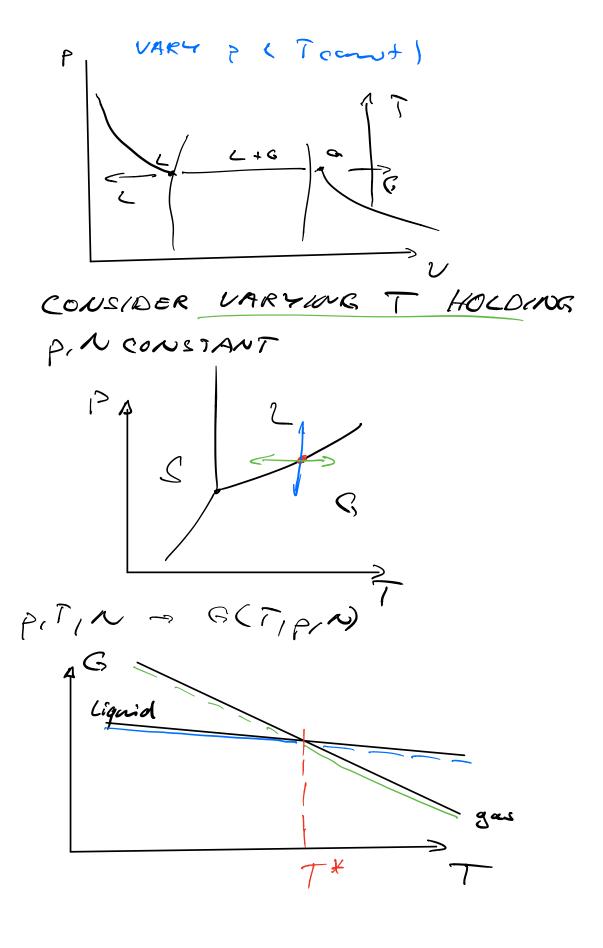
$$\frac{\partial^2 G}{\partial T^2}\Big|_{PM} = -\frac{\partial S}{\partial T}\Big|_{PN} = -\frac{CP}{T} < 0$$

$$\frac{\partial C}{\partial T^2}\Big|_{PM} = -\frac{\partial S}{\partial T}\Big|_{PN} = -\frac{CP}{T} < 0$$

$$\frac{\partial C}{\partial T^2}\Big|_{PM} = -\frac{\partial S}{\partial T}\Big|_{PN} = -\frac{CP}{T} < 0$$

$$\frac{\partial C}{\partial T^2}\Big|_{PM} = -\frac{\partial S}{\partial T}\Big|_{PN} = -\frac{CP}{T} < 0$$

$$\frac{\partial C}{\partial T}\Big|_{PN} = -$$



N = coust, system & extensive

-> Kle = Kla

CALCULATE CHANGE IN ENTROPY

$$\Delta S = \int \frac{dQ}{T} = \frac{1}{T^*} \int dQ = \frac{Q}{T^*} = \frac{L \cdot N}{T^*}$$

$$T = T^* = com t$$

L: Latent heat per partiele.

MOBEL OF A PHASE
TRANSITION

IDEAL GAS pV = NKT no phase transition

goad approximation to DILLITE REAL GASES

REAL GASES: STARTING FROM IDEAL GAS AND ADD CORRECTIONS IN POWERS OF DENETTY R=1, :

P= KT(U)

= KT(U)

= KT(U)

+ B(T) W + B(T)(U)

+ ...

KNOWN as V/R/AL expansion.

Bn (T): V(RIAL COEFFCIENTS.

-> SIMPLE APPROXIMATION FOR A REAL GAS

PU=NGT 1) V -> V - Vexel.

Som finite size
ef molecules
Verd & N.6

2) ATTRACTIVE INTERACTION RETWEEN PARTICLES

potential energy proportion DUF TO THE ATTRATION

Mastr & N moderales

 $U_{\text{olf}} \propto N_{\text{laffr}}$ $= -a N \cdot \frac{N}{V} = -a \frac{N^2}{V}$ a > 0

$$P_{\alpha f h} = -\frac{\partial U}{\partial V}\Big|_{u,s} = -(-a\frac{v^2}{V})(-\frac{1}{V})$$

$$= -a\frac{v^2}{V^2}$$

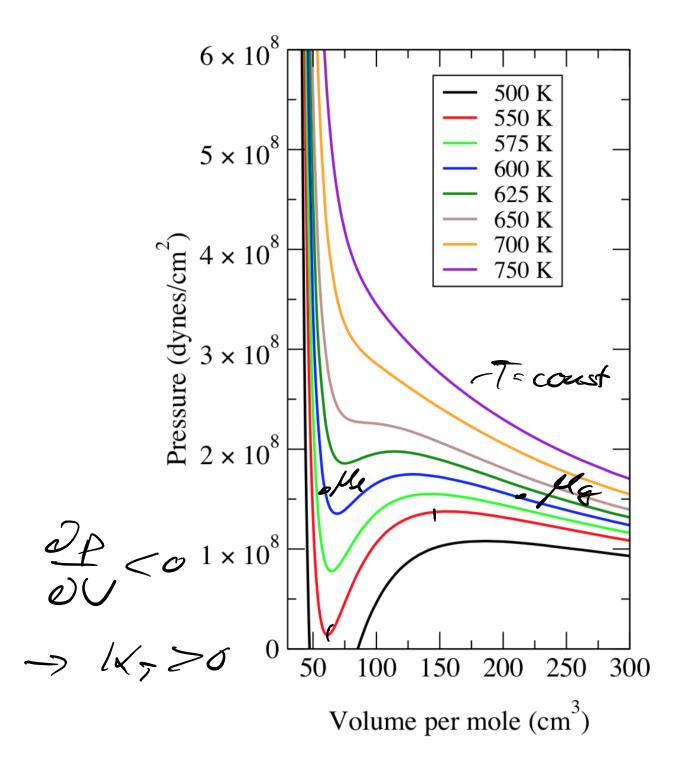
$$P = \frac{NUT}{V-NLS} + Pahr$$

$$\Rightarrow \sqrt{P + a\frac{v^2}{V^2}}(V-NLS) = NUT$$

van de Waals EOS

1. Intro to van der Waaals EOS

Do problems 11.1 and 11.2 a) (not b) in EOPC.



MAXWELL CONSTRUCTION pre (Ve) = mg (Vg) 1-N Go (4) = Gg (4) dG = -SdT +Vdp + udd T = coust , W = coust dG = +Vdp

