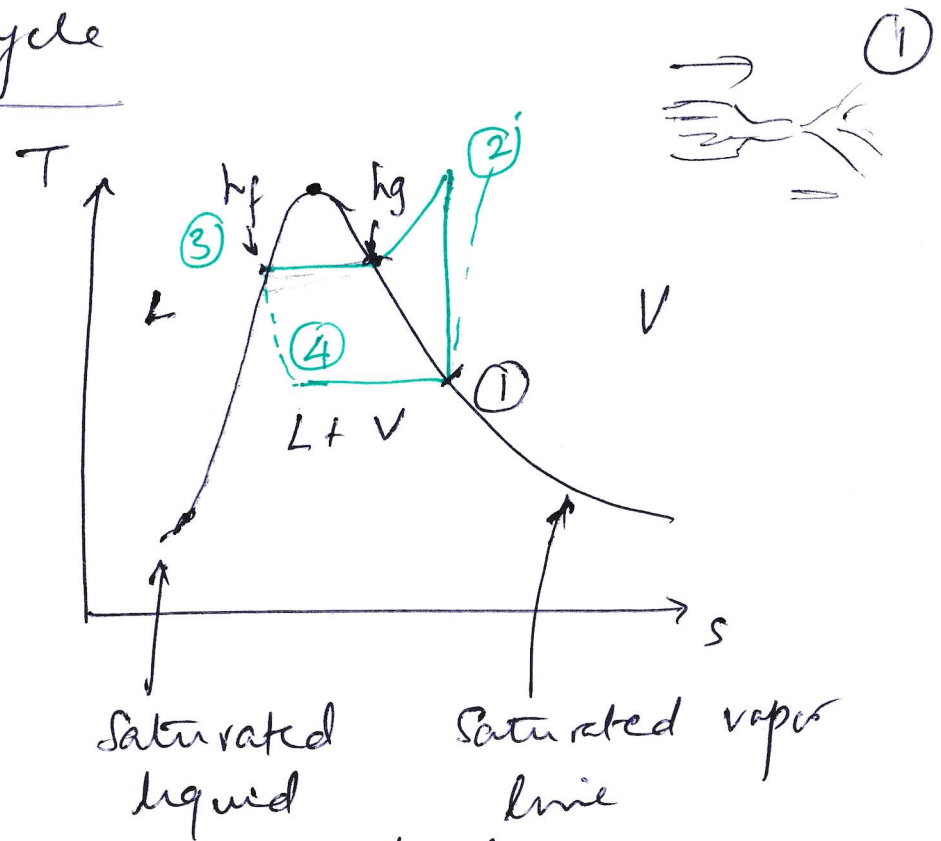
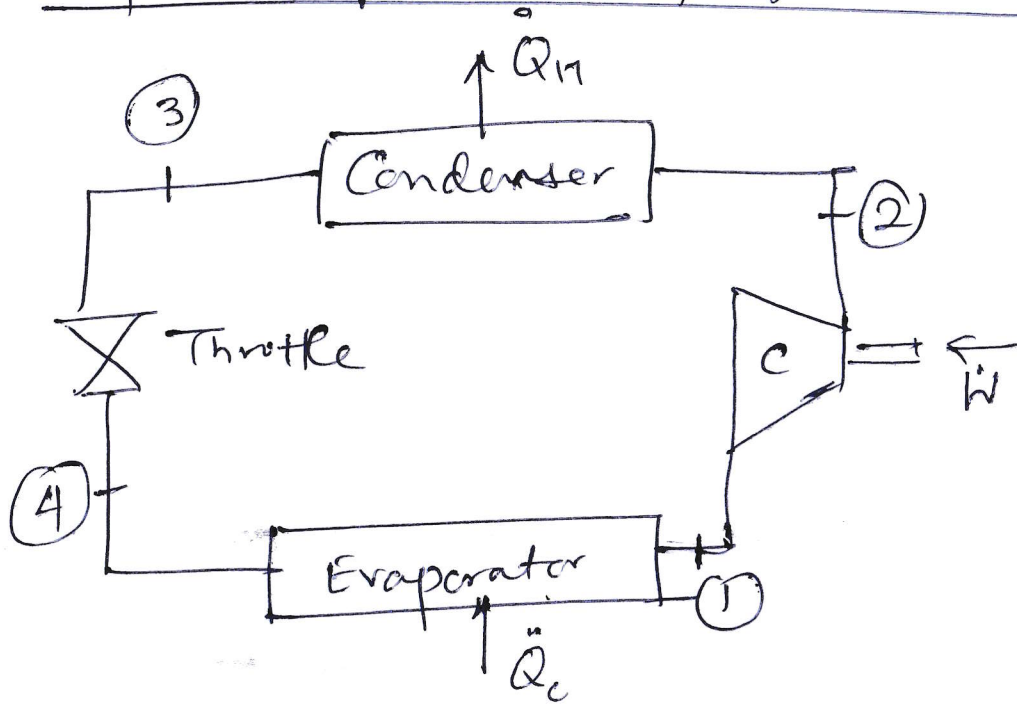


Vapor Compression Refrigeration cycle



1 → 2 : Isentropic compression

2 → 3 : Constant pressure heat rejection

3 → 4 : Constant enthalpy

4 → 1 : Constant pressure heat addition

$$w = h_2 - h_1$$

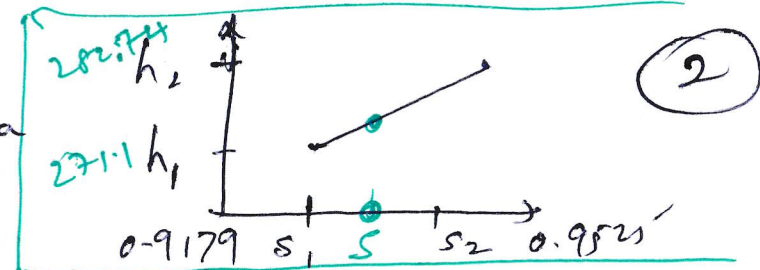
$$q_H = h_2 - h_3$$

$$h_3 = h_4$$

$$q_C = h_1 - h_4$$

$$1 \text{ bar} = 100 \text{ kPa}$$

$$10 \text{ bar} = 1000 \text{ kPa} = 1 \text{ MPa}$$



R-134a is used as refrigerant for a 2 ton refrigeration system. Saturated vapor leaves the evaporator at -12°C . Saturated liquid exits the condenser at 10 bar. The isentropic efficiency of the compressor is 90%. Determine 1) the mass of refrigerant 2) Power required to run the compressor 3) Coefficient of performance 4) Entropy generated in the compressor.

$$\dot{Q}_c = 2 \text{ tons} = 7 \text{ kW}$$

$$1 \text{ ton of refrigeration} = 3.5 \text{ kW}$$

$$s_{2s} = 0.93911 \text{ kJ/kg K}$$

$$\text{State 2: } h_{2s} = \left(\frac{0.93911 - 0.9179}{0.9525 - 0.9179} \right) (282.74 - 271.71) + 271.71 = 278.47 \text{ kJ/kg}$$

$$h_1 = 243.3 \text{ kJ/kg}$$

$$s_1 = 0.93911 \text{ kJ/kg K}$$

$$\eta = \frac{h_{2s} - h_1}{h_2 - h_1} \Rightarrow 0.9 = \frac{278.47 - 243.3}{h_2 - 243.3}$$

$$\Rightarrow h_2 = 282.4 \text{ kJ/kg}$$

$$s_2 \approx 0.9525 \text{ kJ/kg K}$$

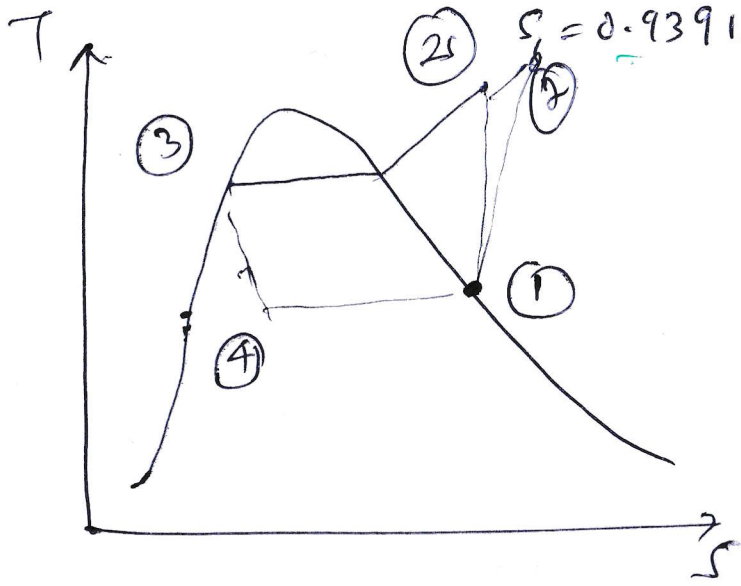
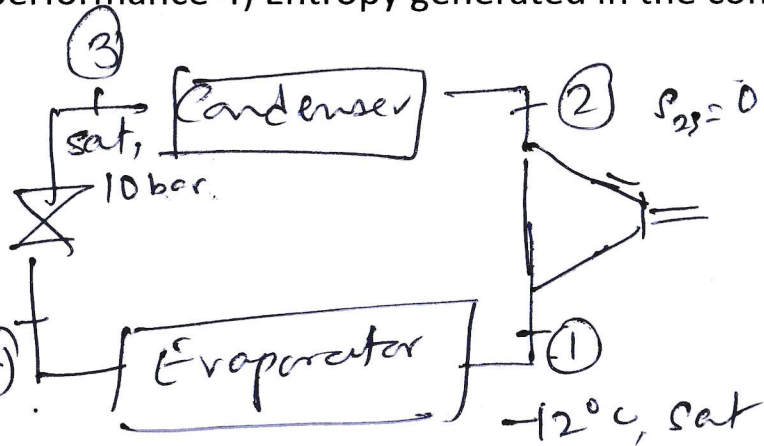
$$\text{State 3: } p = 10 \text{ bar, sat } h_3 = 107.32 \text{ kJ/kg K}$$

$$s_3 = 0.39189$$

$$\text{State 4}$$

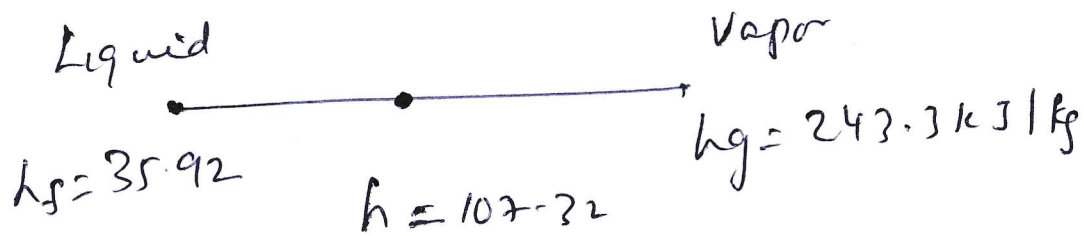
$$h_4 = h_3 = 107.32 \text{ kJ/kg K;}$$

$$At -12^\circ$$



At -12°C , $h_f = 35.92 \text{ kJ/kg}$ $h_g = 243.3 \text{ kJ/kg}$.

(3)



Let x denote the quantity of vapor

(x = condition, quality)

$$h_g = (1-x)h_f + x h_g \Rightarrow x = \frac{h - h_f}{h_g - h_f} = \frac{107.32 - 35.92}{243.3 - 35.92} = 0.344$$

$$s_g = (1-x)s_f + x s_g \Rightarrow s_g = (1-0.344) 0.14504 + (0.344)(0.93911) = 0.4182 \text{ kJ/kg K}$$

$$1) \dot{Q}_c = \dot{m}(h_1 - h_4) \Rightarrow 7 \text{ kW} = \dot{m}(243.3 - 107.32) \Rightarrow \boxed{\dot{m} = 0.0515 \text{ kg/s}}$$

$$(2) \dot{W} = \dot{m}(h_2 - h_1) = (0.0515)(282.4 - 243.3) = \boxed{2.013 \text{ kW}}$$

$$(3) \text{COP} = \frac{\dot{Q}_c}{\dot{W}} = \frac{7}{2.013} = 3.48$$

$$(4) \Delta \dot{S}_{\text{compressor}} = \dot{m}(s_2 - s_1) = (0.0515)(0.9525 - 0.93911) = 6.9 \times 10^{-4} \text{ kW/K}$$

$$\Delta \dot{S}_{\text{throttle}} = \dot{m}(s_4 - s_3) = (0.0515)(0.4182 - 0.39189) = 1.35 \times 10^{-3} \text{ kW/K}$$