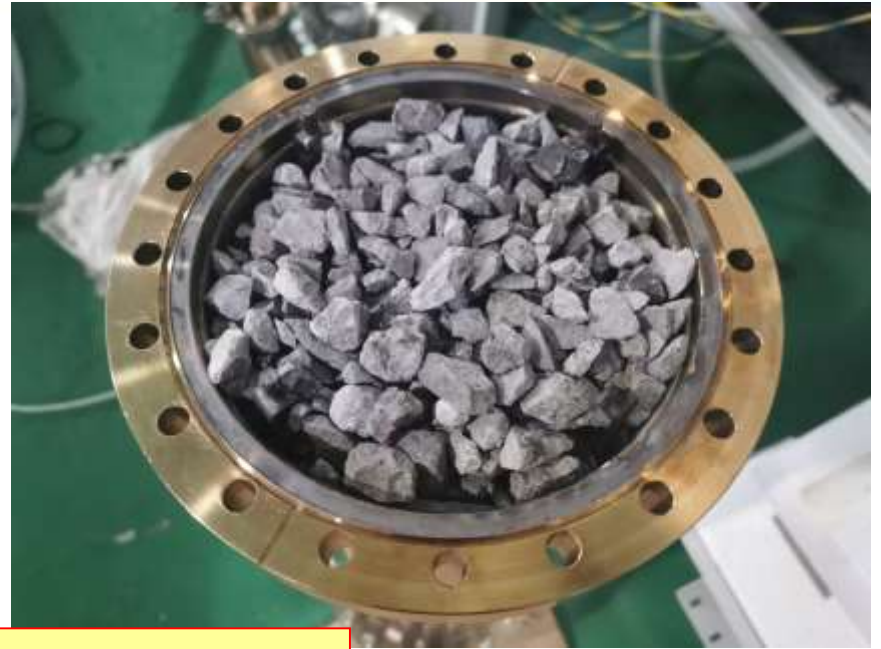




# ALPHA Flux Test



**Zeolites BEFORE & AFTER Flux Test**





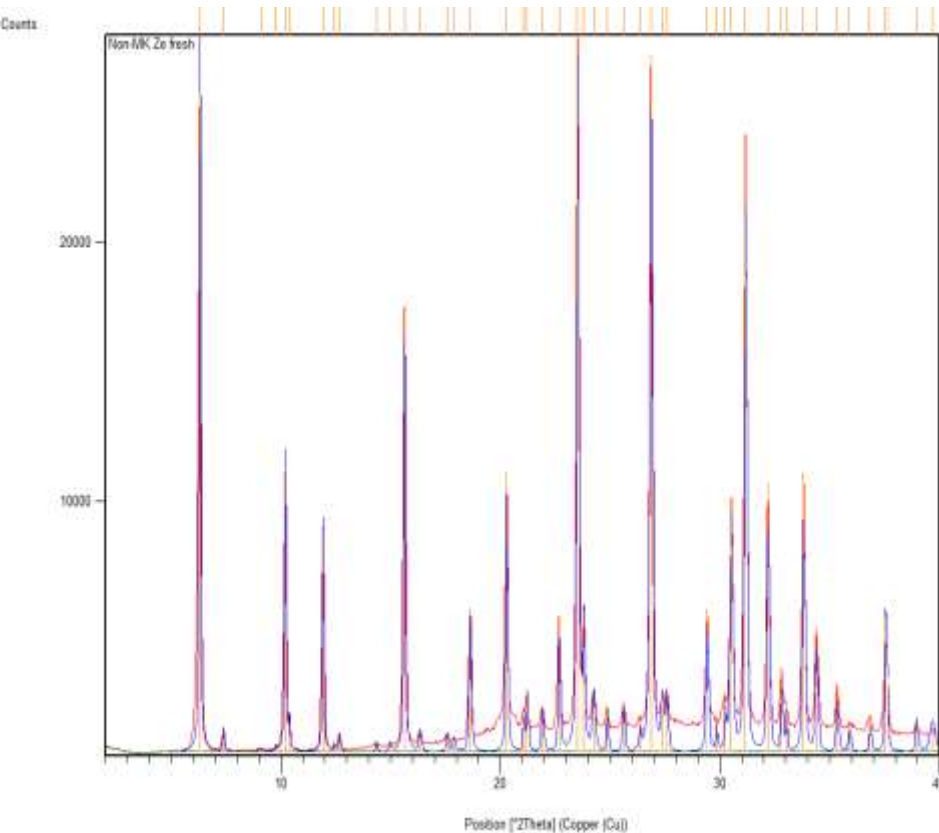
# X13 Zeolite

<p>THE INFORMATION CONTAINED IN THIS DRAWING IS THE SOLE PROPERTY OF &lt;HELLER INDUSTRIES&gt;. ANY REPRODUCTION IN PART OR WHOLE WITHOUT THE WRITTEN PERMISSION OF &lt;HELLER INDUSTRIES&gt; IS PROHIBITED.</p> <h2 style="text-align: center;">Molecular Sieve JZ-ZMS9</h2> <p><b>Description:</b> JZ-ZMS9 is Sodium aluminosilicate, It could absorb the molecular which diameter is not more than 9 angstroms.</p> <p><b>Applications:</b></p> <ol style="list-style-type: none"> <li>Purification of gas in air separation plant, removal of H<sub>2</sub>O, CO<sub>2</sub> and hydrocarbons.</li> <li>Dehydration and desulfurization (removal of H<sub>2</sub>S and mercaptan, etc.) of natural gas, LNG, liquid alkanes (propane, butane, etc.).</li> <li>Deep drying of general gases (e.g. compressed air, permanent gas).</li> <li>Drying and purification of synthetic ammonia.</li> <li>Desulfurization and deodorization of Aerosol.</li> <li>CO<sub>2</sub> removal from pyrolysis gas.</li> </ol> <p><b>Specification:</b></p> <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th>Properties</th> <th>Unit</th> <th colspan="2">sphere</th> <th colspan="2">cylinder</th> </tr> </thead> <tbody> <tr> <td>Diameter</td> <td>mm</td> <td>1.6-2.5</td> <td>3-5</td> <td>1/16"</td> <td>1/8"</td> </tr> <tr> <td>Static Water Adsorption</td> <td>≥%</td> <td>26.5</td> <td>26.5</td> <td>26</td> <td>26</td> </tr> <tr> <td>CO<sub>2</sub> Adsorption</td> <td>≥%</td> <td>18</td> <td>17.5</td> <td>17.5</td> <td>17.5</td> </tr> <tr> <td>Bulk Density</td> <td>≥g/ml</td> <td>0.64</td> <td>0.62</td> <td>0.62</td> <td>0.62</td> </tr> <tr> <td>Crushing Strength</td> <td>≥N/Pc</td> <td>25</td> <td>80</td> <td>25</td> <td>65</td> </tr> <tr> <td>Attrition Rate</td> <td>≤%</td> <td>0.1</td> <td>0.1</td> <td>0.4</td> <td>0.4</td> </tr> <tr> <td>Package Moisture</td> <td>≤%</td> <td>1.5</td> <td>1.5</td> <td>2.0</td> <td>2.0</td> </tr> </tbody> </table> <p><b>Standard Package:</b> sphere: 140kg/steel drum cylinder: 125kg/steel drum</p>	Properties	Unit	sphere		cylinder		Diameter	mm	1.6-2.5	3-5	1/16"	1/8"	Static Water Adsorption	≥%	26.5	26.5	26	26	CO <sub>2</sub> Adsorption	≥%	18	17.5	17.5	17.5	Bulk Density	≥g/ml	0.64	0.62	0.62	0.62	Crushing Strength	≥N/Pc	25	80	25	65	Attrition Rate	≤%	0.1	0.1	0.4	0.4	Package Moisture	≤%	1.5	1.5	2.0	2.0	<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <th colspan="6">REVISIONS LOG</th> </tr> <tr> <th>REV.</th> <th>DATE</th> <th>ECN</th> <th>AUTH</th> <th>DR</th> <th>CK</th> </tr> <tr> <td>-</td> <td>12/2020</td> <td>ECN 200106</td> <td></td> <td>RDONG</td> <td></td> </tr> </table> <div style="text-align: center; margin-top: 20px;"> <span style="font-weight: bold; font-size: 1.2em;">Shanghai Jiuzhou Chemicals Co., Ltd</span> </div> <p style="margin-top: 20px;"><b>NOTE:</b> <b>PROPERTIES: SPHERE</b> <b>DIAMETER : 3-5MM</b></p> <p><b>Attention:</b> The product as desiccant cannot be exposed in the open air and should be stored in dry condition with air-proof package.</p> <hr/> <p style="font-size: 0.8em;">Address: Rm1111, No11, Yujinggang Rd, Jing'an District, Shanghai, China. Tel: +86 21 68769026 Fax: +86 21 68769036 E-mail: <a href="mailto:marketing@jiuzhouchemicals.com">marketing@jiuzhouchemicals.com</a> Website: <a href="http://www.jiuzhouchemicals.com">www.jiuzhouchemicals.com</a></p> <p style="text-align: center; font-weight: bold; font-size: 1.1em; margin-top: 20px;">NO DEVIATIONS, SUBSTITUTIONS, OR EXCEPTIONS</p>	REVISIONS LOG						REV.	DATE	ECN	AUTH	DR	CK	-	12/2020	ECN 200106		RDONG	
Properties	Unit	sphere		cylinder																																																															
Diameter	mm	1.6-2.5	3-5	1/16"	1/8"																																																														
Static Water Adsorption	≥%	26.5	26.5	26	26																																																														
CO <sub>2</sub> Adsorption	≥%	18	17.5	17.5	17.5																																																														
Bulk Density	≥g/ml	0.64	0.62	0.62	0.62																																																														
Crushing Strength	≥N/Pc	25	80	25	65																																																														
Attrition Rate	≤%	0.1	0.1	0.4	0.4																																																														
Package Moisture	≤%	1.5	1.5	2.0	2.0																																																														
REVISIONS LOG																																																																			
REV.	DATE	ECN	AUTH	DR	CK																																																														
-	12/2020	ECN 200106		RDONG																																																															

<p>THIRD ANGLE PROJECTION</p>			<h2 style="margin: 0;">HELLER INDUSTRIES INC</h2>																															
<p><b>DIMENSIONS ARE IN MILLIMETERS</b> UNLESS OTHERWISE SPECIFIED TOLERANCES ARE: (ALL) DECIMAL PLACES: ±0.38 ANGULAR: ±1°</p> <p>BREAK ALL SHARP CORNERS (0.13 - 0.25) TOLERANCE ZONE PROJECTION ± 25.4</p> <p>ALL MACHINED SURFACES </p> <p>DO NOT SCALE DRAWING</p>			<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td>MATERIAL</td> <td colspan="2">SEE NOTES</td> </tr> <tr> <td>FINISH</td> <td colspan="2"></td> </tr> <tr> <td>PROJECT/MFG No:</td> <td colspan="2"></td> </tr> <tr> <td>SCALE</td> <td colspan="2"></td> </tr> </table>	MATERIAL	SEE NOTES		FINISH			PROJECT/MFG No:			SCALE			<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td colspan="3">DESCRIPTION/TITLE</td> </tr> <tr> <td colspan="3">JZ-ZMS9 ZEOLITE, 13X GRADE</td> </tr> <tr> <td>DRAWN BY</td> <td>RDONG</td> <td>DATE 12/19/2020</td> </tr> <tr> <td>APPROVED BY</td> <td>RAVIT</td> <td>DATE 2/24/2021</td> </tr> <tr> <td>SIZE</td> <td colspan="2"> <div style="font-size: 1.5em; font-weight: bold; text-align: center;">A</div> <div style="font-size: 1.5em; font-weight: bold; text-align: center;">2210186</div> </td> </tr> <tr> <td>LAST SAVED BY:</td> <td colspan="2">rpotineni 3/31/2021</td> </tr> </table>	DESCRIPTION/TITLE			JZ-ZMS9 ZEOLITE, 13X GRADE			DRAWN BY	RDONG	DATE 12/19/2020	APPROVED BY	RAVIT	DATE 2/24/2021	SIZE	<div style="font-size: 1.5em; font-weight: bold; text-align: center;">A</div> <div style="font-size: 1.5em; font-weight: bold; text-align: center;">2210186</div>		LAST SAVED BY:	rpotineni 3/31/2021	
MATERIAL	SEE NOTES																																	
FINISH																																		
PROJECT/MFG No:																																		
SCALE																																		
DESCRIPTION/TITLE																																		
JZ-ZMS9 ZEOLITE, 13X GRADE																																		
DRAWN BY	RDONG	DATE 12/19/2020																																
APPROVED BY	RAVIT	DATE 2/24/2021																																
SIZE	<div style="font-size: 1.5em; font-weight: bold; text-align: center;">A</div> <div style="font-size: 1.5em; font-weight: bold; text-align: center;">2210186</div>																																	
LAST SAVED BY:	rpotineni 3/31/2021																																	
QUANTITY:																																		
SO. No:																																		
COMP. DATE:	NEXT ASSY	QTY																																



## XRD – Fresh Zeolite – X13



Candidates:				
No.	Ref. Code	Score	Compound Name	Chemical Formula
1	00-005-0586	22	Calcite, syn	CaCO <sub>3</sub>
2	00-033-1161	15	silica	SiO <sub>2</sub>
3	00-046-1045	15	Quartz, syn	SiO <sub>2</sub>
4	00-010-0173	12	alumina	Al <sub>2</sub> O <sub>3</sub>
5	00-024-0072	7	Hematite	Fe <sub>2</sub> O <sub>3</sub>
6	01-073-1667	6	Bornite	Cu <sub>5</sub> FeS <sub>4</sub>
7	00-006-0329	4	Praseodymium Oxi...	Pr <sub>2</sub> O <sub>3</sub>
8	01-075-0134	3	Uranium Oxide	UO <sub>2</sub>
9	00-005-0628	1	Halite, syn	NaCl
10	01-077-2041	1	Sodium Erbium Flu...	NaErF <sub>4</sub>

Zeolites are microporous, aluminosilicate minerals commonly used as commercial adsorbents and catalysts.

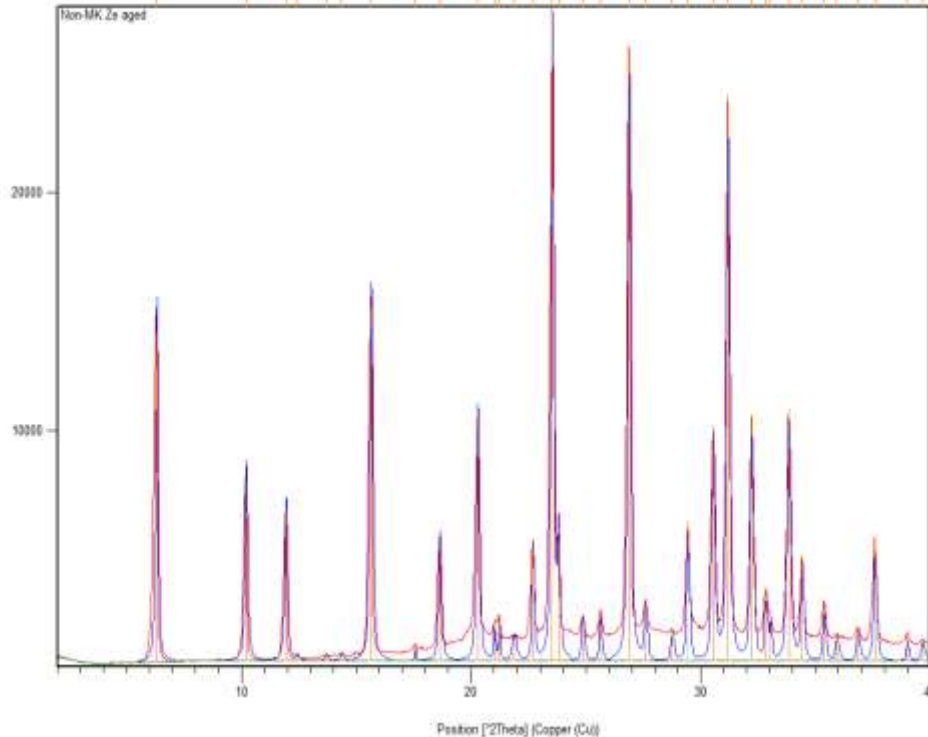
- Presence of Calcite is considered significant
- Zeolite extrudate size ~4mm
- Si –Al ratio ~3 to 1
- Pore size unknown





## XRD – Aged Zeolite – X13 time in service - unknown

Counts



Selected Candidate: 00-005-0586

No.	Ref. Code	Score	Compound Name	Chemical Formula
1	00-005-0586	28	Calcite, syn	Ca C O3
2	00-033-1161	17	silica	Si O2
3	00-046-1045	17	Quartz, syn	Si O2
4	00-010-0173	16	alumina	Al2 O3
5	00-024-0072	13	Hematite	Fe2 O3
6	00-024-1977	13	Sucrose	C12 H22 O11
7	01-073-1667	7	Bornite	Cu5 Fe S4
8	00-033-0664	7	burnt ochre	Fe2 O3
9	00-006-0329	4	Praseodymium Oxide	Pr2 O3
10	01-075-0134	4	Uranium Oxide	U O2
11	01-077-2041	1	Sodium Erbium Flu...	Na Er F4
12	00-005-0628	1	Halite, syn	Na Cl

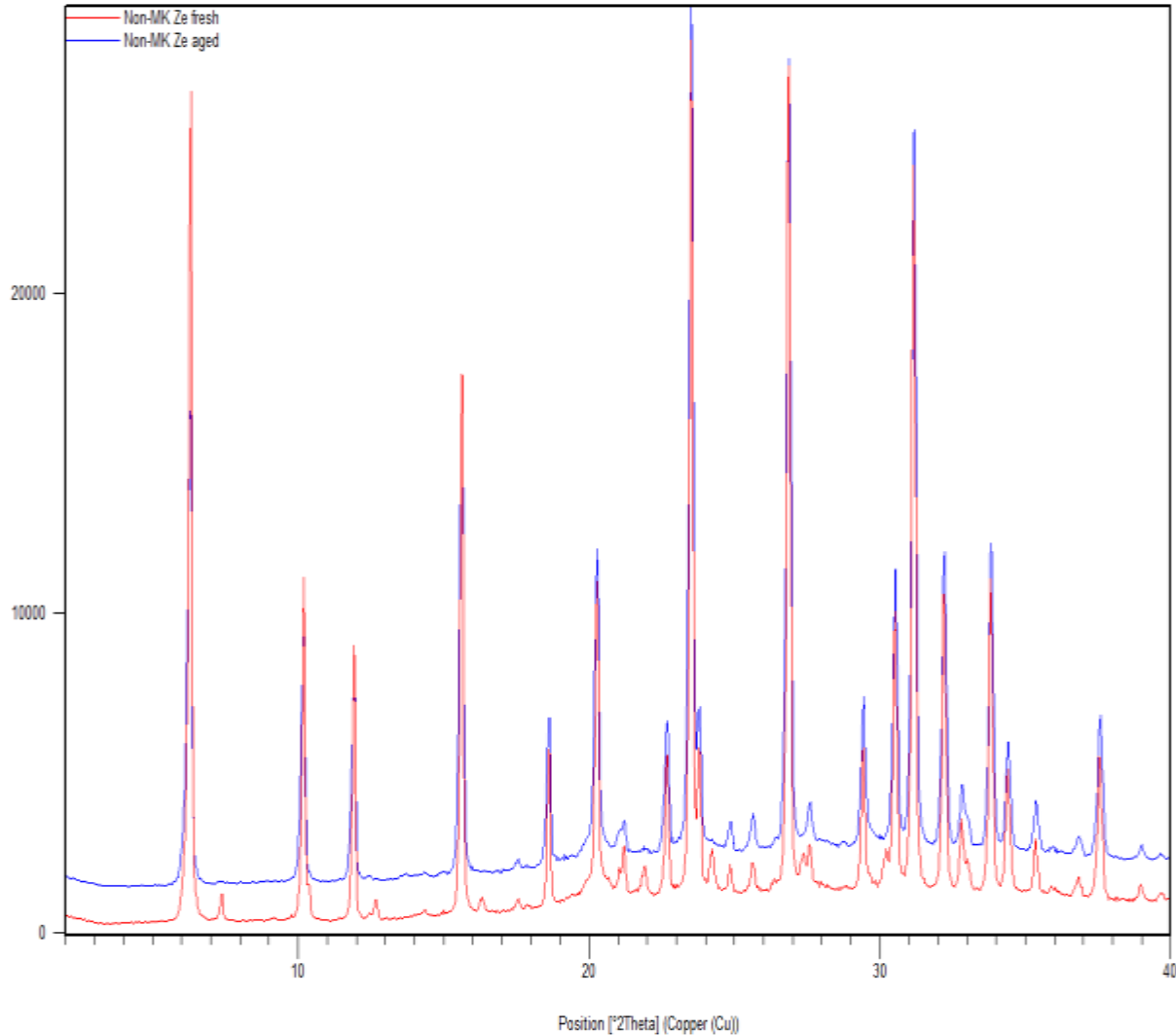
Sucrose, a disaccharide, (C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>), from monomers. glucose and fructose likely originated in a condensation reaction from flux and thixotropic agents. Aged zeolite is 6.0mm in diameter, having adsorbed CO<sub>2</sub> and water.

Burnt Ochre is a family of earth pigments, which includes yellow ochre, red ochre, purple ochre, sienna, and umber. The major ingredient of all the ochres is iron(III) oxide-hydroxide, known as limonite, which gives them a yellow color. Yellow ochre, FeO(OH)·nH<sub>2</sub>O, is a hydrated iron hydroxide (limonite) also called gold ochre.



## XRD comparative spectral overlap – X13 Zeolite – Fresh vs Aged

Counts

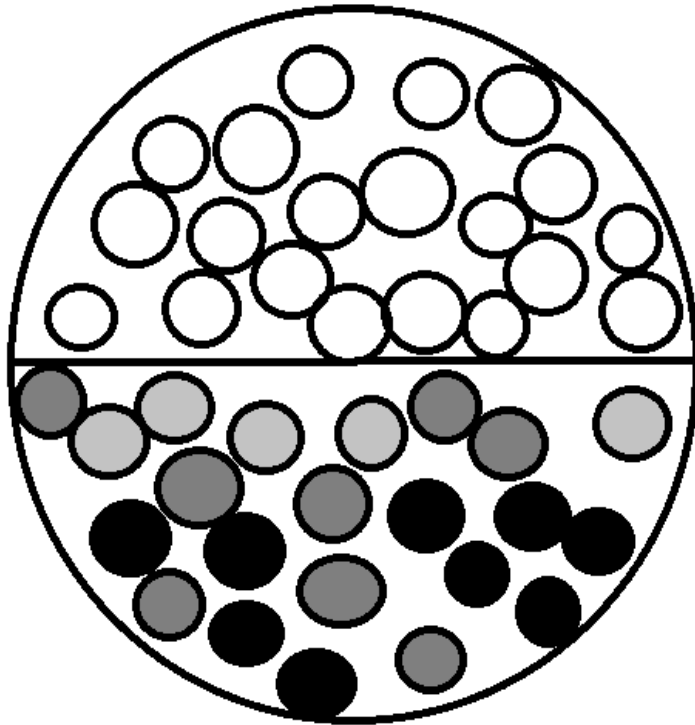


Aged and fresh samples with near identical structures, same zeolite

Structural features observed in the fresh sample, no longer evident in the aged sample



## X13 Zeolite Bed Contaminant Profile



Mal-distribution of contaminants reportedly reside at the bottom region of the incoming zeolite bed

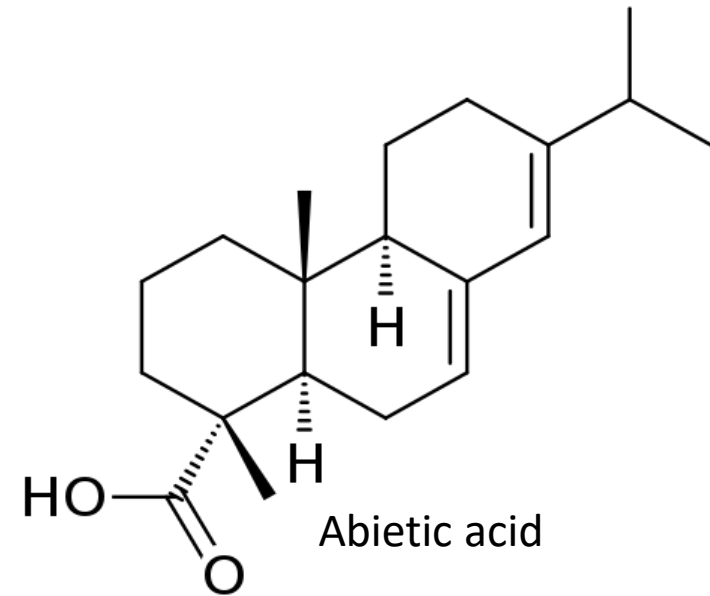
Suggesting non uniform flow and ineffective use of zeolite bed

Inlet flow straightener, eg, wire mesh may mitigate problem. Wire mesh matrix segments could improve zeolite life and performance

Irrespective of matrix design, inlet flow straightening may be necessary in a traditional monolith, metal, ceramic, extrudate



## Flux composition



Rosin (resin) content is upward of 50-60% of a typical flux compound.

Abietic acid is the primary carboxylic acid in flux that serves to remove oxide formation from surfaces.

Rheological modifiers

Solvent – linear glycol

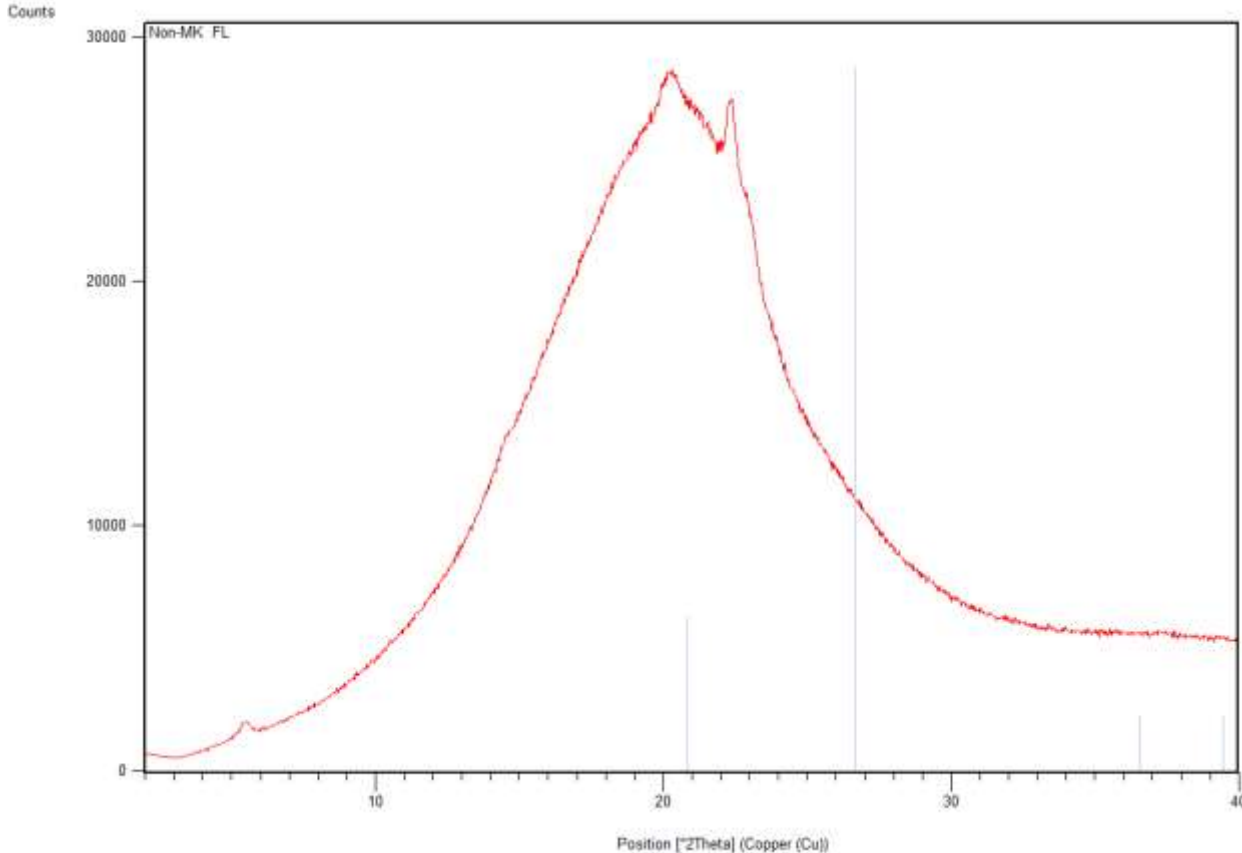
Small molecules

Pyrolysis or devolatilization of flux occurs at elevated temperatures in inert or (O<sub>2</sub>) deprived environment.

In general, pyrolysis of organic substances produces volatile products and leaves a carbon-rich solid residue. Pyrolysis, which leaves carbon compounds as the primary residue, carbonization. Pyrolysis is considered the first step in the processes of gasification or combustion.



## XRD – Flux compound - Fresh



Candidates:				
No.	Ref. Code	Score	Compound Name	Chemical Formula
1	00-033-1161	4	silica	Si O <sub>2</sub>
2	00-046-1045	3	Quartz, syn	Si O <sub>2</sub>
3	00-005-0586	2	Calcite, syn	Ca C O <sub>3</sub>

Sample is largely amorphous suggesting high surface area and high surface energy

Highly viscous

Low crystallinity

Low or non-existent porosity

Flux composition:

- Small molecule
- Rheological Additives
- Abietic acid
- Rosin/resin

CaCO<sub>3</sub> and SiO<sub>2</sub> binders  
Disaccharide formed in  
the furnace





# PSD-FLUX PSD by Microtrac FlowSync with nonpolar solvent Isopar-G Analyzed following 60s sonication

Summary Data   Tabular Data   Notes/Warnings   Raw Data   Setzero   Size/Shape Data

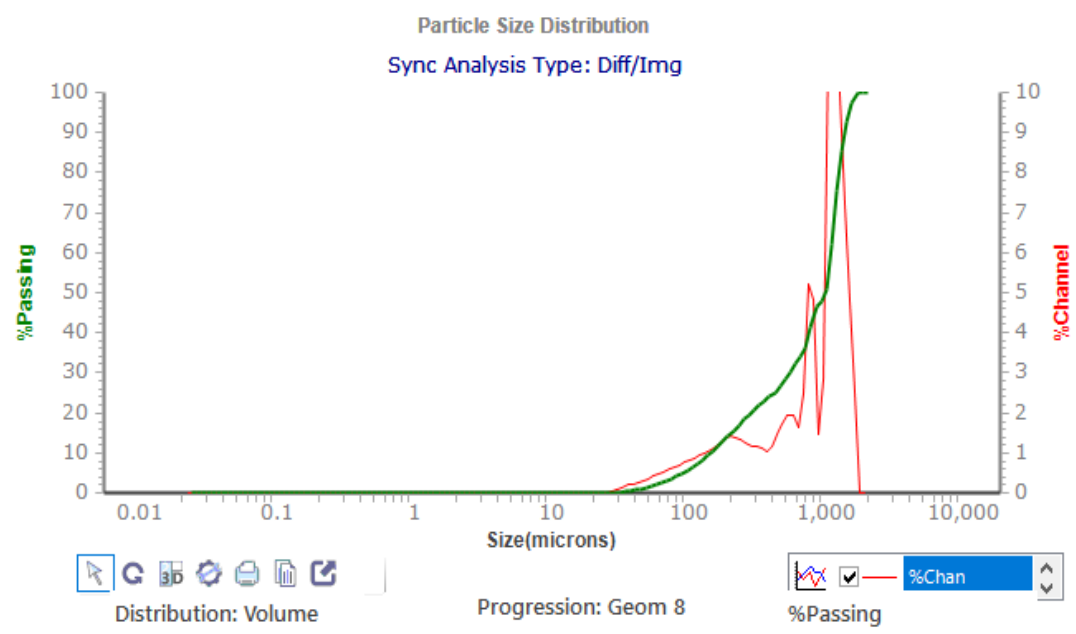
Summary		Percentiles		Size Percent		Peaks Summary		
Data	Value	%Tile	Size(um)	Size(um)	%Tile	Dia(um)	Vol%	Width
MV(um):	828.2	10.00	143.7			1166	53.7	350
MN(um):	59.63	20.00	280.0			741.7	14.1	136.9
MA(um):	361.0	30.00	536.0			466.6	10.3	181.1
CS:	1.66e-2	40.00	749.8			154.6	21.9	176.3
SD:	528.9	50.00	979.6					
		60.00	1072					
Mz:	822.2	70.00	1142					
si:	472.2	80.00	1227					
Ski:	-0.37268	90.00	1358					
Kg:	0.733	95.00	1460					

Wide PSD distribution  
~144U to 1460U

Sample polymer-like  
mildly soluble in Isopar-G

Classification of Flux may  
improve deposition  
furnace contaminant  
profile

- Bar->Line
- Large->Small
- Restore Graph
- Undo Keep
- Retained

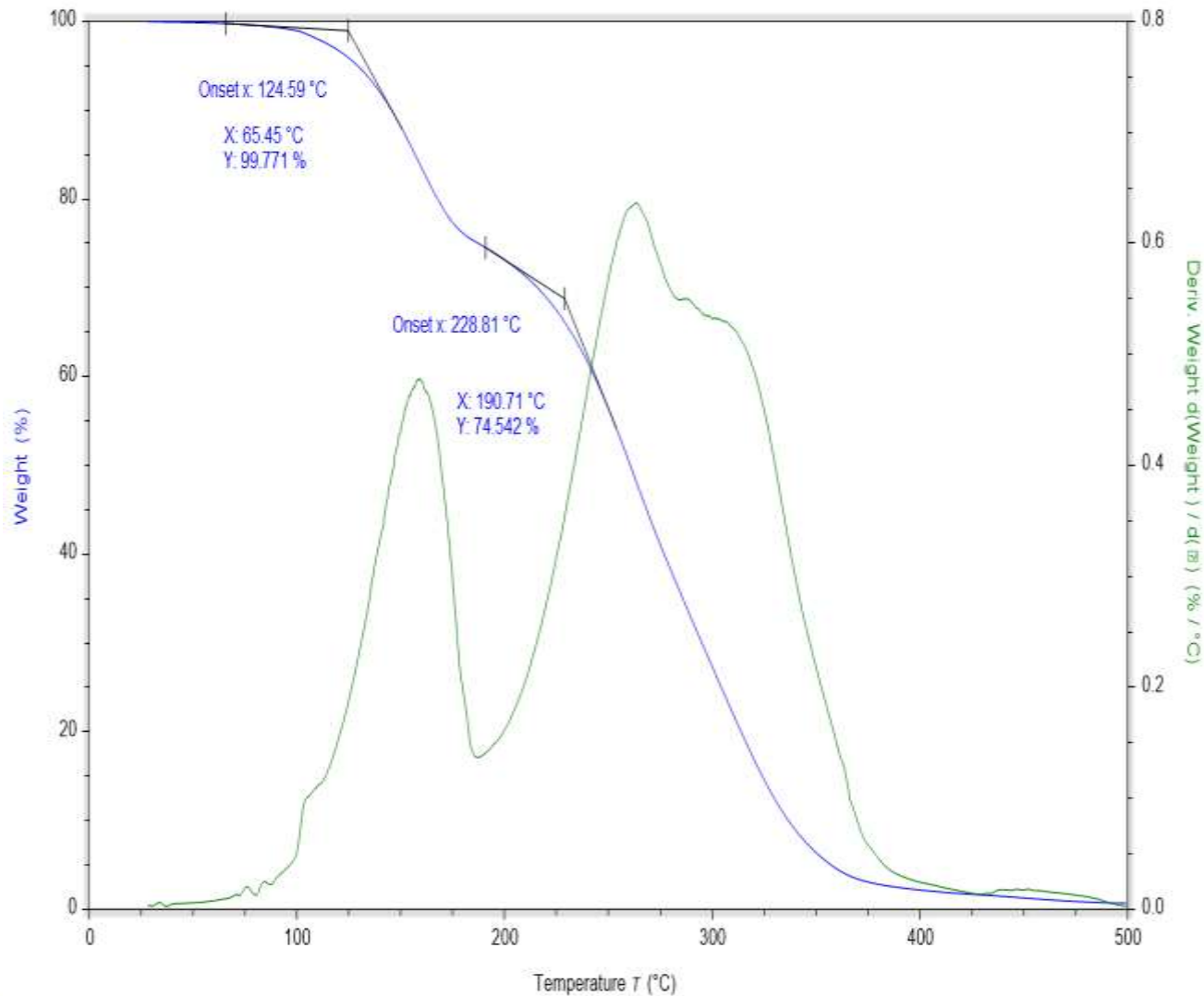




# Flux

TGA/DTA in N<sub>2</sub> : Ramp 5.0c/min to 500c

Non-MK FL nitrogen



Onset  
~65C – 500C

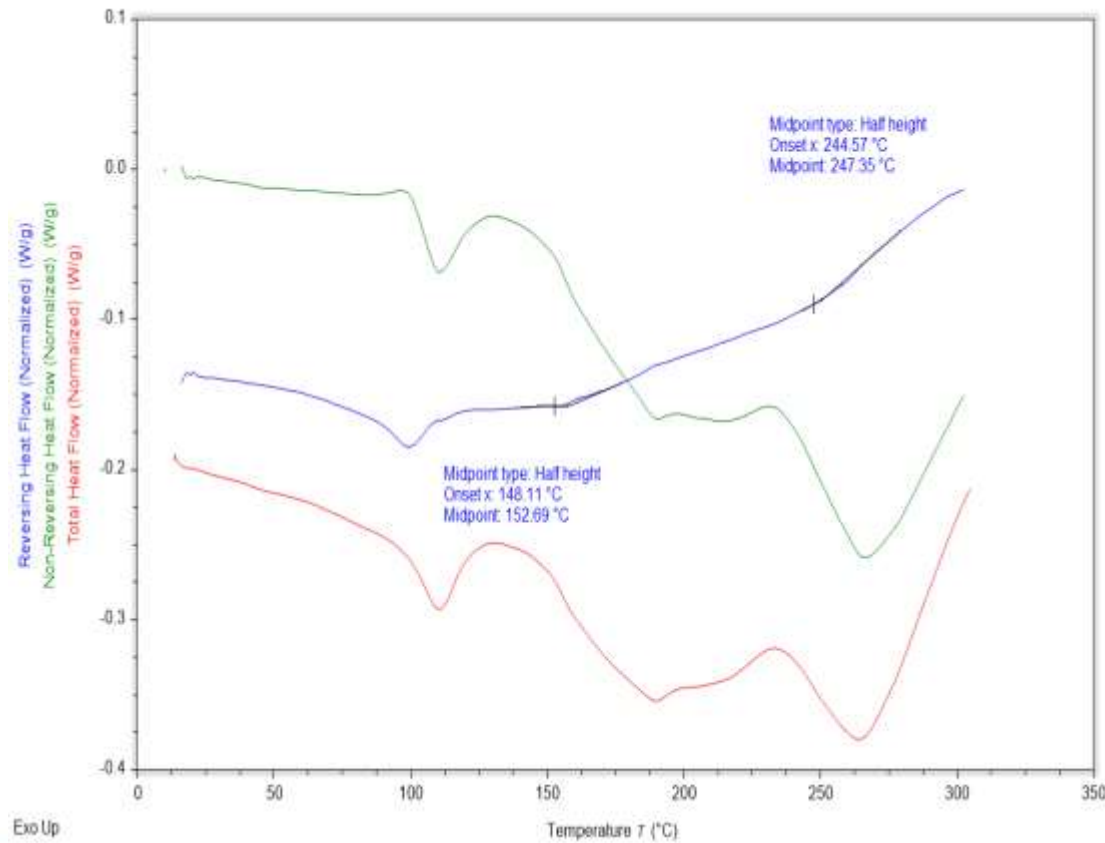
2 exotherms  
1 endotherm



## Flux Characterized by DSC - Differential Scanning Calorimetry ~300C

No.	Description
1	Equilibrate 10.00 °C
2	Isothermal 5.00 min
3	Data On
4	Modulate Temperature Amplitude 0.5000 °C , Period 60.0 s
5	Ramp 5.00 °C/min to 400.00 °C

Non-MK FL



DSC monitors changes in the heat effects associated with physical transitions and chemical reactions as a function of temperature. The DSC principle assumes that under constant pressure conditions the change of heat in any transition is equivalent to the change of enthalpy. The difference in heat flow to the sample and a reference is recorded as a function of temperature. DSC allows the user to study the kinetics of the physical processes and chemical reactions that occur during biodegradation, including enthalpy changes.

Not clear why heat in sample shows an increase with increasing T,



## Characterization in Progress

### SEM

flux – (particles, contaminants, features)  
aged zeolite - (pore size, contaminants)

### TGA/air

flux - (how air and flow influence oxidation of HC's)

### Porosity (N)

SA and micropore measurement  
Comparative, fresh and aged zeolite







# H1 Cross Flow Metal Substrate Development – High Mass Transfer Substrate

**Folded foil matrix**



Material: feccralloy-OCr21A16  
Gauge: 50 micron  
Corrugation height: 0.98mm  
Channel angle: 45 degree  
Foil width: 106.30mm  
Finished diameter: 138mmOD  
5.433"D x 4.185"L  
97.02" = V (total cat V = ~3.2L)  
No mantle

Nonionic wash  
Foil roll with nodal matrix braze  
Heat treat: 920C/2hrs/Air  
Corrugation height 0.98mm  
Corrugation width 2.1mm  
4 wraps of flat foil on perimeter  
Tolerance: +/- 0.38mmD  
Tolerance: +/- 0.13mmL



Technology	Substrate	Type	Dimensions	Quantity Requested	Quantity Received	Date Requested	Invoice Receive Date	PO
Trap-baseline (no coating)	wire mesh		5.66D" x 0.5"L	10	10	22-May	6/3/2022	PO337516/3
Trap-LPA (TH-100)	wire mesh		5.66D" x 0.5"L	10	10	22-May	6/3/2022	PO337516/3
H-Beta	metal monolith	straight channel	138mmD x 106.3mmL	3		6/2 & 7/15/22	NA	
H-Beta w/10% Cu	metal monolith	straight channel	138mmD x 106.3mmL	3		6/2 & 7/15/22	NA	
Cu-Y	metal monolith	straight channel	138mmD x 106.3mmL	3		6/2 & 7/15/22	NA	
ZSM-5	metal monolith	straight channel	138mmD x 106.3mmL	3		6/2 & 7/15/22	NA	
SCR	metal monolith	straight channel	138mmD x 106.3mmL	3		6/2 & 7/15/22	NA	
Cu-ZSM-5	metal monolith	straight channel	138mmD x 106.3mmL	3		6/2 & 7/15/22	NA	
DOC/DPF 5g/2:1:0	Cordierite	DOC/DPF	143.8*152/300	1				
DOC/GPF 5g/2:1:0	Cordierite	DOC/GPF	118.4*127/300	1		6/22/2022	PO375534	
CuSCR/GPF	Cordierite	CuSCR/GPF	118.4*127/300	1				
DOC/DPF 5g/2:1:0	Cordierite	DOC/DPF	143.8*152/300	1				
DOC/GPF 5g/2:1:0	Cordierite	DOC/GPF	118.4*127/300	1		7/15/2022	PC01153	
CuSCR/GPF	Cordierite	CuSCR/GPF	118.4*127/300	1				
H-Beta	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		
H-Beta w/10% Cu	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		
Cu-Y	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		
ZSM-5	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		
SCR	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		
Cu-ZSM-5	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		
DOC-DPF	Cordierite	floThru-300/400 cpsi	TBD	1		25-Jul		



## Zeolyst-Zeolites: Test Candidates for Flux and Formic Applications

Source	Material	Pore SizeD	BET SA	Si-Al Ratio
Zeolyst	CBV-400	24.5A	730	5.1
Zeolyst	CBV-5524G	5.5A	425	50.1
Zeolyst	CBV-2314		425	23.1
Zeolyst	CBV-720	24.28A	780	30.1

### Note:

- 2 Y zeolites and 2 ZSM-5 zeolites
- (CBV-2314, CBV-5524 are ZSM-5)
- (CBV-400, CBV-720 are Y Zeolites)
- CBV 5524G and CBV 2314 must be calcined at 500C for 1 hour in air to activate H form





# Molecular Weight Distribution of Residue in Flux Reactor Cold Traps

## GPC (Gel Permeation Chromatography) Eurofins

GPC separates organic compounds based on the size, (atomic radii), specifically by hydrodynamic volume

Size determination and separation are achieved through packed porous beads in a chromatographic column. Organics can be characterized by molecular wt and average molecular wt within a column.

Gels are used as the stationary phase for GPC. The pore size of a gel are carefully controlled in order to achieve a given separation. Other desirable properties of the gel forming agent are the absence of ionizing groups and, in each solvent, low affinity for the substances to be separated

The column used for GPC is filled with a microporous packing material. The column is filled with the gel.



# XRF result: Wash-coat scrapped from commercial metal monolith

## TEST REPORT

试验结果 Test Result			
委托人	张元	Sample submission date	2022-06-20

XRFUQ 数据					
Pat number	Sample description		Component	Test value %	Test method
2205770	Scratching powder		Al2O3	65.593	ECTC/I-AL-080
			CaO	0.120	
			CeO2	10.768	
			Cr2O3	0.342	
			Fe2O3	2.898	
			La2O3	1.771	
			MgO	0.052	
			MnO	1.881	
			Na2O	0.322	
			P2O5	0.600	
			SO3	0.652	
			SiO2	14.203	

Precious metals deemed too low in concentration to resolve signal to noise interference in providing accurate measurement

Wash-coat loading deemed very low

Wash-coat reported to be very difficult to remove from substrate, suggesting low porosity, low wash coat loading

99.202% wash-coat materials, with PM & contaminant balance



## ICP results suggest low PM and wash-coat content

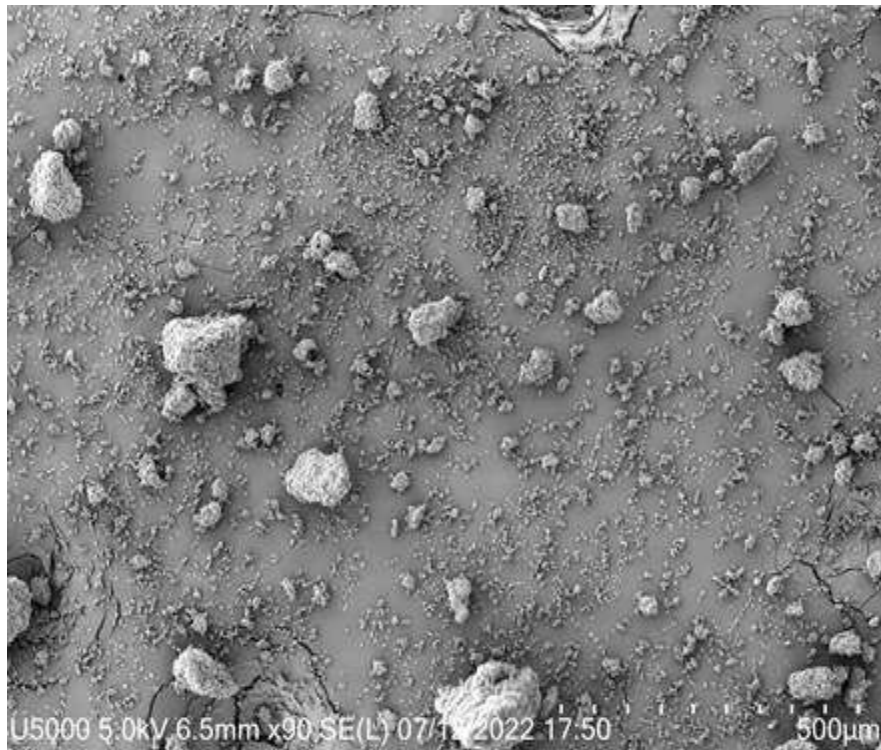
Component	Dark, before regeneration			Light, after regeneration		
	mg/brick	mg/in <sup>3</sup>	g/in <sup>3</sup>	mg/brick	mg/in <sup>3</sup>	g/in <sup>3</sup>
Pd	28.58	0.325883694	0.000325884	30.451	0.347217788	0.000347218
Pt	29.32	0.334321551	0.000334322	22.838	0.26041049	0.00026041
Al	3288.1	37.49258837	0.037492588	4783.968	54.54923603	0.054549236
Ba	2.15	0.024515393	2.45154E-05	2.53	0.028848347	2.88483E-05
Ca	14.53	0.165678449	0.000165678	35.341	0.402976055	0.000402976
Cr	20.87	0.237970353	0.00023797	16.894	0.192633979	0.000192634
Fe	240.73	2.744925884	0.002744926	273.328	3.116624857	0.003116625
Mg	1.15	0.013112885	1.31129E-05	2.115	0.024116306	2.41163E-05
Mn	446.67	5.093158495	0.005093158	479.181	5.46386545	0.005463865
Ni	81.84	0.9331813	0.000933181	363.32	4.142759407	0.004142759
La	626.84	7.147548461	0.007147548	599.95	6.840935006	0.006840935
Ce	749.98	8.551653364	0.008551653	1020.632	11.63776511	0.011637765
			<b>0.063064538</b>			<b>0.087007389</b>



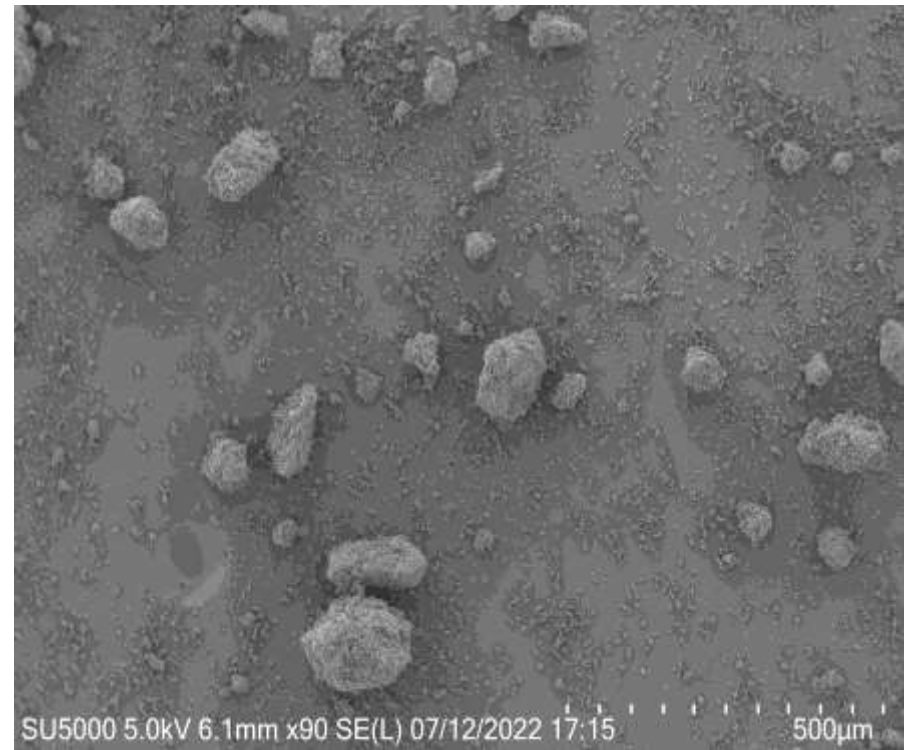
## SEM - Scanning Electron Microscopy for Aged Zeolite

Bulk analysis - Morphology technique - back scattered electrons penetration at various depths  
Irregular Particle Morphology  
Pore size, contamination profile not determined

Fresh 90x



Aged 90x

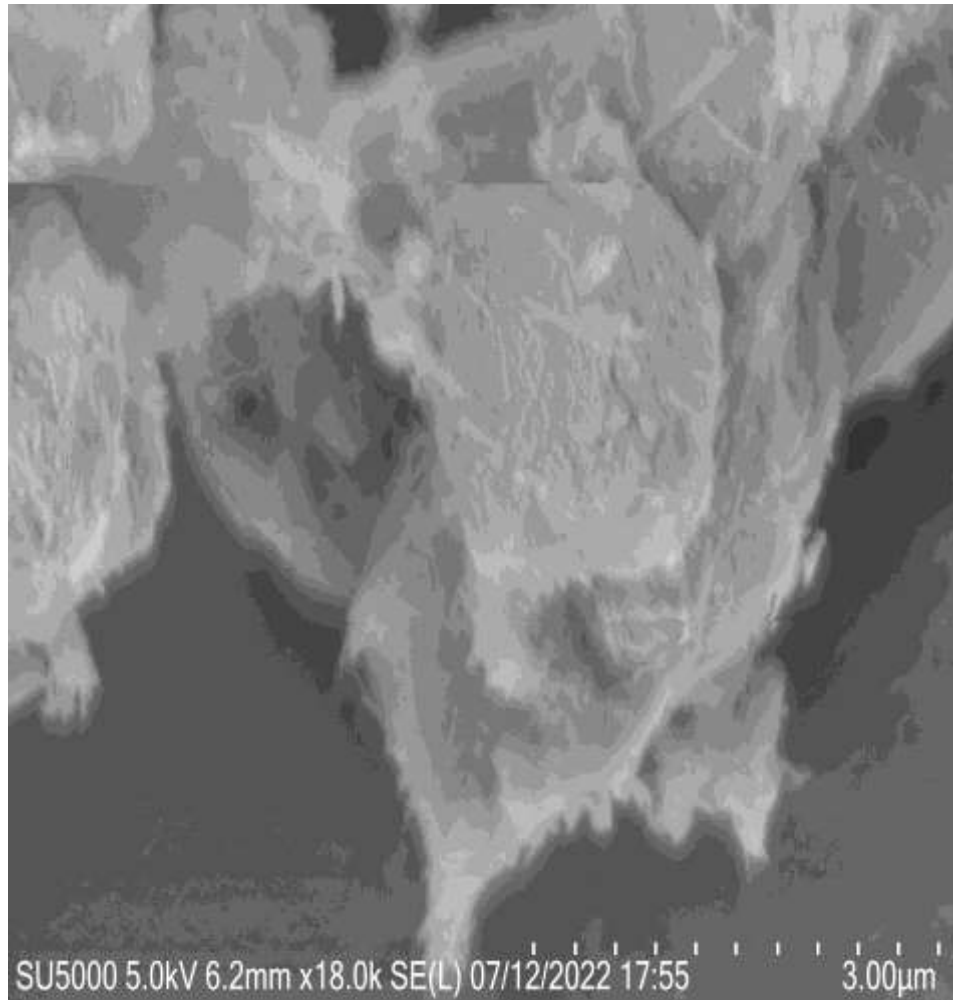




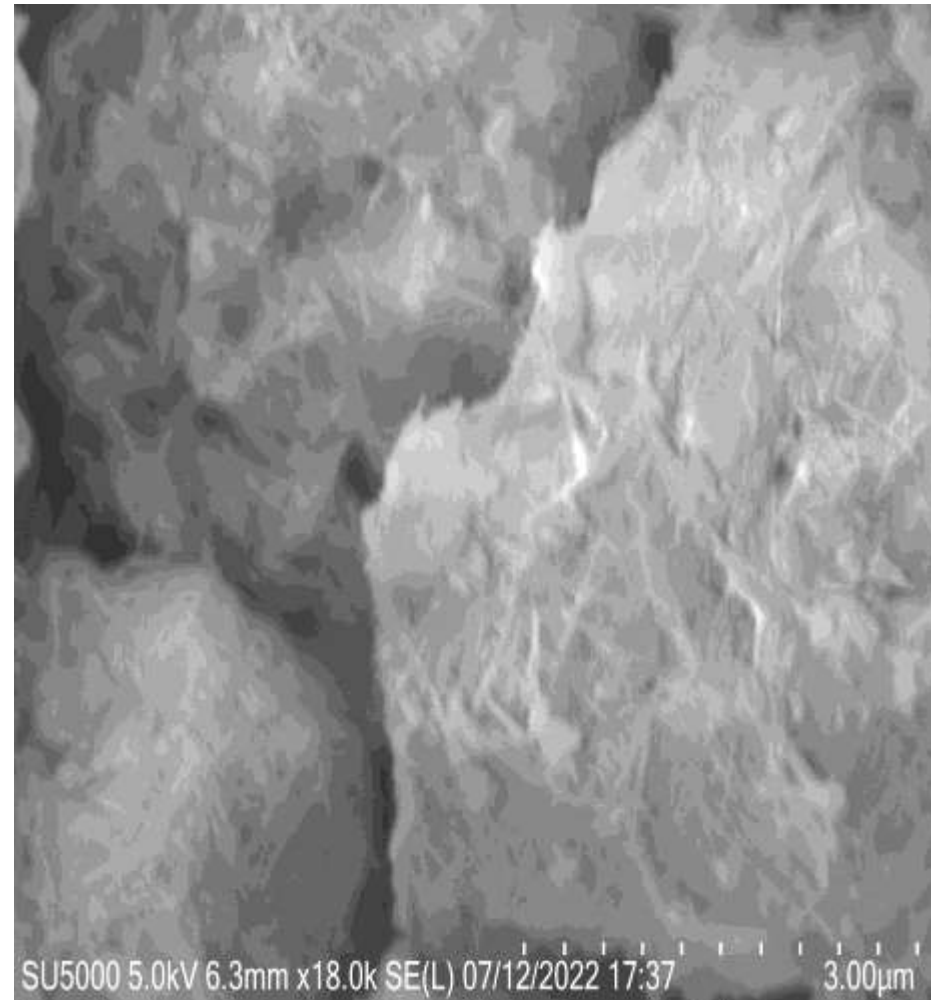


## SEM – X13 Zeolite 18KX

**Fresh**

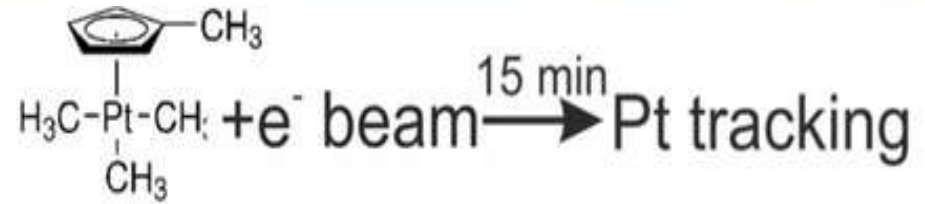


**Aged**



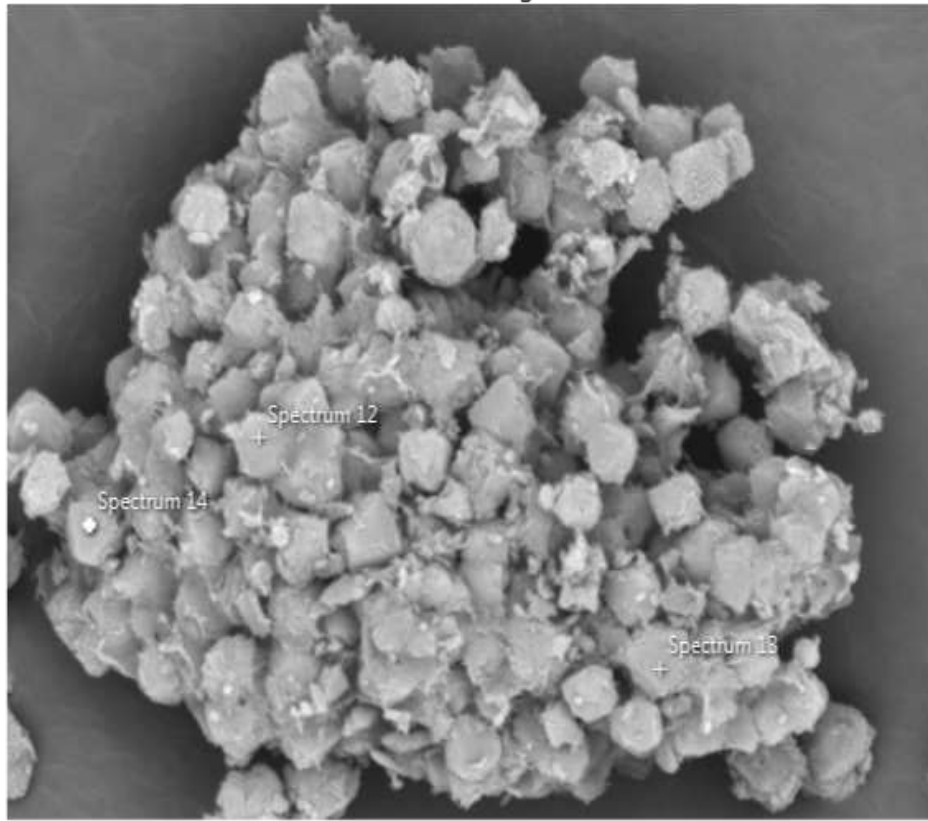


Primary zeolite particles ~ 1 microns D  
Zeolite pores ~ 1-10A D

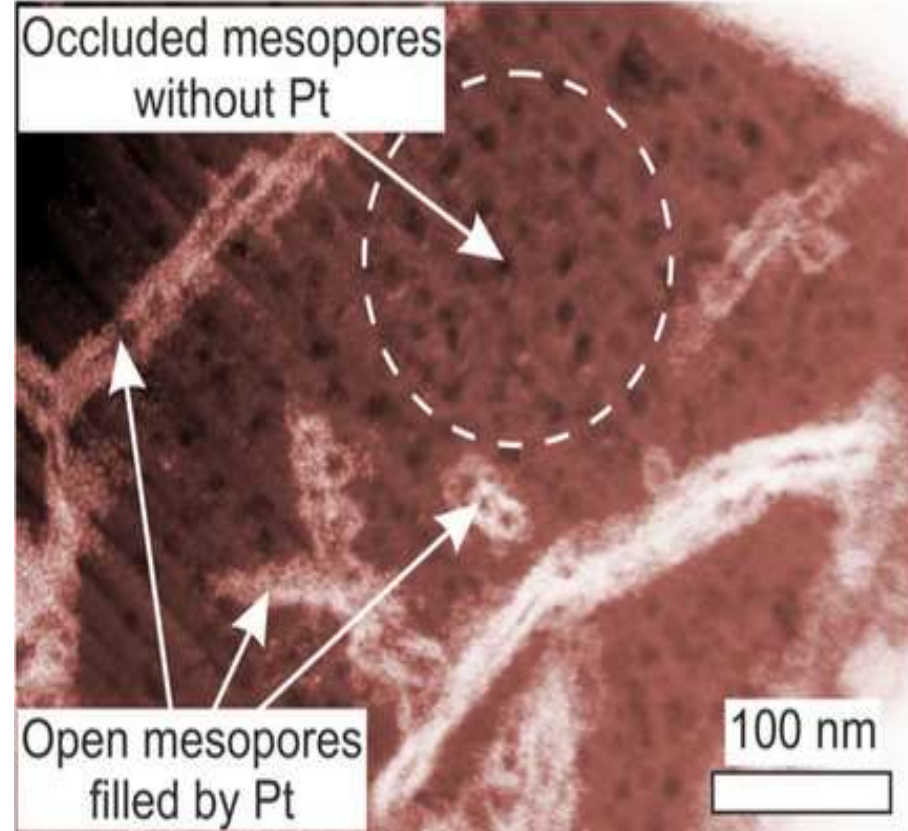


X13 3KX

Electron Image 1

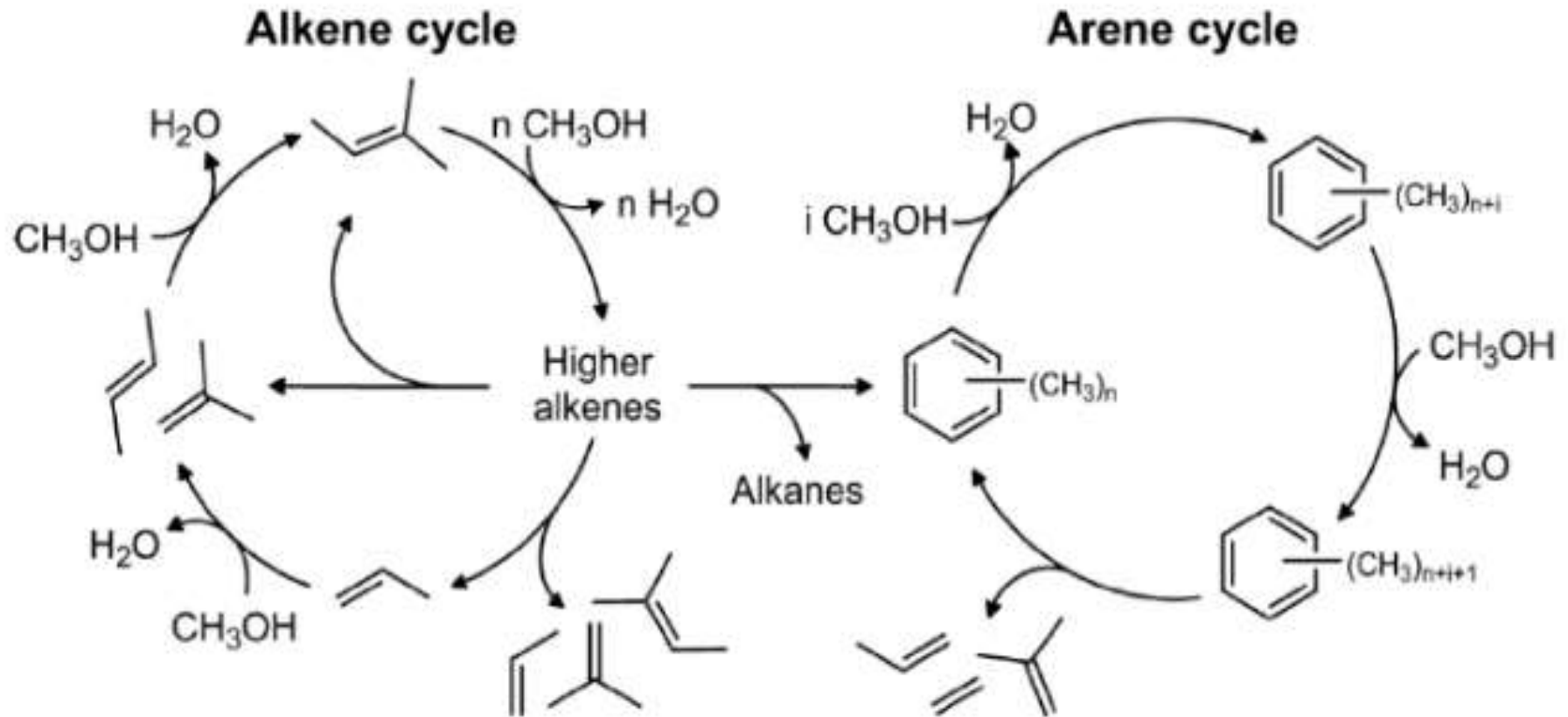


10μm





## Zeolite HC Cracking Kinetics - Condensation (Exothermic) Rx

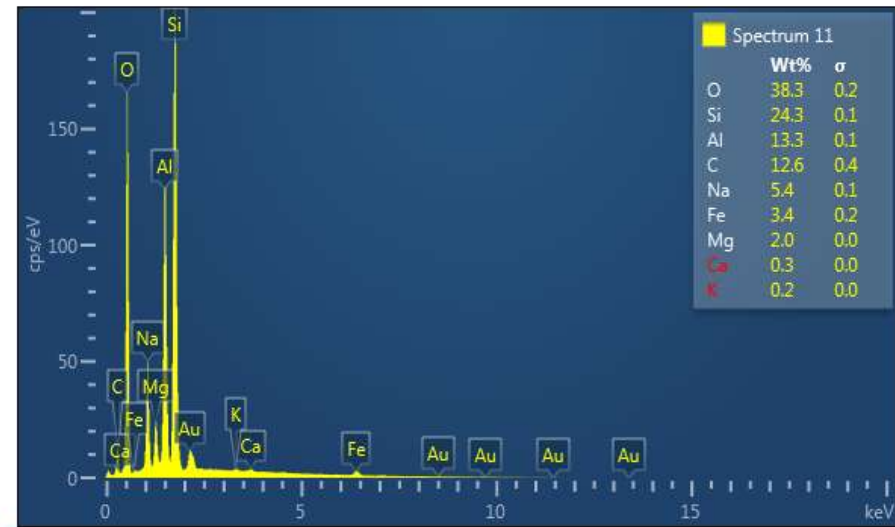
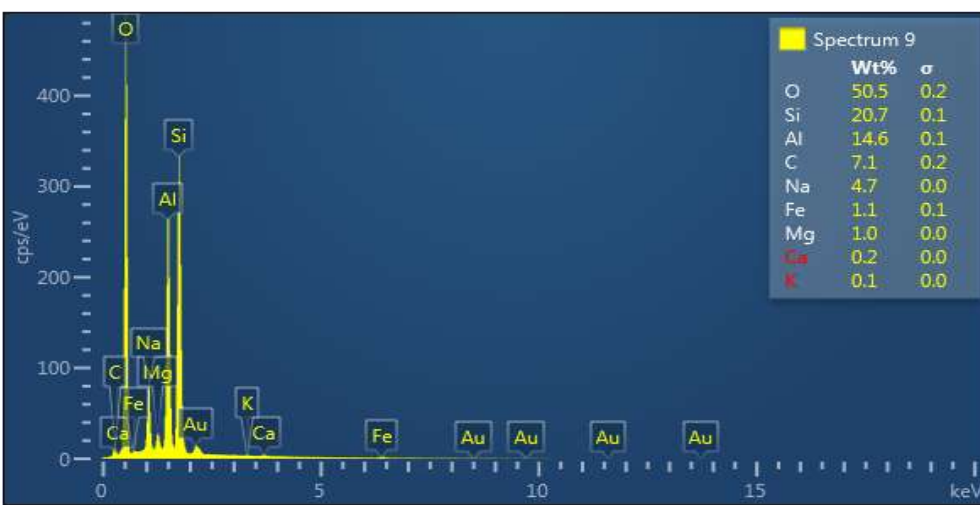
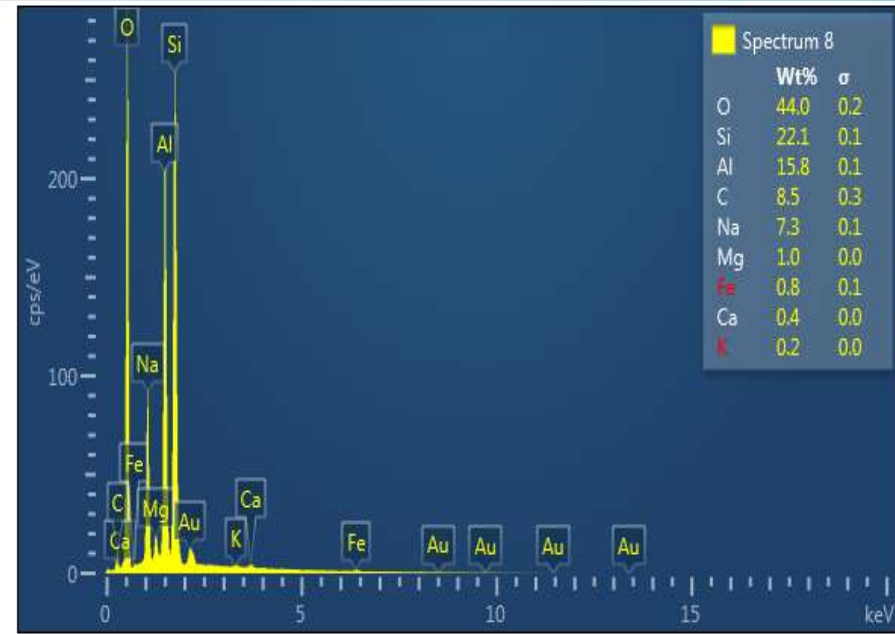
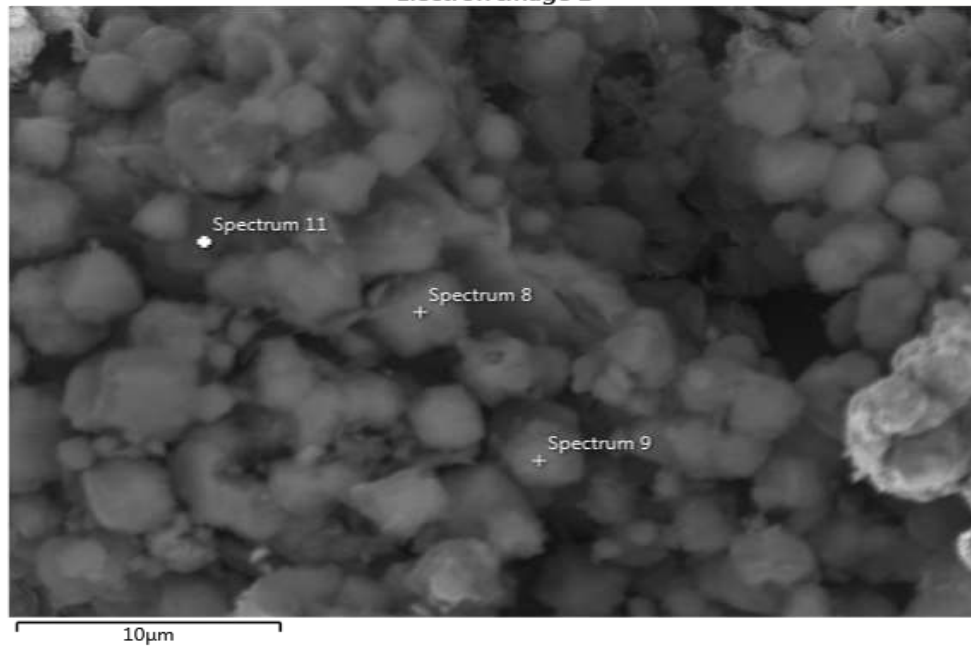






# Aged Zeolite – SEM/EDX 3KX

Electron Image 1

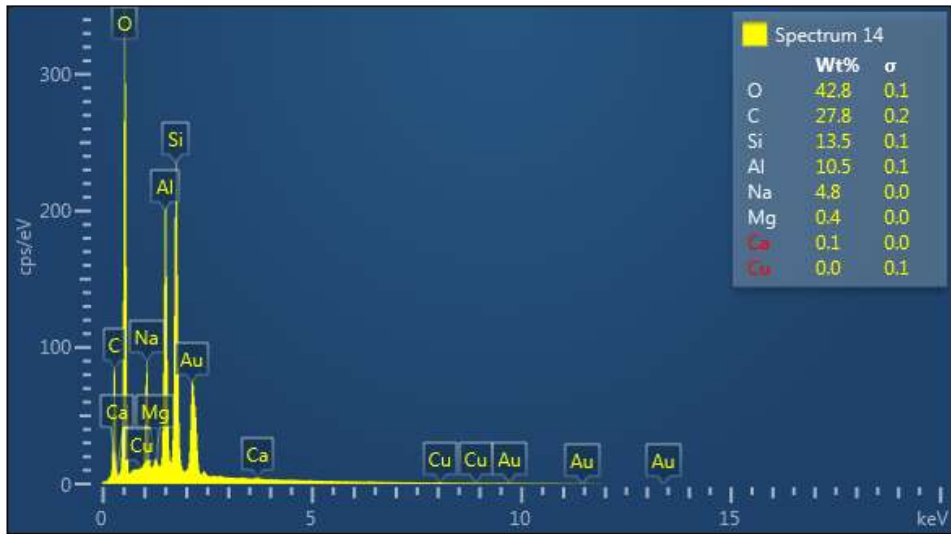
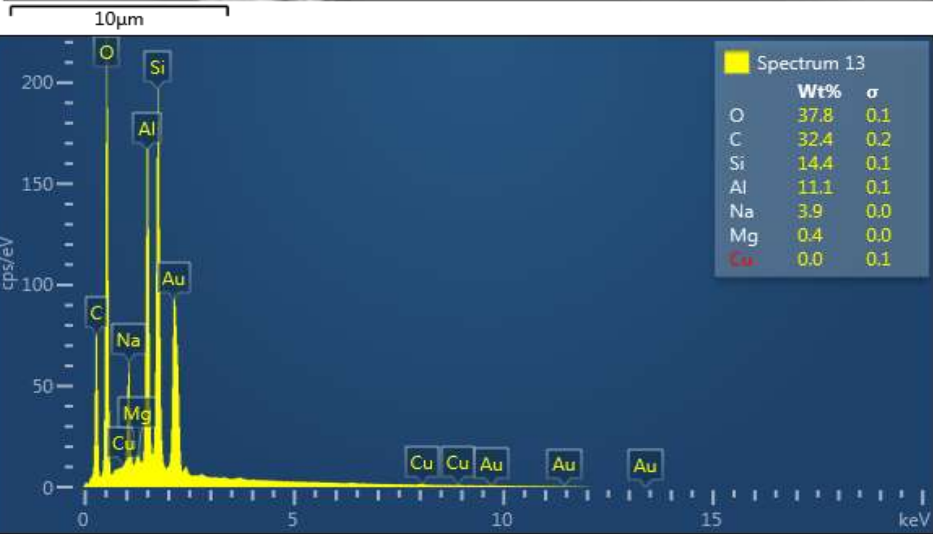
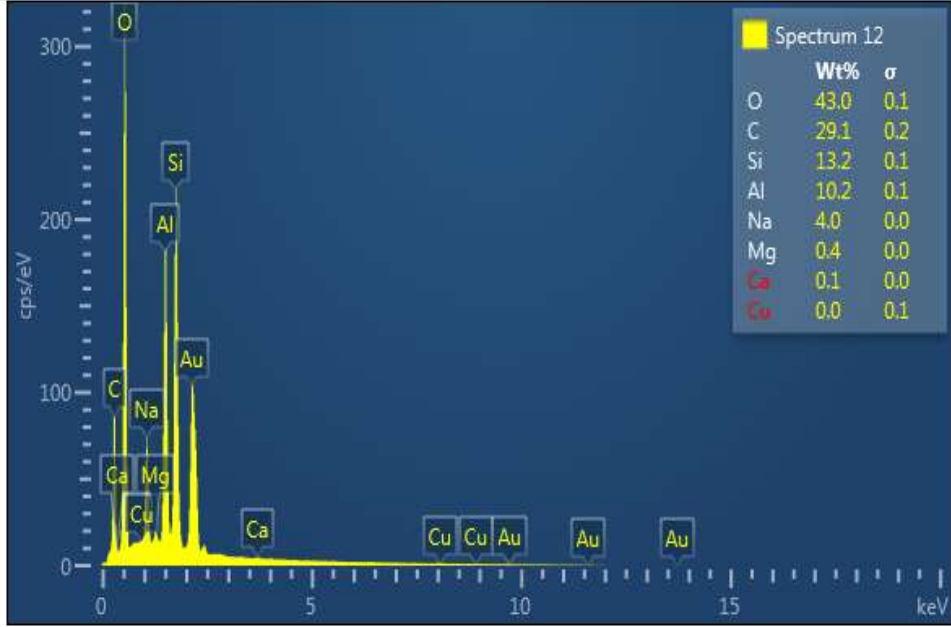
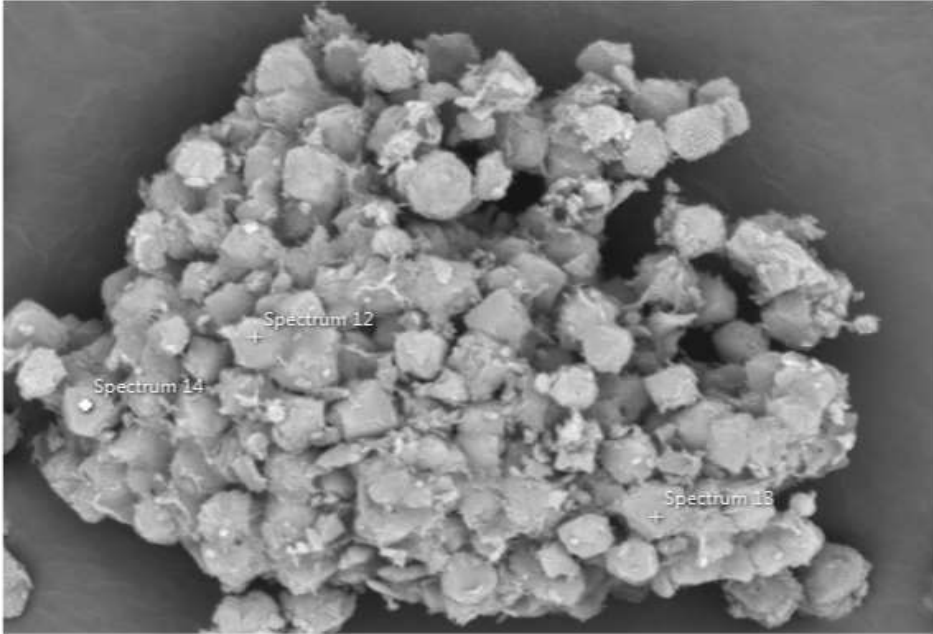






# Fresh Zeolite – SEM/EDX 3KX

Electron Image 1





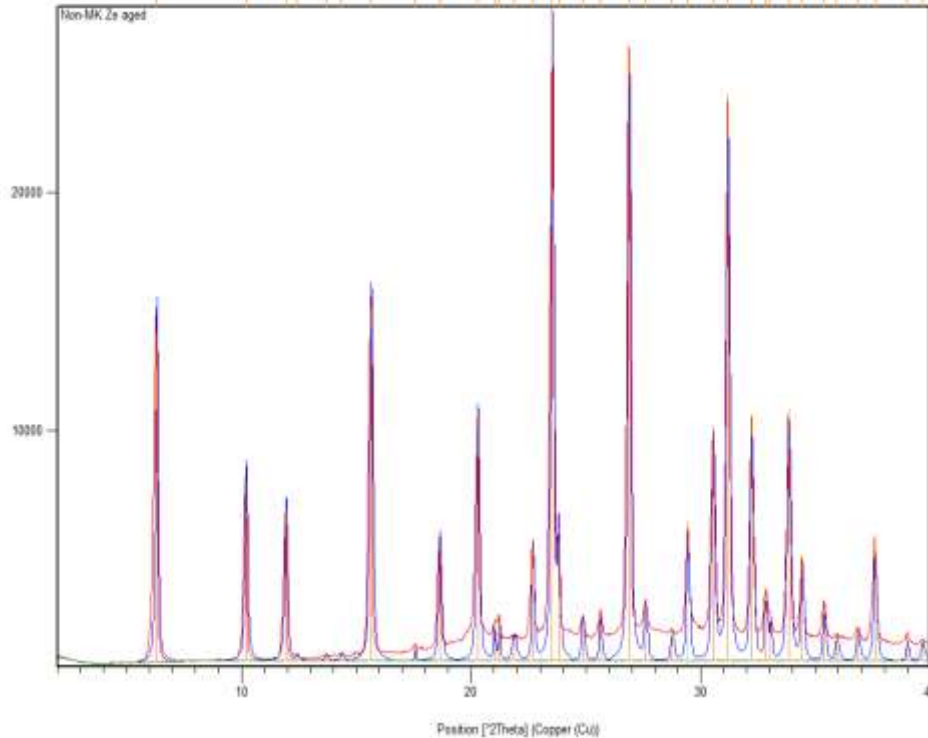
## SEM – EDX Comparative Contaminant Summary - X13 Zeolite

	Zeolite – Fresh				Zeolite - Aged			
				average				average
O	43	37.8	42.8	41.20	40.4	45.3	44.2	43.30
C	29.1	32.4	27.8	29.77	32.7	18.1	17.6	22.80
Si	13.2	14.4	13.5	13.70	11.5	17	18	15.50
Al	10.2	11.1	10.5	10.60	7.9	12.5	12.8	11.07
Na	4	3.9	4.8	4.23	5.7	5.5	5.6	5.60
<b>Mg</b>	<b>0.4</b>	<b>0.4</b>	<b>0.4</b>	<b>0.40</b>	<b>1</b>	<b>1</b>	<b>2</b>	<b>1.33</b>
Ca	0.1	0	0.1	0.07	0.3	0.2	0.3	0.27
Cu	0	0	0	0.00	0	0	0	0.00
<b>Fe</b>	<b>0</b>	<b>0</b>	<b>0</b>	<b>0.00</b>	<b>0.8</b>	<b>1.1</b>	<b>0.95</b>	<b>0.95</b>
K	0	0	0	0.00	0.1	0.2	0.1	0.13



## XRD – Aged Zeolite – X13 (time in service – unknown)

Counts



Selected Candidate: 00-005-0586

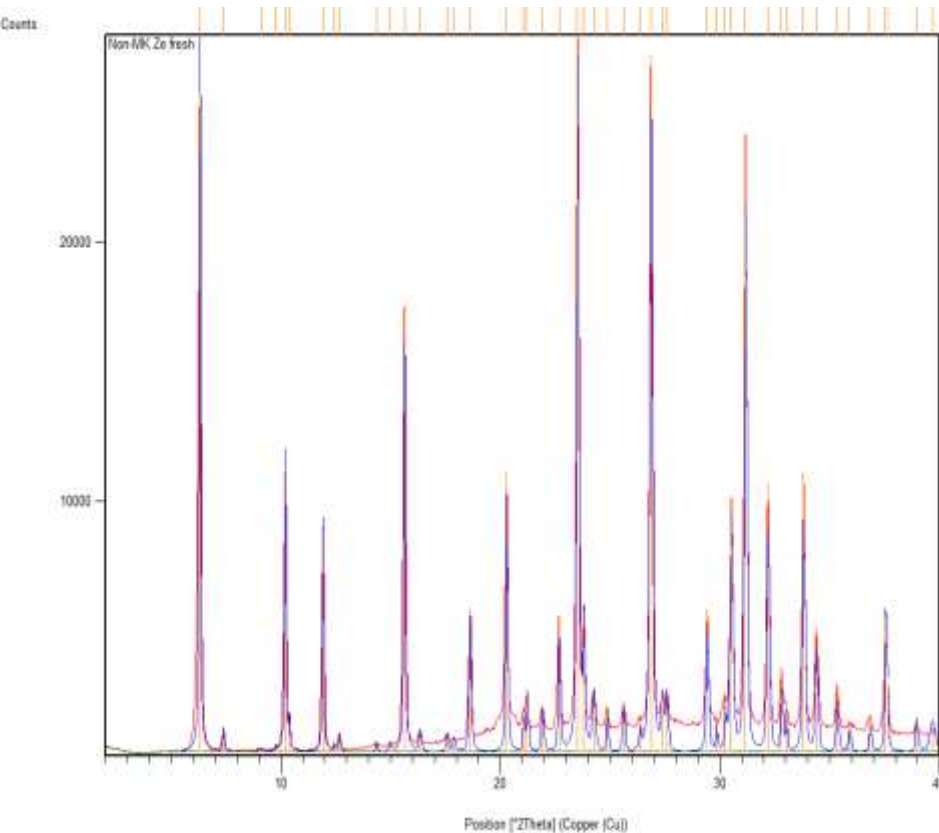
No.	Ref. Code	Score	Compound Name	Chemical Formula
1	00-005-0586	28	Calcite, syn	Ca C O3
2	00-033-1161	17	silica	Si O2
3	00-046-1045	17	Quartz, syn	Si O2
4	00-010-0173	16	alumina	Al2 O3
5	00-024-0072	13	Hematite	Fe2 O3
6	00-024-1977	13	Sucrose	C12 H22 O11
7	01-073-1667	7	Bornite	Cu5 Fe S4
8	00-033-0664	7	burnt ochre	Fe2 O3
9	00-006-0329	4	Praseodymium Oxide	Pr2 O3
10	01-075-0134	4	Uranium Oxide	U O2
11	01-077-2041	1	Sodium Erbium Flu...	Na Er F4
12	00-005-0628	1	Halite, syn	Na Cl

Sucrose, a disaccharide, (C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>), from monomers glucose and fructose like originated in a condensation reaction between the flux and thixotropic agents. Aged Zeolite ~ 6.0mmD originally 4.0mm, absorbed CO<sub>2</sub> and water.

Burnt Ochre = Fe<sub>2</sub>O<sub>3</sub>. FeO.(OH).nH<sub>2</sub>O, is a hydrated iron hydroxide, also referred to a gold ochre



## XRD – Fresh Zeolite – X13



Candidates:					
No.	Ref. Code	Score	Compound Name	Chemical Formula	
1	00-005-0586	22	Calcite, syn	CaCO <sub>3</sub>	
2	00-033-1161	15	silica	SiO <sub>2</sub>	
3	00-046-1045	15	Quartz, syn	SiO <sub>2</sub>	
4	00-010-0173	12	alumina	Al <sub>2</sub> O <sub>3</sub>	
5	00-024-0072	7	Hematite	Fe <sub>2</sub> O <sub>3</sub>	
6	01-073-1667	6	Bornite	Cu <sub>5</sub> FeS <sub>4</sub>	
7	00-006-0329	4	Praseodymium Oxi...	Pr <sub>2</sub> O <sub>3</sub>	
8	01-075-0134	3	Uranium Oxide	UO <sub>2</sub>	
9	00-005-0628	1	Halite, syn	NaCl	
10	01-077-2041	1	Sodium Erbium Flu...	NaErF <sub>4</sub>	

Zeolites are microporous, aluminosilicate minerals commonly used as commercial adsorbents and catalysts.

- Presence of Calcite is considered significant
- Zeolite extrudate size ~4mm
- Si –Al ratio ~3 to 1
- Pore size unknown (TEM being considered)





## LPA (TH-100) w/ 10% Cu Applied to Wire Mesh Trap

Un-coated wire mesh trap

Coated trap





DOC/DPF 143.8\*152/300 5g/2:1:0

DOC/GPF 118.4\*127/300 5g/2:1:0

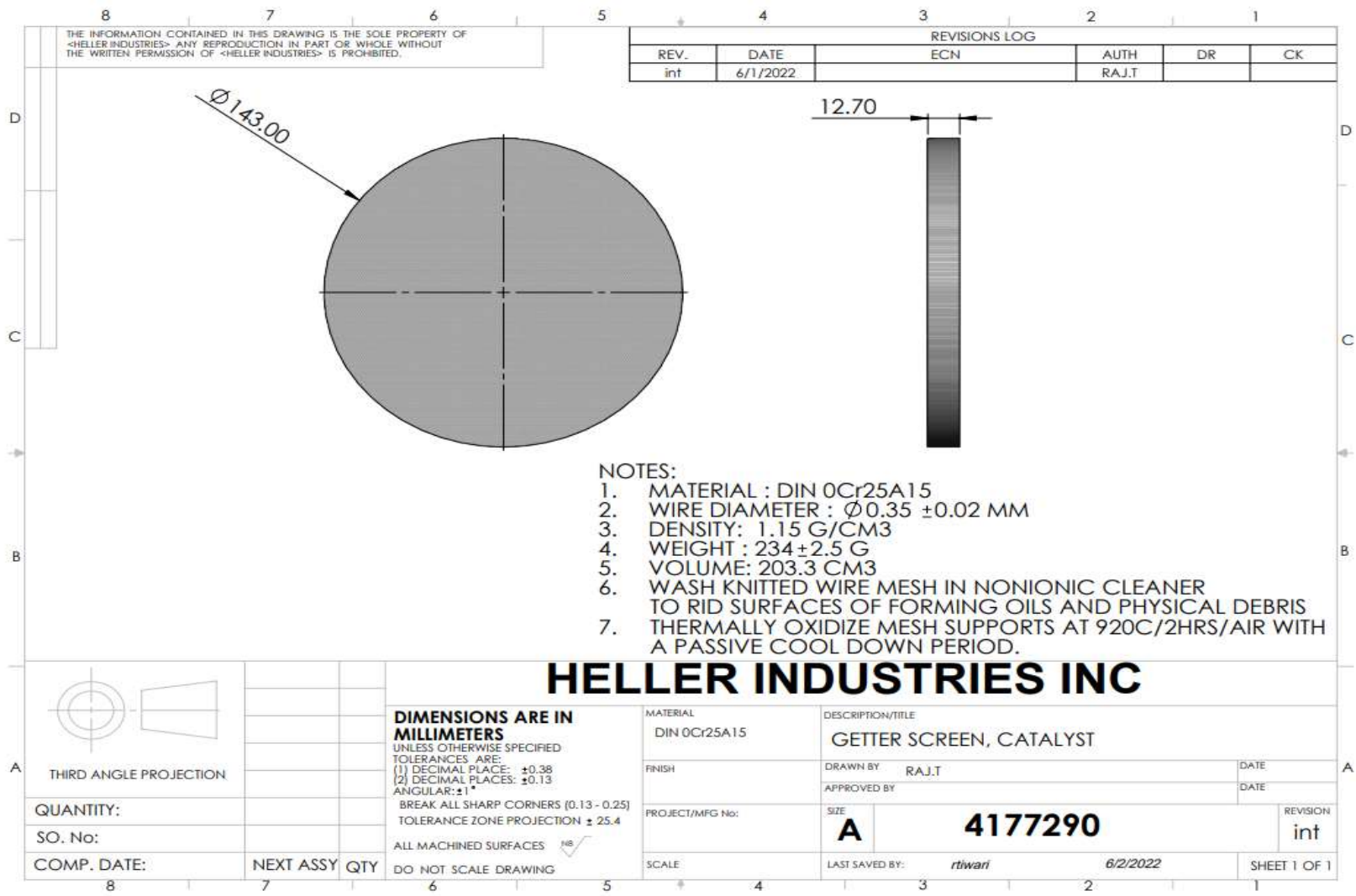
CuSCR/GPF 118.4\*127/300







# Wire Mesh Trap Support





## Atomic Size, Coking, Deactivation, Active Sites

Average zeolite is 1-10A Pore Diam.

10000A = 1 micron-U

C = 1.5A

O = 1.3A

H = 0.5A

H bond = 1.09A

- Silica/Alumina Ratio (SAR) will influence the acidity, thus activity
- Low SAR = more acidity with lower hydrophobicity, undesirable
- Higher Al results in greater coking and deactivation
- Too much acidity can lead to quicker coking and deactivation
- Many reactions require a tradeoff, (FAU and MFI)

Zeolyst: “some reactions can occur on the outer surface/pore mouth of a zeolite. so, molecules don’t always need to fit in the pore”.



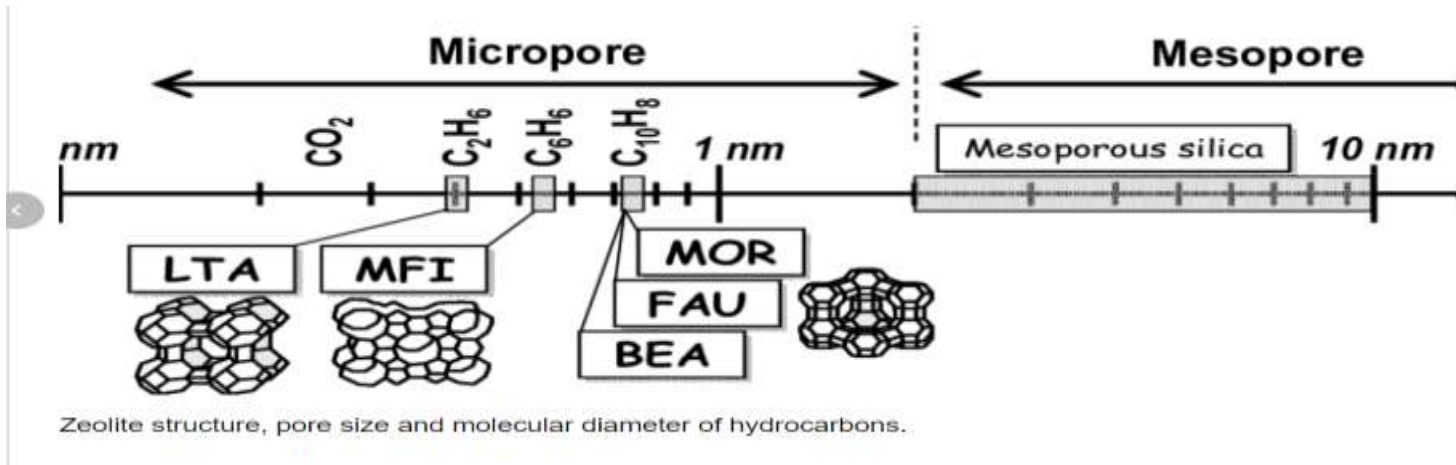


## Zeolites – Pore Size

Pore Size	Number of Tetrahedra (MR <sup>1</sup> )	Pore Diameter (Å)	Example
Small	8	4	PST-1 (NAT)
Medium	10	5.5	ZSM-5 (MFI)
Large	12	7.5	ZSM-12 (MTW)
Extra-large	>12	>7.5	CIT-5 (CFI)

<sup>1</sup> MR: Members of the ring.

Classification of zeolites according to their pore size.



Zeolyst: “some reactions can occur on the outer surface/pore mouth of a zeolite. so, molecules don’t always need to fit in the pore”



## Framework Type **CHA**

### Framework


#### Cell Parameters:

trigonal	R -3 m (# 166)	
$a = 13.6750 \text{ \AA}$	$b = 13.6750 \text{ \AA}$	$c = 14.7670 \text{ \AA}$
$\alpha = 90.000^\circ$	$\beta = 90.000^\circ$	$\gamma = 120.000^\circ$
Volume =	2391.6 $\text{\AA}^3$	
$R_{\text{DLS}} =$	0.0015	

Framework density (FD<sub>Si</sub>):  15.1 T/1000  $\text{\AA}^3$

Topological density:   $\text{TD}_{10} = 677$   $\text{TD} = 0.566667$

Ring sizes (# T-atoms): 8 6 4

Channel dimensionality:  Topological (pore opening > 6-ring): 3-dimensional

Maximum diameter of a sphere: 

that can be included 7.37  $\text{\AA}$

that can diffuse along  $a$ : 3.72  $\text{\AA}$   $b$ : 3.72  $\text{\AA}$   $c$ : 3.72  $\text{\AA}$

Accessible volume: 17.27 %

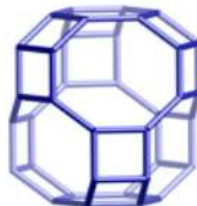
ABC sequence AABBC sequence of 6-rings

Secondary Building Units:  6-6 or 6 or 4-2 or 4

Composite Building Units: 



*d6r (t-hpr)*

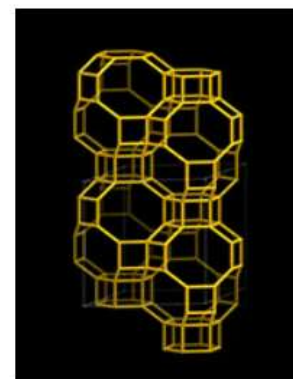


*cha (t-cha)*

Natural Tiling 

*t-cha* *t-hpr*

Framework images  
(click on icon for larger image)



Viewed normal to [001]



projection along [001]



## Framework Type **FAU**

### Framework


#### Cell Parameters:

cubic	F d -3 m (# 227)	
$a = 24.3450 \text{ \AA}$	$b = 24.3450 \text{ \AA}$	$c = 24.3450 \text{ \AA}$
$\alpha = 90.000^\circ$	$\beta = 90.000^\circ$	$\gamma = 90.000^\circ$
Volume =	$14428.8 \text{ \AA}^3$	
$R_{DLS} =$	0.0009	

Framework density ( $FD_{SI}$ ):  13.3 T/1000  $\text{\AA}^3$

Topological density:   $TD_{10} = 579$   $TD = 0.476190$

Ring sizes (# T-atoms): 12 6 4


Channel dimensionality:  Topological (pore opening > 6-ring): 3-dimensional

Maximum diameter of a sphere: 

that can be included 11.24  $\text{\AA}$

that can diffuse along a: 7.35  $\text{\AA}$  b: 7.35  $\text{\AA}$  c: 7.35  $\text{\AA}$

Accessible volume: 27.42 %

Secondary Building Units:  6-6 or 6-2 or 6 or 4-2 or 1-4-1 or 4

Composite Building Units: 



*d6r (t-hpr)*



*sod (t-toc)*

Natural Tiling 

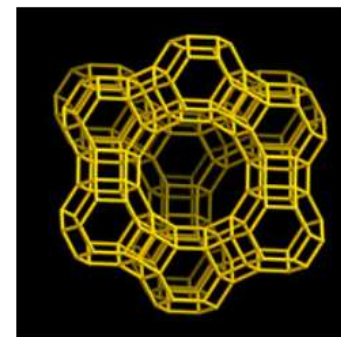
*t-fau*

*t-hpr*

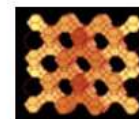
*t-toc*

### Framework images

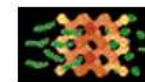
(click on icon for larger image)



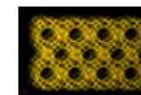
Viewed along [111]



Polyhedral model viewed along [110]



an 'artist impression' of a catalytic reaction



viewed along [110]