

Given:

Air is compressed in a car engine from 22°C and 95 kPa in a reversible and adiabatic manner.

$$T_1 := 22^\circ\text{C} = 295.15\text{ K} \quad P_1 := 95\text{ kPa}$$

Required:

If the compression ratio (v_1/v_2) of the engine is 8, determine the final temperature of the air.

Solution:

The compression ratio is defined as

$$r_c := 8$$

Going to Table A-17 @ $T_1 = 295.15\text{ K}$ shows that interpolation is needed. This is shown below.

$$T_a := 295\text{ K} \quad T_b := 298\text{ K}$$

$$\nu_{ra} := 647.9 \quad \nu_{rb} := 631.9$$

$$\nu_{r1} := \frac{T_1 - T_a}{T_b - T_a} \cdot (\nu_{rb} - \nu_{ra}) + \nu_{ra} = 647.1$$

The relative specific volume at state 2 may then be found by

$$\nu_{r2} = \frac{\nu_2}{\nu_1} \cdot \nu_{r1} \quad \text{or} \quad \nu_{r2} := \frac{\nu_{r1}}{r_c} = 80.9$$

Going to Table A-17 @ $\nu_{r2} = 80.9$ shows that interpolation is need.

$$\nu_{ra} := 85.34 \quad \nu_{rb} := 81.89$$

$$T_a := 650\text{ K} \quad T_b := 660\text{ K}$$

$$T_2 := \frac{\nu_{r2} - \nu_{ra}}{\nu_{rb} - \nu_{ra}} \cdot (T_b - T_a) + T_a = 662.9\text{ K} \quad T_2 = 389.8^\circ\text{C}$$