PHYSICAL REVIEW B 82, 180512(R) (2010)



High-pressure synthesis of the indirectly electron-doped iron pnictide superconductor $Sr_{1-x}La_xFe_2As_2$ with maximum $T_c=22~$ K

Yoshinori Muraba, ¹ Satoru Matsuishi, ¹ Sung-Wng Kim, ² Toshiyuki Atou, ¹ Osamu Fukunaga, ² and Hideo Hosono ^{1,2,*}

¹ Materials and Structures Laboratory, Tokyo Institute of Technology, 4259 Nagatsuta, Midori, Yokohama 226-8503, Japan

² Frontier Research Center, Tokyo Institute of Technology, 4259 Nagatsuta, Midori, Yokohama 226-8503, Japan

(Received 4 May 2010; revised manuscript received 27 August 2010; published 17 November 2010)

Compounds of $Sr_{1-x}La_xFe_2As_2$ were synthesized by solid-state reaction at 1273 K, under pressures of 2 or 3 GPa. The $Sr_{1-x}La_xFe_2As_2$ phase was dominant up to x=0.5 and superconductivity was observed at $x \ge 0.2$. A maximum critical temperature T_c of 22 K and a maximum shielding volume fraction of ~70% were obtained at x=0.4. This is the first experimental demonstration of electron-doped superconductivity in 122-type iron pnictides, where electrons were doped through aliovalent substitution at the site of the alkaline-earth metal. The optimal T_c was slightly higher than that for the Co-doped (directly electron-doped) samples (19 K) and much lower than that for the hole-doped case (37 K). The bulk superconductivity range was narrower than that for the Co-substituted case and both ranges were much narrower than that for the hole-doped case. These observations revealed that differences in the electron-doping mode (direct or indirect) did not have a prominent effect on the optimal T_c or superconductivity range, compared with differences in carrier polarity.

DOI: 10.1103/PhysRevB.82.180512

PACS number(s): 74.70.Xa, 74.62.Bf, 74.25.F-

The discovery of LaFeAsO_{1-x} F_x with a critical temperature (T_c) of 26 K in early 2008 triggered the extensive exploration for new iron-based superconductors. Various compounds with an iron square lattice have since been reported and the T_c was pushed up to 56 K. This exceeded the maximum T_c of other materials with the exception of that for high- T_c cuprates.²⁻⁴ Such superconductors have a twodimensional cylindrical Fermi surface with two hole pockets at the Γ point and two electron pockets at the M point. The origin of the superconductivity is usually discussed in terms of nesting on the Fermi surface.^{5,6} Angle-resolved photoemission spectroscopy experiments⁷ and theoretical calculations^{5,6} suggest that superconductivity is expected to occur by electron or hole doping, providing the hard-band model is valid. The experimental examination of the both electron-doped and hole-doped case of superconductivity is still far from complete. For example, in the 1111 phase it is established that electron doping effectively works to induce superconductivity as exemplified by $LnFeAsO_{1-x}F_x$ (where Ln: lanthanide). However, the hole-doping effect still remains unclear.8 Information on both electron-doped and hole-doped cases in superconductivity is important for examining the modification of the Fermi surface by carrier doping.

The 122 phase⁹ is more favorable than the 1111 phase for examining this issue because the phase forming reaction temperature is sufficiently low that subreactions within container vessels and/or selective vaporization of a component may be suppressed. Direct electron doping to FeAs-layers by substituting the iron site with appropriate transition metals (e.g., Co²⁺ and Ni²⁺) (Refs. 10 and 11) can be achieved and superconductivity was induced by the doping mode. However, to date there are no reports of superconductivity of 122 system induced by indirect electron doping via aliovalent cation substitution to alkaline-earth metal site. While Wu et al.12 examined the La3+ substitution at the alkaline-earth metal site, solid-state reactions of the ingredient mixture for Ba_{1-x}La_xFe₂As₂ did not yield the La-substituted 122 phase by using the conventional glass-tube technique under ambient pressure. Leithe-Jasper et al. 13 reported densityfunctional theory calculations showing a distinct difference in-phase instability and effect of magnetic moment, originating from the electronic structure between Co-doped and Ladoped SrFe₂As₂. The La-substituted phase was electronically unstable but no phase instability was induced by Co doping. This prediction is consistent with the formation of $Sr_{1-x}La_xFe_2As_2$ phases having not been reported to date.

The ionic radius of La³⁺ (116 pm) is smaller than that of Sr^{2+} (126 pm) and so it is likely that a high-pressure synthesis would be effective for obtaining La-substituted $SrFe_2As_2$. In the current study, we report that $Sr_{1-x}La_xFe_2As_2$ ($x \le 0.6$) compounds with negative Seebeck coefficients can be synthesized at high pressure (2–3 GPa) and superconductivity with a maximum T_c of 22 K was observed in the range of $0.2 \le x \le 0.5$.

Samples of nominal chemical composition Sr_{1-x}La_xFe₂As₂ were synthesized by solid-state reaction under high pressure. The starting materials were SrAs, LaAs, and Fe₂As, each of them having been prepared from their respective metals. The mixture of these materials was sealed in a metal Ta tube with a Ta cap, by spot welding in a glove box filled with a dry Ar gas. Tubes were then heated at 1273 K for 2 h under pressures of 2–4 GPa. A belt-type anvil cell was employed for the high-pressure synthesis. Crystalline phases in the resulting samples were identified by powder x-ray diffraction (XRD) using a Bruker diffractometer model D8 ADVANCE (Cu rotating anode). The Rietveld analysis of XRD patterns was performed using RIETAN-FP code. 14 The four-probe resistivity and magnetic susceptibility were measured in the temperature range of 300-2 K, using a PPMS (Physical Property Measurement System, Quantum Design) with a vibrating-sample magnetometer attachment. Thermopower was measured at 300 K to check the polarity of the major carrier in the resulting samples.

Because solubility of La in AeFe₂As₂ (Ae=Sr,Ba) is small, solid-state reactions of the ingredient mixture for $Ae_{1-x}La_x$ Fe₂As₂ did not yield the La-substituted 122 phase by using the conventional glass-tube technique under ambient pressure. LaAs and SrFe₂As₂ phases were identified by

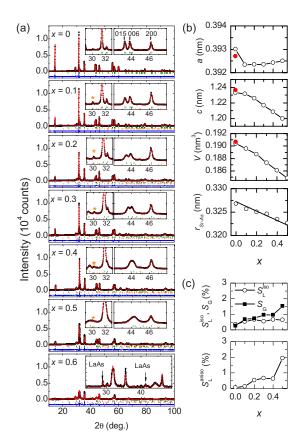


FIG. 1. (Color online) X-ray diffraction patterns and crystallographic data. (a) Powder XRD patterns of $Sr_{1-x}La_xFe_2As_2$ with x=0.0-0.6. A small amount of LaFeAsO phase denoted by asterisk (volume fraction $\sim 3\%$) almost independent of x, was seen for x=0.1-0.5 samples. The remarkable line broadening of 00l reflections was observed upon La substitution, indicating the presence of anisotropic strain along the c axis. (b) Lattice constants a and c, unit-cell volume V, and Sr-As distance r_{Sr-As} as a function of x: dare indicated by the filled circles. (c) Crystallographic strains S_G , S_L^{iso} , and S_L^{aniso} as a function of x: S_G and S_L^{siso} are isotropic strains calculated from Gaussian and Lorentzian-type components of pseudo-Voigt function describing line broadening for Rietveld analysis. S_L^{aniso} is the anisotropic strain along the [001] direction, also calculated from a Lorentzian-type component.

XRD, indicating segregation of La from the 122 phase. When a pressure of 2 GPa was applied during the reaction, the formation of LaAs was suppressed and an XRD peak shift was observed. When the applied pressure was increased to 3 GPa, the volume fraction of impurities increased. The target became a minority phase at 4 GPa, and so reaction conditions were fixed at 2 GPa and 1273 K for 2 h.

Figure 1(a) shows observed and calculated XRD patterns of samples synthesized at 2 GPa. Segregation of La was suppressed up to x=0.5 but the LaAs phase was apparent for x=0.6. A trace amount of LaFeAsO (<3 vol %) was identified in all samples. The same amount of LaFeAsO phase was observed as for samples prepared by the glass-tube technique thus the formation of LaFeAsO was attributed to oxygen contamination in ingredients during preparation and synthesis. Figure 1(b) shows the variation in lattice constants (a and

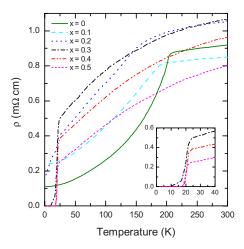


FIG. 2. (Color online) dc resistivity of $Sr_{1-x}La_xFe_2As_2$ as a function of temperature. The inset is the enlarged region around T_c .

b), unit-cell volume V, and the separation between Sr and As ions r_{Sr-As} within the $Sr_{1-x}La_xFe_2As_2$ phase as functions of nominal x. The unit-cell volume continuously decreased from x=0 to 0.5, indicating that the solubility limit of La in $SrFe_2As_2$ was x < 0.6. Providing there was separation between Sr and As ions in the 122 phase, r_{Sr-As} , decreased with the mean size of the cations at the Sr site and the value of r_{Sr-As} should decrease linearly with x. The difference in r_{Sr-As} between x=0 and x=1 (extrapolated) should be the difference in the ionic radius between Sr^{2+} and La^{3+} (~10 pm). The observed trend and value (\sim 10 pm) in Fig. 1(b) agrees with the above assumption, confirming that La³⁺ substitutes the site of Sr^{2+} . With an increase in nominal x, the broadening of diffraction peaks increased, especially for the 00l diffractions. The length of the c axis was more sensitive to xvariations than that of the a axis [see Fig. 1(b)]. A possible reason for the broadening was the modulation of lattice spacing due to inhomogeneity in local La concentration. The modulation of lattice spacing yielded the crystallographic strain. The degree of modulation as crystallographic strain was calculated from the line broadening obtained by the Rietveld analysis of the XRD patterns. Figure 1(c) shows isotropic and anisotropic crystallographic strains as a function of x. S_G and S_L^{iso} are the isotropic strains calculated from Gaussian and Lorentzian-type components of pseudo-Voigt functions, which describe line broadening for the Rietveld analysis. $S_{\rm L}^{aniso}$ is the anisotropic strain along the [001] axis, calculated from the Lorentzian component. 15 In this analysis, the value of parameter corresponding to crystallite size was fixed to that optimized for the sample of x=0. Both isotropic and anisotropic lattice strains start to increase from x=0.1and then increase markedly when x > 0.4. Thus it is presumed that the large strain induced in the crystallites by the La-substitution restricted the solubility limit of La³⁺ at the Sr²⁺ site.

Figure 2 shows the temperature dependence of resistivity for samples with different x. An anomaly arising from the crystallographic transition 16 accompanied the magnetic transition at around 200 K and was suppressed with x. A sharp resistivity drop instead appeared from x=0.2 and zero resistivity was observed at x \ge 0.3.

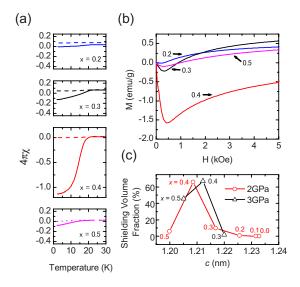


FIG. 3. (Color online) Magnetic properties of $Sr_{1-x}La_xFe_2As_2$. (a) Temperature dependence of magnetic susceptibility χ of the samples subjected to zero-field cooling (bold trace) or field cooling (dotted trace). (b) Magnetic moment vs magnetic field at 2 K. (c) Shielding volume fraction evaluated from dM/dH vs lattice constants c of the samples prepared under pressures of 2 and 3 GPa. The shielding volume fraction trend was almost the same for both series of samples.

Figure 3 shows the temperature dependence of magnetic susceptibility χ for each sample. Pellets were obtained by sintering-synthesized powders and were used for the measurements. Samples all underwent demagnetization at 300 K before magnetic measurements were carried out. A distinct negative shift of χ was observed for x=0.2-0.5 in the temperature range below \sim 22 K. The top of Fig. 3(b) shows *M-H* curves for the samples. The shielding volume fraction calculated from the slope of the M-H curve was $\sim 10\%$ at x=0.3 and reached a maximum of $\sim 70\%$ at x=0.4. It then dropped to \sim 5% at x=0.5. The precipitation of impurity phases was apparent from XRD patterns for the samples prepared at 3 GPa, indicating that the La concentration in the Sr-122 crystals was smaller than the nominal x. Since the c-axis length for samples prepared at 2 GPa was almost proportional to the nominal x, it could be used as a measure for the content of La substituting at the Sr sites. Figure 3(c) shows the shielding volume fraction evaluated from dM/dHvs lattice constant c for samples prepared at 2 and 3 GPa. The samples appeared to also display a maximum volume fraction of superconducting phase at the same c-axis length.

Figure 4 shows the Seebeck coefficients of $Sr_{1-x}M_xFe_2As_2$ (M=K and La) at 300 K. It was evident that the sign of Seebeck coefficients for the La-doped samples was negative and opposite to that observed for the K-doped samples.¹⁷ This result substantiated that electrons were effectively doped to the FeAs layer by the La substitution as intended and that superconductivity was induced by electron doping.

Analysis of XRD patterns indicated that La-substitution induced anisotropic strain along the *c* direction. Saha *et al.* ¹⁸ reported that the superconductivity of nondoped SrFe₂As₂ was promoted by internal strain and caused by stress during

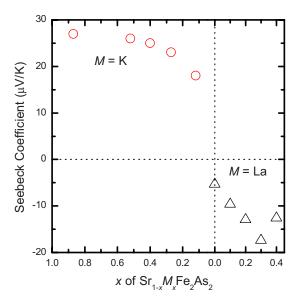


FIG. 4. (Color online) Seebeck coefficients at 300 K in aliovalent cation-substituted SrFe₂As₂. Data on the K-substituted samples were taken from Ref. 16.

crystal growth which disappeared during thermal annealing. In the current study, we examined the annealing at 573 K of $Sr_{0.6}La_{0.4}Fe_2As_2$ prepared at 2 GPa but found no significant changes in superconducting properties (T_c and shielding volume) or XRD profile. This indicated that the superconductivity of our samples was not due to mechanical stress like that reported by Saha *et al.*

The reason for T_c on the ρ -T profile not changing with La concentration was then investigated. The profile analysis of XRD patterns revealed that there is no indication of phase separation but distinct persistent crystalline strain are present in the (Sr, La)Fe₂As₂. This strain is attributed to inhomogeneous replacement of Sr ions with La ions in the SrFe₂As₂ phase. In each sample with the observable T_c , there was a portion in which the La concentration was the same as x =0.4 with a maximum $T_c \sim 22$ K and these portions are percolated throughout the bulk samples. Since the onset T_c is controlled by the highest T_c region (La_{0.4}Sr_{0.6}Fe₂As₂), we may understand the obtained results that the shielding volume fraction changed with x but T_c remained almost unchanged. Here, note that substitution between Sr and La is easy in cuprates even in conventional solid-state reaction processes but impossible in Sr-122 iron arsenide. This makes a sharp contrast between these two representative superconductive systems

Figure 5 summarizes the electronic phase diagram for directly/indirectly, hole/electron-doped $SrFe_2As_2$. Onset T_c and anomaly temperature in ρ -T curves indicating structural/magnetic transition are plotted in this figure. Here, doped carrier numbers per Fe are plotted in place of compositions in the figure. It is of interest to compare the present results (indirect electron doping) with those for $SrFe_{2-y}Co_yAs_2$ (direct doping). Superconductivity in the latter system occurred at $0.2 \le y \le 0.5$. T_c had a maximum of 19 K at $y \sim 0.2$, at which point antiferromagnetic ordering decreased with x and disappeared at y=0.5. In the present system, bulk superconductivity (shielding volume $> \sim 10\%$) was observed

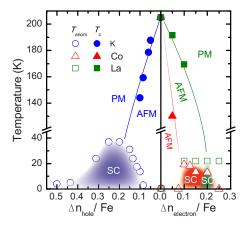


FIG. 5. (Color online) Electronic phase diagrams for hole/electron-doped SrFe₂As₂. This diagram was drawn based on the data reported in Refs. 12 and 16 and present work. $\Delta n_{\rm electron}/{\rm Fe}$ is the injected number of electrons per Fe site and $\Delta n_{\rm hole}/{\rm Fe}$ is the number of holes. Observed data (\square) on T_c in Sr_{1-x}La_xFe₂As₂ is almost constant ~22 K for $\Delta n_{\rm electron}/{\rm Fe}$ =0.1–0.25. Thus, we attribute intrinsic superconducting region to a narrow range around 0.2 on the basis of the presence of persistent crystallographic strain, shielding volume fractions, and onset T_c -x relation.

at x=0.3 and 0.4 for samples prepared at 2 GPa and x=0.4 and 0.5 for those prepared at 3 GPa. The number of electrons injected into the FeAs layer was x/2=y per Fe site. Superconductivity was induced for the La-substituted system if electron numbers of 0.15–0.25 per Fe for samples prepared at 2 GPa and 0.2–0.25 per Fe site for those prepared at 3 GPa, were doped to the parent phase. However, the T_c remains almost constant (22 K) irrespective of x and the shielding volume fraction is very small except at x=0.4. Thus, it is considered that superconducting range in this system is restricted to x=0.4 ($\Delta n_{\rm electron}/{\rm Fe}$ =0.2). This range is

narrower than that for the Co-substituted system and both ranges are much narrower than that for the hole-doped case. In this case the range continues to the end member KFe₂As₂. The maximum T_c for the indirect (La)-doped system was slightly higher than that (19 K) for the direct (Co)-doped system¹³ but was lower than that (37 K) for the optimal hole-doped case.¹⁷

Recently, asymmetry in the electronic phase diagram upon electron and hole doping into Fe pnictide superconductors, was theoretically studied by Ikeda *et al.*¹⁹ They constructed a five-band model using a combination of *ab initio* band calculations and fluctuation exchange approximations and examined the doping dependence of superconductivity. They found that superconductivity was stable over a wider range for hole doping than for electron doping. Electron doping filled the hole pockets, which in turn reduced the nesting between the hole pockets and electron pockets. The disappearance of the electron pockets did not occur by hole doping. The present experimental results on the carrier doped 122 system are consistent with this theoretical prediction.

In summary, indirectly electron-doped $Sr_{1-x}La_xFe_2As_2$ was synthesized by solid-state reaction under pressures of 2–3 GPa. The optimal T_c was slightly higher than that for the indirectly electron (Co)-doped case but much lower than that for the hole-doped case. No significant difference in the superconductivity range was observed between La and Co substitution. Both ranges were much narrower than that for the hole-doped case. It was concluded that the difference in electron-doping mode, either direct or indirect, was much smaller than that of polarity of the doped carrier.

This study was supported by the Funding Program for World-Leading Innovative R&D on Science and Technology, JSPS, Japan. The authors thank T. Kamiya and H. Ikeda for their discussion.

^{*}Corresponding author. FAX +81-45-924-5339; hosono@msl.titech.ac.jp

¹Y. Kamihara, T. Watanabe, M. Hirano, and H. Hosono, J. Am. Chem. Soc. **130**, 3296 (2008).

²H. H. Wen, Adv. Mater. **20**, 3764 (2008).

³K. Ishida, Y. Nakai, and H. Hosono, J. Phys. Soc. Jpn. **78**, 062001 (2009).

⁴I. I. Mazin, Nature (London) **464**, 183 (2010).

⁵I. I. Mazin, D. J. Singh, M. D. Johannes, and M. H. Du, Phys. Rev. Lett. **101**, 057003 (2008).

⁶K. Kuroki, S. Onari, R. Arita, H. Usui, Y. Tanaka, H. Kontani, and H. Aoki, Phys. Rev. Lett. **101**, 087004 (2008).

⁷H. Ding *et al.*, EPL **83**, 47001 (2008); K. Nakayama *et al.*, *ibid.* **85**, 67002 (2009).

⁸ H. H. Wen, G. Mu, L. Fang, H. Yang, and X. Zhu, EPL **82**, 17009 (2008); G. Wu, H. Chen, Y. L. Xie, Y. J. Yan, T. Wu, R. H. Liu, X. F. Wang, D. F. Fang, J. J. Ying, and X. H. Chen, Phys. Rev. B **78**, 092503 (2008).

⁹M. Rotter, M. Tegel, and D. Johrendt, Phys. Rev. Lett. **101**, 107006 (2008).

¹⁰ A. S. Sefat, R. Jin, M. A. McGuire, B. C. Sales, D. J. Singh, and

D. Mandrus, Phys. Rev. Lett. 101, 117004 (2008).

¹¹S. L. Bud'ko, N. Ni, and P. C. Canfield, Phys. Rev. B 79, 220516(R) (2009).

¹²G. Wu, R. H. Liu, H. Chen, Y. J. Yan, T. Wu, Y. L. Xie, J. J. Ying, X. F. Wang, D. F. Fang, and X. H. Chen, EPL **84**, 27010 (2008).

¹³ A. Leithe-Jasper, W. Schnelle, C. Geibel, and H. Rosner, Phys. Rev. Lett. **101**, 207004 (2008).

¹⁴F. Izumi and K. Momma, Solid State Phenom. **130**, 15 (2007).

¹⁵A. C. Larson and R. B. Von Dreele, Los Alamos National Laboratory Report No. LAUR 86-748, 2000 (unpublished).

¹⁶D. Kasinathan, A. Ormeci, K. Koch, U. Burkhardt, W. Schnelle, A. Leithe-Jasper, and H. Rosner, New J. Phys. 11, 025023 (2009).

¹⁷B. Lv, M. Gooch, B. Lorenz, F. Chen, A. M. Guloy, and C. W. Chu, New J. Phys. **11**, 025013 (2009).

¹⁸S. R. Saha, N. P. Butch, K. Kirshenbaum, J. Paglione, and P. Y. Zavalij, Phys. Rev. Lett. **103**, 037005 (2009).

¹⁹H. Ikeda, R. Arita, and J. Kuneš, Phys. Rev. B **81**, 054502 (2010).