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①

$$\frac{dP}{dT} = \frac{g}{K}$$

$$g \equiv -\frac{GM}{r^2}$$

$$dP = \frac{g}{K} dT$$

$$P = \frac{R_g}{\mu} \rho T \leftarrow \text{ideal gas Eos}$$

$$\frac{\mu P}{R_g T} = \rho$$

μ = mean molecular weight.
can calc. this fr. the ~~metals~~ metallicity if we assume all elements are completely ionized w/in the atmos, which I don't believe is entirely true.

$$P = \frac{k}{\mu m_u} \rho T \rightarrow \rho = \frac{\bar{\mu} P}{k_B T}$$

\nwarrow boltzman const.
 \uparrow
 avg. mass, i.e. mean molecular weight, $\bar{\mu}$

$$\frac{1}{\bar{\mu}} = 2X + \frac{3}{4}Y + \frac{1}{2}Z$$

$$X = 0.71 \quad Y = 0.272930 \quad Z = 0.01707$$

$$\bar{\mu} = 0.613308$$