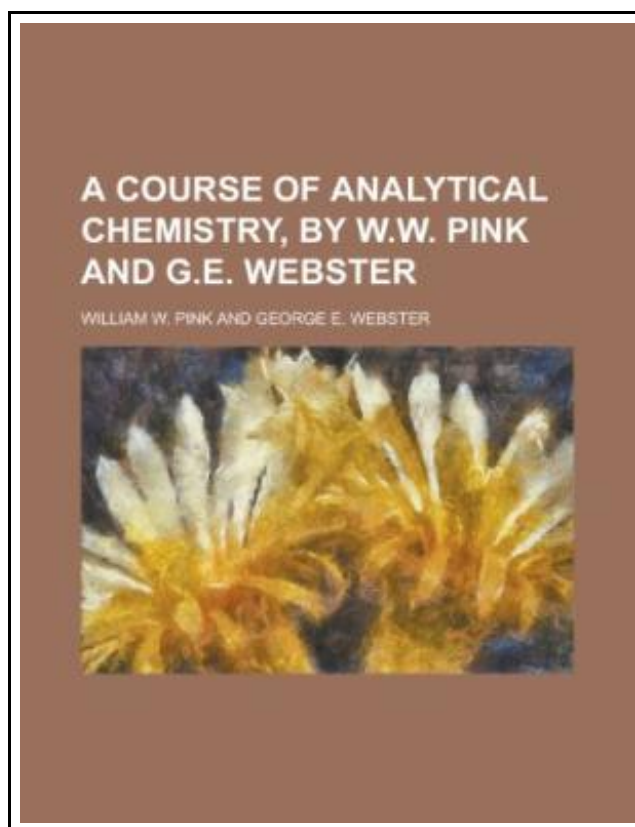


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A COURSE OF ANALYTICAL CHEMISTRY, BY W.W. PINK AND G.E. WEBSTER

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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1874 edition. Excerpt: . Acid. 121. Add MnO_2 , manganic oxide (black oxide of manganese), and SO_2 , then apply heat; evolution of chlorine takes place, known by its smell, colour, c. 122. NO_2 gives a curdy-looking white ppt. of $AgCl$, argentic chloride (chloride of silver), soluble in $AmHo$, but insoluble in weak NO_2Ho ; exposure to the sunlight always darkens this ppt. HBr .--Hydrobromic Acid. 123. Add MnO_2 and SO_2 , then apply heat; evolution of bromine vapour takes place, which colours moistened starch paper ytlhw. 124. NO_2 gives a cream-coloured yellow ppt. of $AgBr$, slightly soluble in $AmHo$, but insoluble in NO_2Ho . HI .--Hydriodic Acid. 125. Treat as in Reaction No. 123; evolution of iodine vapour of a violet or purple colour takes place, which colour moistened starch paper blue. 126. NO_2 gives a light yellow ppt. of AgI , argentic iodide (iodide of silver), soluble in KCy , but insoluble in weak NO_2Ho . No 127. HgO , mercuric nitrate (nitrate of mercury), gives a greenish-yellow ppt., insoluble in excess of the reagent. SH_2 .--Hydrosulphuric Acid. 128. Addition of HCl and application of heat causes evolution of SH_2 , known by its nasty smell; sometimes a piece of zinc is required in the case of those sulphides insoluble in HCl . $C(Me)O$ 129. PbO gives a black ppt. of PlS , plumbic $C(Me)O$ sulphide (sulphide of lead), with soluble sulphides. $AcEXIC$ Acid (This acid is only partly precipitated by NO_2 .) 129a. Add Fe_2Cl_6 , ferric chloride (perchloride of iron), to acetic solutions; it will change them...



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