Name:	Instructor:	ENGR 222 - Quiz
	Soction:	

Allowed Materials: pencils and/or pens.

ExamForm := 45

Honor Statement: On my honor, I promise that I have not received any unauthorized assistance on this exam (I didn't look at another student's paper, I didn't view any unauthorized written materials, I didn't talk or listen to another student, I didn't use an unauthorized calculator, I didn't use any electronic device, any visual or auditory signals, or any other techniques of exchanging information with others.) I have maintained the highest standards of academic integrity while completing this exam.

Cianadi				
Signed:			 	

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## 1. (2 point deduction for failure to complete this problem!)

02 03

Bubble:

01

For Course Section:
001 Hollins
002 Reeves
003 Reis

- Write in all of the indicated information in the boxes of your response form.
  Darken the appropriate circles to encode the corresponding information.
- Write your name on this exam and sign the Honor Statement.

### Notes:

- If your last name is too long, just write the first 10 letters.
- "F.I." and "M.I." are your first and middle initials, respectively
- Your "Username" is the first part of your LATech email address
- For "Section" use the guide provided to the right
- Your "Exam Form" is printed on the upper right corner of this page.
- Indicate "ENGR" as the "Program"

Exam	Т	Program							
Form		<u>.</u>	BIEN						
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Last Name							F.I.	M.I.	. LA Tech Username					Course #			Section (last 2 digits)					
(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(0)	(0)	0	(0)	(0)	(0)	0	(0)

Please put your final answers on the answer sheet that was given to you. You must show your work to receive full credit.

The words "steam" and "water" may be used interchangably. Check the tables to determine the phase of the system.

Unless the problem states otherwise, assume that the atmospheric pressure is 101.325 kPa or 14.7 psia.

Read the questions carefully and CHECK YOUR UNITS.

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Good luck!

2. (10 points) Which of these devices can be used to greatly increase the pressure of a gas?

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3. (10 points) A device commonly used in air conditioners and refrigerators to significantly drop the temperature of a fluid without the aid of another fluid or work input is a \_

**4.** (10 points) A volume balance can be performed on an open system if the flow is \_\_\_\_\_\_.

**5.** (10 points) R-134a flows through a throttling valve at a rate of 7.5 lbm/s. The refrigerant enters the throttling valve at 80 psia and temperature =  $40 \cdot {}^{\circ}$ F and leaves at 30 psia. The quality of the R-134a at the exit is closest to:

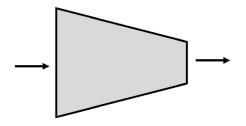


$$Choices = \begin{pmatrix} "A" & 0.0621 \\ "B" & 0.0712 \\ "C" & 0.0801 \\ "D" & 0.0891 \\ "E" & 0.0981 \\ "F" & 0.1070 \\ "G" & "there is no quality" \\ "H" & "not enough information" \\ \end{pmatrix}$$

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**6.** (10 points) Air passes through a nozzle at a rate =  $0.067 \cdot \frac{\text{kg}}{\text{s}}$ . The air exits through the nozzle, which has an (exit)

 $area = 71 \cdot mm^2$ , at 800 kPa and 350 K. The velocity of air at the exit is closest to:



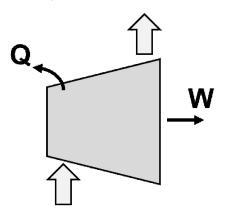
$$Choices = \begin{pmatrix} "A" & 52.91 \\ "B" & 66.03 \\ "C" & 79.10 \\ "D" & 92.28 \\ "E" & 105.34 \\ "F" & 118.49 \\ "G" & 131.61 \\ "H" & "not enough information" \end{pmatrix}. \frac{n}{s}$$

7. (10 points) A rigid tank  $contains = 2.8 \cdot m^3$  of air  $at = 210 \cdot kPa$  and  $22^{\circ}C$ . The tank is connected through a valve to an air supply line carrying air at 600 kPa and 22°C. The valve is opened long enough for the air inside the tank to reach a pressure of 600 kPa and then the valve is closed. At the end of the process, a thermometer inside the tank reads the final air temperature at 77°C. Accounting for variable specific heat, the heat transfer out of the tank is closest to:

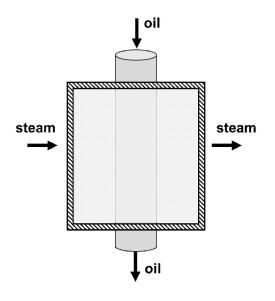
$$\label{eq:Choices} \text{Choices} = \begin{pmatrix} \text{"A"} & 121.6 \\ \text{"B"} & 136.6 \\ \text{"C"} & 151.8 \\ \text{"D"} & 167.0 \\ \text{"E"} & 182.2 \\ \text{"F"} & 197.3 \\ \text{"G"} & 212.4 \\ \text{"H"} & \text{"not enough information"} \end{pmatrix} \cdot \text{kJ}$$

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8. (10 points) Steam enters a turbine at a  $_{rate}=72\cdot\frac{kg}{s}$  at 1.4 MPa and 800 °C. It leaves the turbine at 150 kPa and with a quality of x=0.97. The turbine also  $_{loses}=550\cdot\frac{kJ}{kg}$  of heat. Assuming any changes in kinetic and potential energy are negligible, the rate of work output from the turbine is closest to:



9. (10 points) An insulated heat exchanger uses steam to heat up a liquid stream of (light) oil. The steam enters the heat exchanger at a rate =  $3.1 \cdot \frac{kg}{s}$  and at  $500^{\circ}$ C and 1 MPa. The steam leaves the heat exchanger as a saturated liquid at the same inlet pressure. Oil enters the heat exchanger at a rate of 100 kg/s and at an initial temperature =  $25.^{\circ}$ C. Assuming that the specific heat of oil is constant (and does not experience any phase change), the temperature of the oil when it leaves is closest to:



$$\label{eq:Choices} \text{Choices} = \begin{pmatrix} \text{"A"} & 64.5 \\ \text{"B"} & 68.2 \\ \text{"C"} & 71.8 \\ \text{"D"} & 75.4 \\ \text{"E"} & 79.1 \\ \text{"F"} & 82.7 \\ \text{"G"} & \text{"not enough information"} \end{pmatrix} \cdot ^{\circ} \text{C}$$

**10.** (10 points) Water enters an insulated chamber in two streams (1 and 2) and leaves the chamber in a single stream. Some values are given below. Neglecting changes of kinetic and potential energy, determine the velocity of the outlet stream.

$$T_{1} = 900 \cdot {^{\circ}C} \qquad P_{1} = 400 \cdot {^{\circ}R} \qquad P_{2} = P_{1} = P_{out} \qquad D_{out} = 42 \cdot cm$$
 
$$m_{dot1} = 0.3 \frac{kg}{s} \qquad m_{dot2} = 0.6 \frac{kg}{s} \qquad x_{2} = 36 \cdot \%$$
 
$$Choices = \begin{pmatrix} "A" & 2.27 \\ "B" & 2.50 \\ "C" & 2.73 \\ "D" & 2.95 \\ "E" & 3.18 \\ "F" & 3.41 \\ "G" & 3.64 \end{pmatrix}$$

11. (10 points) Air flows through an adiabatic compressor; the inlet conditions and some exit conditions are listed below. The mechanical power input is  $W_{dot} = 4.2 \cdot kW$ . Find the mass flow rate of the air.

Tip: Do not assume a constant specific heat.

$$T_1 = 210 \, \text{K} \qquad P_1 = 100 \cdot \text{kPa} \qquad P_2 = 750 \cdot \text{kPa} \qquad v_2 = 0.336747 \frac{\text{m}^3}{\text{kg}} \qquad \left( \begin{array}{ccc} \text{"A"} & 5.57 \\ \text{"B"} & 5.99 \\ \text{"C"} & 6.42 \\ \text{"B"} & 7.26 \\ \text{"F"} & 7.69 \\ \text{"G"} & 8.12 \\ \text{"H"} & 9.52 \end{array} \right) \cdot 10^{-3} \frac{\text{kg}}{\text{s}}$$

# **END OF EXAM**

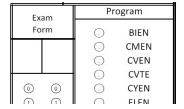
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Bubble:	For Co	urse Section:
01	001	Hollins
02	002	Reeeves
03	003	Reis

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Last Name							F.I.	M.I.	. LA Tech Username					Co	ourse	Section (last 2 digits)						
(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(A)	(0)	0	0	0	0	0	0	0

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