

Masoneilan Control Valve Sizing Handbook

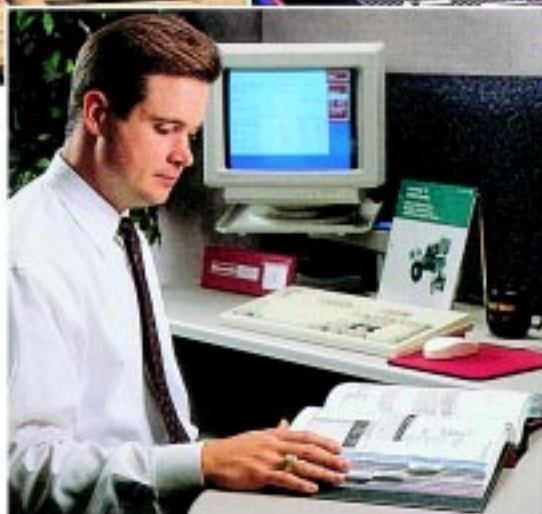


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Note: Tables for C_v , F_L , x_T and K_C vs Travel are found in publication Supplement to Masoneilan Control Valve Sizing Handbook OZ1000.

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Foreword

This handbook on control valve sizing is based on the use of nomenclature and sizing equations from ISA Standard S75.01 and IEC Standard 534-2. Additional explanations and supportive information are provided beyond the content of the standards.

The sizing equations are based on equations for predicting the flow of compressible and incompressible fluids through control valves. The equations are not intended for use when dense slurries, dry solids or non-Newtonian liquids are encountered.

Original equations and methods developed by Masoneilan are included for two-phase flow, multistage flow, and supercritical fluids.

Values of numerical factors are included for commonly encountered systems of units. These are United States customary units and metric units for both kilopascal and bar usage.

The principal use of the equations is to aid in the selection of an appropriate valve size for a specific application. In this procedure, the numbers in the equations consist of values for the fluid and flow conditions and known values for the selected valve at rated opening. With these factors in the equation, the unknown (or product of the unknowns, e.g., $F_p C_v$) can be computed. Although these computed numbers are often suitable for selecting a valve from a series of discrete sizes, they do not represent a true operating condition. Some of the factors are for the valve at rated travel, while others relating to the operating conditions are for the partially open valve.

Once a valve size has been selected, the remaining unknowns, such as F_p , can be computed and a judgement can be made as to whether the valve size is adequate. It is not usually necessary to carry the calculations further to predict the exact opening. To do this, all the pertinent sizing factors must be known at fractional valve openings. A computer sizing program having this information in a database can perform this task.

Flow Coefficient C_v

The use of the flow coefficient, C_v , first introduced by Masoneilan in 1944, quickly became accepted as the universal yardstick of valve capacity. So useful has C_v become, that practically all discussions of valve design and characteristics or flow behavior now employ this coefficient.

By definition, the valve flow coefficient, C_v , is the number of U. S. gallons per minute of water that will pass

through a given flow restriction with a pressure drop of one psi. For example, a control valve that has a maximum flow coefficient, C_v , of 12 has an effective port area in the full open position such that it passes 12 gpm of water with one psi pressure drop. Basically, it is a capacity index upon which the engineer can rapidly and accurately estimate the required size of a restriction in any fluid system.

Operating Conditions

The selection of a correct valve size, as determined by formula, is always premised on the assumption of full knowledge of the actual flowing conditions. Frequently, one or more of these conditions is arbitrarily assumed. It is the evaluation of these arbitrary data that really determines the final valve size. **No formulas, only good common sense combined with experience, can solve this problem.**

There is no substitute for good engineering judgement. Most errors in sizing are due to incorrect assumptions as to actual flowing conditions. Generally speaking, the tendency is to make the valve too large to be on the "safe" side (commonly referred to as "oversizing"). A combination of several of these "safety factors" can result in a valve so greatly oversized it tends to be troublesome.

Specific Gravity

In the flow formulas, the specific gravity is a square root function ; therefore, small differences in gravity have a minor effect on valve capacity. If the specific gravity is not

known accurately, a reasonable assumption will suffice. The use of .9 specific gravity, for example, instead of .8 would cause an error of less than 5 % in valve capacity.

Pressure Drop Across the Valve

On a simple back pressure or pressure reducing application, the drop across the valve may be calculated quite accurately. This may also be true on a liquid level control installation, where the liquid is passing from one vessel at a constant pressure to another vessel at a lower constant pressure. If the pressure difference is relatively small, some allowance may be necessary for line friction. On the other hand, in a large percentage of control applications, the pressure drop across the valve will be chosen arbitrarily.

Any attempt to state a specific numerical rule for such a choice becomes too complex to be practical. The design drop across the valve is sometimes expressed as a percentage of the friction drop in the system, exclusive of the valve. A good working rule is that 50% of this friction drop should be available as drop across the valve. In other words, one-third of the total system drop, including all heat exchangers, mixing nozzles, piping etc., is assumed to be absorbed by the control valve. This may sound excessive, but if the control valve were completely eliminated from such a system, the flow increase would only be about 23%. In pump discharge systems, the head characteristic of the pump becomes a major factor. For valves installed in extremely long or high-pressure drop lines, the percentage of drop across the valve may be somewhat lower, but at least 15% (up to 25% where possible) of the system drop should be taken.

Remember one important fact, the pressure differential absorbed by the control valve in actual operation will be the difference between the total available head and that required to maintain the desired flow through the valve. It is determined by the system characteristics rather than by the theoretical assumptions of the engineer. In the interest of economy, the engineer tries to keep the control valve pressure drop as low as possible. However, a valve can only regulate flow by absorbing and giving up pressure drop to the system. As the proportion of the system drop across the valve is reduced, its ability to further increase flow rapidly disappears.

In some cases, it may be necessary to make an arbitrary choice of the pressure drop across the valve because meager process data are available. For instance, if the valve is in a pump discharge line, having a discharge pressure of 7 bar (100 psi), a drop of 0.7 to 1.7 bar (10 to 25 psi) may be assumed sufficient. This is true if the pump discharge line is not extremely long or complicated by large drops through heat exchangers or other equipment. The tendency should be to use the higher figure.

On more complicated systems, consideration should be given to both maximum and minimum operating conditions. Masoneilan Engineering assistance is available for analysis of such applications.

Flowing Quantity

The selection of a control valve is based on the required flowing quantity of the process. The control valve must be selected to operate under several different conditions. The maximum quantity that a valve should be required to pass is 10 to 15 % above the specified maximum flow. The normal flow and maximum flow used in size calculations should be based on actual operating conditions, whenever possible, without any factors having been applied.

On many systems, a reduction in flow means an increase in pressure drop, and the C_v ratio may be much greater than would be suspected. If, for example, the maximum operating conditions for a valve are 200 gpm at 25 psi

drop, and the minimum conditions are 25 gpm at 100 psi drop, the C_v ratio is 16 to 1, not 8 to 1 as it would first seem. The required change in valve C_v is the product of the ratio of maximum to minimum flow and the square root of the ratio of maximum to minimum pressure drop, e.g.,

$$\frac{200 \times \sqrt{100}}{25 \times \sqrt{25}} = \frac{16}{1}$$

There are many systems where the increase in pressure drop for this same change in flow is proportionally much greater than in this case.

Liquid Flow Equations

Flow of Non-vaporizing Liquid

The following equations are used to determine the required capacity of a valve under fully turbulent, non-vaporizing liquid flow conditions. Note F_p equals unity for the case of valve size equal to line size.

$$\text{volumetric flow} \quad C_v = \frac{q}{N_1 F_p} \sqrt{\frac{G_f}{p_1 - p_2}}$$

$$\text{mass flow} \quad C_v = \frac{w}{N_6 F_p \sqrt{(p_1 - p_2) \gamma_1}}$$

Choked Flow of Vaporizing Liquid

Choked flow is a limiting flow rate. With liquid streams, choking occurs as a result of vaporization of the liquid when the pressure within the valve falls below the vapor pressure of the liquid.

Liquid flow is choked if

$$\Delta p \geq F_L^2 (p_1 - F_F p_v)$$

In this case, the following equations are used.

$$\text{volumetric flow} \quad C_v = \frac{q}{N_1 F_{LP}} \sqrt{\frac{G_f}{p_1 - F_F p_v}}$$

$$\text{mass flow} \quad C_v = \frac{w}{N_6 F_{LP} \sqrt{(p_1 - F_F p_v) \gamma_1}}$$

Nomenclature

- C_v = valve flow coefficient
 N = numerical constants based on units used (see Table 1)
 F_p = piping geometry factor (reducer correction)
 F_F = liquid critical pressure factor = $0.96 - 0.28 \sqrt{\frac{p_v}{p_c}}$
 F_L = liquid pressure recovery factor for a valve
 F_{LP} = combined pressure recovery and piping geometry factor for a valve with attached fittings
 K_i = velocity head factors for an inlet fitting, dimensionless
 p_c = pressure at thermodynamic critical point
 q = volumetric flow rate
 G_f = specific gravity at flowing temperature (water = 1) @ 60°F/15.5°C
 p_1 = upstream pressure
 p_v = vapor pressure of liquid at flowing temperature
 p_2 = downstream pressure
 w = weight (mass) flow rate
 γ_1 = specific weight (mass density) upstream conditions

Numerical Constants for Liquid Flow Equations

Constant		Units Used in Equations				
N		w	q	p, Δp	d, D	γ ₁
N ₁	0.0865	-	m ³ /h	kPa	-	-
	0.865	-	m ³ /h	bar	-	-
	1.00	-	gpm	psia	-	-
N ₂	0.00214	-	-	-	mm	-
	890.0	-	-	-	in	-
N ₆	2.73	kg/h	-	kPa	-	kg/m ³
	27.3	kg/h	-	bar	-	kg/m ³
	63.3	lb/h	-	psia	-	lb/ft ³

Table 1

Liquid Pressure Recovery Factor F_L

The liquid pressure recovery factor is a dimensionless expression of the pressure recovery ratio in a control valve. Mathematically, it is defined as follows:

$$F_L = \sqrt{\frac{p_1 - p_2}{p_1 - p_{vc}}}$$

In this expression, p_{vc} is the pressure at the vena contracta in the valve.

Liquid pressure recovery factors for various valve types at rated travel and at lower valve travel are shown in product bulletins. These values are determined by laboratory test in accordance with prevailing ISA and IEC standards.

Combined Liquid Pressure Recovery Factor F_{LP}

When a valve is installed with reducers, the liquid pressure recovery of the valve reducer combination is not the same as that for the valve alone. For calculations involving choked flow, it is convenient to treat the piping geometry factor F_p and the F_L factor for the valve reducer combination as a single factor F_{LP} . The value of F_L for the combination is then F_{LP}/F_p where :

$$\frac{F_{LP}}{F_p} = \sqrt{\frac{p_1 - p_2}{p_1 - p_{vc}}}$$

The following equation may be used to determine F_{LP} .

$$F_{LP} = F_L \left(\frac{K_i F_L^2 C_v^2}{N_2 d^4} + 1 \right)^{-1/2}$$

where $K_i = K_1 + K_{B1}$ (inlet loss and Bernoulli coefficients)

Cavitation in Control Valves

Cavitation, a detrimental process long associated with pumps, gains importance in control valves due to higher pressure drops for liquids and increased employment of high capacity valves (e.g., butterfly and ball valves).

Cavitation, briefly, is the transformation of a portion of liquid into the vapor phase during rapid acceleration of the fluid in the valve orifice, and the subsequent collapse of vapor bubbles downstream. The collapse of vapor bubbles can cause localized pressure up to 7000 bar (100,000 psi) and are singly, most responsible for the rapid wear of valve trim under high pressure drop conditions. Cavitation leads to rapid deterioration of the valve body plug and seat. It also leads to noise and vibration problems and as well, poses a potential safety hazard.

It is, therefore, necessary to understand and to prevent this phenomenon, particularly when high pressure drop conditions are encountered.

Cavitation in a control valve handling a pure liquid may occur if the static pressure of the flowing liquid decreases to a value less than the fluid vapor pressure. At this point, continuity of flow is broken by the formation of vapor bubbles. Since all control valves exhibit some pressure recovery, the final downstream pressure is generally higher than the orifice throat static pressure. When downstream pressure is higher than vapor pressure of the fluid, the vapor bubbles revert back to liquid. This two-stage transformation is defined as cavitation.

The pressure recovery in a valve is a function of its particular internal geometry. In general, the more streamlined a valve is, the more pressure recovery is experienced. This increases the possibility of cavitation.

The pressure drop in a valve at which cavitation is experienced is termed as critical pressure drop. Full cavitation will exist if actual pressure drop is greater than critical pressure drop, and if the downstream pressure is greater than fluid vapor pressure.

Mathematically, the critical pressure drop can be defined as follows:

$$\Delta p_{crit} = F_L^2 (p_1 - F_F p_v),$$

$$\text{with reducers } \Delta p_{crit} = \left(\frac{F_{LP}}{F_p} \right)^2 (p_1 - F_F p_v),$$

$$\text{where } F_F = 0.96 - 0.28 \sqrt{\frac{p_v}{p_c}}$$

How to Avoid Cavitation

Referring to the relationship for the critical pressure drop, one remedy for a potential application is to decrease the intended pressure drop across the valve to below critical pressure drop. Another possibility is to increase both inlet and outlet pressures by locating a valve at a lower elevation in the piping system : this results in an increase in critical pressure drop.

Another solution is to select a valve that has a higher F_L factor.

For an extremely high pressure drop, a Masoneilan anti-cavitation valve with multiple velocity-headloss trim is recommended.

Effect of Pipe Reducers

When valves are mounted between pipe reducers, there is a decrease in actual valve capacity. The reducers cause an additional pressure drop in the system by acting as contractions and enlargements in series with the valve. The Piping Geometry Factor, F_p , is used to account for this effect.

Summation

$$\Sigma K = K_1 + K_2 + K_{B1} - K_{B2}$$

When inlet and outlet reducers are the same size, the Bernoulli coefficients cancel out.

Piping Geometry Factor

$$F_p = \left(\frac{C_v^2 \Sigma K}{N_2 d^4} + 1 \right)^{-1/2}$$

Pipe Reducer Equations

Loss Coefficients

$$\text{inlet } K_1 = 0.5 \left[1 - \left(\frac{d}{D_1} \right)^2 \right]^2$$

$$\text{outlet } K_2 = \left[1 - \left(\frac{d}{D_2} \right)^2 \right]^2$$

Bernoulli Coefficients

$$K_{B1} = 1 - \left(\frac{d}{D_1} \right)^4$$

$$K_{B2} = 1 - \left(\frac{d}{D_2} \right)^4$$

Nomenclature

C_v = valve flow capacity coefficient

d = valve end inside diameter

D_1 = inside diameter of upstream pipe

D_2 = inside diameter of downstream pipe

F_p = piping geometry factor, dimensionless

K_1 = pressure loss coefficient for inlet reducer, dimensionless

K_2 = pressure loss coefficient for outlet reducer, dimensionless

K_{B1} = pressure change (Bernoulli) coefficient for inlet reducer, dimensionless

K_{B2} = pressure change (Bernoulli) coefficient for outlet reducer, dimensionless

$\Sigma K = K_1 + K_2 + K_{B1} - K_{B2}$, dimensionless

Equations for Nonturbulent Flow

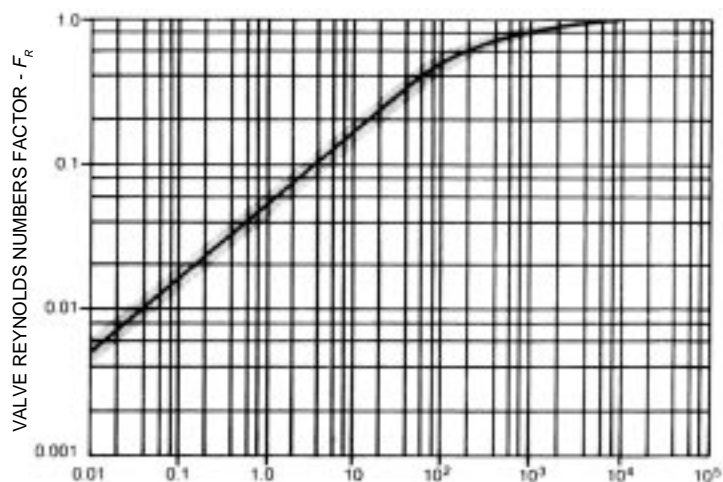
Laminar or transitional flow may result when the liquid viscosity is high, or when valve pressure drop or C_v is small. The Valve Reynolds Number Factor is used in the equations as follows :

volumetric flow
$$C_v = \frac{q}{N_1 F_R} \sqrt{\frac{G_f}{p_1 - p_2}}$$

mass flow
$$C_v = \frac{w}{N_6 F_R \sqrt{(p_1 - p_2) \gamma_1}}$$

The valve Reynolds number is defined as follows :

$$Re_v = \frac{N_4 F_d q}{v F_L^{1/2} C_v} \left(\frac{F_L^2 C_v^2}{N_2 d^4} + 1 \right)^{1/4}$$



Valve Reynolds Number - Re_v
Figure 1 Reynolds Number Factor

Nomenclature

- C_v = valve flow capacity coefficient
- d = nominal valve size
- F_d = valve style modifier, dimensionless
- F_L = Liquid pressure recovery factor
- F_R = Reynolds number correction factor, dimensionless
- G = specific gravity of liquid relative to water
- Δp = valve pressure drop
- q = volumetric flow rate
- Re_v = valve Reynolds number, dimensionless
- w = weight (mass) flow rate
- γ = mass density of liquid
- v = kinematic viscosity, centistokes

Numerical Constants for Liquid Flow Equations

Constant		Units Used in Equations				
	N	w	q	p, Δp	d, D	γ_1
N_1	0.0865	-	m ³ /h	kPa	-	-
	0.865	-	m ³ /h	bar	-	-
	1.00	-	gpm	psia	-	-
N_2	0.00214	-	-	-	mm	-
	890.0	-	-	-	in	-
N_4	76000	-	m ³ /h	-	mm	-
	17300	-	gpm	-	in	-
N_6	2.73	kg/h	-	kPa	-	kg/m ³
	27.3	kg/h	-	bar	-	kg/m ³
	63.3	lb/h	-	psia	-	lb/ft ³

Table 2

Representative F_d Factors

Single Port Globe Valves	$F_d = 1.0$
Double Port Globe Valves	$F_d = 0.7$
Camflex Valves	$F_d = 1.0$
Ball Valves	$F_d = 1.0$
Butterfly Valves	$F_d = 0.7$

In general, an F_d value of 1.0 can be used for valves with one single flow passage. An F_d value of 0.7 can be used for valves with two flow passages, such as double port globe valves and butterfly valves.

Gas and Vapor Flow Equations

volumetric flow

$$C_v = \frac{q}{N_7 F_p p_1 Y} \sqrt{\frac{G_g T_1 Z}{x}}$$

or

$$C_v = \frac{q}{N_9 F_p p_1 Y} \sqrt{\frac{M T_1 Z}{x}} *$$

mass flow

$$C_v = \frac{w}{N_6 F_p Y \sqrt{x p_1 \gamma_1}} *$$

or

$$C_v = \frac{w}{N_8 F_p p_1 Y} \sqrt{\frac{T_1 Z}{x M}} *$$

Gas expansion factor

$$Y = 1 - \frac{x}{3 F_k x_T}$$

Pressure drop ratio

$$x = \frac{\Delta p}{p_1}$$

Ratio of specific heats factor

$$F_k = \frac{k}{1.40}$$

The IEC 534-2 equations are identical to the above ISA equations (marked with an *) except for the following symbols :

k (ISA) corresponds to γ (IEC)
 γ_1 (ISA) corresponds to ρ_1 (IEC)

Nomenclature

C_v = valve flow coefficient
 F_k = ratio of specific heats factor, dimensionless
 F_p = piping geometry factor (reducer correction)
 p_1 = upstream pressure
 p_2 = downstream pressure
 q = volumetric flow rate
 N = numerical constant based on units (see table below)
 G_g = gas specific gravity. Ratio of gas density at standard conditions
 T_1 = absolute inlet temperature
 M = gas molecular weight
 x = pressure drop ratio, $\Delta p/p_1$ Limit $x = F_k x_T$
 Z = gas compressibility factor
 Y = gas expansion factor, $Y = 1 - \frac{x}{3 F_k x_T}$
 x_T = pressure drop ratio factor
 γ_1 = (Gamma) specific weight (mass density), upstream conditions
 w = weight (mass) flow rate
 k = gas specific heat ratio

Numerical Constants for Gas and Vapor Flow Equations

Constant		Units Used in Equations				
	N	w	q*	p, Δp	γ_1	T_1
N ₆	2.73	kg/h	-	kPa	kg/m ³	-
	27.3	kg/h	-	bar	kg/m ³	-
	63.3	lb/h	-	psia	lb/ft ³	-
N ₇	4.17	-	m ³ /h	kPa	-	K
	417.0	-	m ³ /h	bar	-	K
	1360.0	-	scfh	psia	-	R
N ₈	0.948	kg/h	-	kPa	-	K
	94.8	kg/h	-	bar	-	K
	19.3	lb/h	-	psia	-	R
N ₉	22.5	-	m ³ /h	kPa	-	K
	2250.0	-	m ³ /h	bar	-	K
	7320.0	-	scfh	psia	-	R

*q is in cubic feet per hour measured at 14.73 psia and 60°F, or cubic meters per hour measured at 101.3 kPa and 15.6° C.

Table 3

Multistage Valve Gas and Vapor Flow Equations

volumetric flow

$$C_v = \frac{q}{N_7 F_p p_1 Y_M} \sqrt{\frac{G_g T_1 Z}{x_M}}$$

or

$$C_v = \frac{q}{N_9 F_p p_1 Y_M} \sqrt{\frac{M T_1 Z}{x_M}}$$

mass flow

$$C_v = \frac{w}{N_6 F_p Y_M \sqrt{x_M p_1 \gamma_1}}$$

or

$$C_v = \frac{w}{N_8 F_p p_1 Y_M} \sqrt{\frac{T_1 Z}{x_M M}}$$

$$Y_M = 1 - \frac{x_M}{3 F_k x_T}$$

$$x_M = F_M \frac{\Delta p}{p_1}, \quad \text{limit } x_M = F_k x_T$$

$$F_k = \frac{k}{1.40}$$

$$F_M = \text{Multistage Compressible Flow Factor} \\ (F_M = 0.74 \text{ for multistage valves})$$

Ratio of Specific Heats Factor F_k

The flow rate of a compressible fluid through a valve is affected by the ratio of specific heats. The factor F_k accounts for this effect. F_k has a value of 1.0 for air at moderate temperature and pressures, where its specific heat ratio is about 1.40.

For valve sizing purposes, F_k may be taken as having a linear relationship to k . Therefore,

$$F_k = \frac{k}{1.40}$$

Expansion Factor Y

The expansion factor accounts for the changes in density of the fluid as it passes through a valve, and for the change in the area of the vena contracta as the pressure drop is varied. The expansion factor is affected by all of the following influences :

1. Ratio of valve inlet to port area
2. Internal valve geometry
3. Pressure drop ratio, x
4. Ratio of specific heats, k
5. Reynolds Number

The factor x_T accounts for the influence of 1, 2 and 3; factor F_k accounts for the influence of 4. For all practical purposes, Reynolds Number effects may be disregarded for virtually all process gas and vapor flows.

As in the application of orifice plates for compressible flow measurement, a linear relationship of the expansion factor Y to pressure drop ratio x is used as below :

$$Y = 1 - \frac{x}{3 F_k x_T}$$

Two-Phase Flow Equations

Two phase flow can exist as a mixture of a liquid with a non-condensable gas or as a mixture of a liquid with its vapor. The flow equation below applies where the two phase condition exists at the valve inlet.

The flow equation accounts for expansion of the gas or vapor phase, and for possible vaporization of the liquid phase. It utilizes both the gas and liquid limiting sizing pressure drops.

The flow equation for a two phase mixture entering the valve is as follows.

Note : F_p equals unity for the case of valve size equal to line size.

$$C_v = \frac{w}{N_6 F_p} \sqrt{\frac{f_f}{\Delta p_f \gamma_f} + \frac{f_g}{\Delta p_g \gamma_g Y^2}}$$

Use the actual pressure drop for Δp_f and Δp_g , but with the limiting pressure drop for each individually as follows :

$$\Delta p_f = F_L^2 (p_1 - F_F p_v)$$

$$\Delta p_g = F_k x_T p_1$$

The use of this flow equation results in a required C_v greater than the sum of a separately calculated C_v for the liquid plus a C_v for the gas or vapor phase. This increased capacity models published two phase flow data quite well.

For the hypothetical case of all liquid flow ($f_f = 1$), the flow equation reduces to the liquid flow equation for mass flow.

For the hypothetical case of all gas or vapor flow ($f_g = 1$), the flow equation reduces to the gas and vapor flow equation for mass flow.

Nomenclature

- C_v = valve flow coefficient
- f_f = weight fraction of liquid in two phase mixture, dimensionless
- f_g = weight fraction of gas (or vapor) in two phase mixture, dimensionless
- F_F = liquid critical pressure factor = $0.96 - 0.28 \sqrt{\frac{p_v}{p_c}}$
- F_k = ratio of specific heats factor, dimensionless
- F_L = liquid pressure recovery factor
- F_p = piping geometry factor (reducer correction)
- p_1 = upstream pressure
- p_v = vapor pressure of liquid at flowing temperature
- Δp_f = pressure drop for the liquid phase
- Δp_g = pressure drop for the gas phase
- w = weight (mass) flow rate of two phase mixture
- x_T = pressure drop ratio factor
- Y = gas expansion factor, $Y = 1 - \frac{x}{3 F_k x_T}$
- γ_f = specific weight (mass density) of the liquid phase at inlet conditions
- γ_g = specific weight (mass density) of the gas or vapor phase at inlet conditions

Numerical Constants for Liquid Flow Equations

Constant		Units Used in Equations				
	N	w	q	p, Δp	d, D	γ ₁
N ₆	2.73	kg/h	-	kPa	-	kg/m ³
	27.3	kg/h	-	bar	-	kg/m ³
	63.3	lb/h	-	psia	-	lb/ft ³

Table 4

Choked Flow

If all inlet conditions are held constant and pressure drop ratio x is increased by lowering the downstream pressure, mass flow will increase to a maximum limit. Flow conditions where the value of x exceeds this limit are known as choked flow. Choked flow occurs when the jet stream at the vena contracta attains its maximum cross-sectional area at sonic velocity.

Values of x_T for various valve types at rated travel and at lower valve travel are shown in product bulletins. These values are determined by laboratory test.

When a valve is installed with reducers, the pressure ratio factor x_{TP} is different from that of the valve alone x_T . The following equation may be used to calculate x_{TP} :

$$x_{TP} = \frac{x_T}{F_p^2} \left(\frac{x_T K_i C_v^2}{N_5 d^4} + 1 \right)^{-1}, \quad \text{where}$$

$$K_i = K_1 + K_{B1} \text{ (inlet loss and Bernoulli coefficients)}$$

The value of N_5 is 0.00241 for d in mm, and 1000 for d in inches.

Supercritical Fluids

Fluids at temperatures and pressures above both critical temperature and critical pressure are denoted as supercritical fluids. In this region, there is no physical distinction between liquid and vapor. The fluid behaves as a compressible, but near the critical point great deviations from the perfect gas laws prevail. It is very important to take this into account through the use of actual specific weight (mass density) from thermodynamic tables (or the compressibility factor Z), and the actual ratio of specific heats.

Supercritical fluid valve applications are not uncommon. In addition to supercritical fluid extraction processes, some process applications may go unnoticed. For instance, the critical point of ethylene is 10°C (50°F) and 51.1 bar (742 psia). All ethylene applications above this point in both temperature and pressure are supercritical by definition.

In order to size valves handling supercritical fluids, use a compressible flow sizing equation with the weight (mass) rate of flow with actual specific weight (mass density), or the volumetric flow with actual compressibility factor. In addition, the actual ratio of specific heats should be used.

Compressibility Factor Z

For many real gases subjected to commonly encountered temperatures and pressures, the perfect gas laws are not satisfactory for flow measurement accuracy and therefore correction factors must be used.

Following conventional flow measurement practice, the compressibility factor Z, in the equation $PV = ZRT$, will be used. Z can usually be ignored below 7 bar (100 psi) for common gases.

The value of Z does not differ materially for different gases when correlated as a function of the reduced temperature, T_r , and reduced pressure, p_r , found from Figures 2 and 3.

Figure 2 is an enlargement of a portion of Figure 3. Values taken from these figures are accurate to approximately plus or minus two percent.

To obtain the value of Z for a pure substance, the reduced pressure and reduced temperature are calculated as the ratio of the actual absolute gas pressure and its corresponding critical absolute pressure and absolute temperature and its absolute critical temperature.

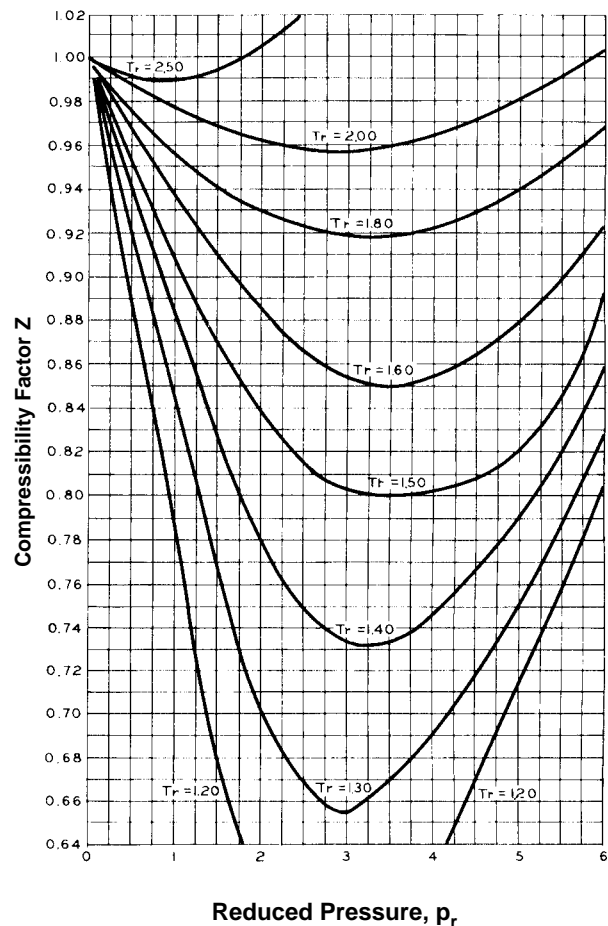
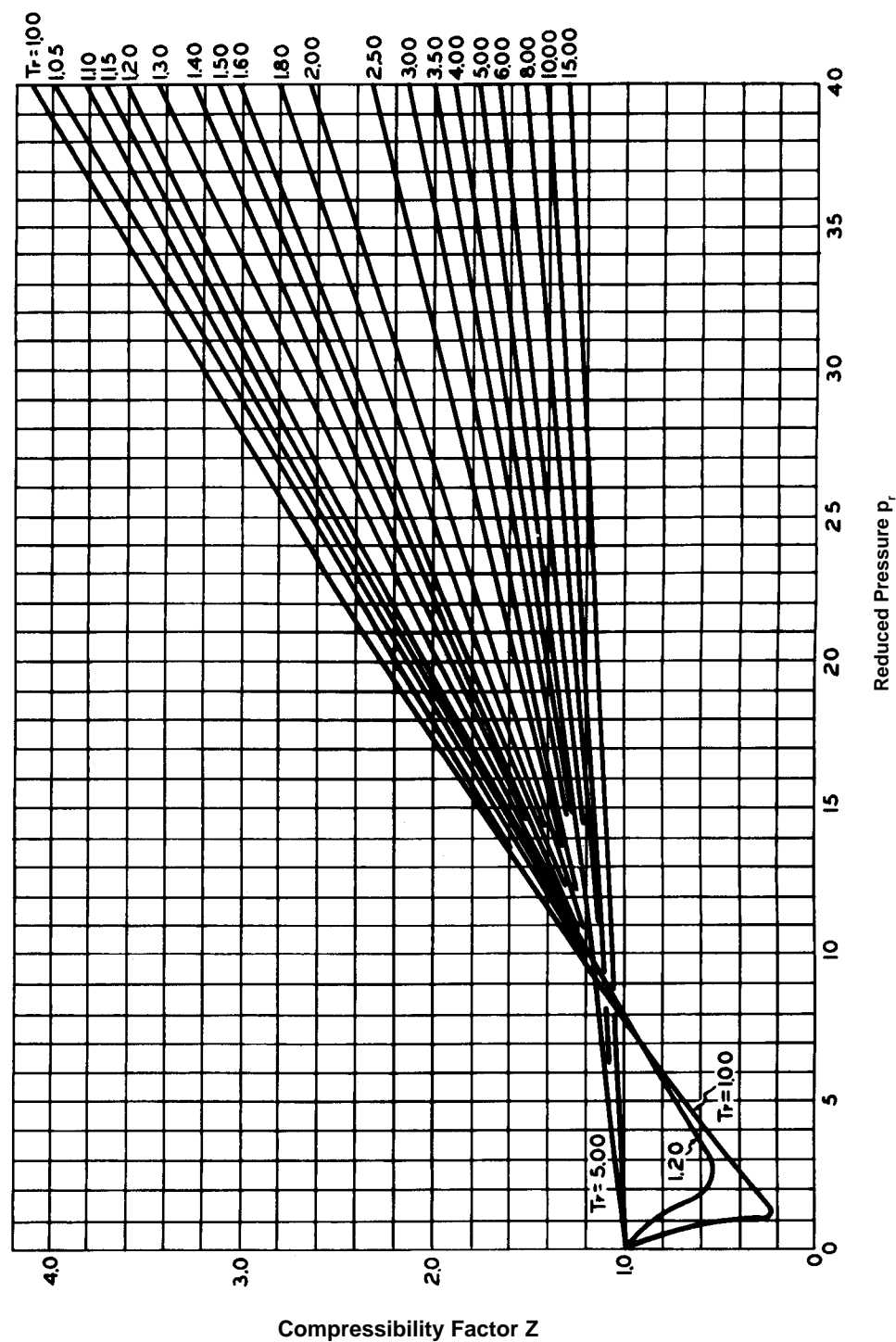


Figure 2
Compressibility Factors for Gases with
Reduced Pressures from 0 to 6

(Data from the charts of L. C. Nelson and E. F. Obert,
Northwestern Technological Institute)

The compressibility factor Z obtained from the Nelson-Obert charts is generally accurate within 3 to 5 percent. For hydrogen, helium, neon and argon, certain restrictions apply. Please refer to specialized literature.

Compressibility



$$p_r = \frac{\text{inlet pressure (absolute)}}{\text{critical pressure (absolute)}}$$

$$T_r = \frac{\text{inlet temperature (absolute)}}{\text{critical temperature (absolute)}}$$

Figure 3
Compressibility Factors for Gases with Reduced Pressures from 0 - 40
See Page 15 for critical pressures and temperatures

(Reproduced from the charts of L. C. Nelson and E. F. Obert, Northwestern Technological Institute)

Thermodynamic Critical Constants and Density of Elements, Inorganic and Organic Compounds

Element or Compound	Critical Pressure - p_c		Critical Temperature - T_c		$k^* \frac{C_p}{C_v}$
	psia	bar (abs)	°F	°C	
Acetic Acid, $\text{CH}_3\text{-CO-OH}$	841	58.0	612	322	1.15
Acetone, $\text{CH}_3\text{-CO-CH}_3$	691	47.6	455	235	-
Acetylene, C_2H_2	911	62.9	97	36	1.26
Air, $\text{O}_2\text{+N}_2$	547	37.8	-222	-141	1.40
Ammonia, NH_3	1638	113.0	270	132	1.33
Argon, A	705	48.6	-188	-122	1.67
Benzene, C_6H_6	701	48.4	552	289	1.12
Butane, C_4H_{10}	529	36.5	307	153	1.09
Carbon Dioxide, CO_2	1072	74.0	88	31	1.30
Carbon Monoxide, CO	514	35.5	-218	-139	1.40
Carbon Tetrachloride, CCl_4	661	45.6	541	283	-
Chlorine, Cl_2	1118	77.0	291	144	1.36
Ethane, C_2H_6	717	49.5	90	32	1.22
Ethyl Alcohol, $\text{C}_2\text{H}_5\text{OH}$	927	64.0	469	243	1.13
Ethylene, $\text{CH}_2\text{=CH}_2$	742	51.2	50	10	1.26
Ethyl Ether, $\text{C}_2\text{H}_5\text{-O-C}_2\text{H}_5$	522	36.0	383	195	-
Fluorine, F_2	367	25.3	-247	-155	1.36
Helium, He	33.2	2.29	-450	-268	1.66
Heptane, C_7H_{16}	394	27.2	513	267	-
Hydrogen, H_2	188	13.0	-400	-240	1.41
Hydrogen Chloride, HCl	1199	82.6	124	51	1.41
Isobutane, $(\text{CH}_3)_2\text{CH-CH}_3$	544	37.5	273	134	1.10
Isopropyl Alcohol, $\text{CH}_3\text{-CHOH-CH}_3$	779	53.7	455	235	-
Methane, CH_4	673	46.4	-117	-83	1.31
Methyl Alcohol, $\text{H-CH}_2\text{OH}$	1156	79.6	464	240	1.20
Nitrogen, N_2	492	34.0	-233	-147	1.40
Nitrous Oxide, N_2O	1054	72.7	99	37	1.30
Octane, $\text{CH}_3\text{-(CH}_2)_6\text{-CH}_3$	362	25.0	565	296	1.05
Oxygen, O_2	730	50.4	-182	-119	1.40
Pentane, C_5H_{12}	485	33.5	387	197	1.07
Phenol, $\text{C}_6\text{H}_5\text{OH}$	889	61.3	786	419	-
Phosgene, COCl_2	823	56.7	360	182	-
Propane, C_3H_8	617	42.6	207	97	1.13
Propylene, $\text{CH}_2\text{=CH-CH}_3$	661	45.6	198	92	1.15
Refrigerant 12, CCl_2F_2	582	40.1	234	112	1.14
Refrigerant 22, CHClF_2	713	49.2	207	97	1.18
Sulfur Dioxide, SO_2	1142	78.8	315	157	1.29
Water, H_2O	3206	221.0	705	374	1.32

* Standard Conditions

Table 5

Thermodynamic Critical Constants and Density of Elements, Inorganic and Organic Compounds

Element or Compound	Density - lb/ft ³ 14.7 psia & 60°F		Density - kg/m ³ 1013 mbar & 15.6°C		Mol Wt
	Liquid	Gas	Liquid	Gas	
Acetic Acid, CH ₃ -CO-OH	65.7		1052.4		66.1
Acetone, CH ₃ -CO-CH ₃	49.4		791.3		58.1
Acetylene, C ₂ H ₂		0.069		1.11	26.0
Air, O ₂ +N ₂		0.0764		1.223	29.0
Ammonia, NH ₃		0.045		0.72	17.0
Argon, A		0.105		1.68	39.9
Benzene, C ₆ H ₆	54.6		874.6		78.1
Butane, C ₄ H ₁₀		0.154		2.47	58.1
Carbon Dioxide, CO ₂		0.117		1.87	44.0
Carbon Monoxide, CO		0.074		1.19	28.0
Carbon Tetrachloride, CCl ₄	99.5		1593.9		153.8
Chlorine, Cl ₂		0.190		3.04	70.9
Ethane, C ₂ H ₆		0.080		1.28	30.1
Ethyl Alcohol, C ₂ H ₅ OH	49.52		793.3		46.1
Ethylene, CH ₂ =CH ₂		0.074		1.19	28.1
Ethyl Ether, C ₂ H ₅ -O-C ₂ H ₅	44.9		719.3		74.1
Fluorine, F ₂		0.097		1.55	38.0
Helium, He		0.011		0.18	4.00
Heptane, C ₇ H ₁₆	42.6		682.4		100.2
Hydrogen, H ₂		0.005		0.08	2.02
Hydrogen Chloride, HCl		0.097		1.55	36.5
Isobutane, (CH ₃) ₂ CH-CH ₃		0.154		2.47	58.1
Isopropyl Alcohol, CH ₃ -CHOH-CH ₃	49.23		788.6		60.1
Methane, CH ₄		0.042		0.67	16.0
Methyl Alcohol, H-CH ₂ OH	49.66		795.5		32.0
Nitrogen, N ₂		0.074		1.19	28.0
Nitrous Oxide, N ₂ O		0.117		1.87	44.0
Octane, CH ₃ -(CH ₂) ₆ -CH ₃	43.8		701.6		114.2
Oxygen, O ₂		0.084		1.35	32.0
Pentane, C ₅ H ₁₂	38.9		623.1		72.2
Phenol, C ₆ H ₅ OH	66.5		1065.3		94.1
Phosgene, COCl ₂		0.108		1.73	98.9
Propane, C ₃ H ₈		0.117		1.87	44.1
Propylene, CH ₂ =CH-CH ₃		0.111		1.78	42.1
Refrigerant 12, CCl ₂ F ₂		0.320		5.13	120.9
Refrigerant 22, CHClF ₂		0.228		3.65	86.5
Sulfur Dioxide, SO ₂		0.173		2.77	64.1
Water, H ₂ O	62.34		998.6		18.0

Table 5

Liquid Velocity in Commercial Wrought Steel Pipe

The velocity of a flowing liquid may be determined by the following expressions :

US Customary Units

$$v = .321 \frac{q}{A}$$

Where

v = velocity, ft/sec
 q = flow, gpm
 A = cross sectional area, sq in

Metric Units

$$v = 278 \frac{q}{A}$$

Where

v = velocity, meters/sec
 q = flow, meters³/hr
 A = cross sectional area, sq mm

Figure 4 gives the solution to these equations for pipes 1" through 12" over a wide flow range on both U. S. Customary and Metric Units.

Steam or Gas Flow in Commercial Wrought Steel Pipe

Steam or Gas (mass basis)

To determine the velocity of a flowing compressible fluid use the following expressions :

US Customary Units

$$v = .04 \frac{WV}{A}$$

Where

v = fluid velocity, ft/sec
 W = fluid flow, lb/hr
 V = specific volume, cu ft/lb
 A = cross sectional area, sq in

Metric Units

$$v = 278 \frac{WV}{A}$$

Where

v = fluid velocity, meters/sec
 W = fluid flow, kg/hr
 V = specific volume, m³/kg
 A = cross sectional area, mm²

Figure 5 is a plot of steam flow versus static pressure with reasonable velocity for Schedule 40 pipes 1" through 12" in US Customary and Metric Units.

Gas (volume basis)

To find the velocity of a flowing compressible fluid with flow in volume units, use the following formulas :

US Customary Units

$$v = .04 \frac{F}{A}$$

Where

v = fluid velocity, ft/sec
 F = gas flow, ft³/hr at flowing conditions*
 A = cross sectional area, sq in

*Note that gas flow must be at flowing conditions. If flow is at standard conditions, convert as follows :

$$F = \frac{\text{std ft}^3}{\text{hr}} \times \frac{14.7}{p} \times \frac{T}{520}$$

Where

p = pressure absolute, psia
 T = temperature absolute, R

Metric Units

$$v = 278 \frac{F}{A}$$

Where

v = fluid velocity, meters/sec
 F = gas flow, meters³/hr at flowing conditions*
 A = cross sectional area, sq mm

*Note that gas flow must be at flowing conditions. If flow is at standard conditions, convert as follows :

$$F = \frac{\text{std meters}^3}{\text{hr}} \times \frac{1.013}{p} \times \frac{T}{288}$$

Where

p = pressure absolute, bar
 T = temperature absolute, K

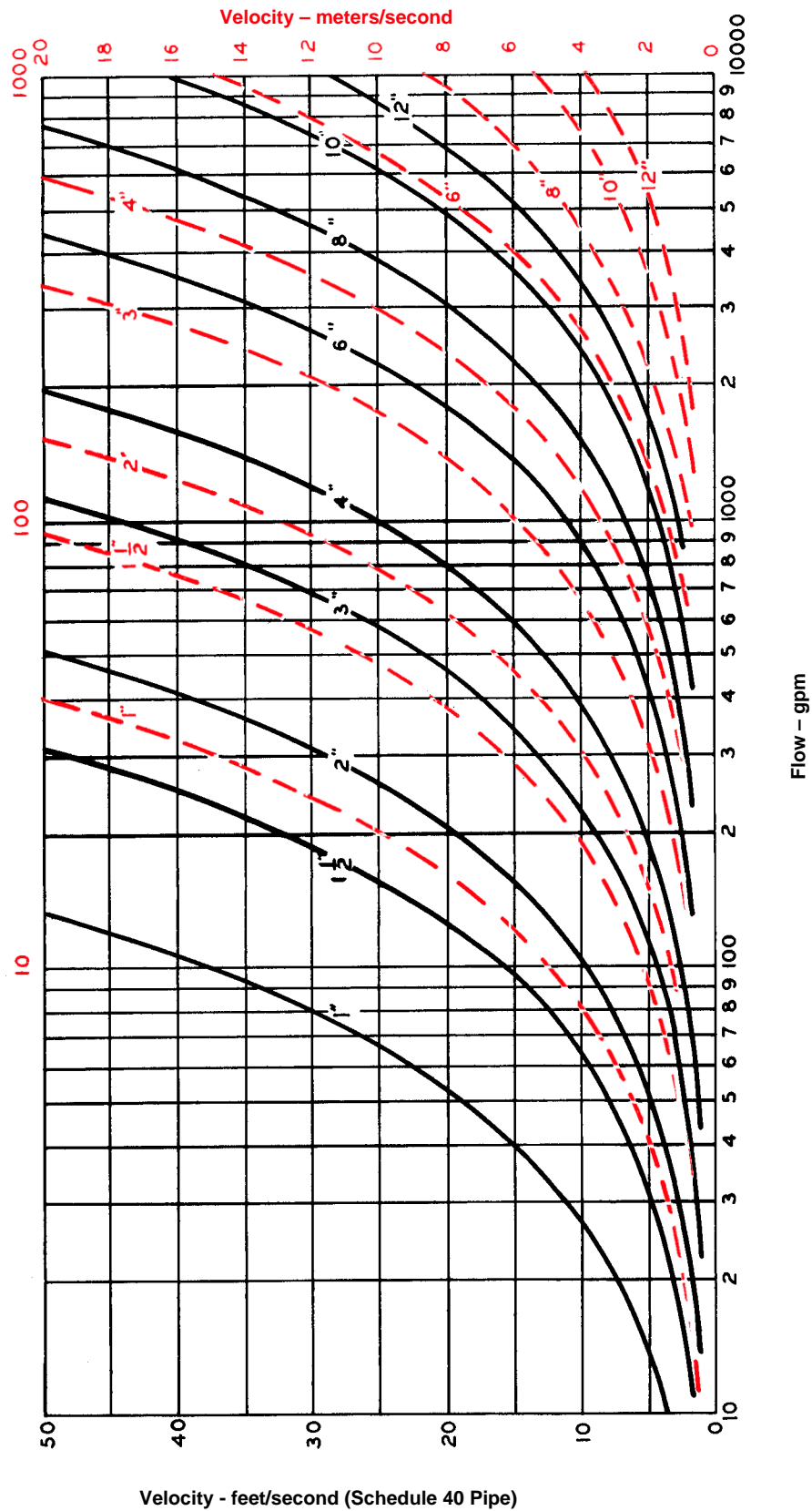


Figure 4

Liquid Velocity vs Flow Rate

US Customary Units

Metric Units

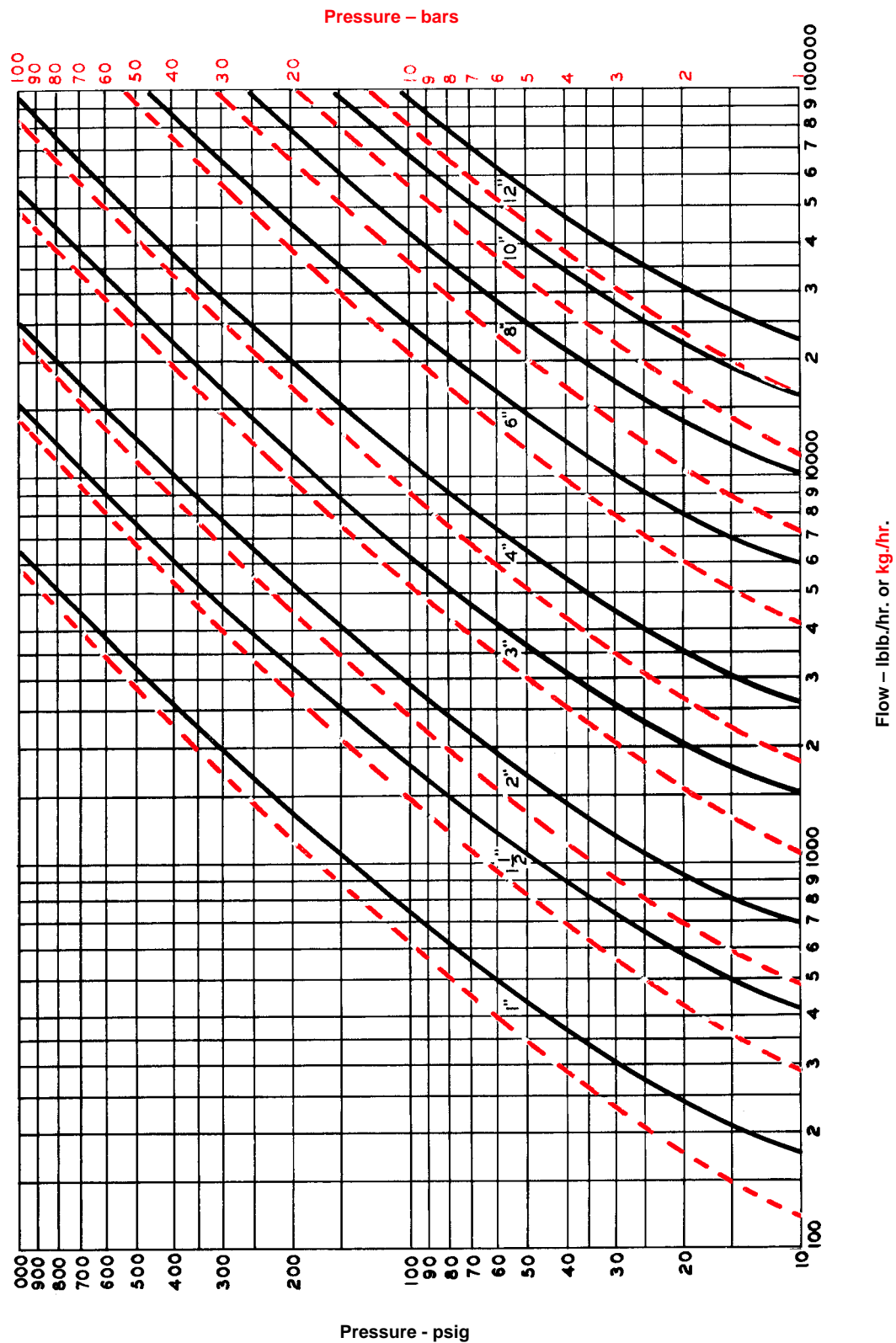


Figure 5
Saturated Steam Flow vs Pressure
for 1" to 12" Schedule 40 Pipe

US Customary Units
 Metric Units

Velocity -- 130 to 170 feet per second --
 -- 50 to 60 meters per second --

Commercial Wrought Steel Pipe Data (ANSI B36.10)

Nominal Pipe Size			O.D.	Wall Thickness		I.D.	Flow Area	
	mm	inches	inches	mm	inches	inches	mm ²	sq in
Schedule 10	350	14	14	6.35	0.250	13.5	92200	143
	400	16	16	6.35	0.250	15.5	121900	189
	450	18	18	6.35	0.250	17.5	155500	241
	500	20	20	6.35	0.250	19.5	192900	299
	600	24	24	6.35	0.250	23.5	280000	434
	750	30	30	7.92	0.312	29.4	437400	678
Schedule 20	200	8	8.63	6.35	0.250	8.13	33500	51.9
	250	10	10.8	6.35	0.250	10.3	53200	82.5
	300	12	12.8	6.35	0.250	12.3	76000	117.9
	350	14	14.0	7.92	0.312	13.4	90900	141
	400	16	16.0	7.92	0.312	15.4	120000	186
	450	18	18.0	7.92	0.312	17.4	152900	237
	500	20	20.0	9.53	0.375	19.3	187700	291
	600	24	24.0	9.53	0.375	23.3	274200	425
	750	30	30.0	12.70	0.500	29.0	426400	661
Schedule 30	200	8	8.63	7.04	0.277	8.07	33000	51.2
	250	10	10.8	7.80	0.307	10.1	52000	80.7
	300	12	12.8	8.38	0.330	12.1	74200	115
	350	14	14.0	9.53	0.375	13.3	89000	138
	400	16	16.0	9.53	0.375	15.3	118000	183
	450	18	18.0	11.13	0.438	17.1	148400	230
	500	20	20.0	12.70	0.500	19.0	183200	284
	600	24	24.0	14.27	0.562	22.9	265100	411
	750	30	30.0	15.88	0.625	28.8	418700	649
Schedule 40*	15	1/2	0.84	2.77	0.109	0.622	190	0.304
	20	3/4	1.05	2.87	0.113	0.824	340	0.533
	25	1	1.32	3.38	0.133	1.05	550	0.864
	32	1 1/4	1.66	3.56	0.140	1.38	970	1.50
	40	1 1/2	1.90	3.68	0.145	1.61	1300	2.04
	50	2	2.38	3.91	0.154	2.07	2150	3.34
	65	2 1/2	2.88	5.16	0.203	2.47	3100	4.79
	80	3	3.50	5.49	0.216	3.07	4700	7.39
	100	4	4.50	6.02	0.237	4.03	8200	12.7
	150	6	6.63	7.11	0.280	6.07	18600	28.9
	200	8	8.63	8.18	0.322	7.98	32200	50.0
	250	10	10.8	9.27	0.365	10.02	50900	78.9
	300	12	12.8	10.31	0.406	11.9	72200	112
	350	14	14.0	11.13	0.438	13.1	87100	135
	400	16	16.0	12.70	0.500	15.0	114200	177
	450	18	18.0	14.27	0.562	16.9	144500	224
	500	20	20.0	15.06	0.593	18.8	179300	278
	600	24	24.0	17.45	0.687	22.6	259300	402

*Standard wall pipe same as Schedule 40 through 10" size. 12" size data follows.

300	12	12.8	9.53	0.375	12.00	72900	113
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Table 6

Commercial Wrought Steel Pipe Data (ANSI B36.10) (continued)

Nominal Pipe Size			O.D.	Wall Thickness		I.D.	Flow Area	
	mm	inches	inches	mm	inches	inches	mm ²	sq in
Schedule 80*	15	1/2	0.84	3.73	0.147	0.546	150	0.234
	20	3/4	1.05	3.91	0.154	0.742	280	0.433
	25	1	1.32	4.55	0.179	0.957	460	0.719
	32	1 1/4	1.66	4.85	0.191	1.28	820	1.28
	40	1 1/2	1.90	5.08	0.200	1.50	1140	1.77
	50	2	2.38	5.54	0.218	1.94	1900	2.95
	65	2 1/2	2.88	7.01	0.276	2.32	2700	4.24
	80	3	3.50	7.62	0.300	2.90	4200	6.61
	100	4	4.50	8.56	0.337	3.83	7400	11.5
	150	6	6.63	10.97	0.432	5.76	16800	26.1
	200	8	8.63	12.70	0.500	7.63	29500	45.7
	250	10	10.8	15.06	0.593	9.56	46300	71.8
	300	12	12.8	17.45	0.687	11.4	65800	102
	350	14	14.0	19.05	0.750	12.5	79300	123
	400	16	16.0	21.41	0.843	14.3	103800	161
	450	18	18.0	23.80	0.937	16.1	131600	204
	500	20	20.0	26.16	1.03	17.9	163200	253
	600	24	24.0	30.99	1.22	21.6	235400	365
Schedule 160	15	1/2	0.84	4.75	0.187	0.466	110	0.171
	20	3/4	1.05	5.54	0.218	0.614	190	0.296
	25	1	1.32	6.35	0.250	0.815	340	0.522
	32	1 1/4	1.66	6.35	0.250	1.16	680	1.06
	40	1 1/2	1.90	7.14	0.281	1.34	900	1.41
	50	2	2.38	8.71	0.343	1.69	1450	2.24
	65	2 1/2	2.88	9.53	0.375	2.13	2300	3.55
	80	3	3.50	11.13	0.438	2.62	3500	5.41
	100	4	4.50	13.49	0.531	3.44	6000	9.28
	150	6	6.63	18.24	0.718	5.19	13600	21.1
	200	8	8.63	23.01	0.906	6.81	23500	36.5
	250	10	10.8	28.70	1.13	8.50	36600	56.8
	300	12	12.8	33.27	1.31	10.1	51900	80.5
	350	14	14.0	35.81	1.41	11.2	63400	98.3
	400	16	16.0	40.39	1.59	12.8	83200	129
	450	18	18.0	45.21	1.78	14.4	105800	164
	500	20	20.0	50.04	1.97	16.1	130900	203
	600	24	24.0	59.44	2.34	19.3	189000	293
Double Extra Strong	15	1/2	0.84	7.47	0.294	0.252	30	0.050
	20	3/4	1.05	7.82	0.308	0.434	90	0.148
	25	1	1.32	9.09	0.358	0.599	180	0.282
	32	1 1/4	1.66	9.70	0.382	0.896	400	0.630
	40	1 1/2	1.90	10.16	0.400	1.10	610	0.950
	50	2	2.38	11.07	0.436	1.50	1140	1.77
	65	2 1/2	2.89	14.02	0.552	1.77	1600	2.46
	80	3	3.50	15.24	0.600	2.30	2700	4.16
	100	4	4.50	17.12	0.674	3.15	5000	7.80
	150	6	6.63	21.94	0.864	4.90	12100	18.8
	200	8	8.63	22.22	0.875	6.88	23900	37.1

*Extra strong pipe same as Schedule 80 through 8" size. 10" & 12" size data follows.

250	10	10.8	12.70	0.500	9.75	48200	74.7
300	12	12.8	12.70	0.500	11.8	69700	108

Table 6

Properties of Steam

US Customary Units

Saturated					Superheated: Total Temperature - °F									
Abs. P'	Gauge P	Sat. Temp.	*	Sat	360	400	440	480	500	600	700	800	900	1000
14.696	0.0	212.00	V hg	26.80 1150.4	33.03 1221.1	34.68 1239.9	36.32 1258.8	37.96 1277.6	38.78 1287.1	42.86 1334.8	46.94 1383.2	51.00 1432.3	55.07 1482.3	59.13 1533.1
20.0	5.3	227.96	V hg	20.08 1156.3	24.21 1220.3	25.43 1239.2	26.65 1258.2	27.86 1277.1	28.46 1286.6	31.47 1334.4	34.47 1382.9	37.46 1432.1	40.45 1482.1	43.44 1533.0
30.0	15.3	250.33	V hg	13.746 1164.1	16.072 1218.6	16.897 1237.9	17.714 1257.0	18.528 1276.2	18.933 1285.7	20.95 1333.8	22.96 1382.4	24.96 1431.7	26.95 1481.8	28.95 1532.7
40.0	25.3	267.25	V hg	10.498 1169.7	12.001 1216.9	12.628 1236.5	13.247 1255.9	13.862 1275.2	14.168 1284.8	15.688 1333.1	17.198 1381.9	18.702 1431.3	20.20 1481.4	21.70 1532.4
50.0	35.3	281.01	V hg	8.515 1174.1	9.557 1215.2	10.065 1235.1	10.567 1254.7	11.062 1274.2	11.309 1283.9	12.532 1332.5	13.744 1381.4	14.950 1430.9	16.152 1481.1	17.352 1532.1
60.0	45.3	292.71	V hg	7.175 1177.6	7.927 1213.4	8.357 1233.6	8.779 1253.5	9.196 1273.2	9.403 1283.0	10.427 1331.8	11.441 1380.9	12.449 1430.5	13.452 1480.8	14.454 1531.9
70.0	55.3	302.92	V hg	6.206 1180.6	6.762 1211.5	7.136 1232.1	7.502 1252.3	7.863 1272.2	8.041 1282.0	8.924 1331.1	9.796 1380.4	10.662 1430.1	11.524 1480.5	12.383 1531.6
80.0	65.3	312.03	V hg	5.472 1183.1	5.888 1209.7	6.220 1230.7	6.544 1251.1	6.862 1271.1	7.020 1281.1	7.797 1330.5	8.562 1379.9	9.322 1429.7	10.077 1480.1	10.830 1531.3
90.0	75.3	320.27	V hg	4.896 1185.3	5.208 1207.7	5.508 1229.1	5.799 1249.8	6.084 1270.1	6.225 1280.1	6.920 1329.8	7.603 1379.4	8.279 1429.3	8.952 1479.8	9.623 1531.0
100.0	85.3	327.81	V hg	4.432 1187.2	4.663 1205.7	4.937 1227.6	5.202 1248.6	5.462 1269.0	5.589 1279.1	6.218 1329.1	6.835 1378.9	7.446 1428.9	8.052 1479.5	8.656 1530.8
120.0	105.3	341.25	V hg	3.728 1190.4	3.844 1201.6	4.081 1224.4	4.307 1246.0	4.527 1266.9	4.636 1277.2	5.165 1327.7	5.683 1377.8	6.195 1428.1	6.702 1478.8	7.207 1530.2
140.0	125.3	353.02	V hg	3.220 1193.0	3.258 1197.3	3.468 1221.1	3.667 1243.3	3.860 1264.7	3.954 1275.2	4.413 1326.4	4.861 1376.8	5.301 1427.3	5.738 1478.2	6.172 1529.7
160.0	145.3	363.53	V hg	2.834 1195.1	----- 1195.1	3.008 1217.6	3.187 1240.6	3.359 1262.4	3.443 1273.1	3.849 1325.0	4.244 1375.7	4.631 1426.4	5.015 1477.5	5.396 1529.1
180.0	165.3	373.06	V hg	2.532 1196.9	----- 1196.9	2.649 1214.0	2.813 1237.8	2.969 1260.2	3.044 1271.0	3.411 1323.5	3.764 1374.7	4.110 1425.6	4.452 1476.8	4.792 1528.6
200.0	185.3	381.79	V hg	2.288 1198.4	----- 1198.4	2.631 1210.3	2.513 1234.9	2.656 1257.8	2.726 1268.9	3.060 1322.1	3.380 1373.6	3.693 1424.8	4.002 1476.2	4.309 1528.0
220.0	205.3	389.86	V hg	2.087 1199.6	----- 1199.6	2.125 1206.5	2.267 1231.9	2.400 1255.4	2.465 1266.7	2.772 1320.7	3.066 1372.6	3.352 1424.0	3.634 1475.5	3.913 1527.5
240.0	225.3	397.37	V hg	1.918 1200.6	----- 1200.6	1.9276 1202.5	2.062 1228.8	2.187 1253.0	2.247 1264.5	2.533 1319.2	2.804 1371.5	3.068 1423.2	3.327 1474.8	3.584 1526.9
260.0	245.3	404.42	V hg	1.774 1201.5	----- 1201.5	----- 1225.7	1.8882 1250.5	2.006 1262.3	2.063 1317.7	2.330 1370.4	2.582 1422.3	2.827 1474.2	3.067 1526.3	3.305
280.0	265.3	411.05	V hg	1.651 1202.3	----- 1202.3	----- 1222.4	1.7388 1247.9	1.8512 1260.0	1.9047 1316.2	2.156 1369.4	2.392 1421.5	2.621 1473.5	2.845 1525.8	3.066
300.0	285.3	417.33	V hg	1.543 1202.8	----- 1202.8	----- 1219.1	1.6090 1245.3	1.7165 1257.6	1.7675 1314.7	2.005 1368.3	2.227 1420.6	2.442 1472.8	2.652 1525.2	2.859
320.0	305.3	423.29	V hg	1.448 1203.4	----- 1203.4	----- 1215.6	1.4950 1242.6	1.5985 1255.2	1.6472 1313.2	1.8734 1367.2	2.083 1419.8	2.285 1472.1	2.483 1524.7	2.678
340.0	325.3	428.97	V hg	1.364 1203.7	----- 1203.7	----- 1212.1	1.3941 1239.9	1.4941 1252.8	1.5410 1311.6	1.7569 1366.1	1.9562 1419.0	2.147 1471.5	2.334 1524.1	2.518
360.0	345.3	434.40	V hg	1.289 1204.1	----- 1204.1	----- 1208.4	1.3041 1237.1	1.4012 1250.3	1.4464 1310.1	1.6533 1365.0	1.8431 1418.1	2.025 1470.8	2.202 1523.5	2.376

* V = specific volume, cubic feet per pound

hg = total heat of steam, Btu per pound

Table 7

Properties of Steam (continued)

US Customary Units

Saturated					Superheated : Total Temperature - °F									
Abs. P'	Gauge P	Sat. Temp.	*	Sat	500	540	600	640	660	700	740	800	900	1000
380.0	365.3	439.60	V hg	1.222 1204.3	1.3616 1247.7	1.4444 1273.1	1.5605 1308.5	1.6345 1331.0	1.6707 1342.0	1.7419 1363.8	1.8118 1385.3	1.9149 1417.3	2.083 1470.1	2.249 1523.0
400.0	385.3	444.59	V hg	1.161 1204.5	1.2851 1245.1	1.3652 1271.0	1.4770 1306.9	1.5480 1329.6	1.5827 1340.8	1.6508 1362.7	1.7177 1384.3	1.8161 1416.4	1.9767 1469.4	2.134 1522.4
420.0	405.3	449.39	V hg	1.106 1204.6	1.2158 1242.5	1.2935 1268.9	1.4014 1305.3	1.4697 1328.3	1.5030 1339.5	1.5684 1361.6	1.6324 1383.3	1.7267 1415.5	1.8802 1468.7	2.031 1521.9
440.0	425.3	454.02	V hg	1.055 1204.6	1.1526 1239.8	1.2282 1266.7	1.3327 1303.6	1.3984 1326.9	1.4306 1338.2	1.4934 1360.4	1.5549 1382.3	1.6454 1414.7	1.7925 1468.1	1.9368 1521.3
460.0	445.3	458.50	V hg	1.009 1204.6	1.0948 1237.0	1.1685 1264.5	1.2698 1302.0	1.3334 1325.4	1.3644 1336.9	1.4250 1359.3	1.4842 1381.3	1.5711 1413.8	1.7124 1467.4	1.8508 1520.7
480.0	465.3	462.82	V hg	0.967 1204.5	1.0417 1234.2	1.1138 1262.3	1.2122 1300.3	1.2737 1324.0	1.3038 1335.6	1.3622 1358.2	1.4193 1380.3	1.5031 1412.9	1.6390 1466.7	1.7720 1520.2
500.0	485.3	467.01	V hg	0.927 1204.4	0.9927 1231.3	1.0633 1260.0	1.1591 1298.6	1.2188 1322.6	1.2478 1334.2	1.3044 1357.0	1.3596 1379.3	1.4405 1412.1	1.5715 1466.0	1.6996 1519.6
520.0	505.3	471.07	V hg	0.891 1204.2	0.9473 1228.3	1.0166 1257.7	1.1101 1296.9	1.1681 1321.1	1.1962 1332.9	1.2511 1355.8	1.3045 1378.2	1.3826 1411.2	1.5091 1465.3	1.6326 1519.0
540.0	525.3	475.01	V hg	0.857 1204.0	0.9052 1225.3	0.9733 1255.4	1.0646 1295.2	1.1211 1319.7	1.1485 1331.5	1.2017 1354.6	1.2535 1377.2	1.3291 1410.3	1.4514 1464.6	1.5707 1518.5
560.0	545.3	478.85	V hg	0.826 1203.8	0.8659 1222.2	0.9330 1253.0	1.0224 1293.4	1.0775 1318.2	1.1041 1330.2	1.1558 1353.5	1.2060 1376.1	1.2794 1409.4	1.3978 1463.9	1.5132 1517.9
580.0	565.3	482.58	V hg	0.797 1203.5	0.8291 1219.0	0.8954 1250.5	0.9830 1291.7	1.0368 1316.7	1.0627 1328.8	1.1131 1352.3	1.1619 1375.1	1.2331 1408.6	1.3479 1463.2	1.4596 1517.3
600.0	585.3	486.21	V hg	0.769 1203.2	0.7947 1215.7	0.8602 1248.1	0.9463 1289.9	0.9988 1315.2	1.0241 1327.4	1.0732 1351.1	1.1207 1374.0	1.1899 1407.7	1.3013 1462.5	1.4096 1516.7
620.0	605.3	489.75	V hg	0.744 1202.9	0.7624 1212.4	0.8272 1245.5	0.9118 1288.1	0.9633 1313.7	0.9880 1326.0	1.0358 1349.9	1.0821 1373.0	1.1494 1406.8	1.2577 1461.8	1.3628 1516.2
640.0	625.3	493.21	V hg	0.719 1202.5	0.7319 1209.0	0.7962 1243.0	0.8795 1286.2	0.9299 1312.2	0.9541 1324.6	1.0008 1348.6	1.0459 1371.9	1.1115 1405.9	1.2168 1461.1	1.3190 1515.6
660.0	645.3	496.58	V hg	0.697 1202.1	0.7032 1205.4	0.7670 1240.4	0.8491 1284.4	0.8985 1310.6	0.9222 1323.2	0.9679 1347.4	1.0119 1370.8	1.0759 1405.0	1.1784 1460.4	1.2778 1515.0
680.0	665.3	499.88	V hg	0.675 1201.7	0.6759 1201.8	0.7395 1237.7	0.8205 1282.5	0.8690 1309.1	0.8922 1321.7	0.9369 1346.2	0.9800 1369.8	1.0424 1404.1	1.1423 1459.7	1.2390 1514.5
700.0	685.3	503.10	V hg	0.655 1201.2	----- 1201.2	0.7134 1235.0	0.7934 1280.6	0.8411 1307.5	0.8639 1320.3	0.9077 1345.0	0.9498 1368.7	1.0108 1403.2	1.1082 1459.0	1.2024 1513.9
750.0	735.3	510.86	V hg	0.609 1200.0	----- 1200.0	0.6540 1227.9	0.7319 1275.7	0.7778 1303.5	0.7996 1316.6	0.8414 1341.8	0.8813 1366.0	0.9391 1400.9	1.0310 1457.2	1.1196 1512.4
800.0	785.3	518.23	V hg	0.568 1198.6	----- 1198.6	0.6015 1220.5	0.6779 1270.7	0.7223 1299.4	0.7433 1312.9	0.7833 1338.6	0.8215 1363.2	0.8763 1398.6	0.9633 1455.4	1.0470 1511.0
850.0	835.3	525.26	V hg	0.532 1197.1	----- 1197.1	0.5546 1212.7	0.6301 1265.5	0.6732 1295.2	0.6934 1309.0	0.7320 1335.4	0.7685 1360.4	0.8209 1396.3	0.9037 1453.6	0.9830 1509.5
900.0	885.3	531.98	V hg	0.500 1195.4	----- 1195.4	0.5124 1204.4	0.5873 1260.1	0.6294 1290.9	0.6491 1305.1	0.6863 1332.1	0.7215 1357.5	0.7716 1393.9	0.8506 1451.8	0.9262 1508.1
950.0	935.3	538.42	V hg	0.471 1193.7	----- 1193.7	0.4740 1195.5	0.5489 1254.6	0.5901 1286.4	0.6092 1301.1	0.6453 1328.7	0.6793 1354.7	0.7275 1391.6	0.8031 1450.0	0.8753 1506.6
1000.0	985.3	544.61	V hg	0.445 1191.8	----- 1191.8	----- -----	0.5140 1248.8	0.5546 1281.9	0.5733 1297.0	0.6084 1325.3	0.6413 1351.7	0.6878 1389.2	0.7604 1448.2	0.8294 1505.1

* V = specific volume, cubic feet per pound

hg = total heat of steam, Btu per pound

Table 7

Properties of Steam (continued)

US Customary Units

Saturated					Superheated : Total Temperature - °F										
Abs. P'	Gauge P	Sat. Temp.	*	Sat	660	700	740	760	780	800	860	900	1000	1100	1200
1100.0	1085.3	556.31	V hg	0.4001 1187.8	0.5110 1288.5	0.5445 1318.3	0.5755 1345.8	0.5904 1358.9	0.6049 1371.7	0.6191 1384.3	0.6601 1420.8	0.6866 1444.5	0.7503 1502.2	0.8117 1558.8	0.8716 1615.2
1200.0	1185.3	567.22	V hg	0.3619 1183.4	0.4586 1279.6	0.4909 1311.0	0.5206 1339.6	0.5347 1353.2	0.5484 1366.4	0.5617 1379.3	0.6003 1416.7	0.6250 1440.7	0.6843 1499.2	0.7412 1556.4	0.7967 1613.1
1300.0	1285.3	577.46	V hg	0.3293 1178.6	0.4139 1270.2	0.4454 1303.4	0.4739 1333.3	0.4874 1347.3	0.5004 1361.0	0.5131 1374.3	0.5496 1412.5	0.5728 1437.0	0.6284 1496.2	0.6816 1553.9	0.7333 1611.0
1400.0	1385.3	587.10	V hg	0.3012 1173.4	0.3753 1260.3	0.4062 1295.5	0.4338 1326.7	0.4468 1341.3	0.4593 1355.4	0.4714 1369.1	0.5061 1408.2	0.5281 1433.1	0.5805 1493.2	0.6305 1551.4	0.6789 1608.9
1500.0	1485.3	596.23	V hg	0.2765 1167.9	0.3413 1249.8	0.3719 1287.2	0.3989 1320.0	0.4114 1335.2	0.4235 1349.7	0.4352 1363.8	0.4684 1403.9	0.4893 1429.3	0.5390 1490.1	0.5862 1548.9	0.6318 1606.8
1600.0	1585.3	604.90	V hg	0.2548 1162.1	0.3112 1238.7	0.3417 1278.7	0.3682 1313.0	0.3804 1328.8	0.3921 1343.9	0.4034 1358.4	0.4353 1399.5	0.4553 1425.3	0.5027 1487.0	0.5474 1546.4	0.5906 1604.6
1700.0	1685.3	613.15	V hg	0.2354 1155.9	0.2842 1226.8	0.3148 1269.7	0.3410 1305.8	0.3529 1322.3	0.3643 1337.9	0.3753 1352.9	0.4061 1395.0	0.4253 1421.4	0.4706 1484.0	0.5132 1543.8	0.5542 1602.5
1800.0	1785.3	621.03	V hg	0.2179 1149.4	0.2597 1214.0	0.2907 1260.3	0.3166 1298.4	0.3284 1315.5	0.3395 1331.8	0.3502 1347.2	0.3801 1390.4	0.3986 1417.4	0.4421 1480.8	0.4828 1541.3	0.5218 1600.4
1900.0	1885.3	628.58	V hg	0.2021 1142.4	0.2371 1200.2	0.2688 1250.4	0.2947 1290.6	0.3063 1308.6	0.3173 1325.4	0.3277 1341.5	0.3568 1385.8	0.3747 1413.3	0.4165 1477.7	0.4556 1538.8	0.4929 1598.2
2000.0	1985.3	635.82	V hg	0.1878 1135.1	0.2161 1184.9	0.2489 1240.0	0.2748 1282.6	0.2863 1301.4	0.2972 1319.0	0.3074 1335.5	0.3358 1381.2	0.3532 1409.2	0.3985 1474.5	0.4311 1536.2	0.4668 1596.1
2100.0	2085.3	642.77	V hg	0.1746 1127.4	0.1962 1167.7	0.2306 1229.0	0.2567 1274.3	0.2682 1294.0	0.2789 1312.3	0.2890 1329.5	0.3167 1376.4	0.3337 1405.0	0.3727 1471.4	0.4089 1533.6	0.4433 1593.9
2200.0	2185.3	649.46	V hg	0.1625 1119.2	0.1768 1147.8	0.2135 1217.4	0.2400 1265.7	0.2514 1286.3	0.2621 1305.4	0.2721 1323.3	0.2994 1371.5	0.3159 1400.8	0.3538 1468.2	0.3887 1531.1	0.4218 1591.8
2300.0	2285.3	655.91	V hg	0.1513 1110.4	0.1575 1123.8	0.1978 1204.9	0.2247 1256.7	0.2362 1278.4	0.2468 1298.4	0.2567 1316.9	0.2835 1366.6	0.2997 1396.5	0.3365 1464.9	0.3703 1528.5	0.4023 1589.6
2400.0	2385.3	662.12	V hg	0.1407 1101.1	----- 1101.1	0.1828 1191.5	0.2105 1247.3	0.2221 1270.2	0.2327 1291.1	0.2425 1310.3	0.2689 1361.6	0.2848 1392.2	0.3207 1461.7	0.3534 1525.9	0.3843 1587.4
2500.0	2485.3	668.13	V hg	0.1307 1091.1	----- 1091.1	0.1686 1176.8	0.1973 1237.6	0.2090 1261.8	0.2196 1283.6	0.2294 1303.6	0.2555 1356.5	0.2710 1387.8	0.3061 1458.4	0.3379 1523.2	0.3678 1585.3
2600.0	2585.3	673.94	V hg	0.1213 1080.2	----- 1080.2	0.1549 1160.6	0.1849 1227.3	0.1967 1252.9	0.2074 1275.8	0.2172 1296.8	0.2431 1351.4	0.2584 1383.4	0.2926 1455.1	0.3236 1520.6	0.3526 1583.1
2700.0	2685.3	679.55	V hg	0.1123 1068.3	----- 1068.3	0.1415 1142.5	0.1732 1216.5	0.1853 1243.8	0.1960 1267.9	0.2059 1289.7	0.2315 1346.1	0.2466 1378.9	0.2801 1451.8	0.3103 1518.0	0.3385 1580.9
2800.0	2785.3	684.99	V hg	0.1035 1054.8	----- 1054.8	0.1281 1121.4	0.1622 1205.1	0.1745 1234.2	0.1854 1259.6	0.1953 1282.4	0.2208 1340.8	0.2356 1374.3	0.2685 1448.5	0.2979 1515.4	0.3254 1578.7
2900.0	2885.3	690.26	V hg	0.0947 1039.0	----- 1039.0	0.1143 1095.9	0.1517 1193.0	0.1644 1224.3	0.1754 1251.1	0.1853 1274.9	0.2108 1335.3	0.2254 1369.7	0.2577 1445.1	0.2864 1512.7	0.3132 1576.5
3000.0	2985.3	695.36	V hg	0.0858 1020.8	----- 1020.8	0.0984 1060.7	0.1416 1180.1	0.1548 1213.8	0.1660 1242.2	0.1760 1267.2	0.2014 1329.7	0.2159 1365.0	0.2476 1441.8	0.2757 1510.0	0.3018 1574.3
3100.0	3085.3	700.31	V hg	0.0753 993.1	----- 993.1	----- -----	0.1320 1166.2	0.1456 1202.9	0.1571 1233.0	0.1672 1259.3	0.1926 1324.1	0.2070 1360.3	0.2382 1438.4	0.2657 1507.4	0.2911 1572.1
3200.0	3185.3	705.11	V hg	0.0580 934.4	----- 934.4	----- -----	0.1226 1151.1	0.1369 1191.4	0.1486 1223.5	0.1589 1251.1	0.1843 1318.3	0.1986 1355.5	0.2293 1434.9	0.2563 1504.7	0.2811 1569.9
3206.0	3191.2	705.40	V	0.0503	-----	-----	0.1220	0.1363	0.1480	0.1583	0.1838	0.1981	0.2288	0.2557	0.2806

* V = specific volume, cubic feet per pound
hg = total heat of steam, Btu per pound

Table 7

Properties of Steam

Metric Units

Pressure (bar abs.)	Saturated			Superheated								
	Temperature (°C)	*	Sat.	250°C	300°C	350°C	400°C	450°C	500°C	550°C	600°C	650°C
1	99.63	V h	1.649 2673	2.406 2973	2.638 3073	2.870 3174	3.102 3277	3.334 3381	3.565 3487	3.796 3594	4.027 3703	4.258 3814
1½	111.37	V h	1.159 2691	1.601 2972	1.757 3072	1.912 3173	2.067 3276	2.221 3380	2.376 3486	2.530 3594	2.684 3703	2.838 3814
2	120.33	V h	0.885 2704	1.198 2970	1.316 3071	1.433 3172	1.549 3275	1.665 3380	1.781 3486	1.897 3593	2.013 3703	2.128 3814
3	133.54	V h	0.605 2723	0.796 2967	0.875 3068	0.953 3170	1.031 3274	1.109 3378	1.186 3485	1.264 3592	1.341 3702	1.418 3813
4	143.63	V h	0.462 2736	1.594 2963	0.654 3065	0.713 3168	0.772 3272	0.830 3377	0.889 3483	0.947 3591	1.005 3701	1.063 3812
5	151.85	V h	0.374 2746	0.474 2960	0.522 3063	0.569 3166	0.617 3270	0.664 3376	0.710 3482	0.757 3590	0.803 3700	0.850 3812
6	158.84	V h	0.315 2755	0.393 2957	0.434 3060	0.474 3164	0.513 3269	0.552 3374	0.591 3481	0.630 3590	0.669 3699	0.708 3811
7	164.96	V h	0.272 2762	0.336 2953	0.371 3058	0.405 3162	0.439 3267	0.473 3373	0.506 3480	0.540 3589	0.573 3699	0.606 3810
8	170.41	V h	0.240 2768	0.293 2950	0.323 3055	0.354 3160	0.384 3265	0.413 3372	0.443 3479	0.472 3588	0.501 3698	0.530 3809
9	175.36	V h	0.214 2773	0.259 2946	0.287 3053	0.314 3158	0.340 3264	0.367 3370	0.393 3478	0.419 3587	0.445 3697	0.471 3809
10	179.88	V h	0.194 2777	0.232 2943	0.257 3050	0.282 3156	0.306 3262	0.330 3369	0.353 3477	0.377 3586	0.400 3696	0.424 3808
11	184.06	V h	0.177 2781	0.210 2939	0.233 3047	0.256 3154	0.278 3261	0.299 3368	0.321 3476	0.342 3585	0.364 3695	0.385 3807
12	187.96	V h	0.163 2784	0.192 2936	0.213 3045	0.234 3152	0.254 3259	0.274 3366	0.294 3475	0.314 3584	0.333 3695	0.353 3807
13	191.60	V h	0.151 2787	0.176 2932	0.196 3042	0.215 3150	0.234 3257	0.253 3365	0.271 3473	0.289 3583	0.307 3694	0.326 3806
14	195.04	V h	0.140 2790	0.163 2928	0.182 3039	0.200 3148	0.217 3256	0.234 3364	0.252 3472	0.268 3582	0.285 3693	0.302 3805
15	198.28	V h	0.131 2792	0.152 2925	0.169 3037	0.186 3146	0.202 3254	0.219 3362	0.235 3471	0.250 3581	0.266 3692	0.282 3805
16	201.37	V h	0.123 2794	0.141 2921	0.158 3034	0.174 3144	0.189 3252	0.205 3361	0.220 3470	0.235 3580	0.249 3691	0.264 3804
17	204.30	V h	0.116 2796	0.133 3917	0.148 3031	0.163 3142	0.178 3251	0.192 3360	0.207 3469	0.221 3579	0.235 3691	0.248 3803
18	207.11	V h	0.110 2798	0.125 2913	0.140 3029	0.154 3140	0.168 3249	0.181 3358	0.195 3468	0.208 3578	0.221 3690	0.235 3803
19	209.79	V h	0.104 2799	0.117 2909	0.132 3026	0.146 3138	0.159 3247	0.172 3357	0.184 3467	0.197 3577	0.210 3689	0.222 3802
20	212.37	V h	0.099 2800	0.111 2905	0.125 3023	0.138 3135	0.151 3246	0.163 3356	0.175 3466	0.187 3576	0.199 3688	0.211 3801
22	217.24	V h	0.090 2802	0.100 2897	0.113 3018	0.125 3131	0.136 3242	0.148 3353	0.159 3463	0.170 3575	0.181 3687	0.191 3800
24	221.78	V h	0.083 2803	0.091 2888	0.103 3012	0.114 3127	0.125 3239	0.135 3350	0.145 3461	0.155 3573	0.165 3685	0.175 3798

*v = specific volume (m³/kg)

h = enthalpy (kJ/kg)

Table 7

Properties of Steam (continued)

Metric Units

Pressure (bar abs.)	Saturated			Superheated								
	Temperature (°C)	*	Sat.	250°C	300°C	350°C	400°C	450°C	500°C	550°C	600°C	650°C
26	226.04	V h	0.076 2804	0.083 2879	0.094 3006	0.105 3123	0.115 3236	0.124 3348	0.134 3459	0.143 3571	0.153 3683	0.162 3797
28	230.04	V h	0.071 2805	0.076 2869	0.087 3000	0.097 3199	0.106 3232	0.115 3345	0.124 3457	0.133 3569	0.141 3682	0.150 3796
30	233.84	V h	0.066 2805	0.070 2859	0.081 2994	0.090 3114	0.099 3229	0.107 3342	0.116 3455	0.124 3567	0.132 3680	0.140 3794
32	237.44	V h	0.062 2805	0.065 2848	0.075 2998	0.084 3110	0.092 3226	0.100 3339	0.108 3452	0.116 3565	0.123 3679	0.131 3793
34	240.88	V h	0.058 2805	0.060 2837	0.070 2982	0.079 3106	0.087 3222	0.094 3337	0.102 3450	0.109 3563	0.116 3677	0.123 3792
36	244.16	V h	0.055 2804	0.056 2826	0.066 2976	0.074 3101	0.081 3219	0.089 3334	0.096 3448	0.103 3561	0.109 3675	0.116 3790
38	247.31	V h	0.052 2803	0.053 2813	0.062 2969	0.070 3097	0.077 3216	0.084 3331	0.091 3446	0.097 3560	0.104 3674	0.110 3789
40	250.33	V h	0.049 2802	----- -----	0.058 2962	0.066 3092	0.073 3212	0.079 3329	0.086 3443	0.092 3558	0.098 3672	0.104 3787
42	253.24	V h	0.047 2801	----- -----	0.055 2956	0.063 3088	0.069 3209	0.075 3326	0.082 3441	0.088 3556	0.093 3671	0.099 3786
44	256.05	V h	0.045 2799	----- -----	0.052 2949	0.059 3083	0.066 3205	0.072 3323	0.078 3439	0.083 3554	0.089 3669	0.095 3785
46	258.76	V h	0.042 2798	----- -----	0.050 2941	0.057 3078	0.063 3202	0.069 3320	0.074 3437	0.080 3552	0.085 3667	0.090 3783
48	261.38	V h	0.041 2796	----- -----	0.047 3934	0.054 3073	0.060 3198	0.066 3318	0.071 3434	0.076 3550	0.081 3666	0.087 3782
50	263.92	V h	0.039 2794	----- -----	0.045 2926	0.051 3069	0.057 3195	0.063 3315	0.068 3432	0.073 3548	0.078 3664	0.083 3781
52	266.38	V h	0.037 2792	----- -----	0.043 2919	0.049 3064	0.055 3191	0.060 3312	0.065 3430	0.070 3546	0.075 3663	0.080 3779
54	268.77	V h	0.036 2790	----- -----	0.041 2911	0.047 3059	0.053 3188	0.058 3309	0.063 3428	0.067 3545	0.072 3661	0.077 3778
56	271.09	V h	0.034 2788	----- -----	0.039 2902	0.045 3054	0.051 3184	0.056 3306	0.060 3425	0.065 3543	0.069 3659	0.074 3776
58	273.36	V h	0.033 2786	----- -----	0.037 2894	0.043 3049	0.049 3181	0.054 3304	0.058 3423	0.063 3541	0.067 3658	0.071 3775
60	275.56	V h	0.032 2783	----- -----	0.036 2885	0.042 3044	0.047 3177	0.052 3301	0.056 3421	0.060 3539	0.065 3656	0.069 3774
65	280.83	V h	0.029 2777	----- -----	0.032 2862	0.038 3031	0.043 3168	0.047 3294	0.052 3415	0.056 3534	0.059 3652	0.063 3770
70	285.80	V h	0.027 2771	----- -----	0.029 2838	0.035 3017	0.039 3158	0.044 3287	0.048 3409	0.051 3529	0.055 3648	0.059 3767
75	290.51	V h	0.025 2764	----- -----	0.026 2812	0.032 3003	0.036 3149	0.040 3279	0.044 3404	0.048 3525	0.051 3644	0.055 3763
80	294.98	V h	0.023 2756	----- -----	0.024 2783	0.029 2988	0.034 3139	0.038 3272	0.041 3398	0.045 3520	0.048 3640	0.051 3760
85	299.24	V h	0.021 2749	----- -----	----- -----	0.027 2972	0.032 3129	0.035 3265	0.039 3392	0.042 3515	0.045 3636	0.048 3757

* v = specific volume (m³/kg)
h = enthalpy (kJ/kg)

Table 7

Properties of Steam (continued)

Metric Units

Pressure (bar abs.)	Saturated			Superheated								
	Temperature (°C)	*	Sat.	250 °C	300 °C	350 °C	400 °C	450 °C	500 °C	550 °C	600 °C	650 °C
90	303.31	V h	0.020 2741	----- -----	----- -----	0.025 2956	0.029 3119	0.033 3257	0.036 3386	0.039 3510	0.042 3632	0.045 3753
95	307.22	V h	0.019 2733	----- -----	----- -----	0.024 2939	0.028 3108	0.031 3250	0.034 3380	0.037 3505	0.040 3628	0.043 3750
100	310.96	V h	0.018 2725	----- -----	----- -----	0.022 2922	0.026 3098	0.029 3242	0.032 3374	0.035 3501	0.038 3624	0.040 3746
105	314.57	V h	0.017 2717	----- -----	----- -----	0.020 2904	0.024 3087	0.028 3235	0.031 3368	0.033 3496	0.036 3620	0.038 3743
110	318.04	V h	0.016 2708	----- -----	----- -----	0.019 2884	0.023 3076	0.026 3227	0.029 3362	0.032 3491	0.034 3616	0.037 3739
115	321.40	V h	0.015 2698	----- -----	----- -----	0.018 2864	0.022 3064	0.025 3219	0.028 3356	0.030 3486	0.033 3612	0.035 3736
120	324.64	V h	0.014 2687	----- -----	----- -----	0.017 2844	0.021 3052	0.024 3211	0.026 3350	0.029 3481	0.031 3608	0.033 3732
125	327.77	V h	0.0135 2675	----- -----	----- -----	0.016 2822	0.020 3040	0.023 3203	0.025 3344	0.027 3476	0.030 3604	0.032 3729
130	330.81	V h	0.0127 2663	----- -----	----- -----	0.015 2799	0.019 3028	0.022 3194	0.024 3338	0.026 3471	0.029 3600	0.031 3725
135	333.76	V h	0.0121 2651	----- -----	----- -----	0.014 2776	0.018 3015	0.021 3186	0.023 3332	0.025 3466	0.027 3596	0.029 3722
140	336.63	V h	0.0114 2637	----- -----	----- -----	0.013 2749	0.017 3002	0.020 3177	0.022 3325	0.024 3462	0.026 3592	0.028 3719
145	339.41	V h	0.0109 2624	----- -----	----- -----	0.012 2722	0.016 2988	0.019 3169	0.021 3319	0.023 3457	0.025 3587	0.027 3715
150	342.12	V h	0.0103 2610	----- -----	----- -----	0.011 2690	0.015 2974	0.018 3160	0.020 3313	0.022 3451	0.024 3583	0.026 3712
155	344.75	V h	0.0098 2596	----- -----	----- -----	0.010 2654	0.014 2960	0.017 3151	0.020 3306	0.022 3446	0.024 3579	0.025 3708
160	347.32	V h	0.0093 2581	----- -----	----- -----	0.009 2614	0.014 2946	0.017 3142	0.019 3300	0.021 3441	0.023 3575	0.024 3705
165	349.82	V h	0.0088 2565	----- -----	----- -----	0.008 2565	0.013 2931	0.016 3133	0.018 3293	0.020 3436	0.022 3571	0.024 3701
170	352.29	V h	0.0083 2547	----- -----	----- -----	----- -----	0.013 2915	0.015 3123	0.018 3287	0.019 3431	0.021 3567	0.023 3698
175	354.64	V h	0.0079 2530	----- -----	----- -----	----- -----	0.012 2899	0.015 3114	0.017 3280	0.019 3426	0.021 3563	0.022 3694
180	356.96	V h	0.0075 2511	----- -----	----- -----	----- -----	0.011 2883	0.014 3104	0.016 3273	0.018 3421	0.020 3558	0.021 3691
190	361.44	V h	0.0067 2468	----- -----	----- -----	----- -----	0.010 2850	0.013 3084	0.015 3259	0.017 3410	0.019 3550	0.020 3684
200	365.71	V h	0.0059 2416	----- -----	----- -----	----- -----	0.009 2815	0.012 3064	0.014 3245	0.016 3400	0.018 3542	0.019 3677
210	369.79	V h	0.0050 2344	----- -----	----- -----	----- -----	0.009 2779	0.011 3042	0.013 3231	0.015 3389	0.017 3533	0.018 3670
220	373.70	V h	0.0038 2218	----- -----	----- -----	----- -----	0.008 2737	0.011 3020	0.013 3216	0.014 3378	0.016 3524	0.017 3662

* v = specific volume (m³/kg)
h = enthalpy (kJ/kg)

Table 7

Temperature Conversion Table

°C		°F	°C		°F
-273	-459.4		43.3	110	230
-268	-450		46.1	115	239
-240	-400		48.9	120	248
-212	-350		54.4	130	266
-184	-300		60.0	140	284
-157	-250	-418	65.6	150	302
-129	-200	-328	71.1	160	320
-101	-150	-238	76.7	170	338
-73	-100	-148	82.2	180	356
-45.6	-50	-58	87.8	190	374
-42.8	-45	-49	93.3	200	392
-40	-40	-40	98.9	210	410
-37.2	-35	-31	104.4	220	428
-34.4	-30	-22	110	230	446
-31.7	-25	-13	115.6	240	464
-28.9	-20	-4	121	250	482
-26.1	-15	5	149	300	572
-23.2	-10	14	177	350	662
-20.6	-5	23	204	400	752
-17.8	0	32	232	450	842
-15	5	41	260	500	932
-12.2	10	50	288	550	1022
-9.4	15	59	316	600	1112
-6.7	20	68	343	650	1202
-3.9	25	77	371	700	1292
-1.1	30	86	399	750	1382
0	32	89.6	427	800	1472
1.7	35	95	454	850	1562
4.4	40	104	482	900	1652
7.2	45	113	510	950	1742
10	50	122	538	1000	1832
12.8	55	131	566	1050	1922
15.6	60	140	593	1100	2012
18.3	65	149	621	1150	2102
21.1	70	158	649	1200	2192
23.9	75	167	677	1250	2282
26.7	80	176	704	1300	2372
29.4	85	185	732	1350	2462
32.2	90	194	762	1400	2552
35	95	203	788	1450	2642
37.8	100	212	816	1500	2732
40.6	105	221			

Note : The temperature to be converted is the figure in the red column. To obtain a reading in °C use the left column ; for conversion to °F use the right column.

Table 8

Masoneilan Control Valve Sizing Formulas

Masoneilan sizing equations have been used for nearly fifty years to determine the capacity requirement of control valves. The most recent version of Masoneilan's sizing equations for liquid and gas/vapor service are presented here as a reference for those who wish to refer to, or continue to use these equations.

For Liquid Service

US Customary Units

A. Subcritical Flow

$$\Delta P < F_L^2 (\Delta P_s)$$

volumetric flow

$$C_v = q \sqrt{\frac{G_f}{\Delta P}}$$

mass flow

$$C_v = \frac{W}{500 \sqrt{G_f \Delta P}}$$

$$* \Delta P_s = P_1 - \left(0.96 - 0.28 \sqrt{\frac{P_v}{P_c}} \right) P_v$$

or for simplicity, if $P_v < 0.5 P_1$, $\Delta P_s = P_1 - P_v$

B. Critical Flow cavitation or flashing

$$\Delta P \geq F_L^2 (\Delta P_s)$$

$$C_v = \frac{q}{F_L} \sqrt{\frac{G_f}{\Delta P_s}}$$

$$C_v = \frac{W}{500 F_L \sqrt{G_f \Delta P_s}}$$

Metric Units

A. Subcritical Flow

$$\Delta P < F_L^2 (\Delta P_s)$$

volumetric flow

$$C_v = 1.16q \sqrt{\frac{G_f}{\Delta P}}$$

mass flow

$$C_v = \frac{1.16 W}{\sqrt{G_f \Delta P}}$$

$$* \Delta P_s = P_1 - \left(0.96 - 0.28 \sqrt{\frac{P_v}{P_c}} \right) P_v$$

or for simplicity, if $P_v < 0.5 P_1$, $\Delta P_s = P_1 - P_v$

B. Critical Flow cavitation or flashing

$$\Delta P \geq F_L^2 (\Delta P_s)$$

$$C_v = \frac{1.16q}{F_L} \sqrt{\frac{G_f}{\Delta P_s}}$$

$$C_v = \frac{1.16 W}{F_L \sqrt{G_f \Delta P_s}}$$

Where:

- C_v = valve flow coefficient
- F_L = critical flow factor
- G_f = specific gravity at flowing temperature (water = 1 @ 60°F)
- P_1 = Upstream pressure, psia
- P_2 = Downstream pressure, psia
- P_c = Pressure at thermodynamic critical point, psia
- P_v = vapor pressure of liquid at flowing temperature, psia
- ΔP = actual pressure drop $P_1 - P_2$, psi
- q = liquid flow rate, U. S. gpm
- W = liquid flow rate, pounds per hour

Where:

- C_v = valve flow coefficient
- F_L = critical flow factor
- G_f = specific gravity at flowing temperature (water = 1 @ 15°C)
- P_1 = Upstream pressure, bar absolute
- P_2 = Downstream pressure, bar absolute
- P_c = Pressure at thermodynamic critical point, bar absolute
- P_v = vapor pressure of liquid at flowing temperature, bar absolute
- ΔP = actual pressure drop $P_1 - P_2$, bar
- q = liquid flow rate, m³/h
- W = liquid flow rate, 1000 kg per hr

For Gas and Vapor Service

US Customary Units

for gas volumetric flow

$$C_v = \frac{Q \sqrt{GTZ}}{834 F_L P_1 (y - 0.148 y^3)}$$

mass flow

$$C_v = \frac{W \sqrt{Z}}{2.8 F_L P_1 \sqrt{G_f} (y - 0.148 y^3)}$$

for saturated steam

$$C_v = \frac{W}{1.83 F_L P_1 (y - 0.148 y^3)}$$

for superheated steam

$$C_v = \frac{W (1 + 0.0007 T_{sh})}{1.83 F_L P_1 (y - 0.148 y^3)}$$

Metric Units

for gas volumetric flow

$$C_v = \frac{Q \sqrt{GTZ}}{257 F_L P_1 (y - 0.148 y^3)}$$

mass flow

$$C_v = \frac{54.5 W \sqrt{Z}}{F_L P_1 \sqrt{G_f} (y - 0.148 y^3)}$$

for saturated steam

$$C_v = \frac{83.7 W}{F_L P_1 (y - 0.148 y^3)}$$

for superheated steam

$$C_v = \frac{83.7 (1 + 0.00126 T_{sh}) W}{F_L P_1 (y - 0.148 y^3)}$$

Where:

$$y = \frac{1.40}{F_L} \sqrt{\frac{\Delta P}{P_1}}$$

(for 77000, LO-DB® cartridges and expansion plates
and two stage 41000 and 72000)

$$y = \frac{1.63}{F_L} \sqrt{\frac{\Delta P}{P_1}}$$

(for all other valves)
with a maximum value of $y = 1.50$
at this value, $y - 0.148y^3 = 1.0$

Where:

- C_v = valve flow coefficient
- F_L = critical flow factor
- G = gas specific gravity (air = 1.0)
- G_f = specific gravity at flowing temperature
= $G \times \frac{520}{T}$
- P_1 = Upstream pressure, psia
- P_2 = Downstream pressure, psia
- ΔP = actual pressure drop $P_1 - P_2$, psi
- Q = gas flow rate at 14.7 psia and 60°F, scfh
- T = flowing temperature, R
- T_{sh} = steam superheat, °F
- W = flow rate, lbs/hr
- Z = compressibility factor

Where:

- C_v = valve flow coefficient
- F_L = critical flow factor
- G = gas specific gravity (air = 1.0)
- G_f = specific gravity at flowing temperature
= $G \times \frac{288}{T}$
- P_1 = Upstream pressure, bar absolute
- P_2 = Downstream pressure, bar absolute
- ΔP = actual pressure drop $P_1 - P_2$, bar
- Q = gas flow rate at 15°C & 1013 millibar absolute, m³/h
- T = flowing temperature, K
- T_{sh} = steam superheat, °C
- W = flow rate, 1000 kg/hr
- Z = compressibility factor

Metric Conversion Tables

Multiply	By	To Obtain	Multiply	By	To Obtain
Length			Flow Rates		
millimeters	0.10	centimeters	cubic feet/minute	60.0	ft ³ /hr
millimeters	0.001	meters	cubic feet/minute	1.699	m ³ /hr
millimeters	0.039	inches	cubic feet/minute	256.5	Barrels/day
millimeters	0.00328	feet	cubic feet/hr	0.1247	GPM
centimeters	10.0	millimeters	cubic feet/hr	0.472	liters/min
centimeters	0.010	meters	cubic feet/hr	0.01667	ft ³ /min
centimeters	0.394	inches	cubic feet/hr	0.0283	m ³ /hr
centimeters	0.0328	feet	cubic meters/hr	4.403	GPM
inches	25.40	millimeters	cubic meters/hr	16.67	liters/min
inches	2.54	centimeters	cubic meters/hr	0.5886	ft ³ /min
inches	0.0254	meters	cubic meters/hr	35.31	ft ³ /hr
inches	0.0833	feet	cubic meters/hr	150.9	Barrels/day
feet	304.8	millimeters	Velocity		
feet	30.48	centimeters	feet per second	60	ft/min
feet	0.304	meters	feet per second	0.3048	meters/second
feet	12.0	inches	feet per second	1.097	km/hr
Area			feet per second	0.6818	miles/hr
sq. millimeters	0.010	sq. centimeters	meters per second	3.280	ft/sec
sq. millimeters	10. ⁻⁶	sq. meters	meters per second	196.9	ft/min
sq. millimeters	0.00155	sq. inches	meters per second	3.600	km/hr
sq. millimeters	1.076 x 10 ⁻⁵	sq. feet	meters per second	2.237	miles/hr
sq. centimeters	100	sq. millimeters	Weight (Mass)		
sq. centimeters	0.0001	sq. meters	pounds	0.0005	short ton
sq. centimeters	0.155	sq. inches	pounds	0.000446	long ton
sq. centimeters	0.001076	sq. feet	pounds	0.453	kilogram
sq. inches	645.2	sq. millimeters	pounds	0.000453	metric ton
sq. inches	6.452	sq. centimeters	short ton	2000.0	pounds
sq. inches	0.000645	sq. meters	short ton	0.8929	long ton
sq. inches	0.00694	sq. feet	short ton	907.2	kilogram
sq. feet	9.29 x 10 ⁴	sq. millimeters	short ton	0.9072	metric ton
sq. feet	929	sq. centimeters	long ton	2240	pounds
sq. feet	0.0929	sq. meters	long ton	1.120	short ton
sq. feet	144	sq. inches	long ton	1016	kilogram
Flow Rates			long ton	1.016	metric ton
gallons US/minute			kilogram	2.205	pounds
GPM	3.785	liters/min	kilogram	0.0011	short ton
gallons US/minute	0.133	ft ³ /min	kilogram	0.00098	long ton
gallons US/minute	8.021	ft ³ /hr	kilogram	0.001	metric ton
gallons US/minute	0.227	m ³ /hr	metric ton	2205	pounds
gallons US/minute	34.29	Barrels/day	metric ton	1.102	short ton
		(42 US gal)	metric ton	0.984	long ton
cubic feet/minute	7.481	GPM	metric ton	1000	kilogram
cubic feet/minute	28.32	liters/minute			

Some units shown on this page are not recommended by SI, e.g., kilogram/sq. cm should be read as kilogram (force) / sq. cm

Table 9

Multiply	By	To Obtain	Multiply	By	To Obtain
Volume & Capacity			Pressure & Head		
cubic cm	0.06102	cubic inches	atmosphere	14.69	psi
cubic cm	3.531×10^{-5}	cubic feet	atmosphere	1.013	bar
cubic cm	10^{-6}	cubic meters	atmosphere	1.033	Kg/cm ²
cubic cm	0.0001	liters	atmosphere	101.3	kPa
cubic cm	2.642×10^{-4}	gallons (US)	atmosphere	33.9	ft of H ₂ O
cubic meters	10^{-6}	cubic cm	atmosphere	10.33	m of H ₂ O
cubic meters	61,023.0	cubic inches	atmosphere	76.00	cm of Hg
cubic meters	35.31	cubic feet	atmosphere	760.0	torr (mm of Hg)
cubic meters	1000.0	liters	atmosphere	29.92	in of Hg
cubic meters	264.2	gallons	bar	14.50	psi
cubic feet	28,320.0	cubic cm	bar	0.9869	atmosphere
cubic feet	1728.0	cubic inches	bar	1.020	Kg/cm ²
cubic feet	0.0283	cubic meters	bar	100.0	kPa
cubic feet	28.32	liters	bar	33.45	ft of H ₂ O
cubic feet	7.4805	gallons	bar	10.20	m of H ₂ O
liters	1000.0	cubic cm	bar	75.01	cm of Hg
liters	61.02	cubic inches	bar	750.1	torr (mm of Hg)
liters	0.03531	cubic feet	bar	29.53	in of Hg
liters	0.001	cubic meters	kilogram/sq. cm	14.22	psi
liters	0.264	gallons	kilogram/sq. cm	0.9807	bar
gallons	3785.0	cubic cm	kilogram/sq. cm	0.9678	atmosphere
gallons	231.0	cubic inches	kilogram/sq. cm	98.07	kPa
gallons	0.1337	cubic feet	kilogram/sq. cm	32.81	ft of H ₂ O (4 DEG C)
gallons	3.785×10^{-3}	cubic meters	kilogram/sq. cm	10.00	m of H ₂ O (4 DEG C)
gallons	3.785	liters	kilogram/sq. cm	73.56	cm of Hg
Pressure & Head			kilogram/sq. cm	735.6	torr (mm of Hg)
pounds/sq. inch	0.06895	bar	kilogram/sq. cm	28.96	in of Hg
pounds/sq. inch	0.06804	atmosphere	kiloPascal	0.145	psi
pounds/sq. inch	0.0703	kg/cm ²	kiloPascal	0.01	bar
pounds/sq. inch	6.895	kPa	kiloPascal	0.00986	atmosphere
pounds/sq. inch	2.307	ft of H ₂ O (4 DEG C)	kiloPascal	0.0102	kg/cm ²
pounds/sq. inch	0.703	m of H ₂ O (4 DEG C)	kiloPascal	0.334	ft of H ₂ O
pounds/sq. inch	5.171	cm of Hg (0 DEG C)	kiloPascal	0.102	m of H ₂ O
pounds/sq. inch	51.71	torr (mm of Hg)	kiloPascal	0.7501	cm of Hg
		(0 DEG C)	kiloPascal	7.501	torr (mm of Hg)
		in of Hg (0 DEG C)	kiloPascal	0.295	in of Hg
pounds/sq. inch	2.036	in of Hg (0 DEG C)	millibar	0.001	bar

Some units shown on this page are not recommended by SI, e.g., kilogram/sq. cm should be read as kilogram (force) /sq. cm

Table 9

Useful List of Equivalents (U. S. Customary Units)

1 U.S. gallon of water = 8.33 lbs @ std cond.
 1 cubic foot of water = 62.34 lbs @ std cond. (= density)
 1 cubic foot of water = 7.48 gallons
 1 cubic foot of air = 0.076 lbs @ std cond. (= air density)
 Air specific volume = 1/density = 13.1 cubic feet /lb
 Air molecular weight M = 29
 Specific gravity of air G = 1 (reference for gases)
 Specific gravity of water = 1 (reference for liquids)
 Standard conditions (US Customary) are at
 14.69 psia & 60 DEG F*
 G of any gas = density of gas/0.076
 G of any gas = molecular wt of gas/29
 G of gas at flowing temp = $\frac{G \times 520}{T + 460}$

Flow conversion of gas

$$\text{scfh} = \frac{\text{lbs/hr}}{\text{density}}$$

$$\text{scfh} = \frac{\text{lbs/hr} \times 379}{M}$$

$$\text{scfh} = \frac{\text{lbs/hr} \times 13.1}{G}$$

Flow conversion of liquid

$$\text{GPM} = \frac{\text{lbs/hr}}{500 \times G}$$

*Normal conditions (metric) are at 1.013 bar and 0 DEG. C & 4 DEG. C water

Note : Within this control valve handbook, the metric factors are at 1.013 bar and 15.6°C.

Universal gas equation

$Pv = mRTZ$ Where P = press lbs/sq ft v = volume in ft ³ m = mass in lbs R = gas constant $= \frac{1545}{M}$ T = temp Rankine Z = gas compressibility factor = Z	Metric P = Pascal v = m ³ m = kg R = gas constant $= \frac{8314}{M}$ T = temp Kelvin
---	---

Gas expansion $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
(perfect gas)

Velocity of sound C (ft/sec) where T = temp DEG F
 M = mol. wt
 k = specific heat ratio Cp/Cv

$$C = 223 \sqrt{\frac{k (T + 460)}{M}}$$

Velocity of Sound C (m/sec) where T = temp DEG C
 M = mol. wt
 k = specific heat ratio Cp/Cv

$$C = 91.2 \sqrt{\frac{k (T + 273)}{M}}$$

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Notes

Notes

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