

QCM-D data analysis GUI mini guide.

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Running QCMDanalysisGUI.m will launch the program for analyzing data from the QSense QCM-D instrument. This code uses the power-law analysis developed by Shull et al., which is a more realistic description of materials than the commonly used Voigt model.[6] The detailed approach is outlined by DeNolf et al. and Martin et al.[1, 2] This file provides a guide on how to use the analysis, import data, and export results. Publication quality plots can be generated directly from this GUI, which also provides checks and balances for accurate viscoelastic analysis of polymer films whose properties can span from glasses to soft gels in air or in liquids and the way to viscous polymer solutions.

1 License Agreement

This code is protected by the open source GNU General Public License v3.0, which lets the user do almost anything with the project, except to distribute closed source versions. Contact the author for closed source versions.

2 Citation

This code is identified by the DOI: . If this code is used for publications, then it needs to be cited appropriately. Additionally, the analysis developed in refs [1, 2, 6] should be cited.

3 Example files

Three example data files are provided.

- 1) The QCM-D response from pure water (Bulk limit)
- 2) A hydrogel equilibrating and then swelling in water (Thin film in water)
- 3) A PMMA film in air (Thin film in air)

4 Buttons

Load QCM-D Data: This button opens a file selection window for you to navigate to your data. Your data should be in the format shown in the example files, where the appropriate referencing for the shifts is already done manually using the absolute frequencies. The program assumes you have referenced your shifts with respect to the bare crystal frequencies in air if your film is in air, or to the bare crystal frequencies in water if your film is in water. For a bulk measurement of a viscous liquid, the reference is the bare crystal in air. The code will read only excel files with the extension .xlsx. If you have a .xls file, simply changing the extension to .xlsx should make things work. Note the dissipation factor is used in the data, but a conversion is made to represent this as $\Delta\Gamma$ using Eq. 1.

$$\Delta\Gamma_n = \frac{\Delta D_n f_n}{2} \quad (1)$$

Bulk Calculation: If you are measuring a bulk viscous liquid, then load your data and hit this button. A pop up figure will appear with your solutions. Here you can use the 'bare crystal in water' file as an example to see the properties of water at different harmonics. The bulkcalc_Callback function near the end of the main code is called, and you can make changes to the plots here according to your needs.

Plot QCM Data: This file will plot the solutions for your film in water or your film in air. Make sure the correct radio button is selected for your situation. Note dp is written in mg/m^2 which is approximately

the film thickness in nanometers is a density of $1g/cm^3$ is assumed. Likewise the $\rho|G_3^*|$ is the complex shear modulus at 15 MHz. This quantity is $\approx 10^5 Pa - g/cm^3$ for water, and $\approx 3 * 10^9$ for polymer glasses. The phase angle ϕ is 90 for a Newtonian liquid like water, and $\approx 1^\circ$ for an elastic solid like a polymer glass.

Viscoelastic plots: This button will show you the viscoelastic plots outside the main GUI. The `export_callback` near the end of the main code is used to make these plots. You can change the way things are plotted here to customize for your publication needs. Only results for the 3:5,5 calculation are shown.

Thickness plots: This button will show you the d/λ_n and also the predicted shifts from your solutions. In this analysis, we exclusively only use the 3:5,5 calculation, therefore we have $\Delta\Gamma_3$ as a free parameter for a consistency check. If the code solved correctly, then $\Delta f_3, \Delta f_5$ and $\Delta\Gamma_5$ (3:5,5) should all have the experimental shifts and predicted shifts match exactly. The difference in the experimental and predicted value of $\Delta\Gamma_3$ should be as small as possible. In most real cases a small discrepancy in the predicted and experimental value is acceptable. Here, the `thicknessplots_Callback` near the end of the main code is called to make the plots. You can customize the output according to your needs.

Refresh: Relaunches the GUI and clears the memory. Useful if an error is encountered, or if you just want a clean new window.

Range to solve: By default we solve every point. However, if the data set is large then it is desirable to solve only a portion of the data with larger intervals. For example, inputting 100:5:500 into the edit text boxes will solve every fifth point between the 100th and 500th data points.

Save plots: If this button is checked, then publication quality vector figures are exported into the working directory when 'Viscoelastic Plots' and 'Thickness Plots' are clicked.

Export Data: This button outputs a tab-delimited text file into the working directory with all the relevant values. This is useful if you want to do your own plotting outside MATLAB.

Handles: This allows you access into MATLAB's handle structure where all the data is stored internally. After you have solved for solutions, click this button and MATLAB will open the workspace in debugging mode. Here, there will be a new variable called "Handles" in the workspace. If you double click this, then the internal data structure is revealed. For example, the variable `phiout{1}` stores all the phase angles from the 3:5:3 calculation, and `phiout{2}` stores all ϕ from the 3:5,5 calculation. Likewise, the variable `d5out` stores d/λ_5 . To leave the handle structure, click "Quit Debugging" in the main MATLAB window. More information on d/λ_n can be found in Anal. Chem. 2018, 90, 4079–4088.[6] Here, d is the film thickness and λ_n is the wavelength of the shear wave at the n^{th} harmonic.

$$\lambda_n = \frac{1}{f_n} \left[\frac{|G_n^*|}{\rho} \right]^{1/2} \frac{1}{\cos(\phi)} \quad (2)$$

Shift Simulator: See the section on shift simulator.

5 General notes

- 1) The code is pretty good at finding missing values in the imported data. However, there is a chance data import may fail if there are too many missing data points, or if the data columns are of varying lengths. You may have to manually clean the data before importing in an error is encountered.
- 2) The GUI has zoom, grab and data cursor functionality enabled. You can select the data cursor, and then click on individual points for the exact values.
- 3) Often, if the shifts do not solve (maybe because your film is too thin), then the calculation will return non-sense for modulus and ϕ . For example, all the ϕ will be 90, which is the upper bound on the numerical solver. However, even if the code does not solve because your film is too thin, the $d\rho$ value is reliable.

- 4) See Figure 6a in Anal. Chem. 2018, 90, 4079–4088 to gain intuition on how ϕ and $\rho|G_n^*|$ are correlated. See Figure 8a to understand the approximate thickness ranges needed for accurate viscoelastic analysis. In general, the stiffer the material, the thicker it has to be for being in the optimal viscoelastic regime. Polymer glasses such as polystyrene need $\approx 5\mu m$ films, while polymer gels/complexes need $\approx 1\mu m$ films. The maximum $\Delta\Gamma_n$ that is trustworthy is $\approx 10,000$ Hz.
- 5) The code is written in MATLAB 2018a. Version specific errors may occur, especially if you are on a much older version. Generally, 2015a and up should work fine.
- 6) A good crystal with a reliable response should have absolute $\Gamma < 100$ Hz for the measured harmonics.
- 7) We plot $d\rho$ in mg/m^2 , which is the areal mass of the film. Here, if you assume your film is $\approx 1g/cm^3$ then $1000\ mg/m^2$ is $\approx 1\mu m$.
- 8) Bug reporting should be done under issues on the main GitHub page under Issues.
- 9) To understand the kinds of experiments which can be performed using the QCM, the reader is referred to several publications.[3, 7, 5, 4, 8]
- 10) Films should be as smooth and homogeneous as possible.
- 11) For high temperature experiments, the bare crystal frequencies at all appropriate temperatures needs to be used as the reference.

6 Shift simulator

One advantage of using this QCM data analysis program is that you can simulate the response from your material before doing any experiments at all. A good QCM experiment requires the film deposited on the crystal to be in the right thickness range. If one has a general idea of what mechanical properties to expect from the film, then these numbers can be inputted into the Shift Simulator, which will provide you with the approximate Δf_n and $\Delta\Gamma_n$, as well as the parameter d/λ_n and the shear wave decay length, $\rho\delta_n$. This is exceptionally useful to minimize the number of trial and error experiments, and also builds intuition about experimental design. For example, the user can keep the material properties $\rho|G_n^*|$ and ϕ constant, and change the $d\rho$ value to see how the QCM response will vary. check Figure 6a in Anal. Chem. 2018, 90, 4079–4088 to see what approximate values to use for ϕ and $\rho|G_n^*|$.

References

- [1] Garret C. DeNolf, Lauren F. Sturdy, and Kenneth R. Shull. High-Frequency Rheological Characterization of Homogeneous Polymer Films with the Quartz Crystal Microbalance. *Langmuir*, 30(32):9731–9740, August 2014.
- [2] Elizabeth J. Martin, Mathew T. Mathew, and Kenneth R. Shull. Viscoelastic Properties of Electrochemically Deposited Protein/Metal Complexes. *Langmuir*, 31(13):4008–4017, April 2015.
- [3] Elizabeth J. Martin, Kazi Sadman, and Kenneth R. Shull. Anodic Electrodeposition of a Cationic Polyelectrolyte in the Presence of Multivalent Anions. *Langmuir*, 32(31):7747–7756, August 2016.
- [4] Kazi Sadman, Qifeng Wang, Shawn H. Chen, David E. Delgado, and Kenneth R. Shull. pH-Controlled Electrochemical Deposition of Polyelectrolyte Complex Films. *Langmuir*, 33(8):1834–1844, February 2017.

- [5] Kazi Sadman, Qifeng Wang, Yaoyao Chen, Bavand Keshavarz, Zhang Jiang, and Kenneth R. Shull. Influence of Hydrophobicity on Polyelectrolyte Complexation. *Macromolecules*, November 2017.
- [6] Kazi Sadman, Clinton G. Wiener, R. A. Weiss, Christopher C. White, Kenneth R. Shull, and Bryan D. Vogt. Quantitative Rheometry of Thin Soft Materials Using the Quartz Crystal Microbalance with Dissipation. *Anal. Chem.*, 90(6):4079–4088, March 2018.
- [7] Lauren F. Sturdy, Alexander Yee, Francesca Casadio, and Kenneth R. Shull. Quantitative characterization of alkyd cure kinetics with the quartz crystal microbalance. *Polymer*, 103:387–396, October 2016.
- [8] C. Joshua Yeh, Michael Hu, and Kenneth R. Shull. Oxygen Inhibition of Radical Polymerizations Investigated with the Rheometric Quartz Crystal Microbalance. *Macromolecules*, 51(15):5511–5518, August 2018.