

# Enzymatic degradation of plastics waste: a geometric model



Michael Schindler and Ludwik Leibler

Laboratoire *Gulliver*, Paris  
UMR 7083 CNRS/ESPCI



# Polyester waste: recycling by enzymatic de-polymerisation?

---



PET bottle waste



96% after 24h



PET textile waste  
swimwear  
lining



82%



PET fiber waste  
automotive scrap



36%



PET/PBT textile  
swimwear  
trunk



15%



PBT



0%

## What do enzymes need?

enzymes act only at the **surface**

they "need space" to act:

**amorphous** substrate (avoid crystals!)

substrate **mobility** (avoid glass!)

natural temperature range (enzymes are proteins)

roughly 40–70°C

water

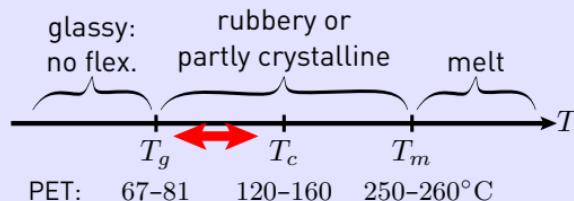
## Prepare the polyesters:

### prepare:

melt + quench

mill + sieve: break into pieces  $\sim 300\mu m$

### de-polymerise:



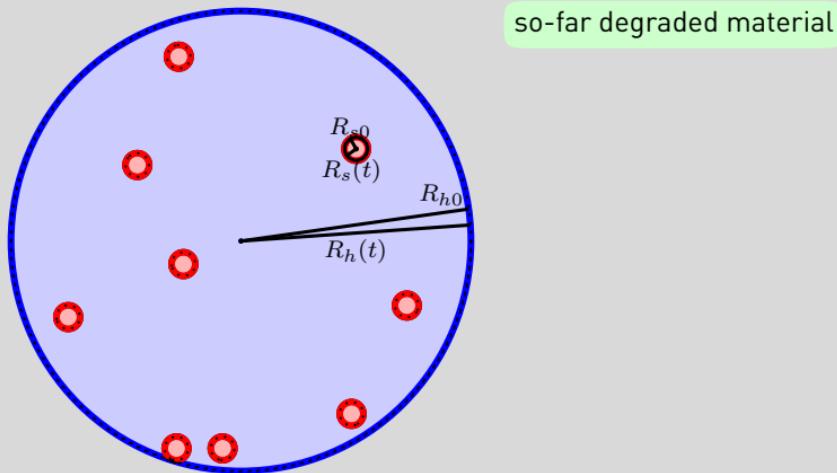
initial crystallinity + nucleation  
**crystal growth!**

**The growing crystallites  
are a problem!**

# Competition: de-polymerisation vs. crystal growth

**model:** one shrinking sphere: amorphous material: degradable

many growing spheres: (partly) crystalline material: not degradable



so-far degraded material

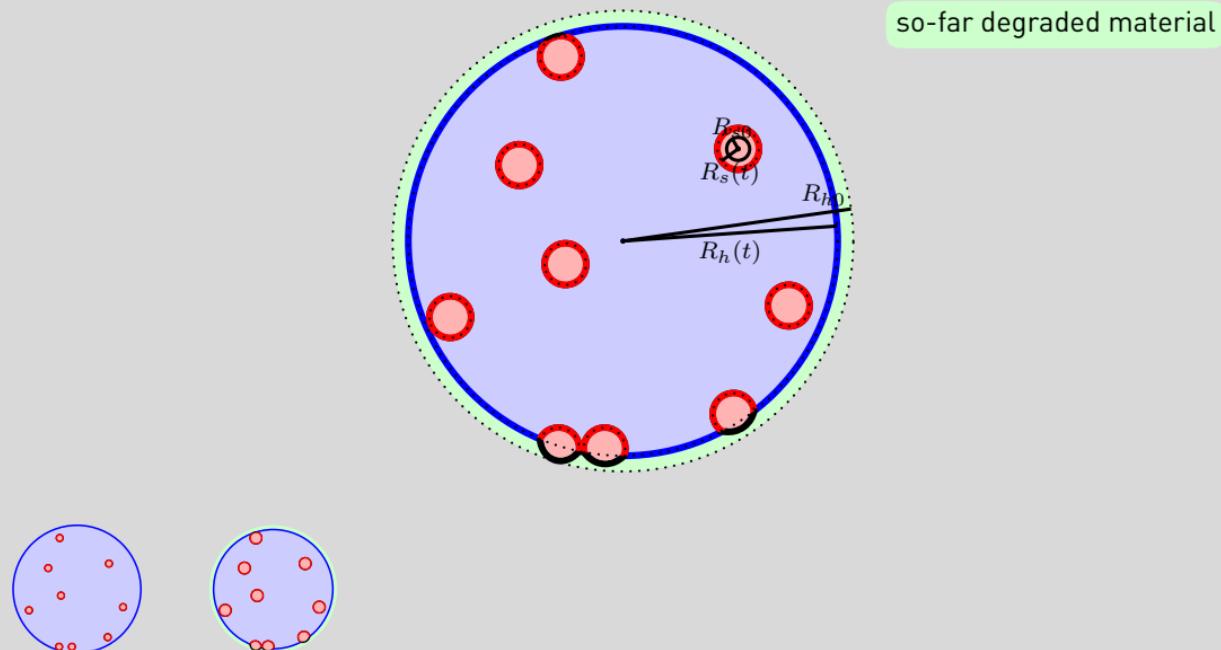
**parameters:**  $N_s, \dot{R}_s/\dot{R}_h, R_{s0}/R_{h0}, (\dot{N}_s)$

depend on material, composition, preparation, processing temperature, ...

# Competition: de-polymerisation vs. crystal growth

**model:** one shrinking sphere: amorphous material: degradable

many growing spheres: (partly) crystalline material: not degradable



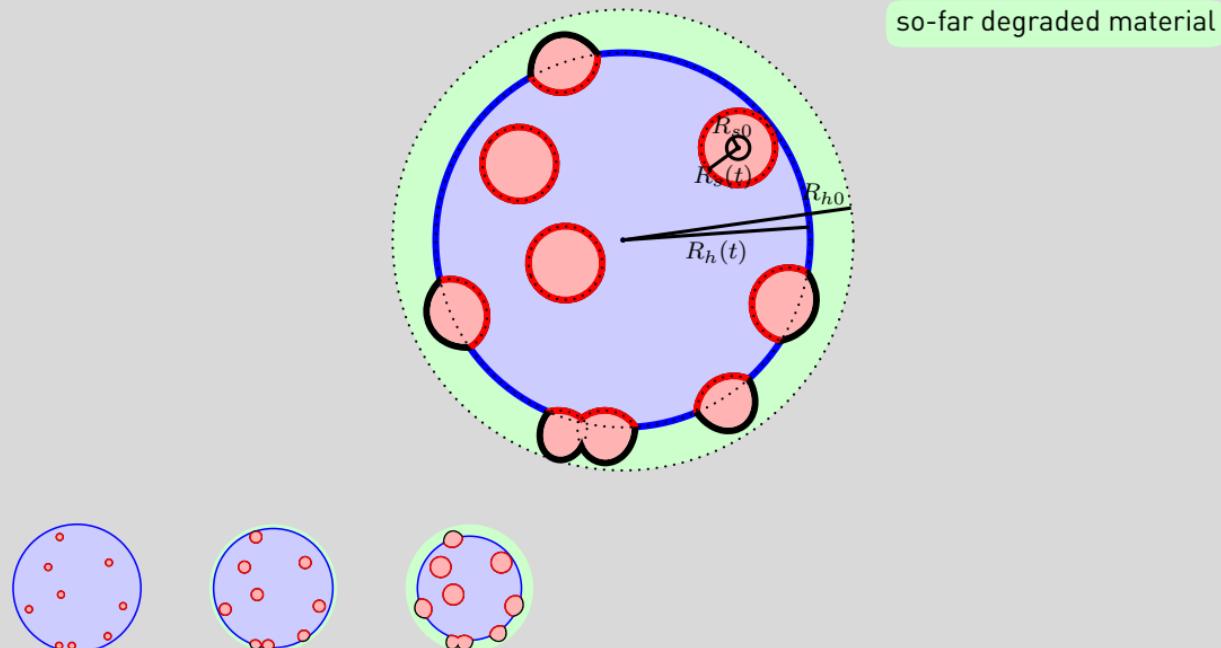
**parameters:**  $N_s, \dot{R}_s/\dot{R}_h, R_{s0}/R_{h0}, [\dot{N}_s]$

depend on material, composition, preparation, processing temperature, ...

# Competition: de-polymerisation vs. crystal growth

**model:** one shrinking sphere: amorphous material: degradable

many growing spheres: (partly) crystalline material: not degradable



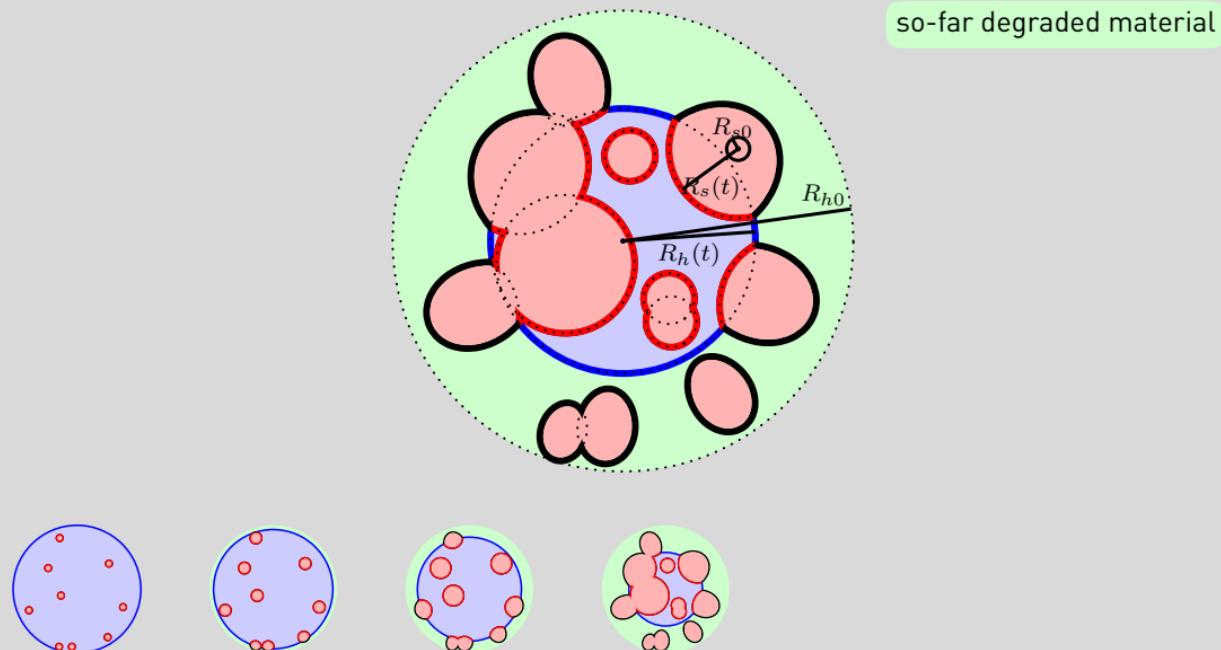
**parameters:**  $N_s, \dot{R}_s/\dot{R}_h, R_{s0}/R_{h0}, [\dot{N}_s]$

depend on material, composition, preparation, processing temperature, ...

# Competition: de-polymerisation vs. crystal growth

**model:** one shrinking sphere: amorphous material: degradable

many growing spheres: (partly) crystalline material: not degradable



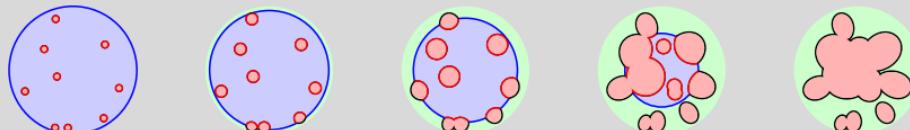
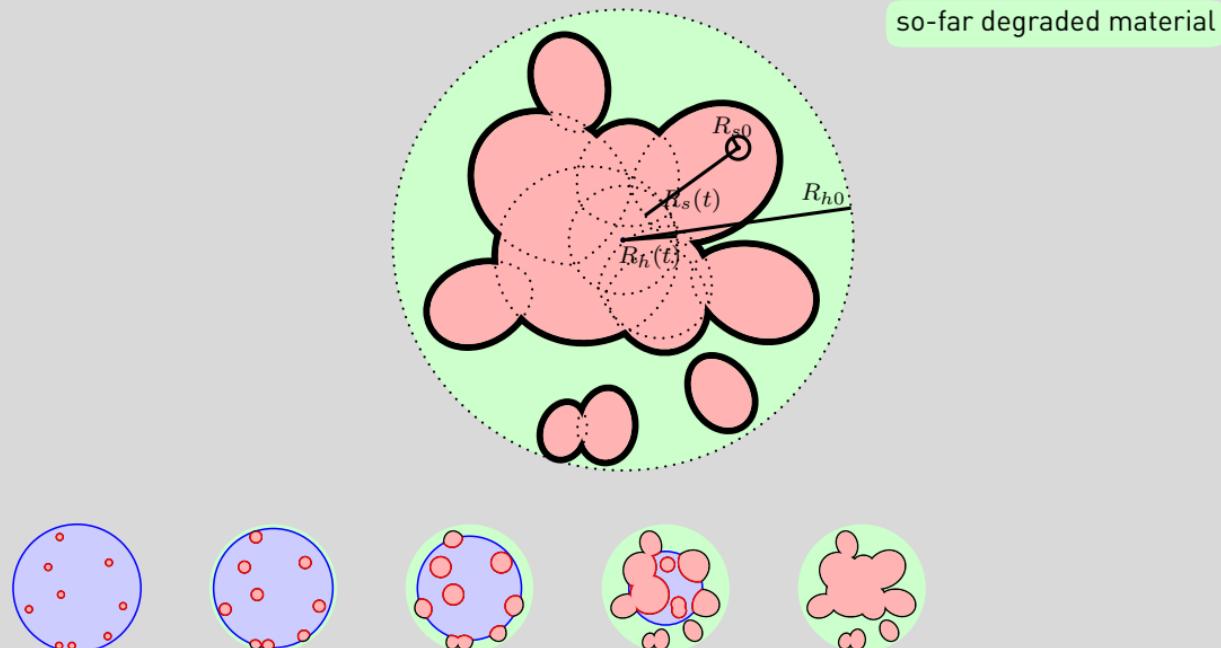
**parameters:**  $N_s, \dot{R}_s/\dot{R}_h, R_{s0}/R_{h0}, (\dot{N}_s)$

depend on material, composition, preparation, processing temperature, ...

# Competition: de-polymerisation vs. crystal growth

**model:** one shrinking sphere: amorphous material: degradable

many growing spheres: (partly) crystalline material: not degradable

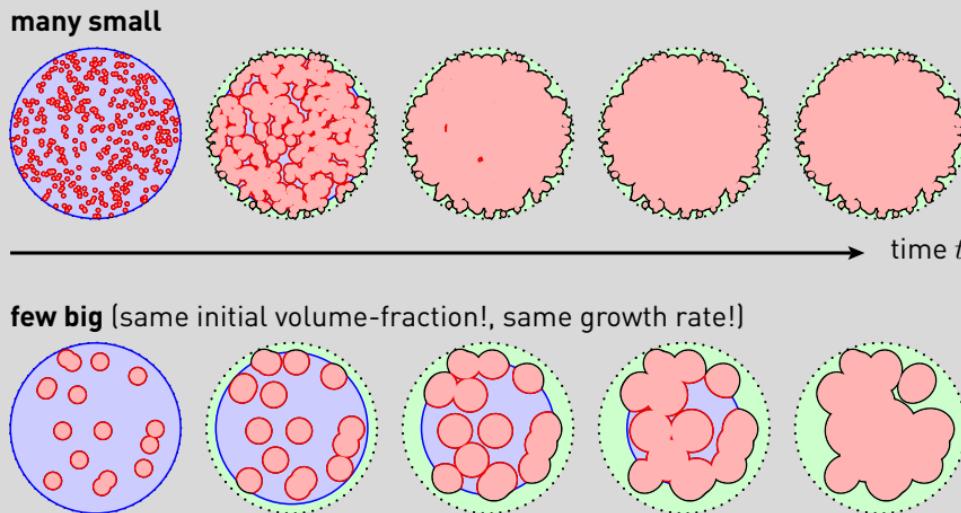


**parameters:**  $N_s, \dot{R}_s/\dot{R}_h, R_{s0}/R_{h0}, [\dot{N}_s]$

depend on material, composition, preparation, processing temperature, ...

# Competition: de-polymerisation vs. crystal growth

**careful!** there is more than just the total initial crystallinity



**parameters:**  $N_s, \dot{R}_s/\dot{R}_h, R_{s0}/R_{h0}, [\dot{N}_s]$

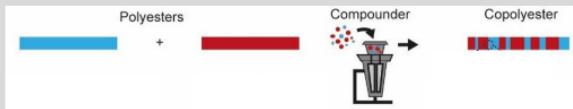
depend on material, composition, preparation, processing temperature, ...

# How to slow down crystallisation?

**"Classic" optimisation:** understand the parameters for every material  
including their temperature dependence, ...  
sort materials (!?)  
optimize parameter compromises  
repeat the procedure also for mixtures (!?)

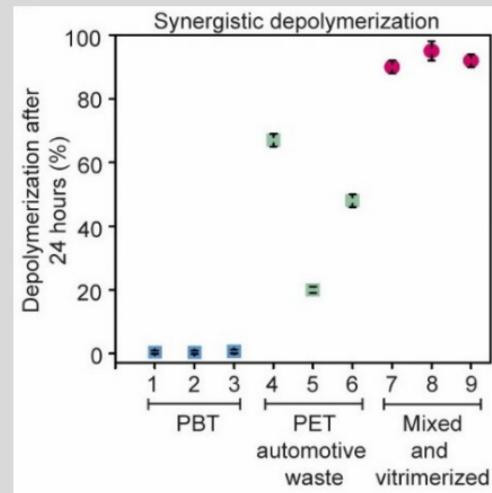
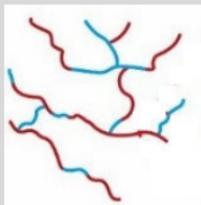
## Radical new idea: Mixing rather than sorting

**melt transesterification:** increase randomness along chains  
works for blends: 75%+25% of PET+PBT or of PET+PTT



## vitrimerisation:

cross-link chains to form a random network



## This was really a group effort:

---



Ludwik Leibler (Gulliver lab, CNRS)



Andrew Griffiths (Biochemistry lab, ESPCI)



Hernan Garate (SIMM lab, ESPCI/CNRS/PSL)



Yannick Rondelez (Gulliver lab, ESPCI)



... and me (Gulliver lab, CNRS)...

and Costantino Creton (SIMM lab, CNRS)

and Clément Freymond (SIMM lab, ESPCI)

and Louise Breloy (SIMM lab, ESPCI)

... plus industrial contacts ...