# Pressure-induced metallization, transition to the pyrite-type structure, and superconductivity in palladium disulfide PdS<sub>2</sub>

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The pressure effect on crystal structures, lattice vibrations, and electrical transport properties of  $PdS_2$  and  $NiSe_2$  were studied under high pressures of up to  $\sim 50$  GPa using Raman spectroscopy and electrical resistivity measurements.  $PdS_2$  undergoes semiconductor-to-metal transition at a pressure of  $\sim 7$  GPa without a structural phase transition. Structural transition to the pyrite-type phase occurs at a pressure of  $\sim 16$  GPa. Superconductivity emerges in the pyrite-type phase of  $PdS_2$  with a well-defined dome-shaped dependence of critical temperature of superconductivity on pressure, with a maximum value of 8.0 K at 37.4 GPa. Although this behavior is similar to that in isostructural  $PdSe_2$ , correlation with anomalous anion bond softening at critical temperature is not observed in  $PdS_2$  in contrast to  $PdSe_2$ . Conversely, the anion bond instability is observed by the softening of the  $Se_2^{2-}$  internal vibrational Raman mode in pyrite  $PdSe_2$  under pressure. Despite the persisting structural instability due to destabilization of  $Se_2^{2-}$  dimers,  $PdSe_2$  remains a normal nonsuperconducting metal.

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### I. INTRODUCTION

Superconductivity with relatively high critical temperature  $(T_c)$  often emerges in compounds with structural instabilities associated with phonon softening, resulting in strongly enhanced electron-phonon coupling. A class of compounds much discussed in this respect are the A15 superconductors with general chemical formula  $A_3B$  (A = group IVb-VIb element, B = group IIIa-Va element) of which some are close to structural distortions [1,2]. The systematic tuning of chemical bonds as a structural parameter (by applied physical or chemical pressure) driving the system toward structural instability was demonstrated to be useful to control physical properties [3] and to enhance or even induce the superconductivity [4–7].

Transition metal dichalcogenides  $TX_2$  (where T denotes the transition metal and  $\bar{X}$  refers to the chalcogenide atom) with a pyrite structure containing quasimolecular  $X_2^{2-}$  dimers are an excellent system to analyze bonding systematics because electronic effects in these compounds are correlated with nonmetal-nonmetal bond strength [8–10]. Superconductivity in pyrite superconductors [11–14] might correlate with structural instability caused by destabilization of anion dimers. Superconductivity with the highest critical temperature emerges in these compounds in the vicinity of the weakest Se<sub>2</sub><sup>2-</sup> bonds tuned by chemical substitution in defective pyrite  $Ir_{0.94-x}Rh_{x}Se_{2}$  (9.6 K for x = 0.36) [11], or by pressure in the high-pressure pyrite phase of PdSe<sub>2</sub> (13.1 K at 23 GPa) [14]. To obtain further insight into the interplay between superconductivity and the bond strength of anion dimers in the pyrite structure, we performed experimental studies of lattice dynamics and electrical resistivity of PdS<sub>2</sub> (isostructural with PdSe<sub>2</sub> at ambient conditions) and pyrite NiSe<sub>2</sub> under high pressure.

### II. EXPERIMENTAL DETAILS

Single crystal samples of PdS<sub>2</sub> and NiSe<sub>2</sub> for this study were grown using the chemical vapor transport method. The compounds PdS<sub>2</sub> and NiSe<sub>2</sub> have first been synthesized by a direct reaction of the elements Pd (Alfa Aesar 99.95%) and S (Alfa Aesar 99.999%) at 550 °C and Ni (Alfa Aesar 99.996%) and Se (Alfa Aesar 99.999%) at 800 °C in evacuated fused silica tubes over 10 days. Phase purity of the obtained microcrystalline powder was proven by x-ray powder diffraction. Starting from the microcrystalline powder, single crystals of PdS<sub>2</sub> and NiSe<sub>2</sub> have been grown by a chemical transport reaction in a temperature gradient from 600 °C (source) to 500 °C (sink) with PdCl<sub>2</sub> (Alfa Aesar 99.9%) as the transport additive and in a temperature gradient from 800 °C (source) to 700 °C (sink) using SeCl<sub>4</sub> as the transport additive, respectively. The stoichiometry of the obtained crystals was proven using energy dispersive x-ray spectroscopy to be close to a nominal composition (33.2 at. % of Pd and 66.8 at. % of S for PdS<sub>2</sub> and 32.9 at. % of Ni and 67.1 at. % of Se for NiSe<sub>2</sub>) uniformly across the sample. The Raman spectra of the samples collected at ambient pressure are in good agreement with literature data proving the crystal structure of obtained single crystals.

A screw-clamped diamond anvil cell equipped with 500- $\mu$ m culet diamond anvils was used for the simultaneous Raman spectroscopy and electrical resistivity measurements. The gasket was made of a 300- $\mu$ m thick tungsten foil. For electrical resistivity measurements, the tungsten gasket was insulated with a cubic BN/epoxy mixture. Single crystal samples of a near-square shape  $\sim$ 120- $\mu$ m size and  $\sim$ 5-10- $\mu$ m thickness were loaded in the sample chamber (diameter 200  $\mu$ m and height 40  $\mu$ m) filled with NaCl as the pressure-

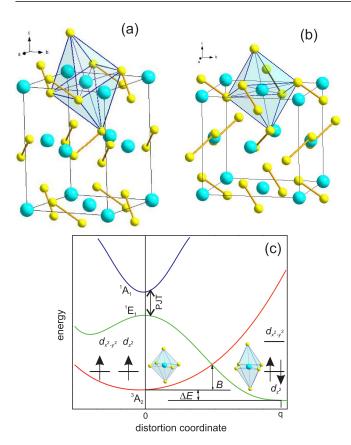


FIG. 1. The ambient-pressure crystal structure of  $PdS_2$  (a) is a distorted pyrite structure (b). (c) Schematic dependence of the energy of the electronic states of the half-filled  $e^2$  system on deformation coordinate q (E mode). Strong vibrational coupling between excited states (hidden pseudo-Jahn-Teller effect, PJT) makes the distorted configuration with the doubly occupied  $d_{z^2}$  orbital energetically favorable (energy separation  $\Delta E$  and energy barrier B). (Adapted from Ref. [16]).

transmitting medium. Four electrical leads fabricated from  $5-\mu$ m-thick Pt foil were contacted by application of pressure to the surface of the sample in the van der Pauw configuration. Electrical resistivity was measured with a direct current, in temperature range from 1.4 K to room temperature.

Raman spectra were collected at room temperature in backscattering geometry using a customary confocal micro-Raman spectrometer equipped with  $20 \times long$  working distance objective, HeNe laser as the excitation source, and a single-grating spectrograph with  $1\,\mathrm{cm}^{-1}$  resolution. Pressure was measured *in situ* with an accuracy of  $\sim 0.1$  GPa using the standard ruby fluorescence scale.

#### III. RESULTS AND DISCUSSION

# A. Suppression of structural distortion, metallization, and transition to the pyrite-type structure

 $PdS_2$  and  $PdSe_2$  at normal pressure crystalize into the unique orthorhombic  $PdS_2$ -type structure belonging to the Pbca space group [Fig. 1(a)], which is a distorted pyrite structure [Fig. 1(b)] with the  $PdS_6$  octahedra elongated along the c axis, leading to a square-planar coordination of Pd. This distortion resembles the cooperative Jahn-Teller (JT) effect

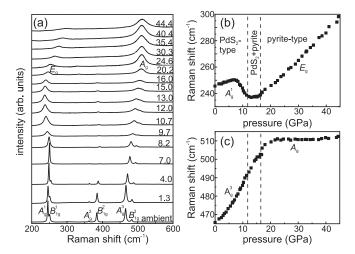


FIG. 2. (a) Selected Raman spectra of  $PdS_2$  collected at a pressure increase run. Numbers denote pressure values in GPa. Mode assignment for the  $PdS_2$ -type structure is presented at the ambient pressure spectrum, and for the pyrite-type structure, at 20.2 GPa. (b) and (c) Pressure dependence of the frequencies of the librational  $A_g^1$  mode (double degenerated  $E_g$  mode in the pyrite phase) and the  $A_g^3$  symmetric stretching mode ( $A_g$  mode in the pyrite phase) of the  $S_2^{2-}$  anions, respectively. Dashed lines indicate tentative phase boundaries drawn by analogy with  $PdSe_2$  (Appendix).

[15]. However, in the symmetrical octahedral coordination, the lowest state according to Hund's rule is the spin-triplet  ${}^{3}A_{2}$ state of the Pd  $(d^8)$  cation. This is an orbitally nondegenerated state and thus not of JT nature. Because the lowest exited states <sup>1</sup>E and <sup>1</sup>A<sub>1</sub> have spin multiplicity different from that of the ground  ${}^{3}A_{2}$  state, there is also no pseudo (secondorder) JT effect for the ground state. If the excited  ${}^{1}A_{1}$  and <sup>1</sup>E states are close in energy with sufficiently strong JT-like vibrational coupling between them [hidden pseudo-JT effect (PJT)] [16], the energy gap to the ground term might be overcome, and the distorted configuration with singlet electronic state becomes energetically favorable [Fig. 1(c)]. Such a situation is apparently realized in PdS2 and PdSe2 (as the only source of instability and distortion of the high-symmetry system configuration in the nondegenerate state [16,17]) due to strong coupling between the half-filled electronic shell system  $e^2$  and the  $E_g$  lattice mode in the undistorted pyritetype structure with symmetric octahedral coordination. This leads to the structural distortion, with electrons filling the lower  $d_{7^2}$  orbitals [Fig. 1(c)]. This configuration accounts for the diamagnetic semiconducting properties of PdS<sub>2</sub> (PdSe<sub>2</sub>) at ambient pressure conditions.

Application of external pressure eliminates the structural distortion of PdSe<sub>2</sub>, rendering the pyrite-type structure stable at pressures above 11 GPa, as established by high-pressure structural [15] and Raman spectroscopy [14] studies. According to the pressure homology principle, the same structural transition can also be expected in PdS<sub>2</sub> at higher pressures. Indeed, pressure evolution of the Raman spectra of PdS<sub>2</sub> [Fig. 2(a)] indicates high-pressure transformation to the pyrite-type structure in PdS<sub>2</sub>.

The Raman spectrum of  $PdS_2$  at the ambient pressure [Fig. 2(a)] is similar to that of bulk isostructural  $PdSe_2$  [18], with all peaks shifted to higher frequencies due to the lighter

atomic mass of the sulfur atoms. Thus, the mode assignment of all peaks can be conducted in the same manner as for PdSe<sub>2</sub> [18]. From 12 Raman active modes following from the group analysis  $(3A_{\rm g}+3B_{\rm 1g}+3B_{\rm 2g}+3B_{\rm 3g})$ , only three  $A_{\rm g}$  and three  $B_{\rm 1g}$  modes have been observed in unpolarized Raman measurements similarly to the experimental observation in related compound PdSe<sub>2</sub> [18], for which theoretical calculations revealed that only these modes have nonzero Raman tensor elements [18]. The atomic vibrations observed in the Raman spectra involve only chalcogen atoms. The  $A_{\rm g}^{\ 1}$  and  $A_{\rm g}^{\ 3}$  modes correspond to the librational and internal stretching vibrational motions of the S dumbbells respectively, whereas the other modes are combinations of these vibrations.

Application of pressure leads to the gradual merging of  $A_g^{1}$  with  $B_{1g}^{1}$  and  $A_g^{3}$  with  $B_{1g}^{3}$  peaks in course of the suppression of structural distortion and the concomitant gradual redistribution of the peak intensities in Raman spectra [Fig. 2(a)]. The high-pressure spectra are dominated by two peaks, which might be assigned to the doubly degenerated  $E_{g}$ librational and  $A_g$  symmetric internal vibrational stretching of the  $S_2^{2-}$  dimers modes in the cubic pyrite-type structure (the less intense  $T_g$  modes expected in the pyrite structure might be screened by broadened  $E_g$  and  $A_g$  peaks) owing to the similarity to the high-pressure Raman spectra of PdSe<sub>2</sub> [14]. However, the continuous gradual changes in the spectra make it difficult to determine the pressure of the structural transition. Analysis of the peculiar pressure dependence of the frequency of the librational  $A_g^1$  mode [Fig. 2(b)] allows one to determine the pressure of the structural phase transition to the pyrite-type phase.

The frequency of the librational  $A_g^{\ 1}$  mode initially increases up to a pressure of 7 GPa (at the rate of  $\sim$ 0.7 cm<sup>-1</sup>/GPa), demonstrating normal behavior upon compression. The mode frequency starts to decrease rapidly upon further pressure increases of up to  $\sim$ 12 GPa, and becomes nearly pressure-independent in the pressure range of 12-16 GPa. Above 16 GPa, the frequency starts monotonously increasing again up to the highest applied pressure in this experiment. Such peculiar dependence of the frequency on pressure in PdS<sub>2</sub> is qualitatively similar to that observed for the same lattice mode in PdSe<sub>2</sub> (see the Appendix). Thus, by analogy with PdSe2, the pressure of 16 GPa, at which the librational frequency starts increasing again, delimits the accomplishment of the structural transition to the pyrite-type phase in PdS<sub>2</sub>, whereas both PdS<sub>2</sub>-type and pyrite-type phases coexist in the pressure range of 12–16 GPa.

The softening of the librational  $A_g^1$  mode frequency above 7 GPa indicates an increase in the intralayer Pd-S distances, similar to what has been established for the intralayer Pd-Se distances in PdSe<sub>2</sub> [15] (see the Appendix). Such atomic displacement corresponds to the filling of the  $d_{x^2-y^2}$  orbitals of the Pd atoms [19]. Indeed, the interlayer cation-anion distances decrease sharply upon compression [15], leading to strong enhancement in the overlap repulsion between the filled  $d_{z^2}$  orbitals of the Pd cation and the  $\pi^*$  shell of the  $S_2^{2-}$  anion from the adjacent layer. This enhanced repulsion modifies the energy landscape [Fig. 1(c)], decreasing the energy difference  $\Delta E$  and barrier height B and enabling the tunneling between the singlet and triplet electronic terms. Thus, both distorted

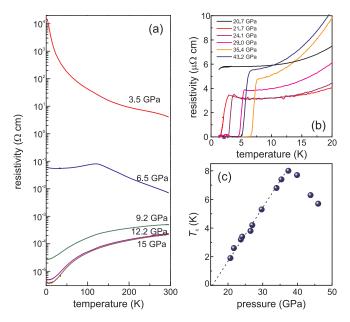


FIG. 3. (a) Application of pressure tunes the electronic properties of  $PdS_2$  from a semiconductor to a metal at pressures above 7 GPa. (b) In the high-pressure pyrite phase of  $PdS_2$ , superconductivity emerges with  $T_c$  showing a dome-shaped dependence on pressure (c) with the maximum  $T_c$  of 8.0 K at a pressure of  $\sim 37.4$  GPa.

(with the doubly occupied  $d_{z^2}$  orbital) and undistorted (with the electrons distributed between the  $d_{z^2}$  and  $d_{x^2-y^2}$  orbitals) configurations coexist dynamically in this situation. This leads to a change in the electronic transport properties of PdS<sub>2</sub>: Semiconducting at low pressures, PdS<sub>2</sub> shows metallic behavior in the whole temperature range of 1.4–300 K at pressures above 7 GPa [Fig. 3(a)]. PdS<sub>2</sub> remains a normal metal (with no superconductivity down to 1.4 K) up to the structural transition to the pyrite-type phase at 16 GPa [Fig. 3(a)].

Such behavior is similar to that of PdSe<sub>2</sub> (see the Appendix) undergoing semiconductor-to-metal transition at 3 GPa within the PdS<sub>2</sub>-type structure [14] concomitantly with the onset of the softening of the  $A_g^{-1}$  mode (Appendix). However, in contrast to PdSe2 showing a softening of the internal stretching  $A_g^3$  mode of the Se dimers (elongation of Se-Se bonds) at pressures above 3 GPa (Appendix), only a monotonic hardening of the stretching mode, without any anomalies due to S-S dimer elongation, is revealed in PdS<sub>2</sub> up to the transition in the pyrite-type structure [Fig. 2(c)]. Thus, the metallization in  $PdX_2$  (X = S, Se) is driven by the population of the  $d_{x^2-y^2}$  orbitals of Pd (which are strongly hybridized with chalcogen p orbitals, forming the conduction band [14]), caused by the suppression of the structural distortion (hidden PJT effect). This might be considered as evidence of orbital selective metallization: While electrons in the  $d_{z^2}$  orbitals remain localized due to persisting structural distortion, the filling of the  $d_{x^2-v^2}$  orbitals drives the semiconductor-to-metal transition.

# B. Superconductivity in the high-pressure pyrite-type phase of PdS<sub>2</sub>

Similar to PdSe<sub>2</sub> [14], PdS<sub>2</sub> in the high-pressure pyritetype structure becomes superconducting at low temperatures [Fig. 3(b)]. The resistivity curve at  $\sim$ 20.7 GPa shows a drop in the resistivity at temperature below  $\sim$ 2 K [Fig. 3(b)], suggesting the onset of a superconducting transition. The superconducting state is clearly observed at higher pressures, with the resistivity dropping abruptly to zero.

The critical temperature of superconductivity  $(T_c)$  demonstrates a well-defined dome-shaped dependence on pressure. Initially,  $T_c$  increases as the pressure increases [Fig. 3(c)] at a rate of  $\sim 0.35$  K/GPa (noticeably, this rate is considerably lower than ~1.2 K/GPa observed in the sister compound  $PdSe_2$  [14]). The increase of the  $T_c$  with pressure in  $PdS_2$  falls within the wide range of dependences observed in transition metals superconductors where a positive sign of  $dT_c/dP$ arises from a rapid increase of the Hopfield parameter  $\eta =$  $N(E_{\rm f})\langle I^2\rangle$  [where  $N(E_{\rm f})$  is the electronic density of states at the Fermi energy and  $\langle I^2 \rangle$  is the average electronic matrix element] [20]. The maximum  $T_c$  value of 8.0 K is reached at a pressure of 37.4 GPa. Above this pressure and up to the highest experimental pressure of 46 GPa, T<sub>c</sub> decreases continuously. The dome-shaped dependence of  $T_c$  is conditioned apparently by the completion of the pressure-induced increase of the Hopfield parameter (reflecting the character of the electrons near Fermi energy) and the stiffening of the lattice vibration spectrum. The later effect dominates apparently at high pressure leading to decrease of  $T_c$  at high pressures. Theoretical calculations of the electronic structure and electron-phonon coupling in dependence on pressure are necessary to understand the dome-shaped dependence of  $T_c$ on pressure.

Thus, the electronic transport properties of PdS<sub>2</sub> in the high-pressure pyrite-type phase are qualitatively similar to those of PdSe<sub>2</sub> [14]. However, in contrast to PdSe<sub>2</sub> which shows an obvious correlation between pressure dependences of  $T_c$  and the bonding strength of Se–Se dumbbells [14], the anions' internal vibration frequency in the pyrite-phase of PdS<sub>2</sub> is nearly independent of pressure, without any anomalies due to the softening of the S-S bonding [Fig. 2(c)]. This behavior of the frequency of the internal vibration of the  $S_2^{2-}$ anions in  $PdS_2$  is similar to the dependence of the  $A_g$  mode frequency on pressure in pyrite NiS<sub>2</sub> [9]. Thus, the destabilization of the chalcogen dumbbells is not the necessary parameter leading to the emergence of superconductivity in the pyrite compounds. However,  $T_c$  for pyrite superconductors PdSe<sub>2</sub> and Ir<sub>0.94-x</sub>Rh<sub>x</sub>Se<sub>2</sub> shows striking correlations with the bond strength of the  $Se_2^{2-}$  anions [11,14]. The bond softening is a result of charge transfer from the cation to Se<sub>2</sub><sup>2-</sup> dimers [10] leading to changes in the electronic system (effective "hole self-doping") and thus influencing the  $T_c$  value and provoking the correlation between  $T_c$  and the bond strength of the anions.

### C. Nonsuperconducting pyrite NiSe<sub>2</sub> under high pressure

NiS<sub>2</sub> and NiSe<sub>2</sub>, which are isoelectronic with PdS<sub>2</sub> and PdSe<sub>2</sub>, crystalize at ambient pressure in the undistorted pyrite structure, with octahedral coordination of the Ni cations in the high spin state. It is proposed that strong on-site repulsion U in the doubly occupied  $d_{z^2}$  orbitals of Ni prevents the structural distortion to the PdS<sub>2</sub>-type structure in the Ni dichalcogenides [15]. NiS<sub>2</sub> is a typical strongly correlated antiferromagnetic insulator undergoing the insulator-to-metal transition under

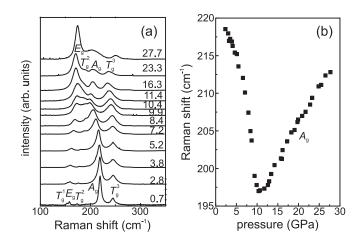


FIG. 4. (a) Selected Raman spectra of NiSe<sub>2</sub> collected at a pressure increase run. Numbers denote pressure values in GPa. Mode assignment for the pyrite-type structure is presented for spectra at 0.7 and 23.3 GPa. (b) Pressure dependence of the frequency of the symmetrical stretching  $A_g$  mode of the Se<sub>2</sub><sup>2-</sup> anion.

application of pressures of  $\sim 2.5-3$  GPa or by Se substitution in NiS<sub>2-x</sub>Se<sub>x</sub> at  $x \sim 0.6$  by preserving the cubic pyrite-type structure [21]. NiSe<sub>2</sub> is a normal metal at ambient conditions [22]. Remarkably, theoretical calculations revealed the anomalous behavior (elongation) of the Se–Se bond length in NiSe<sub>2</sub> under compression [8]. It is interesting to prove this finding experimentally and to check if this bonding instability leads to the pressure-induced superconductivity in NiSe<sub>2</sub>.

The Raman spectra of NiSe<sub>2</sub> in the diamond anvil cell at low pressures [Fig. 4(a)] are in good agreement with data from the literature [23]. All five Raman-active modes  $(A_g + E_g + 3T_g)$  predicted by factor group analysis for a pyrite-type structure could be identified in the spectra. Application of pressure led to shifting of the observed peaks and redistribution of intensities [Fig. 4(a)]. However, all Raman-active modes of the pyrite structure could be identified up to the highest pressures used in this study [Fig. 4(a); the low-frequency  $T_g^1$  and  $E_g$  lines merge into a single band, as in NiS<sub>2</sub>, under pressure [9], indicating structural stability of the pyrite-type structure of NiSe<sub>2</sub> at least up to pressures of ~30 GPa.

Interestingly, the frequency of the internal vibrations ( $A_g$  mode) of the Se dimers demonstrates anomalous softening under compression [Fig. 4(b)], which is in agreement with the theoretically predicted elongation of the Se–Se distance [8]. This elongation is caused by electron transfer from the Ni d orbitals to the antibonding  $\sigma^*$  states of the Se<sub>2</sub><sup>2-</sup> dumbbells [10]. The  $A_g$  mode softening in NiSe<sub>2</sub> is qualitatively similar to the Se–Se bond softening in the high-pressure pyrite phase of PdSe<sub>2</sub> [14]. Remarkably, the intensity of the  $E_g$  mode increases relative to the intensity of the  $A_g$  mode as pressure increases [Fig. 4(a)], which, once again, shows behavior similar to that observed in the pyrite phase of PdSe<sub>2</sub> [14] with the softening of Se–Se bonds, but is in striking contrast to those in the pyrite phase of PdS<sub>2</sub> [Fig. 2(a)] and pyrites FeS<sub>2</sub> [24] and NiS<sub>2</sub> [9] with the rigid S<sub>2</sub><sup>2-</sup> dumbbells.

Thus, NiSe<sub>2</sub> demonstrates Se–Se bond instability under pressure, similar to the superconducting pyrite phase of PdSe<sub>2</sub>, in which the dependence of  $T_c$  on pressure shows

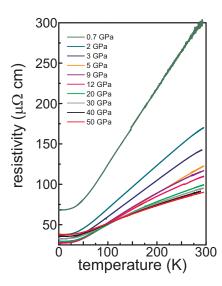


FIG. 5. Temperature dependence of the electrical resistivity of NiSe<sub>2</sub> at different pressures.

a striking correlation with the bonding strength of Se–Se dumbbells [14]. Despite this similarity in lattice dynamics, NiSe<sub>2</sub> remains a normal metal up to 50 GPa (Fig. 5), with no evidence of superconducting transition at  $T_c$  values above 1.4 K. This indicates, once again, that the structural instability due to the destabilization of the chalcogen dumbbells is not the sufficient condition governing the appearance of superconductivity in compounds with a pyrite structure.

It might be assumed that strong electron correlations in Ni prevent superconductivity in NiSe<sub>2</sub>. However, because superconductivity at relatively high  $T_c$  is observed in the pyrite compounds of 4d and 5d transition metals [11–14], the spin-orbit coupling might play an important role in superconductivity in these compounds. So for PdSe2, the ab initio electronic band structure calculations indicate the presence of Dirac and nodal-line fermions in the vicinity of the Fermi energy protected by the pyrite structure symmetry, which can lead to interesting superconducting states [14]. Because superconductivity with qualitatively similar pressure behavior is observed in PdS2 and PdSe2, independent of different anion bond behaviors, it might be suggested that proximity to the structural instability, which leads to the distortion of the pyrite structure (hidden PJT effect), is important for the emergence of superconductivity in the high-pressure pyrite-type phases of Pd dichalcogenides. This structural instability is caused by the strong coupling between the half-filled  $e_{\rm g}$  electronic shell and librational  $E_{\rm g}$ lattice mode in the pyrite structure. This strong coupling manifests in correlation between evolution of the electronic transport properties and anomalous pressure dependence of the librational modes in PdS2 and PdSe2. Therefore, it might be suggested that strong electron-phonon coupling between electrons populating  $e_g$  shells and librational  $E_g$ modes leads to the superconductivity in Pd compounds pyrite structure. It is suggested to be general for pyrite superconductors since the superconductivity in compounds with pyrite structure was observed only in compounds with partially filled electronic  $e_{\rm g}$  orbitals, while the pyrite

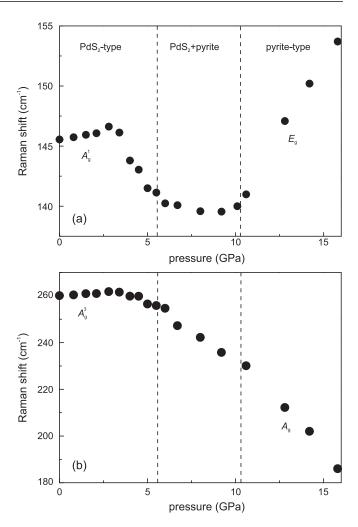


FIG. 6. Pressure dependence of the frequencies of (a) the librational  $A_{\rm g}^{\rm 1}$  mode (double degenerated  $E_{\rm g}$  mode in the pyrite phase) and (b)  $A_{\rm g}^{\rm 3}$  symmetric stretching mode ( $A_{\rm g}$  mode in the pyrite phase) of the Se<sub>2</sub><sup>2-</sup> anions in PdSe<sub>2</sub>. Dashed lines indicate the phase boundaries from high-pressure structural studies of PdSe<sub>2</sub> [15].

compounds with empty  $e_{\rm g}$  orbitals (like RuSe<sub>2</sub>, OsSe<sub>2</sub> and OsTe<sub>2</sub>) undergoing metal to insulator transition at ~20 GPa without any anomalies of the lattice vibrations, remain nonsuperconducting up to pressures beyond 50 GPa [25]. This suggestion however, needs to be proven by studies of lattice dynamic of other pyrite superconductors like  ${\rm Ir}_{0.94-x}{\rm Rh}_x{\rm Se}_2$  in dependence on x and pressure and theoretical calculations.

### IV. CONCLUSIONS

Raman spectroscopy studies of  $PdS_2$  reveal a structural phase transition from the  $PdS_2$  type to the pyrite-type structure at a pressure of approximately 16 GPa. Application of pressure drastically changes the electronic transport properties of  $PdS_2$ . A semiconductor at ambient pressure,  $PdS_2$  becomes a metal at pressures above 7 GPa. The metallization is concomitant with a softening of the vibrational mode due to increasing intraplane cation-anion distances, indicative of the population of the  $PdC_2$  orbitals. Superconductivity emerges in  $PdS_2$ 

after the structural transition to the pyrite-type phase above 16 GPa.  $T_c$  demonstrates dome-shaped dependence on pressure, with the maximum  $T_c$  value being 8.0 K at 37.4 GPa. Thus, the pressure evolution of the electronic transport properties of PdS<sub>2</sub> is qualitatively similar to that of PdSe<sub>2</sub>. Despite this similarity in electronic properties, correlating with  $T_c$ anomalous softening of the anion dumbbells in PdS<sub>2</sub> was not observed, in contrast to PdSe<sub>2</sub>. On the other hand, the Se–Se bond instability was clearly observed in pyrite NiSe<sub>2</sub> under pressure while it remains a normal nonsuperconducting metal. These observations imply insufficiency of a direct correlation between structural instability due to the destabilization of chalcogen dimers and the emergence of superconductivity in the pyrite compounds. Because the electronic properties of the pyrite compounds are depending on the bond strength of the anions, the  $T_c$  in these compounds might be further tuned by tuning the nonmetal-nonmetal bond strength by application of chemical or physical pressure. However, the superconductivity in these compounds is driven most likely by other structural instabilities. The proximity to a structural distortion due to the strong electron-lattice coupling (hidden PJT effect) in PdS2 and PdSe2 might be viewed as the structural instability driving the emergence of superconductivity in the high-pressure pyrite phases of the Pd dichalcogenides.

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# APPENDIX: LATTICE DYNAMICS AND STRUCTURAL TRANSITION TO THE PYRITE-TYPE STRUCTURE IN PdSe<sub>2</sub>

The Raman spectroscopy studies [14] indicate pressureinduced structural phase transition to the pyrite-type structure, which is in good agreement with a structural investigation of PdSe<sub>2</sub> [15]. Pressure dependence of the frequencies of the  $A_g^{\ 1}$  librational (the doubly degenerated  $E_g$  mode in the pyrite-type structure) and the  $A_g^3$  internal stretching of the modes of the Se<sub>2</sub> dumbbells in the PdS<sub>2</sub>-type structure in the pressure region of the structural phase transition are shown in Fig. 6. The frequency of the  $A_g^{-1}$  mode increases slightly when the pressure increases up to  $\sim$ 3 GPa. Above this pressure, the frequency shows significant softening up to a pressure of  $\sim$ 6 GPa. The frequency softening is in agreement with the increase in the intralayer Pd-Se distances observed in the structural studies [15]. Remarkably, at a pressure of  $\sim$ 3 GPa, at which the softening of the  $A_g^{-1}$  mode begins, PdSe<sub>2</sub> undergoes a semiconductor-to-metal transition [14], indicating strong charge-lattice coupling. In a pressure range of 6–11 GPa, the  $A_g^{-1}$  mode is nearly independent of pressure [Fig. 6(a)]. According to high-pressure structural studies, the PdS<sub>2</sub>-type and pyrite-type phases coexist in this pressure range [15]. The  $A_g^{-1}$  mode frequency starts increasing continuously at pressures above 11 GPa [Fig. 6(a)] indicating the completeness of transition to the pyrite-type structure, which is also in agreement with the results of past structural studies [15]. Notably, the frequency of the internal vibrations of  $Se_2^{2-}$ dimers also demonstrates softening at pressures above 3 GPa [Fig. 6(b)]. There are no discontinuities in the pressure dependence of the frequency in this mode at the transition to the pyrite-type structure [Fig. 6(b)] and the pressure dependence of this mode frequency correlates further with the critical temperature of the superconductivity emerging in the pyrite structure [14].

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