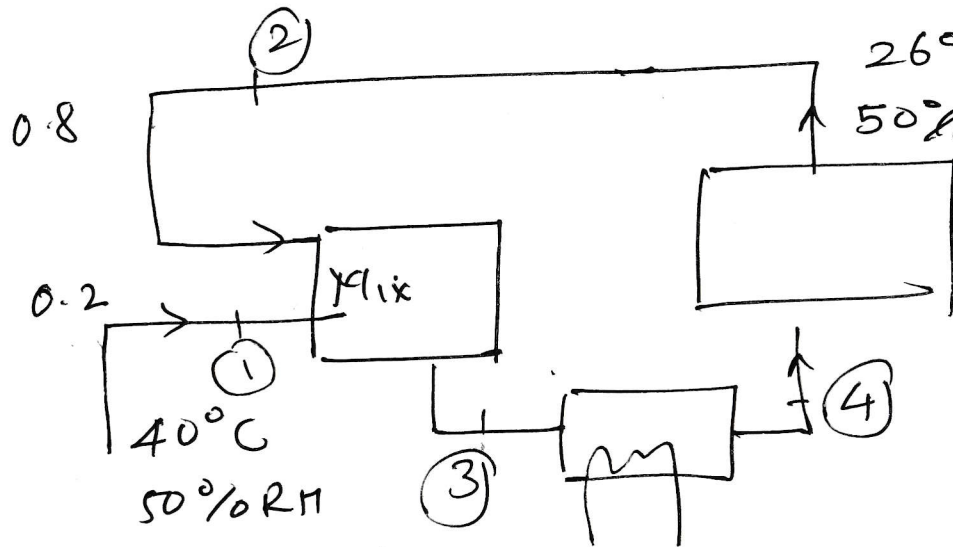


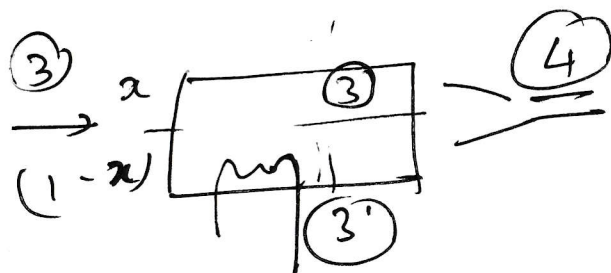
The room sensible and latent heat loads of an AC space is 25 kW and 5 kW respectively. The room condition is ~~26~~ 26°C and 50% RH. The outdoor condition is 40°C and 50% RH. The ventilation requirements is such that on mass flow rate basis 20% fresh air is introduced and 80% supply air is recirculated. The by-pass ratio of the cooling coil is 0.15. Determine the supply air flow rate and the coil temperature.



$$h_1 = 101 \text{ kJ/kg}$$

$$\omega_1 = 23.5 \text{ g/kg d.a.}$$

$$V_1 = 0.92 \text{ m}^3/\text{kg}$$



$$26^\circ\text{C} \quad \omega_2 = 10.8$$

$$50\% \text{ RH} \quad h_2 = 53.5$$

$$SH = 25 \text{ kW} \quad V_2 = 0.862$$

$$LH = 5 \text{ kW}$$

$$\omega_3 = (0.2)\omega_1 + (0.8)\omega_2$$

$$\omega_3 = (0.2)(23.5) + (0.8)(10.8)$$

$$= 13.34 \text{ g/kg d.a.}$$

$$h_3 = (0.2)(101) + (0.8)(53.5)$$

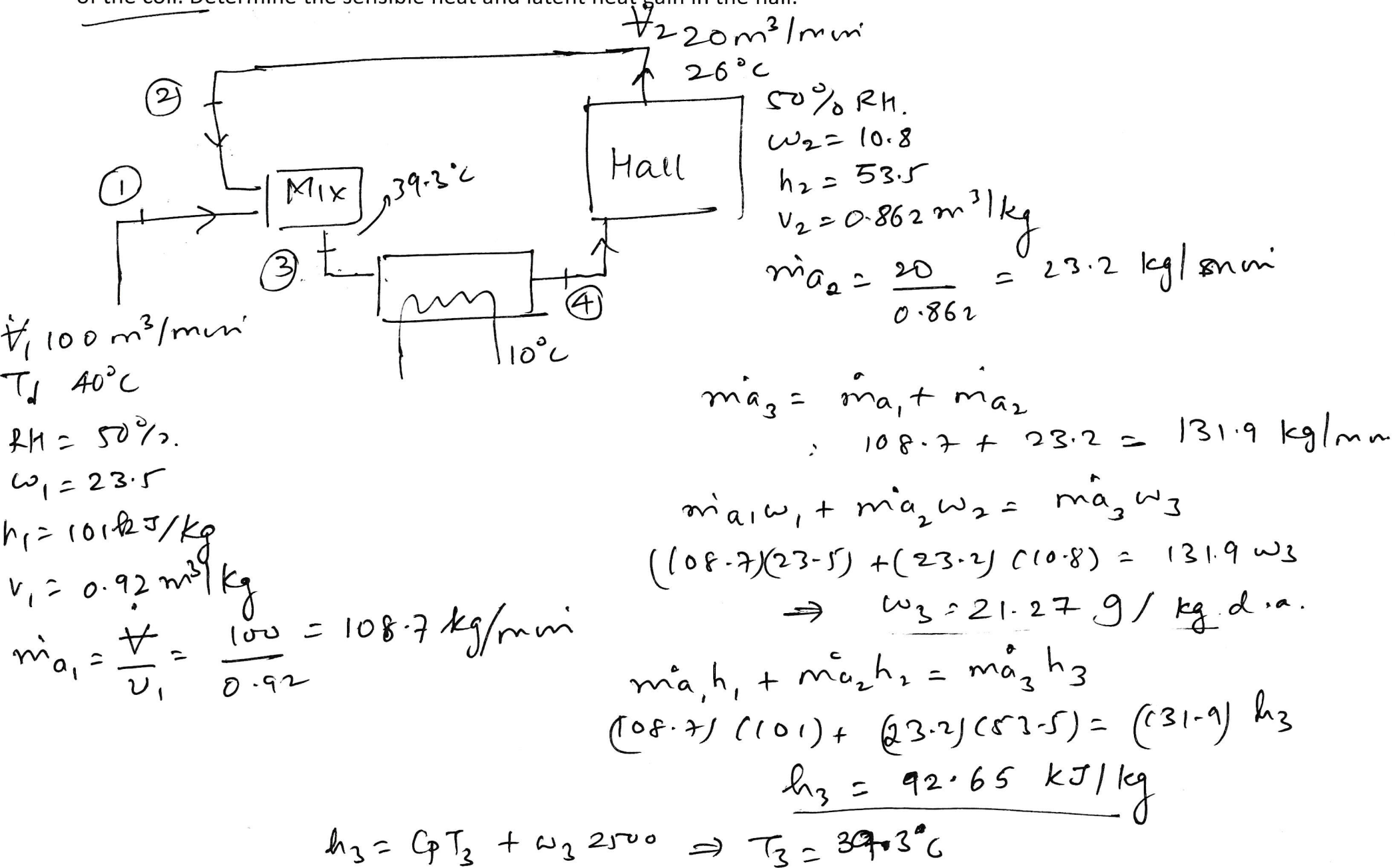
$$= 63 \text{ kJ/kg}$$

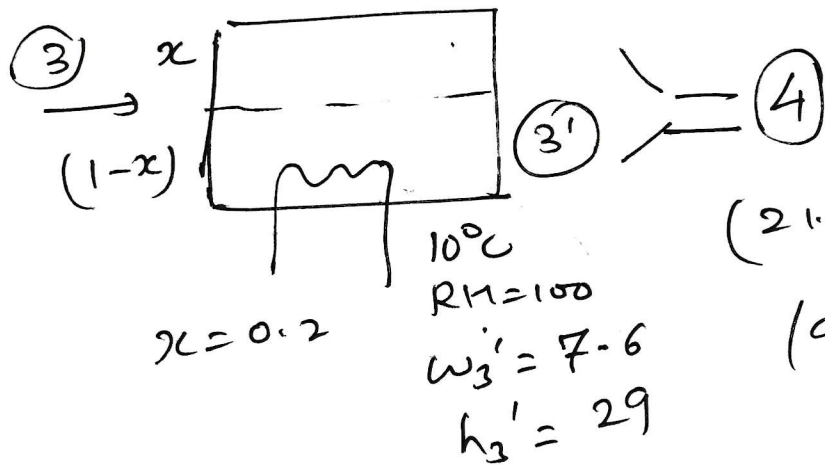
$$T_3 = \frac{63 + 2500\omega_3}{C_p} = 29.5^\circ\text{C}$$

$$x h_3 + (1-x) h_3' = h_4$$

$$\left. \begin{aligned} SH &= \dot{m}_a c_{ph} (T_2 - T_4) \\ LH &= \dot{m}_a h_{fg} (\omega_2 - \omega_4) \end{aligned} \right\} \frac{SH}{LH} = \frac{c_{ph} (T_2 - T_4)}{h_{fg} (\omega_2 - \omega_4)}$$

Air flow at the rate of $100 \text{ m}^3/\text{min}$ at 40°C and $50\% \text{ RH}$ is mixed with another stream flowing at the rate of $20 \text{ m}^3/\text{min}$ at 26°C and $50\% \text{ RH}$ that is recirculated from a hall. The mixture flows over a cooling coil whose surface temperature is 10°C and by-pass factor is 0.2 . Find the temperature and RH leaving the coil. Find the cooling capacity of the coil. Determine the sensible heat and latent heat gain in the hall.





$$\omega_3 x + (1-x) \omega_3' = \omega_4$$

$$h_3 x + (1-x) h_3' = h_4$$

$$(2.27)(\cancel{7.6})(0.2) + (1-0.2)(7.6) = \omega_4$$

$$= 10.33 \frac{\text{g}}{\text{kg da}}$$

$$(92.65)(0.2) + (1-0.2)(29) = h_4$$

$$= 41.73 \text{ kJ/kg}$$

$$h_4 = C_p T_4 + \omega_4 (2500)$$

$$T_4 = 15.9^\circ\text{C}$$

$$p_{v_4} = \frac{p \omega_4}{0.622 + \omega_4} = \frac{(101.325)(0.01033)}{0.622 + 0.01033} = 1.655$$

$$p_{\text{sat}} = 0.01818 \text{ bar} = 1.818 \text{ kPa}$$

$$\text{RH} = \frac{1.655}{1.818} = 91\%$$

$$\dot{m}_a h_3 - \dot{m}_a h_4 + \dot{Q} = 0$$

$$\dot{Q} = \dot{m}_a (h_4 - h_3)$$

$$= (131.9)(41.73 - 92.65)$$

$$= -6716.35 \text{ kJ/min}$$

$$= -111.9 \text{ kW}$$

$$\dot{Q}_H = \dot{m}_a C_{ph} (T_{d2} - T_{d4})$$

$$= (131.9)(1.022)(26 - 15.9) = 1361.5 \text{ kJ/min}$$

$$\dot{Q}_H = \dot{m}_a (\omega_2 - \omega_4) h_{fg}$$

$$= (131.9)(10.8 - 10.33)2500$$

$$= 155 \text{ kJ/min} = 2.6 \text{ kW}$$

$$\frac{LH}{SH} = \frac{h_{fg} (\omega_2 - \omega_4)}{C_{ph} (T_2 - T_4)} \Rightarrow \frac{\omega_2 - \omega_4}{T_2 - T_4} = \frac{C_{ph}}{h_{fg}} \frac{LH}{SH}$$

$$\text{slope} = \frac{\omega_2 - \omega_4}{T_2 - T_4} = \left(\frac{1.022}{2500} \cdot \frac{5}{25} \right) (1000) \downarrow$$

to convert
from g for
the chart

$$\theta = 4.67^\circ$$

Cool temp

From chart, $\boxed{T_3' = 14^\circ\text{C}}$ $h_3' = 39$ $\omega_3' = 10$

Subs in $x h_3 + (1-x) h_3' = h_4$

$$(0.15)(63) + (1-0.15)(39) = h_4$$

$$\Rightarrow h_4 = 42.6 \text{ kJ/kg}$$

$$(0.15)(13.34) + (1-0.15)(10) = \omega_4 \Rightarrow \omega_4 = 10.5g$$

$$\Rightarrow T_4 = 16^\circ\text{C} \quad RH = 90\%$$

$$SH = \dot{m}_a C_{ph} (T_2 - T_4)$$

$$25 = \dot{m}_a (1.022)(26 - 16)$$

$$\Rightarrow \boxed{\dot{m}_a = 2.45 \text{ kg/s}}$$