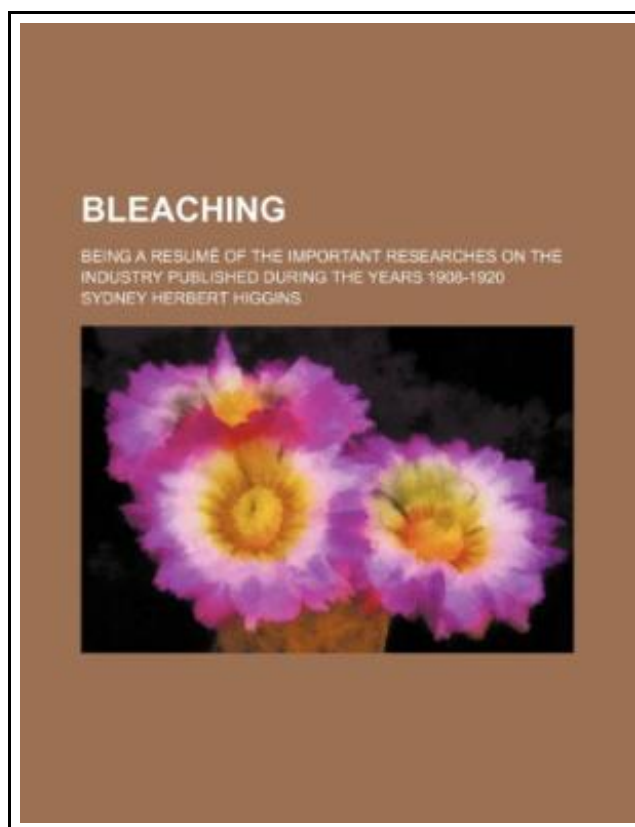


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1921 Excerpt: .directly. It is interesting here to note the conclusion of E. Justin-Mueller (J Pharm. Chim., 1919, 20, 113) that alkali hypochlorites and bleaching powder have similar chemical formulae, a conclusion formed after observations on the reactions of these bodies, and the statement of A. H. Meldrum (Journ. Soc. Chem. Ind., 1919, 80) that bleaching powder is more rapidly dried in desiccators by means of alkaline agents such as sodium hydroxide or basic calcium chloride, the tendency of bleaching powder to lose available chlorine being much reduced by desiccation. Bleaching powder loses chlorine in addition to water during the drying process, and this chlorine is absorbed by the alkaline drying agents. Bleaching Agents and Antiseptics During the war hypochlorites have been found very efficacious as antiseptics, and Higgins (Journ. Soc. Dyers and Col., 1918, 34, 189) shows that all the hypochlorite mixtures which have been used he had previously used as bleaching agents. Lorrain Smith, Drennan, Rettie, and Campbell (Brit. Med. Journ., July 1915) state that hypochlorous acid is a more effective antiseptic than hypochlorites, and, as already arrived at by various investigators, it is the most powerful antiseptic known. It had previously been shown that it is the most powerful of bleaching agents (Journ. Soc. Dyers and Col., 1914, 30, 326-328). H. D. Dakin (Brit. Med. Journ., 1915, 2, 318) described an antiseptic solution containing sodium hypochlorite and sodium bicarbonate, which mixture had previously been used for bleaching (Journ. Soc. Dyers and Col., 1912, 28, 30....



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