Nitrous oxide	N_2O	24.11	5.8632×10^{-2}	-3.562×10^{-5}	10.58×10^{-9}	273-1500	0.59	0.26
Nitrogen	_							
dioxide	NO_2	22.9	5.715×10^{-2}	-3.52×10^{-5}	7.87×10^{-9}	273-1500	0.46	0.18
Ammonia	NH ₃	27.568	2.5630×10^{-2}	0.99072×10^{-5}	-6.6909×10^{-9}	273-1500	0.91	0.36
Sulfur	S_2	27.21	2.218×10^{-2}	-1.628×10^{-5}	3.986×10^{-9}	273-1800	0.99	0.38
Sulfur	2							
dioxide	SO ₂	25.78	5.795×10^{-2}	-3.812×10^{-5}	8.612×10^{-9}	273-1800	0.45	0.24
Sulfur	_							
trioxide	SO ₃	16.40	14.58×10^{-2}	-11.20×10^{-5}	32.42×10^{-9}	273-1300	0.29	0.13
Acetylene	C_2H_2	21.8	9.2143×10^{-2}	-6.527×10^{-5}	18.21×10^{-9}	273-1500	1.46	0.59
Benzene		-36.22	48.475×10^{-2}	-31.57×10^{-5}	77.62×10^{-9}	273-1500	0.34	0.20
Methanol	CH₄Ö	19.0	9.152×10^{-2}	-1.22×10^{-5}	-8.039×10^{-9}	273-1000	0.18	0.08
Ethanol	C_2H_6O	19.9	20.96×10^{-2}	-10.38×10^{-5}	20.05×10^{-9}	273-1500	0.40	0.22
Hydrogen	2 0							
chloride	HCI	30.33	-0.7620×10^{-2}	1.327×10^{-5}	-4.338×10^{-9}	273-1500	0.22	0.08
Methane	CH₄	19.89	5.024×10^{-2}	1.269×10^{-5}	-11.01×10^{-9}	273-1500	1.33	0.57
Ethane	C_2H_6	6.900	17.27×10^{-2}	-6.406×10^{-5}	7.285×10^{-9}	273-1500	0.83	0.28
Propane	C ₃ H ₈	-4.04	30.48×10^{-2}	-15.72×10^{-5}	31.74×10^{-9}	273-1500	0.40	0.12
n−Butane	C_4H_{10}	3.96	37.15×10^{-2}	-18.34×10^{-5}	35.00×10^{-9}	273-1500	0.54	0.24
<i>i</i> -Butane	C ₄ H ₁₀	-7.913	41.60×10^{-2}	-23.01×10^{-5}	49.91×10^{-9}	273-1500	0.25	0.13
<i>n</i> -Pentane	C ₅ H ₁₂	6.774	45.43×10^{-2}	-22.46×10^{-5}	42.29×10^{-9}	273-1500	0.56	0.21
<i>n</i> −Hexane	C ₆ H ₁₄	6.938	55.22×10^{-2}	-28.65×10^{-5}	57.69×10^{-9}	273-1500	0.72	0.20
Ethylene	C_2H_4	3.95	15.64×10^{-2}	-8.344×10^{-5}	17.67×10^{-9}	273-1500	0.54	0.13
Propylene	$C_3^2H_6^4$	3.15	23.83×10^{-2}	-12.18×10^{-5}	24.62×10^{-9}	273-1500	0.73	0.17

Source of Data: B. G. Kyle, Chemical and Process Thermodynamics (Englewood Cliffs, NJ: Prentice-Hall, 1984).

TABLE A-3

Properties of common liquids, solids, and foods

(a) Liquids

	Boiling	data at 1 atm	Freez	zing data		Liquid p	properties
Substance	Normal boiling point, °C	Latent heat of vaporization $h_{\rm fg}$, kJ/kg	Freezing point, °C	Latent heat of fusion h_{if} , kJ/kg	Temperature, °C	Density $ ho$, kg/m ³	Specific heat c_p , kJ/kg·K
Ammonia	-33.3	1357	-77.7	322.4	-33.3 -20 0 25	682 665 639 602	4.43 4.52 4.60 4.80
Argon Benzene Brine (20% sodium	-185.9 80.2	161.6 394	-189.3 5.5	28 126	-185.6 20	1394 879	1.14 1.72
chloride by mass) n—Butane Carbon dioxide	103.9 -0.5 -78.4*	— 385.2 230.5 (at 0°C)	-17.4 -138.5 -56.6	— 80.3	20 -0.5 0	1150 601 298	3.11 2.31 0.59
Ethanol Ethyl alcohol Ethylene glycol	78.2 78.6 198.1	838.3 855 800.1	-114.2 -156 -10.8	109 108 181.1	25 20 20	783 789 1109	2.46 2.84 2.84
Glycerine Helium Hydrogen	179.9 -268.9 -252.8	974 22.8 445.7	18.9 — –259.2	200.6 — 59.5	20 -268.9 -252.8	1261 146.2 70.7	2.32 22.8 10.0
Isobutane Kerosene Mercury	-11.7 204-293 356.7	367.1 251 294.7	-160 -24.9 -38.9	105.7 — 11.4	-11.7 20 25	593.8 820 13,560	2.28 2.00 0.139
Methane Methanol	-161.5 64.5	510.4 1100	-182.2 -97.7	58.4 99.2	-161.5 -100 25	423 301 787	3.49 5.79 2.55
Nitrogen Octane	-195.8 124.8	198.6 306.3	-210 -57.5	25.3 180.7	-195.8 -160 20	809 596 703	2.06 2.97 2.10
Oil (light) Oxygen Petroleum	-183	212.7 230–384	-218.8	13.7	25 -183 20	910 1141 640	1.80 1.71 2.0
Propane	 -42.1	427.8	-187.7	80.0	-42.1 0 50	581 529 449	2.25 2.53 3.13
Refrigerant-134a	-26.1	217.0	-96.6	_	-50 -26.1 0 25	1443 1374 1295 1207	1.23 1.27 1.34 1.43
Water	100	2257	0.0	333.7	0 25 50 75 100	1000 997 988 975 958	4.22 4.18 4.18 4.19 4.22

^{*} Sublimation temperature. (At pressures below the triple—point pressure of 518 kPa, carbon dioxide exists as a solid or gas. Also, the freezing—point temperature of carbon dioxide is the triple—point temperature of -56.5° C.)

513

1.38

TABLE A-3

Properties of common liquids, solids, and foods (Concluded)

7,830

19,400

(b) Solids (values are for	room temperat	ure unless indicate	d otherwise)		
Substance	Density, $ ho$ kg/m 3	Specific heat, $c_p \; \mathrm{kJ/kg \cdot K}$	Substance	Density, $ ho$ kg/m 3	Specific heat, $c_p \text{ kJ/kg} \cdot \text{K}$
Metals			Nonmetals		
Aluminum			Asphalt	2110	0.920
200 K		0.797	Brick, common	1922	0.79
250 K		0.859	Brick, fireclay (500°C)	2300	0.960
300 K	2,700	0.902	Concrete	2300	0.653
350 K		0.929	Clay	1000	0.920
400 K		0.949	Diamond	2420	0.616
450 K		0.973	Glass, window	2700	0.800
500 K		0.997	Glass, pyrex	2230	0.840
Bronze (76% Cu, 2% Zn,	8,280	0.400	Graphite	2500	0.711
2% AI)			Granite	2700	1.017
Brass, yellow (65% Cu,	8,310	0.400	Gypsum or plaster board	800	1.09
35% Zn)			Ice		
Copper			200 K		1.56
−173°C		0.254	220 K		1.71
-100°C		0.342	240 K		1.86
−50°C		0.367	260 K		2.01
0°C		0.381	273 K	921	2.11
27°C	8,900	0.386	Limestone	1650	0.909
100°C		0.393	Marble	2600	0.880
200°C		0.403	Plywood (Douglas Fir)	545	1.21
Iron	7,840	0.45	Rubber (soft)	1100	1.840
Lead	11,310	0.128	Rubber (hard)	1150	2.009
Magnesium	1,730	1.000	Sand	1520	0.800
Nickel	8,890	0.440	Stone	1500	0.800
Silver	10,470	0.235	Woods, hard (maple, oak, etc.)	721	1.26

Tungsten (c) Foods

Steel, mild

	Specific heat,								Specifi	ic heat,	
	Water —		kJ/kg	K	Latent heat of		Water		kJ/kg	Latent heat of	
	content,	Freezing	Above	Below	fusion,		content,	Freezing	Above	Below	fusion,
Food	% (mass)	point, °C	freezing	freezing	kJ/kg	Food	% (mass)	point, °C	freezing	freezing	kJ/kg
Apples	84	-1.1	3.65	1.90	281	Lettuce	95	-0.2	4.02	2.04	317
Bananas	75	-0.8	3.35	1.78	251	Milk, whole	88	-0.6	3.79	1.95	294
Beef round	67	_	3.08	1.68	224	Oranges	87	-0.8	3.75	1.94	291
Broccoli	90	-0.6	3.86	1.97	301	Potatoes	78	-0.6	3.45	1.82	261
Butter	16	_	_	1.04	53	Salmon fish	64	-2.2	2.98	1.65	214
Cheese, swiss	39	-10.0	2.15	1.33	130	Shrimp	83	-2.2	3.62	1.89	277
Cherries	80	-1.8	3.52	1.85	267	Spinach	93	-0.3	3.96	2.01	311
Chicken	74	-2.8	3.32	1.77	247	Strawberries	90	-0.8	3.86	1.97	301
Corn, sweet	74	-0.6	3.32	1.77	247	Tomatoes, ripe	94	-0.5	3.99	2.02	314
Eggs, whole	74	-0.6	3.32	1.77	247	Turkey	64	_	2.98	1.65	214
Ice cream	63	-5.6	2.95	1.63	210	Watermelon	93	-0.4	3.96	2.01	311

Woods, soft (fir, pine, etc.)

0.500

0.130

Source of Data: Values are obtained from various handbooks and other sources or are calculated. Water content and freezing—point data of foods are from ASHRAE, Handbook of Fundamentals, SI version (Atlanta, GA: American Society of Heating, Refrigerating and Air—Conditioning Engineers, Inc., 1993), Chapter 30, Table 1. Freezing point is the temperature at which freezing starts for fruits and vegetables, and the average freezing temperature for other foods.

TABLE A-4

Saturated water—Temperature table

		Specific volume, m³/kg		Int	<i>ernal ene</i> kJ/kg	rgy,	Enthalpy, kJ/kg			Entropy, kJ/kg·K		
Temp.,	Sat. , press., P _{sat} kPa	Sat. liquid, v _f	Sat. vapor, v_g	Sat. liquid, u_f	Evap., u_{fg}	Sat. vapor, u_g	Sat. liquid, h_f	Evap., h_{fg}	Sat. vapor, h_g	Sat. liquid, s_f	Evap., s_{fg}	Sat. vapor, s_g
0.01 5 10 15 20	0.6117 0.8725 1.2281 1.7057 2.3392	0.001000 0.001000 0.001000 0.001001 0.001002	206.00 147.03 106.32 77.885 57.762	0.000 21.019 42.020 62.980 83.913	2374.9 2360.8 2346.6 2332.5 2318.4	2374.9 2381.8 2388.7 2395.5 2402.3	0.001 21.020 42.022 62.982 83.915	2500.9 2489.1 2477.2 2465.4 2453.5	2500.9 2510.1 2519.2 2528.3 2537.4	0.0000 0.0763 0.1511 0.2245 0.2965	9.1556 8.9487 8.7488 8.5559	9.1556 9.0249 8.8999 8.7803 8.6661
25 30 35 40 45	3.1698 4.2469 5.6291 7.3851 9.5953	0.001003 0.001004 0.001006 0.001008 0.001010	43.340 32.879 25.205 19.515 15.251	104.83 125.73 146.63 167.53 188.43	2304.3 2290.2 2276.0 2261.9 2247.7	2409.1 2415.9 2422.7 2429.4 2436.1	104.83 125.74 146.64 167.53 188.44	2441.7 2429.8 2417.9 2406.0 2394.0	2546.5 2555.6 2564.6 2573.5 2582.4	0.3672 0.4368 0.5051 0.5724 0.6386	8.0152 7.8466 7.6832	8.5567 8.4520 8.3517 8.2556 8.1633
50 55 60 65 70	12.352 15.763 19.947 25.043 31.202	0.001012 0.001015 0.001017 0.001020 0.001023	12.026 9.5639 7.6670 6.1935 5.0396	209.33 230.24 251.16 272.09 293.04	2233.4 2219.1 2204.7 2190.3 2175.8	2442.7 2449.3 2455.9 2462.4 2468.9	209.34 230.26 251.18 272.12 293.07	2382.0 2369.8 2357.7 2345.4 2333.0	2591.3 2600.1 2608.8 2617.5 2626.1	0.7038 0.7680 0.8313 0.8937 0.9551	7.2218 7.0769 6.9360	8.0748 7.9898 7.9082 7.8296 7.7540
75 80 85 90 95	38.597 47.416 57.868 70.183 84.609	0.001026 0.001029 0.001032 0.001036 0.001040	4.1291 3.4053 2.8261 2.3593 1.9808	313.99 334.97 355.96 376.97 398.00	2161.3 2146.6 2131.9 2117.0 2102.0	2475.3 2481.6 2487.8 2494.0 2500.1	314.03 335.02 356.02 377.04 398.09	2320.6 2308.0 2295.3 2282.5 2269.6	2634.6 2643.0 2651.4 2659.6 2667.6	1.0158 1.0756 1.1346 1.1929 1.2504	6.5355 6.4089 6.2853	7.6812 7.6111 7.5435 7.4782 7.4151
100 105 110 115 120	101.42 120.90 143.38 169.18 198.67	0.001043 0.001047 0.001052 0.001056 0.001060	1.6720 1.4186 1.2094 1.0360 0.89133	419.06 440.15 461.27 482.42 503.60	2087.0 2071.8 2056.4 2040.9 2025.3	2506.0 2511.9 2517.7 2523.3 2528.9	419.17 440.28 461.42 482.59 503.81	2256.4 2243.1 2229.7 2216.0 2202.1	2675.6 2683.4 2691.1 2698.6 2706.0	1.3072 1.3634 1.4188 1.4737 1.5279	5.9319 5.8193 5.7092	7.3542 7.2952 7.2382 7.1829 7.1292
125 130 135 140 145	232.23 270.28 313.22 361.53 415.68	0.001065 0.001070 0.001075 0.001080 0.001085	0.77012 0.66808 0.58179 0.50850 0.44600	524.83 546.10 567.41 588.77 610.19	2009.5 1993.4 1977.3 1960.9 1944.2	2534.3 2539.5 2544.7 2549.6 2554.4	525.07 546.38 567.75 589.16 610.64	2188.1 2173.7 2159.1 2144.3 2129.2	2713.1 2720.1 2726.9 2733.5 2739.8	1.5816 1.6346 1.6872 1.7392 1.7908	5.3919 5.2901 5.1901	7.0771 7.0265 6.9773 6.9294 6.8827
150 155 160 165 170	476.16 543.49 618.23 700.93 792.18	0.001091 0.001096 0.001102 0.001108 0.001114	0.39248 0.34648 0.30680 0.27244 0.24260	631.66 653.19 674.79 696.46 718.20	1927.4 1910.3 1893.0 1875.4 1857.5	2559.1 2563.5 2567.8 2571.9 2575.7	632.18 653.79 675.47 697.24 719.08	2113.8 2098.0 2082.0 2065.6 2048.8	2745.9 2751.8 2757.5 2762.8 2767.9	1.8418 1.8924 1.9426 1.9923 2.0417	4.9002 4.8066 4.7143	6.8371 6.7927 6.7492 6.7067 6.6650
175 180 185 190 195 200	892.60 1002.8 1123.5 1255.2 1398.8 1554.9	0.001121 0.001127 0.001134 0.001141 0.001149 0.001157	0.21659 0.19384 0.17390 0.15636 0.14089 0.12721	740.02 761.92 783.91 806.00 828.18 850.46	1839.4 1820.9 1802.1 1783.0 1763.6 1743.7	2579.4 2582.8 2586.0 2589.0 2591.7 2594.2	741.02 763.05 785.19 807.43 829.78 852.26	2031.7 2014.2 1996.2 1977.9 1959.0 1939.8	2772.7 2777.2 2781.4 2785.3 2788.8 2792.0	2.0906 2.1392 2.1875 2.2355 2.2831 2.3305	4.4448 4.3572 4.2705 4.1847	6.6242 6.5841 6.5447 6.5059 6.4678 6.4302

TABLE A-4
Saturated water—Temperature table (*Concluded*)

			c <i>volume,</i> ³ /kg	Into	<i>ernal ene</i> kJ/kg	rgy,		<i>Enthalpy,</i> kJ/kg			Entropy kJ/kg·K	
Temp.,	Sat. press., P _{sat} kPa	Sat. liquid, v _f	Sat. vapor, v_g	Sat. liquid, u _f	Evap.,	Sat. vapor, u_g	Sat. liquid, h_f	Evap., h _{fg}	Sat. vapor, h_g	Sat. liquid, s_f	Evap., s_{fg}	Sat. vapor, s_g
205 210 215 220 225	1724.3 1907.7 2105.9 2319.6 2549.7	0.001164 0.001173 0.001181 0.001190 0.001199	0.11508 0.10429 0.094680 0.086094 0.078405	872.86 895.38 918.02 940.79 963.70	1723.5 1702.9 1681.9 1660.5 1638.6	2596.4 2598.3 2599.9 2601.3 2602.3	897.61 920.50 943.55	1920.0 1899.7 1878.8 1857.4 1835.4	2794.8 2797.3 2799.3 2801.0 2802.2	2.3776 2.4245 2.4712 2.5176 2.5639	3.9318 3.8489 3.7664	6.3930 6.3563 6.3200 6.2840 6.2483
230 235 240 245 250	2797.1 3062.6 3347.0 3651.2 3976.2	0.001209 0.001219 0.001229 0.001240 0.001252	0.071505 0.065300 0.059707 0.054656 0.050085	986.76 1010.0 1033.4 1056.9 1080.7	1616.1 1593.2 1569.8 1545.7 1521.1	2602.9 2603.2 2603.1 2602.7 2601.8	990.14 1013.7 1037.5 1061.5 1085.7	1812.8 1789.5 1765.5 1740.8 1715.3	2802.9 2803.2 2803.0 2802.2 2801.0	2.6100 2.6560 2.7018 2.7476 2.7933	3.5216 3.4405 3.3596	6.2128 6.1775 6.1424 6.1072 6.0721
255 260 265 270 275	4322.9 4692.3 5085.3 5503.0 5946.4	0.001263 0.001276 0.001289 0.001303 0.001317	0.045941 0.042175 0.038748 0.035622 0.032767	1104.7 1128.8 1153.3 1177.9 1202.9	1495.8 1469.9 1443.2 1415.7 1387.4	2600.5 2598.7 2596.5 2593.7 2590.3	1110.1 1134.8 1159.8 1185.1 1210.7	1689.0 1661.8 1633.7 1604.6 1574.5	2799.1 2796.6 2793.5 2789.7 2785.2	2.8390 2.8847 2.9304 2.9762 3.0221	3.1169 3.0358 2.9542	6.0369 6.0017 5.9662 5.9305 5.8944
280 285 290 295 300	6416.6 6914.6 7441.8 7999.0 8587.9	0.001333 0.001349 0.001366 0.001384 0.001404	0.030153 0.027756 0.025554 0.023528 0.021659	1228.2 1253.7 1279.7 1306.0 1332.7	1358.2 1328.1 1296.9 1264.5 1230.9	2586.4 2581.8 2576.5 2570.5 2563.6	1236.7 1263.1 1289.8 1317.1 1344.8	1543.2 1510.7 1476.9 1441.6 1404.8	2779.9 2773.7 2766.7 2758.7 2749.6	3.0681 3.1144 3.1608 3.2076 3.2548	2.7066 2.6225 2.5374	5.8579 5.8210 5.7834 5.7450 5.7059
305 310 315 320 325	9209.4 9865.0 10,556 11,284 12,051	0.001425 0.001447 0.001472 0.001499 0.001528	0.019932 0.018333 0.016849 0.015470 0.014183	1360.0 1387.7 1416.1 1445.1 1475.0	1195.9 1159.3 1121.1 1080.9 1038.5	2555.8 2547.1 2537.2 2526.0 2513.4	1373.1 1402.0 1431.6 1462.0 1493.4	1366.3 1325.9 1283.4 1238.5 1191.0	2739.4 2727.9 2715.0 2700.6 2684.3	3.3024 3.3506 3.3994 3.4491 3.4998	2.2737 2.1821 2.0881	5.6657 5.6243 5.5816 5.5372 5.4908
330 335 340 345 350	12,858 13,707 14,601 15,541 16,529	0.001560 0.001597 0.001638 0.001685 0.001741	0.012979 0.011848 0.010783 0.009772 0.008806	1505.7 1537.5 1570.7 1605.5 1642.4	993.5 945.5 893.8 837.7 775.9	2499.2 2483.0 2464.5 2443.2 2418.3	1525.8 1559.4 1594.6 1631.7 1671.2	1140.3 1086.0 1027.4 963.4 892.7	2666.0 2645.4 2622.0 2595.1 2563.9	3.5516 3.6050 3.6602 3.7179 3.7788	1.7857 1.6756 1.5585	5.4422 5.3907 5.3358 5.2765 5.2114
355 360 365 370 373.95	17,570 18,666 19,822 21,044 22,064	0.001808 0.001895 0.002015 0.002217 0.003106	0.007872 0.006950 0.006009 0.004953 0.003106	1682.2 1726.2 1777.2 1844.5 2015.7	706.4 625.7 526.4 385.6 0	2388.6 2351.9 2303.6 2230.1 2015.7	1714.0 1761.5 1817.2 1891.2 2084.3	812.9 720.1 605.5 443.1 0	2526.9 2481.6 2422.7 2334.3 2084.3	3.8442 3.9165 4.0004 4.1119 4.4070	1.1373 0.9489	5.1384 5.0537 4.9493 4.8009 4.4070

Source of Data: Tables A–4 through A–8 are generated using the Engineering Equation Solver (EES) software developed by S. A. Klein and F. L. Alvarado. The routine used in calculations is the highly accurate Steam_IAPWS, which incorporates the 1995 Formulation for the Thermodynamic Properties of Ordinary Water Substance for General and Scientific Use, issued by The International Association for the Properties of Water and Steam (IAPWS). This formulation replaces the 1984 formulation of Haar, Gallagher, and Kell (NBS/NRC Steam Tables, Hemisphere Publishing Co., 1984), which is also available in EES as the routine STEAM. The new formulation is based on the correlations of Saul and Wagner (J. Phys. Chem. Ref. Data, 16, 893, 1987) with modifications to adjust to the International Temperature Scale of 1990. The modifications are described by Wagner and Pruss (J. Phys. Chem. Ref. Data, 22, 783, 1993). The properties of ice are based on Hyland and Wexler, "Formulations for the Thermodynamic Properties of the Saturated Phases of H₂O from 173.15 K to 473.15 K," ASHRAE Trans., Part 2A, Paper 2793, 1983.

TABLE A-5

Saturated water—Pressure table

			c <i>volume,</i> ³ /kg	Int	t <i>ernal ene</i> kJ/kg	rgy,		<i>Enthalpy,</i> kJ/kg		<i>Entropy,</i> kJ/kg∙K		
Press., P kPa	Sat. temp., $T_{\rm sat}$ °C	Sat. liquid, v _f	Sat. vapor, v_g	Sat. liquid, u_f	Evap., u_{fg}	Sat. vapor, u_g	Sat. liquid, h _f	Evap., h_{fg}	Sat. vapor, h_g	Sat. liquid, s_f	Evap., s _{fg}	Sat. vapor, s_g
1.0 1.5 2.0 2.5 3.0	6.97 13.02 17.50 21.08 24.08	0.001000 0.001001 0.001001 0.001002 0.001003		29.302 54.686 73.431 88.422 100.98	2355.2 2338.1 2325.5 2315.4 2306.9	2384.5 2392.8 2398.9 2403.8 2407.9	29.303 54.688 73.433 88.424 100.98	2484.4 2470.1 2459.5 2451.0 2443.9	2513.7 2524.7 2532.9	0.1059 0.1956	8.8690 8.6314 8.4621	8.9749 8.8270 8.7227
4.0 5.0 7.5 10 15	28.96 32.87 40.29 45.81 53.97	0.001004 0.001005 0.001008 0.001010 0.001014	34.791 28.185 19.233 14.670 10.020	121.39 137.75 168.74 191.79 225.93	2293.1 2282.1 2261.1 2245.4 2222.1	2414.5 2419.8 2429.8 2437.2 2448.0	121.39 137.75 168.75 191.81 225.94	2432.3 2423.0 2405.3 2392.1 2372.3	2553.7 2560.7 2574.0 2583.9 2598.3	0.5763	8.0510 7.9176 7.6738 7.4996 7.2522	8.1488
20 25 30 40 50	60.06 64.96 69.09 75.86 81.32	0.001017 0.001020 0.001022 0.001026 0.001030	7.6481 6.2034 5.2287 3.9933 3.2403	251.40 271.93 289.24 317.58 340.49	2204.6 2190.4 2178.5 2158.8 2142.7	2456.0 2462.4 2467.7 2476.3 2483.2	251.42 271.96 289.27 317.62 340.54	2357.5 2345.5 2335.3 2318.4 2304.7	2636.1	0.9441 1.0261	7.0752 6.9370 6.8234 6.6430 6.5019	7.9073 7.8302 7.7675 7.6691 7.5931
75 100 101.325 125 150	91.76 99.61 5 99.97 105.97 111.35	0.001037 0.001043 0.001043 0.001048 0.001053	2.2172 1.6941 1.6734 1.3750 1.1594	384.36 417.40 418.95 444.23 466.97	2111.8 2088.2 2087.0 2068.8 2052.3	2496.1 2505.6 2506.0 2513.0 2519.2	384.44 417.51 419.06 444.36 467.13	2278.0 2257.5 2256.5 2240.6 2226.0	2675.0 2675.6 2684.9	1.2132 1.3028 1.3069 1.3741 1.4337	6.2426 6.0562 6.0476 5.9100 5.7894	7.4558 7.3589 7.3545 7.2841 7.2231
175 200 225 250 275	116.04 120.21 123.97 127.41 130.58	0.001057 0.001061 0.001064 0.001067 0.001070	1.0037 0.88578 0.79329 0.71873 0.65732	486.82 504.50 520.47 535.08 548.57	2037.7 2024.6 2012.7 2001.8 1991.6	2524.5 2529.1 2533.2 2536.8 2540.1	487.01 504.71 520.71 535.35 548.86	2213.1 2201.6 2191.0 2181.2 2172.0	2706.3 2711.7 2716.5	1.4850 1.5302 1.5706 1.6072 1.6408	5.6865 5.5968 5.5171 5.4453 5.3800	7.1716 7.1270 7.0877 7.0525 7.0207
300 325 350 375 400	133.52 136.27 138.86 141.30 143.61	0.001073 0.001076 0.001079 0.001081 0.001084	0.60582 0.56199 0.52422 0.49133 0.46242	561.11 572.84 583.89 594.32 604.22	1982.1 1973.1 1964.6 1956.6 1948.9	2543.2 2545.9 2548.5 2550.9 2553.1	561.43 573.19 584.26 594.73 604.66	2163.5 2155.4 2147.7 2140.4 2133.4	2732.0	1.6717 1.7005 1.7274 1.7526 1.7765	5.3200 5.2645 5.2128 5.1645 5.1191	6.9917 6.9650 6.9402 6.9171 6.8955
450 500 550 600 650	147.90 151.83 155.46 158.83 161.98	0.001088 0.001093 0.001097 0.001101 0.001104	0.41392 0.37483 0.34261 0.31560 0.29260	639.54	1934.5 1921.2 1908.8 1897.1 1886.1	2557.1 2560.7 2563.9 2566.8 2569.4	623.14 640.09 655.77 670.38 684.08	2120.3 2108.0 2096.6 2085.8 2075.5	2752.4 2756.2	1.8205 1.8604 1.8970 1.9308 1.9623	5.0356 4.9603 4.8916 4.8285 4.7699	6.8561 6.8207 6.7886 6.7593 6.7322
700 750	164.95 167.75	0.001108 0.001111		696.23 708.40	1875.6 1865.6	2571.8 2574.0	697.00 709.24	2065.8 2056.4		1.9918 2.0195	4.7153 4.6642	6.7071 6.6837

TABLE A-5Saturated water—Pressure table (*Concluded*)

		Specific m ³ /l		Inte	ernal ener kJ/kg	gy,		<i>Enthalpy,</i> kJ/kg			Entropy, kJ/kg·K	
Press., P kPa	Sat. temp., T_{sat} °C	Sat. liquid, v _f	Sat. vapor, v _g	Sat. liquid, u_f	Evap., u_{fg}	Sat. vapor, u_g	Sat. liquid, h _f	Evap., h_{fg}	Sat. vapor, h_g	Sat. liquid, s_f	Evap., s_{fg}	Sat. vapor, s_g
800 850 900 950 1000	170.41 172.94 175.35 177.66 179.88	0.001115 0.001118 0.001121 0.001124 0.001127	0.24035 0.22690 0.21489 0.20411 0.19436	731.00 741.55 751.67	1856.1 1846.9 1838.1 1829.6 1821.4	2576.0 2577.9 2579.6 2581.3 2582.8	720.87 731.95 742.56 752.74 762.51	2047.5 2038.8 2030.5 2022.4 2014.6	2768.3 2770.8 2773.0	2.0457 2.0705 2.0941 2.1166	4.6160 4.5705 4.5273	6.6616 6.6409 6.6213 6.6027 6.5850
1100 1200 1300 1400 1500	184.06 187.96 191.60 195.04 198.29	0.001133 0.001138 0.001144 0.001149 0.001154	0.17745 0.16326 0.15119 0.14078 0.13171	796.96 813.10 828.35	1805.7 1790.9 1776.8 1763.4 1750.6	2585.5 2587.8 2589.9 2591.8 2593.4	781.03 798.33 814.59 829.96 844.55	1999.6 1985.4 1971.9 1958.9 1946.4	2783.8 2786.5 2788.9	2.2508	4.3058	6.5520 6.5217 6.4936 6.4675 6.4430
1750 2000 2250 2500 3000	205.72 212.38 218.41 223.95 233.85	0.001166 0.001177 0.001187 0.001197 0.001217	0.11344 0.099587 0.088717 0.079952 0.066667	906.12 933.54	1720.6 1693.0 1667.3 1643.2 1598.5	2596.7 2599.1 2600.9 2602.1 2603.2	878.16 908.47 936.21 961.87 1008.3	1917.1 1889.8 1864.3 1840.1 1794.9	2798.3 2800.5 2801.9	2.3844 2.4467 2.5029 2.5542 2.6454		6.3877 6.3390 6.2954 6.2558 6.1856
3500 4000 5000 6000 7000	242.56 250.35 263.94 275.59 285.83	0.001235 0.001252 0.001286 0.001319 0.001352		1045.4 1082.4 1148.1 1205.8 1258.0	1557.6 1519.3 1448.9 1384.1 1323.0	2603.0 2601.7 2597.0 2589.9 2581.0	1087.4 1154.5 1213.8	1753.0 1713.5 1639.7 1570.9 1505.2	2800.8 2794.2 2784.6	2.7253 2.7966 2.9207 3.0275 3.1220	3.3991 3.2731 3.0530 2.8627 2.6927	6.1244 6.0696 5.9737 5.8902 5.8148
8000 9000 10,000 11,000 12,000	295.01 303.35 311.00 318.08 324.68	0.001384 0.001418 0.001452 0.001488 0.001526			1264.5 1207.6 1151.8 1096.6 1041.3	2570.5 2558.5 2545.2 2530.4 2514.3	1363.7 1407.8 1450.2	1441.6 1379.3 1317.6 1256.1 1194.1		3.2077 3.2866 3.3603 3.4299 3.4964	2.5373 2.3925 2.2556 2.1245 1.9975	5.7450 5.6791 5.6159 5.5544 5.4939
13,000 14,000 15,000 16,000 17,000	330.85 336.67 342.16 347.36 352.29	0.001566 0.001610 0.001657 0.001710 0.001770	0.012781 0.011487 0.010341 0.009312 0.008374	1548.4 1585.5 1622.6	985.5 928.7 870.3 809.4 745.1	2496.6 2477.1 2455.7 2432.0 2405.4	1571.0 1610.3 1649.9	1131.3 1067.0 1000.5 931.1 857.4	2637.9 2610.8 2581.0	3.6232 3.6848	1.8730 1.7497 1.6261 1.5005 1.3709	5.4336 5.3728 5.3108 5.2466 5.1791
18,000 19,000 20,000 21,000 22,000 22,064	356.99 361.47 365.75 369.83 373.71 373.95	0.001840 0.001926 0.002038 0.002207 0.002703 0.003106	0.007504 0.006677 0.005862 0.004994 0.003644 0.003106	1740.3 1785.8 1841.6 1951.7	675.9 598.9 509.0 391.9 140.8	2375.0 2339.2 2294.8 2233.5 2092.4 2015.7	1776.8 1826.6 1888.0 2011.1	777.8 689.2 585.5 450.4 161.5	2412.1 2338.4 2172.6	3.8720 3.9396 4.0146 4.1071 4.2942 4.4070	0.2496	5.1064 5.0256 4.9310 4.8076 4.5439 4.4070

TABLE A-6

Superheated water

P = 0.01 MPa (4.5.8) V MPa M MPa M MPa (1.5.8) V P = 0.05 MPa (81.32*C) P = 0.10 MPa (99.6)*C NPa (91.32*C) NPa (91.3		neated wate	er										
P = 0.01 MPa (45.81°C)	T	-											
Sat 14,670	°C	m ³ /kg	kJ/kg	kJ/kg	kJ/kg∙K	m ³ /kg	kJ/kg	kJ/kg	kJ/kg⋅K	m ³ /kg	kJ/kg	kJ/kg	kJ/kg·K
100 17.196 2516.5 2687.5 8.4489 3.4187 2511.5 2682.4 7.6993 1.6999 2506.2 2675.8 7.3611 150 19.513 2587.9 2783.0 8.6893 3.8897 2585.7 2780.2 7.9413 1.9367 2582.9 2776.6 7.6148 2765.2 2765.2 2736.1 2879.6 8.9049 4.356.2 2660.0 2877.8 8.1592 2.1724 2686.8 2.875.7 7.8356 2.500 24.136 2736.1 2977.5 9.1015 4.8206 2735.1 2976.2 8.3568 2.4062 2733.9 2974.5 8.0345 2.400 31.063 2969.3 3200.0 9.6094 6.2094 2968.9 3279.3 8.8659 3.1027 2968.3 3278.6 8.5452 2.500 3.680 312.9 3489.7 9.8998 7.1338 312.6 3489.3 9.1566 3.5655 3132.2 3488.7 8.3862 2.406.0 40.296 3303.3 3706.3 10.1631 8.0577 3303.1 3706.0 9.4201 4.0279 3302.8 3705.5 9.0999 7.00 44.911 3480.8 32929 10.4056 8.9813 3480.6 3929.7 9.6626 4.4900 3480.4 3292.4 9.3424 9.3424 9.3424 9.3424 9.00 5.5755 4055.3 462.8 1.0029 11.503 3.055.2 462.7 10.3000 5.755 4055.3 462.8 1.0029 11.503 3.055.2 462.7 10.3000 5.4755 4055.0 462.6 9.9800 11.00 63.373 4260.0 4893.8 11.2326 12.6745 4259.9 4893.7 10.4897 6.3372 4259.8 4893.9 10.1698 63.972 470.8 51.507 10.6704 687.2 5113.3 10.5229 4888.7 250.0 4887.8 250.0		P =	0.01 MP	a (45.81°	C)*	P =	0.05 MP	a (81.32°	C)	<i>P</i> :	= 0.10 M	Pa (99.61	L°C)
100	Sat.†	14.670	2437.2	2583.9	8.1488	3.2403	2483.2	2645.2	7.5931	1.6941	2505.6	2675.0	7.3589
150 19.513 2587.9 2783.0 8.6893 3.8897 2585.7 2780.2 7.9413 1.9267 2582.9 2776.6 7.6148 2000 21.826 2661.4 2879.6 8.9049 4.5265 2660.0 2877.8 8.1526 2.610.2 2877.8 8.1526 2.610.2 2877.8 8.1526 2.610.2 2877.8 8.1526 2.610.2 2877.8 8.1526 2.610.2 2877.8 8.1526 2.610.2 2873.8 2974.5 8.0346 2873.3 10.63 2963.3 3280.0 9.6094 6.2094 2968.9 3279.3 8.8659 3.1027 2968.3 3278.6 8.5452 303.663 3132.9 3498.7 9.8998 7.1338 3132.6 3489.3 9.1566 3.5565 3132.2 3488.7 8.8362 9.00 40.906 3303.3 3706.3 10.1631 8.0577 3303.1 3706.0 9.4201 4.0279 3302.8 3705.6 9.0999 700 44.911 3480.8 392.9 10.4056 8.9813 3480.6 392.9 7 9.6626 4.4900 3480.4 392.4 9.5420 900 54.143 3856.9 4398.3 10.8429 10.8280 3856.8 4398.2 10.1000 58.758 4055.3 4642.8 11.0429 1.6731 40527 4642.7 10.300 58.758 4055.3 4642.8 11.0429 1.6731 4052.0 4642.7 10.300 5.8755 4055.0 4624.6 9.9800 1100 63.373 4260.0 4893.8 11.2326 12.6745 4259.9 4893.7 10.4897 6.3372 4259.8 4893.6 10.1698 1200 67.989 4470.9 5150.8 11.4132 15.977 4470.8 1510.7 10.670.0 5.8755 4055.0 4624.6 9.9800 1200 67.989 4470.9 5150.8 11.4132 15.977 4470.8 1510.7 10.670.4 6.7898 4470.7 5150.6 10.3504 72.604 4687.4 5413.4 11.5857 14.5209 4687.3 5413.3 10.8429 10.4024 2551.2 2761.2 7.0792 5.0000 1.8049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2377.0 8.2236 1.0155 296.0 1.8049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2360.9 7.1723 5.000 1.81623 2808.8 3072.1 7.8941 0.87555 2807.0 3069.6 7.7037 0.65489 2805.1 3067.1 7.5677 0.000 1.8049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2660.9 7.1723 0.000 1.81623 2808.8 3072.1 7.8941 0.87555 2807.0 3069.6 7.7037 0.65489 2805.1 3067.1 7.5677 0.000 1.8049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2660.9 7.1723 0.000 1.8049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2660.9 7.7123 0.000 1.8049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2660.9 7.7123 0.000 1.8049 2654.6 2870.7 7.75081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2660.9 7.7													
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Texasis	1100	63.373				12.6745		4893.7	10.4897	6.3372	4259.8	4893.6	10.1698
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150		P =	0.20 MF	Pa (120.2)	1°C)	P =	0.30 MP	a (133.52	2°C)	P =	0.40 MF	Pa (143.6	1°C)
200 1.08049 2654.6 2870.7 7.5081 0.71643 2651.0 2865.9 7.3132 0.53434 2647.2 2860.9 7.1723	Sat.	0.88578	2529.1	2706.3	7.1270	0.60582	2543.2	2724.9	6.9917	0.46242	2553.1	2738.1	6.8955
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800 0.98966 3663.6 4158.4 8.8240 0.82457 3663.2 4157.9 8.7395 0.61820 3662.5 4157.0 8.6061 900 1.08227 3855.4 4396.6 9.0362 0.90179 3855.1 4396.2 8.9518 0.67619 3854.5 4395.5 8.8185 1000 1.17480 4054.0 4641.4 9.2364 0.97893 4053.8 4641.1 9.1521 0.73411 4053.3 4640.5 9.0189 1100 1.26728 4259.0 4892.6 9.4263 1.05603 4258.8 4892.4 9.3420 0.79197 4258.3 4891.9 9.2090 1200 1.35972 4470.0 5149.8 9.6071 1.13309 4469.8 5149.6 9.5229 0.84980 4469.4 5149.3 9.3898													
900 1.08227 3855.4 4396.6 9.0362 0.90179 3855.1 4396.2 8.9518 0.67619 3854.5 4395.5 8.8185 1000 1.17480 4054.0 4641.4 9.2364 0.97893 4053.8 4641.1 9.1521 0.73411 4053.3 4640.5 9.0189 1100 1.26728 4259.0 4892.6 9.4263 1.05603 4258.8 4892.4 9.3420 0.79197 4258.3 4891.9 9.2090 1200 1.35972 4470.0 5149.8 9.6071 1.13309 4469.8 5149.6 9.5229 0.84980 4469.4 5149.3 9.3898													
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1100 1.26728 4259.0 4892.6 9.4263 1.05603 4258.8 4892.4 9.3420 0.79197 4258.3 4891.9 9.2090 1200 1.35972 4470.0 5149.8 9.6071 1.13309 4469.8 5149.6 9.5229 0.84980 4469.4 5149.3 9.3898						0.97893	4053.8						
	1100					1.05603	4258.8	4892.4		0.79197	4258.3	4891.9	9.2090
1200 1 45014 4606 6 5410 6 0 7707 1 201010 4606 4 5410 5 0 605 1 4606 1 5410 0 0 5605													
1300 1.45214 4686.6 5412.6 9.7797 1.21012 4686.4 5412.5 9.6955 0.90761 4686.1 5412.2 9.5625	1300	1.45214	4686.6	5412.6	9.7797	1.21012	4686.4	5412.5	9.6955	0.90761	4686.1	5412.2	9.5625

^{*}The temperature in parentheses is the saturation temperature at the specified pressure. † Properties of saturated vapor at the specified pressure.

TABL	E A-6											
Superl	heated wat	er (<i>Concl</i>	luded)									
T	V	и	h	S	V	И	h	S	V	И	h	S
°C	m³/kg	kJ/kg	kJ/kg	kJ/kg∙K	m³/kg	kJ/kg	kJ/kg	kJ/kg∙K	m³/kg	kJ/kg	kJ/kg	kJ/kg∙K
	P =	1.00 MP	a (179.88	°C)	P =	1.20 MP	a (187.96	5°C)	P =	1.40 MPa	(195.04	°C)
Sat.	0.19437	2582.8	2777.1	6.5850	0.16326	2587.8	2783.8	6.5217	0.14078	2591.8	2788.9	6.4675
200	0.20602	2622.3	2828.3	6.6956	0.16934			6.5909	0.14303	2602.7	2803.0	6.4975
250	0.23275	2710.4	2943.1	6.9265	0.19241			6.8313	0.16356	2698.9	2927.9	6.7488
300	0.25799	2793.7	3051.6	7.1246	0.21386			7.0335	0.18233	2785.7	3040.9	6.9553
350	0.28250	2875.7	3158.2	7.3029	0.23455			7.2139	0.20029	2869.7	3150.1	
400	0.30661	2957.9	3264.5	7.4670	0.25482			7.3793	0.21782	2953.1	3258.1	7.3046
500	0.35411	3125.0	3479.1	7.7642	0.29464			7.6779	0.25216	3121.8		7.6047
600 700	0.40111 0.44783	3297.5 3476.3	3698.6 3924.1	8.0311 8.2755	0.33395 0.37297			7.9456 8.1904	0.28597 0.31951	3295.1 3474.4	3695.5 3921.7	7.8730 8.1183
800	0.44783	3661.7	4156.1	8.5024	0.37237			8.4176	0.31331	3660.3	4154.3	8.3458
900	0.54083	3853.9	4394.8	8.7150	0.45059			8.6303	0.38614	3852.7	4393.3	
1000	0.58721	4052.7	4640.0	8.9155	0.48928			8.8310	0.41933	4051.7		8.7595
1100	0.63354	4257.9	4891.4	9.1057	0.52792			9.0212	0.45247	4257.0		8.9497
1200	0.67983	4469.0	5148.9	9.2866	0.56652	4468.7	5148.5	9.2022	0.48558	4468.3	5148.1	9.1308
1300	0.72610	4685.8	5411.9	9.4593	0.60509	4685.5	5411.6	9.3750	0.51866	4685.1	5411.3	9.3036
	P =	1.60 MPa	a (201.37	°C)	P =	1.80 MF	°a (207.1)	1°C)	P =	2.00 MPa	(212.38	°C)
Sat.	0.12374	2594.8	2792.8	6.4200	0.11037	2597.3	2795.9	6.3775	0.09959	2599.1	2798.3	6.3390
225	0.13293		2857.8	6.5537	0.11678				0.10381	2628.5	2836.1	
250	0.14190	2692.9	2919.9	6.6753	0.12502	2686.7	2911.7	6.6088	0.11150	2680.3	2903.3	6.5475
300	0.15866	2781.6	3035.4	6.8864	0.14025	2777.4	3029.9	6.8246	0.12551	2773.2	3024.2	6.7684
350	0.17459	2866.6	3146.0	7.0713	0.15460			7.0120	0.13860	2860.5	3137.7	
400	0.19007		3254.9	7.2394	0.16849			7.1814	0.15122	2945.9	3248.4	
500	0.22029	3120.1	3472.6	7.5410	0.19551			7.4845	0.17568	3116.9	3468.3	
600		3293.9	3693.9	7.8101	0.22200			7.7543	0.19962	3291.5	3690.7	
700	0.27941	3473.5	3920.5	8.0558	0.24822		3919.4		0.22326	3471.7	3918.2	
800 900	0.30865 0.33780	3659.5 3852.1	4153.4 4392.6	8.2834 8.4965	0.27426 0.30020			8.2284	0.24674 0.27012	3658.0 3850.9	4151.5 4391.1	
1000	0.36687	4051.2	4638.2	8.6974	0.32606		4637.6	8.6427	0.27012	4050.2	4637.1	
1100	0.39589	4256.6	4890.0	8.8878	0.35188				0.23542	4255.7	4889.1	
1200	0.42488	4467.9	5147.7	9.0689	0.37766	4467.6	5147.3	9.0143	0.33989	4467.2	5147.0	
1300	0.45383	4684.8	5410.9	9.2418	0.40341	4684.5	5410.6	9.1872	0.36308	4684.2	5410.3	
	P =	2.50 MP	a (223.95	°C)	P =	3.00 MP	a (233.85	i°C)		3.50 MPa		
Sat.	0.07995	2602.1	2801.9	6.2558	0.06667		2803.2		0.05706	2603.0	2802.7	6.1244
225	0.08026	2604.8	2805.5	6.2629								
250	0.08705	2663.3	2880.9	6.4107	0.07063	2644.7	2856.5	6.2893	0.05876	2624.0	2829.7	6.1764
300	0.09894			6.6459			2994.3		0.06845		2978.4	
350	0.10979		3127.0		0.09056				0.07680		3104.9	
400	0.12012			7.0170	0.09938				0.08456		3223.2	
450	0.13015			7.1768	0.10789				0.09198		3338.1	
500	0.13999		3462.8 3686.8	7.3254	0.11620				0.09919		3451.7 3678.9	
600 700	0.15931 0.17835			7.3979	0.13245 0.14841		3912.2		0.11325 0.12702		3909.3	
800	0.17833		4149.2		0.14841				0.12702		4144.6	
900	0.21597		4389.3		0.17988		4387.5		0.15410		4385.7	
1000	0.23466			8.4897			4634.2		0.16751		4632.7	
1100	0.25330			8.6804	0.21105				0.18087		4885.6	
1200	0.27190			8.8618	0.22658				0.19420		5144.1	
1300	0.29048			9.0349					0.20750		5408.0	

TABLE A-6

Superl	heated wate	er (<i>Conti</i>	nued)									
T	V	И	h	S	V	И	h	S	V	И	h	S
°C	m³/kg	kJ/kg	kJ/kg	kJ/kg∙K	m³/kg	kJ/kg	kJ/kg	kJ/kg∙K	m³/kg	kJ/kg	kJ/kg	kJ/kg·K
	P =	4.0 MP	a (250.35°	°C)	P =	= 4.5 MPa	(257.44°C	C)	P =	5.0 MPa	(263.94°	C)
Sat.	0.04978	2601.7	2800.8	6.0696	0.04406	2599.7	2798.0	6.0198	0.03945	2597.0	2794.2	5.9737
275	0.05461		2887.3	6.2312	0.04733	2651.4	2864.4	6.1429	0.04144	2632.3		6.0571
300		2726.2	2961.7	6.3639	0.05138	2713.0	2944.2	6.2854	0.04535	2699.0		6.2111
350	0.06647		3093.3	6.5843	0.05842	2818.6	3081.5	6.5153	0.05197	2809.5		6.4516
400	0.07343		3214.5 3331.2	6.7714	0.06477	2914.2	3205.7	6.7071	0.05784	2907.5		6.6483
450 500		3011.0	3446.0	6.9386 7.0922	0.07076 0.07652	3005.8 3096.0	3324.2 3440.4	6.8770 7.0323	0.06332 0.06858	3000.6 3091.8		6.8210 6.9781
600	0.08844		3674.9	7.0922	0.07652	3276.4	3670.9	7.0323	0.00838	3273.3		7.2605
700	0.11098		3906.3	7.6214	0.09850	3460.0	3903.3	7.5647	0.07870	3457.7		7.5136
800	0.12292		4142.3	7.8523	0.10916	3648.8	4140.0	7.7962	0.09816	3646.9		7.7458
900		3844.8	4383.9	8.0675	0.11972	3843.3	4382.1	8.0118	0.10769	3841.8		7.9619
1000	0.14653		4631.2	8.2698	0.13020	4043.9	4629.8	8.2144	0.11715	4042.6		8.1648
1100	0.15824	4251.4	4884.4	8.4612	0.14064	4250.4	4883.2	8.4060	0.12655	4249.3		8.3566
1200	0.16992	4463.5	5143.2	8.6430	0.15103	4462.6	5142.2	8.5880	0.13592	4461.6	5141.3	8.5388
1300	0.18157	4680.9	5407.2	8.8164	0.16140	4680.1	5406.5	8.7616	0.14527	4679.3	5405.7	8.7124
	P =	6.0 MPa	a (275.59°	C)	P =	= 7.0 MPa	(285.83°C	C)	P =	8.0 MPa	(295.01	°C)
Sat.	0.03245	2589.9	2784.6	5.8902	0.027378	2581.0	2772.6	5.8148	0.023525	2570.5	2758.7	5.7450
300	0.03619	2668.4	2885.6	6.0703	0.029492	2633.5	2839.9	5.9337	0.024279	2592.3	2786.5	5.7937
350	0.04225		3043.9	6.3357	0.035262		3016.9	6.2305	0.029975			6.1321
400	0.04742		3178.3	6.5432	0.039958		3159.2	6.4502	0.034344			6.3658
450		2989.9	3302.9	6.7219	0.044187		3288.3	6.6353	0.038194			6.5579
500	0.05667		3423.1	6.8826	0.048157		3411.4	6.8000	0.041767			6.7266
550	0.06102 0.06527		3541.3	7.0308	0.051966		3531.6	6.9507	0.045172			6.8800
600 700	0.06327		3658.8 3894.3	7.1693 7.4247	0.055665 0.062850		3650.6 3888.3	7.0910 7.3487	0.048463 0.054829			7.0221 7.2822
800		3643.2	4133.1	7.6582	0.062856		4128.5	7.5836	0.054829			7.5185
900		3838.8	4376.6	7.8751	0.003030		4373.0	7.8014	0.067082			7.7372
1000	0.09756		4625.4	8.0786	0.083571		4622.5	8.0055	0.073079			7.9419
1100	0.10543	4247.1	4879.7	8.2709	0.090341	4245.0	4877.4	8.1982	0.079025		4875.0	8.1350
1200	0.11326	4459.8	5139.4	8.4534	0.097075	4457.9	5137.4	8.3810	0.084934	4456.1	5135.5	8.3181
1300	0.12107	4677.7	5404.1	8.6273	0.103781	4676.1	5402.6	8.5551	0.090817	4674.5	5401.0	8.4925
	P =	9.0 MPa	a (303.35°	C)	P =	10.0 MPa	a (311.00°	C)	P =	12.5 MPa	(327.81	°C)
Sat.	0.020489		2742.9	5.6791	0.018028		2725.5	5.6159	0.013496	2505.6	2674.3	5.4638
325	0.023284		2857.1	5.8738	0.019877		2810.3	5.7596 5.9460	0.016138	2624.0	2026.0	5.7130
350 400	0.025816 0.029960		2957.3 3118.8	6.0380 6.2876	0.022440 0.026436		2924.0 3097.5	6.2141	0.016138			6.0433
450	0.029900		3258.0	6.4872	0.020430		3242.4	6.4219	0.020030		3201.5	
500	0.036793				0.023762		3375.1	6.5995	0.025630			
550	0.039885		3512.0	6.8164	0.035655		3502.0	6.7585	0.028033			6.6317
600	0.042861		3634.1	6.9605	0.038378		3625.8	6.9045	0.030306			6.7828
650	0.045755		3755.2	7.0954	0.041018		3748.1	7.0408	0.032491			6.9227
700	0.048589	3438.8	3876.1	7.2229	0.043597		3870.0	7.1693	0.034612			7.0540
800	0.054132		4119.2	7.4606	0.048629		4114.5	7.4085	0.038724			7.2967
900	0.059562		4365.7	7.6802	0.053547		4362.0	7.6290	0.042720			7.5195
1000	0.064919			7.8855	0.058391		4613.8	7.8349	0.046641			7.7269
1100	0.070224		4872.7	8.0791	0.063183		4870.3	8.0289	0.050510			7.9220
1200	0.075492		5133.6	8.2625	0.067938		5131.7	8.2126	0.054342			8.1065
1300	0.080733	40/2.9	5399.5	8.4371	0.072667	40/1.3	5398.0	8.3874	0.058147	4007.3	5594.1	8.2819

TABL	.E A-6											
Super	heated wate	er (<i>Conclu</i>	ıded)									
T	V	U In 1/1	h	S	V	U In 1/15	h	S	V 3/1	U In 1/15	h	S
°C	m ³ /kg	kJ/kg	kJ/kg	kJ/kg⋅K	m ³ /kg	kJ/kg	kJ/kg	kJ/kg·K	m ³ /kg	kJ/kg	kJ/kg	kJ/kg⋅K
	<i>P</i> =	15.0 MPa	a (342.16°			17.5 MPa					a (365.75°	
Sat.	0.010341	2455.7	2610.8	5.3108	0.007932	2390.7	2529.5	5.1435	0.005862	2294.8	2412.1	4.9310
350 400	0.011481 0.015671	2520.9 2740.6	2693.1 2975.7	5.4438 5.8819	0.012463	268/13	2902 /	5.7211	0.009950	2617.9	2816.9	5.5526
450	0.013071	2880.8	3157.9	6.1434	0.012403			6.0212	0.003330		3061.7	5.9043
500	0.020828	2998.4	3310.8	6.3480	0.017385			6.2424	0.014793		3241.2	6.1446
550	0.022945	3106.2	3450.4	6.5230	0.019305				0.016571			6.3390
600	0.024921	3209.3	3583.1	6.6796	0.021073			6.5890	0.018185			6.5075
650	0.026804	3310.1	3712.1 3839.1	6.8233	0.022742				0.019695		36/5.3	6.6593
700 800	0.028621 0.032121	3409.8 3609.3	4091.1	6.9573 7.2037	0.024342 0.027405				0.021134 0.023870		4067.5	6.7991 7.0531
900	0.035503	3811.2	4343.7	7.4288	0.030348	3803.5			0.026484		4325.4	7.2829
1000	0.038808	4017.1	4599.2	7.6378	0.033215				0.029020		4584.7	7.4950
1100	0.042062	4227.7	4858.6	7.8339	0.036029			7.7588	0.031504			7.6933
1200	0.045279	4443.1	5122.3	8.0192	0.038806			7.9449	0.033952		5112.9	7.8802
1300	0.048469	4663.3	5390.3	8.1952	0.041556	4659.2	5386.5	8.1215	0.036371	4655.2	5382.7	8.0574
		P = 25	.0 MPa			P = 30.0) МРа			P = 35.	0 МРа	
375	0.001978	1799.9	1849.4	4.0345	0.001792				0.001701	1702.8	1762.4	3.8724
400	0.006005	2428.5	2578.7	5.1400	0.002798			4.4758	0.002105			4.2144
425	0.007886	2607.8	2805.0	5.4708	0.005299			5.1473	0.003434			4.7751
450 500	0.009176 0.011143	2721.2 2887.3	2950.6 3165.9	5.6759 5.9643	0.006737 0.008691			5.4422 5.7956	0.004957 0.006933		2671.0 2997.9	5.1946 5.6331
550	0.011145	3020.8	3339.2	6.1816	0.000031			6.0403	0.000333		3218.0	5.9093
600	0.014140	3140.0	3493.5	6.3637	0.011445			6.2373	0.009523		3399.0	6.1229
650	0.015430	3251.9	3637.7	6.5243	0.012590			6.4074	0.010565		3560.7	6.3030
700	0.016643	3359.9	3776.0	6.6702	0.013654			6.5599	0.011523		3711.6	6.4623
800 900	0.018922 0.021075	3570.7 3780.2	4043.8 4307.1	6.9322 7.1668	0.015628 0.017473	3551.2			0.013278 0.014904		3996.3	6.7409 6.9853
1000	0.021075	3991.5	4570.2	7.1008	0.017473			7.0695	0.014904		4541.5	7.2069
1100	0.025172	4206.1	4835.4	7.5825	0.020954	4195.2			0.017942			7.4118
1200	0.027157	4424.6	5103.5	7.7710	0.022630			7.6807	0.019398	4406.1	5085.0	7.6034
1300	0.029115	4647.2	5375.1	7.9494	0.024279	4639.2	5367.6	7.8602	0.020827	4631.2	5360.2	7.7841
		P = 40.	0 МРа			P = 50.0) MPa			P = 60.	0 МРа	
375	0.001641	1677.0	1742.6	3.8290	0.001560				0.001503			3.7149
400	0.001911	1855.0	1931.4	4.1145	0.001731			4.0029	0.001633			3.9317
425	0.002538	2097.5	2199.0	4.5044	0.002009				0.001816			
450 500	0.003692 0.005623	2364.2		4.9449 5.4744	0.002487 0.003890				0.002086 0.002952			
550	0.005625		2906.5 3154.4	5.7857	0.003890				0.002952			
600	0.008089		3350.4	6.0170	0.006108				0.003333			
650	0.009053	3159.5	3521.6	6.2078	0.006957	3095.6	3443.5	6.0373	0.005591	3031.3	3366.8	5.8867
700	0.009930	3282.0	3679.2	6.3740	0.007717				0.006265			
800	0.011521		3972.6	6.6613	0.009073				0.007456			
900	0.012980 0.014360		4252.5	6.9107	0.010296 0.011441				0.008519 0.009504			6.6725
1100	0.014360	4173.7	4527.3 4801.1	7.1355 7.3425	0.011441				0.009304			
1200	0.016976	4396.9	5075.9	7.5357	0.013590				0.010433			7.3248
1300	0.018239			7.7175	0.014620				0.012213			

TABLE A-7

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TABI	LE A-7											
Comp	ressed liqui	d water										
T	V	И	h	S	V	И	h	S	V	И	h	S
°C	m³/kg	kJ/kg	kJ/kg	kJ/kg⋅K	m³/kg	kJ/kg	kJ/kg	kJ/kg⋅K	m³/kg	kJ/kg	kJ/kg	kJ/kg·K
	P =	5 MPa	(263.94°C)	P =	10 MPa	(311.00°C)	P =	15 MPa	(342.16°	C)
Sat.	0.0012862	1148.1	1154.5	2.9207	0.0014522	1393.3	1407.9	3.3603	0.0016572	1585.5	1610.3	3.6848
0	0.0009977	0.04	5.03	0.0001	0.0009952	0.12	10.07	0.0003	0.0009928	0.18	15.07	0.0004
20	0.0009996	83.61	88.61	0.2954	0.0009973	83.31	93.28	0.2943	0.0009951	83.01	97.93	0.2932
40	0.0010057	166.92	171.95	0.5705	0.0010035	166.33	176.37	0.5685	0.0010013	165.75	180.77	0.5666
60	0.0010149	250.29	255.36	0.8287	0.0010127	249.43	259.55	0.8260	0.0010105	248.58	263.74	0.8234
80	0.0010267	333.82	338.96	1.0723	0.0010244	332.69	342.94	1.0691	0.0010221	331.59	346.92	1.0659
100	0.0010410	417.65	422.85	1.3034	0.0010385	416.23	426.62	1.2996	0.0010361	414.85	430.39	1.2958
120	0.0010576	501.91	507.19	1.5236	0.0010549	500.18	510.73	1.5191	0.0010522	498.50	514.28	1.5148
140	0.0010769	586.80	592.18	1.7344	0.0010738	584.72	595.45	1.7293	0.0010708	582.69	598.75	1.7243
160	0.0010988	672.55	678.04	1.9374	0.0010954	670.06	681.01	1.9316	0.0010920	667.63	684.01	1.9259
180	0.0011240	759.47	765.09	2.1338	0.0011200	756.48	767.68	2.1271	0.0011160	753.58	770.32	2.1206
200	0.0011531	847.92	853.68	2.3251	0.0011482	844.32	855.80	2.3174	0.0011435	840.84	858.00	2.3100
220	0.0011868	938.39	944.32	2.5127	0.0011809	934.01	945.82	2.5037	0.0011752	929.81	947.43	2.4951
240	0.0012268	1031.6	1037.7	2.6983	0.0012192	1026.2	1038.3	2.6876	0.0012121	1021.0	1039.2	2.6774
260	0.0012755	1128.5	1134.9	2.8841	0.0012653	1121.6	1134.3	2.8710	0.0012560	1115.1	1134.0	2.8586
280					0.0013226	1221.8	1235.0	3.0565	0.0013096	1213.4	1233.0	3.0410
300					0.0013980	1329.4	1343.3	3.2488	0.0013783	1317.6	1338.3	3.2279
320									0.0014733	1431.9	1454.0	3.4263
340									0.0016311	1567.9	1592.4	3.6555
	<i>P</i> =	20 MPa	(365.75°C	C)		P = 30	MPa			P = 50) MPa	
Sat.	0.0020378	1785.8	1826.6	4.0146								
0	0.0009904	0.23	20.03	0.0005	0.0009857	0.29	29.86	0.0003	0.0009767	0.29	49.13	-0.0010
20	0.0009929	82.71	102.57	0.2921	0.0009886	82.11	111.77	0.2897	0.0009805	80.93	129.95	0.2845
40	0.0009992	165.17	185.16	0.5646	0.0009951	164.05	193.90	0.5607	0.0009872	161.90	211.25	0.5528
60	0.0010084	247.75	267.92	0.8208	0.0010042	246.14	276.26	0.8156	0.0009962	243.08	292.88	0.8055
80	0.0010199	330.50	350.90	1.0627	0.0010155	328.40	358.86	1.0564	0.0010072	324.42	374.78	1.0442
100	0.0010337	413.50	434.17	1.2920	0.0010290	410.87	441.74	1.2847	0.0010201	405.94	456.94	1.2705
120	0.0010496	496.85	517.84	1.5105	0.0010445	493.66	525.00	1.5020	0.0010349	487.69	539.43	1.4859
140	0.0010679	580.71	602.07	1.7194	0.0010623	576.90	608.76	1.7098	0.0010517	569.77	622.36	1.6916
160	0.0010886	665.28	687.05	1.9203	0.0010823	660.74	693.21	1.9094	0.0010704	652.33	705.85	1.8889
180	0.0011122	750.78	773.02	2.1143	0.0011049	745.40	778.55	2.1020	0.0010914	735.49	790.06	2.0790
200	0.0011390	837.49	860.27	2.3027	0.0011304	831.11	865.02	2.2888	0.0011149	819.45	875.19	2.2628
220	0.0011697	925.77	949.16	2.4867	0.0011595	918.15	952.93	2.4707	0.0011412	904.39	961.45	2.4414
240	0.0012053	1016.1	1040.2	2.6676	0.0011927	1006.9	1042.7	2.6491	0.0011708	990.55	1049.1	2.6156
260	0.0012472		1134.0	2.8469	0.0012314		1134.7	2.8250	0.0012044		1138.4	2.7864
280	0.0012978		1231.5	3.0265	0.0012770		1229.8	3.0001	0.0012430		1229.9	2.9547
300	0.0013611		1334.4	3.2091	0.0013322		1328.9	3.1761	0.0012879		1324.0	3.1218
320	0.0014450		1445.5	3.3996	0.0014014		1433.7	3.3558	0.0013409		1421.4	3.2888
340	0.0015693		1571.6	3.6086	0.0014932	1502.4	1547.1	3.5438	0.0014049	1452.9	1523.1	3.4575
360	0.0018248	1703.6	1740.1	3.8787	0.0016276		1675.6	3.7499	0.0014848	1556.5	1630.7	3.6301

0.0018729 1782.0 1838.2 4.0026 0.0015884 1667.1 1746.5 3.8102

TABLE A-8

Saturated ice-water vapor

			Specific volume, m³/kg Sat. Sat.		<i>ternal er</i> kJ/kg			<i>Enthalpy</i> kJ/kg	<i>'</i> ,		<i>ntropy,</i> ⟨J/kg⋅K	
	Sat.	Sat.	Sat.	Sat.		Sat.	Sat.		Sat.	Sat.		Sat.
Temp.,	press.,	ice,	vapor,	ice,	Subl.,	vapor,	ice,	Subl.,	vapor,	ice,	Subl.,	vapor,
T °C	$P_{\rm sat}$ kPa	V_i	V_g	U_i	U_{ig}	U_g	h _i	h _{ig}	h_g	S_i	S _{ig}	S_g
0.01	0.61169	0.001091	205.99	-333.40	2707.9	2374.5	-333.40	2833.9	2500.5	-1.2202	10.374	9.154
0	0.61115	0.001091	206.17	-333.43	2707.9	2374.5	-333.43	2833.9	2500.5	-1.2204	10.375	9.154
-2	0.51772	0.001091	241.62	-337.63	2709.4	2371.8	-337.63	2834.5	2496.8	-1.2358	10.453	9.218
-4	0.43748	0.001090	283.84	-341.80	2710.8	2369.0	-341.80	2835.0	2493.2	-1.2513	10.533	9.282
-6	0.36873	0.001090	334.27	-345.94	2712.2	2366.2	-345.93	2835.4	2489.5	-1.2667	10.613	9.347
-8	0.30998	0.001090	394.66	-350.04	2713.5	2363.5	-350.04	2835.8	2485.8	-1.2821	10.695	9.413
-10	0.25990	0.001089	467.17	-354.12	2714.8	2360.7	-354.12	2836.2	2482.1	-1.2976	10.778	9.480
-12	0.21732	0.001089	554.47	-358.17	2716.1	2357.9	-358.17	2836.6	2478.4	-1.3130	10.862	9.549
-14	0.18121	0.001088	659.88	-362.18	2717.3	2355.2	-362.18	2836.9	2474.7	-1.3284	10.947	9.618
-16	0.15068	0.001088	787.51	-366.17	2718.6	2352.4	-366.17	2837.2	2471.0	-1.3439	11.033	9.689
-18	0.12492	0.001088	942.51	-370.13	2719.7	2349.6	-370.13	2837.5	2467.3	-1.3593	11.121	9.761
-20	0.10326	0.001087	1131.3	-374.06	2720.9	2346.8	-374.06	2837.7	2463.6	-1.3748	11.209	9.835
-22	0.08510	0.001087	1362.0	-377.95	2722.0	2344.1	-377.95	2837.9	2459.9	-1.3903	11.300	9.909
-24	0.06991	0.001087	1644.7	-381.82	2723.1	2341.3	-381.82	2838.1	2456.2	-1.4057	11.391	9.985
-26	0.05725	0.001087	1992.2	-385.66	2724.2	2338.5	-385.66	2838.2	2452.5	-1.4212	11.484	10.063
-28	0.04673	0.001086	2421.0	-389.47	2725.2	2335.7	-389.47	2838.3	2448.8	-1.4367	11.578	10.141
-30	0.03802	0.001086	2951.7	-393.25	2726.2	2332.9	-393.25	2838.4	2445.1	-1.4521	11.673	10.221
-32	0.03082	0.001086	3610.9	-397.00	2727.2	2330.2	-397.00	2838.4	2441.4	-1.4676	11.770	10.303
-34	0.02490	0.001085	4432.4	-400.72	2728.1	2327.4	-400.72	2838.5	2437.7	-1.4831	11.869	10.386
-36	0.02004	0.001085	5460.1	-404.40	2729.0	2324.6	-404.40	2838.4	2434.0	-1.4986	11.969	10.470
-38	0.01608	0.001085	6750.5	-408.07	2729.9	2321.8	-408.07	2838.4	2430.3	-1.5141	12.071	10.557
-40	0.01285	0.001084	8376.7	-411.70	2730.7	2319.0	-411.70	2838.3	2426.6	-1.5296	12.174	10.644

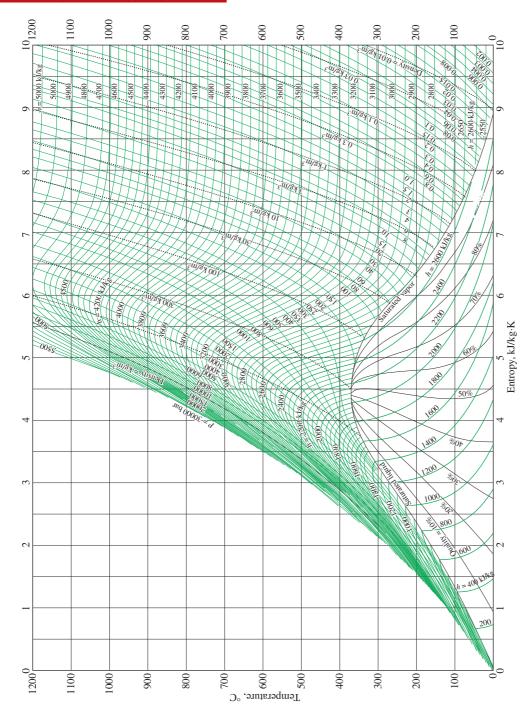


FIGURE A–9 *T-s* diagram for water.

Source of Data: From NBS/NRC Steam Tables/1 by Lester Haar, John S. Gallagher, and George S. Kell. Routledge/Taylor & Francis Books, Inc., 1984.

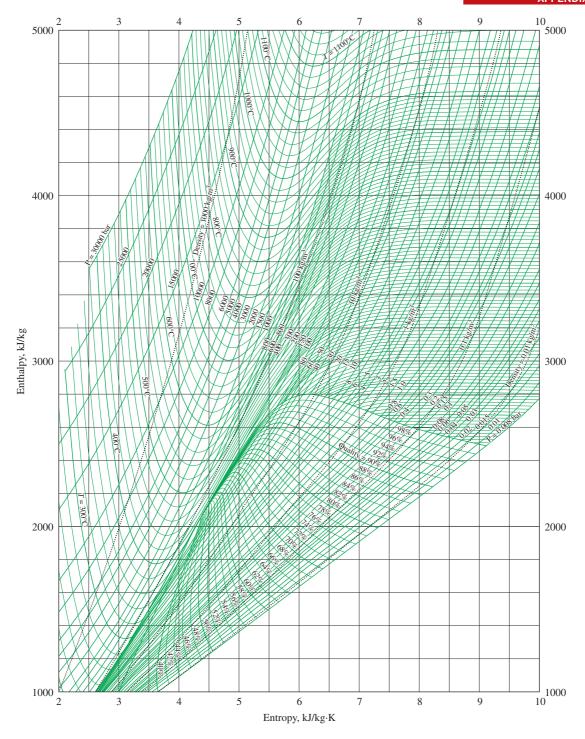


FIGURE A-10 Mollier diagram for water.

Source of Data: From NBS/NRC Steam Tables/1 by Lester Haar, John S. Gallagher, and George S. Kell. Routledge/Taylor & Francis Books, Inc., 1984.

TABLE A-11

Saturated refrigerant-134a—Temperature table

		Specific volume, m³/kg Sat. Sat. liquid, vapor,		Inte	<i>rnal ener</i> kJ/kg	gy,		Enthalpy, kJ/kg			Entropy, kJ/kg·K	
	Sat.	Sat.	Sat.	Sat.		Sat.	Sat.		Sat.	Sat.		Sat.
Temp.,	•	liquid,	vapor,	liquid,	Evap.,	vapor,	liquid,	Evap.,	vapor,	liquid,	Evap.,	vapor,
T°C	P _{sat} kPa	V_f	V_g	U_f	U_{fg}	U_g	h_f	h _{fg}	h _g	S_f	S_{fg}	S_g
-40	51.25	0.0007053	0.36064	-0.036	207.42	207.38	0.00	225.86	225.86	0.00000	0.96869	0.96869
-38	56.86	0.0007082	0.32718	2.472	206.06	208.53	2.512	224.62	227.13	0.01071	0.95516	0.96588
-36	62.95	0.0007111	0.29740	4.987	204.69	209.68	5.032	223.37	228.40	0.02137	0.94182	0.96319
-34	69.56	0.0007141	0.27082	7.509	203.32	210.83	7.559	222.10	229.66	0.03196	0.92867	0.96063
-32	76.71	0.0007171	0.24706	10.04	201.94	211.97	10.09	220.83	230.93	0.04249	0.91569	0.95819
-30	84.43	0.0007201	0.22577	12.58	200.55	213.12	12.64	219.55	232.19	0.05297	0.90289	0.95586
-28	92.76	0.0007232	0.20666	15.12	199.15	214.27	15.19	218.25	233.44	0.06339	0.89024	0.95364
-26	101.73	0.0007264	0.18947	17.67	197.75	215.42	17.75	216.95	234.70	0.07376	0.87776	0.95152
-24	111.37	0.0007296	0.17398	20.23	196.34	216.57	20.31	215.63	235.94	0.08408	0.86542	0.94950
-22	121.72	0.0007328	0.15999	22.80	194.92	217.71	22.89	214.30	237.19	0.09435	0.85323	0.94758
-20	132.82	0.0007361	0.14735	25.37	193.49	218.86	25.47	212.96	238.43	0.10456	0.84119	0.94575
-18	144.69	0.0007394	0.13589	27.96	192.05	220.00	28.07	211.60	239.67	0.11473	0.82927	0.94401
-16	157.38	0.0007428	0.12550	30.55	190.60	221.15	30.67	210.23	240.90	0.12486	0.81749	0.94234
-14	170.93	0.0007463	0.11605	33.15	189.14	222.29	33.28	208.84	242.12	0.13493	0.80583	0.94076
-12	185.37	0.0007498	0.10744	35.76	187.66	223.42	35.90	207.44	243.34	0.14497	0.79429	0.93925
-10	200.74	0.0007533	0.099600	38.38	186.18	224.56	38.53	206.02	244.55	0.15496	0.78286	0.93782
-8	217.08	0.0007570	0.092438	41.01	184.69	225.69	41.17	204.59	245.76	0.16491	0.77154	0.93645
-6	234.44	0.0007607	0.085888	43.64	183.18	226.82	43.82	203.14	246.95	0.17482	0.76033	0.93514
-4	252.85	0.0007644	0.079889	46.29	181.66	227.94	46.48	201.66	248.14	0.18469	0.74921	0.93390
-2	272.36	0.0007683	0.074388	48.94	180.12	229.07	49.15	200.17	249.33	0.19452	0.73819	0.93271
0	293.01	0.0007722	0.069335	51.61	178.58	230.18	51.83	198.67	250.50	0.20432	0.72726	0.93158
2	314.84	0.0007761	0.064690	54.28	177.01	231.30	54.53	197.14	251.66		0.71641	0.93050
4	337.90	0.0007802	0.060412	56.97	175.44	232.40	57.23	195.58	252.82	0.22381	0.70565	
6	362.23	0.0007843	0.056469	59.66	173.84	233.51	59.95	194.01	253.96	0.23351	0.69496	
8	387.88	0.0007886	0.052829	62.37	172.23	234.60	62.68	192.42		0.24318	0.68435	
10	414.89	0.0007929	0.049466	65.09	170.61	235.69	65.42	190.80		0.25282	0.67380	
12	443.31	0.0007973	0.046354	67.82	168.96	236.78	68.17	189.16	257.33	0.26243	0.66331	
14	473.19	0.0008018	0.043471	70.56	167.30	237.86	70.94	187.49		0.27201	0.65289	
16	504.58	0.0008064	0.040798	73.31	165.62	238.93	73.72	185.80	259.51		0.64252	
18	537.52	0.0008112	0.038317	76.07	163.92	239.99	76.51	184.08	260.59	0.29111		0.92330
20	572.07	0.0008160	0.036012	78.85	162.19	241.04	79.32	182.33			0.62192	
22	608.27	0.0008209	0.033867	81.64	160.45	242.09	82.14	180.55	262.69	0.31012	0.61168	
24	646.18	0.0008260	0.031869	84.44	158.68	243.13	84.98	178.74			0.60148	
26	685.84	0.0008312	0.030008	87.26	156.89	244.15	87.83	176.90	264.73	0.32905	0.59131	
28	727.31	0.0008366	0.028271	90.09	155.08	245.17	90.70	175.03			0.58117	
30	770.64	0.0008421	0.026648	92.93	153.24	246.17	93.58	173.13	266.71		0.57105	
32	815.89	0.0008477	0.025131	95.79	151.37	247.17	96.49	171.19	267.67		0.56095	
34	863.11	0.0008535	0.023712	98.67	149.48	248.15	99.41	169.21	268.61		0.55086	
36	912.35	0.0008595	0.022383	101.56	147.55		102.34	167.19	269.53		0.54077	
38	963.68	0.0008657	0.021137	104.47	145.60		105.30	165.13		0.38554	0.53068	
40	1017.1	0.0008720	0.019968	107.39	143.61		108.28	163.03	271.31		0.52059	
42	1072.8	0.0008786	0.018870	110.34	141.59		111.28	160.89	272.17	0.40432	0.51048	
44	1130.7	0.0008854	0.017837	113.30	139.53	252.83	114.30	158.70	273.00	0.41371	0.50036	0.91407

TABLE A-11

Saturated refrigerant-134a—Temperature table (Concluded)

		Specific volume, m³/kg		Internal energy, kJ/kg			Enthalpy, kJ/kg			<i>Entropy,</i> kJ/kg∙K		
T	Sat.	Sat.	Sat.	Sat.	F	Sat.	Sat.	F	Sat.	Sat.	F	Sat.
Temp.,	•	liquid,	vapor,	liquid,	Evap.,	vapor,	liquid,	Evap.,	vapor,	liquid,	Evap.,	vapor,
T°C	P _{sat} kPa	V_f	V_g	U_f	U _{fg}	Ug	h _f	h _{fg}	h _g	S_f	S _{fg}	Sg
46	1191.0	0.0008924	0.016866	116.28	137.43	253.71	117.34	156.46	273.80	0.42311	0.49020	0.91331
48	1253.6	0.0008997	0.015951	119.28	135.30	254.58	120.41	154.17	274.57	0.43251	0.48001	0.91252
52	1386.2	0.0009151	0.014276	125.35	130.89	256.24	126.62	149.41	276.03	0.45136	0.45948	0.91084
56	1529.1	0.0009317	0.012782	131.52	126.29	257.81	132.94	144.41	277.35	0.47028	0.43870	0.90898
60	1682.8	0.0009498	0.011434	137.79	121.45	259.23	139.38	139.09	278.47	0.48930	0.41746	0.90676
65	1891.0	0.0009751	0.009959	145.80	115.06	260.86	147.64	132.05	279.69	0.51330	0.39048	0.90379
70	2118.2	0.0010037	0.008650	154.03	108.17	262.20	156.15	124.37	280.52	0.53763	0.36239	0.90002
75	2365.8	0.0010373	0.007486	162.55	100.62	263.17	165.01	115.87	280.88	0.56252	0.33279	0.89531
80	2635.3	0.0010774	0.006439	171.43	92.22	263.66	174.27	106.35	280.63	0.58812	0.30113	0.88925
85	2928.2	0.0011273	0.005484	180.81	82.64	263.45	184.11	95.39	279.51	0.61487	0.26632	0.88120
90	3246.9	0.0011938	0.004591	190.94	71.19	262.13	194.82	82.22	277.04	0.64354	0.22638	0.86991
95	3594.1	0.0012945	0.003713	202.49	56.25	258.73	207.14	64.94	272.08	0.67605	0.17638	0.85243
100	3975.1	0.0015269	0.002657	218.73	29.72	248.46	224.80	34.22	259.02	0.72224	0.09169	0.81393

Source of Data: Tables A-11 through A-13 are generated using the Engineering Equation Solver (EES) software developed by S. A. Klein and F. L. Alvarado. The routine used in calculations is the R134a, which is based on the fundamental equation of state developed by R. Tillner—Roth and H.D. Baehr, "An International Standard Formulation for the Thermodynamic Properties of 1,1,1,2-Tetrafluoroethane (HFC-134a) for temperatures from 170 K to 455 K and pressures up to 70 MPa," *J. Phys. Chem, Ref. Data*, Vol. 23, No. 5, 1994. The enthalpy and entropy values of saturated liquid are set to zero at -40° C (and -40° F).

TABLE A-12

Saturated refrigerant-134a—Pressure table

		Specific m³/		Inte	<i>rnal enei</i> kJ/kg	gy,		<i>Enthalpy,</i> kJ/kg			Entropy, kJ/kg·K	
Press., P kPa	Sat. temp., T_{sat} °C	Sat. liquid, v _f	Sat. vapor, v _g	Sat. liquid, u _f	Evap., u _{fg}	Sat. vapor, u_g	Sat. liquid, h _f	Evap., h _{fg}	Sat. vapor, h_g	Sat. liquid, s_f	Evap., s _{fg}	Sat. vapor, s _g
60	-36.95	0.0007097	0.31108	3.795	205.34	209.13	3.837	223.96	227.80	0.01633	0.94812	0.96445
70	-33.87	0.0007143	0.26921	7.672	203.23	210.90	7.722	222.02	229.74	0.03264	0.92783	0.96047
80	-31.13	0.0007184	0.23749	11.14	201.33	212.48	11.20	220.27	231.47	0.04707	0.91009	0.95716
90	-28.65	0.0007222	0.21261	14.30	199.60	213.90	14.36	218.67	233.04	0.06003	0.89431	0.95434
100	-26.37	0.0007258	0.19255	17.19	198.01	215.21	17.27	217.19	234.46	0.07182	0.88008	0.95191
120	-22.32	0.0007323	0.16216	22.38	195.15	217.53	22.47	214.52	236.99	0.09269	0.85520	0.94789
140	-18.77	0.0007381	0.14020	26.96	192.60	219.56	27.06	212.13	239.19	0.11080	0.83387	0.94467
160	-15.60	0.0007435	0.12355	31.06	190.31	221.37	31.18	209.96	241.14	0.12686	0.81517	0.94202
180	-12.73	0.0007485	0.11049	34.81	188.20	223.01	34.94	207.95	242.90	0.14131	0.79848	0.93979
200	-10.09	0.0007532	0.099951	38.26	186.25	224.51	38.41	206.09	244.50	0.15449	0.78339	0.93788
240	-5.38	0.0007618	0.083983	44.46	182.71	227.17	44.64	202.68	247.32	0.17786	0.75689	0.93475
280	-1.25	0.0007697	0.072434	49.95	179.54	229.49	50.16	199.61	249.77	0.19822	0.73406	0.93228
320	2.46	0.0007771	0.063681	54.90	176.65	231.55	55.14	196.78	251.93	0.21631	0.71395	0.93026
360	5.82	0.0007840	0.056809	59.42	173.99	233.41	59.70	194.15	253.86	0.23265	0.69591	0.92856
400	8.91	0.0007905	0.051266	63.61	171.49	235.10	63.92	191.68	255.61	0.24757	0.67954	0.92711
450	12.46	0.0007983	0.045677	68.44	168.58	237.03	68.80	188.78	257.58	0.26462	0.66093	0.92555
500	15.71	0.0008058	0.041168	72.92	165.86	238.77	73.32	186.04	259.36	0.28021	0.64399	0.92420
550	18.73	0.0008129	0.037452	77.09	163.29	240.38	77.54	183.44	260.98	0.29460	0.62842	
600	21.55	0.0008198	0.034335	81.01	160.84	241.86	81.50	180.95	262.46	0.30799	0.61398	0.92196
650	24.20	0.0008265	0.031680	84.72	158.51	243.23	85.26	178.56	263.82	0.32052	0.60048	0.92100
700	26.69	0.0008331	0.029392	88.24	156.27	244.51	88.82	176.26	265.08	0.33232	0.58780	0.92012
750	29.06	0.0008395	0.027398	91.59	154.11	245.70	92.22	174.03	266.25	0.34348	0.57582	0.91930
800	31.31	0.0008457	0.025645	94.80	152.02	246.82	95.48	171.86	267.34	0.35408	0.56445	0.91853
850	33.45	0.0008519	0.024091	97.88	150.00	247.88	98.61	169.75	268.36	0.36417	0.55362	0.91779
900	35.51	0.0008580	0.022703		148.03	248.88		167.69	269.31	0.37383	0.54326	0.91709
950	37.48	0.0008640	0.021456		146.11		104.52	165.68	270.20	0.38307	0.53333	0.91641
1000	39.37	0.0008700	0.020329		144.24	250.71		163.70	271.04	0.39196	0.52378	0.91574
1200	46.29	0.0008935	0.016728		137.12	253.84		156.12	273.92	0.42449	0.48870	0.91320
1400	52.40	0.0009167	0.014119		130.44	256.40		148.92	276.17	0.45325	0.45742	0.91067
1600	57.88	0.0009400	0.012134		124.05	258.50		141.96	277.92	0.47921	0.42881	0.90802
1800	62.87	0.0009639	0.010568		117.85	260.21		135.14	279.23	0.50304	0.40213	0.90517
2000	67.45	0.0009887	0.009297		111.75	261.56		128.36	280.15	0.52519	0.37684	0.90204
2500	77.54	0.0010567	0.006941		96.47	263.49		111.18	280.84	0.57542	0.31701	0.89243
3000	86.16	0.0011410	0.005272	183.09	80.17	263.26	186.51	92.57	279.08	0.62133	0.25759	0.87893

			_1	

Supe	rheated ref	rigerant-1	134a									
T	V	и	h	S	V	и	h	S	V	и	h	S
°C	m ³ /kg	kJ/kg	kJ/kg	kJ/kg∙K	m³/kg	kJ/kg	kJ/kg	kJ/kg·K	m³/kg	kJ/kg	kJ/kg	kJ/kg∙K
	P = 0.0	D6 MPa (7	$s_{\text{sat}} = -36$.95°C)	P = 0.	10 MPa (7	$rac{1}{sat} = -26$.37°C)	P=0.	14 MPa (7	$s_{\text{sat}} = -18$.77°C)
Sat.	0.31108		227.80	0.9645	0.19255		234.46		0.14020	219.56	239.19	0.9447
-20	0.33608	220.62	240.78	1.0175	0.19841	219.68	239.52	0.9721				
-10	0.35048	227.57		1.0478	0.20743	226.77		1.0031	0.14605		246.37	0.9724
0	0.36476	234.67		1.0775	0.21630	233.97	255.60	1.0333	0.15263		254.61	1.0032
10 20	0.37893 0.39302	241.94 249.37	264.68	1.1354	0.22506 0.23373	241.32 248.81	263.82 272.18	1.0628 1.0919	0.15908 0.16544		262.95 271.40	1.0331 1.0625
30	0.39302	256.97	281.39		0.23373	256.46	280.69	1.1204	0.10344	255.95	271.40	1.0023
40	0.42102	264.73		1.1916	0.25088	264.27	289.36	1.1485	0.17172		288.72	1.1196
50	0.43495	272.66		1.2192	0.25937	272.24	298.17	1.1762	0.18412		297.59	1.1475
60	0.44883	280.75	307.68		0.26783	280.36	307.15	1.2036	0.19025		306.61	1.1750
70	0.46269	289.01	316.77		0.27626	288.65	316.28	1.2306	0.19635		315.78	1.2021
80	0.47651	297.43	326.02	1.2998	0.28465	297.10	325.57	1.2573	0.20242	296.77	325.11	1.2289
90	0.49032	306.02	335.43	1.3261	0.29303	305.71	335.01	1.2836	0.20847	305.40	334.59	1.2554
100	0.50410	314.76	345.01	1.3521	0.30138	314.48	344.61	1.3097	0.21449	314.19	344.22	1.2815
	P=0.	18 MPa ($T_{\rm sat} = -12$	73°C)	P = 0.	.20 MPa ($T_{\rm sat} = -10$).09°C)	P = 0	.24 MPa ($T_{\rm sat} = -5.$	38°C)
Sat.	0.11049	223.01	242.90	0.9398	0.09995	224.51	244.50	0.9379	0.08398	227.17	247.32	0.9348
-10	0.11189		245.18	0.9485	0.09991	224.57		0.9381				
0	0.11722	232.49	253.59	0.9799	0.10481	232.11	253.07	0.9699	0.08617	231.30	251.98	0.9520
10	0.12240	240.02	262.05	1.0103	0.10955	239.69	261.60	1.0005	0.09026	239.00	260.66	0.9832
20	0.12748		270.60	1.0400	0.11418	247.36	270.20	1.0304	0.09423	246.76	269.38	1.0134
30	0.13248		279.27	1.0691	0.11874	255.16	278.91	1.0596	0.09812	254.63	278.17	1.0429
40	0.13741		288.07	1.0976	0.12322	263.09		1.0882	0.10193	262.61	287.07	1.0718
50	0.14230		297.00	1.1257	0.12766	271.16	296.70	1.1164	0.10570	270.73	296.09	1.1002
60	0.14715		306.07	1.1533	0.13206	279.38	305.79	1.1441	0.10942	278.98	305.24	1.1281
70 80	0.15196 0.15673		315.28 324.65	1.1806 1.2075	0.13641 0.14074	287.75 296.27	315.03 324.41	1.1714 1.1984	0.11310 0.11675	287.38 295.93	314.53 323.95	1.1555 1.1826
90	0.15075		334.16	1.2340	0.14504	304.93	333.94	1.2250	0.11073	304.62	333.51	1.2093
100	0.16622		343.82	1.2603	0.14933	313.75	343.62	1.2513	0.12398		343.22	
	P = 0		$T_{\text{sat}} = -1$			0.32 MPa					$(T_{\rm sat} = 8.9)$	
Sat.		229.49		0.9323	0.06368			0.9303	0.051266			
0	0.07282		250.85	0.9362								
10	0.07646	238.29	259.70	0.9681	0.06609	237.56	258.70	0.9545	0.051506	235.99	256.59	0.9306
20	0.07997	246.15		0.9987	0.06925	245.51	267.67	0.9856	0.054213		265.88	0.9628
30	0.08338		277.42	1.0285	0.07231	253.52	276.66	1.0158	0.056796		275.09	0.9937
40	0.08672	262.12		1.0577			285.72	1.0452	0.059292		284.32	1.0237
50				1.0862	0.07823	269.83		1.0739	0.061724			1.0529
60	0.09324			1.1143	0.08111			1.1022	0.064104			1.0814
70	0.09644			1.1419	0.08395	286.64		1.1299	0.066443		312.45	1.1095
80	0.09961		323.48	1.1690	0.08675	295.24	323.00	1.1572	0.068747		322.04	1.1370
90	0.10275		333.07	1.1958	0.08953	303.99	332.64	1.1841	0.071023			1.1641
100		313.17 322.18		1.2223 1.2484	0.09229	312.87 321.91	342.41 352.31	1.2106 1.2368	0.073274 0.075504		341.59	1.1908 1.2172
120		331.34		1.2742	0.09303	331.08		1.2627	0.075504			1.2172
130		340.65		1.2742	0.09775	340.41	372.55	1.2883	0.077717			1.2689
140		350.11		1.3251	0.10043	349.88		1.3136	0.073313			1.2943
113	0.11010	000.11	000.20	2.0201	0.10017	0.0.00	002.03	2.0100		5.5.12	552.25	1.25 10

TABLE A-13

	Superheated refrigerant-134a (Concluded)											
Supe	rheated refr	igerant-1	34a (<i>Coi</i>	ncluded)								
<i>T</i> °C	v m³/kg	<i>u</i> kJ/kg	<i>h</i> kJ/kg	<i>s</i> kJ/kg⋅K	ν m³/kg	<i>u</i> kJ/kg	<i>h</i> kJ/kg	<i>s</i> kJ/kg⋅K	v m³/kg	<i>u</i> kJ/kg	<i>h</i> kJ/kg	<i>s</i> kJ/kg⋅K
0.1		50 MPa (7				60 MPa (70 MPa (7		
Sat. 20	0.041168 0.042115		259.36 263.48		0.034335	241.86	262.46	0.9220	0.029392	244.51	265.08	0.9201
30	0.042113	250.86		0.9704	0.035984	249.24	270.83	0.9500	0.029966	247.49	268.47	0.9314
40	0.046456	259.27	282.50		0.037865				0.031696			0.9642
50	0.048499		291.98		0.039659				0.033322			0.9955
60	0.050485		301.51		0.041389				0.034875			1.0257
70	0.052427		311.12		0.043069			1.0706	0.036373			1.0550
80 90	0.054331 0.056205		320.82 330.63		0.044710 0.046318				0.037829 0.039250			1.0835 1.1115
100	0.058205		340.55		0.040318				0.039230			1.1113
110	0.059880	320.65	350.59		0.049458				0.042010			1.1659
120	0.061687		360.75		0.050997				0.043358			1.1925
130	0.063479		371.05		0.052519				0.044688			
140	0.065256	348.85	381.47		0.054027		380.68		0.046004			1.2445
150	0.067021		392.04		0.055522				0.047306			1.2700
160	0.068775	368.34	402.73	1.3250	0.057006	367.83	402.03	1.3089	0.048597	367.31	401.32	1.2952
	P = 0.3	80 MPa (7	$T_{\rm sat} = 31.3$	31°C)	P = 0	.90 MPa	$T_{\rm sat} = 35$.51°C)	P = 1.0	00 MPa (7	$T_{\rm sat} = 39.3$	37°C)
Sat.	0.025645				0.022686				0.020319		271.04	
40	0.027035			0.9481	0.023375				0.020406		271.73	
50	0.028547			0.9803	0.024809				0.021796			
60	0.029973		296.82 306.90		0.026146				0.023068		293.40	
70 80	0.031340 0.032659		316.99		0.027413 0.028630				0.024261 0.025398		303.87 314.27	
90	0.032033		327.12		0.028030				0.025338			1.0749
100	0.035193	309.17		1.1259	0.030951				0.027552			1.1032
110	0.036420		347.61		0.032068				0.028584		345.54	1.1309
120	0.037625	327.89		1.1798	0.033164				0.029592		356.08	1.1580
130	0.038813		368.47		0.034241				0.030581			1.1847
140	0.039985		379.07		0.035302				0.031554			
150 160	0.041143 0.042290	356.86 366.78	389.78 400.61		0.036349 0.037384		389.01 399.89	1.2468	0.032512 0.033457		388.24 399.17	
170	0.042290	376.83		1.3081	0.037384				0.033437			
180	0.043427	387.01	422.65		0.039423				0.035317		421.38	
100												
Sat.	P = 1 0.016728	20 MPa (7 253.84	$\frac{1}{1} = 46.2$		P = 1 0.014119	.40 MPa			P = 1.6 0.012134	60 MPa (7		
50	0.010728		278.28		0.014119	250.40	2/0.1/	0.9107	0.012134	230.30	217.32	0.3000
	0.017201				0.015005	264.46	285.47	0.9389	0.012372	260.91	280.71	0.9164
	0.019502				0.016060				0.013430	271.78	293.27	0.9536
	0.020529	286.77	311.40	1.0249	0.017023				0.014362			
90	0.021506		322.09		0.017923				0.015215			
100	0.022442		332.74		0.018778				0.016014			
110 120	0.023348 0.024228		343.41		0.019597 0.020388				0.016773 0.017500			
130	0.024228		364.12		0.020388				0.017500			
140	0.025000		375.74		0.021133				0.018201			
150	0.026753		386.68		0.022636				0.019545			
	0.027566		397.71		0.023355				0.020194			
	0.028367	374.80	408.84	1.2704	0.024061	373.75	407.43	1.2554	0.020830			
180	0.029158	385.10	420.09	1.2955	0.024757	384.12	418.78	1.2808	0.021456	383.13	417.46	1.2677

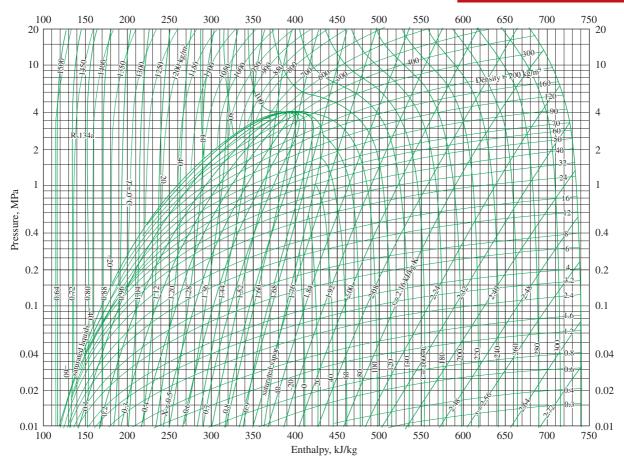


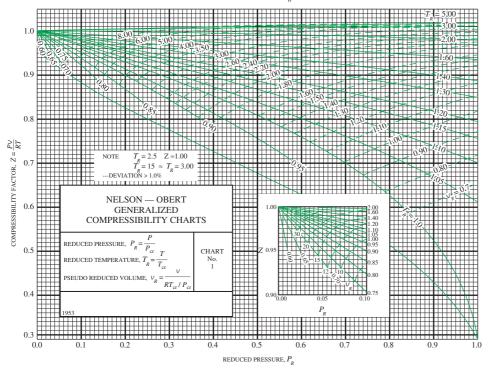
FIGURE A-14

P-h diagram for refrigerant-134a.

Note: The reference point used for the chart is different than that used in the R-134a tables. Therefore, problems should be solved using all property data either from the tables or from the chart, but not from both.

Source of Data: American Society of Heating, Refrigerating, and Air-Conditioning Engineers, Inc., Atlanta, GA.





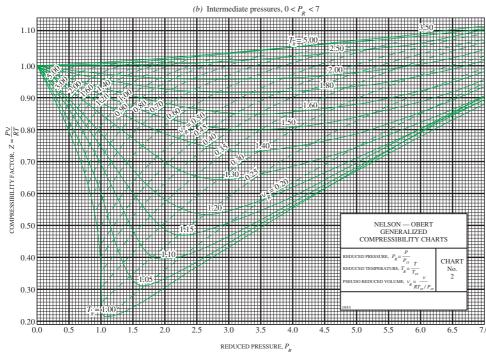


FIGURE A–15Nelson–Obert generalized compressibility chart.

 ${\it Used with permission of Dr. Edward E. Obert, University of Wisconsin.}$

TABLE A-16

Properties of the atmosphere at high altitude

Altitude,	Temperature, °C	Pressure, kPa	Gravity g, m/s ²	Speed of Sound, m/s	Density, kg/m³	Viscosity μ, kg/m·s	Thermal Conductivity, W/m·K
0	15.00	101.33	9.807	340.3	1.225	1.789×10^{-5} 1.783×10^{-5} 1.777×10^{-5} 1.771×10^{-5} 1.764×10^{-5}	0.0253
200	13.70	98.95	9.806	339.5	1.202		0.0252
400	12.40	96.61	9.805	338.8	1.179		0.0252
600	11.10	94.32	9.805	338.0	1.156		0.0251
800	9.80	92.08	9.804	337.2	1.134		0.0250
1000	8.50	89.88	9.804	336.4	1.112	1.758×10^{-5} 1.752×10^{-5} 1.745×10^{-5} 1.739×10^{-5} 1.732×10^{-5}	0.0249
1200	7.20	87.72	9.803	335.7	1.090		0.0248
1400	5.90	85.60	9.802	334.9	1.069		0.0247
1600	4.60	83.53	9.802	334.1	1.048		0.0245
1800	3.30	81.49	9.801	333.3	1.027		0.0244
2000	2.00	79.50	9.800	332.5	1.007	1.726×10^{-5} 1.720×10^{-5} 1.713×10^{-5} 1.707×10^{-5} 1.700×10^{-5}	0.0243
2200	0.70	77.55	9.800	331.7	0.987		0.0242
2400	-0.59	75.63	9.799	331.0	0.967		0.0241
2600	-1.89	73.76	9.799	330.2	0.947		0.0240
2800	-3.19	71.92	9.798	329.4	0.928		0.0239
3000	-4.49	70.12	9.797	328.6	0.909	1.694×10^{-5} 1.687×10^{-5} 1.681×10^{-5} 1.674×10^{-5} 1.668×10^{-5}	0.0238
3200	-5.79	68.36	9.797	327.8	0.891		0.0237
3400	-7.09	66.63	9.796	327.0	0.872		0.0236
3600	-8.39	64.94	9.796	326.2	0.854		0.0235
3800	-9.69	63.28	9.795	325.4	0.837		0.0234
4000	-10.98	61.66	9.794	324.6	0.819	1.661×10^{-5} 1.655×10^{-5} 1.648×10^{-5} 1.642×10^{-5} 1.635×10^{-5}	0.0233
4200	-12.3	60.07	9.794	323.8	0.802		0.0232
4400	-13.6	58.52	9.793	323.0	0.785		0.0231
4600	-14.9	57.00	9.793	322.2	0.769		0.0230
4800	-16.2	55.51	9.792	321.4	0.752		0.0229
5000	-17.5	54.05	9.791	320.5	0.736	1.628×10^{-5} 1.622×10^{-5} 1.615×10^{-5} 1.608×10^{-5} 1.602×10^{-5}	0.0228
5200	-18.8	52.62	9.791	319.7	0.721		0.0227
5400	-20.1	51.23	9.790	318.9	0.705		0.0226
5600	-21.4	49.86	9.789	318.1	0.690		0.0224
5800	-22.7	48.52	9.785	317.3	0.675		0.0223
6000	-24.0	47.22	9.788	316.5	0.660	1.595×10^{-5} 1.588×10^{-5} 1.582×10^{-5} 1.575×10^{-5} 1.568×10^{-5}	0.0222
6200	-25.3	45.94	9.788	315.6	0.646		0.0221
6400	-26.6	44.69	9.787	314.8	0.631		0.0220
6600	-27.9	43.47	9.786	314.0	0.617		0.0219
6800	-29.2	42.27	9.785	313.1	0.604		0.0218
7000	-30.5	41.11	9.785	312.3	0.590	1.561×10^{-5} 1.527×10^{-5} 1.493×10^{-5} 1.458×10^{-5}	0.0217
8000	-36.9	35.65	9.782	308.1	0.526		0.0212
9000	-43.4	30.80	9.779	303.8	0.467		0.0206
10,000	-49.9	26.50	9.776	299.5	0.414		0.0201
12,000	-56.5	19.40	9.770	295.1	0.312	1.422×10^{-5}	0.0195
14,000	-56.5	14.17	9.764	295.1	0.228		0.0195
16,000	-56.5	10.53	9.758	295.1	0.166		0.0195
18,000	-56.5	7.57	9.751	295.1	0.122		0.0195

Source of Data: U.S. Standard Atmosphere Supplements, U.S. Government Printing Office, 1966. Based on year-round mean conditions at 45° latitude and varies with the time of the year and the weather patterns. The conditions at sea level (z=0) are taken to be P=101.325 kPa, T=15°C, $\rho=1.2250$ kg/m³, g=9.80665 m²/s.

TABLE A-17

Ideal-gas properties of air

Ideal-	Ideal-gas properties of air										
T	h		И		s°	Τ	h		И		s°
K	kJ/kg	P_r	kJ/kg	V_r	kJ/kg·K	K	kJ/kg	P_r	kJ/kg	V_r	kJ/kg·K
200	199.97	0.3363	142.56	1707.0	1.29559	580	586.04	14.38	419.55	115.7	2.37348
210	209.97	0.3987	149.69	1512.0	1.34444	590	596.52	15.31	427.15	110.6	2.39140
220	219.97	0.4690	156.82	1346.0	1.39105	600	607.02	16.28	434.78	105.8	2.40902
230	230.02	0.5477	164.00	1205.0	1.43557	610	617.53	17.30	442.42	101.2	2.42644
240	240.02	0.6355	171.13	1084.0	1.47824	620	628.07	18.36	450.09	96.92	2.44356
250	250.05	0.7329	178.28	979.0	1.51917	630	638.63	19.84	457.78	92.84	2.46048
260	260.09	0.8405	185.45	887.8	1.55848	640 650	649.22	20.64	465.50	88.99	2.47716
270 280	270.11 280.13	0.9590 1.0889	192.60 199.75	808.0 738.0	1.59634 1.63279	650 660	659.84 670.47	21.86 23.13	473.25 481.01	85.34 81.89	2.49364 2.50985
285	285.14	1.1584	203.33	706.1	1.65055	670	681.14	24.46	488.81	78.61	2.52589
290	290.16	1.2311	206.91	676.1	1.66802	680	691.82	25.85	496.62	75.50	2.54175
295	290.16	1.2311	210.49	647.9	1.68515	690	702.52	27.29	504.45	73.50	2.54175
298	298.18	1.3543	212.64	631.9	1.69528	700	713.27	28.80	512.33	69.76	2.57277
300	300.19	1.3860	214.07	621.2	1.70203	710	724.04	30.38	520.23	67.07	2.58810
305	305.22	1.4686	217.67	596.0	1.71865	720	734.82	32.02	528.14	64.53	2.60319
310	310.24	1.5546	221.25	572.3	1.73498	730	745.62	33.72	536.07	62.13	2.61803
315	315.27	1.6442	224.85	549.8	1.75106	740	756.44	35.50	544.02	59.82	2.63280
320	320.29	1.7375	228.42	528.6	1.76690	750	767.29	37.35	551.99	57.63	2.64737
325	325.31	1.8345	232.02	508.4	1.78249	760	778.18	39.27	560.01	55.54	2.66176
330	330.34	1.9352	235.61	489.4	1.79783	780	800.03	43.35	576.12	51.64	2.69013
340	340.42	2.149	242.82	454.1	1.82790	800	821.95	47.75	592.30	48.08	2.71787
350	350.49	2.379	250.02	422.2	1.85708	820	843.98	52.59	608.59	44.84	2.74504
360	360.58	2.626	257.24	393.4	1.88543	840	866.08	57.60	624.95	41.85	2.77170
370	370.67	2.892	264.46	367.2	1.91313	860	888.27	63.09	641.40	39.12	2.79783
380	380.77	3.176	271.69	343.4	1.94001	880	910.56	68.98	657.95	36.61	2.82344
390	390.88	3.481	278.93	321.5	1.96633	900	932.93	75.29	674.58	34.31	2.84856
400	400.98	3.806	286.16	301.6	1.99194	920	955.38	82.05	691.28	32.18	2.87324
410	411.12	4.153	293.43	283.3	2.01699	940	977.92	89.28	708.08	30.22	2.89748
420 430	421.26 431.43	4.522 4.915	300.69 307.99	266.6 251.1	2.04142 2.06533	960 980	1000.55 1023.25	97.00 105.2	725.02 741.98	28.40 26.73	2.92128 2.94468
440 450	441.61 451.80	5.332 5.775	315.30 322.62	236.8 223.6	2.08870 2.11161	1000 1020	1046.04 1068.89	114.0 123.4	758.94 776.10	25.17 23.72	2.96770 2.99034
460	462.02	6.245	329.97	211.4	2.11101	1040	1003.85	133.3	793.36	23.72	3.01260
470	472.24	6.742	337.32	200.1	2.15604	1040	1114.86	143.9	810.62	21.14	3.03449
480	482.49	7.268	344.70	189.5	2.17760	1080	1137.89	155.2	827.88	19.98	3.05608
490	492.74	7.824	352.08		2.19876	1100	1161.07		845.33		
500	503.02	8.411	359.49	170.6	2.21952		1184.28	179.7	862.79	17.886	3.09825
510	513.32	9.031	366.92	162.1	2.23993	1140	1207.57	193.1	880.35	16.946	3.11883
520	523.63	9.684	374.36	154.1	2.25997	1160	1230.92	207.2	897.91	16.064	3.13916
530	533.98	10.37	381.84	146.7	2.27967	1180	1254.34	222.2	915.57	15.241	3.15916
540	544.35	11.10	389.34	139.7	2.29906	1200	1277.79	238.0	933.33	14.470	3.17888
550	555.74	11.86	396.86	133.1	2.31809	1220	1301.31	254.7	951.09	13.747	3.19834
560	565.17	12.66	404.42	127.0	2.33685	1240	1324.93	272.3	968.95	13.069	3.21751
570	575.59	13.50	411.97	121.2	2.35531						

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Ideal-gas properties of air (Concluded)

	Para bi abai tii	00 0. 0 (oomoraaoa,								
T	h		и		S°	T	h		и		S°
K	kJ/kg	P_r	kJ/kg	V_r	kJ/kg·K	K	kJ/kg	P_r	kJ/kg	V_r	kJ/kg∙K
1260	1348.55	290.8	986.90	12.435	3.23638	1600	1757.57	791.2	1298.30	5.804	3.52364
1280	1372.24	310.4	1004.76	11.835	3.25510	1620	1782.00	834.1	1316.96	5.574	3.53879
1300	1395.97	330.9	1022.82	11.275	3.27345	1640	1806.46	878.9	1335.72	5.355	3.55381
1320	1419.76	352.5	1040.88	10.747	3.29160	1660	1830.96	925.6	1354.48	5.147	3.56867
1340	1443.60	375.3	1058.94	10.247	3.30959	1680	1855.50	974.2	1373.24	4.949	3.58335
1360	1467.49	399.1	1077.10	9.780	3.32724	1700	1880.1	1025	1392.7	4.761	3.5979
1380	1491.44	424.2	1095.26	9.337	3.34474	1750	1941.6	1161	1439.8	4.328	3.6336
1400	1515.42	450.5	1113.52	8.919	3.36200	1800	2003.3	1310	1487.2	3.994	3.6684
1420	1539.44	478.0	1131.77	8.526	3.37901	1850	2065.3	1475	1534.9	3.601	3.7023
1440	1563.51	506.9	1150.13	8.153	3.39586	1900	2127.4	1655	1582.6	3.295	3.7354
1460	1587.63	537.1	1168.49	7.801	3.41247	1950	2189.7	1852	1630.6	3.022	3.7677
1480	1611.79	568.8	1186.95	7.468	3.42892	2000	2252.1	2068	1678.7	2.776	3.7994
1500	1635.97	601.9	1205.41	7.152	3.44516	2050	2314.6	2303	1726.8	2.555	3.8303
1520	1660.23	636.5	1223.87	6.854	3.46120	2100	2377.7	2559	1775.3	2.356	3.8605
1540	1684.51	672.8	1242.43	6.569	3.47712	2150	2440.3	2837	1823.8	2.175	3.8901
1560	1708.82	710.5	1260.99	6.301	3.49276	2200	2503.2	3138	1872.4	2.012	3.9191
1580	1733.17	750.0	1279.65	6.046	3.50829	2250	2566.4	3464	1921.3	1.864	3.9474

Note: The properties P_r (relative pressure) and v_r (relative specific volume) are dimensionless quantities used in the analysis of isentropic processes, and should not be confused with the properties pressure and specific volume.

Source of Data: Kenneth Wark, Thermodynamics, 4th ed. (New York: McGraw-Hill, 1983), pp. 785–86, table A–5. Originally published in J. H. Keenan and J. Kaye, Gas Tables (New York: John Wiley & Sons, 1948).

TABLE A-18

Ideal-gas properties of nitrogen, N_2

Т	<u></u>	\overline{u}	₹°	Т	<u></u>	ū	<u></u> s°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
0 220	0 6,391	0 4,562	0 182.639	600 610	17,563 17,864	12,574 12,792	212.066 212.564
230	6,683	4,770	183.938	620	18,166	13,011	213.055
240	6,975	4,979	185.180	630	18,468	13,230	213.541
250	7,266	5,188	186.370	640	18,772	13,450	214.018
260	7,558	5,396	187.514	650	19,075	13,671	214.489
270 280	7,849 8,141	5,604 5,813	188.614 189.673	660 670	19,380 19,685	13,892 14,114	214.954 215.413
290	8,432	6,021	190.695	680	19,991	14,337	215.866
298	8,669	6,190	191.502	690	20,297	14,560	216.314
300	8,723	6,229	191.682	700	20,604	14,784	216.756
310	9,014	6,437	192.638	710	20,912	15,008	217.192
320	9,306	6,645	193.562	720	21,220	15,234	217.624
330 340	9,597 9,888	6,853 7,061	194.459 195.328	730 740	21,529 21,839	15,460 15,686	218.059 218.472
350	10,180	7,001	195.328	740 750	22,149	15,080	218.889
360	10,180	7,270 7,478	196.173	760	22,149	16,141	219.301
370	10,763	7,687	197.794	770	22,772	16,370	219.709
380	11,055	7,895	198.572	780	23,085	16,599	220.113
390	11,347	8,104	199.331	790	23,398	16,830	220.512
400	11,640	8,314	200.071	800	23,714	17,061	220.907
410	11,932 12,225	8,523 8,733	200.794	810 820	24,027 24,342	17,292	221.298
420 430	12,225	8,943	201.499 202.189	830	24,658	17,524 17,757	221.684 222.067
440	12,811	9,153	202.863	840	24,974	17,797	222.447
450	13,105	9,363	203.523	850	25,292	18,224	222.822
460	13,399	9,574	204.170	860	25,610	18,459	223.194
470	13,693	9,786	204.803	870	25,928	18,695	223.562
480 490	13,988 14,285	9,997 10,210	205.424 206.033	880 890	26,248 26,568	18,931 19,168	223.927 224.288
500	14,283	10,210	206.630	900	26,890	19,108	224.647
510	14,876	10,423	207.216	910	27,210	19,407	225.002
520	15,172	10,848	207.792	920	27,532	19,883	225.353
530	15,469	11,062	208.358	930	27,854	20,122	225.701
540	15,766	11,277	208.914	940	28,178	20,362	226.047
550	16,064	11,492	209.461	950	28,501	20,603	226.389
560 570	16,363 16,662	11,707 11,923	209.999	960 970	28,826 29,151	20,844	226.728 227.064
580	16,962	12,139	210.528 211.049	970	29,151	21,086 21,328	227.064
590	17,262	12,356	211.562	990	29,803	21,571	227.728

TABLE A-18

Ideal-gas properties of nitrogen, N₂ (Concluded)

T	h	\overline{U}	5°	Т	\overline{h}		<u> </u>
K	kJ/kmol	u kJ/kmol	s kJ/kmol∙K	K	kJ/kmol	u kJ/kmol	s kJ/kmol∙K
1000	30,129	21,815	228.057	1760	56,227	41,594	247.396
1020 1040	30,784 31,442	22,304 22,795	228.706 229.344	1780 1800	56,938 57,651	42,139 42,685	247.798 248.195
1060 1080	32,101 32,762	23,288 23,782	229.973 230.591	1820 1840	58,363 59,075	43,231 43,777	248.589 248.979
1100	33,426	24,280	231.199	1860	59,790	44,324	249.365
1120 1140	34,092 34,760	24,780 25,282	231.799 232.391	1880 1900	60,504 61,220	44,873 45,423	249.748 250.128
1160	35,430	25,786	232.973	1920	61,936	45,973	250.502
1180 1200	36,104 36,777	26,291 26,799	233.549 234.115	1940 1960	62,654 63,381	46,524 47,075	250.874 251.242
1220	37,452	27,308	234.673	1980	64,090	47,627	251.607
1240 1260	38,129 38,807	27,819 28,331	235.223 235.766	2000 2050	64,810 66,612	48,181 49,567	251.969 252.858
1280	39,488	28,845	236.302	2100	68,417	50,957	253.726
1300 1320	40,170 40,853	29,361 29,378	236.831 237.353	2150 2200	70,226 72,040	52,351 53,749	254.578 255.412
1340	41,539	30,398	237.867	2250	73,856	55,149	256.227
1360 1380	42,227 42,915	30,919 31,441	238.376 238.878	2300 2350	75,676 77,496	56,553 57,958	257.027 257.810
1400	43,605	31,964	239.375	2400	79,320	59,366	258.580
1420 1440	44,295 44,988	32,489 33,014	239.865 240.350	2450 2500	81,149 82,981	60,779 62,195	259.332 260.073
1460 1480	45,682 46,377	33,543 34,071	240.827 241.301	2550 2600	84,814 86,650	63,613 65,033	260.799 261.512
1500	47,073	34,601	241.768	2650	88,488	66,455	262.213
1520 1540	47,771 48,470	35,133 35,665	242.228 242.685	2700 2750	90,328 92,171	67,880 69,306	262.902 263.577
1560	49,168	36,197	243.137	2800	94,014	70,734	264.241
1580	49,869	36,732	243.585	2850	95,859	72,163	264.895
1600 1620	50,571 51,275	37,268 37,806	244.028 244.464	2900 2950	97,705 99,556	73,593 75,028	265.538 266.170
1640 1660	51,980 52,686	38,344 38,884	244.896 245.324	3000 3050	101,407 103,260	76,464 77,902	266.793 267.404
1680	53,393	39,424	245.747	3100	105,260	77,902	268.007
1700 1720	54,099 54,807	39,965	246.166	3150 3200	106,972	80,782	268.601
1720	54,807 55,516	40,507 41,049	246.580 246.990	3200 3250	108,830 110,690	82,224 83,668	269.186 269.763

Source of Data: Tables A–18 through A–25 are adapted from Kenneth Wark, Thermodynamics, 4th ed. (New York: McGraw-Hill, 1983), pp. 787–98. Originally published in JANAF, Thermochemical Tables, NSRDS-NBS-37, 1971.

TABLE A-19

Ideal-gas properties of oxygen, O₂

Т	ħ	ū	<u></u> s °	Т	h	ū	₹°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
0	0	0	0	600	17,929	12,940	226.346
220	6,404	4,575	196.171	610	18,250	13,178	226.877
230	6,694	4,782	197.461	620	18,572	13,417	227.400
240	6,984	4,989	198.696	630	18,895	13,657	227.918
250	7,275	5,197	199.885	640	19,219	13,898	228.429
260	7,566	5,405	201.027	650	19,544	14,140	228.932
270	7,858	5,613	202.128	660	19,870	14,383	229.430
280	8,150	5,822	203.191	670	20,197	14,626	229.920
290	8,443	6,032	204.218	680	20,524	14,871	230.405
298	8,682	6,203	205.033	690	20,854	15,116	230.885
300	8,736	6,242	205.213	700	21,184	15,364	231.358
310	9,030	6,453	206.177	710	21,514	15,611	231.827
320	9,325	6,664	207.112	720	21,845	15,859	232.291
330	9,620	6,877	208.020	730	22,177	16,107	232.748
340	9,916	7,090	208.904	740	22,510	16,357	233.201
350	10,213	7,303	209.765	750	22,844	16,607	233.649
360	10,511	7,518	210.604	760	23,178	16,859	234.091
370	10,809	7,733	211.423	770	23,513	17,111	234.528
380	11,109	7,949	212.222	780	23,850	17,364	234.960
390	11,409	8,166	213.002	790	24,186	17,618	235.387
400	11,711	8,384	213.765	800	24,523	17,872	235.810
410	12,012	8,603	214.510	810	24,861	18,126	236.230
420	12,314	8,822	215.241	820	25,199	18,382	236.644
430	12,618	9,043	215.955	830	25,537	18,637	237.055
440	12,923	9,264	216.656	840	25,877	18,893	237.462
450	13,228	9,487	217.342	850	26,218	19,150	237.864
460	13,525	9,710	218.016	860	26,559	19,408	238.264
470	13,842	9,935	218.676	870	26,899	19,666	238.660
480	14,151	10,160	219.326	880	27,242	19,925	239.051
490	14,460	10,386	219.963	890	27,584	20,185	239.439
500	14,770	10,614	220.589	900	27,928	20,445	239.823
510	15,082	10,842	221.206	910	28,272	20,706	240.203
520	15,395	11,071	221.812	920	28,616	20,967	240.580
530	15,708	11,301	222.409	930	28,960	21,228	240.953
540	16,022	11,533	222.997	940	29,306	21,491	241.323
550	16,338	11,765	223.576	950	29,652	21,754	241.689
560	16,654	11,998	224.146	960	29,999	22,017	242.052
570	16,971	12,232	224.708	970	30,345	22,280	242.411
580	17,290	12,467	225.262	980	30,692	22,544	242.768
590	17,609	12,703	225.808	990	31,041	22,809	242.120

TABLE A-19

Ideal-gas properties of oxygen, O₂ (Concluded)

ideal-gas	Ideal-gas properties of oxygen, O ₂ (Concluded)										
T	\overline{h}	\overline{u}	<u></u> s°	Τ	h	\overline{u}	S °				
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K				
1000	31,389	23,075	243.471	1760	58,880	44,247	263.861				
1020	32,088	23,607	244.164	1780	59,624	44,825	264.283				
1040	32,789	24,142	244.844	1800	60,371	45,405	264.701				
1060	33,490	24,677	245.513	1820	61,118	45,986	265.113				
1080	34,194	25,214	246.171	1840	61,866	46,568	265.521				
1100	34,899	25,753	246.818	1860	62,616	47,151	265.925				
1120	35,606	26,294	247.454	1880	63,365	47,734	266.326				
1140	36,314	26,836	248.081	1900	64,116	48,319	266.722				
1160	37,023	27,379	248.698	1920	64,868	48,904	267.115				
1180	37,734	27,923	249.307	1940	65,620	49,490	267.505				
1200	38,447	28,469	249.906	1960	66,374	50,078	267.891				
1220	39,162	29,018	250.497	1980	67,127	50,665	268.275				
1240	39,877	29,568	251.079	2000	67,881	51,253	268.655				
1260	40,594	30,118	251.653	2050	69,772	52,727	269.588				
1280	41,312	30,670	252.219	2100	71,668	54,208	270.504				
1300	42,033	31,224	252.776	2150	73,573	55,697	271.399				
1320	42,753	31,778	253.325	2200	75,484	57,192	272.278				
1340	43,475	32,334	253.868	2250	77,397	58,690	273.136				
1360	44,198	32,891	254.404	2300	79,316	60,193	273.891				
1380	44,923	33,449	254.932	2350	81,243	61,704	274.809				
1400	45,648	34,008	255.454	2400	83,174	63,219	275.625				
1420	46,374	34,567	255.968	2450	85,112	64,742	276.424				
1440	47,102	35,129	256.475	2500	87,057	66,271	277.207				
1460	47,831	35,692	256.978	2550	89,004	67,802	277.979				
1480	48,561	36,256	257.474	2600	90,956	69,339	278.738				
1500	49,292	36,821	257.965	2650	92,916	70,883	279.485				
1520	50,024	37,387	258.450	2700	94,881	72,433	280.219				
1540	50,756	37,952	258.928	2750	96,852	73,987	280.942				
1560	51,490	38,520	259.402	2800	98,826	75,546	281.654				
1580	52,224	39,088	259.870	2850	100,808	77,112	282.357				
1600	52,961	39,658	260.333	2900	102,793	78,682	283.048				
1620	53,696	40,227	260.791	2950	104,785	80,258	283.728				
1640	54,434	40,799	261.242	3000	106,780	81,837	284.399				
1660	55,172	41,370	261.690	3050	108,778	83,419	285.060				
1680	55,912	41,944	262.132	3100	110,784	85,009	285.713				
1700	56,652	42,517	262.571	3150	112,795	86,601	286.355				
1720	57,394	43,093	263.005	3200	114,809	88,203	286.989				
1740	58,136	43,669	263.435	3250	116,827	89,804	287.614				

TABLE A-20

Ideal-gas properties of carbon dioxide, CO₂

T	\overline{h}	ū	\overline{s}°	Т	<u> </u>	\overline{u}	¯s°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
7 K 0 220 230 240 250 260 270 280 290 298 300 310 320 330 340 350 360 370 380 390 400 410 420 430 440 450 460	kJ/kmol 0 6,601 6,938 7,280 7,627 7,979 8,335 8,697 9,063 9,364 9,431 9,807 10,186 10,570 10,959 11,351 11,748 12,148 12,552 12,960 13,372 13,787 14,206 14,628 15,054 15,483 15,916	ū kJ/kmol 0 4,772 5,026 5,285 5,548 5,817 6,091 6,369 6,651 6,885 6,939 7,230 7,526 7,826 8,131 8,439 8,752 9,068 9,392 9,718 10,046 10,378 10,046 10,378 10,714 11,053 11,393 11,742 12,091	0 202.966 204.464 205.920 207.337 208.717 210.062 211.376 212.660 213.685 213.915 215.146 216.351 217.534 218.694 219.831 220.948 222.044 223.122 224.182 225.225 226.250 227.258 228.252 229.230 230.194 231.144	K 600 610 620 630 640 650 660 670 680 690 700 710 720 730 740 750 760 770 780 790 800 810 820 830 840 850 860	kJ/kmol 22,280 22,754 23,231 23,709 24,190 24,674 25,160 25,648 26,138 26,631 27,125 27,622 28,121 28,622 29,124 29,629 30,135 30,644 31,154 31,665 32,179 32,694 33,212 33,730 34,251 34,773 35,296	kJ/kmol 17,291 17,683 18,076 18,471 18,869 19,270 19,672 20,078 20,484 20,894 21,305 21,719 22,134 22,522 22,972 23,393 23,817 24,242 24,669 25,097 25,527 25,959 26,394 26,829 27,267 27,706 28,125	kJ/kmol-K 243.199 243.983 244.758 245.524 246.282 247.032 247.773 248.507 249.233 249.952 250.663 251.368 252.065 252.755 253.439 254.117 254.787 255.452 256.110 256.762 257.408 258.048 258.048 258.048 258.048 258.048 258.048 258.048 258.048
450	15,483 15,916 16,351 16,791	11,742 12,091 12,444 12,800	230.194 231.144 232.080 233.004	850	34,773 35,296 35,821 36,347	27,706 28,125 28,588 29,031	260.551 261.164 261.770 262.371
500 510 520 530 540 550 560 570	17,232 17,678 18,126 18,576 19,029 19,485 19,945 20,407 20,870	13,158 13,521 13,885 14,253 14,622 14,996 15,372 15,751 16,131	233.916 234.814 235.700 236.575 237.439 238.292 239.135 239.962 240.789	900 910 920 930 940 950 960 970	36,876 37,405 37,935 38,467 39,000 39,535 40,070 40,607 41,145	29,476 29,922 30,369 30,818 31,268 31,719 32,171 32,625 33,081	262.968 263.559 264.146 264.728 265.304 265.877 266.444 267.007 267.566
580 590	21,337 21,807	16,515 16,902	241.602 242.405	980 990	41,685 42,226	33,537 33,995	268.119 268.670

TABLE A-20

Ideal-gas properties of carbon dioxide, CO₂ (Concluded)

Ideal-gas	properties of ca	arbon dioxide, Co	J ₂ (Concluded)				
T	\overline{h}	ū	$\overline{\mathcal{S}}^{\circ}$	Τ	\overline{h}	\overline{u}	$\overline{\mathcal{S}}^{\circ}$
K	kJ/kmol	kJ/kmol	kJ/kmol∙K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
1000	42,769	34,455	269.215	1760	86,420	71,787	301.543
1020	43,859	35,378	270.293	1780	87,612	72,812	302.217
1040	44,953	36,306	271.354	1800	88,806	73,840	302.884
1060	46,051	37,238	272.400	1820	90,000	74,868	303.544
1080	47,153	38,174	273.430	1840	91,196	75,897	304.198
1100	48,258	39,112	274.445	1860	92,394	76,929	304.845
1120	49,369	40,057	275.444	1880	93,593	77,962	305.487
1140	50,484	41,006	276.430	1900	94,793	78,996	306.122
1160	51,602	41,957	277.403	1920	95,995	80,031	306.751
1180	52,724	42,913	278.361	1940	97,197	81,067	307.374
1200	53,848	43,871	297.307	1960	98,401	82,105	307.992
1220	54,977	44,834	280.238	1980	99,606	83,144	308.604
1240	56,108	45,799	281.158	2000	100,804	84,185	309.210
1260	57,244	46,768	282.066	2050	103,835	86,791	310.701
1280	58,381	47,739	282.962	2100	106,864	89,404	312.160
1300	59,522	48,713	283.847	2150	109,898	92,023	313.589
1320	60,666	49,691	284.722	2200	112,939	94,648	314.988
1340	61,813	50,672	285.586	2250	115,984	97,277	316.356
1360	62,963	51,656	286.439	2300	119,035	99,912	317.695
1380	64,116	52,643	287.283	2350	122,091	102,552	319.011
1400	65,271	53,631	288.106	2400	125,152	105,197	320.302
1420	66,427	54,621	288.934	2450	128,219	107,849	321.566
1440	67,586	55,614	289.743	2500	131,290	110,504	322.808
1460	68,748	56,609	290.542	2550	134,368	113,166	324.026
1480	66,911	57,606	291.333	2600	137,449	115,832	325.222
1500	71,078	58,606	292.114	2650	140,533	118,500	326.396
1520	72,246	59,609	292.888	2700	143,620	121,172	327.549
1540	73,417	60,613	292.654	2750	146,713	123,849	328.684
1560	74,590	61,620	294.411	2800	149,808	126,528	329.800
1580	76,767	62,630	295.161	2850	152,908	129,212	330.896
1600	76,944	63,741	295.901	2900	156,009	131,898	331.975
1620	78,123	64,653	296.632	2950	159,117	134,589	333.037
1640	79,303	65,668	297.356	3000	162,226	137,283	334.084
1660	80,486	66,592	298.072	3050	165,341	139,982	335.114
1680	81,670	67,702	298.781	3100	168,456	142,681	336.126
1700	82,856	68,721	299.482	3150	171,576	145,385	337.124
1720	84,043	69,742	300.177	3200	174,695	148,089	338.109
1740	85,231	70,764	300.863	3250	177,822	150,801	339.069

TABLE A-21

Ideal-gas properties of carbon monoxide, CO

	$\frac{1}{h}$	<u> </u>	\overline{s}°	7	\overline{h}		
<i>T</i> K	kJ/kmol	\overline{u} kJ/kmol	kJ/kmol·K	<i>T</i> K	rı kJ/kmol	<i>ū</i> kJ/kmol	kJ/kmol·K
	0	0	0				
0 220	6,391	4,562	188.683	600 610	17,611 17,915	12,622 12,843	218.204 218.708
230	6,683	4,771	189.980	620	18,221	13,066	219.205
240	6,975	4,979	191.221	630	18,527	13,289	219.695
250	7,266	5,188	192.411	640	18,833	13,512	220.179
260	7,558	5,396	193.554	650	19,141	13,736	220.656
270	7,849	5,604	194.654	660	19,449	13,962	221.127
280	8,140	5,812	195.713	670	19,758	14,187	221.592
290 298	8,432 8,669	6,020 6,190	196.735 197.543	680 690	20,068 20,378	14,414 14,641	222.052 222.505
300	8,723	6,229	197.723	700	20,578	14,870	222.953
310	9,014	6,437	197.725	700	21,002	15,099	223.396
320	9,306	6,645	199.603	720	21,315	15,328	223.833
330	9,597	6,854	200.500	730	21,628	15,558	224.265
340	9,889	7,062	201.371	740	21,943	15,789	224.692
350	10,181	7,271	202.217	750	22,258	16,022	225.115
360	10,473	7,480	203.040	760	22,573	16,255	225.533
370 380	10,765 11,058	7,689 7,899	203.842 204.622	770 780	22,890 23,208	16,488 16,723	225.947 226.357
390	11,351	8,108	205.383	790	23,526	16,723	226.762
400	11,644	8,319	206.125	800	23,844	17,193	227.162
410	11,938	8,529	206.850	810	24,164	17,429	227.559
420	12,232	8,740	207.549	820	24,483	17,665	227.952
430	12,526	8,951	208.252	830	24,803	17,902	228.339
440	12,821	9,163	208.929	840	25,124	18,140	228.724
450	13,116	9,375	209.593	850	25,446	18,379	229.106
460 470	13,412 13,708	9,587 9,800	210.243 210.880	860 870	25,768 26,091	18,617 18,858	229.482 229.856
480	14,005	10,014	211.504	880	26,415	19,099	230.227
490	14,302	10,228	212.117	890	26,740	19,341	230.593
500	14,600	10,443	212.719	900	27,066	19,583	230.957
510	14,898	10,658	213.310	910	27,392	19,826	231.317
520	15,197	10,874	213.890	920	27,719	20,070	231.674
530	15,497	11,090	214.460	930	28,046	20,314	232.028
540	15,797	11,307	215.020	940	28,375	20,559	232.379
550 560	16,097 16,399	11,524 11,743	215.572 216.115	950 960	28,703 29,033	20,805 21,051	232.727 233.072
570	16,701	11,743	216.115	970	29,033	21,031	233.413
580	17,003	12,181	217.175	980	29,693	21,545	233.752
590	17,307	12,401	217.693	990	30,024	21,793	234.088

TABLE A-21

Ideal-gas properties of carbon monoxide, CO (Concluded)

ideai-gas							
T	\overline{h}	\overline{u}	$\overline{\mathcal{S}}^{\circ}$	Τ	Ī	\overline{u}	$\overline{\mathcal{S}}^{\circ}$
K	kJ/kmol	kJ/kmol	kJ/kmol∙K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
1000	30,355	22,041	234.421	1760	56,756	42,123	253.991
1020	31,020	22,540	235.079	1780	57,473	42,673	254.398
1040	31,688	23,041	235.728	1800	58,191	43,225	254.797
1060	32,357	23,544	236.364	1820	58,910	43,778	255.194
1080	33,029	24,049	236.992	1840	59,629	44,331	255.587
1100	33,702	24,557	237.609	1860	60,351	44,886	255.976
1120	34,377	25,065	238.217	1880	61,072	45,441	256.361
1140	35,054	25,575	238.817	1900	61,794	45,997	256.743
1160	35,733	26,088	239.407	1920	62,516	46,552	257.122
1180	36,406	26,602	239.989	1940	63,238	47,108	257.497
1200	37,095	27,118	240.663	1960	63,961	47,665	257.868
1220	37,780	27,637	241.128	1980	64,684	48,221	258.236
1240	38,466	28,426	241.686	2000	65,408	48,780	258.600
1260	39,154	28,678	242.236	2050	67,224	50,179	259.494
1280	39,844	29,201	242.780	2100	69,044	51,584	260.370
1300	40,534	29,725	243.316	2150	70,864	52,988	261.226
1320	41,226	30,251	243.844	2200	72,688	54,396	262.065
1340	41,919	30,778	244.366	2250	74,516	55,809	262.887
1360	42,613	31,306	244.880	2300	76,345	57,222	263.692
1380	43,309	31,836	245.388	2350	78,178	58,640	264.480
1400	44,007	32,367	245.889	2400	80,015	60,060	265.253
1420	44,707	32,900	246.385	2450	81,852	61,482	266.012
1440	45,408	33,434	246.876	2500	83,692	62,906	266.755
1460	46,110	33,971	247.360	2550	85,537	64,335	267.485
1480	46,813	34,508	247.839	2600	87,383	65,766	268.202
1500	47,517	35,046	248.312	2650	89,230	67,197	268.905
1520	48,222	35,584	248.778	2700	91,077	68,628	269.596
1540	48,928	36,124	249.240	2750	92,930	70,066	270.285
1560	49,635	36,665	249.695	2800	94,784	71,504	270.943
1580	50,344	37,207	250.147	2850	96,639	72,945	271.602
1600	51,053	37,750	250.592	2900	98,495	74,383	272.249
1620	51,763	38,293	251.033	2950	100,352	75,825	272.884
1640	52,472	38,837	251.470	3000	102,210	77,267	273.508
1660	53,184	39,382	251.901	3050	104,073	78,715	274.123
1680	53,895	39,927	252.329	3100	105,939	80,164	274.730
1700	54,609	40,474	252.751	3150	107,802	81,612	275.326
1720	55,323	41,023	253.169	3200	109,667	83,061	275.914
1740	56,039	41,572	253.582	3250	111,534	84,513	276.494

TABLE A-22

Ideal-gas properties of hydrogen, H₂

T	$\frac{\overline{h}}{h}$	\overline{u}		Т	<u></u>	\overline{u}	\overline{s}°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
0	0	0	0	1440	42,808	30,835	177.410
260	7,370	5,209	126.636	1480	44,091	31,786	178.291
270	7,657	5,412	127.719	1520	45,384	32,746	179.153
280	7,945	5,617	128.765	1560	46,683	33,713	179.995
290	8,233	5,822	129.775	1600	47,990	34,687	180.820
298	8,468	5,989	130.574	1640	49,303	35,668	181.632
300	8,522	6,027	130.754	1680	50,622	36,654	182.428
320	9,100	6,440	132.621	1720	51,947	37,646	183.208
340	9,680	6,853	134.378	1760	53,279	38,645	183.973
360	10,262	7,268	136.039	1800	54,618	39,652	184.724
380	10,843	7,684	137.612	1840	55,962	40,663	185.463
400	11,426	8,100	139.106	1880	57,311	41,680	186.190
420	12,010	8,518	140.529	1920	58,668	42,705	186.904
440	12,594	8,936	141.888	1960	60,031	43,735	187.607
460	13,179	9,355	143.187	2000	61,400	44,771	188.297
480	13,764	9,773	144.432	2050	63,119	46,074	189.148
500	14,350	10,193	145.628	2100	64,847	47,386	189.979
520	14,935	10,611	146.775	2150	66,584	48,708	190.796
560	16,107	11,451	148.945	2200	68,328	50,037	191.598
600	17,280	12,291	150.968	2250	70,080	51,373	192.385
640	18,453	13,133	152.863	2300	71,839	52,716	193.159
680	19,630	13,976	154.645	2350	73,608	54,069	193.921
720	20,807	14,821	156.328	2400	75,383	55,429	194.669
760	21,988	15,669	157.923	2450	77,168	56,798	195.403
800	23,171	16,520	159.440	2500	78,960	58,175	196.125
840	24,359	17,375	160.891	2550	80,755	59,554	196.837
880	25,551	18,235	162.277	2600	82,558	60,941	197.539
920	26,747	19,098	163.607	2650	84,368	62,335	198.229
960	27,948	19,966	164.884	2700	86,186	63,737	198.907
1000	29,154	20,839	166.114	2750	88,008	65,144	199.575
1040	30,364	21,717	167.300	2800	89,838	66,558	200.234
1080	31,580	22,601	168.449	2850	91,671	67,976	200.885
1120	32,802	23,490	169.560	2900	93,512	69,401	201.527
1160	34,028	24,384	170.636	2950	95,358	70,831	202.157
1200	35,262	25,284	171.682	3000	97,211	72,268	202.778
1240	36,502	26,192	172.698	3050	99,065	73,707	203.391
1280	37,749	27,106	173.687	3100	100,926	75,152	203.995
1320	39,002	28,027	174.652	3150	102,793	76,604	204.592
1360	40,263	28,955	175.593	3200	104,667	78,061	205.181
1400	41,530	29,889	176.510	3250	106,545	79,523	205.765

TABLE A-23

Ideal-gas properties of water vapor, H₂O

				_	7	_	
T	h	\overline{u}	<u>s</u> °	T	<u> </u>	\overline{u}	<u>s</u> °
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
0	0	0	0	600	20,402	15,413	212.920
220	7,295	5,466	178.576	610	20,765	15,693	213.529
230	7,628	5,715	180.054	620	21,130	15,975	214.122
240 250	7,961 8,294	5,965 6,215	181.471 182.831	630 640	21,495 21,862	16,257 16,541	214.707 215.285
260 270	8,627 8,961	6,466 6,716	184.139 185.399	650 660	22,230 22,600	16,826 17,112	215.856 216.419
280	9,296	6,968	186.616	670	22,970	17,399	216.976
290	9,631	7,219	187.791	680	23,342	17,688	217.527
298	9,904	7,425	188.720	690	23,714	17,978	218.071
300	9,966	7,472	188.928	700	24,088	18,268	218.610
310	10,302	7,725	190.030	710	24,464	18,561	219.142
320	10,639	7,978	191.098	720	24,840	18,854	219.668
330 340	10,976 11,314	8,232 8,487	192.136 193.144	730 740	25,218 25,597	19,148 19,444	220.189 220.707
350	11,652	8,742	194.125	740 750	25,977	19,741	221.215
360	11,992	8,998	195.081	760 760	26,358	20,039	221.720
370	12,331	9,255	196.012	770	26,741	20,339	222.221
380	12,672	9,513	196.920	780	27,125	20,639	222.717
390	13,014	9,771	197.807	790	27,510	20,941	223.207
400	13,356	10,030	198.673	800	27,896	21,245	223.693
410	13,699	10,290	199.521	810	28,284	21,549	224.174
420	14,043	10,551	200.350	820	28,672	21,855	224.651
430 440	14,388 14,734	10,813 11,075	201.160 201.955	830 840	29,062 29,454	22,162 22,470	225.123 225.592
450	15,080	11,339	202.734	850	29,846	22,779	226.057
460	15,428	11,603	202.734	860	30,240	23,090	226.517
470	15,777	11,869	204.247	870	30,635	23,402	226.973
480	16,126	12,135	204.982	880	31,032	23,715	227.426
490	16,477	12,403	205.705	890	31,429	24,029	227.875
500	16,828	12,671	206.413	900	31,828	24,345	228.321
510	17,181	12,940	207.112	910	32,228	24,662	228.763
520	17,534	13,211	207.799	920	32,629	24,980	229.202
530 540	17,889 18,245	13,482 13,755	208.475 209.139	930 940	33,032 33,436	25,300 25,621	229.637 230.070
550	18,601	14,028	209.795	950	33,841	25,943	230.499
560	18,959	14,028	210.440	960	34,247	26,265	230.499
570	19,318	14,579	211.075	970	34,653	26,588	231.347
580	19,678	14,856	211.702	980	35,061	26,913	231.767
590	20,039	15,134	212.320	990	35,472	27,240	232.184

TABLE A-23

Ideal-gas properties of water vapor, H₂O (*Continued*)

T	$\frac{\overline{h}}{h}$	\overline{u}		Т	h	ū	\overline{s}°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
1000	35,882	27,568	232.597	1760	70,535	55,902	258.151
1020	36,709	28,228	233.415	1780	71,523	56,723	258.708
1040	37,542	28,895	234.223	1800	72,513	57,547	259.262
1060	38,380	29,567	235.020	1820	73,507	58,375	259.811
1080	39,223	30,243	235.806	1840	74,506	59,207	260.357
1100	40,071	30,925	236.584	1860	75,506	60,042	260.898
1120	40,923	31,611	237.352	1880	76,511	60,880	261.436
1140	41,780	32,301	238.110	1900	77,517	61,720	261.969
1160	42,642	32,997	238.859	1920	78,527	62,564	262.497
1180	43,509	33,698	239.600	1940	79,540	63,411	263.022
1200	44,380	34,403	240.333	1960	80,555	64,259	263.542
1220	45,256	35,112	241.057	1980	81,573	65,111	264.059
1240	46,137	35,827	241.773	2000	82,593	65,965	264.571
1260	47,022	36,546	242.482	2050	85,156	68,111	265.838
1280	47,912	37,270	243.183	2100	87,735	70,275	267.081
1300	48,807	38,000	243.877	2150	90,330	72,454	268.301
1320	49,707	38,732	244.564	2200	92,940	74,649	269.500
1340	50,612	39,470	245.243	2250	95,562	76,855	270.679
1360	51,521	40,213	245.915	2300	98,199	79,076	271.839
1380	52,434	40,960	246.582	2350	100,846	81,308	272.978
1400	53,351	41,711	247.241	2400	103,508	83,553	274.098
1420	54,273	42,466	247.895	2450	106,183	85,811	275.201
1440	55,198	43,226	248.543	2500	108,868	88,082	276.286
1460	56,128	43,989	249.185	2550	111,565	90,364	277.354
1480	57,062	44,756	249.820	2600	114,273	92,656	278.407
1500	57,999	45,528	250.450	2650	116,991	94,958	279.441
1520	58,942	46,304	251.074	2700	119,717	97,269	280.462
1540	59,888	47,084	251.693	2750	122,453	99,588	281.464
1560	60,838	47,868	252.305	2800	125,198	101,917	282.453
1580	61,792	48,655	252.912	2850	127,952	104,256	283.429
1600	62,748	49,445	253.513	2900	130,717	106,605	284.390
1620	63,709	50,240	254.111	2950	133,486	108,959	285.338
1640	64,675	51,039	254.703	3000	136,264	111,321	286.273
1660	65,643	51,841	255.290	3050	139,051	113,692	287.194
1680	66,614	52,646	255.873	3100	141,846	116,072	288.102
1700	67,589	53,455	256.450	3150	144,648	118,458	288.999
1720	68,567	54,267	257.022	3200	147,457	120,851	289.884
1740	69,550	55,083	257.589	3250	150,272	123,250	290.756

TAB	LE /	4-24				
Ideal	l-gas	properties	of	monatomic	oxygen,	0
		<u>_</u>				

TABLE A-25

2300

2350

73,462

75,236

54,339

55,697

Т	<u></u>	ū	<u></u> s°	Т	Ī	ū	 s°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol∙K
0	0	0	0	2400	50,894	30,940	204.932
298	6,852	4,373	160.944	2450	51,936	31,566	205.362
300	6,892	4,398	161.079	2500	52,979	32,193	205.783
500	11,197	7,040	172.088	2550	54,021	32,820	206.196
1000	21,713	13,398	186.678	2600	55,064	33,447	206.601
1500	32,150	19,679	195.143	2650	56,108	34,075	206.999
1600	34,234	20,931	196.488	2700	57,152	34,703	207.389
1700	36,317	22,183	197.751	2750	58,196	35,332	207.772
1800	38,400	23,434	198.941	2800	59,241	35,961	208.148
1900	40,482	24,685	200.067	2850	60,286	36,590	208.518
2000	42,564	25,935	201.135	2900	61,332	37,220	208.882
2050	43,605	26,560	201.649	2950	62,378	37,851	209.240
2100	44,646	27,186	202.151	3000	63,425	38,482	209.592
2150	45,687	27,811	202.641	3100	65,520	39,746	210.279
2200	46,728	28,436	203.119	3200	67,619	41,013	210.945
2250	47,769	29,062	203.588	3300	69,720	42,283	211.592
2300	48,811	29,688	204.045	3400	71,824	43,556	212.220
2350	49,852	30,314	204.493	3500	73,932	44,832	212.831

Ideal-gas	properties of h	ydroxyl, OH					
Т	h	\overline{u}	₹°	Т	\overline{h}	\overline{u}	₹°
K	kJ/kmol	kJ/kmol	kJ/kmol·K	K	kJ/kmol	kJ/kmol	kJ/kmol·K
0	0	0	0	2400	77,015	57,061	248.628
298	9,188	6,709	183.594	2450	78,801	58,431	249.364
300	9,244	6,749	183.779	2500	80,592	59,806	250.088
500	15,181	11,024	198.955	2550	82,388	61,186	250.799
1000	30,123	21,809	219.624	2600	84,189	62,572	251.499
1500	46,046	33,575	232.506	2650	85,995	63,962	252.187
1600	49,358	36,055	234.642	2700	87,806	65,358	252.864
1700	52,706	38,571	236.672	2750	89,622	66,757	253.530
1800	56,089	41,123	238.606	2800	91,442	68,162	254.186
1900	59,505	43,708	240.453	2850	93,266	69,570	254.832
2000	62,952	46,323	242.221	2900	95,095	70,983	255.468
2050	64,687	47,642	243.077	2950	96,927	72,400	256.094
2100	66,428	48,968	243.917	3000	98,763	73,820	256.712
2150	68,177	50,301	244.740	3100	102,447	76,673	257.919
2200	69,932	51,641	245.547	3200	106,145	79,539	259.093
2250	71,694	52,987	246.338	3300	109,855	82,418	260.235

3400

3500

113,578

117,312

85,309

88,212

261.347

262.429

247.116

247.879

TABLE A-26

Enthalpy of formation, Gibbs function of formation, and absolute entropy at 25°C. 1 atm

		$\overline{h}_{\scriptscriptstyle f}^{\circ}$	$\overline{g}_{\scriptscriptstyle f}^{\circ}$	<u></u> s°
Substance	Formula	kJ/kmol	kJ/kmol	kJ/kmol∙K
Carbon	C(s)	0	0	5.74
Hydrogen	$H_2(g)$	0	0	130.68
Nitrogen	$N_2(g)$	0	0	191.61
Oxygen	$O_{2}^{2}(g)$	0	0	205.04
Carbon monoxide	CO(g)	-110,530	-137,150	197.65
Carbon dioxide	CO ₂ (g)	-393,520	-394,360	213.80
Water vapor	$H_2\bar{O}(g)$	-241,820	-228,590	188.83
Water	H ₂ O(ℓ)	-285,830	-237,180	69.92
Hydrogen peroxide	$H_2O_2(g)$	-136,310	-105,600	232.63
Ammonia	$NH_3(g)$	-46,190	-16,590	192.33
Methane	$CH_4(g)$	-74,850	-50,790	186.16
Acetylene	$C_2H_2(g)$	+226,730	+209,170	200.85
Ethylene	$C_2H_4(g)$	+52,280	+68,120	219.83
Ethane	$C_2H_6(g)$	-84,680	-32,890	229.49
Propylene	$C_3H_6(g)$	+20,410	+62,720	266.94
Propane	$C_3H_8(g)$	-103,850	-23,490	269.91
<i>n</i> -Butane	$C_4H_{10}(g)$	-126,150	-15,710	310.12
<i>n</i> -Octane	$C_8H_{18}(g)$	-208,450	+16,530	466.73
<i>n</i> -Octane	$C_8H_{18}(\ell)$	-249,950	+6,610	360.79
<i>n</i> -Dodecane	$C_{12}H_{26}(g)$	-291,010	+50,150	622.83
Benzene	$C_6H_6(g)$	+82,930	+129,660	269.20
Methyl alcohol	CH ₃ OH(<i>g</i>)	-200,670	-162,000	239.70
Methyl alcohol	CH ₃ OH(ℓ)	-238,660	-166,360	126.80
Ethyl alcohol	$C_2H_5OH(g)$	-235,310	-168,570	282.59
Ethyl alcohol	$C_2H_5OH(\ell)$	-277,690	-174,890	160.70
Oxygen	O(g)	+249,190	+231,770	161.06
Hydrogen	H(g)	+218,000	+203,290	114.72
Nitrogen	N(g)	+472,650	+455,510	153.30 183.70
Hydroxyl	OH(<i>g</i>)	+39,460	+34,280	183.70

Source of Data: From JANAF, Thermochemical Tables (Midland, MI: Dow Chemical Co., 1971); Selected Values of Chemical Thermodynamic Properties, NBS Technical Note 270-3, 1968; and API Research Project 44 (Carnegie Press, 1953).

TABLE A-27

Properties of some common fuels and hydrocarbons

Fuel (phase)	Formula	Molar mass, kg/kmol	Density,¹ kg/L	Enthalpy of vaporization, ² kJ/kg	Specific heat, ¹ c _p kJ/kg·K	Higher heating value, ³ kJ/kg	Lower heating value, ³ kJ/kg
Carbon (s)	С	12.011	2	_	0.708	32,800	32,800
Hydrogen (g)	H ₂	2.016	_	_	14.4	141,800	120,000
Carbon monoxide (g)	CO	28.013	_	_	1.05	10,100	10,100
Methane (g)	CH₄	16.043	_	509	2.20	55,530	50,050
Methanol (ℓ)	CH₄O	32.042	0.790	1168	2.53	22,660	19,920
Acetylene (g)	C_2H_2	26.038	_	_	1.69	49,970	48,280
Ethane (g)	C_2H_6	30.070	_	172	1.75	51,900	47,520
Ethanol (ℓ)	C ₂ H ₆ O	46.069	0.790	919	2.44	29,670	26,810
Propane (ℓ)	C ₃ H ₈	44.097	0.500	335	2.77	50,330	46,340
Butane (ℓ)	C_4H_{10}	58.123	0.579	362	2.42	49,150	45,370
1-Pentene (ℓ)	C_5H_{10}	70.134	0.641	363	2.20	47,760	44,630
Isopentane (ℓ)	C_5H_{12}	72.150	0.626	_	2.32	48,570	44,910
Benzene (ℓ)	C_6H_6	78.114	0.877	433	1.72	41,800	40,100
Hexene (ℓ)	C_6H_{12}	84.161	0.673	392	1.84	47,500	44,400
Hexane (ℓ)	C_6H_{14}	86.177	0.660	366	2.27	48,310	44,740
Toluene (ℓ)	C ₇ H ₈	92.141	0.867	412	1.71	42,400	40,500
Heptane (ℓ)	C_7H_{16}	100.204	0.684	365	2.24	48,100	44,600
Octane (ℓ)	C_8H_{18}	114.231	0.703	363	2.23	47,890	44,430
Decane (ℓ)	$C_{10}H_{22}$	142.285	0.730	361	2.21	47,640	44,240
Gasoline (ℓ)	$C_n H_{1.87n}$	100–110	0.72–0.78	350	2.4	47,300	44,000
Light diesel (ℓ)	C _n H _{1.8n}	170	0.78–0.84	270	2.2	46,100	43,200
Heavy diesel (ℓ)	$C_nH_{1.7n}$	200	0.82–0.88	230	1.9	45,500	42,800
Natural gas (g)	$C_n H_{3.8n} N_{0.1n}$	18	_		2	50,000	45,000

 $^{^1\}text{At}$ 1 atm and 20°C. ^2At 25°C for liquid fuels, and 1 atm and normal boiling temperature for gaseous fuels. ^3At 25°C. Multiply by molar mass to obtain heating values in kJ/kmol.

TABLE A-28

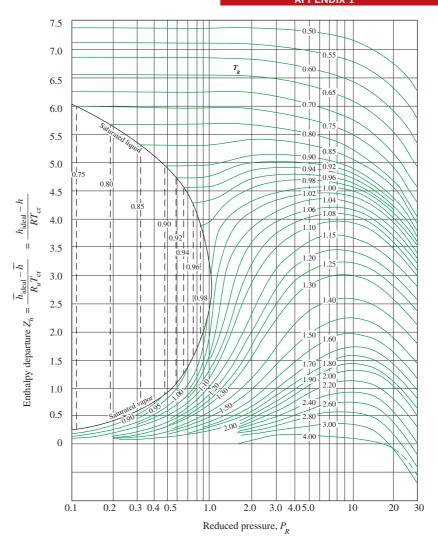
Natural logarithms of the equilibrium constant K_p

The equilibrium constant K_p for the reaction $\nu_A A + \nu_B B \Longrightarrow \nu_C C + \nu_D D$ is defined as $K_p \equiv \frac{P_C^{\nu_C} P_D^{\nu_D}}{P_A^{\nu_A} P_B^{\nu_B}}$

Temp	Temp.,											
Κ .	$H_2 \rightleftharpoons 2H$	$O_2 \rightleftharpoons 20$	$N_2 \rightleftharpoons 2N$	$H_2O \rightleftharpoons H_2 + \frac{1}{2}O_2$	$H_2O \rightleftharpoons {}^1/_2H_2 + OH$	$CO_2 \rightleftharpoons CO + \frac{1}{2}O_2$	$^{1}/_{2}N_{2} + ^{1}/_{2}O_{2} \rightleftharpoons NO$					
298	-164.005	-186.975	-367.480	-92.208	-106.208	-103.762	-35.052					
500	-92.827	-105.630	-213.372	-52.691	-60.281	-57.616	-20.295					
1000	-39.803	-45.150	-99.127	-23.163	-26.034	-23.529	-9.388					
1200	-30.874	-35.005	-80.011	-18.182	-20.283	-17.871	-7.569					
1400	-24.463	-27.742	-66.329	-14.609	-16.099	-13.842	-6.270					
1600	-19.637	-22.285	-56.055	-11.921	-13.066	-10.830	-5.294					
1800	-15.866	-18.030	-48.051	-9.826	-10.657	-8.497	-4.536					
2000	-12.840	-14.622	-41.645	-8.145	-8.728	-6.635	-3.931					
2200	-10.353	-11.827	-36.391	-6.768	-7.148	-5.120	-3.433					
2400	-8.276	-9.497	-32.011	-5.619	-5.832	-3.860	-3.019					
2600	-6.517	-7.521	-28.304	-4.648	-4.719	-2.801	-2.671					
2800	-5.002	-5.826	-25.117	-3.812	-3.763	-1.894	-2.372					
3000	-3.685	-4.357	-22.359	-3.086	-2.937	-1.111	-2.114					
3200	-2.534	-3.072	-19.937	-2.451	-2.212	-0.429	-1.888					
3400	-1.516	-1.935	-17.800	-1.891	-1.576	0.169	-1.690					
3600	-0.609	-0.926	-15.898	-1.392	-1.088	0.701	-1.513					
3800	0.202	-0.019	-14.199	-0.945	-0.501	1.176	-1.356					
4000	0.934	0.796	-12.660	-0.542	-0.044	1.599	-1.216					
4500	2.486	2.513	-9.414	0.312	0.920	2.490	-0.921					
5000	3.725	3.895	-6.807	0.996	1.689	3.197	-0.686					
5500	4.743	5.023	-4.666	1.560	2.318	3.771	-0.497					
6000	5.590	5.963	-2.865	2.032	2.843	4.245	-0.341					

Source of Data: Gordon J. Van Wylen and Richard E. Sonntag, Fundamentals of Classical Thermodynamics, English/SI Version, 3rd ed. (New York: John Wiley & Sons, 1986), p. 723, table A.14. Based on thermodynamic data given in JANAF, Thermochemical Tables (Midland, MI: Thermal Research Laboratory, The Dow Chemical Company, 1971).

941 APPENDIX



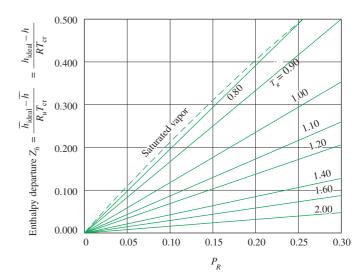


FIGURE A-29 Generalized enthalpy departure chart.

Source of Data: Redrawn from Gordon van Wylen and Richard Sontag, Fundamentals of Classical Thermodynamics, (SI version), 2d ed., Wiley, New York, 1976.

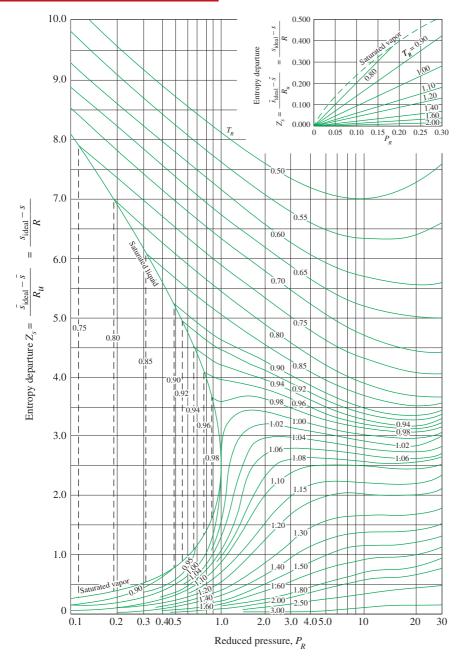


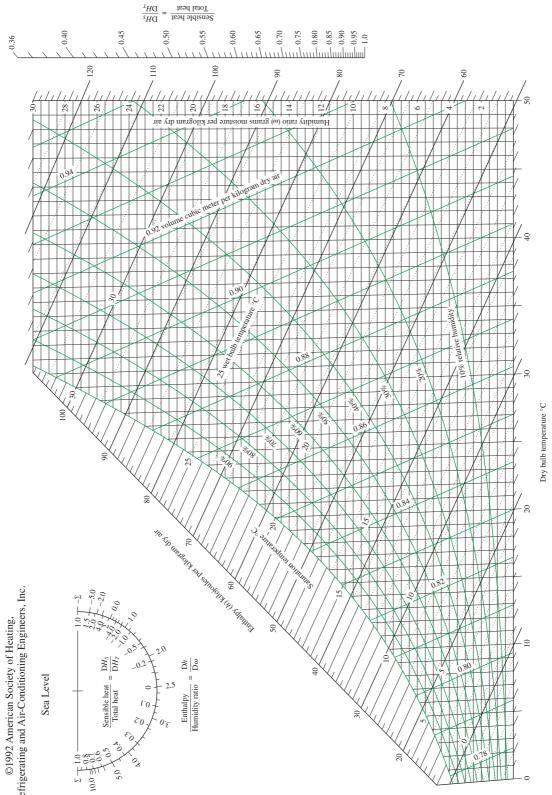
FIGURE A-30

Generalized entropy departure chart.

Source of Data: Redrawn from Gordon van Wylen and Richard Sontag, Fundamentals of Classical Thermodynamics, (SI version), 2d ed., Wiley, New York, 1976.

ASHRAE Psychrometric Chart No. 1 Normal Temperature Barometric Pressure: 101.325 kPa

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Prepared by Center for Applied Thermodynamic Studies, University of Idaho.

FIGURE A-31

Psychrometric chart at 1 atm total pressure.

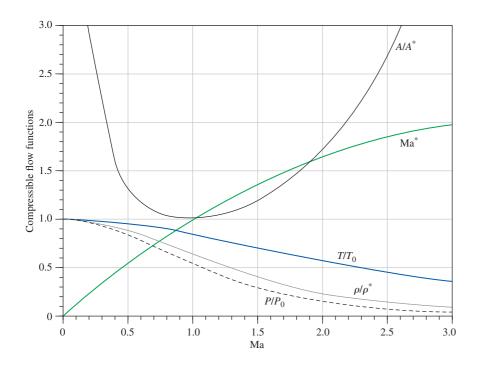
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TABLE A-32

One-dimensional isentropic compressible-flow functions for an ideal gas with k = 1.4

Ма	Ma*	A/A*	P/P_0	$ ho/ ho_0$	<i>T/T</i> ₀
0	0	∞	1.0000	1.0000	1.0000
0.1	0.1094	5.8218	0.9930	0.9950	0.9980
0.2	0.2182	2.9635	0.9725	0.9803	0.9921
0.3	0.3257	2.0351	0.9395	0.9564	0.9823
0.4	0.4313	1.5901	0.8956	0.9243	0.9690
0.5	0.5345	1.3398	0.8430	0.8852	0.9524
0.6	0.6348	1.1882	0.7840	0.8405	0.9328
0.7	0.7318	1.0944	0.7209	0.7916	0.9107
0.8	0.8251	1.0382	0.6560	0.7400	0.8865
0.9	0.9146	1.0089	0.5913	0.6870	0.8606
1.0	1.0000	1.0000	0.5283	0.6339	0.8333
1.2	1.1583	1.0304	0.4124	0.5311	0.7764
1.4	1.2999	1.1149	0.3142	0.4374	0.7184
1.6	1.4254	1.2502	0.2353	0.3557	0.6614
1.8	1.5360	1.4390	0.1740	0.2868	0.6068
2.0	1.6330	1.6875	0.1278	0.2300	0.5556
2.2	1.7179	2.0050	0.0935	0.1841	0.5081
2.4	1.7922	2.4031	0.0684	0.1472	0.4647
2.6	1.8571	2.8960	0.0501	0.1179	0.4252
2.8	1.9140	3.5001	0.0368	0.0946	0.3894
3.0	1.9640	4.2346	0.0272	0.0760	0.3571
5.0	2.2361	25.000	0.0019	0.0113	0.1667
∞	2.2495	∞	0	0	0

$$\begin{split} \mathbf{Ma}^* &= \mathbf{Ma} \sqrt{\frac{k+1}{2+(k-1)\mathbf{Ma}^2}} \\ \frac{A}{A^*} &= \frac{1}{\mathbf{Ma}} \left(\frac{2}{k+1}\right) \left(1 + \frac{k-1}{2} \mathbf{Ma}^2\right) \right]^{0.5(k+1)/(k-1)} \\ \frac{P}{P_0} &= \left(1 + \frac{k-1}{2} \mathbf{Ma}^2\right)^{-k/(k-1)} \\ \frac{\rho}{\rho_0} &= \left(1 + \frac{k-1}{2} \mathbf{Ma}^2\right)^{-1/(k-1)} \\ \frac{T}{T_0} &= \left(1 + \frac{k-1}{2} \mathbf{Ma}^2\right)^{-1} \end{split}$$



$$\begin{split} T_{01} &= T_{02} \\ \mathrm{Ma}_2 &= \sqrt{\frac{(k-1)\mathrm{Ma}_1^2 + 2}{2k\mathrm{Ma}_1^2 - k + 1}} \\ \frac{P_2}{P_1} &= \frac{1 + k\mathrm{Ma}_1^2}{1 + k\mathrm{Ma}_2^2} = \frac{2k\mathrm{Ma}_1^2 - k + 1}{k + 1} \\ \frac{\rho_2}{\rho_1} &= \frac{P_2/P_1}{T_2/T_1} = \frac{(k+1)\mathrm{Ma}_1^2}{2 + (k-1)\mathrm{Ma}_1^2} = \frac{V_1}{V_2} \\ \frac{T_2}{T_1} &= \frac{2 + \mathrm{Ma}_1^2(k-1)}{2 + \mathrm{Ma}_2^2(k-1)} \\ \frac{P_{02}}{P_{01}} &= \frac{\mathrm{Ma}_1}{\mathrm{Ma}_2} \bigg[\frac{1 + \mathrm{Ma}_2^2(k-1)/2}{1 + \mathrm{Ma}_1^2(k-1)/2} \bigg]^{(k+1)/[2(k-1)]} \\ \frac{P_{02}}{P_1} &= \frac{(1 + k\mathrm{Ma}_1^2)[1 + \mathrm{Ma}_2^2(k-1)/2]^{k/(k-1)}}{1 + k\mathrm{Ma}_2^2} \end{split}$$

TABLE A-33

One-dimensional		normal-shock function		for an ideal	gas with k =	= 1.4
Ma_1	Ma ₂	P_2/P_1	ρ_2/ρ_1	T_2/T_1	P_{02}/P_{01}	P_{02}/P_1
1.0	1.0000	1.0000	1.0000	1.0000	1.0000	1.8929
1.1	0.9118	1.2450	1.1691	1.0649	0.9989	2.1328
1.2	0.8422	1.5133	1.3416	1.1280	0.9928	2.4075
1.3	0.7860	1.8050	1.5157	1.1909	0.9794	2.7136
1.4	0.7397	2.1200	1.6897	1.2547	0.9582	3.0492
1.5	0.7011	2.4583	1.8621	1.3202	0.9298	3.4133
1.6	0.6684	2.8200	2.0317	1.3880	0.8952	3.8050
1.7	0.6405	3.2050	2.1977	1.4583	0.8557	4.2238
1.8	0.6165	3.6133	2.3592	1.5316	0.8127	4.6695
1.9	0.5956	4.0450	2.5157	1.6079	0.7674	5.1418
2.0	0.5774	4.5000	2.6667	1.6875	0.7209	5.6404
2.1	0.5613	4.9783	2.8119	1.7705	0.6742	6.1654
2.2	0.5471	5.4800	2.9512	1.8569	0.6281	6.7165
2.3	0.5344	6.0050	3.0845	1.9468	0.5833	7.2937
2.4	0.5231	6.5533	3.2119	2.0403	0.5401	7.8969
2.5	0.5130	7.1250	3.3333	2.1375	0.4990	8.5261
2.6	0.5039	7.7200	3.4490	2.2383	0.4601	9.1813
2.7	0.4956	8.3383	3.5590	2.3429	0.4236	9.8624
2.8	0.4882	8.9800	3.6636	2.4512	0.3895	10.5694
2.9	0.4814	9.6450	3.7629	2.5632	0.3577	11.3022
3.0	0.4752	10.3333	3.8571	2.6790	0.3283	12.0610
4.0	0.4350	18.5000	4.5714	4.0469	0.1388	21.0681
5.0	0.4152	29.000	5.0000	5.8000	0.0617	32.6335
∞	0.3780	∞	6.0000	∞	0	∞

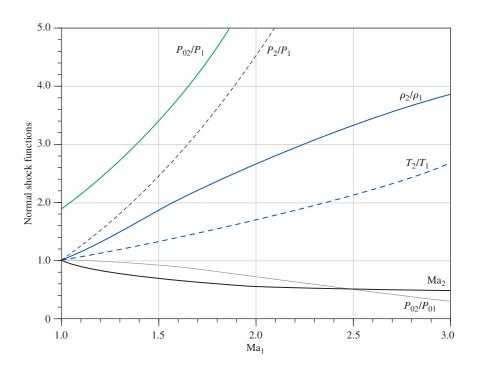


TABLE A-34

Mayleigi	I HOW TUITCHOIL	s ioi aii ideal gi	35 WILII N — 1.4		
Ма	T_{0}/T_{0}^{*}	P_0/P_0^*	T/ T*	P/P*	
0.0	0.0000	1.0670	0.0000	0.4000	

Rayleigh flow functions for an ideal gas with k = 1.4

Ма	T_0/T_0^*	P_0/P_0^*	T/T*	PIP*	<i>V/V</i> *
0.0	0.0000	1.2679	0.0000	2.4000	0.0000
0.1	0.0468	1.2591	0.0560	2.3669	0.0237
0.2	0.1736	1.2346	0.2066	2.2727	0.0909
0.3	0.3469	1.1985	0.4089	2.1314	0.1918
0.4	0.5290	1.1566	0.6151	1.9608	0.3137
0.5	0.6914	1.1141	0.7901	1.7778	0.4444
0.6	0.8189	1.0753	0.9167	1.5957	0.5745
0.7	0.9085	1.0431	0.9929	1.4235	0.6975
8.0	0.9639	1.0193	1.0255	1.2658	0.8101
0.9	0.9921	1.0049	1.0245	1.1246	0.9110
1.0	1.0000	1.0000	1.0000	1.0000	1.0000
1.2	0.9787	1.0194	0.9118	0.7958	1.1459
1.4	0.9343	1.0777	0.8054	0.6410	1.2564
1.6	0.8842	1.1756	0.7017	0.5236	1.3403
1.8	0.8363	1.3159	0.6089	0.4335	1.4046
2.0	0.7934	1.5031	0.5289	0.3636	1.4545
2.2	0.7561	1.7434	0.4611	0.3086	1.4938
2.4	0.7242	2.0451	0.4038	0.2648	1.5252
2.6	0.6970	2.4177	0.3556	0.2294	1.5505
2.8	0.6738	2.8731	0.3149	0.2004	1.5711
3.0	0.6540	3.4245	0.2803	0.1765	1.5882

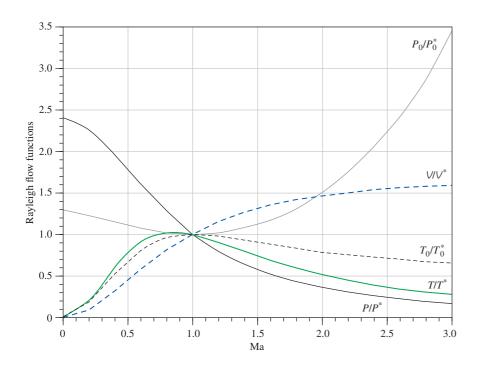
$$\frac{T_0}{T_0^*} = \frac{(k+1)\text{Ma}^2[2+(k-1)\text{Ma}^2]}{(1+k\text{Ma}^2)^2}$$

$$= \frac{P_0}{P_0^*} = \frac{k+1}{1+k\text{Ma}^2} \left(\frac{2+(k-1)\text{Ma}^2}{k+1}\right)^{k/(k-1)}$$

$$\frac{T}{T^*} = \left(\frac{\text{Ma}(1+k)}{1+k\text{Ma}^2}\right)^2$$

$$\frac{P}{P^*} = \frac{1+k}{1+k\text{Ma}^2}$$

$$\frac{V}{V^*} = \frac{\rho^*}{\rho} = \frac{(1+k)\text{Ma}^2}{1+k\text{Ma}^2}$$



2

PROPERTY TABLES AND CHARTS (ENGLISH UNITS)

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TABLE A-1E

Molar mass, gas constant, and critical-point properties

		Molar	Gas con	stant, R	Critical-p	oint propertie	s
Substance	Formula	mass, <i>M</i> Ibm/Ibmol	Btu/ Ibm·R*	psia∙ft³/ Ibm∙R*	Temperature, R	Pressure, psia	Volume, ft³/lbmol
Air	_	28.97	0.06855	0.3704	238.5	547	1.41
Ammonia	NH_3	17.03	0.1166	0.6301	729.8	1636	1.16
Argon	Ar	39.948	0.04971	0.2686	272	705	1.20
Benzene	C_6H_6	78.115	0.02542	0.1374	1012	714	4.17
Bromine	Br ₂	159.808	0.01243	0.06714	1052	1500	2.17
n-Butane	C_4H_{10}	58.124	0.03417	0.1846	765.2	551	4.08
Carbon dioxide	CO_2	44.01	0.04513	0.2438	547.5	1071	1.51
Carbon monoxide	CO	28.011	0.07090	0.3831	240	507	1.49
Carbon tetrachloride	CCI ₄	153.82	0.01291	0.06976	1001.5	661	4.42
Chlorine	Cl ₂	70.906	0.02801	0.1517	751	1120	1.99
Chloroform	CHCI ₃	119.38	0.01664	0.08988	965.8	794	3.85
Dichlorodifluoromethane (R-12)	CCI ₂ F ₂	120.91	0.01643	0.08874	692.4	582	3.49
Dichlorofluoromethane (R-21)	CHCl ₂ F	102.92	0.01930	0.1043	813.0	749	3.16
Ethane	C_2H_6	30.020	0.06616	0.3574	549.8	708	2.37
Ethyl alcohol	C ₂ H ₅ OH	46.07	0.04311	0.2329	929.0	926	2.68
Ethylene	C_2H_4	28.054	0.07079	0.3825	508.3	742	1.99
Helium	He	4.003	0.4961	2.6809	9.5	33.2	0.926
n-Hexane	C_6H_{14}	86.178	0.02305	0.1245	914.2	439	5.89
Hydrogen (normal)	H_2	2.016	0.9851	5.3224	59.9	188.1	1.04
Krypton	Kr	83.80	0.02370	0.1280	376.9	798	1.48
Methane	CH ₄	16.043	0.1238	0.6688	343.9	673	1.59
Methyl alcohol	CH ₃ OH	32.042	0.06198	0.3349	923.7	1154	1.89
Methyl chloride	CH ₃ CI	50.488	0.03934	0.2125	749.3	968	2.29
Neon	Ne	20.183	0.09840	0.5316	80.1	395	0.668
Nitrogen	N_2	28.013	0.07090	0.3830	227.1	492	1.44
Nitrous oxide	N_2O	44.013	0.04512	0.2438	557.4	1054	1.54
Oxygen	02	31.999	0.06206	0.3353	278.6	736	1.25
Propane	C_3H_8	44.097	0.04504	0.2433	665.9	617	3.20
Propylene	C_3H_6	42.081	0.04719	0.2550	656.9	670	2.90
Sulfur dioxide	SO ₂	64.063	0.03100	1.1675	775.2	1143	1.95
Tetrafluoroethane (R-134a)	CF ₃ CH ₂ F	102.03	0.01946	0.1052	673.6	588.7	3.19
Trichlorofluoromethane (R-11)	CCĬ ₃ F	137.37	0.01446	0.07811	848.1	635	3.97
Water	H_2O	18.015	0.1102	0.5956	1164.8	3200	0.90
Xenon	Xe	131.30	0.01513	0.08172	521.55	852	1.90

^{*}Calculated from $R=R_u/M$, where $R_u=1.98588~{\rm Btu/lbmol\cdot R}=10.7316~{\rm psia\cdot ft^3/lbmol\cdot R}$ and M is the molar mass.

Source of Data: K. A. Kobe and R. E. Lynn, Jr., Chemical Review 52 (1953), pp. 117–236, and ASHRAE, Handbook of Fundamentals (Atlanta, GA: American Society of Heating, Refrigerating, and Air-Conditioning Engineers, Inc., 1993), pp. 16.4 and 36.1.

TABLE A-2E

Ideal-gas specific heats of various common gases (a) At 80°F

Gas	Formula	Gas constant, <i>R</i> Btu/lbm·R	<i>c_p</i> Btu/lbm∙R	<i>c_v</i> Btu/lbm∙R	k
Air	_	0.06855	0.240	0.171	1.400
Argon	Ar	0.04971	0.1253	0.0756	1.667
Butane	C_4H_{10}	0.03424	0.415	0.381	1.09
Carbon dioxide	CO_2	0.04513	0.203	0.158	1.285
Carbon monoxide	CO	0.07090	0.249	0.178	1.399
Ethane	C_2H_6	0.06616	0.427	0.361	1.183
Ethylene	C_2H_4	0.07079	0.411	0.340	1.208
Helium	He	0.4961	1.25	0.753	1.667
Hydrogen	H_2	0.9851	3.43	2.44	1.404
Methane	CH₄	0.1238	0.532	0.403	1.32
Neon	Ne	0.09840	0.246	0.1477	1.667
Nitrogen	N_2	0.07090	0.248	0.177	1.400
Octane	C ₈ H ₁₈	0.01742	0.409	0.392	1.044
Oxygen	02	0.06206	0.219	0.157	1.395
Propane	C ₃ H ₈	0.04504	0.407	0.362	1.124
Steam	$H_2^{\circ}O^{\circ}$	0.1102	0.445	0.335	1.329

Source of Data: Gordon J. Van Wylen and Richard E. Sonntag, Fundamentals of Classical Thermodynamics, English/SI Version, 3rd ed. (New York: John Wiley & Sons, 1986), p. 687, Table A–8E.

TABLE A-2E

Ideal-gas specific heats of various common gases (*Continued*) (*b*) At various temperatures

Temp.,	c_p Btu/Ibm \cdot R	<i>c</i> _v Btu/lbm⋅R	k	<i>c_p</i> Btu/lbm⋅R	<i>c</i> _v Btu/lbm⋅R	k	c_p Btu/lbm \cdot R	<i>c</i> _v Btu/Ibm⋅R	k		
		Air		Cari	oon dioxide, C		Car	Carbon monoxide, CO			
40	0.240	0.171	1.401	0.195	0.150	1.300	0.248	0.177	1.400		
100	0.240	0.172	1.400	0.205	0.160	1.283	0.249	0.178	1.399		
200	0.241	0.173	1.397	0.217	0.172	1.262	0.249	0.179	1.397		
300	0.243	0.174	1.394	0.229	0.184	1.246	0.251	0.180	1.394		
400	0.245	0.176	1.389	0.239	0.193	1.233	0.253	0.182	1.389		
500	0.248	0.179	1.383	0.247	0.202	1.223	0.256	0.185	1.384		
600	0.250	0.182	1.377	0.255	0.210	1.215	0.259	0.188	1.377		
700	0.254	0.185	1.371	0.262	0.217	1.208	0.262	0.191	1.371		
800	0.257	0.188	1.365	0.269	0.224	1.202	0.266	0.195	1.364		
900	0.259	0.191	1.358	0.275	0.230	1.197	0.269	0.198	1.357		
1000	0.263	0.195	1.353	0.280	0.235	1.192	0.273	0.202	1.351		
1500	0.276	0.208	1.330	0.298	0.253	1.178	0.287	0.216	1.328		
2000	0.286	0.217	1.312	0.312	0.267	1.169	0.297	0.226	1.314		
		Hydrogen, H_2			Nitrogen, N ₂			Oxygen, O ₂			
40	3.397	2.412	1.409	0.248	0.177	1.400	0.219	0.156	1.397		
100	3.426	2.441	1.404	0.248	0.178	1.399	0.220	0.158	1.394		
200	3.451	2.466	1.399	0.249	0.178	1.398	0.223	0.161	1.387		
300	3.461	2.476	1.398	0.250	0.179	1.396	0.226	0.164	1.378		
400	3.466	2.480	1.397	0.251	0.180	1.393	0.230	0.168	1.368		
500	3.469	2.484	1.397	0.254	0.183	1.388	0.235	0.173	1.360		
600	3.473	2.488	1.396	0.256	0.185	1.383	0.239	0.177	1.352		
700	3.477	2.492	1.395	0.260	0.189	1.377	0.242	0.181	1.344		
800	3.494	2.509	1.393	0.262	0.191	1.371	0.246	0.184	1.337		
900	3.502	2.519	1.392	0.265	0.194	1.364	0.249	0.187	1.331		
1000	3.513	2.528	1.390	0.269	0.198	1.359	0.252	0.190	1.326		
1500	3.618	2.633	1.374	0.283	0.212	1.334	0.263	0.201	1.309		
2000	3.758	2.773	1.355	0.293	0.222	1.319	0.270	0.208	1.298		

Note: The unit Btu/lbm·R is equivalent to Btu/lbm·°F.

Source of Data: Kenneth Wark, Thermodynamics, 4th ed. (New York: McGraw-Hill, 1983), p. 830, Table A-4. Originally published in Tables of Properties of Gases, NBS Circular 564, 1955.

TABLE A-2E

Ideal-gas specific heats of various common gases (*Concluded*) (c) As a function of temperature

$$\bar{c}_p = a + bT + cT^2 + dT^3$$

(*T* in R, c_p in Btu/Ibmol·R)

						Temperature	% er	ror
Substance	Formula	а	b	С	d	range, R	Max.	Avg.
Nitrogen	N ₂	6.903	-0.02085×10^{-2}	0.05957×10^{-5}	-0.1176×10^{-9}	491-3240	0.59	0.34
Oxygen	02	6.085	0.2017×10^{-2}	-0.05275×10^{-5}	0.05372×10^{-9}	491-3240	1.19	0.28
Air	_	6.713	0.02609×10^{-2}	0.03540×10^{-5}	-0.08052×10^{-9}	491-3240	0.72	0.33
Hydrogen	H_2	6.952	-0.02542×10^{-2}	0.02952×10^{-5}	-0.03565×10^{-9}	491-3240	1.02	0.26
Carbon monoxide	CO	6.726	0.02222×10^{-2}	0.03960×10^{-5}	-0.09100×10^{-9}	491-3240	0.89	0.37
Carbon dioxide	CO ₂	5.316	0.79361×10^{-2}	-0.2581×10^{-5}	0.3059×10^{-9}	491-3240	0.67	0.22
Water vapor	$H_2\bar{O}$	7.700	0.02552×10^{-2}	0.07781×10^{-5}	-0.1472×10^{-9}	491-3240	0.53	0.24
Nitric oxide	NO	7.008	-0.01247×10^{-2}	0.07185×10^{-5}	-0.1715×10^{-9}	491-2700	0.97	0.36
Nitrous oxide	N_2O	5.758	0.7780×10^{-2}	-0.2596×10^{-5}	0.4331×10^{-9}	491-2700	0.59	0.26
Nitrogen dioxide	\overline{NO}_2	5.48	0.7583×10^{-2}	-0.260×10^{-5}	0.322×10^{-9}	491-2700	0.46	0.18
Ammonia	NH_3	6.5846	0.34028×10^{-2}	0.073034×10^{-5}	-0.27402×10^{-9}	491-2700	0.91	0.36
Sulfur	S_2	6.499	0.2943×10^{-2}	-0.1200×10^{-5}	0.1632×10^{-9}	491-3240	0.99	0.38
Sulfur dioxide	SO ₂	6.157	0.7689×10^{-2}	-0.2810×10^{-5}	0.3527×10^{-9}	491-3240	0.45	0.24
Sulfur trioxide	SO ₃	3.918	1.935×10^{-2}	-0.8256×10^{-5}	1.328×10^{-9}	491-2340	0.29	0.13
Acetylene	C_2H_2	5.21	1.2227×10^{-2}	-0.4812×10^{-5}	0.7457×10^{-9}	491-2700	1.46	0.59
Benzene	C_6H_6	-8.650	6.4322×10^{-2}	-2.327×10^{-5}	3.179×10^{-9}	491-2700	0.34	0.20
Methanol	CH ₄ O	4.55	1.214×10^{-2}	-0.0898×10^{-5}	-0.329×10^{-9}	491-1800	0.18	0.08
Ethanol	C ₂ H ₆ O	4.75	2.781×10^{-2}	-0.7651×10^{-5}	0.821×10^{-9}	491-2700	0.40	0.22
Hydrogen chloride	HCI	7.244	-0.1011×10^{-2}	0.09783×10^{-5}	-0.1776×10^{-9}	491-2740	0.22	0.08
Methane	CH ₄	4.750	0.6666×10^{-2}	0.09352×10^{-5}	-0.4510×10^{-9}	491-2740	1.33	0.57
Ethane	C ₂ H ₆	1.648	2.291×10^{-2}	-0.4722×10^{-5}	0.2984×10^{-9}	491-2740	0.83	0.28
Propane	C_3H_8	-0.966	4.044×10^{-2}	-1.159×10^{-5}	1.300×10^{-9}	491-2740	0.40	0.12
<i>n</i> -Butane	C ₄ H ₁₀	0.945	4.929×10^{-2}	-1.352×10^{-5}	1.433×10^{-9}	491-2740	0.54	0.24
<i>i</i> -Butane	C ₄ H ₁₀	-1.890	5.520×10^{-2}	-1.696×10^{-5}	2.044×10^{-9}	491-2740	0.25	0.13
<i>n</i> -Pentane	C ₅ H ₁₂	1.618	6.028×10^{-2}	-1.656×10^{-5}	1.732×10^{-9}	491-2740	0.56	0.21
<i>n</i> -Hexane	C ₆ H ₁₄	1.657	7.328×10^{-2}	-2.112×10^{-5}	2.363×10^{-9}	491-2740	0.72	0.20
Ethylene	C_2H_4	0.944	2.075×10^{-2}	-0.6151×10^{-5}	0.7326×10^{-9}	491-2740	0.54	0.13
Propylene	C ₃ H ₆	0.753	3.162×10^{-2}	-0.8981×10^{-5}	1.008×10^{-9}	491-2740	0.73	0.17

Source of Data: B.G. Kyle, Chemical and Process Thermodynamics, 3rd ed. (Upper Saddle River, NJ: Prentice Hall, 2000).

TABLE A-3E

Properties of common liquids, solids, and foods (a) Liquids

	Boiling	data at 1 atm	Freezing	g data	Liquid pi	roperties	
Substance	Normal boiling point, °F	Latent heat of vaporization, h_{fg} Btu/Ibm	Freezing point, °F	Latent heat of fusion, h_{if} Btu/Ibm	Temperature,	Density, ρ Ibm/ft ³	Specific heat, c_p Btu/lbm·R
Ammonia	-27.9	24.54	-107.9	138.6	-27.9 0 40 80	42.6 41.3 39.5 37.5	1.06 1.083 1.103 1.135
Argon Benzene Brine (20% sodium chloride	-302.6 176.4	69.5 169.4	-308.7 41.9	12.0 54.2	-302.6 68	87.0 54.9	0.272 0.411
by mass) n-Butane Carbon dioxide Ethanol Ethyl alcohol Ethylene glycol Glycerine Helium Hydrogen Isobutane Kerosene Mercury Methane Methanol Nitrogen Octane Oil (light)	219.0 31.1 -109.2* 172.8 173.5 388.6 355.8 -452.1 -423.0 10.9 399-559 674.1 -258.7 148.1 -320.4 256.6 - -297.3		0.7 -217.3 -69.8 -173.6 -248.8 12.6 66.0434.5 -255.5 -12.8 -38.0 296.0 -143.9 -346.0 -71.5		68 31.1 32 77 68 68 68 -452.1 -423.0 10.9 68 77 -258.7 -160 77 -320.4 -260 68 77 -3297.3	71.8 37.5 57.8 48.9 49.3 69.2 78.7 9.13 4.41 37.1 51.2 847 26.4 20.0 49.1 50.5 38.2 43.9 56.8	0.743 0.552 0.583 0.588 0.678 0.678 0.554 5.45 2.39 0.545 0.478 0.033 0.834 1.074 0.609 0.492 0.643 0.502
Oxygen Petroleum Propane	-297.3 -43.7	91.5 99–165 184.0	-361.8 -305.8	34.4	68 -43.7 32 100	71.2 40.0 36.3 33.0 29.4	0.408 0.478 0.538 0.604 0.673
Refrigerant-134a	-15.0	93.3	-141.9	_	-40 -15 32 90	88.5 86.0 80.9 73.6	0.283 0.294 0.318 0.348
Water	212	970.1	32	143.5	32 90 150 212	62.4 62.1 61.2 59.8	1.01 1.00 1.00 1.01

^{*}Sublimation temperature. (At pressures below the triple-point pressure of 75.1 psia, carbon dioxide exists as a solid or gas. Also, the freezing-point temperature of carbon dioxide is the triple-point temperature of -69.8° F.)

TABLE A-3E

Properties of common liquids, solids, and foods (*Concluded*) (b) Solids (values are for room temperature unless indicated otherwise)

Substance	Density, $ ho$ lbm/ft 3	Specific heat, c_p Btu/Ibm·R	Substance	Density, ρ lbm/ft ³	Specific heat, c_p Btu/lbm·R
Metals			Nonmetals		
Aluminum			Asphalt	132	0.220
-100°F		0.192	Brick, common	120	0.189
32°F		0.212	Brick, fireclay (500°C)	144	0.229
100°F	170	0.218	Concrete	144	0.156
200°F		0.224	Clay	62.4	0.220
300°F		0.229	Diamond	151	0.147
400°F		0.235	Glass, window	169	0.191
500°F		0.240	Glass, pyrex	139	0.200
Bronze (76% Cu, 2% Zn,	517	0.0955	Graphite	156	0.170
2% AI)			Granite	169	0.243
Brass, yellow (65% Cu, 35% Zn)	519	0.0955	Gypsum or plaster board Ice	50	0.260
Copper			−50°F		0.424
−60°F		0.0862	0°F		0.471
0°F		0.0893	20°F		0.491
100°F	555	0.0925	32° F	57.5	0.502
200°F		0.0938	Limestone	103	0.217
390°F		0.0963	Marble 162	0.210	
Iron	490	0.107	Plywood (Douglas fir)	34.0	
Lead	705	0.030	Rubber (hard)	68.7	
Magnesium	108	0.239	Rubber (soft)	71.8	
Nickel	555	0.105	Sand	94.9	
Silver	655	0.056	Stone	93.6	
Steel, mild	489	0.119	Woods, hard (maple, oak, etc.)		
Tungsten	1211	0.031	Woods, soft (fir, pine, etc.)	32.0	

(c) Foods

	Water	Freezing	Specific Btu/Ib	and the second second	Latent heat of		Water content,	Freezing	Specific Btu/It		Latent heat of
Food	content, % (mass)	point, °F	Above freezing	Below freezing	fusion, Btu/Ibm	Food	% (mass)	point, °F	Above freezing	Below freezing	fusion, Btu/Ibm
Apples	84	30	0.873	0.453	121	Lettuce	95	32	0.961	0.487	136
Bananas	75	31	0.801	0.426	108	Milk, whole	88	31	0.905	0.465	126
Beef round	67	_	0.737	0.402	96	Oranges	87	31	0.897	0.462	125
Broccoli	90	31	0.921	0.471	129	Potatoes	78	31	0.825	0.435	112
Butter	16	_	_	0.249	23	Salmon fish	64	28	0.713	0.393	92
Cheese, Swiss	39	14	0.513	0.318	56	Shrimp	83	28	0.865	0.450	119
Cherries	80	29	0.841	0.441	115	Spinach	93	31	0.945	0.481	134
Chicken	74	27	0.793	0.423	106	Strawberries	90	31	0.921	0.471	129
Corn, sweet	74	31	0.793	0.423	106	Tomatoes, ripe	94	31	0.953	0.484	135
Eggs, whole	74	31	0.793	0.423	106	Turkey	64	_	0.713	0.393	92
Ice cream	63	22	0.705	0.390	90	Watermelon	93	31	0.945	0.481	134

Source of Data: Values are obtained from various handbooks and other sources or are calculated. Water content and freezing-point data of foods are from ASHRAE, Handbook of Fundamentals, I-P version (Atlanta, GA: American Society of Heating, Refrigerating, and Air-Conditioning Engineers, Inc., 1993), Chap. 30, Table 1. Freezing point is the temperature at which freezing starts for fruits and vegetables, and the average freezing temperature for other foods.

TABLE A-4E

Saturated water—Temperature table

			<i>c volume,</i> /lbm		<i>nal energy,</i> tu/lbm			<i>Inthalpy,</i> Btu/lbm			E <i>ntropy,</i> tu/Ibm·R	
Temp <i>T</i> °F	Sat. ., press., $P_{\rm sat}$ psia	Sat. liquid, v _f	Sat. vapor, v _g	Sat. liquid, u _f	Evap., u _{fg}	Sat. vapor, u _g	Sat. liquid, h _f	Evap., <i>h_{fg}</i>	Sat. vapor, h_g	Sat. liquid, s _f	Evap., S _{fg}	Sat. vapor, s_g
32.0 35 40 45 50	18 0.08871 0.09998 0.12173 0.14756 0.17812	0.01602 0.01602 0.01602 0.01602 0.01602	3299.9 2945.7 2443.6 2035.8 1703.1	3.004	1021.0 1019.0 1015.6 1012.2 1008.9	1021.0 1022.0 1023.7 1025.3 1026.9	0.000 3.004 8.032 13.05 18.07	1075.2 1073.5 1070.7 1067.8 1065.0	1075.2 1076.5 1078.7 1080.9 1083.1	0.00000 0.00609 0.01620 0.02620 0.03609	2.18672 2.17011 2.14271 2.11587 2.08956	2.1762 2.1589 2.1421
55	0.21413	0.01603	1430.4	23.07	1005.5	1028.6	23.07	1062.2	1085.3	0.04586	2.06377	2.0940
60	0.25638	0.01604	1206.1	28.08	1002.1	1030.2	28.08	1059.4	1087.4	0.05554	2.03847	
65	0.30578	0.01604	1020.8	33.08	998.76	1031.8	33.08	1056.5	1089.6	0.06511	2.01366	
70	0.36334	0.01605	867.18	38.08	995.39	1033.5	38.08	1053.7	1091.8	0.07459	1.98931	
75	0.43016	0.01606	739.27	43.07	992.02	1035.1	43.07	1050.9	1093.9	0.08398	1.96541	
80	0.50745	0.01607	632.41	48.06	988.65	1036.7	48.07	1048.0	1096.1	0.09328	1.94196	
85	0.59659	0.01609	542.80	53.06	985.28	1038.3	53.06	1045.2	1098.3	0.10248	1.91892	
90	0.69904	0.01610	467.40	58.05	981.90	1040.0	58.05	1042.4	1100.4	0.11161	1.89630	
95	0.81643	0.01612	403.74	63.04	978.52	1041.6	63.04	1039.5	1102.6	0.12065	1.87408	
100	0.95052	0.01613	349.83	68.03	975.14	1043.2	68.03	1036.7	1104.7	0.12961	1.85225	
110 120 130 140 150	1.2767 1.6951 2.2260 2.8931 3.7234	0.01617 0.01620 0.01625 0.01629 0.01634	264.96 202.94 157.09 122.81 96.929	78.01 88.00 97.99 107.98 117.98	968.36 961.56 954.73 947.87 940.98	1046.4 1049.6 1052.7 1055.9 1059.0	78.02 88.00 97.99 107.99 117.99	1031.0 1025.2 1019.4 1013.6 1007.8	1109.0 1113.2 1117.4 1121.6 1125.7	0.14728 0.16466 0.18174 0.19855 0.21508	1.76856 1.72877 1.69024	1.9570 1.9332 1.9105 1.8888 1.8680
160	4.7474	0.01639	77.185	127.98	934.05	1062.0	128.00	1001.8	1129.8	0.23136	1.61670	1.8481
170	5.9999	0.01645	61.982	138.00	927.08	1065.1	138.02	995.88	1133.9	0.24739	1.58155	1.8289
180	7.5197	0.01651	50.172	148.02	920.06	1068.1	148.04	989.85	1137.9	0.26318	1.54741	1.8106
190	9.3497	0.01657	40.920	158.05	912.99	1071.0	158.08	983.76	1141.8	0.27874	1.51421	1.7930
200	11.538	0.01663	33.613	168.10	905.87	1074.0	168.13	977.60	1145.7	0.29409	1.48191	1.7760
210	14.136	0.01670	27.798	178.15	898.68	1076.8	178.20	971.35	1149.5	0.30922	1.45046	1.7597
212	14.709	0.01671	26.782	180.16	897.24	1077.4	180.21	970.09	1150.3	0.31222	1.44427	1.7565
220	17.201	0.01677	23.136	188.22	891.43	1079.6	188.28	965.02	1153.3	0.32414	1.41980	1.7439
230	20.795	0.01684	19.374	198.31	884.10	1082.4	198.37	958.59	1157.0	0.33887	1.38989	1.7288
240	24.985	0.01692	16.316	208.41	876.70	1085.1	208.49	952.06	1160.5	0.35342	1.36069	1.7141
250	29.844	0.01700	13.816	218.54	869.21	1087.7	218.63	945.41	1164.0	0.36779	1.33216	1.6999
260	35.447	0.01708	11.760	228.68	861.62	1090.3	228.79	938.65	1167.4	0.38198	1.30425	1.6862
270	41.877	0.01717	10.059	238.85	853.94	1092.8	238.98	931.76	1170.7	0.39601	1.27694	1.6730
280	49.222	0.01726	8.6439	249.04	846.16	1095.2	249.20	924.74	1173.9	0.40989	1.25018	1.6601
290	57.573	0.01735	7.4607	259.26	838.27	1097.5	259.45	917.57	1177.0	0.42361	1.22393	1.6475
300	67.028	0.01745	6.4663	269.51	830.25	1099.8	269.73	910.24	1180.0	0.43720	1.19818	1.6354
310	77.691	0.01755	5.6266	279.79	822.11	1101.9	280.05	902.75	1182.8	0.45065	1.17289	1.6235
320	89.667	0.01765	4.9144	290.11	813.84	1104.0	290.40	895.09	1185.5	0.46396	1.14802	1.6120
330	103.07	0.01776	4.3076	300.46	805.43	1105.9	300.80	887.25	1188.1	0.47716	1.12355	1.6007
340	118.02	0.01787	3.7885	310.85	796.87	1107.7	311.24	879.22	1190.5	0.49024	1.09945	1.5897
350	134.63	0.01799	3.3425	321.29	788.16	1109.4	321.73	870.98	1192.7	0.50321	1.07570	1.5478
360	153.03	0.01811	2.9580	331.76	779.28	1111.0	332.28	862.53	1194.8	0.51607	1.05227	
370	173.36	0.01823	2.6252	342.29	770.23	1112.5	342.88	853.86	1196.7	0.52884	1.02914	
380	195.74	0.01836	2.3361	352.87	761.00	1113.9	353.53	844.96	1198.5	0.54152	1.00628	
390	220.33	0.01850	2.0842	363.50	751.58	1115.1	364.25	835.81	1200.1	0.55411	0.98366	

TABLE A-4E

Saturated water—Temperature table (Concluded)

			<i>volume,</i> Ibm					<i>Enthalpy,</i> Btu/lbm			<i>Entropy,</i> Btu/lbm∙R	
Temp., <i>T°</i> F	Sat. press., P _{sat} psia	Sat. liquid, v _f	Sat. vapor, v _g	Sat. liquid, u _f	Evap., u_{fg}	Sat. vapor, u_g	Sat. liquid, h _f	Evap., h _{fg}	Sat. vapor, h_g	Sat. liquid, s _f	Evap., s _{fg}	Sat. vapor, s_g
400	247.26	0.01864	1.8639	374.19	741.97	1116.2	375.04	826.39	1201.4	0.56663	0.96127	1.5279
410	276.69	0.01878	1.6706	384.94	732.14	1117.1	385.90	816.71	1202.6	0.57907	0.93908	1.5182
420	308.76	0.01894	1.5006	395.76	722.08	1117.8	396.84	806.74	1203.6	0.59145	0.91707	1.5085
430	343.64	0.01910	1.3505	406.65	711.80	1118.4	407.86	796.46	1204.3	0.60377	0.89522	1.4990
440	381.49	0.01926	1.2178	417.61	701.26	1118.9	418.97	785.87	1204.8	0.61603	0.87349	1.4895
450	422.47	0.01944	1.0999	428.66	690.47	1119.1	430.18	774.94	1205.1	0.62826	0.85187	1.4801
460	466.75	0.01962	0.99510	439.79	679.39	1119.2	441.48	763.65	1205.1	0.64044	0.83033	1.4708
470	514.52	0.01981	0.90158	451.01	668.02	1119.0	452.90	751.98	1204.9	0.65260	0.80885	1.4615
480	565.96	0.02001	0.81794	462.34	656.34	1118.7	464.43	739.91	1204.3	0.66474	0.78739	1.4521
490	621.24	0.02022	0.74296	473.77	644.32	1118.1	476.09	727.40	1203.5	0.67686	0.76594	1.4428
500	680.56	0.02044	0.67558	485.32	631.94	1117.3	487.89	714.44	1202.3	0.68899	0.74445	1.4334
510	744.11	0.02067	0.61489	496.99	619.17	1116.2	499.84	700.99	1200.8	0.70112	0.72290	1.4240
520	812.11	0.02092	0.56009	508.80	605.99	1114.8	511.94	687.01	1199.0	0.71327	0.70126	1.4145
530	884.74	0.02118	0.51051	520.76	592.35	1113.1	524.23	672.47	1196.7	0.72546	0.67947	1.4049
540	962.24	0.02146	0.46553	532.88	578.23	1111.1	536.70	657.31	1194.0	0.73770	0.65751	1.3952
550	1044.8	0.02176	0.42465	545.18	563.58	1108.8	549.39	641.47	1190.9	0.75000	0.63532	1.3853
560	1132.7	0.02207	0.38740	557.68	548.33	1106.0	562.31	624.91	1187.2	0.76238	0.61284	1.3752
570	1226.2	0.02242	0.35339	570.40	532.45	1102.8	575.49	607.55	1183.0	0.77486	0.59003	1.3649
580	1325.5	0.02279	0.32225	583.37	515.84	1099.2	588.95	589.29	1178.2	0.78748	0.56679	1.3543
590	1430.8	0.02319	0.29367	596.61	498.43	1095.0	602.75	570.04	1172.8	0.80026	0.54306	1.3433
600	1542.5	0.02362	0.26737	610.18	480.10	1090.3	616.92	549.67	1166.6	0.81323	0.51871	1.3319
610	1660.9	0.02411	0.24309	624.11	460.73	1084.8	631.52	528.03	1159.5	0.82645	0.49363	1.3201
620	1786.2	0.02464	0.22061	638.47	440.14	1078.6	646.62	504.92	1151.5	0.83998	0.46765	1.3076
630	1918.9	0.02524	0.19972	653.35	418.12	1071.5	662.32	480.07	1142.4	0.85389	0.44056	1.2944
640	2059.3	0.02593	0.18019	668.86	394.36	1063.2	678.74	453.14	1131.9	0.86828	0.41206	1.2803
650	2207.8	0.02673	0.16184	685.16	368.44	1053.6	696.08	423.65	1119.7	0.88332	0.38177	1.2651
660	2364.9	0.02767	0.14444	702.48	339.74	1042.2	714.59	390.84	1105.4	0.89922	0.34906	1.2483
670	2531.2	0.02884	0.12774	721.23	307.22	1028.5	734.74	353.54	1088.3	0.91636	0.31296	1.2293
680	2707.3	0.03035	0.11134	742.11	269.00	1011.1	757.32	309.57	1066.9	0.93541	0.27163	1.2070
690	2894.1	0.03255	0.09451	766.81	220.77	987.6	784.24	253.96	1038.2	0.95797	0.22089	1.1789
700	3093.0	0.03670	0.07482	801.75	146.50	948.3	822.76	168.32	991.1	0.99023	0.14514	1.1354
705.10	3200.1	0.04975	0.04975	866.61	0	866.6	896.07	0	896.1	1.05257	0	1.0526

Source of Data: Tables A-4E through A-8E are generated using the Engineering Equation Solver (EES) software developed by S. A. Klein and F. L. Alvarado. The routine used in calculations is the highly accurate Steam_IAPWS, which incorporates the 1995 Formulation for the Thermodynamic Properties of Ordinary Water Substance for General and Scientific Use, issued by The International Association for the Properties of Water and Steam (IAPWS). This formulation replaces the 1984 formulation of Haar, Gallagher, and Kell (NBS/NRC Steam Tables, Hemisphere Publishing Co., 1984), which is also available in EES as the routine STEAM. The new formulation is based on the correlations of Saul and Wagner (J. Phys. Chem. Ref. Data, 16, 893, 1987) with modifications to adjust to the International Temperature Scale of 1990. The modifications are described by Wagner and Pruss (J. Phys. Chem. Ref. Data, 22, 783, 1993). The properties of ice are based on Hyland and Wexler, "Formulations for the Thermodynamic Properties of the Saturated Phases of H₂O from 173.15 K to 473.15 K," ASHRAE Trans., Part 2A, Paper 2793, 1983.

TABLE A-5E

Saturated water—Pressure table

		Specific ft ³ /lb		Inte	ernal energ Btu/Ibm	gy,		<i>nthalpy,</i> Btu/Ibm			<i>ntropy,</i> u/lbm∙R	
Press., P psia	Sat. temp., T _{sat} °F	Sat. liquid, v _f	Sat. vapor, v _g	Sat. liquid, u_f	Evap., u _{fg}	Sat. vapor, u_g	Sat. liquid, h _f	Evap., h _{fg}	Sat. vapor, h _g	Sat. liquid, s_f	Evap., s_{fg}	Sat. vapor, s_g
1	101.69	0.01614	333.49	69.72	973.99	1043.7	69.72	1035.7	1105.4	0.13262	1.84495	1.9776
2	126.02	0.01623	173.71	94.02	957.45	1051.5	94.02	1021.7	1115.8	0.17499	1.74444	1.9194
3	141.41	0.01630	118.70	109.39	946.90	1056.3	109.40	1012.8	1122.2	0.20090	1.68489	1.8858
4	152.91	0.01636	90.629	120.89	938.97	1059.9	120.90	1006.0	1126.9	0.21985	1.64225	1.8621
5	162.18	0.01641	73.525	130.17	932.53	1062.7	130.18	1000.5	1130.7	0.23488	1.60894	1.8438
6	170.00	0.01645	61.982	138.00	927.08	1065.1	138.02	995.88	1133.9	0.24739	1.58155	1.8289
8	182.81	0.01652	47.347	150.83	918.08	1068.9	150.86	988.15	1139.0	0.26757	1.53800	1.8056
10	193.16	0.01659	38.425	161.22	910.75	1072.0	161.25	981.82	1143.1	0.28362	1.50391	1.7875
14.696	211.95	0.01671	26.805	180.12	897.27	1077.4	180.16	970.12	1150.3	0.31215	1.44441	1.7566
15	212.99	0.01672	26.297	181.16	896.52	1077.7	181.21	969.47	1150.7	0.31370	1.44441	1.7549
20	227.92	0.01683	20.093	196.21	885.63	1081.8	196.27	959.93	1156.2	0.33582	1.39606	1.7319
25	240.03	0.01692	16.307	208.45	876.67	1085.1	208.52	952.03	1160.6	0.35347	1.36060	1.7141
30	250.30	0.01700	13.749	218.84	868.98	1087.8	218.93	945.21	1164.1	0.36821	1.33132	1.6995
35	259.25	0.01708	11.901	227.92	862.19	1090.1	228.03	939.16	1167.2	0.38093	1.30632	1.6872
40	267.22	0.01715	10.501	236.02	856.09	1092.1	236.14	933.69	1169.8	0.39213	1.28448	1.6766
45	274.41	0.01721	9.4028	243.34	850.52	1093.9	243.49	928.68	1172.2	0.40216	1.26506	1.6672
50	280.99	0.01727	8.5175	250.05	845.39	1095.4	250.21	924.03	1174.2	0.41125	1.24756	1.6588
55	287.05	0.01732	7.7882	256.25	840.61	1096.9	256.42	919.70	1176.1	0.41958	1.23162	1.6512
60	292.69	0.01738	7.1766	262.01	836.13	1098.1	262.20	915.61	1177.8	0.42728	1.21697	1.6442
65	297.95	0.01743	6.6560	267.41	831.90	1099.3	267.62	911.75	1179.4	0.43443	1.20341	1.6378
70	302.91	0.01748	6.2075	272.50	827.90	1100.4	272.72	908.08	1180.8	0.44112	1.19078	1.6319
75	307.59	0.01752	5.8167	277.31	824.09	1101.4	277.55	904.58	1182.1	0.44741	1.17895	1.6264
80	312.02	0.01757	5.4733	281.87	820.45	1102.3	282.13	901.22	1183.4	0.45335	1.16783	1.6212
85	316.24	0.01761	5.1689	286.22	816.97	1103.2	286.50	898.00	1184.5	0.45897	1.15732	1.6163
90	320.26	0.01765	4.8972	290.38	813.62	1104.0	290.67	894.89	1185.6	0.46431	1.14737	1.6117
95	324.11	0.01770	4.6532	294.36	810.40	1104.8	294.67	891.89	1186.6	0.46941	1.13791	1.6073
100	327.81	0.01774	4.4327	298.19	807.29	1105.5	298.51	888.99	1187.5	0.47427	1.12888	1.6032
110	334.77	0.01781	4.0410	305.41	801.37	1106.8	305.78	883.44	1189.2	0.48341	1.11201	1.5954
120	341.25	0.01789	3.7289	312.16	795.79	1107.9	312.55	878.20	1190.8	0.49187	1.09646	1.5883
130	347.32	0.01796	3.4557	318.48	790.51	1109.0	318.92	873.21	1192.1	0.49974	1.08204	1.5818
140	353.03	0.01802	3.2202	324.45	785.49	1109.9	324.92	868.45	1193.4	0.50711	1.06858	1.5757
150	358.42	0.01809	3.0150	330.11	780.69	1110.8	330.61	863.88	1194.5	0.51405	1.05595	1.5700
160	363.54	0.01815	2.8347	335.49	776.10	1111.6	336.02	859.49	1195.5	0.52061	1.04405	1.5647
170	368.41	0.01821	2.6749	340.62	771.68	1112.3	341.19	855.25	1196.4	0.52682	1.03279	1.5596
180	373.07	0.01827	2.5322	345.53	767.42	1113.0	346.14	851.16	1197.3	0.53274	1.02210	1.5548
190 200 250 300 350	377.52 381.80 400.97 417.35 431.74	0.01833 0.01839 0.01865 0.01890 0.01912	2.4040 2.2882 1.8440 1.5435 1.3263	350.24 354.78 375.23 392.89 408.55	763.31 759.32 741.02 724.77 709.98	1113.6 1114.1 1116.3 1117.7 1118.5	350.89 355.46 376.09 393.94 409.79	847.19 843.33 825.47 809.41 794.65	1198.1 1198.8 1201.6 1203.3 1204.4		1.01191 1.00219 0.95912 0.92289 0.89143	1.5111
400 450 500 550 600	444.62 456.31 467.04 476.97 486.24	0.01934 0.01955 0.01975 0.01995 0.02014	1.1617 1.0324 0.92819 0.84228 0.77020	422.70 435.67 447.68 458.90 469.46	696.31 683.52 671.42 659.91 648.88	1119.0 1119.2 1119.1 1118.8 1118.3	424.13 437.30 449.51 460.93 471.70	780.87 767.86 755.48 743.60 732.15	1205.0 1205.2 1205.0 1204.5 1203.9	0.63595 0.64900 0.66107	0.86350 0.83828 0.81521 0.79388 0.77400	1.4742 1.4642 1.4550
700 800	503.13 518.27	0.02051 0.02087	0.65589 0.56920	488.96 506.74	627.98 608.30	1116.9 1115.0	491.62 509.83	710.29 689.48	1201.9 1199.3		0.73771 0.70502	

TABLE A-5E

Saturated water—Pressure table (Concluded)

	Specific volume, ft³/lbm				<i>al energy,</i> u/lbm			<i>nthalpy,</i> Btu/Ibm		<i>Entropy,</i> Btu/lbm∙R		
Press., P psia	Sat. temp., $T_{\rm sat}$ °F	Sat. liquid, v_f	Sat. vapor, v_g	Sat. liquid, u_f	Evap., u _{fg}	Sat. vapor, u_g	Sat. liquid, h_f	Evap., h_{fg}	Sat. vapor, h_g	Sat. liquid, s_f	Evap., s _{fg}	Sat. vapor, s_g
900	532.02	0.02124	0.50107	523.19	589.54	1112.7	526.73	669.46	1196.2	0.72793	0.67505	1.4030
1000	544.65	0.02159	0.44604	538.58	571.49	1110.1	542.57	650.03	1192.6	0.74341	0.64722	1.3906
1200	567.26	0.02232	0.36241	566.89	536.87	1103.8	571.85	612.39	1184.2	0.77143	0.59632	1.3677
1400	587.14	0.02307	0.30161	592.79	503.50	1096.3	598.76	575.66	1174.4	0.79658	0.54991	1.3465
1600	604.93	0.02386	0.25516	616.99	470.69	1087.7	624.06	539.18	1163.2	0.81972	0.50645	1.3262
1800	621.07	0.02470	0.21831	640.03	437.86	1077.9	648.26	502.35	1150.6	0.84144	0.46482	1.3063
2000	635.85	0.02563	0.18815	662.33	404.46	1066.8	671.82	464.60	1136.4	0.86224	0.42409	1.2863
2500	668.17	0.02860	0.13076	717.67	313.53	1031.2	730.90	360.79	1091.7	0.91311	0.31988	1.2330
3000	695.41	0.03433	0.08460	783.39	186.41	969.8	802.45	214.32	1016.8	0.97321	0.18554	1.1587
3200.1	705.10	0.04975	0.04975	866.61	0	866.6	896.07	0	896.1	1.05257	0	1.0526

TABLE A-6E

Superl	heated wat	er										
T	V	и	h	S	V	и	h	S	V	и	h	S
°F	ft ³ /lbm	Btu/Ibm	Btu/lbm	Btu/Ibm·R	ft ³ /lbm	Btu/lbm	Btu/lbm	Btu/Ibm·R	ft ³ /lbm	Btu/Ibm	Btu/lbm	Btu/ Ibm·R
	P =	= 1.0 psia	(101.69°	F)*	Р	= 5.0 ps	ia (162.18	3°F)	Р	= 10 psi	a (193.16	5°F)
Sat.†	333.49		1105.4			1062.7		1.8438		1072.0		
200	392.53		1150.1			1076.2		1.8716	38.849		1146.4	
240	416.44		1168.3			1090.3		1.8989	41.326			1.8207
280	440.33		1186.5			1104.3		1.9246		1103.4		
320	464.20		1204.8			1118.4		1.9490	46.205 48.624	1117.6		1.8716
360 400	488.07 511.92		1223.3 1241.8			1132.5 1146.7		1.9722 1.9944		1131.9 1146.2		1.8950
440	535.77	1147.1	1260.4		102.23	1140.7		2.0156	53.441		1259.4	
500	571.54	1182.8	1288.6		114.21	1182.6		2.0461	57.041	1182.2		1.9693
600	631.14		1336.2			1219.2		2.0933		1219.0		
700	690.73		1384.6				1384.4	2.1371	69.007		1384.2	
800	750.31		1433.9				1433.7	2.1778	74.980		1433.5	
1000	869.47	1374.2	1535.1	2.4299	173.86	1374.2	1535.0	2.2524	86.913	1374.1	1534.9	2.1760
1200	988.62	1457.1	1640.0	2.4972	197.70	1457.0	1640.0	2.3198	98.840	1457.0	1639.9	2.2433
1400	1107.8	1543.7	1748.7	2.5590	221.54	1543.7	1748.7	2.3816	110.762	1543.6	1748.6	2.3052
	Р	= 15 psia	(212.99°	F)	P	' = 20 psi	a (227.92	2°F)	Р	= 40 psi	a (267.22	2°F)
Sat.	26.297	1077.7	1150.7	1.7549	20.093	1081.8	1156.2	1.7319	10.501	1092.1	1169.8	1.6766
	27.429		1163.9		20.478	1086.5		1.7406				
280	29.085	1102.4	1183.2	1.8010	21.739	1101.4		1.7679	10.713	1097.3	1176.6	1.6858
320	30.722	1116.9	1202.2	1.8260	22.980	1116.1	1201.2	1.7933	11.363	1112.9	1197.1	1.7128
360	32.348	1131.3	1221.1	1.8496	24.209	1130.7	1220.2	1.8171	11.999	1128.1	1216.9	1.7376
	33.965	1145.7	1239.9		25.429		1239.3	1.8398	12.625			1.7610
	35.576		1258.8				1258.3	1.8614	13.244		1256.0	
	37.986		1287.3		28.458		1286.9	1.8922	14.165		1285.0	
	41.988	1218.7	1335.3		31.467		1334.9	1.9398	15.686	1217.5		1.8625
	45.981		1383.9		34.467	1256.1 1294.5	1383.7	1.9837	17.197		1382.6 1432.3	
	49.967 57.930	1294.6 1374.0	1433.3 1534.8		37.461 43.438		1433.1 1534.6	2.0247 2.0994	18.702 21.700		1534.1	1.9478 2.0227
	65.885		1639.8		49.407	1456.8	1639.7	2.1668	24.691		1639.3	
	73.836	1543.6	1748.5		55.373	1543.5	1748.4	2.2287	27.678		1748.1	2.1522
	81.784	1634.0	1861.0		61.335		1860.9	2.2861	30.662		1860.7	
		= 60 psia					a (312.02			= 100 ps		
Sat	7.1766	1098.1		1.6442			1183.4			1105.5		
Sat. 320	7.1766	11098.1	1177.8			1102.3	1183.4	1.6212	4.4327	1105.5	1167.5	1.0032
360	7.4003	1125.5		1.6897		1122.7	1209.9	1.6545	4.6628	1119.8	1206.1	1.6263
400	8.3548	1140.9	1233.7					1.6794		1136.4		
440		1156.1		1.7364			1251.2		5.2006		1248.7	
	9.4005	1178.8		1.7682			1281.2		5.5876		1279.3	
	10.4256	1216.5		1.8168		1215.4		1.7841	6.2167	1214.4	1329.4	1.7586
700	11.4401	1254.5	1381.6	1.8613	8.5616	1253.8	1380.5	1.8289	6.8344	1253.0	1379.5	1.8037
	12.4484	1293.3	1431.5		9.3218	1292.6	1430.6	1.8704	7.4457	1292.0	1429.8	1.8453
	14.4543	1373.0	1533.5	-	10.8313		1532.9		8.6575		1532.4	
	16.4525	1456.2	1638.9		12.3331		1638.5		9.8615		1638.1	
	18.4464	1543.0	1747.8		13.8306		1747.5		11.0612			
	20.438	1633.5	1860.5		15.3257		1860.2		12.2584		1860.0	
	22.428	1727.6	1976.6		16.8192		1976.5		13.4541		1976.3	
2000	24.417	1825.2	2096.3	2.2694	18.3117	1825.0	2096.1	2.23/6	14.6487	1824.9	2096.0	2.2130

^{*}The temperature in parentheses is the saturation temperature at the specified pressure.

 $^{^{\}dagger}$ Properties of saturated vapor at the specified pressure.

	В			т

Superheated	water	(Concluded)

Super	neated war	lei (Conch	Jueu)									
T	V	И	h	S	V	И	h	S	V	И	h	S
°F	ft ³ /lbm	Btu/lbm	Btu/lbm	Btu/Ibm·R	ft ³ /lbm	Btu/Ibm	Btu/lbm	Btu/Ibm·R	ft ³ /lbm	Btu/lbm	Btu/lbm	Btu/ Ibm·R
	P	= 120 psia	(341.25	°F)	Р	= 140 ps	ia (353.0	3°F)	P	r = 160 ps	ia (363.5	4°F)
Sat.	3.7289	1107.9	1190.8	1.5883	3.2202	1109.9	1193.4	1.5757	2.8347	1111.6	1195.5	1.5647
360	3.8446	1116.7	1202.1			1113.4						
400	4.0799	1134.0	1224.6			1131.5			3.0076	1129.0		1.5914
450	4.3613	1154.5	1251.4			1152.6		1.6403	3.2293		1246.3	1.6234
500	4.6340	1174.4	1277.3			1172.9			3.4412		1273.2	1.6522
550	4.9010	1193.9	1302.8			1192.7			3.6469		1299.4	1.6788
600	5.1642	1213.4	1328.0			1212.3			3.8484		1325.2	
700	5.6829	1252.2	1378.4			1251.4			4.2434	1250.6		1.7498
800	6.1950	1291.4	1429.0			1290.8		1.8072	4.6316	1290.2		1.7920
1000	7.2083	1371.7	1531.8			1371.3			5.3968		1530.7	1.8682
1200	8.2137	1455.3	1637.7			1455.0			6.1540	1454.7		1.9363
1400	9.2149	1542.3	1746.9			1542.1			6.9070	1541.8		1.9986
	10.2135	1633.0	1859.8			1632.8			7.6574	1632.6		2.0563
	11.2106	1727.2	1976.1					2.1250	8.4063		1975.7	2.1102
2000	12.2067	1824.8	2095.8	2.1928	10.4624	1824.6	2095.7	2.1758	9.1542	1824.5	2095.5	2.1610
	P	= 180 psia	(373.07°	°F)	Р	= 200 ps	ia (381.8	0°F)	P	r = 225 ps	ia (391.8	0°F)
Sat.	2.5322	1113.0	1197.3	1.5548	2.2882	1114.1	1198.8	1.5460	2.0423	1115.3	1200.3	1.5360
400	2.6490	1126.3	1214.5	1.5752	2.3615	1123.5	1210.9	1.5602	2.0728	1119.7	1206.0	1.5427
450	2.8514	1148.7	1243.7		2.5488	1146.7		1.5943	2.2457		1237.6	1.5783
500	3.0433	1169.8	1271.2		2.7247	1168.2		1.6243	2.4059	1166.2		1.6091
550	3.2286	1190.2	1297.7		2.8939		1296.0		2.5590	1187.2		1.6370
600	3.4097	1210.2	1323.8		3.0586		1322.3		2.7075	1207.7		1.6628
700	3.7635	1249.8	1375.2		3.3796		1374.1		2.9956		1372.7	1.7099
800	4.1104	1289.5	1426.5		3.6934		1425.6		3.2765		1424.5	1.7528
900	4.4531	1329.7	1478.0		4.0031	1329.2		1.8059	3.5530		1476.5	1.7925
1000	4.7929	1370.5	1530.1		4.3099	1370.1		1.8430	3.8268		1528.9	1.8296
1200	5.4674	1454.3	1636.5		4.9182		1636.1		4.3689		1635.6	1.8981
1400	6.1377	1541.6	1746.0		5.5222		1745.7		4.9068		1745.4	1.9606
1600	6.8054	1632.4	1859.1		6.1238		1858.8		5.4422		1858.6	2.0184
1800 2000	7.4716 8.1367	1726.7 1824.4	1975.6		6.7238 7.3227	1726.5 1824.3	1975.4 2095.3	2.0855	5.9760 6.5087	1726.4 1824.1	1975.2 2095.1	2.0724 2.1232
2000	0.1307	1024.4	2095.4	2.14/9	1.3221	1024.3	2095.5	2.1303	0.3067	1024.1	2095.1	2.1232
	P	= 250 psia	(400.97°	°F)	Р	= 275 ps	ia (409.4	5°F)		P = 30	00 psia (4	17.35°F)
Sat.	1.8440		1201.6			1117.0			1.5435		1203.3	
	2.0027	1141.3	1234.0			1138.5		1.5499	1.6369		1226.4	
	2.1506	1164.1		1.5953	1.9415	1162.0		1.5825	1.7670		1257.9	
	2.2910	1185.6		1.6237	2.0715		1289.3	1.6115	1.8885		1287.0	
	2.4264	1206.3		1.6499		1204.9		1.6380	2.0046		1314.8	
	2.5586		1345.1			1225.6			2.1172	1224.4	10.110	1.0020
	2.6883	1247.0	1371.4			1246.0			2.2273		1368.6	
	2.9429	1287.3	1423.5			1286.5			2.4424		1421.3	
	3.1930	1327.9	1475.6			1327.3			2.6529		1473.9	
	3.4403	1369.0	1528.2			1368.5			2.8605		1526.7	
	3.9295	1453.3	1635.0			1452.9			3.2704		1634.0	
	4.4144	1540.8		1.9488 2.0066		1540.5 1631.5			3.6759		1744.2	
	4.8969	1631.7 1726.2							4.0789 4.4803		1857.7	
	5.3777 5.8575	1726.2	1974.9	2.0607		1726.0 1823.8			4.4803		1974.5 2094.6	
2000	3.6575	1023.9	2094.9	2.1110	5.5247	1023.8	2094.7	2.1010	4.0007	1023.0	2094.0	2.0913

TABL	E A-6E											
Super	heated wat	ter <i>(Contir</i>	nued)									
T	V	и	h	s	V	и	h	S	V	и	h	S
°F	ft ³ /lbm	Btu/lbm	Btu/lbm	Btu/lbm·R	ft ³ /lbm	Btu/Ibm	Btu/lbm	Btu/lbm·R	ft ³ /lbm	Btu/lbm	Btu/Ibm	Btu/ Ibm⋅R
	P	= 350 psia	(431.74	°F)	Р	= 400 ps	sia (444.6	2°F)	Р	= 450 ps	ia (456.3	1°F)
Sat.	1.3263		1204.4		1.1617			1.4852	1.0324	1119.2	1205.2	1.4742
	1.3739		1218.3		1.1747	1122.5	1209.4					
	1.4921	1155.2	1251.9		1.2851	1150.4	1245.6	1.5288	1.1233	1145.4	1238.9	
	1.6004	1178.6	1282.2		1.3840	1174.9		1.5610	1.2152	1171.1	1272.3	
	1.7030 1.8018		1310.9 1338.6		1.4765 1.5650	1197.6 1219.4	1306.9 1335.3	1.6158	1.3001 1.3807	1194.6 1216.9	1302.8	1.5737
	1.8979	1242.8	1365.8		1.6507	1240.7		1.6401	1.4584	1238.5		1.6253
	2.0848	1284.1	1419.1		1.8166	1282.5		1.6849	1.6080	1280.8	1414.7	
	2.2671	1325.3	1472.2		1.9777	1324.0	1470.4	1.7257	1.7526	1322.7	1468.6	
	2.4464	1366.9	1525.3		2.1358	1365.8	1523.9	1.7636	1.8942	1364.7	1522.4	
	2.7996	1451.7	1633.0	1.8483	2.4465	1450.9	1632.0	1.8331	2.1718	1450.1	1631.0	1.8196
1400	3.1484	1539.6	1743.5	1.9111	2.7527	1539.0	1742.7	1.8960	2.4450	1538.4	1742.0	1.8827
1600	3.4947	1630.8	1857.1	1.9691	3.0565	1630.3	1856.5	1.9541	2.7157	1629.8	1856.0	1.9409
	3.8394	1725.4	1974.0		3.3586	1725.0		2.0084	2.9847	1724.6		1.9952
2000	4.1830	1823.3	2094.2	2.0742	3.6597	1823.0	2093.9	2.0594	3.2527	1822.6	2093.5	2.0462
	Р	= 500 psia	(467.04	°F)	Р	= 600 ps	sia (486.2	4°F)	Р	= 700 ps	ia (503.1	3°F)
Sat.	0.92815	1119.1	1205.0	1.4642	0.77020	1118.3	1203.9	1.4463	0.65589	1116.9	1201.9	1.4305
	0.99304	1140.1		1.4928	0.79526			1.4596				
	1.07974	1167.1		1.5284	0.87542			1.4996	0.72799			1.4730
	1.15876	1191.4		1.5590	0.94605			1.5325	0.79332		1280.7	1.5087
	1.23312	1214.3		1.5865	1.01133			1.5614	0.85242			1.5393
	1.30440	1236.4		1.6117	1.07316			1.5877	0.90769			1.5666
	1.44097	1279.2		1.6576	1.19038			1.6348	1.01125			1.6150
	1.57252 1.70094	1321.4 1363.6		1.6992 1.7376	1.30230 1.41097			1.6771 1.7160	1.10921 1.20381	1316.0		1.6581 1.6974
	1.82726	1406.2		1.7735	1.51749			1.7522	1.20361	1402.5	1515.2	1.7341
	1.95211	1449.4		1.7733	1.62252			1.7865	1.38709	1402.5	1625.9	1.7685
	2.1988	1537.8		1.8708	1.82957			1.8501		1535.4		1.8324
	2.4430	1629.4		1.9291	2.0340	1628.4		1.9085	1.74192		1853.1	1.8911
	2.6856	1724.2		1.9834	2.2369	1723.4		1.9630	1.91643	1722.7	1970.9	1.9457
	2.9271	1822.3		2.0345	2.4387		2092.4	2.0141	2.08987		2091.7	
	Р	= 800 psia	(518.27	°F)	P =	= 1000 p	sia (544.6	55°F)	P	= 1250 p	sia (572.4	 -5°F)
Sat.	0.56920	1115.0	1199.3	1.4162	0.44604	1110.1	1192.6	1.3906	0.34549			
550	0.61586	1139.4		1.4476	0.45375			1.3972				
600	0.67799	1170.5		1.4866	0.51431			1.4457	0.37894		1217.2	
	0.73279	1197.6	1306.0		0.56411				0.42703			1.4414
	0.78330	1222.4		1.5476	0.60844					1198.7		
	0.83102	1246.0		1.5735	0.64944			1.5418	0.50344			
	0.87678	1268.9		1.5975	0.68821				0.53687			1.5347
	0.96434	1313.3		1.6413	0.76136			1.6126		1300.5		1.5826
	1.04841 1.13024	1357.0 1400.7		1.6812	0.83078 0.89783			1.6535	0.65656 0.71184	1346.7		1.6249
	1.13024	1400.7		1.7181 1.7528	0.89783			1.6911 1.7263	0.71184			1.6635
	1.36797	1534.2		1.7528	1.09101			1.7263	0.76545			
	1.52283	1626.5		1.8759	1.21610				0.86944			1.7649
	1.67606	1721.9		1.9306	1.33956					1718.4		1.8799
	1.82823	1820.4		1.9819	1.46194				1.16892			

	В			
				Е

Superheated water (Concluded) Τ h h °F ft³/lbm ft³/lbm Btu/lbm Btu/lbm ·R ft³/lbm Btu/lbm Btu/lbm Btu/lbm·R Btu/lbm Btu/lbm Btu/lbm·R $P = 1500 \text{ psia } (596.26^{\circ}\text{F})$ $P = 1750 \text{ psia } (617.17^{\circ}\text{F})$ $P = 2000 \text{ psia } (635.85^{\circ}\text{F})$ 0.27695 1092.1 1169.0 1.3362 0.18815 1066.8 1136.4 1.2863 Sat. 0.22681 1080.5 1153.9 1.3112 600 0.28189 1097.2 1175.4 1.3423 650 0.33310 0.26292 1122.8 1207.9 1.3607 0.20586 1091.4 1167.6 1.3146 1147.2 1239.7 1.4016 700 0.37198 0.24894 1147.6 1239.8 1.3783 1183.6 1286.9 1.4433 0.30252 1166.8 1264.7 1.4108 750 0.40535 1214.4 1326.9 1.4771 0.33455 1201.5 1309.8 1.4489 0.28074 1187.4 1291.3 1.4218 800 0.43550 1242.2 1363.1 1.5064 0.36266 1231.7 1349.1 1.4807 0.30763 1220.5 1334.3 1.4567 850 0.46356 1268.2 1396.9 1.5328 0.38835 1259.3 1385.1 1.5088 0.33169 1250.0 1372.8 1.4867 900 0.49015 1293.1 1429.2 1.5569 0.41238 1285.4 1419.0 1.5341 0.35390 1277.5 1408.5 1.5134 0.39479 1328.7 1474.9 1.5606 1000 0.54031 1340.9 1490.8 1.6007 0.45719 1334.9 1482.9 1.5796 1100 0.58781 1387.3 1550.5 1.6402 0.49917 1382.4 1544.1 1.6201 0.43266 1377.5 1537.6 1.6021 1200 0.63355 1433.3 1609.2 1.6767 0.53932 1429.2 1603.9 1.6572 0.46864 1425.1 1598.5 1.6400 1400 0.72172 1525.7 1726.0 1.7432 0.61621 1522.6 1722.1 1.7245 0.53708 1519.5 1718.3 1.7081 0.60269 1615.0 1838.0 1.7693 1600 0.80714 1619.8 1843.8 1.8033 0.69031 1617.4 1840.9 1.7852 1800 0.89090 1716.4 1963.7 1.8589 0.76273 1714.5 1961.5 1.8410 0.66660 1712.5 1959.2 1.8255 1815.9 2086.1 1.9108 0.83406 1814.2 2084.3 1.8931 0.72942 1812.6 2082.6 1.8778 2000 0.97358 $P = 2500 \text{ psia } (668.17^{\circ}\text{F})$ $P = 3000 \text{ psia } (695.41^{\circ}\text{F})$ P = 3500 psiaSat. 0.13076 1031.2 1091.7 1.2330 0.08460 969.8 1016.8 1.1587 650 679.9 0.8632 0.02492 663.7 700 0.16849 1098.4 1176.3 1.3072 0.09838 1005.3 1059.9 1.1960 0.03065 760.0 779.9 0.9511 750 0.20327 1154.9 1249.0 1.3686 0.14840 1114.1 1196.5 1.3118 0.10460 1057.6 1125.4 1.2434 800 0.22949 1195.9 1302.0 1.4116 0.17601 1167.5 1265.3 1.3676 0.13639 1134.3 1222.6 1.3224 850 0.25174 1230.1 1346.6 1.4463 0.19771 1208.2 1317.9 1.4086 0.15847 1183.8 1286.5 1.3721 900 0.27165 1260.7 1386.4 1.4761 0.21640 1242.8 1362.9 1.4423 0.17659 1223.4 1337.8 1.4106 950 0.29001 1289.1 1423.3 1.5028 0.23321 1273.9 1403.3 1.4716 0.19245 1257.8 1382.4 1.4428 1000 0.30726 1316.1 1458.2 1.5271 0.24876 1302.8 1440.9 1.4978 0.20687 1289.0 1423.0 1.4711 1100 0.33949 1367.3 1524.4 1.5710 0.27732 1356.8 1510.8 1.5441 0.23289 1346.1 1496.9 1.5201 1200 0.36966 1416.6 1587.6 1.6103 0.30367 1408.0 1576.6 1.5850 0.25654 1399.3 1565.4 1.5627 0.35249 1507.0 1702.7 1.6567 0.29978 1500.7 1694.8 1.6364 1400 0.42631 1513.3 1710.5 1.6802 1600 0.48004 1610.1 1832.2 1.7424 0.39830 1605.3 1826.4 1.7199 0.33994 1600.4 1820.5 1.7006 1800 0.53205 1708.6 1954.8 1.7991 0.44237 1704.7 1950.3 1.7773 0.37833 1700.8 1945.8 1.7586 2000 0.58295 1809.4 2079.1 1.8518 0.48532 1806.1 2075.6 1.8304 0.41561 1802.9 2072.1 1.8121 P = 4000 psiaP = 5000 psiaP = 6000 psia650 0.02448 657.9 676.1 0.8577 0.02379 648.3 670.3 0.8485 0.02325 640.3 666.1 0.8408 700 0.02871 742.3 763.6 0.9347 0.02678 721.8 746.6 0.9156 708.1 0.02564 736.5 0.9028 750 0.06370 962.1 1009.2 1.1410 0.03373 821.8 853.0 1.0054 0.02981 788.7 821.8 0.9747 0.05937 986.9 1041.8 1.1581 1172.1 1.2734 897.1 941.0 1.0711 800 0.10520 1094.2 0.03949 850 0.12848 1156.7 1251.8 1.3355 0.08551 1092.4 1171.5 1.2593 0.05815 1018.6 1083.1 1.1819 900 0.14647 1202.5 1310.9 1.3799 0.10390 1155.9 1252.1 1.3198 0.07584 1103.5 1187.7 1.2603 950 0.16176 1240.7 1360.5 1.4157 0.11863 1203.9 1313.6 1.3643 0.09010 1163.7 1263.7 1.3153 1000 0.17538 1274.6 1404.4 1.4463 0.13128 1244.0 1365.5 1.4004 0.10208 1211.4 1324.7 1.3578 1482.8 1.4983 1100 0.19957 1335.1 0.15298 1312.2 1453.8 1.4590 0.12211 1288.4 1424.0 1.4237 1390.3 1554.1 1.5426 0.17185 1372.1 1531.1 1.5070 0.13911 1353.4 1507.8 1.4758 1200 0.22121 1300 0.24128 1443.0 1621.6 1.5821 0.18902 1427.8 1602.7 1.5490 0.15434 1412.5 1583.8 1.5203 1400 0.26028 1494.3 1687.0 1.6182 0.20508 1481.4 1671.1 1.5868 0.16841 1468.4 1655.4 1.5598 1600 0.29620 1595.5 1814.7 1.6835 0.23505 1585.6 1803.1 1.6542 0.19438 1575.7 1791.5 1.6294 1800 0.33033 1696.8 1941.4 1.7422 0.26320 1689.0 1932.5 1.7142 0.21853 1681.1 1923.7 1.6907 2000 0.36335 1799.7 2068.6 1.7961 0.29023 1793.2 2061.7 1.7689 0.24155 1786.7 2054.9 1.7463

TABLE A-7E

Com	pressed lic	juid wate	er									
Т	V	и	h	S	V	и	h	S	V	и	h	S
°F	ft ³ /lbm	Btu/lbm	Btu/Ibm	Btu/Ibm·R	ft ³ /lbm	Btu/lbm	Btu/Ibm	Btu/lbm⋅R	ft ³ /lbm	Btu/lbm	Btu/Ibm	Btu/Ibm·R
	P =	500 psia	(467.04°	F)	P =	1000 psia	(544.65	°F)	P =	1500 psi	a (596.26	°F)
Sat.	0.019750	447.68	449.51	0.64900	0.021595	538.58	542.57	0.74341	0.023456	605.07	611.58	0.80836
32	0.015994	0.01	1.49	0.00001	0.015966	0.03	2.99	0.00005	0.015939	0.05	4.48	0.00008
50	0.015998	18.03	19.51	0.03601	0.015972	17.99	20.95	0.03593	0.015946	17.95	22.38	0.03584
100	0.016107	67.86	69.35	0.12930	0.016083	67.69	70.67	0.12899	0.016059	67.53	71.98	0.12869
150	0.016317	117.70	119.21	0.21462	0.016292	117.42	120.43	0.21416	0.016267	117.14	121.66	0.21369
200	0.016607	167.70	169.24	0.29349	0.016580	167.31	170.38	0.29289	0.016553	166.92	171.52	0.29229
250	0.016972		219.61	0.36708	0.016941	217.51	220.65	0.36634	0.016911	217.00	221.69	0.36560
300	0.017417		270.53	0.43641	0.017380	268.24	271.46	0.43551	0.017345	267.57	272.39	0.43463
350	0.017954		322.30	0.50240	0.017910	319.77	323.08	0.50132	0.017866	318.91	323.87	0.50025
400	0.018609		375.33	0.56595	0.018552		375.91	0.56463	0.018496	371.37	376.51	0.56333
450	0.019425	428.44	430.24	0.62802	0.019347		430.51	0.62635	0.019271	425.47	430.82	0.62472
500					0.020368	484.03	487.80	0.68764	0.020258	482.01	487.63	0.68550
550									0.021595	542.50	548.50	0.74731
	P = 20)00 psia (635.85°F	-)	P =	3000 ps	ia (695.4)	l°F)		P = 500	00 psia	
Sat.	0.025634	662.33	671.82	0.86224	0.034335	783.39	802.45	0.97321				
32	0.015912	0.07	5.96	0.00010	0.015859	0.10	8.90	0.00011	0.015756	0.13	14.71	0.00002
50	0.015921	17.91	23.80	0.03574	0.015870	17.83	26.64	0.03554	0.015773	17.65	32.25	0.03505
100	0.016035	67.36	73.30	0.12838	0.015988	67.04	75.91	0.12776	0.015897	66.41	81.12	0.12652
200	0.016527	166.54	172.66	0.29170	0.016475	165.79	174.94	0.29053	0.016375	164.36	179.51	0.28824
300	0.017310	266.92	273.33	0.43376	0.017242	265.65	275.22	0.43204	0.017112	263.24	279.07	0.42874
400	0.018442	370.30	377.12	0.56205	0.018338	368.22	378.41	0.55959	0.018145	364.35	381.14	0.55492
450	0.019199	424.06	431.16	0.62314	0.019062	421.36	431.94	0.62010	0.018812	416.40	433.80	0.61445
500	0.020154	480.08	487.54	0.68346	0.019960	476.45	487.53	0.67958	0.019620	469.94	488.10	0.67254
560	0.021739	552.21	560.26	0.75692	0.021405	546.59	558.47	0.75126	0.020862	537.08	556.38	0.74154
600	0.023317	605.77	614.40	0.80898	0.022759	597.42	610.06	0.80086	0.021943	584.42	604.72	0.78803
640					0.024765	654.52	668.27	0.85476	0.023358	634.95	656.56	0.83603
680					0.028821	728.63	744.64	0.92288	0.025366	690.67	714.14	0.88745
700									0.026777	721.78	746.56	0.91564

TABLE A-8E

Saturated ice—water vapor

			<i>c volume,</i> ³/lbm	Internal energy, Btu/lbm				E <i>nthalpy,</i> Btu/lbm		Entropy, Btu/lbm·R		
Temp.,	Sat. press., P_{sat} psia	Sat. ice,	Sat. vapor, v_g	Sat. ice,	Subl.,	Sat. vapor, u _g	Sat. ice, h _i	Subl., h _{ig}	Sat. vapor, h_g	Sat. ice, s _i	Subl., s_{ig}	Sat. vapor, s_g
	- Gut											
	0.08871	0.01747	3299.6	-143.34		1020.9	-143.34	1218.3	1075.0	-0.29146	2.4779	2.1864
32	0.08864	0.01747	3302.6	-143.35	1164.2	1020.9	-143.35	1218.4	1075.0	-0.29148	2.4779	2.1865
30	0.08086	0.01747	3605.8	-144.35	1164.6	1020.2	-144.35	1218.5	1074.2	-0.29353	2.4883	2.1948
25	0.06405	0.01746	4505.8	-146.85	1165.4	1018.6	-146.85	1218.8	1072.0	-0.29865	2.5146	2.2160
20	0.05049	0.01746	5657.6	-149.32		1016.9	-149.32	1219.1	1069.8	-0.30377	2.5414	2.2376
15	0.03960	0.01745	7138.9	-151.76	1167.0	1015.2	-151.76	1219.3	1067.6	-0.30889	2.5687	2.2598
10	0.03089	0.01744	9054.0	-154.18	1167.8	1013.6	-154.18	1219.5	1065.4	-0.31401	2.5965	2.2825
5	0.02397	0.01743	11,543	-156.57	1168.5	1011.9	-156.57	1219.7	1063.1	-0.31913	2.6248	2.3057
0	0.01850	0.01743	14,797	-158.94	1169.2	1010.3	-158.94	1219.9	1060.9	-0.32426	2.6537	2.3295
-5	0.01420	0.01742	19,075	-161.28	1169.9	1008.6	-161.28	1220.0	1058.7	-0.32938	2.6832	2.3538
-10	0.01083	0.01741	24,731	-163.60	1170.6	1007.0	-163.60	1220.1	1056.5	-0.33451	2.7133	2.3788
-15	0.00821	0.01740	32,257	-165.90	1171.2	1005.3	-165.90	1220.2	1054.3	-0.33964	2.7440	2.4044
-20	0.00619	0.01740	42,335	-168.16	1171.8	1003.6	-168.16	1220.3	1052.1	-0.34478	2.7754	2.4306
-25	0.00463	0.01739	55,917	-170.41	1172.4	1002.0	-170.41	1220.3	1049.9	-0.34991	2.8074	2.4575
-30	0.00344	0.01738	74,345	-172.63	1173.0	1000.3	-172.63	1220.3	1047.7	-0.35505	2.8401	2.4850
-35	0.00254	0.01738	99,526	-174.83	1173.5	998.7	-174.83	1220.3	1045.5	-0.36019	2.8735	2.5133
-40	0.00186	0.01737	,		1174.0	997.0	-177.00	1220.3	1043.3	-0.36534	2.9076	2.5423

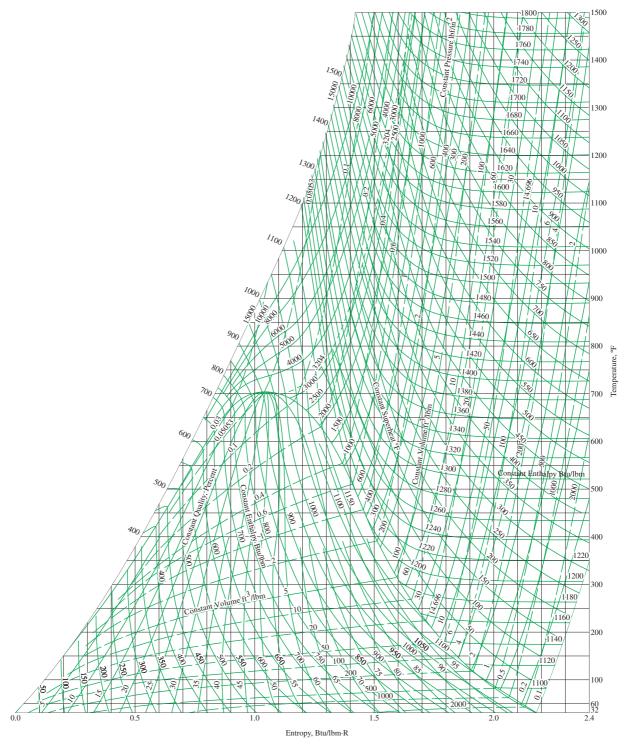


FIGURE A-9E

T-s diagram for water.

Source of Data: Joseph H. Keenan, Frederick G. Keyes. Philip G. Hill. and Joan G. Moore. Steam Tables (New York: John Wiley & Sons, 1969)

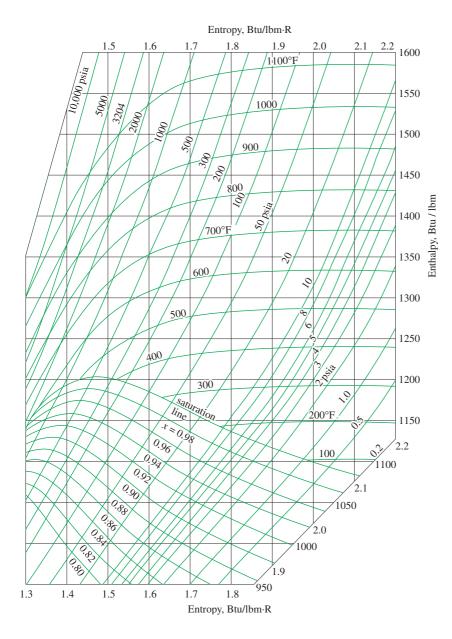


FIGURE A-10E

Mollier diagram for water.

Source of Data: Joseph H. Keenan, Frederick G. Keyes. Philip G. Hill. and Joan G. Moore. Steam Tables (New York: John Wiley & Sons, 1969)

TABLE A-11E

Saturated refrigerant-134a—Temperature table

		Specific ft ³ /II		Inte	ernal ene Btu/lbm	ergy,		Enthalpy Btu/lbm	,		Entropy, Btu/lbm·F	
	Sat.	Sat.	Sat.	Sat.		Sat.	Sat.		Sat.	Sat.		Sat.
Temp.		liquid,	vapor,	liquid,	Evap.,	vapor,	liquid,	Evap.,	vapor,	liquid,	Evap.,	vapor,
<i>T</i> °F	P _{sat} psia	V_f	V_g	U_f	U_{fg}	U_g	h_f	h_{fg}	h_g	S_f	S_{fg}	S_g
-40	7.432	0.01130	5.7769	-0.016	89.174	89.16	0.000	97.104	97.10	0.00000	0.23136	0.23136
-35	8.581	0.01136	5.0489	1.483	88.360	89.84	1.501	96.360	97.86	0.00355	0.22689	0.23044
-30	9.869	0.01143	4.4286	2.987	87.542	90.53	3.008	95.608	98.62	0.00707	0.22250	0.22957
-25	11.306	0.01149	3.8980	4.497	86.717	91.21	4.522	94.849	99.37	0.01057	0.21819	0.22876
-20	12.906	0.01156	3.4424	6.014	85.887	91.90	6.041	94.080	100.12	0.01404	0.21396	0.22800
-15	14.680	0.01163	3.0495	7.536	85.050	92.59	7.568	93.303	100.87	0.01748	0.20981	0.22729
-10	16.642	0.01170	2.7097	9.065	84.206	93.27	9.102	92.515	101.62	0.02090	0.20572	0.22662
-5	18.806		2.4146	10.601	83.355	93.96	10.642	91.717	102.36	0.02430	0.20171	0.22600
0	21.185	0.01185	2.1575	12.143	82.496	94.64	12.190	90.907	103.10	0.02767	0.19775	0.22542
5	23.793	0.01193	1.9328	13.693	81.628	95.32	13.745	90.085	103.83	0.03103	0.19385	0.22488
10	26.646	0.01200	1.7358	15.249	80.751	96.00	15.308	89.251	104.56	0.03436	0.19001	0.22437
15	29.759	0.01208	1.5625	16.813	79.865	96.68	16.879	88.403	105.28	0.03767	0.18623	0.22390
20	33.147	0.01216	1.4097	18.384	78.969	97.35	18.459	87.541	106.00	0.04097	0.18249	0.22345
25	36.826	0.01225	1.2746	19.963	78.062	98.03	20.047	86.665	106.71	0.04424	0.17880	0.22304
30	40.813	0.01233	1.1548	21.550	77.144	98.69	21.643	85.772	107.42	0.04750	0.17515	0.22265
35	45.124	0.01242	1.0482	23.145	76.214	99.36	23.249	84.863	108.11	0.05074	0.17154	0.22228
40	49.776	0.01251	0.95323	24.749		100.02	24.864	83.937	108.80	0.05397	0.16797	0.22194
45	54.787	0.01261	0.86837	26.361	74.317	100.68	26.489	82.993	109.48	0.05718	0.16443	0.22162
50	60.175	0.01270	0.79236	27.983	73.347	101.33	28.124	82.029	110.15	0.06038	0.16093	0.22131
55	65.957	0.01280	0.72414	29.614		101.98	29.770	81.046	110.82	0.06357	0.15746	0.22103
60	72.152	0.01290	0.66277	31.254		102.62	31.426	80.041	111.47	0.06674	0.15401	0.22075
65	78.780	0.01301	0.60744	32.904		103.25	33.094	79.014	112.11	0.06991	0.15058	0.22049
70	85.858	0.01311	0.55746	34.565		103.88	34.773	77.964	112.74	0.07306	0.14718	0.22024
75	93.408	0.01323	0.51222	36.237		104.50	36.465	76.889	113.35	0.07621	0.14379	0.22000
80	101.45	0.01334	0.47119	37.920		105.11	38.170	75.788	113.96	0.07934	0.14042	0.21976
	110.00	0.01346	0.43391	39.614		105.72	39.888	74.660	114.55	0.08247	0.13706	0.21953
	119.08	0.01359	0.39997	41.321	64.989	106.31	41.620	73.503	115.12	0.08560	0.13371	0.21931
	128.72 138.93	0.01372 0.01386	0.36902 0.34074	43.041 44.774		106.89 107.46	43.367 45.130	72.315 71.094	115.68 116.22	0.08872 0.09183	0.13036 0.12702	0.21908 0.21885
	149.73	0.01386	0.34074	46.521		107.46	46.909	69.838	116.22	0.09165	0.12702	0.21862
	149.73	0.01400	0.29113	48.284	60.284	108.02	48.706	68.544	117.25	0.09495	0.12307	0.21838
	173.23	0.01413	0.26933	50.063		108.37	50.521	67.210	117.23	0.10118	0.12031	0.21813
	185.96	0.01430	0.24928	51.858		109.10	52.356	65.833	117.73	0.10118	0.11054	0.21813
	213.53	0.01440	0.24328	55.505		110.58	56.091	62.935	119.03	0.10430	0.11330	0.21788
140	244.06	0.01462	0.21373	59.237	52.221	111.46	59.925	59.813	119.03	0.11686	0.10072	0.21728
	277.79	0.01522	0.15331	63.070		112.22	63.875	56.419	120.29	0.11000	0.09973	0.21500
	314.94	0.01507	0.13707	67.022		112.22	67.965	52.690	120.29	0.12324	0.09233	0.21377
	355.80	0.01613	0.13423	71.139		113.24	72.246	48.509	120.75	0.12371	0.00302	0.21473
	400.66	0.01002	0.09619	75.464		113.24	76.768	43.721	120.73	0.13037	0.06834	0.21340
	449.90	0.01753	0.03013	80.093		113.02		38.025	119.67	0.14327	0.05852	0.21101
200	504.00	0.02010	0.06441	85.297	26.629	111.93	87.172	30.761	117.93	0.15872	0.03652	0.20534
210	563.76	0.02309	0.04722	91.993		108.49	94.402	19.015	113.42	0.16924	0.02839	0.19763
			,,									

Source of Data: Tables A-11E through A-13E are generated using the Engineering Equation Solver (EES) software developed by S. A. Klein and F. L. Alvarado. The routine used in calculations is the R134a, which is based on the fundamental equation of state developed by R. Tillner-Roth and H.D. Baehr, "An International Standard Formulation for the Thermodynamic Properties of 1,1,1,2-Tetrafluoroethane (HFC-134a) for temperatures from 170 K to 455 K and pressures up to 70 MPa," *J. Phys. Chem, Ref. Data,* Vol. 23, No. 5, 1994. The enthalpy and entropy values of saturated liquid are set to zero at -40°C (and -40°F).

TABLE A-12E

Saturated refrigerant-134a—Pressure table

Sat. Sat. Iquid, Press., temp., 1 liquid, 2 vapor, 1 liquid, 4 vap	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Sat.
5 -53.09 0.01113 8.3740 -3.914 91.283 87.37 -3.903 99.021 95.12 -0.00944 0.24353 10 -29.52 0.01143 4.3740 3.132 87.463 90.59 3.153 95.536 98.69 0.00741 0.22208 15 -14.15 0.01161 2.9882 7.796 84.907 92.70 7.828 93.170 101.00 0.01806 0.20911 20 -2.43 0.01181 2.2781 11.393 82.915 94.31 11.436 91.302 102.74 0.02603 0.19967 25 7.17 0.01209 1.5506 16.929 79.799 96.73 16.996 88.340 105.34 0.032247 0.12921 35 22.57 0.01221 1.3382 191.95 78.504 97.70 19.274 87.093 106.37 0.04265 0.18068 40 29.01 0.01221 1.1773 21.266 98.56 21.327 85.950 107.28	vapor,
10	S _g
15 -14.15 0.01164 2.9882 7.796 84.907 92.70 7.828 93.170 101.00 0.0186 0.20911 25 7.17 0.01196 1.8442 14.367 81.249 95.62 14.422 89.725 104.15 0.03247 0.19218 30 15.37 0.01209 1.5506 16.929 79.799 96.73 16.996 88.340 105.34 0.03792 0.1805 35 22.57 0.01221 1.3382 19.195 78.504 97.70 19.274 87.093 106.37 0.04686 0.17586 45 34.86 0.01242 1.0510 23.101 76.240 99.34 23.205 84.889 108.09 0.05655 0.17164 50 40.23 0.01250 0.84929 24.824 75.228 100.05 24.939 83.894 108.09 0.05655 0.17164 60 49.84 0.01270 0.79462 27.932 73.378 101.31 28.054 10.951	0.23409
20 -2.43 0.01181 2.2781 11.393 82.915 94.31 11.436 91.302 102.74 0.02603 0.19967 25 7.17 0.01196 1.8442 14.367 81.249 95.62 14.422 89.725 104.15 0.03247 0.19218 30 15.37 0.01221 1.3382 19.195 78.504 97.70 19.274 87.093 106.37 0.04265 0.18595 40 29.01 0.01232 1.1773 21.236 77.326 98.56 21.327 85.950 107.28 0.04686 0.17586 45 34.86 0.01242 1.0510 23.101 76.240 99.34 23.205 84.889 108.09 0.05065 0.17164 55 45.20 0.01261 0.86509 24.224 75.228 100.05 24.939 83.894 108.83 0.05732 0.16429 60 49.84 0.01278 0.73462 29.351 72.523 101.87 29.505 81.205 <td>0.22949</td>	0.22949
25 7.17 0.01196 1.8442 14.367 81.249 95.62 14.422 89.725 104.15 0.03247 0.19218 30 15.37 0.01209 1.5506 16.929 79.799 96.73 16.996 88.340 105.34 0.03792 0.18595 35 22.57 0.01221 1.3382 19.195 78.504 97.70 19.274 87.093 106.37 0.04265 0.18058 40 29.01 0.01232 1.1773 21.236 77.326 98.56 21.327 85.950 107.28 0.04686 0.17586 45 34.86 0.01242 1.0510 23.101 76.240 99.34 23.205 84.889 108.09 0.05055 0.17164 55 45.20 0.01261 0.86509 26.428 74.277 100.70 26.556 82.954 109.51 0.05732 0.16429 60 49.84 0.01270 0.79462 27.932 73.378 101.31 280.060 110.13 </td <td>0.22717</td>	0.22717
30	0.22570 0.22465
35 22.57 0.01221 1.3382 19.195 78.504 97.70 19.274 87.093 106.37 0.04265 0.18058 40 29.01 0.01232 1.1773 21.236 77.326 98.56 21.327 85.950 107.28 0.04686 0.17586 45 34.86 0.01242 1.0510 23.101 76.240 99.34 23.205 84.889 108.09 0.05065 0.17164 50 40.23 0.01252 0.94909 24.824 75.228 100.05 24.939 83.894 108.83 0.05112 0.16780 55 45.20 0.01270 0.79462 27.932 73.378 101.31 28.073 101.13 0.06028 0.16104 65 54.20 0.01278 0.73462 29.351 72.523 101.87 29.505 81.205 110.71 0.06306 0.15801 70 58.30 0.01278 0.63784 31.975 70.921 102.90 32.155 79.594 111.	0.22386
40 29.01 0.01232 1.1773 21.236 77.326 98.56 21.327 85.950 107.28 0.04686 0.17586 45 34.86 0.01242 1.0510 23.101 76.240 99.34 23.205 84.889 108.09 0.05065 0.17164 50 40.23 0.01261 0.86509 26.428 74.277 100.70 26.556 82.954 109.51 0.05732 0.16429 60 49.84 0.01270 0.79462 29.351 72.523 101.87 29.505 81.205 110.71 0.06028 0.16104 65 54.20 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 75 62.19 0.01295 0.63784 31.975 70.921 102.90 32.155 79.594 111.75 0.06813 0.15251 80 65.89 0.01303 0.59622 33.198 70.167 103.36 33.391 78	0.22324
50 40.23 0.01252 0.94909 24.824 75.228 100.05 24.939 83.894 108.83 0.05412 0.16780 55 45.20 0.01261 0.86509 26.428 74.277 100.70 26.556 82.954 109.51 0.05732 0.16429 60 49.84 0.01270 0.79462 27.932 73.378 101.31 28.073 82.060 110.13 0.06028 0.16104 65 54.20 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 70 58.30 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 75 62.19 0.01303 0.59822 33.198 70.167 103.36 33.391 78.830 112.25 0.06567 0.15580 85 69.41 0.01310 0.56309 34.369 69.438 103.31 34.575 <t< td=""><td>0.22272</td></t<>	0.22272
55 45.20 0.01261 0.86509 26.428 74.277 100.70 26.556 82.954 109.51 0.05732 0.16429 60 49.84 0.01270 0.79462 27.932 73.378 101.31 28.073 82.060 110.13 0.06028 0.16104 65 54.20 0.01278 0.73462 29.351 72.523 101.87 29.505 81.205 110.71 0.06306 0.15801 70 58.30 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 75 62.19 0.01303 0.59822 33.198 70.167 103.36 33.391 78.830 112.22 0.07047 0.14998 85 69.41 0.01310 0.56309 34.369 69.438 103.81 34.575 78.089 112.66 0.07269 0.14758 90 72.78 0.01318 0.55173 35.494 68.733 104.23 35.713 <t< td=""><td>0.22229</td></t<>	0.22229
60 49.84 0.01270 0.79462 27.932 73.378 101.31 28.073 82.060 110.13 0.06028 0.16104 65 54.20 0.01278 0.73462 29.351 72.523 101.87 29.505 81.205 110.71 0.06306 0.15801 70 58.30 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 75 62.19 0.01295 0.63784 31.975 70.921 102.90 32.155 79.594 111.75 0.06813 0.15251 80 65.89 0.01303 0.59822 33.198 70.167 103.36 33.391 78.830 112.22 0.07047 0.14998 85 69.41 0.01318 0.53173 35.494 68.733 104.23 35.713 77.369 113.08 0.07481 0.14529 95 76.02 0.01332 0.47811 37.623 67.383 105.01 37.870 <t< td=""><td>0.22192</td></t<>	0.22192
65 54.20 0.01278 0.73462 29.351 72.523 101.87 29.505 81.205 110.71 0.06306 0.15801 70 58.30 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 75 62.19 0.01295 0.63784 31.975 70.21 102.90 32.155 79.594 111.75 0.06813 0.15251 80 65.89 0.01303 0.59822 33.198 70.167 0.07047 0.14998 85 69.41 0.01310 0.56309 34.369 69.438 103.81 34.575 78.089 112.66 0.07269 0.14758 90 72.78 0.01318 0.53356 36.577 68.048 104.63 36.810 76.668 113.48 0.07684 0.14311 100 79.12 0.01332 0.47811 37.623 67.383 105.01 37.870 75.984 113.85 0.07879 0.14101 <td>0.22160</td>	0.22160
70 58.30 0.01287 0.68290 30.696 71.705 102.40 30.862 80.385 111.25 0.06567 0.15518 75 62.19 0.01295 0.63784 31.975 70.921 102.90 32.155 79.594 111.75 0.06813 0.15251 80 65.89 0.01303 0.59822 33.198 70.167 103.36 33.391 78.830 112.22 0.07047 0.14998 85 69.41 0.01310 0.56309 34.369 69.438 103.81 34.575 78.089 112.66 0.07269 0.14758 90 72.78 0.01318 0.53173 35.494 68.733 104.23 35.713 77.369 113.08 0.07481 0.14529 95 76.02 0.01325 0.50356 36.577 68.048 104.63 36.810 76.668 113.48 0.07684 0.14311 100 79.12 0.01332 0.47811 37.623 67.383 105.01 37.870 <	0.22132
75 62.19 0.01295 0.63784 31.975 70.921 102.90 32.155 79.594 111.75 0.06813 0.15251 80 65.89 0.01303 0.59822 33.198 70.167 103.36 33.391 78.830 112.22 0.07047 0.14998 85 69.41 0.01310 0.56309 34.369 69.438 103.81 34.575 78.089 112.66 0.07269 0.14758 90 72.78 0.01318 0.53173 35.494 68.733 104.23 35.713 77.369 113.08 0.07481 0.14529 95 76.02 0.01325 0.50356 36.577 68.048 104.63 36.810 76.668 113.48 0.07684 0.14311 100 79.12 0.01326 0.47811 37.623 67.383 105.01 37.870 75.984 113.85 0.07879 0.14101 110 85.00 0.01346 0.43390 39.614 66.102 105.72 39.888	0.22107
80 65.89 0.01303 0.59822 33.198 70.167 103.36 33.391 78.830 112.22 0.07047 0.14998 85 69.41 0.01310 0.56309 34.369 69.438 103.81 34.575 78.089 112.66 0.07269 0.14758 90 72.78 0.01318 0.53173 35.494 68.733 104.23 35.713 77.369 113.08 0.07481 0.14529 95 76.02 0.01325 0.50356 36.577 68.048 104.63 36.810 76.668 113.48 0.07684 0.14311 100 79.12 0.01332 0.47811 37.623 67.383 105.01 37.870 75.984 113.85 0.07879 0.14101 110 85.00 0.01346 0.43390 39.614 66.102 105.72 39.888 74.660 114.55 0.08247 0.13706 120 90.49 0.01360 0.39681 41.489 64.878 106.37 41.791	0.22084 0.22064
85 69.41 0.01310 0.56309 34.369 69.438 103.81 34.575 78.089 112.66 0.07269 0.14758 90 72.78 0.01318 0.53173 35.494 68.733 104.23 35.713 77.369 113.08 0.07481 0.14529 95 76.02 0.01325 0.50356 36.577 68.048 104.63 36.810 76.668 113.48 0.07684 0.14311 100 79.12 0.01332 0.47811 37.623 67.383 105.01 37.870 75.984 113.85 0.07879 0.14101 110 85.00 0.01346 0.43390 39.614 66.102 105.72 39.888 74.660 114.55 0.08247 0.13706 120 90.49 0.01360 0.39681 41.489 64.878 106.37 41.791 73.388 115.18 0.08590 0.13338 130 95.64 0.01374 0.36523 43.263 63.704 106.97 43.594	0.22045
90 72.78 0.01318 0.53173 35.494 68.733 104.23 35.713 77.369 113.08 0.07481 0.14529 95 76.02 0.01325 0.50356 36.577 68.048 104.63 36.810 76.668 113.48 0.07684 0.14311 100 79.12 0.01332 0.47811 37.623 67.383 105.01 37.870 75.984 113.85 0.07879 0.14101 110 85.00 0.01346 0.43390 39.614 66.102 105.72 39.888 74.660 114.55 0.08247 0.13706 120 90.49 0.01360 0.39681 41.489 64.878 106.37 41.791 73.388 115.18 0.08590 0.13338 130 95.64 0.01374 0.36523 43.263 63.704 106.97 43.594 72.159 115.75 0.08912 0.12993 140 100.51 0.01387 0.33800 44.951 62.570 107.52 45.311 70.967 116.28 0.09215 0.12668 150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.20 0.09776 0.12064 170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 65.402 118.32 0.10526 0.11252 200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.22027
100 79.12 0.01332 0.47811 37.623 67.383 105.01 37.870 75.984 113.85 0.07879 0.14101 110 85.00 0.01346 0.43390 39.614 66.102 105.72 39.888 74.660 114.55 0.08247 0.13706 120 90.49 0.01360 0.39681 41.489 64.878 106.37 41.791 73.388 115.18 0.08590 0.13338 130 95.64 0.01374 0.36523 43.263 63.704 106.97 43.594 72.159 115.75 0.08912 0.12993 140 100.51 0.01387 0.33800 44.951 62.570 107.52 45.311 70.967 116.28 0.09215 0.12668 150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527	0.22011
110 85.00 0.01346 0.43390 39.614 66.102 105.72 39.888 74.660 114.55 0.08247 0.13706 120 90.49 0.01360 0.39681 41.489 64.878 106.37 41.791 73.388 115.18 0.08590 0.13338 130 95.64 0.01374 0.36523 43.263 63.704 106.97 43.594 72.159 115.75 0.08912 0.12993 140 100.51 0.01487 0.33800 44.951 62.570 107.52 45.311 70.967 116.28 0.09215 0.12668 150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507	0.21995
120 90.49 0.01360 0.39681 41.489 64.878 106.37 41.791 73.388 115.18 0.08590 0.13338 130 95.64 0.01374 0.36523 43.263 63.704 106.97 43.594 72.159 115.75 0.08912 0.12993 140 100.51 0.01387 0.33800 44.951 62.570 107.52 45.311 70.967 116.28 0.09215 0.12668 150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.20 0.09776 0.12064 170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 <td>0.21981</td>	0.21981
130 95.64 0.01374 0.36523 43.263 63.704 106.97 43.594 72.159 115.75 0.08912 0.12993 140 100.51 0.01387 0.33800 44.951 62.570 107.52 45.311 70.967 116.28 0.09215 0.12668 150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.20 0.09776 0.12064 170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 <td>0.21953</td>	0.21953
140 100.51 0.01387 0.33800 44.951 62.570 107.52 45.311 70.967 116.28 0.09215 0.12668 150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.20 0.09776 0.12064 170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 65.402 118.32 0.10526 0.11252 200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 </td <td>0.21928</td>	0.21928
150 105.12 0.01400 0.31426 46.563 61.473 108.04 46.952 69.807 116.76 0.09502 0.12359 160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.20 0.09776 0.12064 170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 65.402 118.32 0.10526 0.11252 200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 </td <td>0.21905 0.21883</td>	0.21905 0.21883
160 109.50 0.01413 0.29339 48.109 60.406 108.51 48.527 68.674 117.20 0.09776 0.12064 170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 65.402 118.32 0.10526 0.11252 200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 </td <td>0.21861</td>	0.21861
170 113.69 0.01426 0.27487 49.595 59.366 108.96 50.043 67.564 117.61 0.10036 0.11783 180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 65.402 118.32 0.10526 0.11252 200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09627 280 150.62 </td <td>0.21840</td>	0.21840
180 117.69 0.01439 0.25833 51.027 58.349 109.38 51.507 66.475 117.98 0.10286 0.11513 190 121.53 0.01452 0.24346 52.412 57.353 109.76 52.922 65.402 118.32 0.10526 0.11252 200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09627 280 150.62 0.01570 0.15555 63.313 48.951 112.26 64.126 56.197 120.32 0.12364 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800 </td <td>0.21819</td>	0.21819
200 125.22 0.01464 0.23001 53.753 56.375 110.13 54.295 64.345 118.64 0.10757 0.11000 220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09627 280 150.62 0.01570 0.15555 63.313 48.951 112.26 64.126 56.197 120.32 0.12364 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.21799
220 132.21 0.01490 0.20662 56.321 54.462 110.78 56.927 62.267 119.19 0.11195 0.10519 240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09627 280 150.62 0.01570 0.15555 63.313 48.951 112.26 64.126 56.197 120.32 0.12364 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.21778
240 138.73 0.01516 0.18694 58.757 52.596 111.35 59.430 60.225 119.65 0.11606 0.10063 260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09627 280 150.62 0.01570 0.15555 63.313 48.951 112.26 64.126 56.197 120.32 0.12364 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.21757
260 144.85 0.01543 0.17012 61.082 50.763 111.84 61.824 58.205 120.03 0.11994 0.09627 280 150.62 0.01570 0.15555 63.313 48.951 112.26 64.126 56.197 120.32 0.12364 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.21714
280 150.62 0.01570 0.15555 63.313 48.951 112.26 64.126 56.197 120.32 0.12364 0.09207 300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.21669
300 156.09 0.01598 0.14279 65.460 47.154 112.61 66.347 54.195 120.54 0.12717 0.08800	0.21622 0.21571
	0.21571
330 100.04 0.01072 0.11073 70.307 42.032 113.20 71.031 49.109 120.70 0.13343 0.07813	0.21317
400 179.86 0.01758 0.09643 75.401 37.957 113.36 76.702 43.794 120.50 0.14317 0.06847	0.21164
450 190.02 0.01860 0.07979 80.112 32.909 113.02 81.662 38.003 119.67 0.15060 0.05849	0.20909
500 199.29 0.01997 0.06533 84.900 27.096 112.00 86.748 31.292 118.04 0.15810 0.04748	0.20558

TABLE A-13E

Superheated refrigerant-134a	Super	heated	refrie	gerant-1	L34a
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Superi	neated re	ingerant	154a									
T	V	И	h	S	V	И	h	S	V	и	h	S
°F	ft ³ /lbm	Btu/lbm	Btu/lbm	Btu/Ibm·R	ft ³ /lbm	Btu/Ibm	Btu/Ibm	Btu/lbm⋅R	ft ³ /lbm	Btu/Ibm	Btu/lbm	Btu/Ibm·R
	$P = 10 \text{ psia } (T_{\text{sat}} = -29.52^{\circ}\text{F})$				$P = 15 \text{ psia } (T_{\text{sat}} = -14.15^{\circ}\text{F})$			$P = 20 \text{ psia } (T_{\text{sat}} = -2.43^{\circ}\text{F})$				
Sat.	4.3740	90.59		0.22949	2.9882	92.70	101.00	0.22717	2.2781	94.31	102.74	0.22570
-20 0	4.4856 4.7135			0.23351 0.24175	3.1001	95.08	103.69	0.23312	2.2922	94 73	103.21	0.22673
20	4.9380			0.24978	3.2551	98.49	107.52	0.24129	2.4130	98.19		0.23506
40				0.25763	3.4074	101.96		0.24924	2.5306	101.71		0.24313
60				0.26533	3.5577	105.51		0.25702		105.29		
80				0.27290	3.7064	109.14		0.26465		108.94		
100	5.8165	113.02	123.78	0.28035	3.8540	112.85	123.54	0.27214	2.8726	112.67		0.26623
120	6.0331	116.80	127.96	0.28768	4.0006	116.64	127.75	0.27952	2.9842	116.48	127.53	0.27364
140	6.2490	120.66	132.23	0.29492	4.1464	120.52	132.03	0.28678	3.0950	120.38	131.83	0.28094
160	6.4642	124.62	136.58	0.30205	4.2915	124.49	136.40	0.29395	3.2051	124.35	136.22	0.28814
180	6.6789	128.66	141.01	0.30910	4.4361	128.53	140.85	0.30102		128.41		0.29523
200	6.8930	132.78	145.54	0.31606	4.5802	132.67	145.38	0.30800		132.56		0.30223
220	7.1068	136.99	150.14	0.32293	4.7239	136.89	150.00	0.31489	3.5324	136.78	149.86	0.30914
	$P = 30 \text{ psia } (T_{\text{sat}} = 15.37^{\circ}\text{F})$				P =	$P = 40 \text{ psia } (T_{\text{sat}} = 29.01^{\circ}\text{F})$			$P = 50 \text{ psia } (T_{\text{sat}} = 40.23^{\circ}\text{F})$			
Sat.	1.5506			0.22386	1.1773	98.56	107.28	0.22272	0.9491	100.05	108.83	0.22192
20	1.5691			0.22583	1 0100	100.61	100 50	0.00740				
40				0.23416		100.61		0.22740	1 0010	102.05	110 10	0.02022
60				0.24220	1.2768	104.35		0.23567				0.23033
80				0.25003 0.25769	1.3389 1.3995	108.12 111.94		0.24365	1.0540 1.1043		117.44 121.78	
100 120				0.25769			126.63	0.25142 0.25902	1.1534	111.36		0.24639
140				0.20319		119.79		0.26646		119.48		
160				0.27230		123.82		0.27377		123.54		
180				0.28695			140.00	0.28096	1.2955	127.67		0.27622
200				0.29399	1.6887		144.60	0.28805		131.87		0.28335
220			149.57			136.37		0.29503		136.15		0.29037
240			154.30	0.30780	1.8007		154.03	0.30192		140.51		0.29730
260			159.11	0.31458			158.86	0.30873		144.94		0.30413
280				0.32128		149.62		0.31545		149.45		0.31087
	$P = 60 \text{ psia } (T_{\text{sat}} = 49.84^{\circ}\text{F})$				$P = 70 \text{ psia } (T_{\text{sat}} = 58.30^{\circ}\text{F})$			$P = 80 \text{ psia } (T_{\text{sat}} = 65.89^{\circ}\text{F})$				
Sat.				0.22132			111.25	0.22084				0.22045
60				0.22572		102.74		0.22157	0.5362	105.50	112.22	0.22043
80				0.23408		106.77		0.23018	0.6243	106 27	115 51	0.22663
100				0.24212		110.77		0.23838	0.6601			0.23501
120		115.14		0.24992			125.20	0.24630	0.6941	114.43		0.24305
140				0.25753			129.74	0.25399		118.53		0.25084
160				0.26497					0.7589			
180	1.0709	127.42	139.31	0.27227	0.9105	127.16	138.95	0.26886	0.7900	126.89	138.59	0.26585
200				0.27945	0.9447	131.40		0.27608	0.8206		143.31	
220				0.28651	0.9785	135.72		0.28318	0.8507		148.09	
240				0.29346	1.0118		153.22	0.29017	0.8803		152.94	
260				0.30032	1.0449	144.57		0.29706	0.9096		157.85	
280				0.30709		149.10		0.30386	0.9386		162.82	
300	1.3004	153.88	168.31	0.31378	1.1101	153.71	168.09	0.31057	0.9674	153.55	167.87	0.30775
320	1.3377	158.55	173.40	0.32039	1.1424	158.40	173.20	0.31720	0.9959	158.25	172.99	0.31440

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Superheated refrigerant-134a (Concluded)

Super	heated ref	rigerant-	134a (<i>C</i>	oncluded)								
T	V	и	h	S	V	и	h	S	V	и	h	S
°F	ft ³ /lbm	Btu/Ibm	Btu/lbm	Btu/lbm⋅R	ft ³ /lbm	Btu/Ibm	Btu/Ibm	Btu/Ibm∙R	ft ³ /lbm	Btu/lbm	Btu/lbm	Btu/Ibm·R
	P = 9	90 psia (<i>T</i>	$r_{\rm sat} = 72.7$	78°F)	P = 1	100 psia ($T_{\rm sat} = 79.$.12°F)	P = 1	120 psia ($T_{\rm sat} = 90.$	49°F)
Sat.				0.22011				0.21981		106.37		0.21928
80				0.22332	0.47906				0.03001	100.07	110.10	0.21520
100				0.23191	0.51076			0.22902	0.41013	108.49	117.59	0.22364
120		114.05		0.24009	0.54022			0.23735	0.43692			0.23234
140		118.20		0.24799	0.56821			0.24535	0.46190			0.24059
160	0.66796	122.39	133.51	0.25565	0.59513	122.09	133.10	0.25310	0.48563	121.47	132.25	0.24853
180	0.69629	126.63	138.22	0.26313	0.62122	126.36	137.85	0.26065	0.50844	125.80	137.09	0.25621
200	0.72399	130.92	142.98	0.27045	0.64667	130.68	142.64	0.26802	0.53054	130.18	141.96	0.26370
220	0.75119	135.28	147.79	0.27763	0.67158	135.05	147.48	0.27525	0.55206	134.60	146.86	0.27102
240	0.77796	139.70	152.66	0.28469	0.69605	139.50	152.38	0.28234	0.57312	139.08	151.80	0.27819
260		144.19		0.29164	0.72016			0.28932	0.59379			0.28523
280		148.75		0.29849	0.74396		162.34	0.29620	0.61413			0.29216
300		153.39		0.30524	0.76749		167.42	0.30297	0.63420			0.29898
320	0.88195	158.09	172.78	0.31191	0.79079	157.94	172.57	0.30966	0.65402	157.62	172.15	0.30571
	P = 14	10 psia (<i>1</i>	$s_{sat} = 100$).51°F)	P = 1	60 psia ($T_{\rm sat} = 109$	9.50°F)	P = 1	80 psia ($T_{\rm sat} = 117$	7.69°F)
Sat.	0.33800	107.52	116.28	0.21883	0.29339	108.51	117.20	0.21840	0.25833	109.38	117.98	0.21799
120	0.36243	111.97	121.36	0.22775	0.30578	111.01	120.07	0.22339	0.26083	109.95	118.64	0.21912
140	0.38551	116.42	126.40	0.23630	0.32774	115.63	125.33	0.23232	0.28231	114.78	124.18	0.22852
160	0.40711	120.82	131.37	0.24444	0.34790	120.14	130.44	0.24070	0.30154	119.43	129.47	0.23720
180		125.23			0.36686				0.31936			0.24542
200		129.66		0.25990	0.38494			0.25647	0.33619			0.25332
220				0.26731	0.40234			0.26399	0.35228			0.26095
240				0.27457	0.41921			0.27133	0.36779			0.26838
260		143.22		0.28168	0.43564			0.27851	0.38284			0.27564
280		147.85 152.55		0.28866 0.29553	0.45171 0.46748		160.86	0.28555	0.39751 0.41186			0.28275
300 320					0.48299		166.05 171.29	0.29248 0.29929	0.41166			0.28972
340		157.31 162.14		0.30230 0.30898	0.48299			0.29929	0.42394			0.29658 0.30333
360		167.03		0.30898	0.43628		181.95	0.30000	0.45347			0.30333
300												
•		00 psia (7				00 psia (00 psia (
Sat.				0.21757	0.142/9	112.61	120.54	0.21517	0.09643	113.36	120.50	0.21164
140				0.22483	0.14656	112.00	101.00	0.01747				
160		118.67		0.23386 0.24231	0.14656 0.16355			0.21747	0.00650	112 42	120 56	0.01174
180 200		123.36 128.01		0.24231	0.16333		134.66	0.22803 0.23734	0.09658 0.11440			0.21174 0.22473
220		132.65		0.25037	0.17776			0.23734	0.11440			0.22473
240		137.31		0.26566	0.19044			0.24396	0.12746			0.23302
260	0.32038			0.20300	0.20211		151.60	0.25412	0.13833			0.25272
280		146.73		0.27300	0.21300		157.11	0.26949	0.14844			0.26079
300		151.50		0.28720	0.23346		162.62	0.27683	0.16611			0.26853
320		156.34		0.29410	0.24310		168.13	0.28399	0.17423			0.27601
340		161.23		0.30089	0.25246		173.66	0.29100	0.17423			0.28328
360		166.18			0.26159				0.18951			0.29037

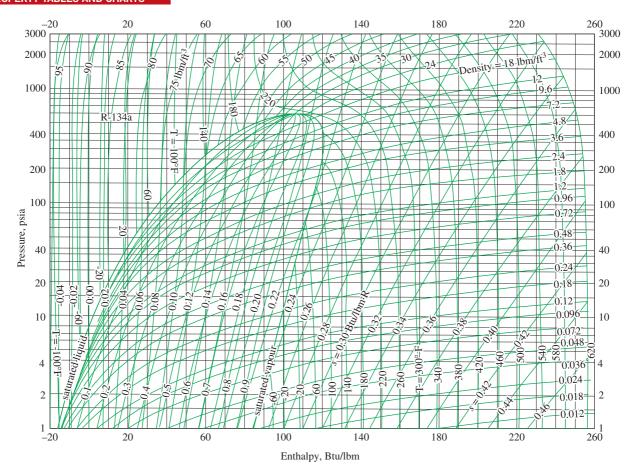


FIGURE A–14E *P-h* diagram for refrigerant-134a.

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TABLE A-16E

Properties of the atmosphere at high altitude

Altitude, ft	Temperature, °F	Pressure, psia	Gravity, g, ft/s ²	Speed of sound, ft/s	Density, Ibm/ft ³	Viscosity μ, lbm/ft·s	Thermal conductivity, Btu/h·ft·R
0	59.00	14.7	32.174	1116	0.07647	1.202×10^{-5}	0.0146
500	57.22	14.4	32.173	1115	0.07536	1.199×10^{-5}	0.0146
1000	55.43	14.2	32.171	1113	0.07426	1.196×10^{-5}	0.0146
1500	53.65	13.9	32.169	1111	0.07317	1.193×10^{-5}	0.0145
2000	51.87	13.7	32.168	1109	0.07210	1.190×10^{-5}	0.0145
2500	50.09	13.4	32.166	1107	0.07104	1.186×10^{-5}	0.0144
3000	48.30	13.2	32.165	1105	0.06998	1.183×10^{-5}	0.0144
3500	46.52	12.9	32.163	1103	0.06985	1.180×10^{-5}	0.0143
4000	44.74	12.7	32.162	1101	0.06792	1.177×10^{-5}	0.0143
4500	42.96	12.5	32.160	1099	0.06690	1.173×10^{-5}	0.0142
5000	41.17	12.2	32.159	1097	0.06590	1.170×10^{-5}	0.0142
5500	39.39	12.0	32.157	1095	0.06491	1.167×10^{-5}	0.0141
6000	37.61	11.8	32.156	1093	0.06393	1.164×10^{-5}	0.0141
6500	35.83	11.6	32.154	1091	0.06296	1.160×10^{-5}	0.0141
7000	34.05	11.3	32.152	1089	0.06200	1.157×10^{-5}	0.0140
7500	32.26	11.1	32.151	1087	0.06105	1.154×10^{-5}	0.0140
8000	30.48	10.9	32.149	1085	0.06012	1.150×10^{-5}	0.0139
8500	28.70	10.7	32.148	1083	0.05919	1.147×10^{-5}	0.0139
9000	26.92	10.5	32.146	1081	0.05828	1.144×10^{-5}	0.0138
9500	25.14	10.3	32.145	1079	0.05738	1.140×10^{-5}	0.0138
10,000	23.36	10.1	32.145	1077	0.05648	1.137×10^{-5}	0.0137
11,000	19.79	9.72	32.140	1073	0.05473	1.130×10^{-5}	0.0136
12,000	16.23	9.34	32.137	1069	0.05302	1.124×10^{-5}	0.0136
13,000	12.67	8.99	32.134	1065	0.05135	1.117×10^{-5}	0.0135
14,000	9.12	8.63	32.131	1061	0.04973	1.110×10^{-5}	0.0134
15,000	5.55	8.29	32.128	1057	0.04814	1.104×10^{-5}	0.0133
16,000	+1.99	7.97	32.125	1053	0.04659	1.097×10^{-5}	0.0132
17,000	-1.58	7.65	32.122	1049	0.04508	1.090×10^{-5}	0.0132
18,000	-5.14	7.34	32.119	1045	0.04361	1.083×10^{-5}	0.0130
19,000	-8.70	7.05	32.115	1041	0.04217	1.076×10^{-5}	0.0129
20,000	-12.2	6.76	32.112	1037	0.04077	1.070×10^{-5}	0.0128
22,000	-19.4	6.21	32.106	1029	0.03808	1.056×10^{-5}	0.0126
24,000	-26.5	5.70	32.100	1020	0.03553	1.042×10^{-5}	0.0124
26,000	-33.6	5.22	32.094	1012	0.03311	1.028×10^{-5}	0.0122
28,000	-40.7	4.78	32.088	1003	0.03082	1.014×10^{-5}	0.0121
30,000	-47.8	4.37	32.082	995	0.02866	1.000×10^{-5}	0.0119
32,000	-54.9	3.99	32.08	987	0.02661	0.986×10^{-5}	0.0117
34,000	-62.0	3.63	32.07	978	0.02468	0.971×10^{-5}	0.0115
36,000	-69.2	3.30	32.06	969	0.02285	0.956×10^{-5}	0.0113
38,000	-69.7	3.05	32.06	968	0.02079	0.955×10^{-5}	0.0113
40,000	-69.7	2.73	32.05	968	0.01890	0.955×10^{-5}	0.0113
45,000	-69.7	2.148	32.04	968	0.01487	0.955×10^{-5}	0.0113
50,000	-69.7	1.691	32.02	968	0.01171	0.955×10^{-5}	0.0113
55,000	-69.7	1.332	32.00	968	0.00922	0.955×10^{-5}	0.0113
60,000	-69.7	1.048	31.99	968	0.00726	0.955×10^{-5}	0.0113

Source of Data: U.S. Standard Atmosphere Supplements, U.S. Government Printing Office, 1966. Based on year-round mean conditions at 45° latitude and varies with the time of the year and the weather patterns. The conditions at sea level (z=0) are taken to be P=14.696 psia, $T=59^{\circ}$ F, $\rho=0.076474$ lbm/ft³, g=32.1741 ft²/s.

TABLE A-17E

Ideal-gas properties of air

T R	h Btu/Ibm	И	om v _r	s° Btu/Ibm∙R	<i>T</i>	<i>h</i> Btu/lbm	P_r	u Btu/lbm	V_r	s° Btu/lbm⋅R
360 380 400 420 440	85.97 90.75 95.53 100.32 105.11	0.3363 61. 0.4061 64. 0.4858 68. 0.5760 71.	29 396.6 70 346.6 11 305.0 52 270.1 93 240.6	0.50369 0.51663 0.52890 0.54058 0.55172	1600 1650 1700 1750 1800	395.74 409.13 422.59 436.12 449.71	71.13 80.89 90.95 101.98 114.0	286.06 296.03 306.06 316.16 326.32	8.263 7.556 6.924 6.357 5.847	0.87130 0.87954 0.88758 0.89542 0.90308
460 480 500 520 537 540	109.90 114.69 119.48 124.27 128.10 129.06	0.9182 81. 1.0590 85. 1.2147 88. 1.3593 91.	36 215.33 77 193.65 20 174.90 52 158.58 53 146.34 04 144.32	0.56235 0.57255 0.58233 0.59173 0.59945 0.60078	1850 1900 1950 2000 2050 2100	463.37 477.09 490.88 504.71 518.71 532.55	127.2 141.5 157.1 174.0 192.3 212.1	336.55 346.85 357.20 367.61 378.08 388.60	5.388 4.974 4.598 4.258 3.949 3.667	0.91056 0.91788 0.92504 0.93205 0.93891 0.94564
560 580 600 620 640	133.86 138.66 143.47 148.28 153.09	1.7800 98. 2.005 102.	78 102.12	0.60950 0.61793 0.62607 0.63395 0.64159	2150 2200 2250 2300 2350	546.54 560.59 574.69 588.82 603.00	223.5 256.6 281.4 308.1 336.8	399.17 409.78 420.46 431.16 441.91	3.410 3.176 2.961 2.765 2.585	0.95222 0.95919 0.96501 0.97123 0.97732
660 680 700 720 740	157.92 162.73 167.56 172.39 177.23	2.801 112. 3.111 116. 3.446 119. 3.806 123. 4.193 126.	12 80.96 58 75.25 04 70.07	0.64902 0.65621 0.66321 0.67002 0.67665	2400 2450 2500 2550 2600	617.22 631.48 645.78 660.12 674.49	367.6 400.5 435.7 473.3 513.5	452.70 463.54 474.40 485.31 496.26	2.419 2.266 2.125 1.996 1.876	0.98331 0.98919 0.99497 1.00064 1.00623
760 780 800 820 840	182.08 186.94 191.81 196.69 201.56	4.607 129. 5.051 133. 5.526 136. 6.033 140. 6.573 143.	57.20 53.63 50.35	0.68312 0.68942 0.69558 0.70160 0.70747	2650 2700 2750 2800 2850	688.90 703.35 717.83 732.33 746.88	556.3 601.9 650.4 702.0 756.7	507.25 518.26 529.31 540.40 551.52	1.765 1.662 1.566 1.478 1.395	1.01172 1.01712 1.02244 1.02767 1.03282
860 880 900 920 940	206.46 211.35 216.26 221.18 226.11	7.149 147. 7.761 151. 8.411 154. 9.102 158. 9.834 161.	02 42.01 57 39.64 12 37.44	0.71323 0.71886 0.72438 0.72979 0.73509	2900 2950 3000 3050 3100	761.45 776.05 790.68 805.34 820.03	814.8 876.4 941.4 1011 1083	562.66 573.84 585.04 596.28 607.53	1.318 1.247 1.180 1.118 1.060	1.03788 1.04288 1.04779 1.05264 1.05741
960 980 1000 1040 1080	231.06 236.02 240.98 250.95 260.97	10.61 165. 11.43 168. 12.30 172. 14.18 179. 16.28 186.	33 31.76 43 30.12 56 27.17	0.74030 0.74540 0.75042 0.76019 0.76964	3150 3200 3250 3300 3350	834.75 849.48 864.24 879.02 893.83	1161 1242 1328 1418 1513	618.82 630.12 641.46 652.81 664.20		1.06212 1.06676 1.07134 1.07585 1.08031
1120 1160 1200 1240 1280	271.03 281.14 291.30 301.52 311.79	18.60 194. 21.18 201. 24.01 209. 27.13 216. 30.55 224.	20.29 5 18.51 5 16.93	0.77880 0.78767 0.79628 0.80466 0.81280	3400 3450 3500 3550 3600	908.66 923.52 938.40 953.30 968.21			0.7436 0.7087 0.6759	1.08470 1.08904 1.09332 1.09755 1.10172
1320 1360 1400 1440 1480	322.11 332.48 342.90 353.37 363.89	34.31 231. 38.41 239. 42.88 246. 47.75 254. 53.04 262.	25 13.12 93 12.10 66 11.17	0.82075 0.82848 0.83604 0.84341 0.85062	3800	983.15 998.11 1013.1 1028.1 1043.1	2196 2330 2471 2618 2773	732.95 744.48 756.04 767.60 779.19	0.5882 0.5621 0.5376	1.10584 1.10991 1.11393 1.11791 1.12183
1520 1560	374.47 385.08	58.78 270. 65.00 278.		3 0.85767 0 0.86456		1058.1 1073.2	2934 3103	790.80 802.43		1.12571 1.12955

TABLE A-17E	-17E	Α	LΕ	В	A	T.	
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Ideal-gas properties of air (Concluded)

T R	<i>h</i> Btu/lbm	P_r	<i>u</i> Btu/lbm	V _r	s° Btu/lbm⋅R	<i>T</i> R	<i>h</i> Btu/lbm	P_r	<i>u</i> Btu/lbm	V _r	s° Btu/lbm⋅R
4000	1088.3	3280	814.06	0.4518	1.13334	4600	1270.4	6089	955.04	0.2799	1.17575
4050 4100 4150 4200 4300	1133.6 1148.7	3464 3656 3858 4067 4513	825.72 837.40 849.09 860.81 884.28	0.4331 0.4154 0.3985 0.3826 0.3529	1.13709 1.14079 1.14446 1.14809 1.15522	4700 4800 4900 5000 5100	1300.9 1331.5 1362.2 1392.9 1423.6	6701 7362 8073 8837 9658	978.73 1002.5 1026.3 1050.1 1074.0	0.2415 0.2248 0.2096	1.18232 1.18876 1.19508 1.20129 1.20738
4400 4500	1209.4 1239.9	4997 5521	907.81 931.39	0.3262 0.3019	1.16221 1.16905	5200 5300	1454.4 1485.3	10,539 11,481	1098.0 1122.0		1.21336 1.2192

Note: The properties P_r (relative pressure) and v_r (relative specific volume) are dimensionless quantities used in the analysis of isentropic processes, and should not be confused with the properties pressure and specific volume.

Source of Data: Kenneth Wark, Thermodynamics, 4th ed. (New York: McGraw-Hill, 1983), pp. 832–33, Table A–5. Originally published in J. H. Keenan and J. Kaye, Gas Tables (New York: John Wiley & Sons, 1948).

TABLE A-18E

Ideal-gas properties of nitrogen, N₂

T R	h Btu/lbmol	\overline{u} Btu/Ibmol	<i>s</i> ° Btu/lbmol⋅R	T R		ū Btu/lbmol	\overline{s}° Btu/lbmol·R
300	2,082.0	1,486.2	41.695	1080	7,551.0	5,406.2	50.651
320	2,221.0	1,585.5	42.143	1100	7,695.0	5,510.5	50.783
340	2,360.0	1,684.4	42.564	1120	7,839.3	5,615.2	50.912
360	2,498.9	1,784.0	42.962	1140	7,984.0	5,720.1	51.040
380	2,638.0	1,883.4	43.337	1160	8,129.0	5,825.4	51.167
400	2,777.0	1,982.6	43.694	1180	8,274.4	5,931.0	51.291
420	2,916.1	2,082.0	44.034	1200	8,420.0	6,037.0	51.143
440	3,055.1	2,181.3	44.357	1220	8,566.1	6,143.4	51.534
460	3,194.1	2,280.6	44.665	1240	8,712.6	6,250.1	51.653
480	3,333.1	2,379.9	44.962	1260	8,859.3	6,357.2	51.771
500	3,472.2	2,479.3	45.246	1280	9,006.4	6,464.5	51.887
520	3,611.3	2,578.6	45.519	1300	9,153.9	6,572.3	51.001
537	3,729.5	2,663.1	45.743	1320	9,301.8	6,680.4	52.114
540	3,750.3	2,678.0	45.781	1340	9,450.0	6,788.9	52.225
560	3,889.5	2,777.4	46.034	1360	9,598.6	6,897.8	52.335
580	4,028.7	2,876.9	46.278	1380	9,747.5	7,007.0	52.444
600	4,167.9	2,976.4	46.514	1400	9,896.9	7,116.7	52.551
620	4,307.1	3,075.9	46.742	1420	10,046.6	7,226.7	52.658
640	4,446.4	3,175.5	46.964	1440	10,196.6	7,337.0	52.763
660	4,585.8	3,275.2	47.178	1460	10,347.0	7,447.6	52.867
680	4,725.3	3,374.9	47.386	1480	10,497.8	7,558.7	52.969
700	4,864.9	3,474.8	47.588	1500	10,648.0	7,670.1	53.071
720	5,004.5	3,574.7	47.785	1520	10,800.4	7,781.9	53.171
740	5,144.3	3,674.7	47.977	1540	10,952.2	7,893.9	53.271
760	5,284.1	3,774.9	48.164	1560	11,104.3	8,006.4	53.369
780	5,424.2	3,875.2	48.345	1580	11,256.9	8,119.2	53.465
800	5,564.4	3,975.7	48.522	1600	11,409.7	8,232.3	53.561
820	5,704.7	4,076.3	48.696	1620	11,562.8	8,345.7	53.656
840	5,845.3	4,177.1	48.865	1640	11,716.4	8,459.6	53.751
860	5,985.9	4,278.1	49.031	1660	11,870.2	8,573.6	53.844
880	6,126.9	4,379.4	49.193	1680	12,024.3	8,688.1	53.936
900	6,268.1	4,480.8	49.352	1700	12,178.9	8,802.9	54.028
920	6,409.6	4,582.6	49.507	1720	12,333.7	8,918.0	54.118
940	6,551.2	4,684.5	49.659	1740	12,488.8	9,033.4	54.208
960	6,693.1	4,786.7	49.808	1760	12,644.3	9,149.2	54.297
980	6,835.4	4,889.3	49.955	1780	12,800.2	9,265.3	54.385
1000	6,977.9	4,992.0	50.099	1800	12,956.3	9,381.7	54.472
1020	7,120.7	5,095.1	50.241	1820	13,112.7	9,498.4	54.559
1040	7,263.8	5,198.5	50.380	1840	13,269.5	9,615.5	54.645
1060	7,407.2	5,302.2	50.516	1860	13,426.5	9,732.8	54.729

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Ideal-gas properties of nitrogen, N₂ (Concluded)

<i>T</i>	<i>h</i>	<i>ū</i>	<i>s</i> °	<i>T</i>	<i>h</i>	ū	ਤ°
R	Btu∕lbmol	Btu∕lbmol	Btu/lbmol∙R	R	Btu∕lbmol	Btu∕lbmol	Btu/lbmol∙R
1900	13,742	9,968	54.896	3500	27,016	20,065	59.944
1940	14,058	10,205	55.061	3540	27,359	20,329	60.041
1980	14,375	10,443	55.223	3580	27,703	20,593	60.138
2020	14,694	10,682	55.383	3620	28,046	20,858	60.234
2060	15,013	10,923	55.540	3660	28,391	21,122	60.328
2100	15,334	11,164	55.694	3700	28,735	21,387	60.422
2140	15,656	11,406	55.846	3740	29,080	21,653	60.515
2180	15,978	11,649	55.995	3780	29,425	21,919	60.607
2220	16,302	11,893	56.141	3820	29,771	22,185	60.698
2260	16,626	12,138	56.286	3860	30,117	22,451	60.788
2300	16,951	12,384	56.429	3900	30,463	22,718	60.877
2340	17,277	12,630	56.570	3940	30,809	22,985	60.966
2380	17,604	12,878	56.708	3980	31,156	23,252	61.053
2420	17,392	13,126	56.845	4020	31,503	23,520	61.139
2460	18,260	13,375	56.980	4060	31,850	23,788	61.225
2500	18,590	13,625	57.112	4100	32,198	24,056	61.310
2540	18,919	13,875	57.243	4140	32,546	24,324	61.395
2580	19,250	14,127	57.372	4180	32,894	24,593	61.479
2620	19,582	14,379	57.499	4220	33,242	24,862	61.562
2660	19,914	14,631	57.625	4260	33,591	25,131	61.644
2700	20,246	14,885	57.750	4300	33,940	25,401	61.726
2740	20,580	15,139	57.872	4340	34,289	25,670	61.806
2780	20,914	15,393	57.993	4380	34,638	25,940	61.887
2820	21,248	15,648	58.113	4420	34,988	26,210	61.966
2860	21,584	15,905	58.231	4460	35,338	26,481	62.045
2900	21,920	16,161	58.348	4500	35,688	26,751	62.123
2940	22,256	16,417	58.463	4540	36,038	27,022	62.201
2980	22,593	16,675	58.576	4580	36,389	27,293	62.278
3020	22,930	16,933	58.688	4620	36,739	27,565	62.354
3060	23,268	17,192	58.800	4660	37,090	27,836	62.429
3100	23,607	17,451	58.910	4700	37,441	28,108	62.504
3140	23,946	17,710	59.019	4740	37,792	28,379	62.578
3180	24,285	17,970	59.126	4780	38,144	28,651	62.652
3220	24,625	18,231	59.232	4820	38,495	28,924	62.725
3260	24,965	18,491	59.338	4860	38,847	29,196	62.798
3300	25,306	18,753	59.442	4900	39,199	29,468	62.870
3340	25,647	19,014	59.544	5000	40,080	30,151	63.049
3380	25,989	19,277	59.646	5100	40,962	30,834	63.223
3420	26,331	19,539	59.747	5200	41,844	31,518	63.395
3460	26,673	19,802	59.846	5300	42,728	32,203	63.563

Source of Data: Tables A-18E through A-23E are adapted from Kenneth Wark, Thermodynamics, 4th ed. (New York: McGraw-Hill, 1983), pp. 834-44. Originally published in J. H. Keenan and J. Kaye, Gas Tables (New York: John Wiley & Sons, 1945).

TABLE A-19E

Ideal-gas properties of oxygen, O₂

T	$\frac{\overline{h}}{h}$	\overline{u}	\overline{s}°	Т	\overline{h}	\overline{u}	¯§°
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R
300	2,073.5	1,477.8	44.927	1080	7,696.8	5,552.1	54.064
320	2,212.6	1,577.1	45.375	1100	7,850.4	5,665.9	54.204
340	2,351.7	1,676.5	45.797	1120	8,004.5	5,780.3	54.343
360	2,490.8	1,775.9	46.195	1140	8,159.1	5,895.2	54.480
380	2,630.0	1,875.3	46.571	1160	8,314.2	6,010.6	54.614
400	2,769.1	1,974.8	46.927	1180	8,469.8	6,126.5	54.748
420	2,908.3	2,074.3	47.267	1200	8,625.8	6,242.8	54.879
440	3,047.5	2,173.8	47.591	1220	8,782.4	6,359.6	55.008
460	3,186.9	2,273.4	47.900	1240	8,939.4	6,476.9	55.136
480	3,326.5	2,373.3	48.198	1260	9,096.7	6,594.5	55.262
500	3,466.2	2,473.2	48.483	1280	9,254.6	6,712.7	55.386
520	3,606.1	2,573.4	48.757	1300	9,412.9	6,831.3	55.508
537	3,725.1	2,658.7	48.982	1320	9,571.9	6,950.2	55.630
540	3,746.2	2,673.8	49.021	1340	9,730.7	7,069.6	55.750
560	3,886.6	2,774.5	49.276	1360	9,890.2	7,189.4	55.867
580	4,027.3	2,875.5	49.522	1380	10,050.1	7,309.6	55.984
600	4,168.3	2,976.8	49.762	1400	10,210.4	7,430.1	56.099
620	4,309.7	3,078.4	49.993	1420	10,371.0	7,551.1	56.213
640	4,451.4	3,180.4	50.218	1440	10,532.0	7,672.4	56.326
660	4,593.5	3,282.9	50.437	1460	10,693.3	7,793.9	56.437
680	4,736.2	3,385.8	50.650	1480	10,855.1	7,916.0	56.547
700	4,879.3	3,489.2	50.858	1500	11,017.1	8,038.3	56.656
720	5,022.9	3,593.1	51.059	1520	11,179.6	8,161.1	56.763
740	5,167.0	3,697.4	51.257	1540	11,342.4	8,284.2	56.869
760	5,311.4	3,802.4	51.450	1560	11,505.4	8,407.4	56.975
780	5,456.4	3,907.5	51.638	1580	11,668.8	8,531.1	57.079
800	5,602.0	4,013.3	51.821	1600	11,832.5	8,655.1	57.182
820	5,748.1	4,119.7	52.002	1620	11,996.6	8,779.5	57.284
840	5,894.8	4,226.6	52.179	1640	12,160.9	8,904.1	57.385
860	6,041.9	4,334.1	52.352	1660	12,325.5	9,029.0	57.484
880	6,189.6	4,442.0	52.522	1680	12,490.4	9,154.1	57.582
900	6,337.9	4,550.6	52.688	1700	12,655.6	9,279.6	57.680
920	6,486.7	4,659.7	52.852	1720	12,821.1	9,405.4	57.777
940	6,636.1	4,769.4	53.012	1740	12,986.9	9,531.5	57.873
960	6,786.0	4,879.5	53.170	1760	13,153.0	9,657.9	57.968
980	6,936.4	4,990.3	53.326	1780	13,319.2	9,784.4	58.062
1000	7,087.5	5,101.6	53.477	1800	13,485.8	9,911.2	58.155
1020	7,238.9	5,213.3	53.628	1820	13,652.5	10,038.2	58.247
1040	7,391.0	5,325.7	53.775	1840	13,819.6	10,165.6	58.339
1060	7,543.6	5,438.6	53.921	1860	13,986.8	10,293.1	58.428

TABLE A-	

Ideal-gas properties of oxygen, O₂ (Concluded)

T	$\frac{\overline{h}}{\overline{h}}$	\overline{u} oxygen, O_2 (Cor	<u></u>	Т		ū	
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol∙R
1900	14,322	10,549	58.607	3500	28,273	21,323	63.914
1940	14,658	10,806	58.782	3540	28,633	21,603	64.016
1980	14,995	11,063	58.954	3580	28,994	21,884	64.114
2020	15,333	11,321	59.123	3620	29,354	22,165	64.217
2060	15,672	11,581	59.289	3660	29,716	22,447	64.316
2100	16,011	11,841	59.451	3700	30,078	22,730	64.415
2140	16,351	12,101	59.612	3740	30,440	23,013	64.512
2180	16,692	12,363	59.770	3780	30,803	23,296	64.609
2220	17,036	12,625	59.926	3820	31,166	23,580	64.704
2260	17,376	12,888	60.077	3860	31,529	23,864	64.800
2300	17,719	13,151	60.228	3900	31,894	24,149	64.893
2340	18,062	13,416	60.376	3940	32,258	24,434	64.986
2380	18,407	13,680	60.522	3980	32,623	24,720	65.078
2420	18,572	13,946	60.666	4020	32,989	25,006	65.169
2460	19,097	14,212	60.808	4060	33,355	25,292	65.260
2500	19,443	14,479	60.946	4100	33,722	25,580	65.350
2540	19,790	14,746	61.084	4140	34,089	25,867	64.439
2580	20,138	15,014	61.220	4180	34,456	26,155	65.527
2620	20,485	15,282	61.354	4220	34,824	26,144	65.615
2660	20,834	15,551	61.486	4260	35,192	26,733	65.702
2700	21,183	15,821	61.616	4300	35,561	27,022	65.788
2740	21,533	16,091	61.744	4340	35,930	27,312	65.873
2780	21,883	16,362	61.871	4380	36,300	27,602	65.958
2820	22,232	16,633	61.996	4420	36,670	27,823	66.042
2860	22,584	16,905	62.120	4460	37,041	28,184	66.125
2900	22,936	17,177	62.242	4500	37,412	28,475	66.208
2940	23,288	17,450	62.363	4540	37,783	28,768	66.290
2980	23,641	17,723	62.483	4580	38,155	29,060	66.372
3020	23,994	17,997	62.599	4620	38,528	29,353	66.453
3060	24,348	18,271	62.716	4660	38,900	29,646	66.533
3100	24,703	18,546	62.831	4700	39,274	29,940	66.613
3140	25,057	18,822	62.945	4740	39,647	30,234	66.691
3180	25,413	19,098	63.057	4780	40,021	30,529	66.770
3220	25,769	19,374	63.169	4820	40,396	30,824	66.848
3260	26,175	19,651	63.279	4860	40,771	31,120	66.925
3300	26,412	19,928	63.386	4900	41,146	31,415	67.003
3340	26,839	20,206	63.494	5000	42,086	32,157	67.193
3380	27,197	20,485	63.601	5100	43,021	32,901	67.380
3420	27,555	20,763	63.706	5200	43,974	33,648	67.562
3460	27,914	21,043	63.811	5300	44,922	34,397	67.743

TABLE A-20E

Ideal-gas properties of carbon dioxide, CO₂

T	ħ	ū	¯S°	Т	h	ū	₹°
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R
300	2,108.2	1,512.4	46.353	1080	9,575.8	7,431.1	58.072
320	2,256.6	1,621.1	46.832	1100	9,802.6	7,618.1	58.281
340	2,407.3	1,732.1	47.289	1120	10,030.6	7,806.4	58.485
360	2,560.5	1,845.6	47.728	1140	10,260.1	7,996.2	58.689
380	2,716.4	1,961.8	48.148	1160	10,490.6	8,187.0	58.889
400	2,874.7	2,080.4	48.555	1180	10,722.3	8,379.0	59.088
420	3,035.7	2,201.7	48.947	1200	10,955.3	8,572.3	59.283
440	3,199.4	2,325.6	49.329	1220	11,189.4	8,766.6	59.477
460	3,365.7	2,452.2	49.698	1240	11,424.6	8,962.1	59.668
480	3,534.7	2,581.5	50.058	1260	11,661.0	9,158.8	59.858
500	3,706.2	2,713.3	50.408	1280	11,898.4	9,356.5	60.044
520	3,880.3	2,847.7	50.750	1300	12,136.9	9,555.3	60.229
537	4,027.5	2,963.8	51.032	1320	12,376.4	9,755.0	60.412
540	4,056.8	2,984.4	51.082	1340	12,617.0	9,955.9	60.593
560	4,235.8	3,123.7	51.408	1360	12,858.5	10,157.7	60.772
580	4,417.2	3,265.4	51.726	1380	13,101.0	10,360.5	60.949
600	4,600.9	3,409.4	52.038	1400	13,344.7	10,564.5	61.124
620	4,786.6	3,555.6	52.343	1420	13,589.1	10,769.2	61.298
640	4,974.9	3,704.0	52.641	1440	13,834.5	10,974.8	61.469
660	5,165.2	3,854.6	52.934	1460	14,080.8	11,181.4	61.639
680	5,357.6	4,007.2	53.225	1480	14,328.0	11,388.9	61.800
700	5,552.0	4,161.9	53.503	1500	14,576.0	11,597.2	61.974
720	5,748.4	4,318.6	53.780	1520	14,824.9	11,806.4	62.138
740	5,946.8	4,477.3	54.051	1540	15,074.7	12,016.5	62.302
760	6,147.0	4,637.9	54.319	1560	15,325.3	12,227.3	62.464
780	6,349.1	4,800.1	54.582	1580	15,576.7	12,439.0	62.624
800	6,552.9	4,964.2	54.839	1600	15,829.0	12,651.6	62.783
820	6,758.3	5,129.9	55.093	1620	16,081.9	12,864.8	62.939
840	6,965.7	5,297.6	55.343	1640	16,335.7	13,078.9	63.095
860	7,174.7	5,466.9	55.589	1660	16,590.2	13,293.7	63.250
880	7,385.3	5,637.7	55.831	1680	16,845.5	13,509.2	63.403
900	7,597.6	5,810.3	56.070	1700	17,101.4	13,725.4	63.555
920	7,811.4	5,984.4	56.305	1720	17,358.1	13,942.4	63.704
940	8,026.8	6,160.1	56.536	1740	17,615.5	14,160.1	63.853
960	8,243.8	6,337.4	56.765	1760	17,873.5	14,378.4	64.001
980	8,462.2	6,516.1	56.990	1780	18,132.2	14,597.4	64.147
1000	8,682.1	6,696.2	57.212	1800	18,391.5	14,816.9	64.292
1020	8,903.4	6,877.8	57.432	1820	18,651.5	15,037.2	64.435
1040	9,126.2	7,060.9	57.647	1840	18,912.2	15,258.2	64.578
1060	9,350.3	7,245.3	57.861	1860	19,173.4	15,479.7	64.719

	1-2	

Ideal-gas properties of carbon dioxide, CO₂ (Concluded)

		carbon dioxide,	CO ₂ (Concluded)				
T	Ī	\overline{u}	$\overline{\mathcal{S}}^{\circ}$	T	h	\overline{u}	<u></u> s°
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R
1900	19,698	15,925	64.999	3500	41,965	35,015	73.462
1940	20,224	16,372	65.272	3540	42,543	35,513	73.627
1980	20,753	16,821	65.543	3580	43,121	36,012	73.789
2020	21,284	17,273	65.809	3620	43,701	36,512	73.951
2060	21,818	17,727	66.069	3660	44,280	37,012	74.110
2100	22,353	18,182	66.327	3700	44,861	37,513	74.267
2140	22,890	18,640	66.581	3740	45,442	38,014	74.423
2180	23,429	19,101	66.830	3780	46,023	38,517	74.578
2220	23,970	19,561	67.076	3820	46,605	39,019	74.732
2260	24,512	20,024	67.319	3860	47,188	39,522	74.884
2300	25,056	20,489	67.557	3900	47,771	40,026	75.033
2340	25,602	20,955	67.792	3940	48,355	40,531	75.182
2380	26,150	21,423	68.025	3980	48,939	41,035	75.330
2420	26,699	21,893	68.253	4020	49,524	41,541	75.477
2460	27,249	22,364	68.479	4060	50,109	42,047	75.622
2500	27,801	22,837	68.702	4100	50,695	42,553	75.765
2540	28,355	23,310	68.921	4140	51,282	43,060	75.907
2580	28,910	23,786	69.138	4180	51,868	43,568	76.048
2620	29,465	24,262	69.352	4220	52,456	44,075	76.188
2660	30,023	24,740	69.563	4260	53,044	44,584	76.327
2700	30,581	25,220	69.771	4300	53,632	45,093	76.464
2740	31,141	25,701	69.977	4340	54,221	45,602	76.601
2780	31,702	26,181	70.181	4380	54,810	46,112	76.736
2820	32,264	26,664	70.382	4420	55,400	46,622	76.870
2860	32,827	27,148	70.580	4460	55,990	47,133	77.003
2900	33,392	27,633	70.776	4500	56,581	47,645	77.135
2940	33,957	28,118	70.970	4540	57,172	48,156	77.266
2980	34,523	28,605	71.160	4580	57,764	48,668	77.395
3020	35,090	29,093	71.350	4620	58,356	49,181	77.581
3060	35,659	29,582	71.537	4660	58,948	49,694	77.652
3100	36,228	30,072	71.722	4700	59,541	50,208	77.779
3140	36,798	30,562	71.904	4740	60,134	50,721	77.905
3180	37,369	31,054	72.085	4780	60,728	51,236	78.029
3220	37,941	31,546	72.264	4820	61,322	51,750	78.153
3260	38,513	32,039	72.441	4860	61,916	52,265	78.276
3300	39,087	32,533	72.616	4900	62,511	52,781	78.398
3340	39,661	33,028	72.788	5000	64,000	54,071	78.698
3380	40,236	33,524	72.960	5100	65,491	55,363	78.994
3420	40,812	34,020	73.129	5200	66,984	56,658	79.284
3460	41,388	34,517	73.297	5300	68,471	57,954	79.569

TABLE A-21E

Ideal-gas properties of carbon monoxide, CO

<i>T</i> R	<i>h</i> Btu∕lbmol	\overline{u} Btu/lbmol	ड∙ Btu/Ibmol∙R	<i>T</i> R	<i>h</i> Btu∕lbmol	ū Btu∕lbmol	\overline{s}° Btu/Ibmol·R
300	2,081.9	1,486.1	43.223	1080	7,571.1	5,426.4	52.203
320	2,220.9	1,585.4	43.672	1100	7,716.8	5,532.3	52.337
340	2,359.9	1,684.7	44.093	1120	7,862.9	5,638.7	52.468
360	2,498.8	1,783.9	44.490	1140	8,009.2	5,745.4	52.598
380	2,637.9	1,883.3	44.866	1160	8,156.1	5,851.5	52.726
400	2,776.9	1,982.6	45.223	1180	8,303.3	5,960.0	52.852
420	2,916.0	2,081.9	45.563	1200	8,450.8	6,067.8	52.976
440	3,055.0	2,181.2	45.886	1220	8,598.8	6,176.0	53.098
460	3,194.0	2,280.5	46.194	1240	8,747.2	6,284.7	53.218
480	3,333.0	2,379.8	46.491	1260	8,896.0	6,393.8	53.337
500	3,472.1	2,479.2	46.775	1280	9,045.0	6,503.1	53.455
520	3,611.2	2,578.6	47.048	1300	9,194.6	6,613.0	53.571
537	3,725.1	2,663.1	47.272	1320	9,344.6	6,723.2	53.685
540	3,750.3	2,677.9	47.310	1340	9,494.8	6,833.7	53.799
560	3,889.5	2,777.4	47.563	1360	9,645.5	6,944.7	53.910
580	4,028.7	2,876.9	47.807	1380	9,796.6	7,056.1	54.021
600	4,168.0	2,976.5	48.044	1400	9,948.1	7,167.9	54.129
620	4,307.4	3,076.2	48.272	1420	10,100.0	7,280.1	54.237
640	4,446.9	3,175.9	48.494	1440	10,252.2	7,392.6	54.344
660	4,586.6	3,275.8	48.709	1460	10,404.8	7,505.4	54.448
680	4,726.2	3,375.8	48.917	1480	10,557.8	7,618.7	54.522
700	4,886.0	3,475.9	49.120	1500	10,711.1	7,732.3	54.665
720	5,006.1	3,576.3	49.317	1520	10,864.9	7,846.4	54.757
740	5,146.4	3,676.9	49.509	1540	11,019.0	7,960.8	54.858
760	5,286.8	3,777.5	49.697	1560	11,173.4	8,075.4	54.958
780	5,427.4	3,878.4	49.880	1580	11,328.2	8,190.5	55.056
800	5,568.2	3,979.5	50.058	1600	11,483.4	8,306.0	55.154
820	5,709.4	4,081.0	50.232	1620	11,638.9	8,421.8	55.251
840	5,850.7	4,182.6	50.402	1640	11,794.7	8,537.9	55.347
860	5,992.3	4,284.5	50.569	1660	11,950.9	8,654.4	55.411
880	6,134.2	4,386.6	50.732	1680	12,107.5	8,771.2	55.535
900	6,276.4	4,489.1	50.892	1700	12,264.3	8,888.3	55.628
920	6,419.0	4,592.0	51.048	1720	12,421.4	9,005.7	55.720
940	6,561.7	4,695.0	51.202	1740	12,579.0	9,123.6	55.811
960	6,704.9	4,798.5	51.353	1760	12,736.7	9,241.6	55.900
980	6,848.4	4,902.3	51.501	1780	12,894.9	9,360.0	55.990
1000	6,992.2	5,006.3	51.646	1800	13,053.2	9,478.6	56.078
1020	7,136.4	5,110.8	51.788	1820	13,212.0	9,597.7	56.166
1040	7,281.0	5,215.7	51.929	1840	13,371.0	9,717.0	56.253
1060	7,425.9	5,320.9	52.067	1860	13,530.2	9,836.5	56.339

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Ideal-gas properties of carbon monoxide, CO (Concluded)

T	$\frac{\overline{h}}{h}$	Ū	$\frac{\overline{s}^{\circ}}{\overline{s}}$	Т	<u> </u>	ū	<u> </u>
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol⋅R
1900	13,850	10,077	56.509	3500	27,262	20,311	61.612
1940	14,170	10,318	56.677	3540	27,608	20,576	61.710
1980	14,492	10,560	56.841	3580	27,954	20,844	61.807
2020	14,815	10,803	57.007	3620	28,300	21,111	61.903
2060	15,139	11,048	57.161	3660	28,647	21,378	61.998
2100	15,463	11,293	57.317	3700	28,994	21,646	62.093
2140	15,789	11,539	57.470	3740	29,341	21,914	62.186
2180	16,116	11,787	57.621	3780	29,688	22,182	62.279
2220	16,443	12,035	57.770	3820	30,036	22,450	62.370
2260	16,722	12,284	57.917	3860	30,384	22,719	62.461
2300	17,101	12,534	58.062	3900	30,733	22,988	62.511
2340	17,431	12,784	58.204	3940	31,082	23,257	62.640
2380	17,762	13,035	58.344	3980	31,431	23,527	62.728
2420	18,093	13,287	58.482	4020	31,780	23,797	62.816
2460	18,426	13,541	58.619	4060	32,129	24,067	62.902
2500	18,759	13,794	58.754	4100	32,479	24,337	62.988
2540	19,093	14,048	58.885	4140	32,829	24,608	63.072
2580	19,427	14,303	59.016	4180	33,179	24,878	63.156
2620	19,762	14,559	59.145	4220	33,530	25,149	63.240
2660	20,098	14,815	59.272	4260	33,880	25,421	63.323
2700	20,434	15,072	59.398	4300	34,231	25,692	63.405
2740	20,771	15,330	59.521	4340	34,582	25,934	63.486
2780	21,108	15,588	59.644	4380	34,934	26,235	63.567
2820	21,446	15,846	59.765	4420	35,285	26,508	63.647
2860	21,785	16,105	59.884	4460	35,637	26,780	63.726
2900	22,124	16,365	60.002	4500	35,989	27,052	63.805
2940	22,463	16,225	60.118	4540	36,341	27,325	63.883
2980	22,803	16,885	60.232	4580	36,693	27,598	63.960
3020	23,144	17,146	60.346	4620	37,046	27,871	64.036
3060	23,485	17,408	60.458	4660	37,398	28,144	64.113
3100	23,826	17,670	60.569	4700	37,751	28,417	64.188
3140	24,168	17,932	60.679	4740	38,104	28,691	64.263
3180	24,510	18,195	60.787	4780	38,457	28,965	64.337
3220	24,853	18,458	60.894	4820	38,811	29,239	64.411
3260	25,196	18,722	61.000	4860	39,164	29,513	64.484
3300	25,539	18,986	61.105	4900	39,518	29,787	64.556
3340	25,883	19,250	61.209	5000	40,403	30,473	64.735
3380	26,227	19,515	61.311	5100	41,289	31,161	64.910
3420	26,572	19,780	61.412	5200	42,176	31,849	65.082
3460	26,917	20,045	61.513	5300	43,063	32,538	65.252

TABLE A-22E

Ideal-gas properties of hydrogen, H₂

T	\overline{h}	\overline{u}	<u>s</u> °	T	<u>h</u>	Ū	<u>s</u> °
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R
300	2,063.5	1,467.7	27.337	1400	9,673.8	6,893.6	37.883
320	2,189.4	1,553.9	27.742	1500	10,381.5	7,402.7	38.372
340	2,317.2	1,642.0	28.130	1600	11,092.5	7,915.1	38.830
360	2,446.8	1,731.9	28.501	1700	11,807.4	8,431.4	39.264
380	2,577.8	1,823.2	28.856	1800	12,526.8	8,952.2	39.675
400	2,710.2	1,915.8	29.195	1900	13,250.9	9,477.8	40.067
420	2,843.7	2,009.6	29.520	2000	13,980.1	10,008.4	40.441
440	2,978.1	2,104.3	29.833	2100	14,714.5	10,544.2	40.799
460	3,113.5	2,200.0	30.133	2200	15,454.4	11,085.5	41.143
480	3,249.4	2,296.2	20.424	2300	16,199.8	11,632.3	41.475
500	3,386.1	2,393.2	30.703	2400	16,950.6	12,184.5	41.794
520	3,523.2	2,490.6	30.972	2500	17,707.3	12,742.6	42.104
537	3,640.3	2,573.9	31.194	2600	18,469.7	13,306.4	42.403
540	3,660.9	2,588.5	31.232	2700	19,237.8	13,876.0	42.692
560	3,798.8	2,686.7	31.482	2800	20,011.8	14,451.4	42.973
580	3,937.1	2,785.3	31.724	2900	20,791.5	15,032.5	43.247
600	4,075.6	2,884.1	31.959	3000	21,576.9	15,619.3	43.514
620	4,214.3	2,983.1	32.187	3100	22,367.7	16,211.5	43.773
640	4,353.1	3,082.1	32.407	3200	23,164.1	16,809.3	44.026
660	4,492.1	3,181.4	32.621	3300	23,965.5	17,412.1	44.273
680	4,631.1	3,280.7	32.829	3400	24,771.9	18,019.9	44.513
700	4,770.2	3,380.1	33.031	3500	25,582.9	18,632.4	44.748
720	4,909.5	3,479.6	33.226	3600	26,398.5	19,249.4	44.978
740	5,048.8	3,579.2	33.417	3700	27,218.5	19,870.8	45.203
760	5,188.1	3,678.8	33.603	3800	28,042.8	20,496.5	45.423
780	5,327.6	3,778.6	33.784	3900	28,871.1	21,126.2	45.638
800	5,467.1	3,878.4	33.961	4000	29,703.5	21,760.0	45.849
820	5,606.7	3,978.3	34.134	4100	30,539.8	22,397.7	46.056
840	5,746.3	4,078.2	34.302	4200	31,379.8	23,039.2	46.257
860	5,885.9	4,178.0	34.466	4300	32,223.5	23,684.3	46.456
880	6,025.6	4,278.0	34.627	4400	33,070.9	24,333.1	46.651
900	6,165.3	4,378.0	34.784	4500	33,921.6	24,985.2	46.842
920	6,305.1	4,478.1	34.938	4600	34,775.7	25,640.7	47.030
940	6,444.9	4,578.1	35.087	4700	35,633.0	26,299.4	47.215
960	6,584.7	4,678.3	35.235	4800	36,493.4	26,961.2	47.396
980	6,724.6	4,778.4	35.379	4900	35,356.9	27,626.1	47.574
1000	6,864.5	4,878.6	35.520	5000	38,223.3	28,294.0	47.749
1100	7,564.6	5,380.1	36.188	5100	39,092.8	28,964.9	47.921
1200	8,265.8	5,882.8	36.798	5200	39,965.1	29,638.6	48.090
1300	8,968.7	6,387.1	37.360	5300	40,840.2	30,315.1	48.257

TABLE A-23E

Ideal-gas properties of water vapor, H₂O

$\frac{10001 \text{ gas}}{T}$	$\frac{\overline{h}}{h}$	\overline{u}	<u>s</u> °	Т	<u>Б</u>		<u> </u>
R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R	R	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R
300	2,367.6	1,771.8	40.439	1080	8,768.2	6,623.5	50.854
320	2,526.8	1,891.3	40.952	1100	8,942.0	6,757.5	51.013
340	2,686.0	2,010.8	41.435	1120	9,116.4	6,892.2	51.171
360	2,845.1	2,130.2	41.889	1140	9,291.4	7,027.5	51.325
380	3,004.4	2,249.8	42.320	1160	9,467.1	7,163.5	51.478
400	3,163.8	2,369.4	42.728	1180	9,643.4	7,300.1	51.360
420	3,323.2	2,489.1	43.117	1200	9,820.4	7,437.4	51.777
440	3,482.7	2,608.9	43.487	1220	9,998.0	7,575.2	51.925
460	3,642.3	2,728.8	43.841	1240	10,176.1	7,713.6	52.070
480	3,802.0	2,848.8	44.182	1260	10,354.9	7,852.7	52.212
500	3,962.0	2,969.1	44.508	1280	10,534.4	7,992.5	52.354
520	4,122.0	3,089.4	44.821	1300	10,714.5	8,132.9	52.494
537	4,258.0	3,191.9	45.079	1320	10,895.3	8,274.0	52.631
540	4,282.4	3,210.0	45.124	1340	11,076.6	8,415.5	52.768
560	4,442.8	3,330.7	45.415	1360	11,258.7	8,557.9	52.903
580	4,603.7	3,451.9	45.696	1380	11,441.4	8,700.9	53.037
600	4,764.7	3,573.2	45.970	1400	11,624.8	8,844.6	53.168
620	4,926.1	3,694.9	46.235	1420	11,808.8	8,988.9	53.299
640	5,087.8	3,816.8	46.492	1440	11,993.4	9,133.8	53.428
660	5,250.0	3,939.3	46.741	1460	12,178.8	9,279.4	53.556
680	5,412.5	4,062.1	46.984	1480	12,364.8	9,425.7	53.682
700	5,575.4	4,185.3	47.219	1500	12,551.4	9,572.7	53.808
720	5,738.8	4,309.0	47.450	1520	12,738.8	9,720.3	53.932
740	5,902.6	4,433.1	47.673	1540	12,926.8	9,868.6	54.055
760	6,066.9	4,557.6	47.893	1560	13,115.6	10,017.6	54.117
780	6,231.7	4,682.7	48.106	1580	13,305.0	10,167.3	54.298
800	6,396.9	4,808.2	48.316	1600	13,494.4	10,317.6	54.418
820	6,562.6	4,934.2	48.520	1620	13,685.7	10,468.6	54.535
840	6,728.9	5,060.8	48.721	1640	13,877.0	10,620.2	54.653
860	6,895.6	5,187.8	48.916	1660	14,069.2	10,772.7	54.770
880	7,062.9	5,315.3	49.109	1680	14,261.9	10,925.6	54.886
900	7,230.9	5,443.6	49.298	1700	14,455.4	11,079.4	54.999
920	7,399.4	5,572.4	49.483	1720	14,649.5	11,233.8	55.113
940	7,568.4	5,701.7	49.665	1740	14,844.3	11,388.9	55.226
960	7,738.0	5,831.6	49.843	1760	15,039.8	11,544.7	55.339
980	7,908.2	5,962.0	50.019	1780	15,236.1	11,701.2	55.449
1000	8,078.2	6,093.0	50.191	1800	15,433.0	11,858.4	55.559
1020	8,250.4	6,224.8	50.360	1820	15,630.6	12,016.3	55.668
1040	8,422.4	6,357.1	50.528	1840	15,828.7	12,174.7	55.777
1060	8,595.0	6,490.0	50.693	1860	16,027.6	12,333.9	55.884

TABLE A-23E

Ideal-gas properties of water vapor, H₂O (Continued)

T	<i>h</i> Btu/Ibmol	ū	₹°	<i>T</i>	<i>h</i>	ū	\overline{s}°
R		Btu/lbmol	Btu/lbmol·R	R	Btu∕lbmol	Btu∕lbmol	Btu/Ibmol∙R
1900	16,428	12,654	56.097	3500	34,324	27,373	62.876
1940	16,830	12,977	56.307	3540	34,809	27,779	63.015
1980	17,235	13,303	56.514	3580	35,296	28,187	63.153
2020	17,643	13,632	56.719	3620	35,785	28,596	63.288
2060	18,054	13,963	56.920	3660	36,274	29,006	63.423
2100	18,467	14,297	57.119	3700	36,765	29,418	63.557
2140	18,883	14,633	57.315	3740	37,258	29,831	63.690
2180	19,301	14,972	57.509	3780	37,752	30,245	63.821
2220	19,722	15,313	57.701	3820	38,247	30,661	63.952
2260	20,145	15,657	57.889	3860	38,743	31,077	64.082
2300	20,571	16,003	58.077	3900	39,240	31,495	64.210
2340	20,999	16,352	58.261	3940	39,739	31,915	64.338
2380	21,429	16,703	58.445	3980	40,239	32,335	64.465
2420	21,862	17,057	58.625	4020	40,740	32,757	64.591
2460	22,298	17,413	58.803	4060	41,242	33,179	64.715
2500	22,735	17,771	58.980	4100	41,745	33,603	64.839
2540	23,175	18,131	59.155	4140	42,250	34,028	64.962
2580	23,618	18,494	59.328	4180	42,755	34,454	65.084
2620	24,062	18,859	59.500	4220	43,267	34,881	65.204
2660	24,508	19,226	59.669	4260	43,769	35,310	65.325
2700	24,957	19,595	59.837	4300	44,278	35,739	65.444
2740	25,408	19,967	60.003	4340	44,788	36,169	65.563
2780	25,861	20,340	60.167	4380	45,298	36,600	65.680
2820	26,316	20,715	60.330	4420	45,810	37,032	65.797
2860	26,773	21,093	60.490	4460	46,322	37,465	65.913
2900	27,231	21,472	60.650	4500	46,836	37,900	66.028
2940	27,692	21,853	60.809	4540	47,350	38,334	66.142
2980	28,154	22,237	60.965	4580	47,866	38,770	66.255
3020	28,619	22,621	61.120	4620	48,382	39,207	66.368
3060	29,085	23,085	61.274	4660	48,899	39,645	66.480
3100	29,553	23,397	61.426	4700	49,417	40,083	66.591
3140	30,023	23,787	61.577	4740	49,936	40,523	66.701
3180	30,494	24,179	61.727	4780	50,455	40,963	66.811
3220	30,967	24,572	61.874	4820	50,976	41,404	66.920
3260	31,442	24,968	62.022	4860	51,497	41,856	67.028
3300	31,918	25,365	62.167	4900	52,019	42,288	67.135
3340	32,396	25,763	62.312	5000	53,327	43,398	67.401
3380	32,876	26,164	62.454	5100	54,640	44,512	67.662
3420	33,357	26,565	62.597	5200	55,957	45,631	67.918
3460	33,839	26,968	62.738	5300	57,279	46,754	68.172

TABLE A-26E

Enthalpy of formation, Gibbs function of formation, and absolute entropy at 77°F, 1 atm

		\overline{h}_f°	\overline{g}_f°	<u></u> $ \overline{S}^{\circ} $
Substance	Formula	Btu/Ibmol	Btu/Ibmol	Btu/Ibmol·R
Carbon	C(s)	0	0	1.36
Hydrogen	$H_2(g)$	0	0	31.21
Nitrogen	$N_2(g)$	0	0	45.77
Oxygen	O ₂ (g)	0	0	49.00
Carbon monoxide	CO(<i>g</i>)	-47,540	-59,010	47.21
Carbon dioxide	CO ₂ (g)	-169,300	-169,680	51.07
Water vapor	H ₂ O(g)	-104,040	-98,350	45.11
Water	$H_2O(\ell)$	-122,970	-102,040	16.71
Hydrogen peroxide	$H_2O_2(g)$	-58,640	-45,430	55.60
Ammonia	$NH_3(g)$	-19,750	-7,140	45.97
Methane	$CH_4(g)$	-32,210	-21,860	44.49
Acetylene	$C_2H_2(g)$	+97,540	+87,990	48.00
Ethylene	$C_2H_4(g)$	+22,490	+29,306	52.54
Ethane	$C_2H_6(g)$	-36,420	-14,150	54.85
Propylene	$C_3H_6(g)$	+8,790	+26,980	63.80
Propane	$C_3H_8(g)$	-44,680	-10,105	64.51
<i>n</i> -Butane	$C_4H_{10}(g)$	-54,270	-6,760	74.11
<i>n</i> -Octane	$C_8H_{18}(g)$	-89,680	+7,110	111.55
<i>n</i> -Octane	$C_8H_{18}(\ell)$	-107,530	+2,840	86.23
<i>n</i> -Dodecane	$C_{12}H_{26}(g)$	-125,190	+21,570	148.86
Benzene	$C_6H_6(g)$	+35,680	+55,780	64.34
Methyl alcohol	CH ₃ OH(<i>g</i>)	-86,540	-69,700	57.29
Methyl alcohol	CH ₃ OH(ℓ)	-102,670	-71,570	30.30
Ethyl alcohol	$C_2H_5OH(g)$	-101,230	-72,520	67.54
Ethyl alcohol	$C_2H_5OH(\ell)$	-119,470	-75,240	38.40
Oxygen	O(<i>g</i>)	+107,210	+99,710	38.47
Hydrogen	H(<i>g</i>)	+93,780	+87,460	27.39
Nitrogen	N(<i>g</i>)	+203,340	+195,970	36.61
Hydroxyl	OH(<i>g</i>)	+16,790	+14,750	43.92

Source of Data: From JANAF, Thermochemical Tables (Midland, MI: Dow Chemical Co., 1971), Selected Values of Chemical Thermodynamic Properties, NBS Technical Note 270-3, 1968; and API Research Project 44 (Carnegie Press, 1953).

TABLE A-27E

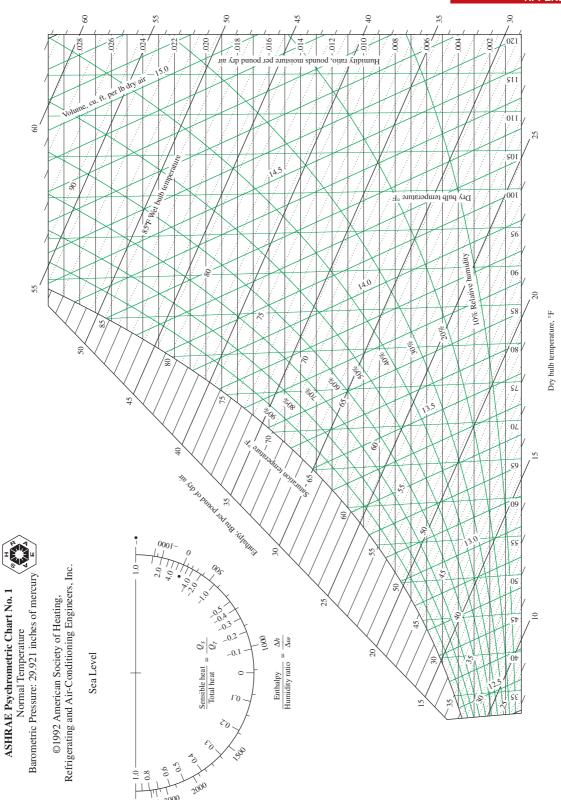
Properties of some common fuels and hydrocarbons

Fuel (phase)	Formula	Molar mass, lbm/lbmol	Density, ¹ Ibm/ft ³	Enthalpy of vaporization, ² Btu/lbm	Specific heat, c_p Btu/lbm·°F	Higher heating value, ³ Btu/lbm	Lower heating value, ³ Btu/Ibm
Carbon (s)	С	12.011	125	_	0.169	14,100	14,100
Hydrogen (g)	H_2	2.016	_	_	3.44	60,970	51,600
Carbon monoxide (g)	CÔ	28.013	_	_	0.251	4,340	4,340
Methane (g)	CH₄	16.043	_	219	0.525	23,880	21,520
Methanol (ℓ)	CH₄O	32.042	49.3	502	0.604	9,740	8,570
Acetylene (g)	C_2H_2	26.038	_	_	0.404	21,490	20,760
Ethane (g)	C_2H_6	30.070	_	74	0.418	22,320	20,430
Ethanol (ℓ)	C_2H_6O	46.069	49.3	395	0.583	12,760	11,530
Propane (ℓ)	C_3H_8	44.097	31.2	144	0.662	21,640	19,930
Butane (ℓ)	C_4H_{10}	58.123	36.1	156	0.578	21,130	19,510
1-Pentene (ℓ)	C ₅ H ₁₀	70.134	40.0	156	0.525	20,540	19,190
Isopentane (ℓ)	C ₅ H ₁₂	72.150	39.1	_	0.554	20,890	19,310
Benzene (ℓ)	C_6H_6	78.114	54.7	186	0.411	17,970	17,240
Hexene (ℓ)	C_6H_{12}	84.161	42.0	169	0.439	20,430	19,090
Hexane (ℓ)	C_6H_{14}	86.177	41.2	157	0.542	20,770	19,240
Toluene (ℓ)	C ₇ H ₈	92.141	54.1	177	0.408	18,230	17,420
Heptane (ℓ)	C ₇ H ₁₆	100.204	42.7	157	0.535	20,680	19,180
Octane (ℓ)	C ₈ H ₁₈	114.231	43.9	156	0.533	20,590	19,100
Decane (ℓ)	$C_{10}H_{22}$	142.285	45.6	155	0.528	20,490	19,020
Gasoline (ℓ)	$C_n H_{1.87n}$	100–110	45–49	151	0.57	20,300	18,900
Light diesel (ℓ)	$C_nH_{1.8n}$	170	49–52	116	0.53	19,800	18,600
Heavy diesel (ℓ)	$C_nH_{1.7n}$	200	51–55	99	0.45	19,600	18,400
Natural gas (g)	$C_n H_{3.8n} N_{0.1n}$	18	_	_	0.48	21,500	19,400

¹At 1 atm and 68°F.

 $^{^2\}mbox{At }77^{\circ}\mbox{F}$ for liquid fuels, and 1 atm and normal boiling temperature for gaseous fuels.

 $^{^3\}mbox{At }77\mbox{°F}.$ Multiply by molar mass to obtain heating values in Btu/lbmol.



5000 3000

Prepared by Center for Applied Thermodynamic Studies, University of Idaho.

FIGURE A-31E

Psychrometric chart at 1 atm total pressure.

From the American Society of Heating Refrigerating and Air-Conditioning Engineers, Atlanta, GA: used with permission.