# My Thermal & Statistical Physics Notes for Jomo Kenyatta University of Agriculture and Technology

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### Introduction

Thermal and Statistical Physics is the best of the Sciences and I aim to cover a lot in my Notes.

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### Chapter 1

### Heat and Temperature

#### 1.1 What is Heat?

We could think of heat as:

- The transfer of thermal energy from a region of high thermal energy to a region of low thermal energy, provided that the two regions are in contact, is known as **heat**.
- The transfer of thermal energy between molecules within a system

The SI unit for Heat is the joule J

#### Just a heads up...

A body can only lose or gain heat but not possess heat as heat is a measure of thermal energy and not a property of matter.

#### 1.2 What is Temperature?

We can define temperature as:

- (*layman*) the hotness or coldness of a body.
- (*more technical*) a measure of the <u>mean</u> kinetic energy of a body in a thermal energy system.

The SI unit for Temperature is the kelvin K.

## Appendices

## Bibliography