

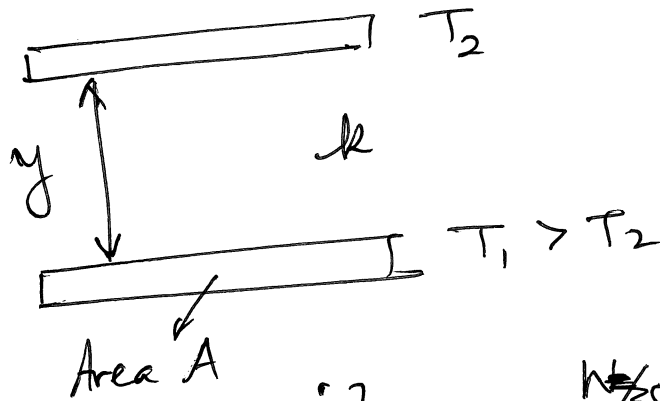
Transport.

- 1) Heat transport
- 2) Mass transport
- 3) Momentum transport.

} $\begin{cases} \text{Molecular transport} \\ \text{Convective transport.} \end{cases}$

①

Heat transport by conduction



$$[k] = \frac{[\dot{q}]}{[dT/dy]} = \frac{\text{W/m}^2}{\text{K/m}} = \frac{\text{W}}{\text{K m}} = \text{W K}^{-1} \text{m}^{-1}$$

$$\frac{\text{J}}{\text{s m}^2} = [\alpha] \frac{\text{J}}{\text{m}^3} \Rightarrow [\alpha] = \frac{\text{m}^2}{\text{s}}$$

$$\dot{Q} = \frac{k A (T_1 - T_2)}{y}$$

$$\dot{q} = \frac{\dot{Q}}{A} = -k \left(\frac{T_2 - T_1}{y} \right)$$

$$\dot{q} = -k \frac{dT}{dy} \leftarrow \text{Fourier's law of conduction}$$


$$\dot{q} = -k \frac{d}{dy} (\rho C_p T)$$


$$\text{Let } \alpha \triangleq \frac{k}{\rho C_p} \leftarrow \text{Thermal diffusivity.}$$

$$\dot{q} = -\alpha \frac{d}{dy} (\rho C_p T) \quad \begin{matrix} \text{Energy} \\ \text{Enthalpy per unit} \\ \text{volume} \end{matrix}$$

Rate of energy flow per unit area

Momentum transfer

 $u=0$
 Dynamic viscosity μ

 $u=V$
 Area A

$$[\tau] = [\mu] \left(\frac{du}{dy} \right)$$

$$\text{Pa} = [\mu] \frac{\text{m}}{\text{s}} \cdot \frac{1}{\text{m}}$$

$$[\mu] = \text{Pa} \cdot \text{s}$$

$$[\mu] = \mu_{\text{water}} \sim 10^{-3} \text{ Pa} \cdot \text{s}$$

$$[\nu] = \frac{[\mu]}{[\rho]} = \frac{\text{Pa} \cdot \text{s}}{\text{kg}/\text{m}^3}$$

$$[\nu] = \frac{\text{m}^2}{\text{s}}$$

$$F = \mu \frac{V}{y} A \Rightarrow \frac{F}{A} = \mu \frac{V}{y}$$

$$\Rightarrow \tau = \mu \frac{V}{y}$$

$$\Rightarrow \boxed{\tau = -\mu \frac{du}{dy}}$$

$$\tau = -\frac{\mu}{\rho} \frac{d(\rho u)}{dy} \downarrow$$

Momentum
per volume

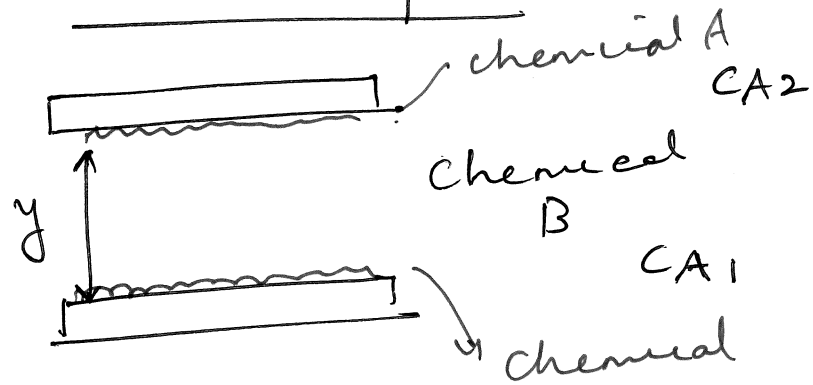
Define $\nu \triangleq \frac{\mu}{\rho}$ Kinematic
viscosity.

$$\boxed{\tau = -\nu \frac{d}{dy} (\rho u)}$$

$$= \frac{\text{N}}{\text{m}^2} \frac{\text{s}}{\text{kg}/\text{m}^3}$$

$$= \left(\frac{\text{kg}}{\text{s}^2} \frac{\text{m}}{\text{m}^2} \right) \frac{1}{\text{m}^2} \cdot \frac{\text{s}}{\frac{\text{kg}}{\text{m}^3}} = \frac{\text{m}^2}{\text{s}}$$

Mass transfer



CA1, CA2 = concentrations of chemical A

$$\dot{m}_A = D_{AB} \frac{CA_1 - CA_2}{y} A$$

Diffusion coefficient depends on both A and B

$$\frac{\dot{m}_A}{A} = D_{AB} \frac{CA_1 - CA_2}{y}$$

$j_A = \text{mass flux} \triangleq \frac{\dot{m}_A}{A}$

$$j_A = -D_{AB} \frac{dCA}{dy}$$

$$[j_A] = [D_{AB}] \frac{[dCA]}{[dy]}$$

$$\frac{\text{kg}}{\text{s m}^2} = [D_{AB}] \frac{\text{kg}}{\text{m}^3} \cdot \frac{1}{\text{m}}$$

$$[D_{AB}] = \frac{\text{m}^2}{\text{s}}$$

Heat transfer

$$\dot{q} = -\frac{k}{\rho c_p} \frac{d}{dy} (\rho c_p T)$$

$$= -\alpha \frac{d}{dy} (\rho c_p T)$$

Momentum transfer

$$\tau = -\mu \frac{du}{dy}$$

$$= -\frac{\mu}{\rho} \frac{d}{dy} (\rho u)$$

Mass transfer (4)

$$j_A = -D_{AB} \frac{dc_A}{dy}$$

Flux = - (transport quantity) (Gradient of a quantity)

↓
energy
momentum
mass

Transport quantity

\propto

$$m^2/s$$

ν

$$m^2/s$$

D_{AB}

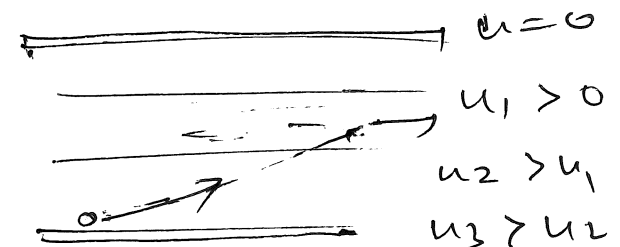
$$m^2/s$$

$$Pr \triangleq \frac{\nu}{\alpha}$$

Prandtl number

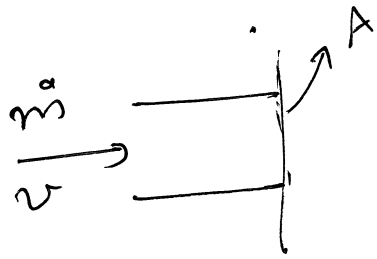
$$Sc \triangleq \frac{\nu}{D_{AB}}$$

Schmidt Number



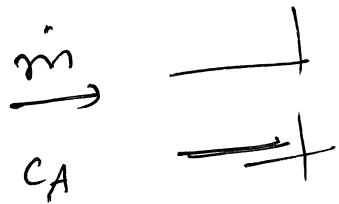
Convective transport

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$$\begin{aligned}\dot{E} &= \dot{m} h \\ &= \dot{m} C_p T \\ &= \rho A v C_p T\end{aligned}$$

$$\dot{q}_{conv} = \frac{\dot{E}}{A} = (\rho C_p T) v$$



$$\dot{m}_A = \rho A v C_A$$

$$\dot{j}_{conv} = \frac{\dot{m}_A}{A} = (C_A) v$$

$$\begin{aligned}\dot{M} &= \dot{m} v \\ &= (\rho A v) v\end{aligned}$$

$$\tau = \frac{\dot{M}}{A} = \rho v^2$$