Oxygen isotope effect in Bi₂Sr₂Ca_{n-1}Cu_nO_{2n+4+ δ} (n=1,2,3) single crystals

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We report measurements of the oxygen isotope effect in $\text{Bi}_2\text{Sr}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2n+4+\delta}$ (n=1,2,3) single crystals. For optimally doped materials the isotope exponent is found to decrease with increasing the number of CuO_2 layers in a manner of inversely correlating with the superconducting transition temperature in this homologous family. This behavior is in contrast to the general belief that the isotope effect can be negligible for the optimally doped cuprates. Our results highlight the important role played by phonons and interlayer coupling in the high-temperature superconductivity.

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The coupling mechanism which causes high-temperature superconductivity in cuprates remains a challenging issue in condensed matter physics. Although it is generally believed that antiferromagnetism is important for superconductivity, there has been growing experimental evidence^{1–5} for the role of phonons in the electronic state. Isotope effects provide a good indication for electron-phonon interaction. Gweon et al.³ reported a large isotope effect on the electronic structure involving shifts of the band bottom as large as 10-40 meV in optimally doped Bi₂Sr₂CaCu₂O_{8+δ} (Bi2212) using angleresolved photoemission spectroscopy (ARPES), indicating that the coupling between the lattice and electrons plays an important role. Douglas et al.4 in contract found no large isotope effect in the same material, but concluded that their measurements are not inconsistent with a conventional isotope shift of the order of 3 meV. Based on d^2I/dV^2 imaging measurements using scanning tunneling microscopy (STM), Lee et al.⁵ obtained a bosonic mode of Bi2212 irrespective of doping level with energy of 52 meV which was identified as a lattice mode from the observed 6% energy shift upon substitution for ¹⁶O by ¹⁸O. The fact that their samples are poorly characterized because the superconducting transition temperature T_c is 76 K for the 16 O sample and 88 K for the ¹⁸O sample, far beyond the expected change solely through the isotope exchange, leaves a possibility for other interpretations. On the other hand, an inelastic neutron-scattering experiment⁶ showed a negligible shift of the magnetic resonance energy in YBa₂Cu₃O_{6.89} upon oxygen isotope substitution ($^{16}O \rightarrow ^{18}O$), thus placing a severe limit on the role of electron-phonon coupling in cuprate superconductors.

The isotope effect on T_c as a direct experimental probe in revealing electron-phonon coupling supported phonon-mediated pairing in conventional superconductors. Such studies in cuprate superconductors, particularly in YBa₂Cu₃O_{7- δ} and La_{2-x}Sr_xCuO₄, have established that T_c exhibits a weak shift upon isotope substitution at optimal doping where T_c is maximal. Recent experiments suggested a sharp transition in quasiparticle dynamics precisely at optimal doping. Neutron diffraction data slo showed a sharp change of the in-plane Cu–O bond distribution at op-

timal doping, underscoring a strong electron-phonon coupling. It is apparently crucial to investigate superconducting properties at this magic doping level. The striking feature of cuprate superconductors is the appearance of the CuO₂ plane. Even at optimal doping, T_c initially increases with increasing the number of CuO₂ layers within a unit cell in a homologous layered system. It is still unknown how the CuO₂-layer number affects the isotope effect in a homologous series. Although mercury- and thallium-based families have relatively high T_c 's, the oxygen-isotope studies in both systems are very difficult due to the complexity of the selfdoping defect involving the partial substitution of copper for mercury¹¹ or thallium.¹² Among the various layered families discovered so far, the bismuth-based series then becomes the only promotional candidate for such a purpose. Unlike other families, all members of the bismuth-based family have high-quality single crystals available. Moreover, most systematic data of ARPES1-4 and STM5 were collected in bismuth-based single crystals, allowing direct comparison for clarifying the phonon contribution to superconductivity.

In this work we report measurements of the oxygen isotope effect in $\mathrm{Bi_2Sr_2Ca_{n-1}Cu_nO_{2n+4+\delta}}$ (n=1,2,3) single crystals. We find that the optimally doped monolayer material has a sizeable oxygen isotope exponent α . However, α decreases monotonically with the number of $\mathrm{CuO_2}$ layers in a way opposite to T_c in this homologous family at optimal doping. This behavior directly points to the importance of electron-phonon interactions as well as the interlayer coupling effect in layered cuprates.

Single crystals of $Bi_2Sr_{1.6}La_{0.4}CuO_{6+\delta}$ (Bi2201), $Bi_2Sr_2CaCu_2O_{8+\delta}$ (Bi2212), and $Bi_2Sr_2Ca_2Cu_3O_{10+\delta}$ (Bi2223) were grown by the traveling-solvent floating-zone method with very slow growth rates and steep temperature gradients, as described in detail elsewhere. ^{13–15} Plate-shaped crystals were cleaved from crystal ingots. The purity and crystallinity of the samples were checked out by x-ray diffraction (Fig. 1). The fact that only sharp (00*l*) reflection peaks can be observed in each pattern confirms the high crystalline quality and phase purity of these samples. The three Bi-based cuprates used for oxygen isotope studies have simi-

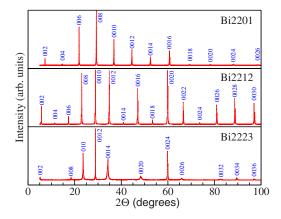


FIG. 1. (Color online) Single-crystal x-ray diffraction patterns of nearly optimally doped $Bi_2Sr_{1.6}La_{0.4}CuO_{6+\delta}$, $Bi_2Sr_2CaCu_2O_{8+\delta}$, and $Bi_2Sr_2Ca_2Cu_3O_{10+\delta}$.

lar structures of the tetragonal I4/mmm space group $(a \approx b \approx 5.4 \text{ Å})$. Their unit cells differ only in the number (n-1) of $\text{CuO}_2\text{-Ca-CuO}_2$ slabs packed along the c axis with lattice parameters of 24.33(7), 30.80(6), and 37.07(7) Å for the monolayer, bilayer, and trilayer Bi-based cuprates, respectively.

The c-axis length in Bi-based cuprates has been found to be a sensitive measure of oxygen doping level. 15,16 The c-axis length monotonically decreases with increasing δ in these materials. For the monolayer and bilayer cuprates, T_c increases with decreasing c-axis length, reaching its maximum, and then decreases with decreasing c-axis length, reflecting the bell-shaped doping dependence of T_c . In the case of Bi2223, T_c first increases with decreasing c-axis length similar to other members of this family. However, once the maximum value is achieved T_c remains fixed irrespective of the further decrease of the c-axis length. Measurements of optical¹⁵ and transport¹⁶ properties clearly show that the carrier has been doped continuously even in the constant T_c region. Our measured c-axis lattice parameters for Bi2201 and Bi2223 indicate that these materials are in reality optimally doped, although the maximum T_c 's are scattered for both materials in the literature. 15–17 The Bi2212 sample was found to be nearly optimally doped but in the underdoped regime based on its c-axis length. Using the correlation between δ and c-axis length difference, ¹⁶ we may estimate the oxygen content $\delta \sim 0.12$ for our present Bi2212 sample. The oxygen content δ of the present Bi2201 samples was estimated to be 0.36 as described previously, 14 and the value of δ for Bi2223 is believed to be \sim 0.28.

For the oxygen isotope substitution, we chose each pair of Bi-based cuprate single crystals. The selected samples were annealed in isotopically enriched oxygen under identical conditions following a similar procedure described in Ref. 18. Each sample of crystals of Bi-based cuprates was cut into halves. A pair of them was sealed in separated quartz tubes. One tube was charged with high-purity ¹⁶O gas while the other was charged with 99% ¹⁸O enriched oxygen. All exchange procedures were performed simultaneously in two adjacent quartz tubes at 650 °C and 1 bar oxygen pressure for 96 h to ensure the same content. Then both tubes were

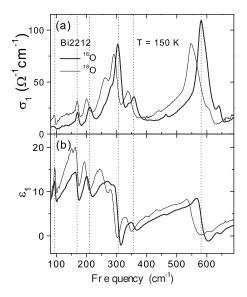


FIG. 2. The *c*-polarized response of (a) the real part of the optical conductivity $\sigma_1(\omega)$ and (b) the dielectric function $\epsilon_1(\omega)$ for ^{16}O and ^{18}O exchanged pairs of nearly optimally doped $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$ measured at 150 K (Ref. 19).

slowly cooled to room temperature at a rate of 20° /h in order to oxidize them completely. The annealing was repeated with a fresh charge of ¹⁸O. Estimates based on the mass change upon isotope replacement indicate that the ¹⁸O contents were 80(5)% in the exchanged samples.

The isotope exchange process was examined by very accurate ellipsometric measurements performed at the U4IR beamline of the National Synchrotron Light Source at Brookhaven National Laboratory detailed in Ref. 19. Figure 2 shows the real parts of the c-axis optical conductivity $\sigma_1(\omega)$ and the dielectric function $\epsilon_1(\omega)$ for the Bi2212 single crystals for ¹⁶O and ¹⁸O substitution measured at 150 K and in the energy range from about 100 to 700 cm⁻¹. Six main peaks with energies of 97, 168, 210, 304, 358, and 583 cm⁻¹ are observed in the ¹⁶O standard sample. The peak assignments¹⁹ are as follows: The two low-frequency modes at 97 and 168 cm⁻¹ arise from motion of the heavy Bi atoms and the Cu ions, respectively. The A_{2u} mode at 210 cm⁻¹ is due to the vibrations of Ca and O(1) atoms in the Cu-O planes, and from the O(3) atoms in the Bi-O planes to the caxis oscillations. The three high-energy vibrations are predominantly oxygen related. The mode at 304 cm⁻¹ is basically due to O(3) and O(1) vibrations, with a predominant contribution from the O(3) atoms. The Cu-O bond-bending character mode at 358 cm⁻¹ involves significant contribution from the O(1) atoms. The highest frequency mode at 583 cm⁻¹ is mainly from the vibrations of O(2) apical atoms in the Sr-O planes. These phonon modes are clearly identified especially due to the resonance features in $\epsilon_1(\omega)$ as shown in Fig. 2(b).

Upon 18 O substitution, the two low-frequency modes at 97 and 168 cm⁻¹ show no shift at all, thus confirming their heavy metal ion feature. With 18 O exchange the four A_{2u} phonons at 210, 304, 358, and 583 cm⁻¹ soften in frequency by 4.1, 4.4, 5.3, and 5.5%, respectively, yielding an average isotope exchange ratio consistent with the estimate from

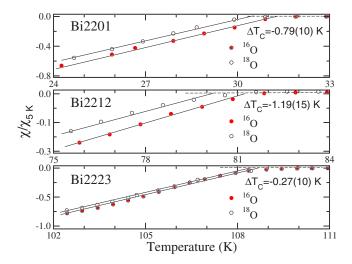


FIG. 3. (Color online) Temperature dependence of the magnetic susceptibility χ (normalized to the value at 5 K) for ^{16}O and ^{18}O exchanged pairs in nearly optimally doped $Bi_2Sr_{1.6}La_{0.4}CuO_{6+\delta}$, $Bi_2Sr_2CaCu_2O_{8+\delta}$, and $Bi_2Sr_2Ca_2Cu_3O_{10+\delta}$.

mass change. The relatively large frequency shifts of the 358 and 583 cm⁻¹ modes suggest that the O(1) site in the Cu-O planes and the apical O(2) site are nearly fully occupied by ¹⁸O. The observed sizeable redshift of the 210 cm⁻¹ mode upon ¹⁸O substitution further indicates that this mode indeed contains appreciable contribution from the O(1) and O(3) atoms. It is also interesting to note that there are two more small peaks at 465 and 630 cm⁻¹. Raman studies²⁰ on Bi2212 have shown that the 465 cm⁻¹ mode is associated to *c*-axis vibration of O(3) atoms in the Bi-O layers and the 630 cm⁻¹ mode to the O(2) apical oxygen atoms in the Sr-O layers. It can be seen that in the ¹⁸O exchanged sample these two peaks all move to low frequency by 4.7% again consistent with 80% ¹⁸O exchange.

Superconducting transition temperatures were determined by magnetic susceptibility χ in a Quantum Design superconducting quantum interference device magnetometer. All samples were measured in an applied field of 10 Oe with $H \| c$. To compare the temperature dependence of the isotope pairs, the data points are normalized by taking χ at each value of temperature and dividing by χ at 5 K. In Fig. 3, we show the normalized zero-field-cooled susceptibility versus temperature curves for the ¹⁶O and ¹⁸O exchanged samples. T_c was taken as the temperature where the linearly extrapolated transition slope intersects the zero line, which has been generally used in isotope-effect studies.²¹ We find a lower T_c for all the ¹⁸O exchanged samples. Oxygen isotope shift of ΔT_c was obtained to be -0.79(10), -1.19(15), and -0.27(15) K for the Bi2201, Bi2212, and Bi2223, respectively.

Using the definition of the isotope exponent $\alpha = -(\Delta T_c/T_c)/(\Delta M/M)$ with M being the oxygen isotopic mass, we have $\alpha = 0.25(4)$ for Bi2201, 0.14(3) for Bi2212, and 0.02(1) for Bi2223. The present α of Bi2201 is the first determined for the isotope effect in this material, and its value lies in the α range for other optimally doped low- T_c monolayer compounds. This observation indicates that the optimally doped monolayer cuprates indeed possess a non-

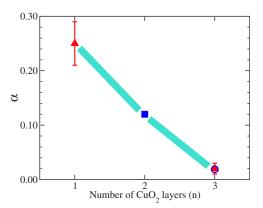


FIG. 4. (Color online) Dependence of the oxygen isotope exponent α on the number of CuO₂ layers in the optimally doped Bibased cuprates. Triangles are the experimental data points in the present work. The square denotes the data of the Bi₂Sr₂CaCu₂O_{8+ δ} sample having T_c =92 K from Refs. 3 and 24. The circle represents the data of the optimally doped Bi_{1.6}Pb_{0.4}Sr₂Ca₂Cu₃O_{10+ δ} with T_c ~ 107 K from Ref. [23]. The thick lines are to guide the eyes.

negligible isotope effect. The trilayer Bi2223 has a very small but positive oxygen isotope effect. This finding concludes the early controversy regarding whether the oxygen isotope effect in this material is negative. 22,23 Importantly, the samples used in our experiments are single crystals and lead-free. Lead-containing powder samples were used in previous experiments. ^{22,23} In the case of Bi2212, Gweon *et al.*³ noticed that an $^{16}\text{O} \rightarrow ^{18}\text{O}$ substitution leads to T_c shifting from 92 to 91 K. Taking the exchange rate of their ¹⁸O sample of 75%, 3,24 one would have $\alpha = 0.12$ for the optimally doped Bi2212, which is close to but smaller than our result for the nearly optimally doped sample on the underdoped side. Previous isotope studies on the overdoped side give α =0±0.01 and 0.03 for samples having T_c =84.7 K (Ref. 18) and 75.5 K,²⁵ respectively. One then readily gets the information on how α evolves around optimal doping in Bi2212: α slightly decreases with doping, reaching a minimum on the overdoped side, and then increases for heavily overdoped materials. In Fig. 4, we plot the CuO_2 layer dependence of α for the optimally doped Bi-based materials. It is quite clear that the oxygen isotope exponent α systematically decreases with the increase of the number of CuO₂ layers, which is the central message provided in the present work.

The sizable α of the monolayer Bi2201 indicates that phonons are directly or indirectly involved in determining superconducting state. Just like the T_c behavior, α also has a systematic CuO_2 layer dependence in the Bi-based cuprates. Therefore the same physical parameter(s) should control the Cu-O layer effect on both the T_c and α . Keeping in mind that the trilayer material has the highest T_c in this homologous series but it has very small isotope effect. Clearly, the argument that a mechanism other than an electron-phonon interaction dominates the superconductivity based only on a small α in a cuprate with a relatively high T_c is inappropriate. The observed CuO_2 -layer dependence of the isotope effect indicates that the interlayer coupling between the adjacent CuO_2 planes is necessary for superconductivity in layered cuprates. For the monolayer materials having lower

 T_c 's, the interlayer coupling plays a less important role and T_c can be mainly controlled by the phonon coupling, yielding a larger size of isotope exponent. With increasing the number of CuO_2 layers in a unit cell, the interlayer coupling begins to play an important role in enhancing T_c and the isotope effect is expected to be small.²⁶

In summary, our experimental finding of the non-negligible oxygen isotope effect in optimally doped $\mathrm{Bi}_2\mathrm{Sr}_2\mathrm{Ca}_{n-1}\mathrm{Cu}_n\mathrm{O}_{2n+4+\delta}$ (n=1,2,3) highlights the role of phonons played in the high- T_c superconductivity. Most important is the systematic reduction of the isotope exponent with the number of CuO_2 layers in a way opposite to T_c in

this homologous family. Our results also suggest that the interlayer coupling effect should be included in any quantitative analysis of layered cuprates.

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- ¹ A. Lanzara, P. V. Bogdanov, X. J. Zhou, S. A. Keller, D. L. Feng, E. D. Lu, T. Yoshida, H. Eisaki, A. Fujimori, K. Kishio, J. I. Shimoyama, T. Noda, S. Uchida, Z. Hussain, and Z.-X. Shen, Nature (London) 412, 510 (2001).
- ²T. Cuk, F. Baumberger, D. H. Lu, N. Ingle, X. J. Zhou, H. Eisaki, N. Kaneko, Z. Hussain, T. P. Devereaux, N. Nagaosa, and Z.-X. Shen, Phys. Rev. Lett. **93**, 117003 (2004).
- ³G. H. Gweon, T. Sasagawa, S. Y. Zhou, J. Graf, H. Takagi, D. H. Lee, and A. Lanzara, Nature (London) **430**, 187 (2004).
- ⁴J. F. Douglas, H. Iwasawa, Z. Sun, A. V. Fedorov, M. Ishikado, T. Saitoh, H. Eisaki, H. Bando, T. Iwase, A. Ino, M. Arita, K. Shimada, H. Namatame, M. Taniguchi, T. Masui, S. Tajima, K. Fujita, S. Uchida, Y. Aiura, and D. S. Dessau, Nature (London) 446, E5 (2007).
- ⁵ J. Lee, K. Fujita, K. McElroy, J. A. Slezak, M. Wang, Y. Aiura, H. Bando, M. Ishikado, T. Masui, J.-X. Zhu, A. V. Balatsky, H. Eisaki, S. Uchida, and J. C. Davis, Nature (London) 442, 546 (2006).
- ⁶S. Pailhès, P. Bourges, Y. Sidis, C. Bernhard, B. Keimer, C. T. Lin, and J. L. Tallon, Phys. Rev. B **71**, 220507(R) (2005).
- ⁷J. P. Franck, in *Physical Properties of High Temperature Super-conductors IV*, edited by D. M. Ginzberg (World Scientific, Singapore, 1994), p. 189.
- ⁸F. F. Balakirev, J. B. Betts, A. Migliori, S. Ono, Y. Ando, and G. S. Boebinger, Nature (London) 424, 912 (2003).
- ⁹N. Gedik, M. Langner, J. Orenstein, S. Ono, Y. Abe, and Y. Ando, Phys. Rev. Lett. **95**, 117005 (2005).
- ¹⁰E. S. Bozin, G. H. Kwei, H. Takagi, and S. J. L. Billinge, Phys. Rev. Lett. **84**, 5856 (2000).
- ¹¹ J. L. Wagner, P. G. Radaelli, D. G. Hinks, J. D. Jorgensen, J. F. Mitchell, B. Dabrowski, G. S. Knapp, and M. A. Beno, Physica C 210, 447 (1993).
- ¹²Y. Shimakawa, Y. Kubo, T. Manako, H. Igarashi, F. Izumi, and H.

- Asano, Phys. Rev. B 42, 10165 (1990).
- ¹³B. Liang and C. T. Lin, J. Cryst. Growth **267**, 510 (2004); **237-239**, 756 (2002).
- ¹⁴C. T. Lin, M. Freiberg, and E. Schonherr, Physica C **337**, 270 (2000).
- ¹⁵B. Liang, C. Bernhard, Th. Wolf, and C. T. Lin, Supercond. Sci. Technol. 17, 731 (2004).
- ¹⁶T. Fujii, I. Terasaki, T. Watanabe, and A. Matsuda, Phys. Rev. B 66, 024507 (2002).
- ¹⁷ M. Karppinen, S. Lee, J. M. Lee, J. Poulsen, T. Nomura, S. Tajima, J. M. Chen, R. S. Liu, and H. Yamauchi, Phys. Rev. B **68**, 054502 (2003).
- ¹⁸D. J. Pringle, G. V. M. Williams, and J. L. Tallon, Phys. Rev. B 62, 12527 (2000).
- ¹⁹N. N. Kovaleva, A. V. Boris, T. Holden, C. Ulrich, B. Liang, C. T. Lin, B. Keimer, C. Bernhard, J. L. Tallon, D. Munzar, and A. M. Stoneham, Phys. Rev. B 69, 054511 (2004).
- ²⁰ A. E. Pantoja, D. M. Pooke, H. J. Trodahl, and J. C. Irwin, Phys. Rev. B **58**, 5219 (1998).
- ²¹R. Khasanov, A. Shengelaya, K. Conder, E. Morenzoni, I. M. Savić, J. Karpinski, and H. Keller, Phys. Rev. B 74, 064504 (2006).
- ²²H. J. Bornemann, D. E. Morris, and H. B. Liu, Physica C 182, 132 (1991).
- ²³G. M. Zhao, V. Kirtikar, and D. E. Morris, Phys. Rev. B 63, 220506(R) (2001).
- ²⁴T. Sasagawa, A. Lanzara, G.-H. Gweon, S. Zhou, J. Graf, Suryadijaya, and H. Takagi, Physica C **426-431**, 436 (2005).
- ²⁵H. J. Bornemann, D. E. Morris, H. B. Liu, and P. K. Narwankar, Physica C **191**, 211 (1992).
- ²⁶X. J. Chen, V. V. Struzhkin, Z. G. Wu, R. J. Hemley, H. K. Mao, and H. Q. Lin, Phys. Rev. B **75**, 134504 (2007).