Suppression of superconductivity in the $L_{1-x} Pr_x Ba_{1.7} Sr_{0.3} Cu_3 O_7$ (L = Yb and Lu) system: Observation of the hole localization effect

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Compounds in the system $L_{1-x} \Pr_x \operatorname{Ba}_{1.7} \operatorname{Sr}_{0.3} \operatorname{Cu}_3 \operatorname{O}_7$ ($L = \operatorname{Yb}$ and Lu) were synthesized to study the effect of Pr substitution on superconductivity. The nature of variation of T_c with Pr concentration is significantly different from that in other Pr-doped L-123 systems and deviates from Abrikosov-Gor'kov-like behavior. The results are explained on the basis of hole localization and magnetic pair-breaking mechanisms. The ion size effect on T_c is attributed to the dependence of the extent of orthorhombicity and oxygen ordering on the L ion leading to retention of mobile holes in the CuO_2 planes.

INTRODUCTION

The compound PrBa₂Cu₃O₇ (Pr-123) crystallizes orthorhombically and is isostructural to 123 structure. However, unlike other L-123 (L = rare earth) compounds, that show superconductivity with $T_c = 90$ K, Pr-123 is neither metallic nor superconducting. The absence of superconductivity in Pr-123 has been extensively investigated in the literature. The superconducting transition temperature T_c is strongly suppressed by the partial substitution of Pr at the L site in other L-123. Several studies have been made using a variety of experimental techniques on the system $Y_{1-x}Pr_xBa_2Cu_3O_7$ to explain the T_c suppression mechanism.^{4,5} Upon Pr substitution the superconducting transition temperature T_c decreases monotonically form 90 K for x = 0.0, and undergoes a metal to semiconductor transition at $x \ge 0.6$. Antiferromagnetism also appears for x > 0.6 and reaches a maximum T_N of 17 K for x = 1.0 (PrBa₂Cu₃O₇), which is an order of magnitude higher than that observed for other L-123 systems.

Mechanisms involving hole filling (or localization) and/or depairing of superconducting electrons via exchange scattering of the mobile holes in the CuO₂ sheets by the Pr magnetic moment such as Abrikosov-Gor'kov pair-breaking theory have been proposed to account for these observations.⁶⁻⁹ There is now strong evidence that Pr 4f and O 2p orbitals of the CuO₂ planes strongly hybridize and therefore the mobile holes in the CuO₂ planes are localized.⁹ Transport, thermal, and magnetic measurements suggest that the valence of Pr is close to 4+ while several spectroscopic studies reveal the valence of Pr to be +3.³ The valence state of Pr is still a controversial issue that remains to be solved.

Studies on other $L_{1-x} Pr_x Ba_2 Cu_3 O_7$ (L = La-Yb) systems suggest that ionic size of the L^{3+} plays an important role in the rate of suppression of T_c . ¹⁰⁻¹⁴ Substitution of Pr for L^{3+} in L-123 containing larger L^{3+} appears to decrease T_c at a much faster rate compared to the substitution in L-123 with smaller L^{3+} . The critical concentration of Pr, x_{cr} , at which T_c completely vanishes

is x=0.33 for Nd-123 and x=0.56 for Y-123 (small discrepancies in the value of $x_{\rm cr}$ between different groups in the literature could be due to differences in the synthetic procedure and/or the oxygen content that can also affect T_c). The decrease in T_c is linear with x in $L_{1-x} \Pr_x \text{Ba}_2 \text{Cu}_3 \text{O}_7$ (L=La-Dy) until $x_{\rm cr}$. For L=Ho-Yb and Y, the T_c reduction shows a deviation from linear decrease in T_c at the initial concentrations of Pr. For L=Yb, the deviation from linearity is significantly higher compared to the Y system. However, at any given composition x, T_c decreases approximately linearly with increasing radius of the L ion.

In the case of L = Yb the x = 0.0 composition viz. YbBa₂Cu₃O₇ does not yield a single-phase compound. It always results in multiphase formation with Yb2BaCuO5 (211) as a major impurity besides the 123 phase. For L = Lu the percentage of 123 phase itself is very low. This constraint restricts the study of superconducting properties at lower concentration of $L_{1-x}Pr_xBa_2Cu_3O_7$ (L=Yb and Lu) system. However, we have shown that single-phase Yb-123 can be achieved easily by partial substitution of Sr for Ba (as low as 5%). Lu-123 can also be synthesized as a major phase with impurity (211 phase) concentration less than 5%. 15 Therefore partial substitution of Sr at the Ba site in L-123 (L = Yb and Lu) facilitates the study of the effect of Pr on the T_c even at lower concentrations of Pr.

In the present work, we have undertaken a study on the structural and superconducting behavior of $L_{1-x} \operatorname{Pr}_x \operatorname{Ba}_{1.7} \operatorname{Sr}_{0.3} \operatorname{Cu}_3 \operatorname{O}_7$ ($L = \operatorname{Yb}$ and Lu) system. Our results show that in the phases with $L = \operatorname{Yb}$, single-phase 123 could be obtained for $0.0 \le x \le 1.0$. For phases with $L = \operatorname{Lu}$, however, single-phase formation is seen from $x \ge 0.1$. The plot of normalized T_c (i.e., $T_c/T_{c(0)}$) vs composition x shows that T_c remains almost constant at initial concentration of x (up to $x \approx 0.2$ in both the systems).

EXPERIMENT

Compounds of the general formula, $L_{1-x}Pr_xBa_{1.7}Sr_{0.3}Cu_3O_7$ (L=Yb and Lu) with

 $0.0 \le x \le 1.0$ were prepared by employing a high-temperature solid-state reaction. Respective high-purity oxides (>99.9%) and carbonates (99.99%) were thoroughly mixed according to the desired stoichiometric ratio and heated initially at 900 °C for 24 h. Later the powder was reground and again heated at 910–920 °C for 48 h with several intermittent grindings. The resultant powder was then pelletized and sintered at 930 °C for 24 h. Finally, all the compositions were given oxygen treatment at 900 °C for 24 h and 600 °C for 48 h followed by furnace cooling to room temperature under O_2 flow.

The phase formation was established by powder x-ray diffraction (Rich Seifert unit, Germany; XRD 3000P). The XRD patterns were indexed based on orthorhombic unit cell and the lattice parameters were calculated by the LSQ fit of high angle lines. The oxygen content for the single-phase materials was estimated by the iodometric method. Superconductivity was established by both resistivity and susceptibility techniques. dc resistivity measurements were carried out by the four-probe method (van der Pauw method) from room temperature down to 13 K using a closed-cycle He refrigerator. ac magnetic susceptibility measurements were carried out by mutual inductance method in the range 300-13 K using Sumitomo Superconducting Property measuring equipment (SCR 204-T; Japan; field of the order of 0.1 Oe, frequency 300 Hz).

RESULTS AND DISCUSSION

A total of about 32 compositions were synthesized in the present study. X-ray-diffraction (XRD) studies reveal that for L = Yb, in the range $0.0 \le x \le 1.0$ all the compo-

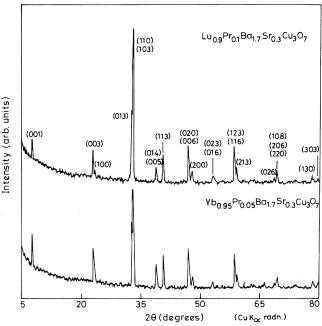


FIG. 1. XRD pattern for the composition that is single phase with the lowest concentration of Pr for L = Yb and Lu in the system, $L_{1-x}Pr_xBa_{1.7}Sr_{0.3}Cu_3O_7$.

sitions are single phasic, while for L = Lu a single-phase region exists for $0.1 \le x \le 1.0$. X-ray-diffraction patterns for representative compositions that are single phasic with the lowest concentration of Pr in both the systems are shown in Fig. 1. The multiphasic compositions contain mainly $\text{Lu}_2\text{BaCuO}_5$ (211) and BaCuO_2 as impurities.

The LSQ fitted orthorhombic lattice parameters are presented in Tables I and II for L=Yb and Lu, respectively. There is a systematic increase in a and decrease in b parameters with increase in x in both the systems. Consequently there is a decrease in the degree of orthorhombicity defined as (b-a)/a. This is consistent with the observation that the degree of orthorhombicity increases from LaBa₂Cu₃O₇ to TmBa₂Cu₃O₇. This trend can be expected for the Sr-substituted phases also. In the L-123 system it is known that although Sr substitution at the Ba site decreases the overall unit cell volume, the degree of orthorhombicity, however, remains unchanged. The c lattice parameter in the present study does not vary according to Vegard's law at lower concentrations, up to x=0.2.

The oxygen content for all compositions in the single-phase region remains close to 6.92 ± 0.03 suggesting that the phases are well oxygenated and hence the variation in physical properties studied presently are not due to variations in oxygen content.

To establish superconductivity, resistivity, and susceptibility were measured in the range 300-13 K for all the compositions. Figure 2 shows the variation in resistivity vs temperature for L = Yb phases. The values of $\rho_{300 \text{ K}}$ lie in the range 1-2 m Ω cm for metallic and superconducting samples. Superconductivity is observed for phases with x = 0.0-0.55 for L = Yb and x = 0.10-0.55for L = Lu (not shown in figure). In both the systems phases with low Pr concentrations, i.e., x = 0.0-0.2, exhibit sharp transition from metallic behavior to superconductivity. The transition width increases for phases with x > 0.3. Figure 3 shows the susceptibility plots for the L = Lu system. For all the compositions showing T_c , in both the systems, the value of $T_{c, {
m zero}}$ from ho-T measurements matches well with $T_{c, {
m onset}}$ from susceptibility measurements. Phases with $x \ge 0.6$ in both the systems exhibits semiconducting behavior.

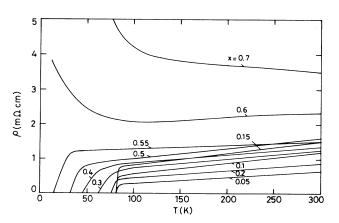


FIG. 2. ρ -T plots for the Yb_{1-x}Pr_xBa_{1.7}Sr_{0.3}Cu₃O₇ system.

TABLE I. Structural and T_c data for the $Yb_{1-x}Pr_xBa_{1.7}Sr_{0.3}Cu_3O_{7-\delta}$ system.

Composition x	Lattice parameter in Å			(b-a)/a	δ	ρ _{300 K}	$T_{c, m zero} \ ({f K})$
	a	b	c	$\times 10^{-2}$	± 0.03	mΩ cm	±1 K
0.00	3.794	3.870	11.600	2.00	0.10	0.50	81
0.05	3.792	3.867	11.589	1.98	0.09	0.66	81
0.10	3.794	3.863	11.593	1.82	0.10	1.18	81
0.15	3.796	3.869	11.608	1.92	0.09	1.51	80
0.20	3.797	3.869	11.617	1.90	0.08	0.85	76
0.25	3.800	3.869	11.619	1.82	0.08	0.95	71
0.30	3.803	3.869	11.628	1.74	0.07	1.25	63
0.35	3.804	3.870	11.635	1.74	0.05	1.30	55
0.40	3.815	3.874	11.652	1.55	0.07	1.36	46
0.45	3.815	3.868	11.650	1.39	0.12	1.60	41
0.50	3.815	3.873	11.655	1.52	0.07	1.51	30
0.55	3.819	3.875	11.662	1.46	0.09	1.49	15
0.60	3.826	3.873	11.674	1.23	0.13	2.31	
0.70	3.830	3.882	11.670	1.35	0.09	3.51	
0.80	3.837	3.881	11.716	1.15	0.08	4.41	
0.90	3.855	3.883	11.721	0.73	0.12	7.32	
1.00	3.861	3.890	11.722	0.75	0.13	13.6	

Isovalent substitution of Sr for Ba in YBa₂Cu₃O₇ decreases the superconducting transition temperature T_c from 90 K for the pure phase to 79 K for YBaSrCu₃O₇. Neutron-diffraction and extended x-ray absorption fine structure (EXAFS) studies on the $YBa_{2-x}Sr_xCu_3O_7$ reveal that the Cu(2)-O(2) plane and the CuO2-Y-CuO2 sandwich structure is well preserved. There is a strong distortion about the local structure at the Ba/Sr-O plane. 17 The reduction in the T_c values observed upon Sr substitution is attributed to the subtle changes in the band structure and the density of electronic states at the Fermi level caused by such local structural distortions. We have proved that, on changing the rare earth from Y to Yb the rate of T_c decrease (\sim 0.4 K/atom %) remains unaffected. This suggests that in the present study, the suppression of T_c is attributed to the effect caused by Pr substitution alone and there is no additional contribution from Sr substitution. Thus, Sr substitution in the present case helps only to stabilize the 123 phase.

Figure 4 shows the plot of normalized $T_{c, {\rm zero}}$ values for $L={\rm Yb}$ and Lu with respect to composition x. For comparison, we have also plotted $T_c/T_{c(0)}$ for ${\rm Y}_{1-x}{\rm Pr}_x{\rm Ba}_2{\rm Cu}_3{\rm O}_7$ (Ref. 5) and ${\rm Nd}_{1-x}{\rm Pr}_x{\rm Ba}_2{\rm Cu}_3{\rm O}_7$. As can be seen from the plot, at lower concentrations of Pr the superconducting transition temperature T_c

TABLE II. Structural and T_c data for the $Lu_{1-x}Pr_xBa_{1.7}Sr_{0.3}Cu_3O_{7-\delta}$ system.

Composition	Lattice parameter in Å			(b-a)/a	δ	ρ _{300 K}	$T_{c, \mathrm{zero}}$ (\mathbf{K})
x	а	b	c	$\times 10^{-2}$	±0.03	$m\Omega$ cm	$\pm 1~K$
0.00	3.790	3.868	11.640	2.06			
0.10	3.794	3.869	11.609	1.98	0.05	1.02	79
0.15	3.794	3.868	11.609	1.95	0.09	0.78	79
0.20	3.795	3.870	11.611	1.97	0.07	0.88	77
0.25	3.797	3.872	11.613	1.97	0.10	0.71	75
0.30	3.799	3.871	11.618	1.89	0.11	0.82	66
0.35	3.807	3.876	11.622	1.81	0.07	1.11	61
0.40	3.812	3.877	11.625	1.70	0.09	1.25	52
0.45	3.811	3.875	11.635	1.68	0.08	1.79	47
0.50	3.813	3.874	11.643	1.60	0.08	1.49	35
0.55	3.823	3.878	11.661	1.44	0.09	1.85	20
0.60	3.825	3.878	11.664	1.38	0.07	2.26	
0.70	3.833	3.875	11.674	1.09	0.10	2.94	
0.80	3.846	3.883	11.693	0.96	0.12	4.30	
0.90	3.856	3.887	11.705	0.85	0.11	6.40	
1.00	3.861	3.890	11.722	0.75	0.13	13.6	

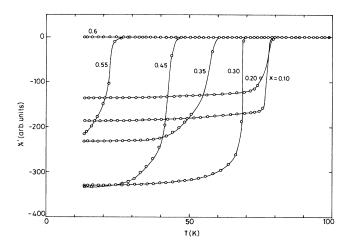


FIG. 3. χ -T plots for the Lu_{1-x}Pr_xBa_{1.7}Sr_{0.3}Cu₃O₇ system.

remains unaffected up to x=0.15 for L=Yb and x=0.20 for L=Lu. In the case of L=Y also such an invariant behavior is observed up to x=0.05. However, this invariant region is not evident in the case of larger L's. Thus, as the ionic size of L decreases there exists a region where Pr appears to have negligible effect on superconductivity. Also, it is evident form the plot that at any given concentration of Pr the value of T_c is higher for smaller L's compared to that for larger L's. Figure 5 shows the c lattice parameter variation with x in comparison with T_c variation. The c axis remains unchanged in the range of x over which the T_c is weakly dependent on Pr concentration.

The suppression of T_c by Pr in L_{1-x} Pr_xBa₂Cu₃O₇ sys-

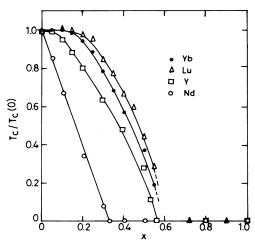


FIG. 4. Normalized T_c values $(T_c/T_{c(0)})$ vs x plots for the system, $L_{1-x}\Pr_x \text{Ba}_{1.7}\text{Sr}_{0.3}\text{Cu}_3\text{O}_7$ (L=Yb and Lu). Plots for the system, $L_{1-x}\Pr_x \text{Ba}_2\text{Cu}_3\text{O}_7$ [L=Y (Ref. 5) and Nd (Ref. 17)] are also shown for comparison.

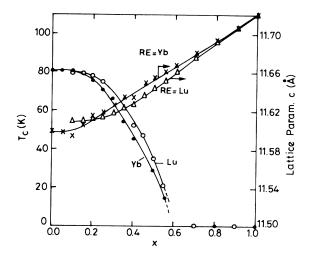


FIG. 5. Variation of T_c and c lattice parameter vs x. The invariance of T_c and c parameter at low Pr concentrations can be clearly seen.

tem for other L's was explained in terms of pair-breaking theory such as Abrikosov-Gor'kov (AG) theory, in which the depairing of superconducting electrons occurs via exchange scattering of mobile holes in CuO_2 plane by Pr magnetic moment. AG theory predicts that the depression of T_c with paramagnetic impurity concentration is linear in the low concentration region (concentration corresponding to $\approx 50\%$ reduction of T_c). ¹⁹ Here, Pr is the only impurity ion. The decrease in T_c with composition x is given by the relation

$$\ln \left[\frac{T_c}{T_{c(0)}} \right] = \psi \left[\frac{1}{2} \right] - \psi \left[\frac{1}{2} + \frac{0.14xT_{c(0)}}{x_{cr}T_c} \right] ,$$

where x_{cr} corresponds to the critical concentration of Pr for the complete suppression of T_c and ψ is the digamma function. This expression takes a linear form as

$$\frac{T_c}{T_{c(0)}} = 1 - \left[\frac{\pi^2}{4k_B} N(E_F) j_{\text{ex}}^2 (g-1)^2 J(J+1) \right] x$$

$$= 1 - Ax ,$$

where $N(E_F)$ is the density of states at the Fermi level, g the Lande g factor, J the total angular momentum of the Hund's rule ground state for the \Pr^{3+} ion, and $j_{\rm ex}$ is the exchange interaction parameter.

Hence, the linear dependence of $T_c/T_{c(0)}$ vs composition (x) observed in the case of L=La-Dy can be viewed as an evidence for AG pair-breaking mechanism of T_c suppression. In the present study, for concentrations of Pr up to x=0.15 for L=Yb and x=0.20 for L=Lu, the experimental data of $T_c/T_{c(0)}$ vs x deviate from AG-like behavior with an initial slope close to zero as $x\to 0$. At higher concentrations of Pr also, there is a significant deviation from linearity and the curve is not fully AG like. Hence, the suppression of T_c cannot be exclusively

attributed to the AG pair-breaking mechanism alone.

The two most important explanations among many proposed for the L-ion size effect on T_c in $L_{1-x} \Pr_x \operatorname{Ba_2Cu_3O_7}$ system are (i) the extent of hybridization between 4f states of Pr ion and the conduction band states depends on the ionic radius of the L ion¹² and (ii) the tendency of the system to phase separate into $L\operatorname{Ba_2Cu_3O_7}$ and $\operatorname{PrBa_2Cu_3O_7}$ is greater for larger L's while smaller L's tend to distribute Pr statistically. It is assumed that localization of holes occurs only when there are two neighboring Pr ions. ¹³ In the present system, $L_{1-x}\operatorname{Pr}_x\operatorname{Ba_{1.7}Sr_{0.3}Cu_3O_7}, L=\operatorname{Yb}$ and Lu , due to the smaller size of the polyhedron around the L ion, the structure can facilitate the stabilization of more highly hibridized Pr ion.

A careful study by Neumeier and Maple⁵ on the system $Y_{1-x}Pr_xBa_2Cu_3O_7$ indicates that the experimental T_c vs x curve is not linear at low concentrations of Pr. The results are explained in terms of two competing effects: (a) hole localization leading to increase in T_c and (b) magnetic pair breaking. A successful attempt to deconvolute the two mechanisms was carried out in the system (Y_{1-x-y}Pr_xCa_y)Ba₂Cu₃O₇.⁶ The mobile holes filled or localized by the Pr ion have been compensated by the creation of additional holes in the CuO2 planes by the simultaneous substitution of Ca2+. The experimental data, $T_c(x,y)$, are best fitted into the expression $T_c(x,y) = T_{c(0)} - A (\alpha - \beta x + y)^2 - Bx$. The quantity $T_{c(0)}$ is the maximum possible value of T_c attainable for the optimum hole concentration in the CuO₂ planes in the absence of magnetic pair breaking. The second term is an empirical relation that represents the effect of hole generation by the Ca(y) ions and the localization by the Pr(x) ions where β is the fraction of holes localized per Pr ion and α is the optimal hole concentration. The -Bx term represents the depression of T_c with x due to the pair-breaking effect. The values obtained for the best fit of the experimental data are $T_{c(0)}$ =97 K, A =425 K, $\alpha = 0.1$, $\beta = 0.95 \pm 0.2$ and B = 96.5 K.

The value of $T_{c\,(0)}=97$ K presumes that the T_c of YBa₂Cu₃O_{7- δ} could be increased to 97 K by a reduction in the hole concentration of 0.1 holes per formula unit. Indeed, it has been shown that the T_c is enhanced by ~ 2 K in the system YBa_{2-x}La_xCu₃O₇ for x=0.05-0.06. Also, with the increase of δ in YBa₂Cu₃O_{7- δ}, T_c passes through a maximum (an enhancement of ~ 2 K) near $\delta=0.12$. Thus, the weak dependence of T_c on x in Y_{1-x}Pr_xBa₂Cu₃O₇ at low concentrations of Pr is due to the two opposing effects viz. a reduction in mobile hole concentration tending to enhance the T_c and the magnetic pair-breaking effect reducing the T_c .

The experimental data in the present study provide further evidence for the above model. The weak dependence of T_c on x at low concentrations of Pr in both the systems, $L_{1-x} \operatorname{Pr}_x \operatorname{Ba}_{1.7} \operatorname{Sr}_{0.3} \operatorname{Cu}_3 \operatorname{O}_7$ ($L = \operatorname{Yb}$ and Lu), similar to $\operatorname{Y}_{1-x} \operatorname{Pr}_x \operatorname{Ba}_2 \operatorname{Cu}_3 \operatorname{O}_7$ system, is due to the competing effects, the hole localization and the pair-breaking effects. However, a distinct feature is that the region of x where there is a weak dependence of T_c on the concentration of Pr increases with decreasing size of the L ion. Also, the

values of T_c for a given x (Pr concentration) are higher compared to those of Pr-substituted L-123 with larger L's and the behavior of T_c vs x is non-AG like even at higher values of x. These can be attributed to an increase in mobile hole concentration in the CuO_2 planes with decrease in L-ion size. The origin of the higher mobile hole concentration for smaller L's can be explained as follows.

When superconductivity is suppressed in LBa₂Cu₃O₇ by chemical substitution, oxygen removal, etc. (leading to hole filling or localization), there appears a systematic dependence of the T_c suppression on the ionic size of the L^{3+} . The larger the size of the L ion, the higher is the rate of suppression of T_c , with respect to chemical substitution as well as oxygen depletion. 22,23 Indeed, the T_c value for a given dopant concentration increases systematically with decrease in size of the L ion. This is explained based on the fact that the degree of orthorhombicity increases with decrease in size of the L ion. In the case of smaller L's, the unit cell is smaller and hence the repulsion between the oxygens in the O(1) and O(5) positions is more. This would result in negligible occupancy of oxygen in the O(5) position. Thus, for L-123 with smaller L's, the retention of orthorhombicity and thereby the chain integrity results in an effective transfer of holes from the chain copper to the plane copper. This results in higher mobile hole concentration in the CuO₂ planes. This is further illustrated by the fact that the T_c vs δ plots for various $L \operatorname{Ba_2Cu_3O_{7-\delta}}$ (Ref. 23) reveal that in the case of L = Y, Er, Yb, etc., the range of δ over which the T_c is almost invariant at 60 K is wide $(0.25 \le \delta \le 0.55)$, attributed to oxygen ordering in the basal plane and effective charge transfer to the CuO₂ planes. The hole concentration is retained in spite of the oxygen depletion. In the case of L = La, Nd, Sm, etc., the range corresponding to the 60 K plateau region decreases with increase in the size of L. In the LaBa₂Cu₃O₇₋₈ compound, there is no existence of the 60 K plateau region.²³ Thus, in the present system, the retention of mobile holes leads to both the observed effects viz. weak dependence of T_c on x at low Pr concentrations and higher values of T_c for a given Pr concentration compared to the case of other L-123 with larger L's.

SUMMARY AND CONCLUSIONS

Partial substitution of Sr at the Ba site stabilizes $L=\mathrm{Yb}$ and Lu in the single-phase 123 structure and facilitates study of the effect of low concentrations of Pr in the system $L_{1-x}\mathrm{Pr}_x\mathrm{Ba}_{1.7}\mathrm{Sr}_{0.3}\mathrm{Cu}_3\mathrm{O}_{7-\delta}$. The single-phase region exists in the range $0.0 \le x \le 1.0$ for $L=\mathrm{Yb}$ and $0.1 \le x \le 1.0$ for $L=\mathrm{Lu}$. The degree of orthorhombicity decreases with increase in x. T_c remains unaffected in the range $0.0 \le x \le 0.15$ for $L=\mathrm{Yb}$ and $0.1 \le x \le 0.2$ for $L=\mathrm{Lu}$. For a given x, $T_c/T_{c(0)}$ values are significantly higher in the present case when compared to Prsubstituted L-123 with larger L's. The c lattice parameter remains almost unaffected in the range x where the T_c remains unaffected in the respective systems.

The behavior of the present systems

 $L_{1-x} Pr_x Ba_{1.7} Sr_{0.3} Cu_3 O_7$, L=Yb and Lu, is akin to that of the system $Y_{1-x-y} Pr_x Ca_y Ba_2 Cu_3 O_7$. In the latter, the generation of additional mobile holes is realized externally by Ca^{2+} substitution while in the former this is achieved internally by the increased degree of orthorhombicity and effective charge transfer from the CuO chains to the CuO_2 planes. This also explains our observations of higher values of T_c for the phases with smaller L's compared to those with larger L's for a given value of x and the deviations from linear AG-like

behavior. Hence, the size of the L ions plays a vital role in determining the observed physical properties.

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