

MATERIAL DE CONSULTA

Equação diferencial da continuidade de A:

Coordenadas retangulares:

$$\frac{\partial C_A}{\partial t} + \frac{\partial N_{A,x}}{\partial x} + \frac{\partial N_{A,y}}{\partial y} + \frac{\partial N_{A,z}}{\partial z} = R_A''' \quad \frac{\partial \rho_A}{\partial t} + \frac{\partial n_{A,x}}{\partial x} + \frac{\partial n_{A,y}}{\partial y} + \frac{\partial n_{A,z}}{\partial z} = r_A''''$$

Coordenadas cilíndricas:

$$\frac{\partial C_A}{\partial t} + \left[\frac{1}{r} \frac{\partial(rN_{A,r})}{\partial r} + \frac{1}{r} \frac{\partial N_{A,\theta}}{\partial \theta} + \frac{\partial N_{A,z}}{\partial z} \right] = R_A''' \quad \frac{\partial \rho_A}{\partial t} + \left[\frac{1}{r} \frac{\partial(rn_{A,r})}{\partial r} + \frac{1}{r} \frac{\partial n_{A,\theta}}{\partial \theta} + \frac{\partial n_{A,z}}{\partial z} \right] = r_A''''$$

Coordenadas esféricas:

$$\frac{\partial C_A}{\partial t} + \left[\frac{1}{r^2} \frac{\partial(r^2 N_{A,r})}{\partial r} + \frac{1}{r \sin \theta} \frac{\partial(N_{A,\theta} \sin \theta)}{\partial \theta} + \frac{1}{r \sin \theta} \frac{\partial N_{A,\phi}}{\partial \phi} \right] = R_A'''$$

$$\frac{\partial \rho_A}{\partial t} + \left[\frac{1}{r^2} \frac{\partial(r^2 n_{A,r})}{\partial r} + \frac{1}{r \sin \theta} \frac{\partial(n_{A,\theta} \sin \theta)}{\partial \theta} + \frac{1}{r \sin \theta} \frac{\partial n_{A,\phi}}{\partial \phi} \right] = r_A''''$$

Equação diferencial da continuidade em termos da lei da Fick:

$$\frac{\partial C_A}{\partial t} + \vec{\nabla}(-D_{AB} \vec{\nabla} C_A) + \vec{\nabla} C_A \vec{V} - R_A''' = 0 \quad \frac{\partial \rho_A}{\partial t} + \vec{\nabla}(-D_{AB} \vec{\nabla} \rho_A) + \vec{\nabla} \rho_A \vec{v} - r_A'''' = 0$$

Equação do fluxo global de A

$$\vec{N}_A = -D_{AB} \vec{\nabla} C_A + y_A (\vec{N}_A + \vec{N}_B) \quad \vec{n}_A = -D_{AB} \vec{\nabla} \rho_A + w_A (\vec{n}_A + \vec{n}_B)$$

Constante universal dos gases ideais:

$$R = 8,314 \frac{J}{mol.K} = 8,314 \frac{m^3 \cdot Pa}{mol.K} = 8,314 \cdot 10^{-2} \frac{L \cdot bar}{mol.K} = 82,06 \frac{cm^3 atm}{mol.K} = 62,36 \frac{L \cdot mmHg}{mol.K}$$

Conversões de unidades:

$$1 atm = 1,013 bar = 1,013 \cdot 10^5 Pa = 101,3 kPa = 760 mmHg = 1,013 \frac{kgf}{cm^2} = 14,7 psi$$

$$1 m^3 = 1000 L = 1000 dm^3 = 10^6 cm^3$$