Fundamentals of Engineering Sample Examination **Morning Section**

1.	(A)	B	C	D	31.	(A)	B	(C)	D	61.	(A)	B	C	D	91.	A	B	C	D
2.	A	B		D	32.	A	B	©	D	62.	A	B	C	D	92.	A	B	C	D
3.	A	B	<u> </u>	D	33.	A	B	C	D	63.	A	B	C	D	93.	A	B	<u>C</u>	D
4.	A	B	C	D	34.	A	B	(C)	D	64.	A	\bigcirc B	C	D	94.	A	\bigcirc B	\bigcirc	D
5.	A	B	C	D	35.	A	B	C	D	65.	A	B	C	D	95.	A	\bigcirc B	C	D
6.	A	B	C	D	36.	A	B	C	D	66.	A	\bigcirc B	\bigcirc	D	96.	\bigcirc A	B	C	D
7.	A	\bigcirc B	\bigcirc	D	37.	\bigcirc A	B	\bigcirc	D	67.	\bigcirc A	\bigcirc B	C	D	97.	\bigcirc	\bigcirc B	\bigcirc	D
8.	A	B	C	D	38.	A	B	C	D	68.	\bigcirc A	B	C	D	98.	A	B	C	D
9.	A	B	C	D	39.	A	B	C	D	69.	A	\bigcirc B	C	D	99.	A	B	\bigcirc	D
10.	A	B	C	D	40.	A	\bigcirc B	C	D	70.	A	\bigcirc B	C	D	100	. (A)	B	(C)	D
11.	A	\bigcirc B	\bigcirc	D	41.	\bigcirc A	B	\bigcirc	D	71.	\bigcirc A	B	C	D	101	. (A)	B	C	D
12.	A	B	C	D	42.	\bigcirc A	\bigcirc B	\bigcirc	D	72.	A	B	C	D	102	. (A)	B	C	D
13.	A	\bigcirc B	\bigcirc	D	43.	\bigcirc A	B	C	D	73.	\bigcirc A	B	(C)	D	103	. (A)	B	C	D
14.	\bigcirc A	\bigcirc B	\bigcirc	D	44.	A	B	C	D	74.	\bigcirc A	B	C	D		. (A)	B	C	D
15.	A	\bigcirc B	\bigcirc	D	45.	\bigcirc A	B	0	D	75.	\bigcirc	B	C	D		. (A)	B	C	D
16.	\bigcirc A	\bigcirc B	C	D	46.	A	B	C	D	76.	A	B	©	D		. (A)	B	C	D
17.	A	\bigcirc B	\bigcirc	D	47.	\bigcirc A	B	C	D	77.	A	B	<u></u>	(D)		. (A)	B	©	D
18.	A	B	\bigcirc	D	48.	A	B	C	D	78.	A	B	C	D		. (A)	B	<u> </u>	D
19.	A	\bigcirc B	C	D	49.	A	B	C	D	79.	A	B	©	D		. (A)	B	C	D
20.	\bigcirc A	B	(C)	D	50.	A	B	C	D	80.	A	B	(C)	D _		. (A)	B	C	D
21.	A	\bigcirc B	\bigcirc	D	51.	A	B	C	D	81.	A	B	©	D		. (A)	B	©	D
22.	A	B	C	D	52.	A	B	C	D	82.	\bigcirc	(B)	C	D		. (A)	B		(D)
23.	A	B	C	D	53.	A	B	C	D	83.	A	B	©	(D)		. (A)	B	0	(D)
24.	A	B	C	D	54.	A	B	©	(D)	84.		B	0	D		. (A)	B	(0)	D
25.	A	B	C	D	55.	A	B	©	(D)	85.	A	B		(D)		. (A)	B	C	D
26.	A	B	C	D	56.	A	B	C	D	86.	A	B	©	(D)		. (A)	B		D
27.	A	B	C	D	57.	A	B	<u> </u>	(D)	87.		B		(D)		(. (A)	B	C	D
28.	A	B	C	D	58.	A	B	0	(D)	88.	A	B		(D)		. (A)	B	©	D
29.	A	B	C	D	59.	A	B	©	D	89.	A	B		(D)		. (A)	B	©	D (D)
30.	\bigcirc A	\bigcirc B	C	D	60.	\bigcirc A	B	\bigcirc	D	90.	A	B	C	D	120	. (A)	B	C	D

- 1. The half-life of a radioactive isotope is 4.3 days. Most nearly, how long will it take to reduce the original amount to 1%?
 - (A) 3.2 days
 - (B) 16 days
 - (C) 29 days
 - (D) 40 days
- **2.** What is the determinant of the following 2×2 matrix?

$$\begin{bmatrix} 1 & 4 \\ 3 & 2 \end{bmatrix}$$

- (A) -10
- (B) -5
- (C) 9
- (D) 10
- 3. What is most nearly the value of the following limit?

$$\lim_{x\to 3} \frac{x^2-9}{x-3}$$

- (A) -6
- (B) 1
- (C) 6
- (D) ∞
- 4. Given the vector V = i+2j+k, what is most nearly the angle between V and the x-axis?
 - (A) 22°
 - (B) 24°
 - (C) 66°
 - (D) 80°
- 5. Evaluate the following definite integral.

$$\int_{2}^{\infty} \frac{1}{x^2} dx$$

- (A) 1/24
- (B) 1/8
- (C) 1/2
- (D) 2
- **6.** Which is a true statement about the two vectors?

$$V_1 = i + 2j + k$$
$$V_2 = i + 3j - 7k$$

- (A) Both vectors pass through (0, -1, 6).
- (B) The vectors are parallel.
- (C) The vectors are orthogonal.
- (D) The angle between the vectors is 17.4° .

- 7. What is most nearly the area bounded by y = 0, $y = e^x$, x = 0, and x = 1?
 - (A) 1.4
 - (B) 1.7
 - (C) 2.7
 - (D) 3.4
- **8.** A function of x is given below. Which (x, y) point is a relative maximum or minimum?

$$y = \frac{1}{4}x^4 - 1.5x^2 + 2x + 5$$

- (A) (-2, -1)
- (B) (-2, -2)
- (C) (2,-2)
- (D) (-1, -1.75)
- 9. The slope of a line is $^{1}/_{2}$. The slope of a second line is $-^{2}/_{3}$. The lines intercept at the point (3,1). What is most nearly the acute angle between the lines?
 - (A) 27°
 - (B) 50°
 - (C) 60°
 - (D) 80°
- 10. A function is given. What value of x maximizes y?

$$y^2 + y + x^2 - 2x = 5$$

- (A) -1
- (B) 1/2
- (C) 1
- (D) 5
- 11. An elephant is chained to a corner of a 30 m \times 35 m building. If the chain is 40 m long and the elephant can reach 1 m farther than the chain length, what is the maximum area the elephant can cover?
 - (A) 3870 m^2
 - (B) 3960 m^2
 - (C) 3970 m^2
 - (D) 4080 m^2
- **12.** What is a point of inflection for the following equation?

$$y = 9x^3 + x^2 - 15x + 32$$

- (A) (-27, -176,000)
- (B) (-21.2, -84,400)
- (C) (-0.037, 32.6)
- (D) (19.2, 63,300)

6 FE/EIT Sample Examinations

13. Assume that three force vectors are applied at a single point.

$$\mathbf{F}_1 = \mathbf{i} + 3\mathbf{j} + 4\mathbf{k}$$

$$\mathbf{F}_2 = 2\mathbf{i} + 7\mathbf{j} - \mathbf{k}$$

$$\mathbf{F}_3 = -\mathbf{i} + 4\mathbf{j} + 2\mathbf{k}$$

What is most nearly the magnitude of the resultant force vector, \mathbf{R} ?

- (A) 13
- (B) 14
- (C) 15
- (D) 16
- 14. What is the solution to the differential equation if x = 1 at t = 0, and dx/dt = 0 at t = 0?

$$\frac{1}{2}\frac{d^2x}{dt^2} + 4\frac{dx}{dt} + 8x = 5$$

- (A) $e^{-4t} + 4te^{-4t}$
- (B) $\frac{3}{8}e^{-2t}(\cos 2t + \sin 2t) + \frac{5}{8}$
- (C) $e^{-4t} + 4te^{-4t} + \frac{5}{8}$
- (D) $\frac{3}{8}e^{-4t} + \frac{3}{2}te^{-4t} + \frac{5}{8}$
- **15.** What is the smallest positive value of y on the curve $y = 7x^2 3x + 8$?
 - (A) 3/14
 - (B) 3/7
 - (C) 14/3
 - (D) 215/28
- **16.** Which of the following is equivalent to $\sin 2\theta$?
 - (A) $2\sin\theta\cos\theta$
 - (B) $\cos^2\theta \sin^2\theta$
 - (C) $\sin\theta\cos\theta$
 - (D) $\frac{1 \cos 2\theta}{2}$
- 17. If $i \equiv \sqrt{-1}$, what is the value of $(i)^i$?
 - (A) i^2
 - (B) e^{2i}
 - (C) -1
 - (D) $e^{-\frac{\pi}{2}}$
- 18. What is most nearly the volume of the object created when the area bounded by y = 0, x = 0, and $y = \sqrt{4 x^2}$ is rotated about the y-axis?

- (A) 3.1
- (B) 8.4
- (C) 17
- (D) 34
- 19. Five fair coins are each flipped once. What is the probability that at least two of the coins will show heads?
 - (A) 0.19
 - (B) 0.80
 - (C) 0.81
 - (D) 0.84
- **20.** What is most nearly the standard deviation of 1, 4, and 7?
 - (A) 2.5
 - (B) 3.0
 - (C) 5.7
 - (D) 6.0
- 21. What is most nearly the probability of picking an orange ball out of a bag containing seven orange balls, eight green balls, and two white balls?
 - (A) 0.059
 - (B) 0.15
 - (C) 0.24
 - (D) 0.41
- 22. Five dice are thrown. The probability of rollling at least one six is most nearly
 - (A) 0.50
 - (B) 0.52
 - (C) 0.60
 - (D) 0.67
- **23.** What is most nearly the sample standard deviation for the data set $\{2.0, 7.0, 9.0, 12, 34\}$?
 - (A) 11
 - (B) 12
 - (C) 13
 - (D) 17
- 24. On average, a piece of machinery jams five times a week. Assuming a Poisson distribution for the jams, the probability that the machine will jam exactly three times in a given week is most nearly

- (A) 0.0033
- (B) 0.14
- (C) 0.33
- (D) 1.40
- 25. The least-squares method is used to plot a straight line through the data points (2,2), (4,7), (1,11), and (5,9). The slope of the line is most nearly
 - (A) -8
 - (B) -4
 - (C) 0
 - (D) 3.5
- **26.** The method of determining the graph of a curve that best approximates a given collection of points is called
 - (A) integration by parts
 - (B) curve fitting
 - (C) nonlinear interpolation
 - (D) hypothesis testing
- 27. Which of the following is the formula for acetic acid?
 - (A) COOH
 - (B) CH₂COOH
 - (C) CH₃CH₂COOH
 - (D) CH₃COOH
- 28. What is oxidized and what is reduced in the following reaction?

$$Zn + H_2SO_4 \rightarrow ZnSO_4 + H_2$$

- (A) Zinc is oxidized only.
- (B) Zinc is reduced only.
- (C) Zinc is oxidized and hydrogen is reduced.
- (D) Zinc is reduced and hydrogen is oxidized.
- 29. How many moles of NaOH will be neutralized by 1 mole of H₃PO₄?
 - (A) 1/3
 - (B) 1
 - (C) 2
 - (D) 3
- 30. An anode is
 - (A) the endpoint of a directed network
 - (B) an electrode at which oxidation occurs
 - (C) an electrode at which reduction occurs
 - (D) the electrode to which the cation would be attracted during an electrolytic reaction

- 31. Beryllium, magnesium, and calcium all belong to which elemental grouping?
 - (A) noble elements
 - (B) halogens
 - (C) alkali metals
 - (D) alkaline earth metals
- 32. If 1 mol of gaseous chlorine combines with 1 mol of calcium to form calcium chloride (CaCl₂), which of the following statements is true?
 - (A) Calcium has a valence of +1.
 - (B) 1 lbm of chlorine will combine with 1 lbm of calcium.
 - (C) Chlorine has a valence of -2.
 - (D) Calcium has a valence of +2.
- **33.** Balance the following equation.

$$\underline{\hspace{0.5cm}}\operatorname{PbO}_2 + \underline{\hspace{0.5cm}}\operatorname{H}_2\operatorname{SO}_4 + \underline{\hspace{0.5cm}}\operatorname{Pb} \rightleftharpoons \underline{\hspace{0.5cm}}\operatorname{H}_2\operatorname{O} + \underline{\hspace{0.5cm}}\operatorname{PbSO}_4$$

- (A) $3PbO_2 + 6H_2SO_4 + 3Pb \rightleftharpoons 6H_2O + 5PbSO_4$
- (B) $3PbO_2 + 6H_2SO_4 + Pb \rightleftharpoons 6H_2O + 4PbSO_4$
- (C) $2\text{PbO}_2 + \text{H}_2\text{SO}_4 + 2\text{Pb} \rightleftharpoons 4\text{H}_2\text{O} + 4\text{PbSO}_4$
- (D) $PbO_2 + 2H_2SO_4 + Pb \rightleftharpoons 2H_2O + 2PbSO_4$
- **34.** What is the most likely formula of a compound with the following gravimetric analysis?

oxygen: 13.7% carbon: 20.5% hydrogen: 5.1% chlorine: 60.7%

- (A) CH₃OCl
- (B) C_2H_6OCl
- (C) $C_2H_6OCl_2$
- (D) CH_6O_2Cl
- 35. How many moles of hydrochloric acid will be required to neutralize 1 mol of sodium hydroxide?
 - (A) 0.3 mol
 - (B) 0.5 mol
 - (C) 1.0 mol
 - (D) 1.5 mol
- **36.** What is most nearly the mass of 1 atom of carbon-12?
 - (A) 5.0×10^{-24} g
 - (B) 2.0×10^{-23} g
 - (C) 4.0×10^{-23} g
 - (D) 7.6×10^{-23} g

8 FE/EIT Sample Examinations

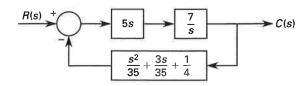
37. A system is found to have the following transfer function.

$$H(s) = \frac{7+s}{2s^2 + 14s - 36}$$

s is the frequency in Hz. What frequency defines the system as inherently unstable?

- (A) -7 Hz
- (B) 2 Hz
- (C) 4 Hz
- (D) 9 Hz

Problems 38–43 are based on the following illustration. All values are in non-dimensionalized units.



- **38.** What is most nearly the natural frequency of the control system?
 - (A) 0.707
 - (B) 3.12
 - (C) 5.92
 - (D) 9.75

Problems 39–42 assume that the natural frequency of the control system shown in Prob. 38 is 5q Hz, where q is some scalar.

- **39.** What is most nearly the steady-state gain of the control system?
 - (A) $1.4/q^2$
 - (B) $7.0/q^2$
 - (C) $16/q^2$
 - (D) $180/q^2$
- **40.** What is most nearly the damping ratio of the control system?
 - (A) 0.06/q
 - (B) 0.3/q
 - (C) 0.6/q
 - (D) 0.8/q

Problems 41 and 42 assume a damping ratio of 0.5.

41. What is most nearly the damped natural frequency (in Hz) of the control system?

- (A) 0.866q
- (B) 3.54q
- (C) 3.75q
- (D) 4.33q
- **42.** What is most nearly the damped resonant frequency (in Hz) of the control system?
 - (A) 0.866q
 - (B) 3.54q
 - (C) 3.75q
 - (D) 4.33q
- **43.** If the control system is given a unit step input, how will the response be characterized?
 - (A) underdamped
 - (B) overdamped
 - (C) critically damped
 - (D) unstable

44. A baud is

- (A) a measure of the intervals between transmitted digital data
- (B) a device used to demodulate audio data signals
- (C) a unit of magnitude of electronically transmitted audio signals
- (D) a unit of speed in digital data transmission measuring number of signals per second
- **45.** The following code is an example of what programming technique?

- (A) branching
- (B) function call
- (C) looping
- (D) subroutine
- **46.** When can a professional provide services if conflicts of interest are involved?
 - (A) if no compensation is received
 - (B) if doing so does not personally benefit the professional
 - (C) if full disclosure of potential conflicts are provided
 - (D) never

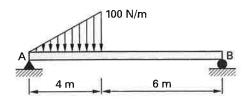
- **47.** When is the receipt of compensation from more than one party for the same project ethical?
 - (A) never
 - (B) when the total of all compensation received does not exceed "reasonable" compensation for the type of work performed
 - (C) always
 - (D) when all parties involved know about and agree to the arrangement
- 48. When may professionals make political donations?
 - (A) at no time
 - (B) if it is not for current, past, or future influence or favors
 - (C) if all donations are made as an individual and do not represent a firm or entity
 - (D) only below specified amounts
- **49.** Which of the following is not always an ethics violation?
 - (A) signing plans or blueprints without having first designed and/or checked the plans
 - (B) revealing confidential information about a product without first obtaining permission
 - (C) granting a contract to a company for which the professional is an officer while concurrently serving on the board issuing the grants
 - (D) any individual accepting fees from contractors hired for a project
- **50.** In the event of an ethical conflict, to whom does the professional hold the least ethical responsibility?
 - (A) the employer
 - (B) the client
 - (C) the consumer
 - (D) society
- 51. Ethical behavior is officially regulated by whom?
 - (A) individual employers
 - (B) registered engineers
 - (C) professional societies
 - (D) state enforcement agencies
- **52.** Whistle blowing is
 - (A) an ethical practice
 - (B) an illegal practice
 - (C) an unethical practice
 - (D) a career-enhancing practice

- **53.** Which of the following is not a reason professionals must adhere to a code of ethics?
 - (A) Professions are self-regulating.
 - (B) Professionals receive above-average compensation.
 - (C) Professionals have autonomy.
 - (D) Professions are responsible for training other professionals.
- **54.** The National Society of Professional Engineers' (NSPE) Code of Ethics addresses competitive bidding. Which of the following is NOT stipulated?
 - (A) Engineers and their firms may refuse to bid competitively on engineering services.
 - (B) Clients are required to seek competitive bids for design services.
 - (C) Federal laws governing procedures for procuring engineering services (e.g., competitive bidding) remain in full force.
 - (D) Engineers and their societies may actively lobby for legislation that would prohibit competitive bidding for design services.
- **55.** A credit card offers 1.2% effective monthly interest. What is most nearly the effective annual rate with monthly compounding?
 - (A) 7.9%
 - (B) 8.9%
 - (C) 14%
 - (D) 15%
- **56.** A bank charges 12% simple interest on a \$300 loan. Most nearly, how much will be repaid if the loan is paid back in one lump sum after three years?
 - (A) \$108
 - (B) \$408
 - (C) \$415
 - (D) \$421
- **57.** Most nearly, how long will it take a sum of money to double at a 5% annual percentage rate?
 - (A) 6 years
 - (B) 10 years
 - (C) 11 years
 - (D) 14 years
- **58.** A piece of machinery has an initial cost of \$40,000 and results in an increase in annual maintenance costs of \$2000. If the machinery saves the company \$10,000

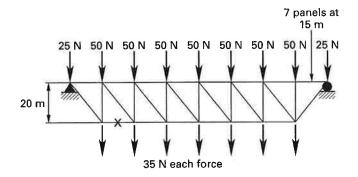
per year, in approximately how many years will the machine pay for itself if compounding is considered? The effective annual interest rate is 6%.

- (A) 5.2 years
- (B) 6.1 years
- (C) 7 years
- (D) 8 years
- **59.** Funds are deposited in a savings account at an interest rate of 8% per annum. If the interest is compounded semi-annually, what is most nearly the initial amount that must be deposited to yield a total of \$10,000 in 10 years?
 - (A) \$4530
 - (B) \$4560
 - (C) \$6730
 - (D) \$8200
- **60.** \$500 is deposited into a bank savings account with 6% interest compounded annually. Most nearly how much is in the account at the end of three years?
 - (A) \$550
 - (B) \$600
 - (C) \$650
 - (D) \$700
- **61.** At the end of each year for five years, \$500 is deposited into a credit union account. The credit union pays 5% interest compounded annually. At the end of five years (immediately following the fifth deposit) most nearly how much will be in the account?
 - (A) \$640
 - (B) \$1750
 - (C) \$2760
 - (D) \$3550
- **62.** On January 1, \$5000 is deposited into a high-interest savings account that pays 8% interest compounded annually. If all of the money is withdrawn in five equal end-of-year sums beginning December 31 of the first year, most nearly how much will each withdrawal be?
 - (A) \$1008
 - (B) \$1150
 - (C) \$1210
 - (D) \$1250
- **63.** If you needed to have \$800 in savings at the end of four years and your savings account yielded 5% interest paid annually, most nearly how much would you need to deposit today?

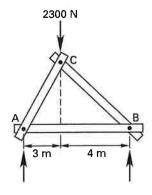
- (A) \$570
- (B) \$600
- (C) \$660
- (D) \$770
- **64.** What is most nearly the reaction at A?



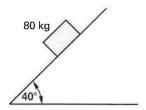
- (A) 50 N
- (B) 100 N
- (C) 130 N
- (D) 150 N
- **65.** Find the magnitude of the force in the member marked with an "X." All members are pin-connected.



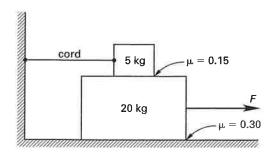
- (A) 9.70 N
- (B) 27.3 N
- (C) 85.0 N
- (D) 223 N
- **66.** The approximate vertical force component in member BC is



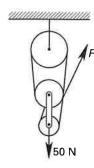
- (A) 990 N
- (B) 1300 N
- (C) 2300 N
- (D) 3600 N
- **67.** What is most nearly the frictional force between the 80 kg block and the ramp? The coefficient of static friction is 0.2, and the coefficient of dynamic friction is 0.15.



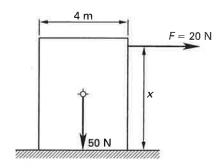
- (A) 60 N
- (B) 80 N
- (C) 90 N
- (D) 120 N
- **68.** Force F is gradually increased until the 20 kg block begins moving to the right. The 5 kg block is prevented from moving by a cord. What is most nearly the minimum force F for which movement is possible?



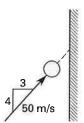
- (A) 7.4 N
- (B) 59 N
- (C) 74 N
- (D) 81 N
- **69.** Most nearly, what force, F, is required to lift a 50 N load? All pulleys are frictionless. Assume all strands are parallel.



- (A) 8.3 N
- (B) 13 N
- (C) 17 N
- (D) 25 N
- **70.** What is most nearly the maximum value of x such that F can be applied without tipping the block? ($\mu = 0.4$.)



- (A) 1.5 m
- (B) 3.5 m
- (C) 4.4 m
- (D) 5.0 m
- 71. What is most nearly the component of velocity perpendicular to the wall after impact if the coefficient of restitution is 0.8?



- (A) 24 m/s
- (B) 30 m/s
- (C) 32 m/s
- (D) 40 m/s
- **72.** The velocity of a particle at time t is

$$\mathbf{v}(t) = 12t^4 + \frac{7}{t}$$

Most nearly, what total distance is traveled between t=0.2 and t=0.3?

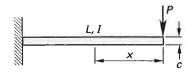
- (A) 0.98
- (B) 1.8
- (C) 2.8
- (D) 8.4
- **73.** A spring has a spring constant of 10 N/cm. It is compressed 5 cm. The spring is released and pushes against a free projectile with a mass of 1 kg. The projectile velocity immediately after losing contact with the spring is most nearly
 - (A) 0.32 m/s
 - (B) 1.6 m/s
 - (C) 32 m/s
 - (D) 50 m/s
- **74.** A rocket is moving through a vacuum. It changes its velocity from 9020 m/s to 5100 m/s in 48 s. The power required to accomplish this if the rocket's mass is 213 000 kg is most nearly
 - (A) 34 GW
 - (B) 120 GW
 - (C) 170 GW
 - (D) 250 GW
- 75. A projectile is launched at 52 degrees from horizontal with an initial velocity of 3600 m/s. If the mass of the projectile is 32 kg, what is most nearly the total kinetic and potential energy possessed by the projectile at t=13 s? Neglect all forms of friction.
 - (A) 5.9 kJ
 - (B) 0.58 MJ
 - (C) 210 MJ
 - (D) 420 MJ

Problems 76 and 77 are based on the following statement.

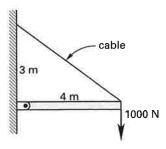
A 30 cm long rod ($E=3\times10^7$ N/cm², $\alpha=6\times10^{-6}$ cm/cm·°C) with a 2 cm² cross section is fixed at both ends.

- **76.** If the rod is heated to 60°C above the neutral temperature, what is most nearly the stress?
 - (A) 110 N/cm^2
 - (B) $11\,000\ N/cm^2$
 - (C) $36\,000\,\mathrm{N/cm^2}$
 - (D) 57 000 N/cm²

- **77.** If one end of the rod is free to expand the elongation is most nearly
 - (A) 5.4×10^{-4} cm
 - (B) 3.6×10^{-4} cm
 - (C) 0.01 cm
 - (D) 0.03 cm
- 78. Vickers, Knoop, and Brinell are all names of
 - (A) Nobel prize winners in metallurgy
 - (B) thermodynamic constants
 - (C) hardness tests
 - (D) chi-squared statistics
- **79.** What is the maximum flexural stress at a distance x from the free end of a cantilever beam supporting a tip load, P?



- (A) $\frac{Pxc}{2I}$
- (B) $\frac{Pc}{2I}$
- (C) $\frac{PcL}{2I}$
- (D) $\frac{Pxc}{2EI}$
- 80. What is most nearly the elongation in the cable if F=1000 N? The cable's effective cross-sectional area is $2~\rm cm^2$. Its modulus of elasticity is 1.5×10^6 N/cm².

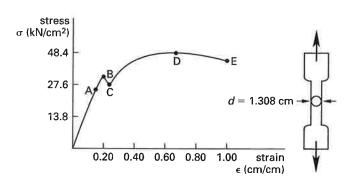


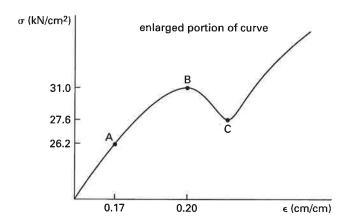
- (A) 0.0028 cm
- (B) 0.14 cm
- (C) 0.28 cm
- (D) 0.56 cm

- 81. A simply-supported beam carries a single concentrated load at its center. If its slenderness ratio is 150, it is most likely to fail
 - (A) where the moment is zero
 - (B) where the shear is maximum
 - (C) at a support
 - (D) where the slope of the deflection curve is zero

Problems 82 and 83 are based on the following statement and illustrations.

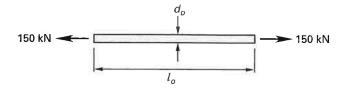
The results of a tensile test on a round specimen of a given material are shown.





- 82. What is most nearly the yield stress?
 - (A) 26 kN/cm^2
 - (B) 28 kN/cm^2
 - (C) 29 kN/cm²
 - (D) 31 kN/cm^2
- **83.** What is most nearly the elastic limit of the material?
 - (A) 26 kN/cm^2
 - (B) 28 kN/cm^2
 - (C) 31 kN/cm^2
 - (D) 34 kN/cm^2

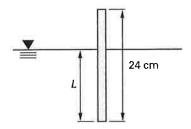
84. The aluminum rod shown in the following illustration has an initial diameter, d_o , of 30 mm and an initial gauge length, ℓ_o , of 200 mm. The yield strength is 420 MPa. What is the modulus of elasticity if a force of 150 kN elongates the rod by 1.0 mm?



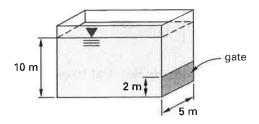
- (A) 21 GPa
- (B) 31 GPa
- (C) 42 GPa
- (D) 53 GPa
- **85.** Which type of hardening will work to some extent in all metals?
 - (A) work-hardening
 - (B) annealing
 - (C) martempering
 - (D) austenitizing
- **86.** What do impact tests determine?
 - (A) hardness
 - (B) yield strength
 - (C) toughness
 - (D) creep strength
- 87. What is one of the main differences between cast iron and steel?
 - (A) Steel has a lower carbon content.
 - (B) Steel always contains alloying metals such as nickel, chromium, manganese, and vanadium.
 - (C) Steel cannot be annealed, whereas cast iron can.
 - (D) Steel contains a large amount of uncombined carbon.
- 88. For a fixed curing time, the ultimate strength of concrete
 - (A) increases with a decrease in water content
 - (B) decreases with a decrease in water content
 - (C) is independent of water content if cured for a sufficiently long time
 - (D) is independent of curing pressure

- 89. The radius of a hypothetical electron orbit is known to be 0.75 Å. What is most nearly the de Broglie wavelength of the electron if four complete cycles constitute a stable pattern around the nucleus?
 - (A) 0.19 Å
 - (B) 1.2 Å
 - (C) 2.4 Å
 - (D) 4.7 Å
- 90. An orbital
 - (A) may have 2, 8, 18, or 32 electrons
 - (B) may have 2 electrons with the same spin direction
 - (C) may be photographed with an electron microscope
 - (D) may be unoccupied
- **91.** For corrosion to occur, which of the following items must be present?
 - I. anode
 - II. cathode
 - III. electrolyte
 - (A) I and II
 - (B) I and III
 - (C) II and III
 - (D) I, II, and III
- **92.** A system consisting of an open bucket containing a mixture of ice and water is to be warmed from 0°C to 20°C. How many degrees of freedom does the system have?
 - (A) 0
 - (B) 1
 - (C) 2
 - (D) 3
- 93. A venturi meter is used to measure air velocity. A one-fifth scale model of the venturi meter is built, and water is used as the test fluid. Viscosity of the air is $1.82 \times 10^{-5} \ \mathrm{N \cdot s/m^2}$. Viscosity of the water is $9.82 \times 10^{-4} \ \mathrm{N \cdot s/m^2}$. What will be the approximate ratio of the model to the actual velocities observed?
 - (A) 0.32
 - (B) 3.1
 - (C) 11
 - (D) 54
- **94.** A 150 m long surface vessel is modeled at 1:50. Most nearly, what speed must the model travel if a 40 kph similarity is desired?

- (A) 0.22 m/s
- (B) 1.5 m/s
- (C) 1.6 m/s
- (D) 2.2 m/s
- 95. What is most nearly the hydraulic radius of an equilateral triangle (vertex down) open channel flowing at full capacity with a maximum depth of 3 m?
 - (A) 0.60 m
 - (B) 0.65 m
 - (C) 0.70 m
 - (D) 0.75 m
- **96.** A 24 cm long rod floats vertically in water. It has a 1 cm² cross section and a specific gravity of 0.6. Most nearly, what length, L, is submerged?



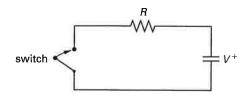
- (A) 9.6 cm
- (B) 14 cm
- (C) 18 cm
- (D) 19 cm
- **97.** A tank is filled with water to a depth of 10 m. The total force on the gate is most nearly



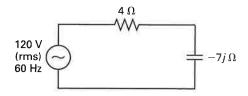
- (A) 90 kN
- (B) 440 kN
- (C) 880 kN
- (D) 980 kN
- **98.** When a thin-bore, hollow glass tube is inserted into a container of mercury, the surface of the mercury in the tube

- (A) is level with the surface of the mercury in the container
- (B) is below the container surface due to cohesion
- (C) is below the container surface due to adhesion
- (D) is above the container surface due to cohesion
- **99.** The fluid in a manometer tube is 60% water and 40% alcohol (specific gravity = 0.8). What is most nearly the manometer fluid height difference if a 42.7 kPa pressure difference is applied across the two ends of the manometer?
 - (A) 23 cm
 - (B) 47 cm
 - (C) 470 cm
 - (D) 550 cm
- 100. Carbon dioxide (CO_2) gas has a molecular weight of 44. The density of STP air is 1.29 kg/m³. The specific gravity of CO_2 gas at 66°C and 138 kPa, using STP air as reference, is most nearly
 - (A) 0.67
 - (B) 1.1
 - (C) 1.7
 - (D) 2.1
- 101. A vacuum pump is used to drain a basement of 20°C water. The vapor pressure of water at this temperature is 2.34 kPa. The pump is incapable of lifting water higher than 10.5 m. The atmospheric pressure is most nearly
 - (A) 100 kPa
 - (B) 150 kPa
 - (C) 210 kPa
 - (D) 270 kPa
- **102.** Six coulombs of charge pass through a wire in 2 s. What is most nearly the average current flowing?
 - (A) 1.6 A
 - (B) 3 A
 - (C) 4.8 A
 - (D) 6 A
- 103. An ideal transformer has 200 primary turns and 20 secondary turns. What is most nearly the secondary voltage if the primary voltage is 120 V?
 - (A) 1.2 V
 - (B) 12 V
 - (C) 120 V
 - (D) 1200 V

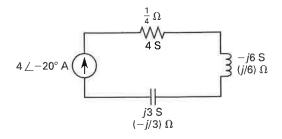
- 104. If 0.3 A flows in the secondary and 30 A flows in the primary of a perfectly matched, ideal transformer, what is most nearly the primary to secondary turns ratio?
 - (A) 1:100
 - (B) 1:10
 - (C) 10:1
 - (D) 100:1
- 105. Five watts are dissipated in a primary (input) circuit that includes a perfectly matched, ideal transformer with a primary-to-secondary turns ratio of 15:1. If the input resistance is 2000 Ω , what is most nearly the load resistance?
 - (A) 8.9Ω
 - (B) 130 Ω
 - (C) $6.2 \text{ k}\Omega$
 - (D) $30 \text{ k}\Omega$
- **106.** What is the total energy dissipated in the resistor when the capacitor discharges?



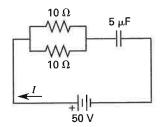
- (A) V^2R
- (B) $\frac{1}{2}CV$
- (C) CV
- (D) $\frac{1}{2}CV^2$
- 107. What is most nearly the phase angle difference between the current and the voltage? Take the voltage as the reference.



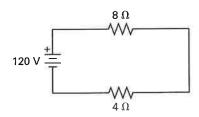
- (A) -60°
- (B) -35°
- (C) -30°
- (D) $+30^{\circ}$
- **108.** What is most nearly the phasor voltage drop across the current source?



- (A) 0.80\(\perp16.8\circ\) V
- (B) 1.2∠-53.7° V
- (C) 5.0\(\angle 36.8\circ V\)
- (D) $6.0\angle -20.0^{\circ} \text{ V}$
- 109. Two 10 Ω resistances are connected in parallel. This combination is connected in series with a capacitor of 5 μ F. The circuit is connected across a DC source voltage of 50 V. What is most nearly the maximum steady-state current through the battery?



- (A) 0
- (B) 1 A
- (C) 5 A
- (D) 7 A
- 110. The voltage appearing across the 4 Ω resistor is msot nearly



- (A) 27 V
- (B) 40 V
- (C) 80 V
- (D) 120 V
- 111. A series RLC circuit driven by an AC voltage contains reactances $X_L = 20 \Omega$ and $X_C = 14 \Omega$ and resistance $R = 10 \Omega$. What is most nearly the impedance at resonant frequency?

- (A) 10Ω
- (B) 14 Ω
- (C) 20 Ω
- (D) 24Ω
- 112. What are the changes in internal energy for reversible adiabatic and isothermal processes, respectively?
 - (A) $C_p \Delta T$ and 0
 - (B) 0 and $C_{\rm v}\Delta T$
 - (C) $C_{\mathbf{v}}\Delta T$ and $C_{\mathbf{p}}\Delta T$
 - (D) $C_{\mathbf{v}}\Delta T$ and 0
- 113. A steam engine operates between 150°C and 550°C. What is most nearly its theoretical maximum thermal efficiency?
 - (A) 27%
 - (B) 49%
 - (C) 73%
 - (D) 95%
- 114. Regardless of the process, the change in enthalpy (ΔH) for n moles of an ideal gas is
 - (A) $\frac{n\overline{R}T}{V}$
 - (B) 0
 - (C) $nC_{\rm v}\Delta T$
 - (D) $nC_p\Delta T$
- 115. Which of the following statements is true?
 - (A) Entropy always decreases.
 - (B) Entropy increases up to the critical temperature, and then it decreases.
 - (C) Theoretically, entropy approaches zero as the temperature approaches zero.
 - (D) None of the above statements are true.
- 116. When is the equation TdS = pdV + dU valid?
 - (A) only in constant temperature processes
 - (B) only in constant pressure processes
 - (C) only in reversible processes
 - (D) always
- 117. Which of the following statements is true for a perfect gas flowing through an insulated valve?
 - (A) Enthalpy is essentially unchanged.
 - (B) Entropy decreases.
 - (C) Temperature increases greatly.
 - (D) Flow is isentropic.

- 118. If an initial volume of an ideal gas is compressed to one-half its original volume and to twice its original temperature, the pressure
 - (A) remains constant
 - (B) doubles
 - (C) quadruples
 - (D) halves
- 119. If an initial volume of saturated steam is expanded isothermally to twice the initial volume, the pressure
 - (A) decreases
 - (B) increases
 - (C) halves
 - (D) doubles

- **120.** Which of the following is true of an adiabatic process?
 - (A) It allows heat transfer into the system but not out of the system.
 - (B) It may be reversible.
 - (C) It is one in which enthalpy remains unchanged.
 - (D) It is one in which the equation W = Q is valid.