TABLE 6-5
The Complete Hydrogenlike Atomic Wave Functions for n = 1, 2, and 3. The Quantity The Complete Hydrogenlike Atomic Wave Functions for n = 1, 2, and 3. The Quantity Number of the Nucleus, and $\sigma = Zr/a_0$, Where a_0 is the Bohr Partity The Complete Hydrogenlike Atomic Values, and $\sigma = Zr/a_0$, Where a_0 is the Bohr Radius, Z Is the Atomic Number of the Nucleus, and $\sigma = Zr/a_0$, Where a_0 is the Bohr Radius,

he Atomic France
$$n = 1; l = 0, m = 0$$

$$\psi_{100} = \frac{1}{\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} e^{-\sigma}$$

$$n = 2; l = 0, m = 0$$

$$\psi_{200} = \frac{1}{\sqrt{32\pi}} \left(\frac{Z}{a_0}\right)^{3/2} (2 - \sigma) e^{-\sigma/2}$$

$$l = 1, m = 0$$

$$\psi_{210} = \frac{1}{\sqrt{32\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \sigma e^{-\sigma/2} \cos \theta$$

$$l = 1, m = \pm 1$$

$$\psi_{21\pm 1} = \frac{1}{\sqrt{64\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \sigma e^{-\sigma/2} \sin \theta e^{\pm i\phi}$$

$$n = 3; l = 0, m = 0$$

$$\psi_{300} = \frac{1}{81\sqrt{3\pi}} \left(\frac{Z}{a_0}\right)^{3/2} (27 - 18\sigma + 2\sigma^2) e^{-\sigma/3}$$

$$l = 1, m = 0$$

$$\psi_{310} = \frac{1}{81} \left(\frac{Z}{\pi}\right)^{1/2} \left(\frac{Z}{a_0}\right)^{3/2} (6\sigma - \sigma^2) e^{-\sigma/3} \cos \theta$$

$$l = 1, m = \pm 1$$

$$\psi_{31\pm 1} = \frac{1}{81\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} (6\sigma - \sigma^2) e^{-\sigma/3} \sin \theta e^{\pm i\phi}$$

$$l = 2, m = 0$$

$$\psi_{320} = \frac{1}{81\sqrt{6\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \sigma^2 e^{-\sigma/3} (3\cos^2 \theta - 1)$$

$$l = 2, m = \pm 1$$

$$\psi_{32\pm 1} = \frac{1}{81\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \sigma^2 e^{-\sigma/3} \sin \theta \cos \theta e^{\pm i\phi}$$

$$l = 2, m = \pm 2$$

$$\psi_{32\pm 2} = \frac{1}{162\sqrt{\pi}} \left(\frac{Z}{a_0}\right)^{3/2} \sigma^2 e^{-\sigma/3} \sin^2 \theta e^{\pm 2i\phi}$$

6-1

TABLE 0-0
The Complete Hydrogenlike Atomic Wave Functions Expressed as Real Functions for The Complete Hydrogenlike Atomic Number of the Nucleus and The Complete Hydrogeniake Atomic Number of the Nucleus and $\sigma = Zr/a_0$, n = 1, 2, and 3. The Quantity Z is the Atomic Number of the Nucleus and $\sigma = Zr/a_0$, Where a_0 is the Bohr Radius.

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$$a_0$$
 is since a_1 is since a_2 is since a_1 is a_2 is a_3 is a_4 in a_4