

# AE 234/711 Aircraft Propulsion

## Tutorial 5 - Thermodynamics

1. Show that  $c_p - c_v = \mathcal{R}_g$ .
2. Derive the expressions that relate  $T - s$  and  $p - v$  for isobaric, isochoric, isothermal and isentropic processes. Use these expressions to represent them on  $T - s$  and  $p - v$  axes.
3. Consider the compression of a system from a volume of  $\mathcal{V}_i$  to a smaller volume  $\mathcal{V}_f$ . Compare the change in internal energy, pressure gain, work and heat interactions with surroundings isobaric, isothermal and isentropic processes. Which of these is the most efficient way to compress a gas?
4. Derive the expression for the ideal efficiency of an Otto Cycle. Plot the dependence of this efficiency as a function of the compression ratio.
5. Derive the expressions for the efficiencies of Carnot and Diesel cycles.
6. The temperature at the beginning of the compression process of an air-standard Otto cycle with a compression ratio of 8 is 300 K, the pressure is 1 bar, and the cylinder volume is  $560 \text{ cm}^3$ . The maximum temperature during the cycle is 2000 K. Determine (a) the temperature and pressure at the end of each process of the cycle, and (b) the thermal efficiency.

Stage 1 (100 kPa, 560 cc, 300 K, 0 J/kg-K) → Stage 2 (1837.9 kPa, 70 cc, 689.2 K, 0 J/K) → Stage 3 (5333.3 kPa, 70 cc, 2000 K, 8.9 J/K) → Stage 4 (290.2 kPa, 560 cc, 870.6 K, 8.9 J/K), Efficiency is 56.5 %

How would these change if there was a energy loss of 1% during compression and expansion? Assume that the compression ratio and maximum temperature are maintained as before.

Stage 1 (100 kPa, 560 cc, 300 K, 0 J/K) → Stage 2 (1853.3 kPa, 70 cc, 695 K, 0.1 J/K) → Stage 3 (5333.3 kPa, 70 cc, 2000 K, 8.9 J/K) → Stage 4 (290.2 kPa, 560 cc, 863.6 K, 8.8 J/K), Efficiency is 55.7 %