

Given:

The Mars Curiosity Rover uses a radioisotope thermoelectric generator (RTG) which uses the heat from nuclear decay to generate electricity. Curiosity uses 4.8 kg of plutonium-238 dioxide which has a power density of 0.54 W/g. The average temperature of the heat source is 550°C and the average temperature of the external sink is 230°C. **Note:** ^{238}Pu has a half-life of 87.7 years; which means the power output will decrease by 0.787% per year.

$$m_{\text{fuel}} := 4.8 \text{ kg} \quad q'_{\text{fuel}} := 0.54 \frac{\text{W}}{\text{gm}} \quad T_H := 550^\circ\text{C} \quad T_L := 230^\circ\text{C}$$

Required:

What is the maximum rate of power production for this thermoelectric generator? Curiosity actually produces 110 W of electrical power and outputs 2 kW of heat waste. What is the actual thermal efficiency of the RTG?

Solution:

The actual work output of the RTG is defined as

$$W'_{\text{out,rtg}} := 110 \text{ W}$$

The actual heat rejected by the RTG is defined as

$$Q'_{L,\text{rtg}} := 2 \text{ kW}$$

The Carnot efficiency of the system is given by

$$\eta_{\text{th,rev}} := 1 - \frac{T_L}{T_H} = 38.88\%$$

The theoretical heat supplied to the RTG is given by

$$Q'_{H,\text{fuel}} := m_{\text{fuel}} \cdot q'_{\text{fuel}} = 2.592 \cdot \text{kW}$$

The maximum rate of power production for the RTG is then found by

$$\eta_{\text{th}} = \frac{W'_{\text{out}}}{Q'_H} \quad \text{or} \quad \boxed{W'_{\text{out,max}} := \eta_{\text{th,rev}} \cdot Q'_{H,\text{fuel}} = 1.008 \cdot \text{kW}}$$

The actual heat supplied to the RTG is given by

$$Q'_{H,\text{rtg}} := W'_{\text{out,rtg}} + Q'_{L,\text{rtg}} = 2.11 \cdot \text{kW}$$

The actual thermal efficiency is then given by

$$\boxed{\eta_{\text{th,rtg}} := \frac{W'_{\text{out,rtg}}}{Q'_{H,\text{rtg}}} = 5.21\%}$$