

# **1,3,5-Triazine-Functionalized Thermally Activated Delayed Fluorescence Emitters for Organic Light-Emitting Diodes**

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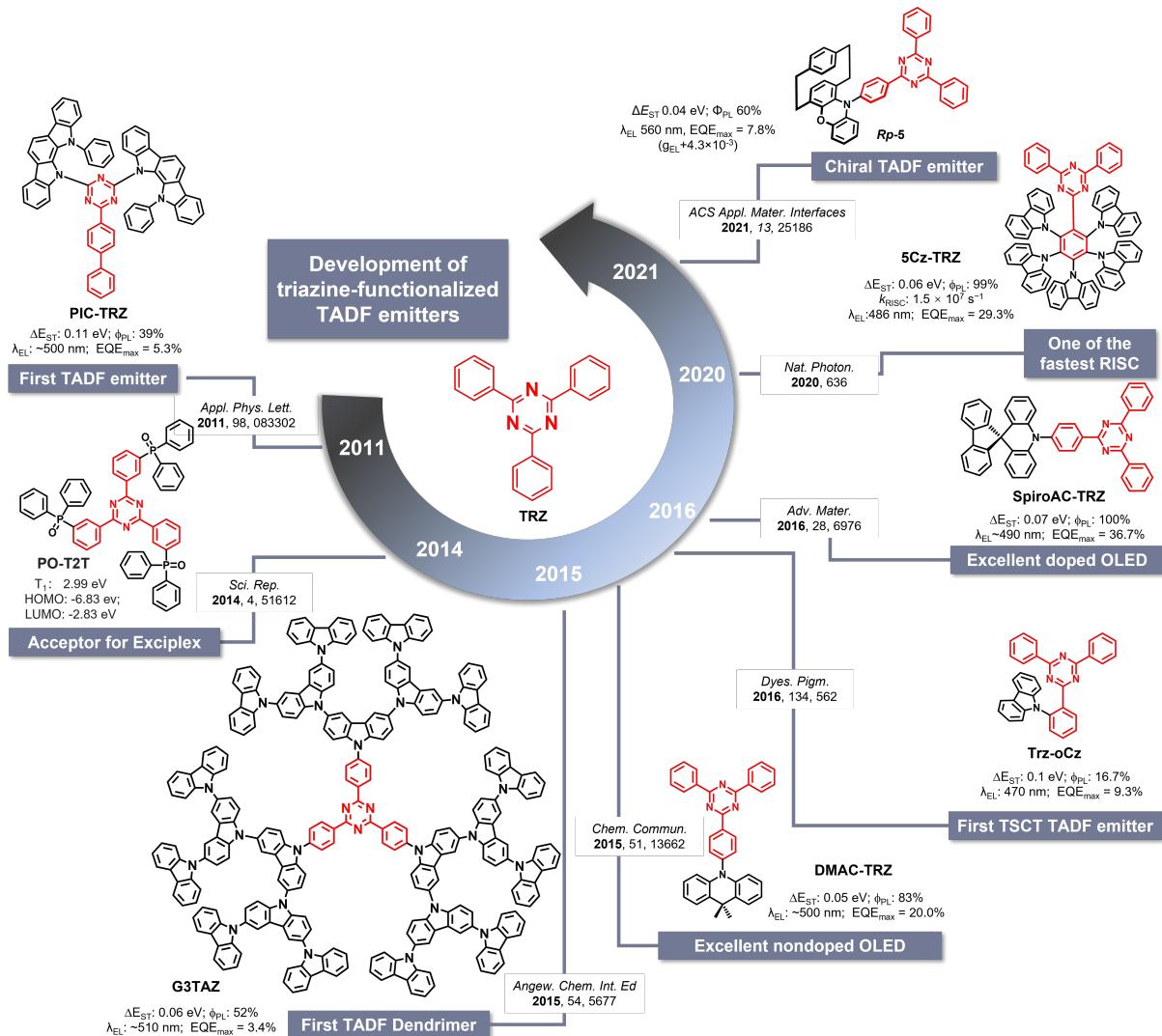
Keywords: triazine, TADF, OLED, carbazole, emitter

The 1,3,5-triazine electron-acceptor has become one of the most popular building blocks for the design of thermally activated delayed fluorescence (TADF) materials. Many TADF design strategies are first applied in compounds that contain triazines, and there are numerous examples of OLEDs with triazine-containing emitters that show high efficiencies and long operating lifetimes. We provide a comprehensive review of triazine-containing TADF emitters. This review is organized according to the triazine-derived structural motifs, such as number and position of electron-donor groups in donor-acceptor type emitters, the  $\pi$ -bridging linkers employed, and the design of through-space charge-transfer (TSCT) emitter. We correlate the structure of the compounds with their optoelectronic properties and the corresponding performance of the organic light-emitting diode devices.

## **1. Introduction**

Since 1987, when the first viable electroluminescent device was reported by Tang and Van Slyke,<sup>[1]</sup> significant progress has been made in terms of materials development such that organic light-emitting diodes (OLEDs) have now been commercialized in both the solid-state lighting and display markets. The current mature OLED products heavily rely on the phosphorescent

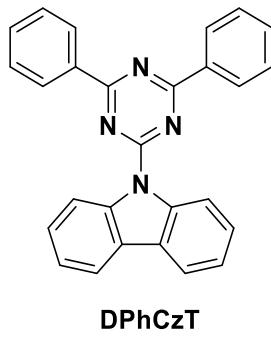
emitters that contain noble metals such as iridium because these materials permit 100% utilization of the electrogenerated singlet and triplet excitons to produce light. The breakthrough of thermally activated delayed fluorescence (TADF) materials in 2012 by Adachi and co-workers<sup>[2]</sup> provides an alternative route to reach equally efficient OLEDs, and without the use of these scarce metals. To date, thousands of TADF materials have been reported and their use in red, green and blue OLEDs has been demonstrated. Importantly, there are now many examples of devices that show high external quantum efficiency (EQE), device stability and color purity, these reports demonstrate that TADF OLEDs are attractive alternatives to phosphorescent devices. Indeed, the first commercialized TADF OLED devices are now on the market.<sup>[3]</sup> For instance, one of the best red TADF OLED, which uses **TPA-PZCN** as the emitter, has realized a record high maximum EQE ( $\text{EQE}_{\max}$ ) of 28.1% with an electroluminescence maximum,  $\lambda_{\text{EL}}$ , of 648 nm.<sup>[4]</sup> For green OLEDs, the use of **CzDBA** as the emitter resulted in a device with an  $\text{EQE}_{\max}$  of 37.8% with the  $\lambda_{\text{EL}}$  at 528 nm.<sup>[5]</sup> Blue OLED employing **TDBA-DI** as the emitter showed a comparable high  $\text{EQE}_{\max}$  of 38.2% with CIE coordinates of (0.15, 0.28),<sup>[6]</sup> and an OLED using a multiple resonance TADF (MR-TADF) emitter, **v-DABNA**, achieved an  $\text{EQE}_{\max}$  of 34.4% with a  $\lambda_{\text{EL}}$  at 469 nm and a full-width at half-maximum (FWHM) of only 18 nm.<sup>[7]</sup>



**Figure 1.** Timeline of the development of 1,3,5-triazine-based TADF materials for OLEDs

All these achievements in terms of device performance are inextricably linked to the development of high-performance highly twisted donor-acceptor emitter architectures. The vast majority of electron-donor groups used within TADF emitters are hole-transporting *N*-heterocycles and are typically chosen from carbazole and its derivatives (e.g., bicarbazole, benzofurocarbazole, thienocarbaazole, indolocarbazole), arylamine, acridan, phenoxazine, phenothiazine, and phenazine, and their derivatives. There is more structural diversity in the electron-acceptor moiety, with popular motifs containing borane, sulfone, ketone, pyrimidine, benzonitrile, phthalonitrile, triazole, oxadiazole, thiadiazole, benzothiazole, benzoxazoles,

quinoxaline, anthraquinone, heptazine, and of course, triazine. In particular, 1,3,5-triazine has been one of the most popular electron-deficient heteroaromatic acceptors used in green and blue TADF emitters and is also a popular component used in host materials design due to its moderate electron affinity with the LUMO values in the range of -2.7 to 3.1 eV.<sup>[8]</sup> Most commonly, the **1,3,5-triazine** is decorated by three phenyl groups attached to the 2, 4, 6-position of triazine in a triphenyltriazine structure (TRZ). There are also examples of triazine-containing materials where the electron-donor is directly attached to the triazine. However, in many of these materials, as there are negligible steric interactions between the triazine and the donor, thus leading to a predominantly co-planar conformation, there is significant overlap between the highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO), thereby resulting in a correspondingly large singlet and triplet energy splitting ( $\Delta E_{ST}$ ) and no TADF. For example, **DPhCzT**,<sup>[9]</sup> a compound containing a carbazole directly coupled to a diphenyltriazine, was reported as a host material, but did not show any TADF properties due to its large  $\Delta E_{ST}$  of 0.3 eV (**Figure 2**).

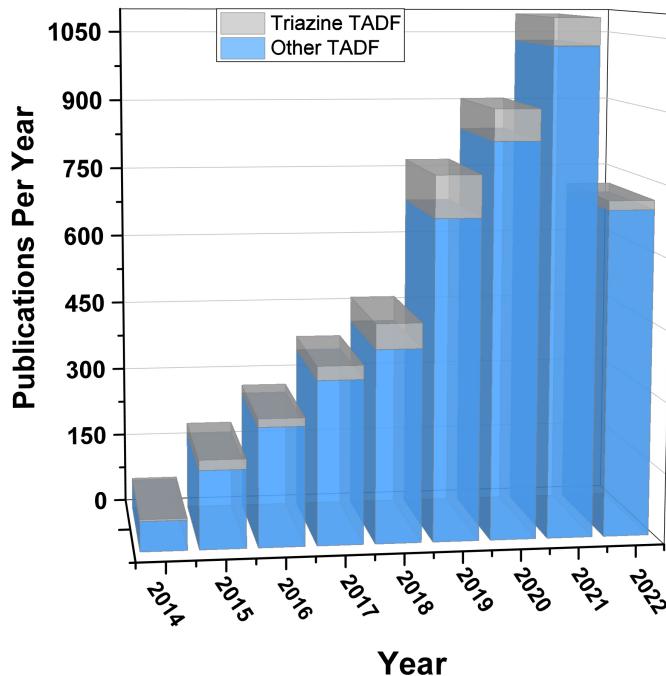


**Figure 2.** Chemical structure of **DPhCzT**

The genesis of organic TADF materials as emitters for OLEDs was in 2011 and relied on a triazine-based compound (**Figure 1**). A sufficient separation between HOMO and LUMO was induced by strengthening the electron-donor by extending the conjugation length as was done in **PIC-TRZ** ( $\Delta E_{ST} = 0.11$  eV), the first example of a purely organic TADF emitter. The design of **PIC-TRZ** also paved the way for the adoption of one of the most successful design strategies

for TADF emitters, which is based on a strongly twisted donor- $\pi$ -acceptor motif. Exciplex systems are another route to small  $\Delta E_{\text{ST}}$  as the weak intermolecular interactions between electron-donor and electron-acceptor compounds ensure suitable physical separation of the HOMO and LUMO. Triazine-containing electron-acceptors appear prominently in exciplex systems due to their largely planar conformation. **PO-T2T**, for instance, has been widely used as the acceptor for exciplex since it was first reported in 2014.<sup>[10]</sup> Triazine electron-acceptors were incorporated into some of the first TADF dendrimer emitters in 2015, such as **G3TAZ**, this due in part to the  $D_3$ -symmetric structure of the TRZ and the ease with which chemical modification of this core can occur. High-efficiency non-doped OLEDs ( $\text{EQE}_{\text{max}} = 20\%$ ) using the small molecule emitter **DMAC-TRZ** showed comparable performance to the doped device, indicating that it is possible to design a compound that shows significantly suppressed concentration quenching in neat films. **DMAC-TRZ** is still one of the most widely studied donor-acceptor type TADF emitters. Disposing the electron-donor adjacent to the TRZ results in improved performance in the OLED, as exemplified by **TRZ-oCz** ( $\text{EQE}_{\text{max}} = 9.3\%$ ). This is partially due to the more strongly twisted conformation that must be adopted for this compound and the additional through space charge transfer (TSCT) state that is induced. Since the first example in 2017, TRZ-containing TSCT TADF emitters have developed rapidly. Presently, the highest performing TRZ-based TADF emitter is **SpiroAC-TRZ**. Compared to the OLED with **DMAC-TRZ**, the  $\text{EQE}_{\text{max}}$  of the OLED with **SpiroAC-TRZ** is improved to 36.7%. Here, by employing a related electron-donor in 10H-spiro[acridine-9,9'-fluorene] (**SpiroAC**), a highly horizontally orientated transition dipole moment (TDM) was achieved, leading to enhanced light outcoupling.<sup>[11]</sup> Recently, a triazine-containing TADF emitter **5Cz-TRZ** with five carbazoles closely packed onto a central phenylene bridge afforded a material with one of the fastest reverse intersystem crossing (RISC) rate constants of  $k_{\text{RISC}} = 1.5 \times 10^7 \text{ s}^{-1}$ .

There are several advantages for the use of triazine within TADF materials. These include: (1) triazine and its derivatives are easy to synthesize from inexpensive starting materials; (2) triazine is easily functionalized at the 2,4,6 positions, and can be done both symmetrically or asymmetrically, thereby increasing structural diversity; (3) the linkage between the electron-donor and the triazine core can be rationally adjusted; (4) triazine is a stable aromatic structure that contributes to improved operational lifetimes of the devices. Documenting the popularity of this moiety, in this review, we summarize the recent progress of triazine-containing TADF materials. First, the TADF emitters based on carbazole-triazine structures are introduced and classified according to: i) the number and substitution position of the carbazole; ii) the influence of the bridge between the donor and the triazine acceptor; iii) the influence of the structure on the orientation of the emitter within the emitter layer, which will affect the orientation of the TDM and hence the light outcoupling efficiency in the device; iv) trisubstitution about the triazine core; v) dimerization strategies where there are at least two triazines within the material; vi) examples where the linker group goes beyond a phenylene; vii) adjustment of the triazine electron-withdrawing strength through peripheral decoration; viii) examples where there is TSCT in addition to through-bond charge transfer. Finally, moving beyond carbazole-based electron-donors, examples of emitters using other donors will be summarized.



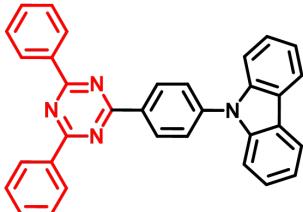
**Figure 3.** Number of publications for TADF materials and the proportion that contain triazine.

Keyword: “triazine” or “TADF”. Scifinder search conducted: 30/06/2022.

## 2. Influence of carbazole (number and position) on the properties of TRZ-based TADF emitters

Carbazole (Cz) has been widely used as an electron-donating group in optoelectronic materials because of its moderate donor strength and the ease with which chemical derivatization can occur.<sup>[12]</sup> Carbazole is also a fully aromatic structure, with no bonds to a sp<sup>3</sup>-centre as exists in acridine, phenoxazine and phenothiazine, bonds that are prone to scission and routes to degradation in the device.

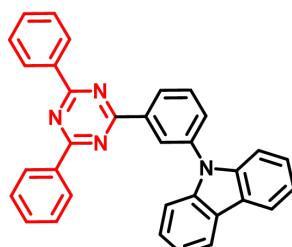
### Single carbazole substitution



CzTRZ / p-CzTRZ

$\lambda_{PL} = N/A$ ,  $\tau_{DF} = N/A$   
 $\Delta E_{ST} = 0.36 \text{ eV}$ ,  $\Phi_{PL} = 71\%$  (10 wt% DPEPO)  
 $\lambda_{EL} = 449 \text{ nm}$ ,  $EQE_{max/100/1000} = 4.2/\sim 1.5/0.6\%$

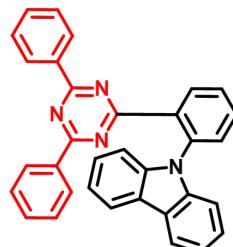
*Adv. Optical Mater.*, 2018, 6, 1701340



m-CzTRZ

$\lambda_{PL} = \sim 425 \text{ nm}$ ,  $\tau_{DF} = 6.44 \mu\text{s}$ , (10 wt% mCP)  
 $\Delta E_{ST} = 0.22 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = \sim 500 \text{ nm}$ ,  $EQE_{max/500/1000} = 19.2/15.7\%/\text{N/A}$

*J. Mater. Chem. C.*, 2019, 7, 12470

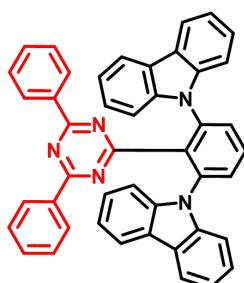


CzTRZ / o-CzTRZ

$\lambda_{PL} = 455 \text{ nm}$ ,  $\tau_{DF} = 3.90 \mu\text{s}$  (5wt % mCP)  
 $\Delta E_{ST} = 0.1 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = 470 \text{ nm}$ ,  $EQE_{max/100/1000} = 9.3/\sim 8.5/\sim 2.0\%$

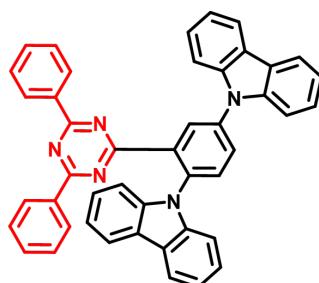
*Dyes. Pigments.*, 2016, 134, 562

### Double carbazole substitution



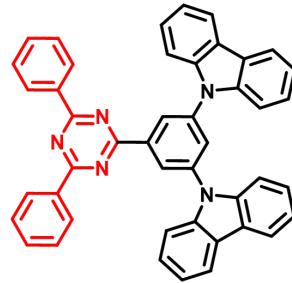
26CT

$\lambda_{PL} = \sim 465 \text{ nm}$ ,  $\tau_{DF} = 16.4 \mu\text{s}$  (1 wt% Zeonex)  
 $\Delta E_{ST} = -0.07 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = N/A$ ,  $EQE_{max/100/1000} = N/A$



25CT

$\lambda_{PL} = \sim 465 \text{ nm}$ ,  $\tau_{DF} = 8.0 \mu\text{s}$  (1 wt% Zeonex)  
 $\Delta E_{ST} = 0.07 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = N/A$ ,  $EQE_{max/100/1000} = N/A$

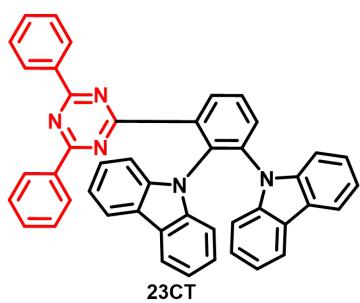


DCzTRZ / 35CT

$\lambda_{PL} = 429 \text{ nm}$ ,  $\tau_{DF} = 3.0 \mu\text{s}$  (polystyrene)  
 $\Delta E_{ST} = 0.25 \text{ eV}$ ,  $\Phi_{PL} = 43\%$  (PhMe)  
 $\lambda_{EL} = 459 \text{ nm}$ ,  $EQE_{max/100/1000} = 17.8/\sim 16.5/\sim 14.0\%$

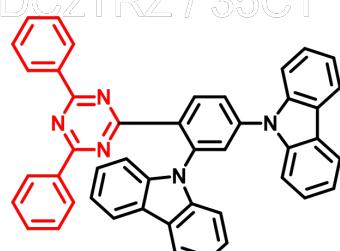
*ACS Appl. Mater. Interfaces* 2018, 10, 35420

*Adv. Mater.* 2015, 27, 2515



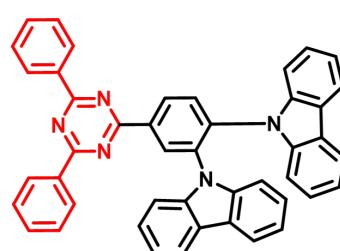
23CT

$\lambda_{PL} = \sim 485 \text{ nm}$ ,  $\tau_{DF} = 8.7 \mu\text{s}$  (1 wt% Zeonex)  
 $\Delta E_{ST} = -0.02 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = \sim 485 \text{ nm}$ ,  $EQE_{max/100/1000} = 21.8/21.8/20.8\%$



24CT

$\lambda_{PL} = \sim 465 \text{ nm}$ ,  $\tau_{DF} = 21.4 \mu\text{s}$  (1 wt% Zeonex)  
 $\Delta E_{ST} = 0.11 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = \sim 475 \text{ nm}$ ,  $EQE_{max/100/1000} = 22.4/20.4/14.5\%$



34CT

$\lambda_{PL} = \sim 450 \text{ nm}$ ,  $\tau_{DF} = N/A$   
 $\Delta E_{ST} = 0.29 \text{ eV}$ ,  $\Phi_{PL} = N/A$   
 $\lambda_{EL} = \sim 460 \text{ nm}$ ,  $EQE_{max/100/1000} = 13.3/9.4/5.6\%$

*ACS Appl. Mater. Interfaces* 2018, 10, 35420

Small  $\Delta E_{ST}$

Large  $\Delta E_{ST}$

CT Strength

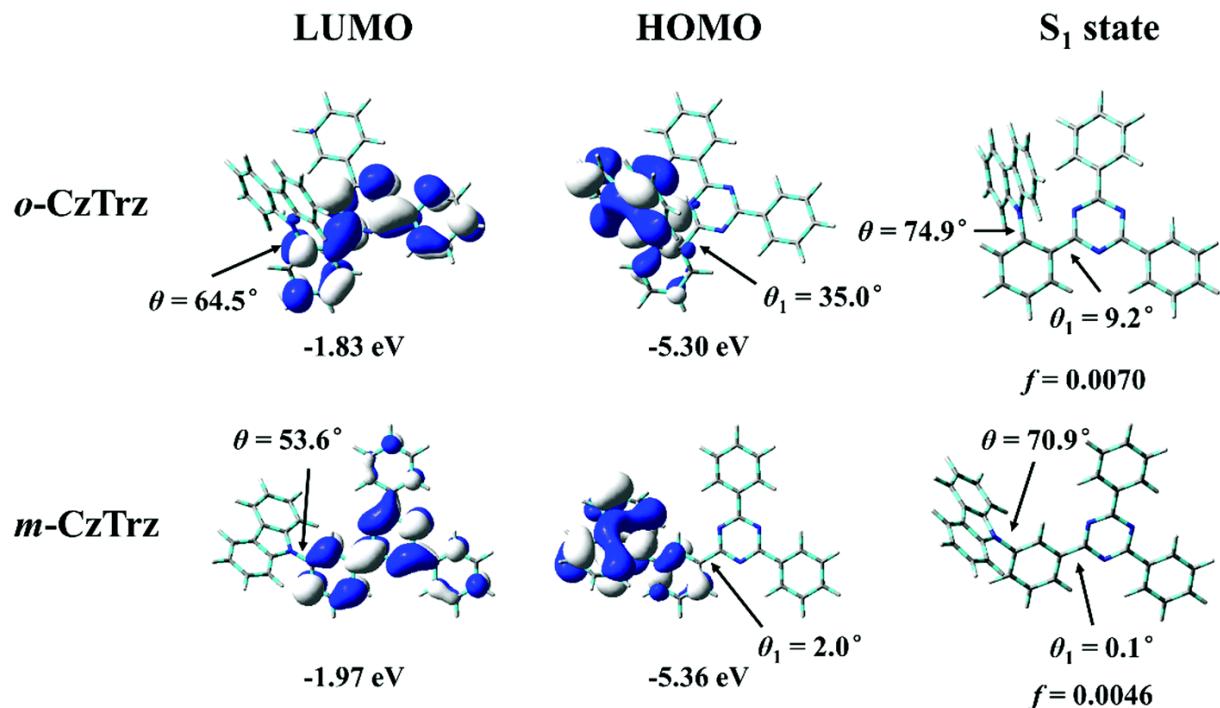
**Figure 3.** Molecular structures and properties of TADF emitters containing carbazole-triazine motifs.

First, we compare the optoelectronic properties of Cz-TRZ derivatives where the Cz is either *ortho*, *meta* or *para* disposed with respect to the diphenyltriazine, as shown in **Figure 3**. The

*para*-analogue **CzTRZ** (here, it is renamed as ***p*-CzTRZ**), first reported by Lee *et al.*,<sup>[13]</sup> is not a TADF emitter due to its too strong conjugation between Cz and TRZ, leading to a large  $\Delta E_{ST}$  of 0.36 eV; the compound is a blue emitter with a  $\lambda_{PL}$  of 449 nm and a photoluminescence quantum yield,  $\Phi_{PL}$ , of 71% (10 wt% DPEPO). A blue OLED with CIE coordinates of (0.17, 0.11) showed only an EQE<sub>max</sub> of 4.2%. A slightly improved OLED efficiency (EQE<sub>max</sub> of 5.8%) was later reported by our group,<sup>[14]</sup> which provided further confirmation that in the device ***p*-CzTRZ** acts as a fluorescent emitter.

The *meta*- and *ortho*- analogues, by contrast, are efficient TADF emitters. **CzTRZ** (renamed as ***o*-CzTRZ**;  $\lambda_{PL} = 455$  nm,  $\tau_d = 3.90$   $\mu$ s in 5 wt% mCP film) possesses a much smaller overlap between the HOMO and LUMO as a result of the more strongly twisted conformation (87.3° for the dihedral angle between carbazole and phenylene bridge determined from the crystal structure<sup>[15]</sup>), which translates into a compound with a  $\Delta E_{ST}$  of 0.1 eV in 5 wt% mCP film. By doping 5 wt% ***o*-CzTRZ** in mCP host as the emitting layer, the OLED exhibited a blue emission with a  $\lambda_{EL}$  at 470 nm and an EQE<sub>max</sub> of 9.3% despite the  $\Phi_{PL}$  of ***o*-CzTRZ** being only 16.7% (5 wt% mCP), which likely implies that the  $\Phi_{PL}$  has been underestimated, considering the device performance. Interestingly, the full width at half maximum (FWHM) of the blue TADF device is only 66 nm, which we attribute to a much reduced geometric reorganization in the excited state to the small degree of conformational freedom present in the *ortho*-linked system. Indeed, many other compounds containing *ortho*-linked donors and acceptors also show narrow emission spectra (in section 9 related to TSCT). Normally, *meta*-disposed electron-donors and electron-acceptors in D-A type emitters show the weakest conjugation compared to their *ortho*- and *para*- analogues, which should lead to a correspondingly smaller  $\Delta E_{ST}$ . However, it was observed by Wu *et al.*<sup>[16]</sup> that the  $\Delta E_{ST}$  of ***m*-CzTRZ** (0.22 eV in 10 wt% in mCP film) is indeed larger than 0.03 eV for ***o*-CzTRZ** (determined again by Wu *et al.*<sup>[16]</sup> although these values are

extracted from the energy difference between the PL at room temperature and the phosphorescence spectra at 77 K in 2-MeTHF and so differ from those of Gong *et al.* [17]). The authors contended that the smaller observed  $\Delta E_{ST}$  for ***o*-CzTRZ** resulted from the near orthogonal conformation of the donor, leading to near complete suppression of the conjugation between Cz and TRZ. The DFT calculations (**Figure 4**) from their work suggest that the  $^3\text{LE}$  state of ***m*-CzTRZ** is much more stabilized than the  $^3\text{CT}$  state, while the  $^3\text{LE}$  state of ***o*-CzTRZ** is much closer in energy to its  $^3\text{CT}$  state. The stabilized  $^3\text{LE}$  state explains the greater  $\Delta E_{ST}$  of ***m*-CzTRZ** than that of ***o*-CzTRZ** due to their similar energy in  $^1\text{CT}$  state, resulting in calculated  $\Delta E_{ST}$  of 0.06 eV and 0.23 eV for ***o*-CzTRZ** and ***m*-CzTRZ**, respectively, which are consistent with the experimental values. The OLEDs exhibited high  $\text{EQE}_{\max}$  of 17.5% and 19.2% for the devices with ***o*-CzTRZ** and ***m*-CzTRZ**, respectively (10 wt% doped in mCP host). Further, the  $\text{EQE}_{\max}$  of the corresponding non-doped OLEDs remain nearly as high at 17.6% and 16.7% for the devices with ***o*-CzTRZ** and ***m*-CzTRZ**, respectively.

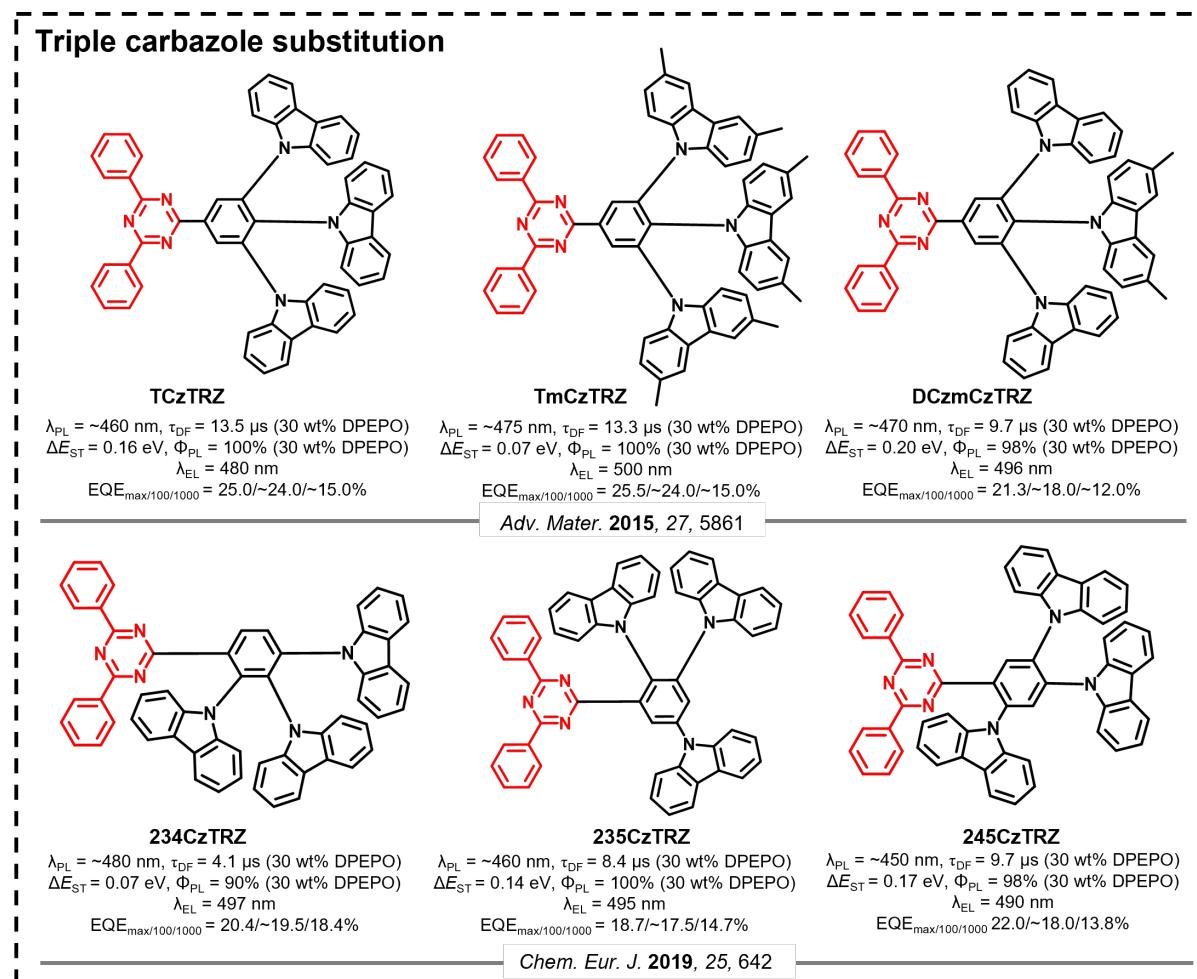


**Figure 4.** HOMO and LUMO distribution and their energy levels, optimized S<sub>1</sub> geometry with oscillator strengths (*f*) for *o*-CzTRZ and *m*-CzTRZ. Reproduced with permission.<sup>[16]</sup> Copyright © 2019, Royal Society of Chemistry.

In comparison to the three aforementioned triazine-based TADF emitters, each possessing only one carbazole donor, emitters with more than one carbazole have also been investigated. Lee *et al.*<sup>[18]</sup> reported a stable deep blue emitter **DCzTRZ** ( $\lambda_{PL}$ : 459 nm;  $\Phi_{PL}$ : 43% in toluene) in which the two carbazole donor group are disposed mutually *meta* to each other and to the triazine acceptor. The  $\Delta E_{ST}$  of **DCzTRZ** is 0.25 eV, which is similar to that of **m-CzTRZ** (0.22 eV in 10 wt% mCP). The blue OLED has an EQE<sub>max</sub> of 17.8% at CIE coordinates of (0.15, 0.15), which is slight less efficient than that of **m-CzTRZ** (19.2%). It is surprising to observe a much bluer emission spectrum in comparison with the device based on **m-CzTRZ**, although the solution PL spectrum in toluene is exactly the same. The blue-shift of the EL may be caused by the different device configurations, choice of host and doping concentration between the two devices.

A number of other analogues containing two carbazole donors were also prepared by Lee *et al.*<sup>[19]</sup> The different substitution patterns of the two carbazoles for six emitters **23CT**, **24CT**, **25CT**, **26CT**, **34CT** and **35CT** (here **DCzTRZ** was renamed as **35CT**) are correlated with their photophysical properties. *Ortho*-substitution to the triazine acceptor induces a large dihedral angle between the carbazole and triazine moieties, thus leading to a stronger charge transfer state and smaller  $\Delta E_{ST}$ , which is evident by comparing **23CT** ( $\Delta E_{ST}$ : -0.02 eV, 1 wt% in Zeonex), **24CT** ( $\Delta E_{ST}$ : 0.11 eV, 1 wt% in Zeonex), **25CT** ( $\Delta E_{ST}$ : 0.07, 1 wt% in Zeonex), **26CT** ( $\Delta E_{ST}$ : -0.07 eV, 1 wt% in Zeonex) with **34CT** ( $\Delta E_{ST}$ : 0.29 eV, 1 wt% in Zeonex) and **35CT** ( $\Delta E_{ST}$ : 0.24 eV, 1 wt% in Zeonex). The difference of  $\Delta E_{ST}$  may result from different conformers,

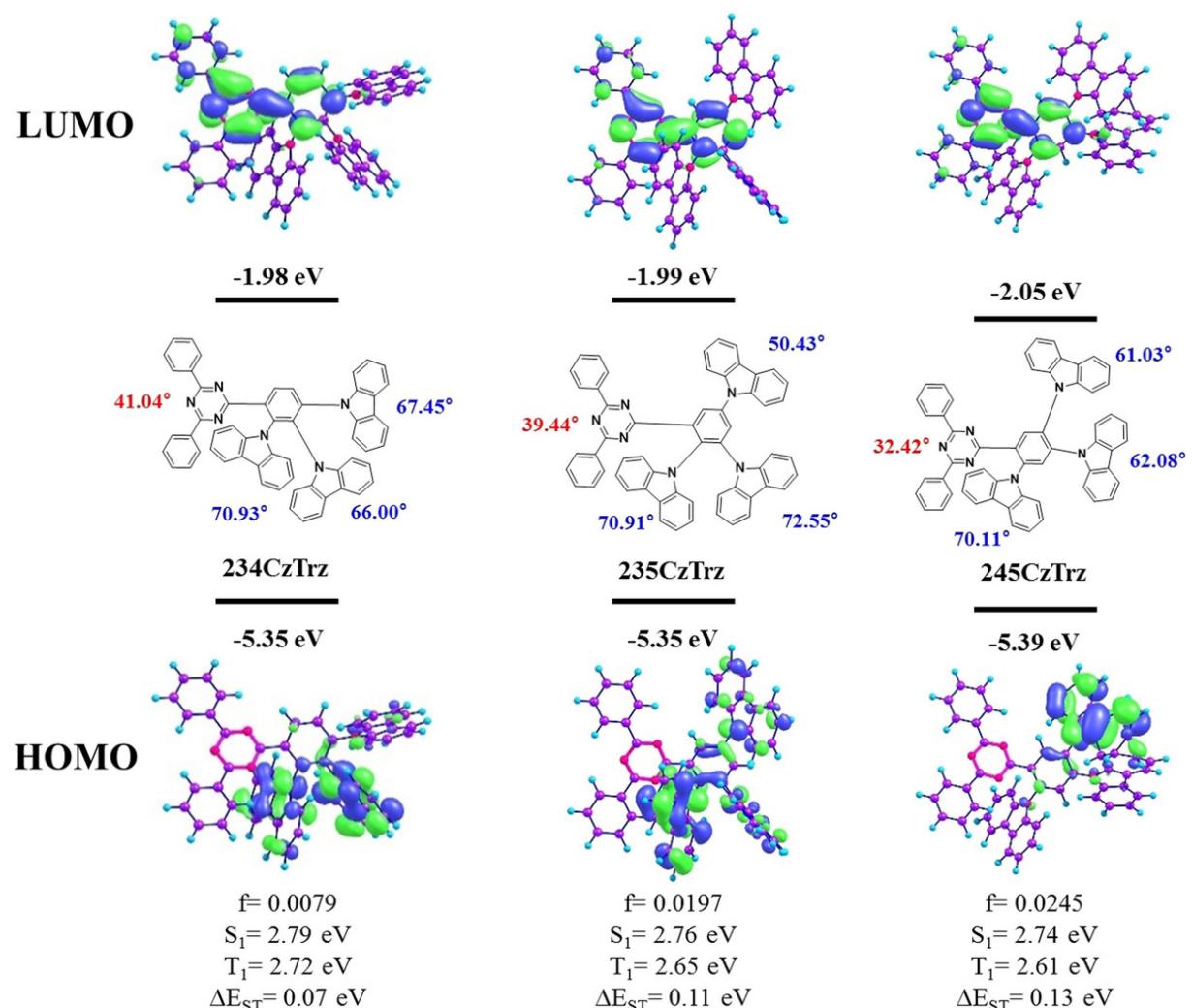
showing the various CT strength. The emitters with no *ortho*-substituted carbazole have larger  $\Delta E_{ST}$  and long-to-negligible delayed fluorescence. The EQE<sub>max</sub> of the OLEDs with **23CT**, **24CT** and **34CT** as emitters are 21.8, 22.4, and 13.3%, respectively, which align with both the smaller  $\Delta E_{ST}$  and faster RISC rates of the former two emitters.



**Figure 5.** Molecular structures and properties of TADF emitters containing three carbazole donors and a triazine acceptor.

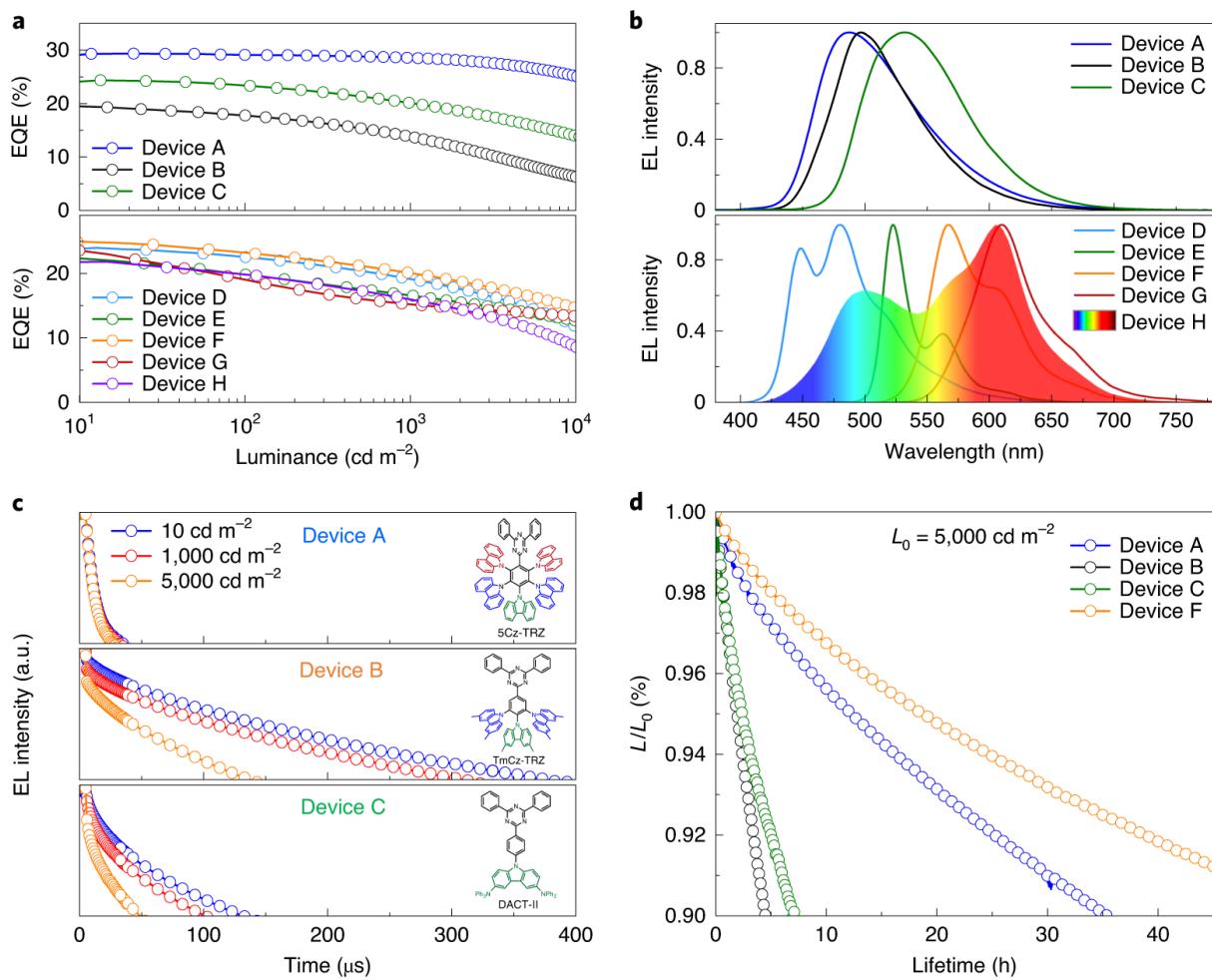
There are also a number of analogue emitters possessing three carbazole donors. Compared with **35CT** ( $\Delta E_{ST}$ : 0.23 eV;  $\Phi_{PL}$ : 43% in toluene), **TCzTRZ**<sup>[20]</sup> ( $\Delta E_{ST}$ : 0.16 eV;  $\Phi_{PL}$ : 100% in toluene), as shown in **Figure 5**, possesses a smaller  $\Delta E_{ST}$  coupled with a significantly improved  $\Phi_{PL}$ . The structural difference between these two compounds is the addition of a third carbazole

*para* to the triazine, which results in a HOMO that is distributed over the three donor carbazoles. The even distribution of the electron density of the HOMO across multiple donors also is in operation in **TmCzTRZ** ( $\Delta E_{ST}$ : 0.07 eV;  $\Phi_{PL}$ : 99% in toluene), which also possesses a smaller  $\Delta E_{ST}$  and a higher  $\Phi_{PL}$  than **DCzmCzTRZ** ( $\Delta E_{ST}$ : 0.20 eV;  $\Phi_{PL}$ : 84% in toluene), which has two different carbazole-based donors. OLEDs with these three emitters achieved comparably high efficiencies, with  $EQE_{max}$  of 25% and CIE coordinates of (0.18, 0.33) for the device with **TCzTRZ** and an  $EQE_{max}$  of 25.5% and CIE coordinates of (0.25, 0.50) for the device with **TmCzTRZ**; the red-shift in the EL of the latter is due to the use of a stronger dimethylcarbazole donor. The  $EQE_{max}$  decreased to 21.3%, with CIE coordinates of (0.23, 0.46), for the device with **DCzmCzTRZ**, reflecting to the lower  $\Phi_{PL}$  for the emitter.



**Figure 6.** Results of molecular orbital calculations showing frontier molecular orbitals distribution and geometry of **234CzTRZ**, **235CzTRZ**, and **245CzTRZ**. Reproduced with permission.<sup>[21]</sup>

Lee *et al.*<sup>[21]</sup> later probed the effect of the regiochemistry of the three carbazole donors on the photophysical properties of the emitters. A cross-comparison of the photophysical properties of compounds **234CzTRZ** ( $\Delta E_{ST}$ : 0.07 eV;  $\Phi_{PL}$ : 90% in DPEPO;  $\tau_d$ : 4.1  $\mu$ s), **235CzTRZ** ( $\Delta E_{ST}$ : 0.14 eV;  $\Phi_{PL}$ : 100% in DPEPO ;  $\tau_d$ : 8.4  $\mu$ s) and **245CzTRZ** ( $\Delta E_{ST}$ : 0.17 eV;  $\Phi_{PL}$ : 98.4% in DPEPO ;  $\tau_d$ : 9.7  $\mu$ s) revealed that the substitution pattern of the three carbazole donors affects both  $\Delta E_{ST}$  and  $\tau_d$ ; all three compounds possess a shorter delayed fluorescence lifetime than that of **TCzTRZ** ( $\tau_d$ : 13.5  $\mu$ s). As shown in the **Figure 6**, the *ortho*-substituted carbazoles to the diphenyltriazine acceptor adopt a large dihedral angle, which leads to a shortened delayed fluorescence lifetime, compound **234CzTRZ**, with three carbazoles packed closely to each other, showed the shortest delayed lifetime of 4.1  $\mu$ s. The EQE<sub>max</sub> of the OLEDs using **234CzTRZ** ( $\lambda_{EL}$ : 497 nm; CIE = (0.20, 0.44)), **235CzTRZ** ( $\lambda_{EL}$ : 495 nm; CIE = (0.20, 0.45)) and **245CzTRZ** ( $\lambda_{EL}$ : 490 nm; CIE = (0.17, 0.39)) are all quite similar at 20.4%, 18.7%, and 22.0%, respectively. At a luminance of 1000 cd m<sup>-2</sup>, the device with **234CzTRZ** can still maintain 18.4% while the EQE<sub>1000</sub> of the OLEDs with **235CzTRZ** and **245CzTRZ** drop to 14.7% and 13.8%, respectively. However, even though these emitters exhibited shorter  $\tau_{DF}$  and smaller  $\Delta E_{ST}$  than those of **TCzTRZ**, their devices performance are inferior to the OLED based on **TCzTRZ**.



**Figure 7.** **a**, EQE versus luminance characteristics. **b**, Normalized EL spectra. **c**, Transient EL decay characteristics of devices A, B and C at 10, 1,000 and 5,000  $\text{cd m}^{-2}$ . a.u., arbitrary units. **d**, Operational lifetimes of the TADF and hyperfluorescent OLEDs. Reproduced with permission.<sup>[22]</sup> Copyright © 2020, Nature Publishing Group.

Recently, a TADF emitter **5Cz-TRZ** ( $\Delta E_{\text{ST}}$ : 0.06;  $\Phi_{\text{PL}}$ : 99% (15wt% mCBP);  $\tau_d$ : 0.8  $\mu\text{s}$  in DMF) was reported (Figure 7c) that contains five carbazole donors.<sup>[22]</sup> Compound **5Cz-TRZ** possesses a very high  $k_{\text{RISC}}$  of  $1.5 \times 10^7 \text{ s}^{-1}$ , which is in part due to the large density of excited states and the small  $\Delta E_{\text{ST}}$ . The authors contended that the fast RISC in **5Cz-TRZ** would contribute to a smaller efficiency roll-off in the device as efficiency roll-off is typically caused by the accumulation of long-lived triplet excitons. The resulting device showed an outstanding

performance with an EQE<sub>max</sub> of 29.3%, which decreased a mere 2.3% at 1,000 cd m<sup>-2</sup>, and a device lifetime, T<sub>90</sub>, of ~600 h from an initial brightness of 1,000 cd m<sup>-2</sup>. All the photophysics and electrochemical characteristics of the aforementioned materials are summarized in **Table 1**.

1. Representative device performance is summarized in **Table 2**.

**Table 1.** Summary of photophysical and electrochemical properties

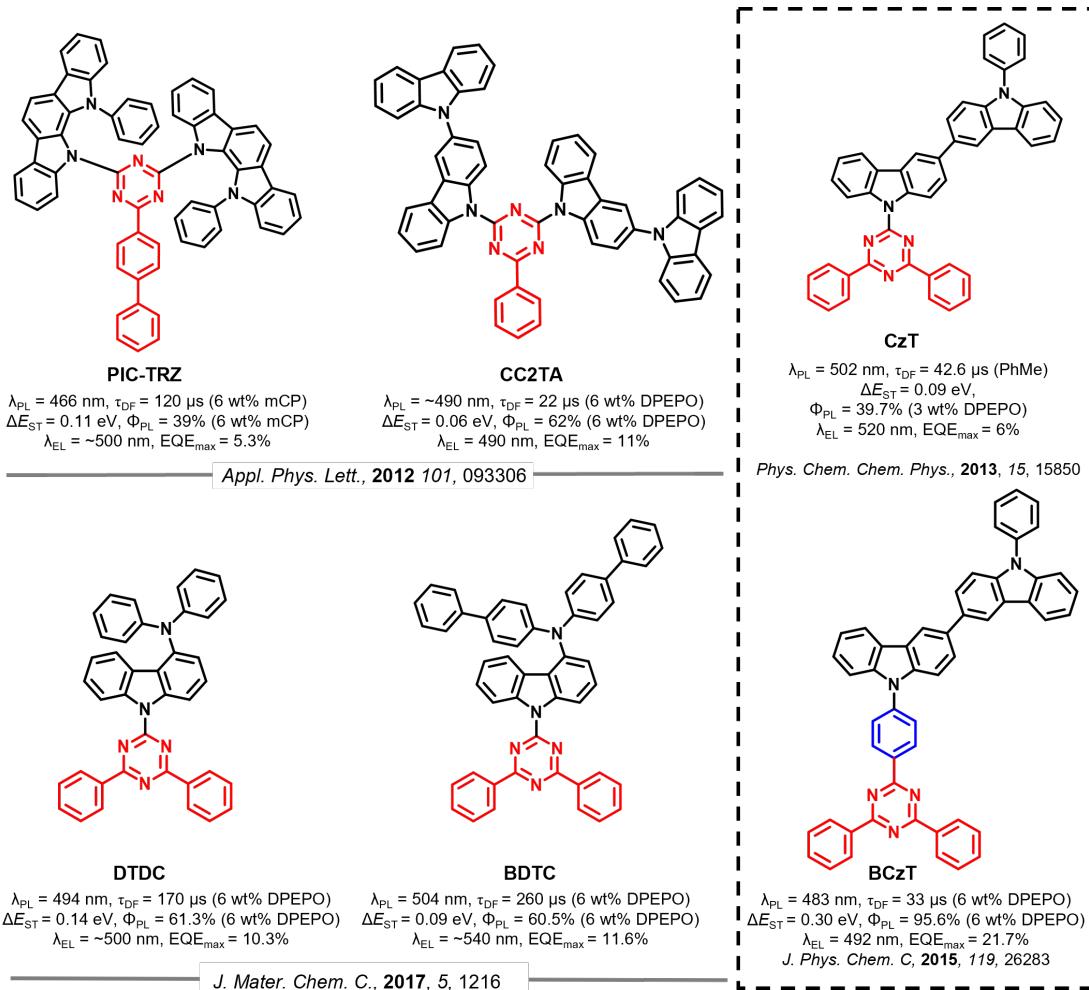
Emitter	Solution λ <sub>PL</sub> /Φ <sub>PL</sub> /τ <sub>d</sub> (medium) [nm/%/μs]	Solid State λ <sub>PL</sub> /Φ <sub>PL</sub> /τ <sub>d</sub> (medium) [nm/%/μs]	ΔE <sub>ST</sub> [eV]	HOMO [eV]	LUMO [eV]	Ref.
CzTrz	/-/	-/71/- (10 wt% in DPEPO)	0.36	-6.04	-3.39	[13]
m-CzTrz	431/-/(PhMe)	~425/-/6.44(10wt% in mCP)	0.22	-5.62	-2.77	[16]
o-CzTRZ	-/-	455/-/3.9(5wt% in mCP)	0.10	-6.12	-3.26	[15]
26CT	~465/-/(PhMe)	-/-16.4 ± 0.9(1% wt in ZEONEX)	-0.07	-	-	[19]
25CT	~465/-/(PhMe)	-/-8.0 ± 1.2(1% wt in ZEONEX)	0.07	-	-	[18]
DCzTrz	~440/0.43/-/(PhMe)	429/-/3.1(polystyrene)	0.25	-5.88	-2.86	[19]
23CT	~485/-/(PhMe)	-/-8.7 ± 0.3(1% wt in ZEONEX)	-0.02	-6.12	-3.52	[19]
24CT	~465/-/(PhMe)	-/-21.4 ± 1.9(1% wt in ZEONEX)	0.11	-6.14	-3.55	[19]
34CT	~450/-/(PhMe)	-/-/(1% wt in ZEONEX)	0.29	-6.16	-3.50	[19]
TCzTrz	-/100/(PhMe)	~460/100/13.5(30 wt% in DPEPO)	0.16	-	-	[20]
TmCzTrz	-/99/(PhMe)	~475/100/13.3(30 wt% in DPEPO)	0.07	-	-	[20]
DCzmCzTrz	-/89/(PhMe)	~470/98/9.7(30 wt% in DPEPO)	0.20	-	-	[20]
235CzTrz	~480/-/(PhMe)	~460/100/8.4	0.14	-	-	[21]
245CzTrz	~500/-/(PhMe)	~450/98.4/9.7	0.17	-	-	[21]
5Cz-TRZ	486/92/-/(PhMe)	486/99/2.1(15wt% in mCBP)	0.03	-5.92	-	[22]

**Table 2.** Summary of device structures and performance

Emitter	Device Structure	λ <sub>EL</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a</sup> ) [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> [cd m <sup>-2</sup> ]	Ref.
CzTRZ	ITO/PEDOT:PSS/TAPC/mCP/ CzTrz /TSPO1/TPBi/LiF/AI	449	(0.17, 0.11)	-	4.2/2.7/3.8	~1.5/0.6	[13]
m-CzTRZ	ITO/PEDOT:PSS/TAPC/10 wt% m-CzTrz:mCP/TmPyPB/LiF/AI	~500	(0.25,0.44)	3.3	19.2/40.3/51.3	15.7/N/A	[16]
o-CzTRZ	ITO/PEDOT:PSS/TAPC/mCP/mCP:5%CzTrz/TSP O1/LiF/AI	470	(0.15, 0.22)	-	9.3/-/14.59	-/-	[15]
DCzTRZ	ITO/PEDOT:PSS/TAPC/mCP/25 wt% DCzTrz:DPEPO/TSPO1/TPBi/LiF/AI	459	(0.15, 0.15)	5.9	17.8/22.4/26.8	~8.5/~2.0	[18]
23CT	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30 wt% 23CT /TSPO1/TPBi/LiF/AI	-	(0.17,0.33)	~3.8	21.8/30.9/45.9	21.8/20.8	[19]
24CT	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30 wt% 24CT /TSPO1/TPBi/LiF/AI	-	(0.15,0.26)	~3.8	22.4/-/40	20.4/14.5	[19]
34CT	ITO (120 nm)/PEDOT:PSS/TAPC/mCP/DPEPO:30 wt% 34CT /TSPO1/TPBi/LiF/AI	-	(0.15,0.17)	~3.8	13.3/-/21.5	9.4/5.6	[19]
TCzTRZ	ITO/PEDOT:PSS/TAPC/mCP/40 wt% TCzTrz:DPEPO/TSPO1/TPBi/LiF/AI	480	(0.18, 0.33)	-	25/42.7/-	~24.0/~15.0	[20]
234CzTRZ	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30% 234CzTrz/TSPO1/TPBi/LiF/AI	497	(0.20, 0.44)	~3.6	20.4/-/-	~19.5/18.4	[21]
235CzTRZ	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30% 235CzTrzor/TSPO1/TPBi/LiF/AI	495	(0.20, 0.45)	~3.6	18.7/-/-	~17.5/14.7	[21]
245CzTRZ	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30% 235CzTrzor/TSPO1/TPBi/LiF/AI	490	(0.17, 0.39)	~3.6	22/-/-	~18.0/13.8	[21]

5Cz-TRZ	ITO/HAT-CN/ $\alpha$ -NPD/Tris-PCz/mCBP /15 wt%	486	-	-	29.3/-/-	$\sim$ 29.0/28.6	[22]
	5Cz-TRZ: mCBP/CF3-TRZ/30 wt%						
	Liq:BPPB/Liq/AI						

### 3. Importance of conjugated bridge between TRZ and carbazole/carbazole derivatives



**Figure 8.** Comparison of TADF emitters with and without a phenylene bridge between triazine and carbazole derivatives.

**PIC-TRZ** [23] ( $\lambda_{PL}$ : 466 nm;  $\tau_d$ : 120  $\mu$ s;  $\Delta E_{ST}$ : 0.11 eV;  $\Phi_{PL}$ : 39%; 6 wt% in mCP) was the first reported purely organic TADF emitter for use in OLEDs (Figure 1). Unlike the most commonly used design strategy for TADF emitters, which is to separate electron-donor and electron-acceptor groups by a conjugated phenylene bridge, **PIC-TRZ** contains two carbazole derivatives that are directly connected to a triazine core. A relatively small  $\Delta E_{ST}$  (0.11 eV) was obtained due to the large twist angle between the carbazole-based donors and the triazine

acceptor. Even though the reported EQE<sub>max</sub> could only reach 5.3% for the sky-blue OLED, this nevertheless indicated that up-conversion of triplet excitons to singlet excitons was occurring as the theoretical EQE<sub>max</sub> of the device was limited to 2% based on its relatively low Φ<sub>PL</sub> of 39% and assuming no triplet exciton harvesting.

In lieu of an indolocarbazole donor, compound **CC2TA** [24] ( $\lambda_{PL}$ : 490 nm;  $\tau_d$ : 22 μs;  $\Delta E_{ST}$ : 0.06 eV;  $\Phi_{PL}$ : 62%; 6 wt% in DPEPO) contains a carbazole dendron type donor structure. **CC2TA** exhibited a much improved  $\Phi_{PL}$  of 62% (6 wt% in DPEPO) as well as a reduced  $\Delta E_{ST}$  (0.06 eV) compared to **PIC-TRZ**, and the corresponding device showed an EQE<sub>max</sub> of 11%. **DTDC** ( $\Delta E_{ST}$ : 0.14;  $\tau_d$ : 170 μs;  $\Phi_{PL}$ : 61.3%; 6 wt% in DPEPO) and **BDTC** ( $\Delta E_{ST}$ : 0.09;  $\tau_d$ : 260 μs;  $\Phi_{PL}$ : 60.5%; 6 wt% in DPEPO), each containing only one functionalized carbazole donor, were also prepared by Adachi and co-workers.<sup>[25]</sup> These two compounds are randomly oriented in the neat films; however, the orientation of the TDMs (orientation order parameter θ: -0.18 for DTDC; -0.12 for BDTC) of these two TADF emitters is significantly more horizontally aligned when they are dispersed in the DPEPO matrix. The best OLEDs showed comparable efficiencies to that of the **CC2TA** device with EQE<sub>max</sub> values of 10.3% and 11.6% for the **DTDC** and **BDTC** devices, respectively.

Adachi *et al.*<sup>[26]</sup> also performed a study to elucidate the importance of the bridging phenylene by comparing **CzT** ( $\lambda_{PL}$ : 502 nm;  $\Delta E_{ST}$ : 0.09 eV;  $\tau_d$ : 43 μs;  $\Phi_{PL}$ : 40%; 3 wt% in DPEPO) with **BCzT** ( $\lambda_{PL}$ : 483 nm;  $\Delta E_{ST}$ : 0.3 eV;  $\tau_d$ : 33 μs;  $\Phi_{PL}$ : 96%; 6 wt% in DPEPO) and demonstrated that the bridging phenylene contributes to increasing the overlap density between the electronic wave functions of the ground state and the lowest excited singlet state, leading to both a larger  $\Delta E_{ST}$  and higher  $\Phi_{PL}$ , yet surprisingly a blue-shifted  $\lambda_{PL}$ . **CzT**, first reported in 2013 by Adachi *et al.*,<sup>[27]</sup> shows a small  $\Delta E_{ST}$  of 0.09 eV. The reported OLED shows an EQE<sub>max</sub> of only 6% due

to the relatively low  $\Phi_{PL}$  of 40%. The  $S_1 \rightarrow S_0$  radiative decay of **BCzT** is significantly enhanced, leading to a much higher  $\Phi_{PL}$  of 96% (6 wt% in DPEPO), while the emission blue-shifted by 19 nm with a higher doping concentration in DPEPO matrix. Despite **BCzT** obtaining a near unity  $\Phi_{PL}$ , the observed  $\Delta E_{ST}$  was dramatically raised to 0.3 eV. Despite the larger  $\Delta E_{ST}$ , the delayed fluorescence lifetimes are of a similar magnitude, which can only be explained by the implication of intermediate triplet states in the RISC process of **BCzT**. The **BCzT**-based sky-blue OLED exhibits an  $EQE_{max}$  of 21.7% with  $\lambda_{EL}$  at 492 nm, which is significantly improved compared to the 6%  $EQE_{max}$  of the **CzT**-based OLED. The photophysical and electrochemical data of the aforementioned materials are summarized in **Table 3**. Representative device performance is summarized in **Table 4**.

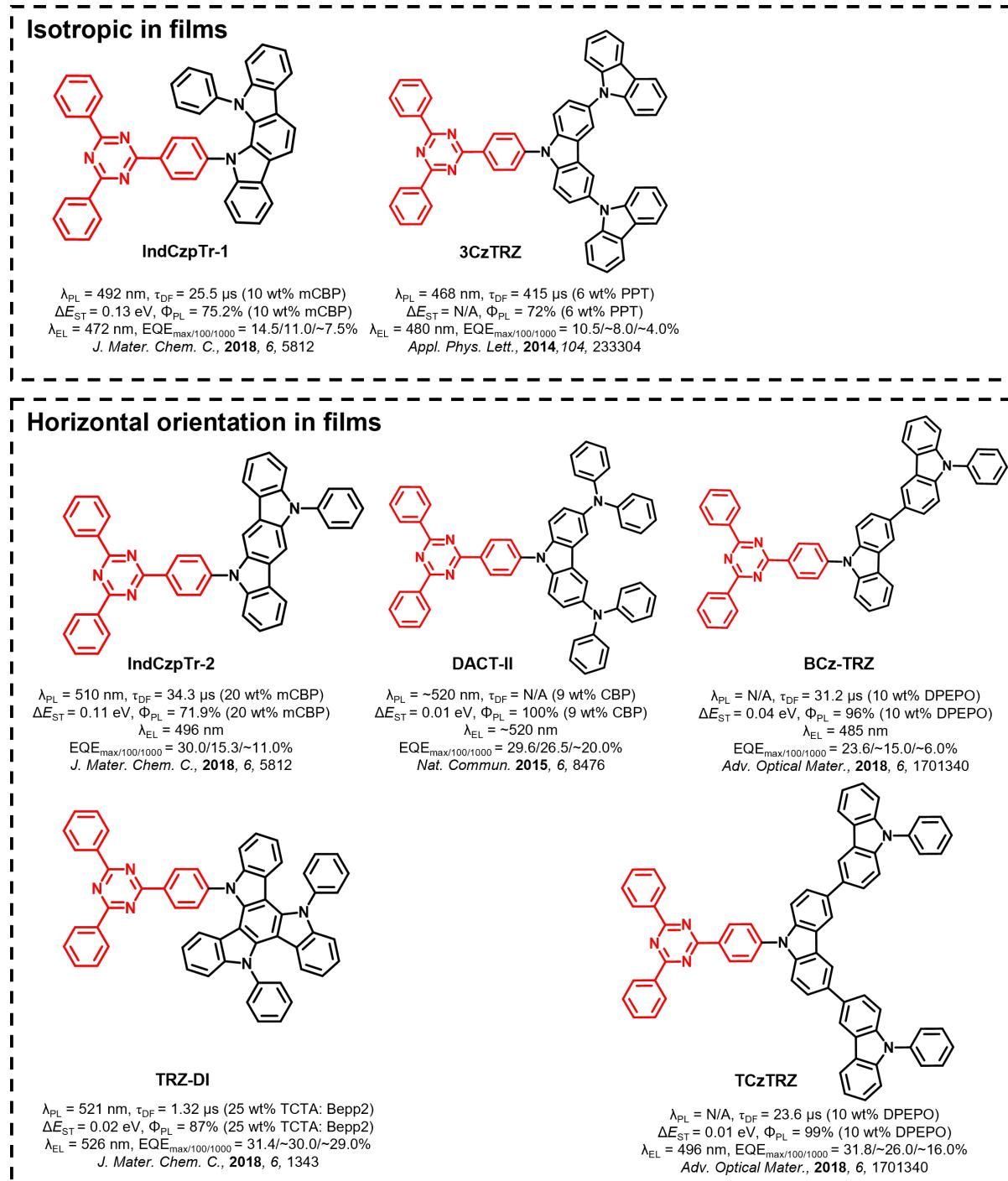
**Table 3.** Summary of photophysical and electrochemical properties

Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
PIC-TRZ	-/35/120(PhMe)	/39/120(6 wt % in mCP)	0.11	-	-	[23]
CC2TA	435/-/(cyclohexane)	~490/62/22(6 wt% in DPEPO)	0.06			[24]
CzT	512/45.6/42.6(PhMe)	502/39.7/-(3 wt% in DPEPO)	0.09			[26]
DTDC	-/-/-	494/61.3/0.17 ms(6 wt% in DPEPO)	0.14	5.80	2.60	[25]
BDTC	-/-/-	504/60.5/0.26 ms(6 wt% in DPEPO)	0.09	5.70	2.70	[25]
BCzT	460/-/(PhMe)	483/95.6/33(6 wt % in DPEPO)	~0.30	-	-	[26]

**Table 4.** Summary of device structures and performances

Emitter	Device Structure	$EL_{max}$ [nm]	CIE	$V_{on}$ [V]	$EQE/PE/CE^a$ [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	$EQE_{100/1000}$ cd m <sup>-2</sup>	Ref.
PIC-TRZ	ITO/a-NPD/m-CP/6 wt %-PIC-TRZ:m-CP/BP4mPy/LiF/Al	~500	-	-	0.053/-/-	NA	[23]
CC2TA	ITO/a-NPD/ 6 wt. % CC2TA:mCP/6 wt. % CC2TA:DPEPO/DPEPO/TPBi /LiF/Al	490	-	-	0.11/-/-	NA	[24]
CzT	ITO/a-NPD/TCTA/CzSi/3 wt% CzT:DPEPO/DPEPO/TPBi/LiF/Al	520	(0.23, 0.40)		0.06/9.7-	NA	[26]
DTDC	glass/ITO/a-NPD/TCTA/mCP /6 wt% DTDC:DPEPO/DPEPO/TPBi/LiF/Al	~500	-	3.8	10.3/-/-	NA	[25]
BDTC	glass/ITO/a-NPD/TCTA/mCP/6 wt% BDTC:DPEPO /DPEPO/TPBi/LiF/Al	~540	-	4.7	11.6/-/-	NA	[25]
BCzT	ITO/a-NPD/m-CBP/6 wt% BCzT:DPEPO/TPBi/LiF/Al	492	-	-	21.7/-/-	NA	[26]

#### 4. Orientation of TADF materials containing TRZ and carbazole/carbazole derivatives as donors

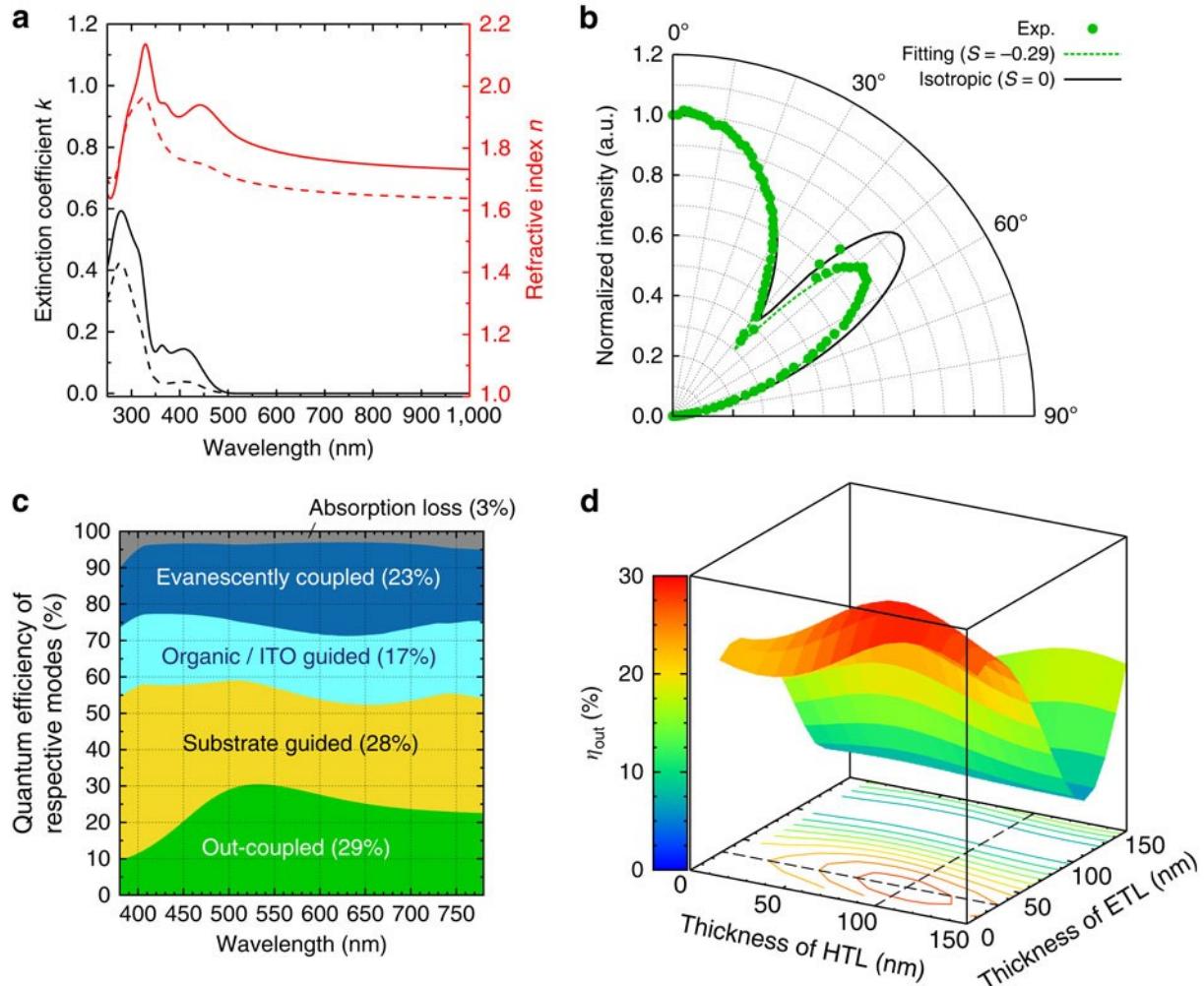


**Figure 9.** Molecular structures of triazine based TADF emitters with/without preference of horizontal orientation in films.

The EQE of the OLEDs also depend on the orientation of the TDM of the emitters, which affects the light out-coupling efficiency of the device.<sup>[28]</sup> Wang *et al.*<sup>[29]</sup> prepared two TADF emitters **IndCzpTr-1** ( $\Phi_{PL}$ : 75.2%, 10 wt% in mCBP) and **IndCzpTr-2** ( $\Phi_{PL}$ : 71.9%, 20 wt% in mCBP) containing an indolocarbazole-based donor (**Figure 9**). **IndCzpTr-2** possesses a slightly smaller  $\Delta E_{ST}$  of 0.11 eV than the 0.13 eV for **IndCzpTr-1**. The more extended structure of **IndCzpTr-2** leads to a preferential horizontal orientation of the TDM in the neat film with an orientation order parameter, S, of -0.264 (estimated by VASE). The S value of **IndCzpTr-1** in the neat film is only -0.1, which is far away from the theoretical value of S = -0.5 for a perfectly horizontal orientation of the TDM. From these measurements in neat films, the orientation factor  $\theta$  was calculated as 0.70 for **IndCzpTr-1** and 0.77 for **IndCzpTr-2**, where a value of 1 indicates a perfectly horizontally aligned TDM and a value of 0.66 indicates an isotropic orientation. As a result of greater horizontal orientation of the TDM of **IndCzpTr-2** there is a strong divergence in the efficiencies of the OLEDs. The optimized OLEDs within this study realized  $EQE_{max}$  values of 14.5% and 30.0% for **IndCzpTr-1** ( $\lambda_{EL}$ : 472 nm; CIE = (0.17, 0.27)) and **IndCzpTr-2** ( $\lambda_{EL}$ : 496 nm; CIE = (0.23, 0.50)), respectively. Even though the  $EQE_{max}$  of the device with **IndCzpTr-1** only reached 14.5%, it is still much improved compared with **PIC-TRZ**, demonstrating the link between improved performance and the presence of the phenylene bridge, as both compounds contain the same donor and diphenyltriazine acceptor.

The emitter **DACT-II** ( $\Delta E_{ST}$ : 0.009 eV;  $\Phi_{PL}$ : 100% as 9 wt% doped film in CBP))<sup>[30]</sup> is a derivative of **Cz-TRZ** but contains peripheral diphenylamino groups decorating the Cz donor. Their addition significantly strengthens the donor and turns on TADF compared to **Cz-TRZ**. **DACT-II** exhibits a near zero  $\Delta E_{ST}$  of 0.009 eV yet a large oscillator strength,  $f$ , reflected in the unity  $\Phi_{PL}$  (9 wt% in CBP film). These desirable traits translate into a green OLED that shows an impressive  $EQE_{max}$  of 29.6% with  $\lambda_{EL}$  at ~520 nm. The high efficiency of the device

is also due to the preferential horizontal orientation of the TDM of **DACT-II** as determined by VASE measurements ( $S = -0.32$ ) in **Figure 10a** and angular-dependent PL measurements ( $S = -0.29$ ) in **Figure 10b**. Kim *et al.*<sup>[31]</sup> later showed that efficiency roll-off can be suppressed in the OLED if the CBP host is replaced by a TCTA:B3YPMPM exciplex-forming host, as the authors found a 1.5-fold increased RISC rate in this system ( $k_{\text{risc}}$ : CBP  $1.37 \times 10^5 \text{ s}^{-1}$ ; TCTA:B3PYMPM  $2.06 \times 10^5 \text{ s}^{-1}$ ). As a result, the device gives a higher efficiency with an  $\text{EQE}_{\text{max}}$  of 34.2%, which only slightly decreased to 31% at a luminescence of  $1000 \text{ cd m}^{-2}$ .

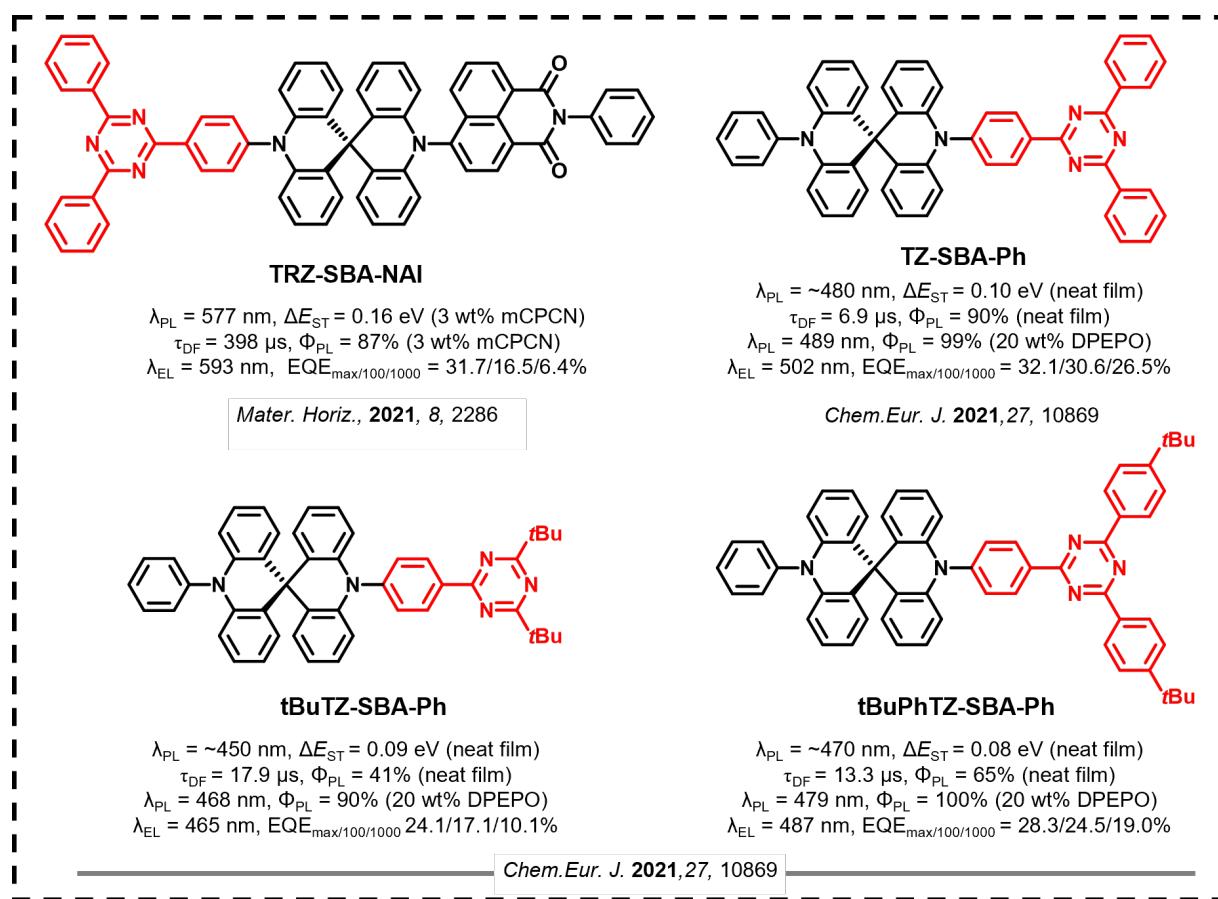


**Figure 10.** (a) Results of VASE measurements; extinction coefficient  $k$  (black) and refractive index  $n$  (red). The solid and dashed lines represent spectra for the ordinary and extraordinary optical constants, respectively. (b) Results of angular-dependent PL experiments. (c) Optical

simulations for the **DACT-II-9** device. The numbers in parentheses for respective modes are obtained by the integration with respect to emission wavelengths with weighting of the PL intensity. (d) Dependence of the out-coupling mode on the thicknesses of hole and electron-transport layers. [30]

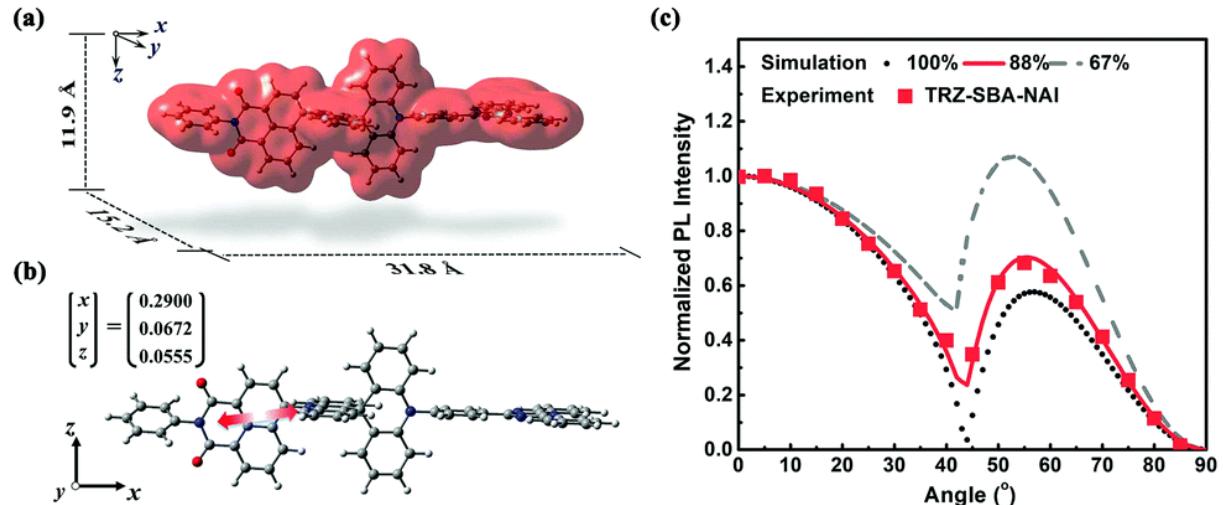
However, the close structural analogue, **3CzTRZ**<sup>[32]</sup> ( $\lambda_{PL}$ : 468 nm;  $\tau_d$ : 415  $\mu$ s;  $\Delta E_{ST}$ : not reported;  $\Phi_{PL}$  72%; 6 wt% in PPT), seems to be oriented randomly in the solid state. **3CzTRZ** presents a blue-shifted emission at 468 nm and a lower  $\Phi_{PL}$  of 72% as a 6 wt% doped PPT film compared with **DACT-II** (9 wt% in CBP). The OLED with **3CzTRZ** thus shows much poorer performance with an  $EQE_{max}$  of 10.5% [ $\lambda_{EL}$ : 480 nm; CIE = (0.17, 0.26)], which implies that there is not even a 100% IQE, assuming an out-coupling efficiency of 0.2~0.25 based on an isotropic orientation of the TDM. However, a structural analog of **3CzTRZ**, **TCzTRZ** ( $\Delta E_{ST}$ : 0.01 eV;  $\Phi_{PL}$ : 99%; 10 wt% in DPEPO), developed by Lee *et al.*,<sup>[13]</sup> who replaced the C-N bonded tercarbazole dendron of **3CzTRZ** with a C-C bonded tercarbazole dendron, does show 95% horizontal orientation of the TDM; a similar analogue, **BCzTRZ**, also shows a horizontal dipole orientation of its TDM as high as 89%, indicating the preferential in-plane alignment is formed both in tercarbazole and bicarbazole containing compounds. Thus, subtle changes in molecular design can have profound effects on both the propensity for horizontal orientation of the TDM and the efficiency of the device. The  $\Phi_{PL}$  values of the 10 wt% doped DPEPO film are 99% and 96% for **TCzTRZ** and **BCzTRZ**, respectively, both of which are significantly higher than that of **3CzTRZ** (72% in DPEPO matrix). Thanks to the near unity  $\Phi_{PL}$  and near perfect horizontal orientation, the blue OLED shows an outstanding  $EQE_{max}$  of 31.8% at a  $\lambda_{EL}$  of 496 nm and CIE coordinates of (0.20, 0.44).

**TRZ-DI** ( $\Delta E_{ST}$ : 0.023 eV), designed by Kwon *et al.*,<sup>[33]</sup> incorporates a  $D_3$ -symmetric triazatruxene donor. Compared with Cz in **Cz-TRZ**, the extended conjugation within triazatruxene effectively localizes the HOMO only on the donor group, leading to a very small  $\Delta E_{ST}$  of 0.02 eV. A small dihedral angle ( $24.5^\circ$ ) between the plane of triazatruxene and the TRZ acceptor, along with the rigid molecular structure, contribute to a  $\Phi_{PL}$  of 87% (25 wt% in TCTA: Bepp2 exciplex host). A green OLED using **TRZ-DI** [ $\lambda_{EL}$ : 526 nm; CIE = (0.31, 0.57)] as the emitter exhibits an outstanding EQE<sub>max</sub> of 31.4%. The orientation of the TDM of **TRZ-DI** was not investigated, but it is not unreasonable to hypothesize that **TRZ-DI** possesses a highly horizontally orientated TDM based on the analysis of the  $\Phi_{PL}$  and the EQE<sub>max</sub>.



**Figure 11.** Molecular structures and properties of TADF emitters with preference of horizontal orientation in films.

**TRZ–SBA–NAI** ( $\Delta E_{\text{ST}}$ : 0.16 eV;  $\Phi_{\text{PL}}$ : 87%; 3 wt% in mCPCN), designed by Yang *et al.*,<sup>[34]</sup> contains an A–D–A' motif, where a spirobiacridine (SBA) donor is used to link the acceptors NAI and TRZ, as shown in **Figure 11**. Because of the weak electronic communication through the central *sp*<sup>3</sup> carbon on the SBA donor, two distinct CT emission bands at  $\lambda_{\text{PL}}$  of around 474 and 586 nm were observed in toluene, with the former assigned to **TRZ–SBA** and the later to **SBA–NAI**. By contrast, the solid-state emission 3.0 wt% doped in mCPCN matrix only showed a band at  $\lambda_{\text{PL}}$  of 577 nm from the **SBA–NAI** moiety, suggesting efficient energy conversion from the high-lying excited states of **TRZ–SBA** to **SBA–NAI**. The different PL character in toluene and film was ascribed to the higher intramolecular charge transfer rate via FRET to the radiative decay rate of the **TRZ–SBA** counterpart. The TDM vector aligns with the molecular long axis, which contributes to the high horizontal dipole ratios ( $\Theta_{\parallel}$ ) of 88% in the mCPCN host matrix (**Figure 12**). Finally, the orange-red OLED based on **TRZ–SBA–NAI** showed an EQE<sub>max</sub> of 31.7% at  $\lambda_{\text{EL}}$  of 593 nm and CIE coordinates of (0.55, 0.45).

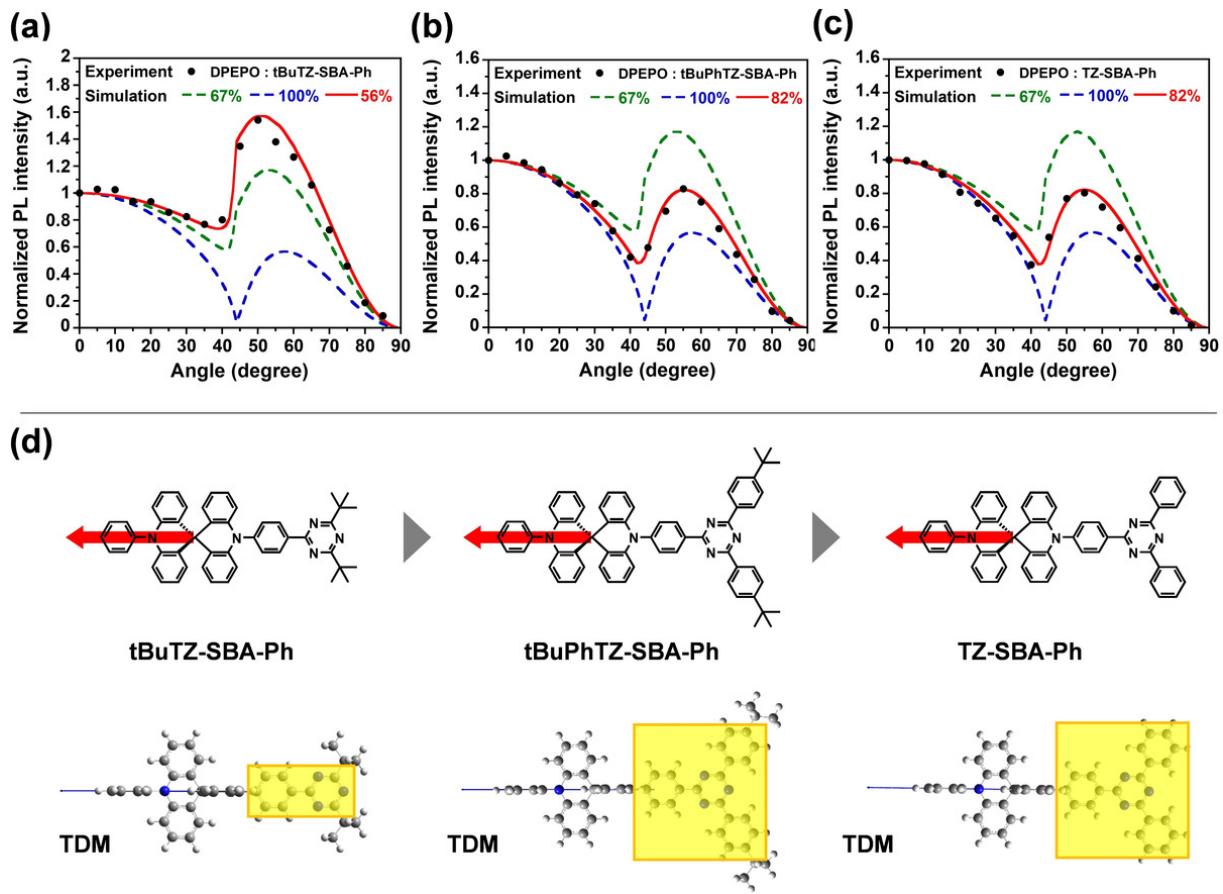


**Figure 12.** (a) Isodensity surface of **TRZ–SBA–NAI** with  $\rho = 0.001 \text{ e bohr}^{-3}$  and their spatial range in the cartesian axes based on optimized S<sub>0</sub> structure. (b) The direction of the calculated S<sub>0</sub>–S<sub>1</sub> transition dipole moment (as indicated by the arrow) of **TRZ–SBA–NAI** based on the optimized S<sub>1</sub> structure. (c) Measured *p*-polarized PL intensity (at PL peak

wavelength) *versus* emission angle curve of 3.0 wt% **TRZ–SBA–NAI** doped into mCPCN host.

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A similar design using SBA as the donor and triazine as the acceptor has also been explored by Kido *et al.*<sup>[35]</sup> to enable the horizontal orientation of the TDM of emitters in OLEDs. Three molecules, **tBuTZ-SBA-Ph**, **TZ-SBA-Ph** and **tBuPhTZ-SBA-Ph** were prepared with the difference in structure associated with the choice of distal groups on the triazine acceptor unit. The  $\Phi_{PL}$  values of **tBuTZ-SBA-Ph**, **TZ-SBA-Ph** and **tBuPhTZ-SBA-Ph** in toluene are modest at 34%, 55% and 43%, respectively. However, the  $\Phi_{PL}$  values in neat films increased to 41% for **tBuTZ-SBA-Ph**, 90% for **TZ-SBA-Ph**, and 65% for **tBuPhTZ-SBA-Ph**. By comparing with the  $\Phi_{PL}$  values of the 10 wt% doped films in DPEPO matrix (87% for **tBuTZ-SBA-Ph**, 100% for **TZ-SBA-Ph**, and 97% for **tBuPhTZ-SBA-Ph**), the similarly high  $\Phi_{PL}$  values for **TZ-SBA-Ph** in both neat and doped films indicate a strong suppression of intermolecular concentration quenching. The orientation ratio ( $\Theta$ ) in evaporated doped films was quantified to be 56%, 82% and 82% for **tBuTZ-SBA-Ph**, **TZ-SBA-Ph** and **tBuPhTZ-SBA-Ph**, respectively, as shown in **Figure 13**. The larger  $\pi$ -plane in **TZ-SBA-Ph** and **tBuPhTZ-SBA-Ph**, as asserted by the authors, would be more likely to interact with the deposition surface, thus enhancing the horizontal orientation. The OLEDs achieved  $EQE_{max}$  of 24.1%, 31.2% and 28.3% for the devices with **tBuTZ-SBA-Ph**, **TZ-SBA-Ph** and **tBuPhTZ-SBA-Ph**, respectively. The performance is line with the properties in doped films in terms of  $\Phi_{PL}$  and  $\Theta$  values. The photophysics and electrochemical characteristics of the aforementioned materials are summarized in **Table 5** and representative device performance metrics are summarized in **Table 6**.



**Figure 13.** PL intensity of 10 wt% a) tBuTZ-SBA-Ph, b) tBuPhTZ-SBA-Ph, and c) TZ-SBA-Ph-doped host films at different angles. The experimental data are compared with the fitting curve for different horizontal dipole ratios ( $\Theta$ ) for X-SBA-Ph derivatives doped in a host film of **DPEPO**. d) The direction of the calculated S<sub>0</sub>-S<sub>1</sub> transition dipole moment. Reproduced with permission.<sup>[35]</sup>

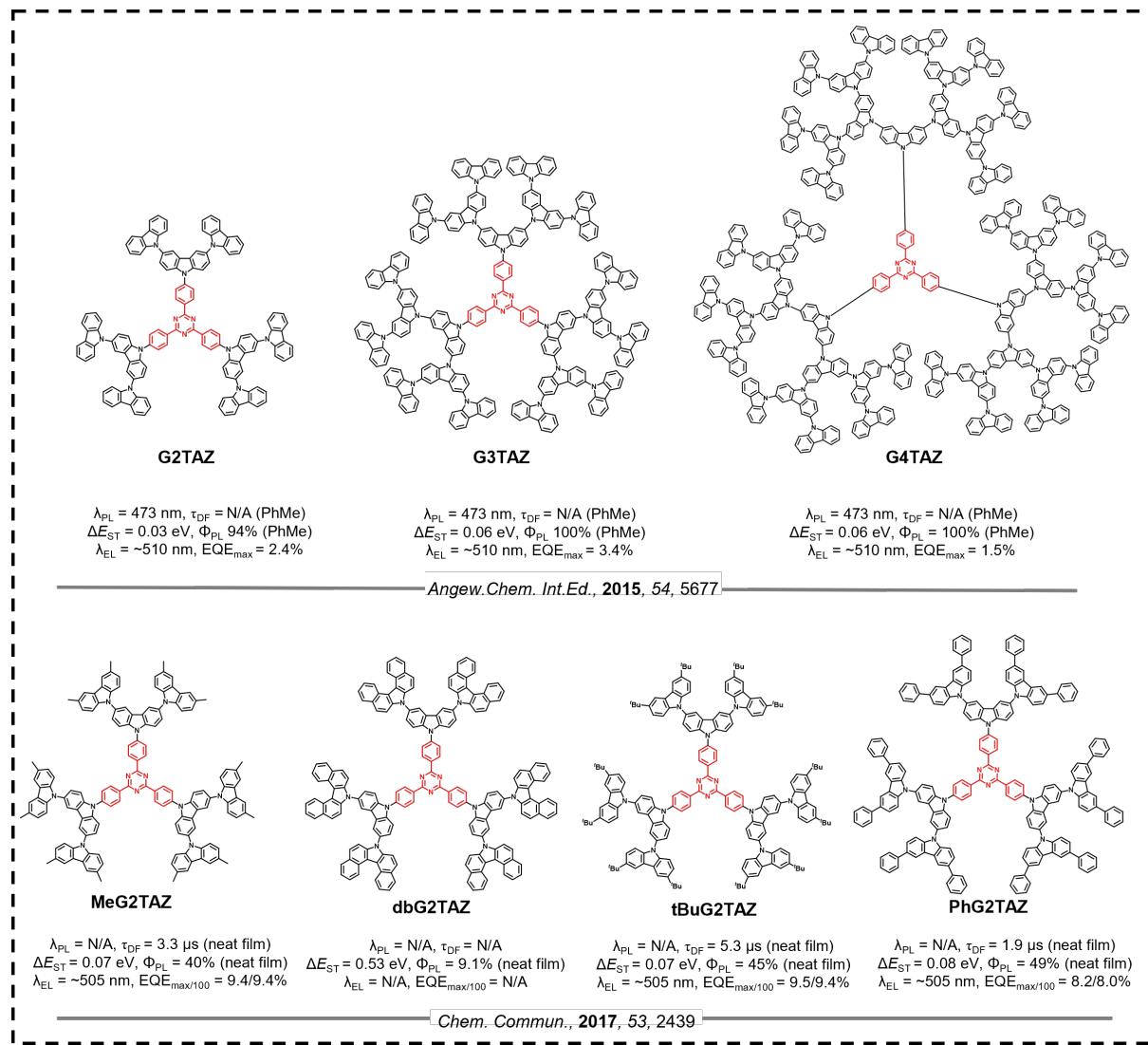
**Table 5.** Summary of photophysical and electrochemical properties

Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
IndCzPTr-1	480//25.48(2Me-THF)	492/75.2/-(mCBP)	0.13	-5.47	-2.91	[29]
IndCzPTr-2	490//34.31(2Me-THF)	510/71.9/-(mCBP)	0.11	-5.29	-2.95	[29]
3CzTRZ	-/-/-	468/72/415(6 wt. % in PPT)	-	-	-	[32]
DACT-II	-/-/-	-/96/-(TCTA:B3YPMMPM (1:1):DACT-II (7 wt%))				[30]
DACT-II	-/63.7/-(PhMe)	~520/100/-(9 wt% in CBP)	0.009	5.5	3.2	[30]
BCzTrz	-/-/-	-/96/31.2(10 wt% in DPEPO)	0.04	-5.8	-3.24	[13]
TCzTrz	-/-/-	-/99/23.6(10 wt% in DPEPO)	0.01	-5.75	-3.41	[13]
TRZ-DI	506/-/1.61(PhMe)	521/87/1.32(25wt% TCTA:Bepp2)	0.023	5.7	3	[33]

**Table 6.** Summary of device structures and performance

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a)</sup> [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref
IndCzpTr-1	ITO/MoO <sub>3</sub> /TAPC/mCP/EML/DPEPO/TmPyPB/LiF/Al	472	0.17, 0.27	3.8	14.5/23.2/28.1	11.0/~7.5	[29]
IndCzpTr-2	ITO/MoO <sub>3</sub> /TAPC/mCP/EML/DPEPO/TmPyPB/LiF/Al	496	0.23, 0.50	4	30/61.8/82.6	15.3/~11	[29]
3CzTRZ	ITO/HAT-CN/Tris-PCz /10 wt. % 3CzTRZ:PPT/PPT/LiF/Al	480	(0.17, 0.26)	-	10.5/-/-	~8./~4	[32]
DACT-II	ITO/TAPC/TCTA/TCTA:B3PYMPM:7 wt% DACT-II/B3PYMPM/LiF/Al				34.2/121.3/114	26.5/~20	[30]
DACT-II	ITO/TAPC/9 wt% DACT-II:CBP/BAlq/Liq/Al	≈520	-	-	29.6/-/-	NA	[30]
BCzTrz	ITO/PEDOT:PSS/TAPC/mCP/ 10 wt% BCzTrz in DPEPO /TSP01/TBPI/LiF/Al	485	(0.23, 0.42)	-	23.6/46.6/53.3	~15/~6	[13]
TCzTrz	ITO/PEDOT:PSS/TAPC/mCP/ 10 wt% TCzTrz in DPEPO /TSP01/TBPI/LiF/Al	496	(0.20, 0.44)	-	31.8/61.5/86.4	~26/~16	[13]
TRZ-DI	ITO/HATCN/NPB/TAPC/TCTA:Bepp2:25% TRZ-DI/TmPyPB/ LiF/A	526	0.31,0.57	2.6	31.4/91.6/87.5	~30/29	[33]

## 5. Symmetric substitution of carbazole/carbazole dendrons on triphenyltriazine core

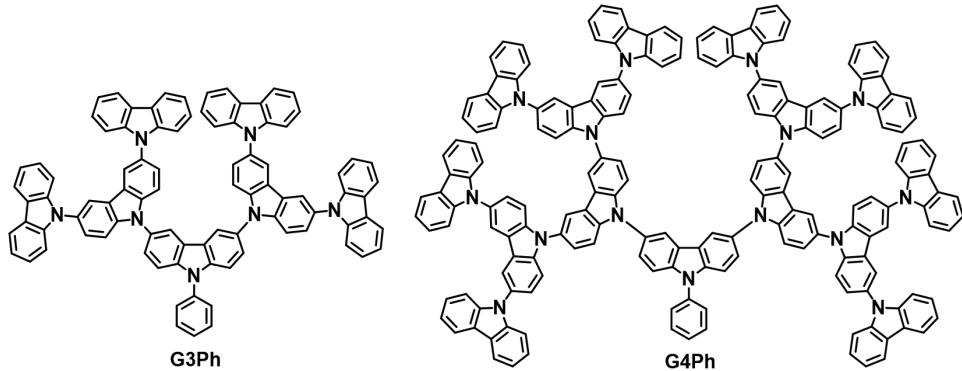


**Figure 14.** Molecular structures and properties of TADF dendrimer emitters with symmetric tri-substitution of carbazole dendrons.

In this section we focus on trisubstituted triazine-based dendrimer emitters. The very first TADF dendrimers, **G<sub>n</sub>TAZ** (n refers to the generation of the dendrons) (**Figure 14**), were developed by Yamamoto *et al.*<sup>[36]</sup> These compounds contain three symmetrical carbazole donor dendrons (from 2<sup>nd</sup> generation to 4<sup>th</sup> generation; 1<sup>st</sup> generation with only Cz as the donor was not studied due to the insolubility of this compound) attached to a central TRZ acceptor. All three dendrimers showed near unity  $\Phi_{PL}$  and very small  $\Delta E_{ST}$  in toluene solution: **G2TAZ** ( $\Delta E_{ST}$ : 0.03 eV;  $\Phi_{PL}$ : 94% in toluene), **G3TAZ** ( $\Delta E_{ST}$ : 0.06 eV;  $\Phi_{PL}$ : 100% in toluene) and **G4TAZ** ( $\Delta E_{ST}$ : 0.06 eV;  $\Phi_{PL}$ : 94% in toluene). However, the  $\Phi_{PL}$  decreased significantly for the neat films to 52%, 31% and 8.5%, respectively, which was rationalized in terms of increasing concentration quenching in the neat films along the series. OLEDs with neat films of **G<sub>n</sub>TAZ** dendrimers as the emitting layer performed poorly, with the highest EQE<sub>max</sub> of 3.4% at CIE coordinates of (0.27, 0.49) for the device with **G3TAZ**.

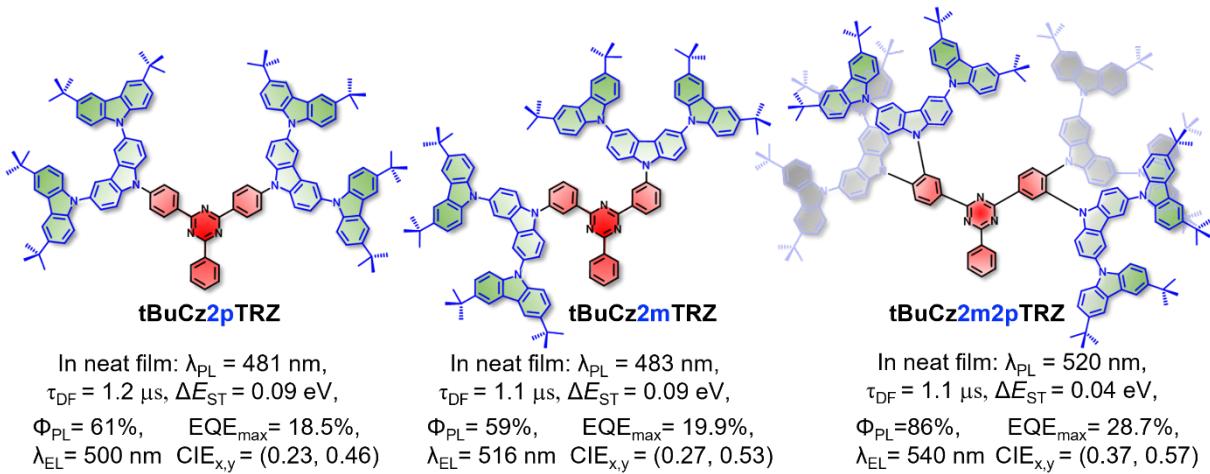
Yamamoto *et al.*<sup>[37]</sup> showed how substitution about the periphery of the donor dendron in **G2TAZ** analogs can affect the  $\Phi_{PL}$ . Compared to **G2TAZ**, methyl (**MeG2TAZ**) ( $\Phi_{PL}$ : 40%;  $\tau_d$ : 3.3  $\mu$ s;  $\Delta E_{ST}$ : 0.07 eV; in neat film), *tert*-butyl (**tBuG2TAZ**) ( $\Phi_{PL}$ : 45%;  $\tau_d$ : 5.3  $\mu$ s;  $\Delta E_{ST}$ : 0.07 eV; in neat film), phenyl (**PhG2TAZ**) ( $\Phi_{PL}$ : 49%;  $\tau_d$ : 1.9  $\mu$ s;  $\Delta E_{ST}$ : 0.08 eV; in neat film), and 7H-dibenzo[c,g]carbazole (**dbG2TAZ**) ( $\Phi_{PL}$ : 9.1%;  $\Delta E_{ST}$ : 0.53 eV; in neat film) analogues all showed similar PL spectra and comparably high  $\Phi_{PL}$  in toluene, with **tBuG2TAZ** showing the highest  $\Phi_{PL}$  of 96%. These dendrimers suffered a similarly lower  $\Phi_{PL}$  as neat films compared to toluene solutions. Apart from **dbG2TAZ**, which has the lowest  $\Phi_{PL}$  in the neat film, all of the other dendrimers have only modestly lower  $\Phi_{PL}$  than **G2TAZ** (52%). However,

the EQE<sub>max</sub> based on the solution-processed OLEDs improved from 6.0% (**G2TAZ**) to 9.4% (**MeG2TAZ**), 9.5% (**tBuG2TAZ**) and 8.2% (**PhG2TAZ**), respectively.



**Figure 15.** Molecular structures of dendrimer host materials **G3Ph** and **G4Ph**

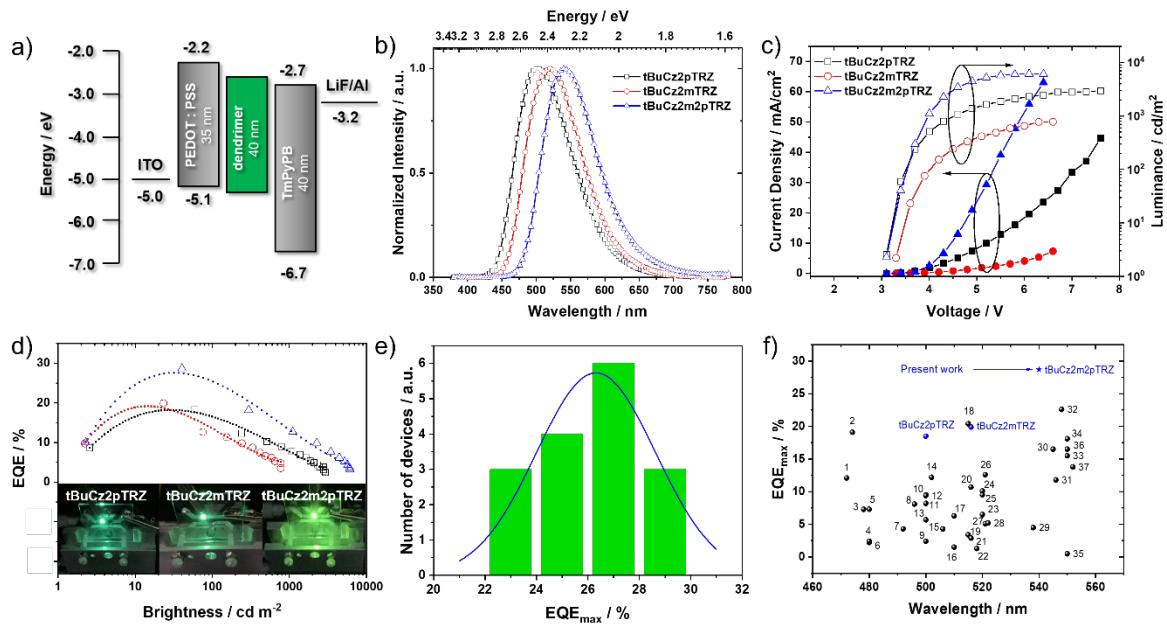
Although the EQE<sub>max</sub> of the OLED based on **tBuG2TAZ** improved to 9.5%, this performance is still not comparable to the efficiencies reported for non-doped vacuum-deposited devices based on small molecule TADF emitters. This is due to the low  $\Phi_{PL}$  (<50%) in the neat film that results from concentration quenching. To solve the problem and further improve the device performance, Yamamoto *et al.*<sup>[38]</sup> developed two carbazole-based dendrimers (**G3Ph** and **G4Ph** as shown in **Figure 15**) as host materials to suppress the observed concentration quenching of **tBuG2TAZ** ( $\lambda_{PL}$ : 500 nm;  $\Phi_{PL}$ : 76%;  $\Delta E_{ST}$ : 0.07 eV; 15 wt% in **G3Ph**). The doping concentration of 15% was chosen to optimize the  $\Phi_{PL}$  in the blended film. The emission spectra remain unchanged at 500 nm between the neat film and the films doped in **G3Ph** and **G4Ph**. However, the  $\Phi_{PL}$  increased from 44% in neat film to 76% and 70% in **G3Ph** and **G4Ph** (15 wt%), respectively. The EQE<sub>max</sub> of the green-emitting OLED was significantly improved to 16.1% at CIE coordinates (0.28, 0.49).



**Figure 16.** Molecular structures and properties of TADF dendrimer emitters with symmetric di-substitution of carbazole dendrons. Reproduced with permission.<sup>[39]</sup>

Recently, our group<sup>[39]</sup> has developed a series of green TADF dendrimers that contain carbazole dendron donors surrounding a TRZ core acceptor unit, linked via either a *para-/meta*-phenylene (**tBuCz2pTRZ/ tBuCz2mTRZ**), or combination of both connections (**tBuCz2m2pTRZ**), as shown in **Figure 16**. The design strategy of **tBuCz2m2pTRZ** ( $\lambda_{PL}$ : 520 nm;  $\Phi_{PL}$ : 86%;  $\Delta E_{ST}$ : 0.04 eV; neat film) demonstrates that **tBuCz2m2pTRZ** is capable of inheriting key properties from both **tBuCz2pTRZ** ( $\lambda_{PL}$ : 481 nm;  $\Phi_{PL}$ : 61%;  $\Delta E_{ST}$ : 0.09 eV; neat film) and **tBuCz2mTRZ** ( $\lambda_{PL}$ : 483 nm;  $\Phi_{PL}$ : 59%;  $\Delta E_{ST}$ : 0.09 eV; neat film), with the excited-state behavior being modulated as a result of the interactions between the two adjacent donor dendrons. The *para*-connection of the donor dendrons in **tBuCz2m2pTRZ** leads to strong electronic coupling between donor and acceptor, as evidenced by the strong molar absorption for the ICT transition, whilst the additional *meta*-connection in **tBuCz2m2pTRZ** results in a small  $\Delta E_{ST}$  as the *meta*-disposed donor and acceptor groups are electronically decoupled. The  $\lambda_{PL}$  of **tBuCz2pTRZ** is similar to that of **tBuCz2mTRZ**, indicating similar energies of their <sup>1</sup>CT states regardless of the electronic coupling between donor and acceptor. The spectral red-shift of **tBuCz2m2pTRZ** indicates the stabilized <sup>1</sup>CT states with more dendritic moieties than **tBuCz2pTRZ** and

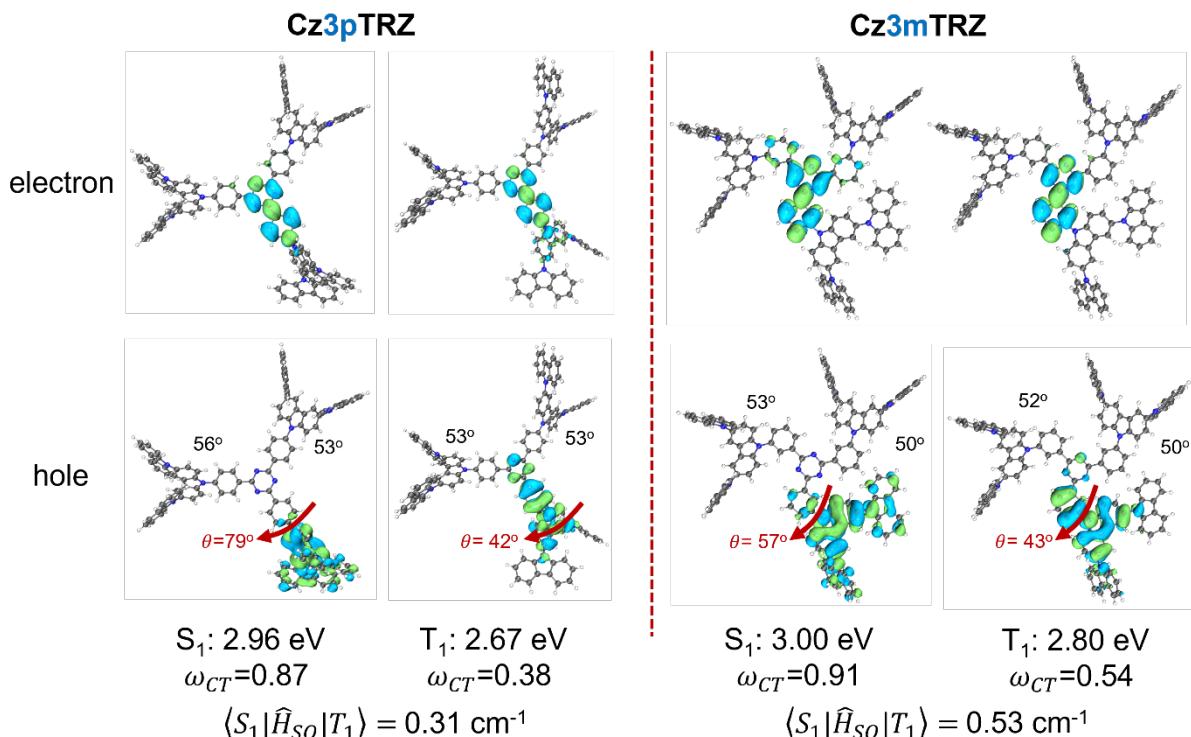
**tBuCz2mTRZ.** Nondoped solution-processed OLEDs using a simple device configuration (**Figure 17**) without exciton barrier layers and containing only the dendrimers in the emissive layer exhibited the  $\text{EQE}_{\max}$  of 18.5%, 19.9% and 28.7% for **tBuCz2pTRZ**, **tBuCz2mTRZ** and **tBuCz2m2pTRZ**, respectively. The performance of the device with **tBuCz2m2pTRZ** represents a step change in the efficiency of the nondoped solution-processed OLEDs. Importantly, the efficiency roll-off of the OLED based on **tBuCz2m2pTRZ** is significantly improved by doping 30 wt% OXD-7, an electron-transporting material, into the emissive layer. As a result of the improved charge balance, the EQE of the optimized device not only reached a similar  $\text{EQE}_{\max}$  of 28.4%, but also maintained its efficiency of 22.7% at a luminance of 500  $\text{cd m}^{-2}$ .



**Figure 17.** Electroluminescence characteristics of host-free OLEDs using **tBuCz2pTRZ**, **tBuCz2mTRZ** and **tBuCz2m2pTRZ** as emitters. (a) Device configuration. (b) Normalized electroluminescence spectra. (c) Current density and luminance versus driving voltage characteristics. (d) EQE versus brightness for **tBuCz2pTRZ** (black), **tBuCz2mTRZ** (red) and **tBuCz2m2pTRZ** (blue) based devices. (The photos shown from bottom left to right for **tBuCz2pTRZ**, **tBuCz2mTRZ** and **tBuCz2m2pTRZ** based devices, respectively) (e) Histogram of maximum EQE (%) for various devices. (f) Scatter plot of maximum EQE (%) vs wavelength (nm) for the present work and **tBuCz2m2pTRZ**. The plot shows numerous data points labeled with numbers corresponding to the devices in (e).

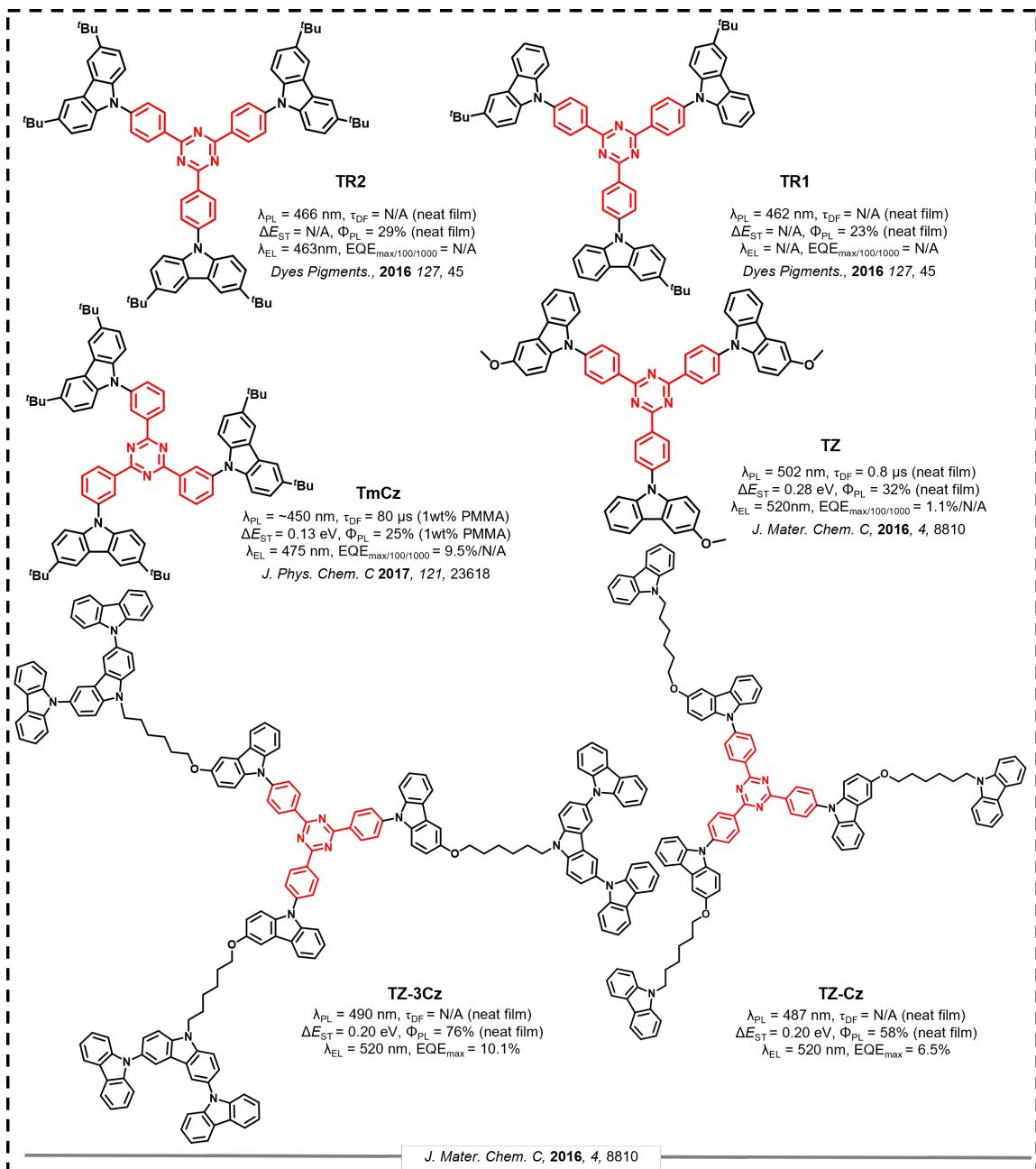
Statistical histogram of  $\text{EQE}_{\max}$  for **tBuCz2m2pTRZ** based OLEDs. (f) The  $\text{EQE}_{\max}$  of all reported solution-processed host-free TADF OLEDs as a function of wavelength.

A related work<sup>[40]</sup> documents a detailed photophysical investigation to rationalize the structure-property relationship of the TADF dendrimers **tBuCz3pTRZ** and **tBuCz3mTRZ** (the DFT optimized structures without peripheral tertbutyl groups are shown in **Figure 18**). Both dendrimers exhibit high  $\Phi_{PL}$  values (89% for **tBuCz3pTRZ** and 81% for **tBuCz3mTRZ**, 10 wt% doped in mCP). While **tBuCz3mTRZ** (2.92 eV) possesses a slightly higher  $S_1$  state than that of **tBuCz3pTRZ** (2.84 eV) due to the weak electron coupling of *meta*-connection, both have a similarly small  $\Delta E_{ST}$  (0.1 eV for **tBuCz3pTRZ** and 0.08 eV for **tBuCz3mTRZ**). The activation energy barriers ( $E_{act}$ ) for RISC are approximately half those of the corresponding  $\Delta E_{ST}$  values. However, the contribution of the delayed emission to the total emission (19% for **tBuCz3pTRZ** and 63% for **tBuCz3mTRZ**) and the  $k_{\text{RISC}}$  ( $0.5 \times 10^5 \text{ s}^{-1}$  for **tBuCz3pTRZ** and  $3.7 \times 10^5 \text{ s}^{-1}$  for **tBuCz3mTRZ**), both reflecting the efficiency of the RISC process, are much greater/faster in the case of the *meta*-connected dendrimer **tBuCz3mTRZ**.



**Figure 18.** Natural transition orbital (NTO) pairs for the S<sub>1</sub> and T<sub>1</sub> states of **Cz3pTRZ** and **Cz3mTRZ**. The spin-orbit coupling (SOC) matrix elements are also provided.

The comparison of the DFT calculation for the model dendrimers without *tert*-butyl groups are shown in **Figure 18**. Unlike **Cz3pTRZ**, **Cz3mTRZ** has negligible overlap between the hole and electron NTO distribution in the T<sub>1</sub> states. This leads to a greater CT character in the T<sub>1</sub> state of **Cz3mTRZ** ( $\omega_{CT} = 0.54$ ) as compared to **Cz3pTRZ** ( $\omega_{CT} = 0.38$ ). The calculated spin orbital coupling (SOC) matrix element was found to be higher for **Cz3mTRZ** ( $0.53 \text{ cm}^{-1}$ ) as compared to **Cz3pTRZ** ( $0.31 \text{ cm}^{-1}$ ). The intramolecular ( $\lambda_{intra}$ ) reorganization energy for **Cz3pTRZ** (275 meV) was also calculated to be higher than that of **Cz3mTRZ** (155 meV). Most TADF molecule design strategies focus on the minimization the  $\Delta E_{ST}$ , however, it is observed that the enhancement of the SOC for the T<sub>1</sub>→S<sub>1</sub> transition and the reduced reorganization energy also significantly contribute to the faster  $k_{RISC}$  in **tBuCz3mTRZ** compared to **tBuCz3pTRZ**, despite their similar  $\Delta E_{ST}$  values. This work demonstrates the importance of the regiochemistry of the donor dendrons on the control of the SOC and reorganization energies, which is a heretofore unexploited strategy that is distinct from the involvement of intermediate triplet states through a non-adiabatic (vibronic) coupling with the lowest singlet charge transfer state.



**Figure 19.** Molecular structures and properties of TADF emitters with tri-substitution of carbazole/carbazole derivatives.

The introduction of *tert*-butyl groups to carbazole donor dendrons in **G1TAZ** was reported by Ulanski *et al.*<sup>[41]</sup> Dendrimers **TR1** and **TR2** (Figure 19) contain one and two *tert*-butyl groups on each carbazole unit, respectively, their inclusion was designed to improve the solubility of the dendrimers. Similar to **Cz-TRZ**, both **TR1** ( $\lambda_{PL}$ : 398 nm;  $\Phi_{PL}$ : 83% in hexane) and **TR2** ( $\lambda_{PL}$ : 407 nm;  $\Phi_{PL}$ : 74% in hexane) are deep blue emitters, with only very short lifetimes of a

few nanoseconds, suggesting that these dendrimers do not show TADF. A subsequent study<sup>[42]</sup> compared the *para*-substituted compound **TR2** (renamed as **TpCz**) with a newly designed *meta*-linked donor dendron dendrimer **TmCz**. The photophysical properties of **TmCz** ( $\lambda_{PL}$ : ~450 nm;  $\Phi_{PL}$ : 25%;  $\tau_d$ : 80  $\mu$ s (1 wt% in PMMA);  $\Delta E_{ST}$ : 0.125 eV) and **TpCz** ( $\lambda_{PL}$ : ~440 nm;  $\Phi_{PL}$ : 35%  $\tau_d$ : 500  $\mu$ s (1 wt% in PMMA);  $\Delta E_{ST}$ : 0.249 eV) revealed that **TpCz** possesses a larger oscillator strength leading to a higher  $\Phi_{PL}$ , but at the expense of a lower triplet energy, stronger charge transfer character and larger  $\Delta E_{ST}$ . A blue vacuum-deposited OLED using **TmCz** as the emitter showed an  $EQE_{max}$  of 9.5% at CIE coordinates of (0.16, 0.23) and  $\lambda_{EL}$  at 475 nm, demonstrating that **TmCz** emits via TADF in the device.

Another design strategy, developed by Sun *et al.*,<sup>[43]</sup> for TADF dendrimers incorporates non-conjugated aliphatic chains linked to distal carbazole moieties that act as host units. Compounds **TZ-Cz** and **TZ-3Cz** (**Figure 19**) are based on a *para*-connected  $C_3$  symmetric carbazole-triazine reference compound, **Tz**. The authors contend that the emissive core is encapsulated by the non-conjugated dendrons, thus suppressing concentration quenching. Similar to **TR1** and **TR2**, **Tz** ( $\lambda_{PL}$ : 502 nm;  $\Phi_{PL}$ : 25%;  $\Delta E_{ST}$ : 0.28 eV; in neat film) is only a fluorescent emitter and does not show TADF. However, TADF turns on in **TZ-Cz** ( $\lambda_{PL}$ : 490 nm;  $\Phi_{PL}$ : 58% in neat film;  $\Delta E_{ST}$ : 0.20 eV) and **TZ-3Cz** ( $\lambda_{PL}$ : 487 nm;  $\Phi_{PL}$ : 76% in neat film;  $\Delta E_{ST}$ : 0.20 eV), which the authors assert is due to the change in local environment that results from the encapsulation. The best solution-processed OLED was achieved using **TZ-3Cz** ( $\lambda_{EL}$ : 520 nm; CIE = (0.24, 0.51)) in a host-free configuration and showed a much-improved  $EQE_{max}$  of 10.1% compared to the OLED based on **Tz** (1.09%). The photophysics and electrochemical characteristics of the aforementioned materials are summarized in **Table 7** and representative device performance metrics are summarized in **Table 8**.

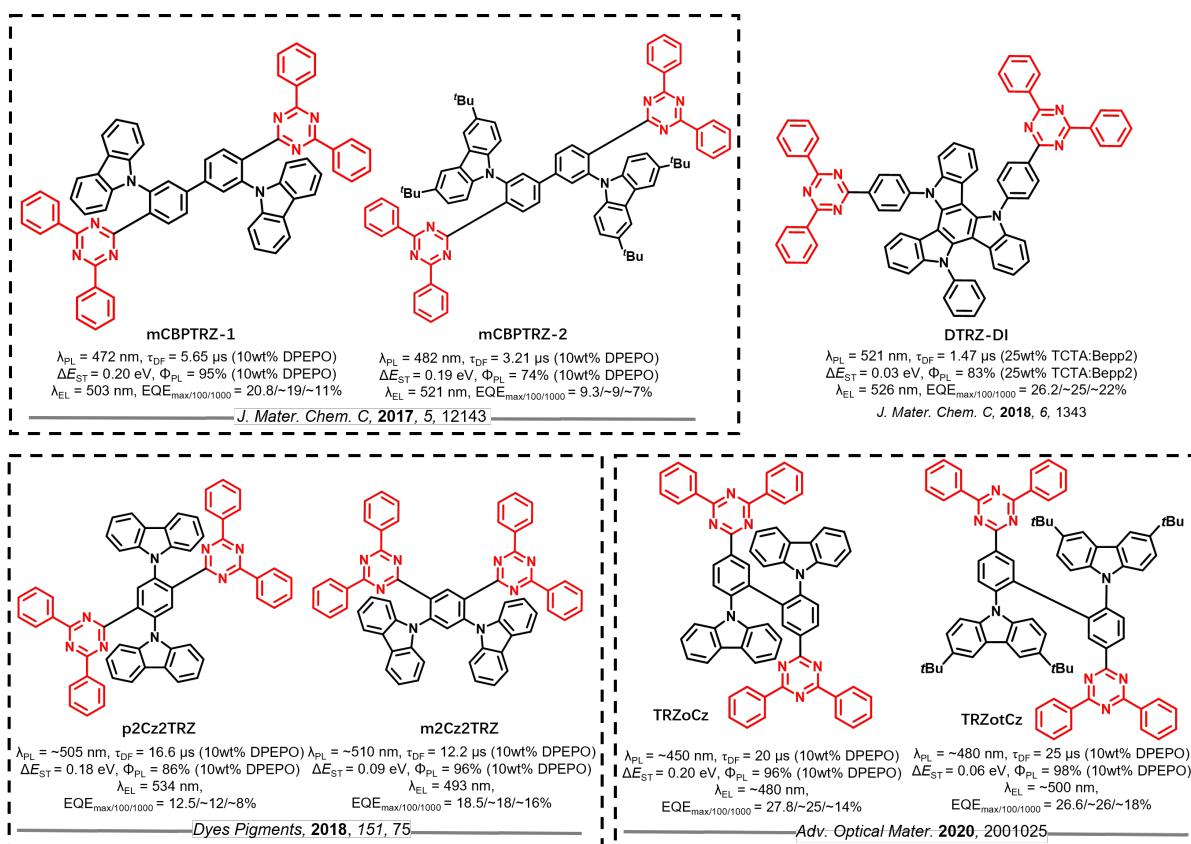
**Table 7.** Summary of photophysical and electrochemical properties

Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
G2TAZ	473/94/-(PhMe)	-/52/-(in neat film)	0.03	-5.76	-3.01	[36]
G3TAZ	473/100/-(PhMe)	-/31/-(in neat film)	0.06	-5.72	-2.97	[36]
G4TAZ	473/94/-(PhMe)	-/8.5/-(in neat film)	0.06	-5.68	-2.80	[36]
MeG2TAZ	-/92/-(PhMe)	-/40/3.3(in neat film)	0.07	-5.73	-2.85	[37]
tBuG2TAZ	-/96/-(PhMe)	-/45/5.3(in neat film)	0.07	-5.79	-2.88	[37]
PhG2TAZ	-/95/-(PhMe)	-/49/1.9(in neat film)	0.08	-5.79	-2.90	[37]
dBG2TAZ	-/86/-(PhMe)	-/9.1/-(in neat film)	0.53	-5.76	-2.99	[37]
TR1	398/83/-(hexane)	462/23/-(neat film)	-	5.49	2.86	[41]
TR2	407/74/-(hexane)	466/29/-(neat film)	-	5.70	2.83	[41]
TmCZ	~475/-/(PhMe)	~450/25/80 (in PMMA)	0.13	-5.65	-2.78	[42]
Tz	535/-/(DCM)	502/32/-(neat )	0.28	-5.11	-2.22	[43]
TZ-Cz	535/-/(DCM)	490/58/-(neat film)	0.20	-5.20	-2.31	[43]
TZ-3Cz	535/-/(DCM)	487/76/-(neat film)	0.20	-5.00	2.11	[43]

**Table 8.** Summary of device structures and performance

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a</sup> [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref
G2TAZ	ITO/PEDOT:PSS/G2TAZ/TPBI/Ca/Al	-	(0.252,0.493)	3.3	2.4/-/-	N/A	[36]
G3TAZ	ITO/PEDOT:PSS/G3TAZ/TPBI/Ca/Al	-	(0.266,0.485)	3.5	3.4/-/-	N/A	[36]
G4TAZ	ITO/PEDOT:PSS/G4TAZ/TPBI/Ca/Al	-	(0.232,0.368)	3.5	1.5/-/-	N/A	[36]
G2TAZ	ITO/PEDOT-PSS/PVK/G2TAZ/TPBi/Ca/Al	-	-	3.7	6.2/-/-	N/A	[37]
MeG2TAZ	ITO/PEDOT-PSS/PVK/TPBi/Ca/Al	MeG2TAZ	-	3.0	9.4/-/-	9.4/N/A	[37]
tBuG2TAZ	ITO/PEDOT-PSS/PVK/TPBi/Ca/Al	tBuG2TAZ	-	3.5	9.5/-/-	9.4/N/A	[37]
PhG2TAZ	ITO/PEDOT-PSS/PVK/TPBi/Ca/Al	PhG2TAZ	-	3.2	8.2/-/-	8.0/N/A	[37]
tBuG2TAZ	A:ITO/PEDOTPSS/15 wt% tBuG2TAZ in Gn3Ph/SPP013/LiF/Al;	503	(0.28, 0.48)	2.9	16.1/19.3/22.7		[38]
tBuG2TAZ	A:ITO/PEDOTPSS/15 wt% tBuG2TAZ in Gn4Ph/SPP013/LiF/Al	508	(0.29, 0.51)	2.9	10.5/27.3/28.6		[38]
TR1	ITO/PEDOT:PSS/PVK:PBD (40 wt%):TR2 (3 wt%)/LiF/Al	463	(0.15, 0.13)	8	-/-/0.51	N/A	[41]
TR2	ITO/PEDOT:PSS/PVK:PBD (40 wt%):TR3 (3 wt%)/LiF/Al	515	(0.21, 0.33)	9.5	-/-/0.46	N/A	[41]
TmCZ	ITO/m-MTDATA/NPB/mCP/5 wt %-TmCZ:DPEPO/TPBi/LiF/Al	475	(0.16, 0.23)	5.0~6.0	9.5/-/-	N/A	[42]
TZ	ITO/PEDOT:PSS/TZ-Cz/TPBI/Cs <sub>2</sub> CO <sub>3</sub> /Al	520	(0.32, 0.51)	4.7	1.09/-/3.4	N/A	[43]
TZ-Cz	ITO/PEDOT:PSS/TZ-Cz/TPBI/Cs <sub>2</sub> CO <sub>3</sub> /Al	520	(0.24, 0.51)	4	6.5/-/20	N/A	[43]
TZ-3Cz	ITO/PEDOT:PSS/TZ-3Cz/TPBI/Cs <sub>2</sub> CO <sub>3</sub> /Al	520	(0.24, 0.51)	3.6	10.1/-/30.5	N/A	[43]

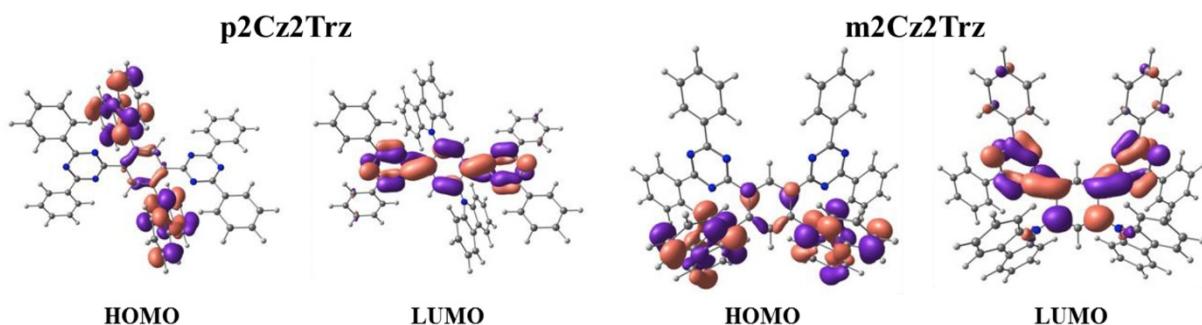
## 6. TADF emitters containing two TRZ and carbazole/carbazole derivative donors



**Figure 20.** Molecular structures and properties of TADF emitters based on two diphenyltriazine TRZ and carbazole/carbazole derivative donors.

There are a small number of examples of emitters containing two triazine acceptors. Many of these examples are based on a dimerization strategy. As shown in **Figure 20**, Lee *et al.*<sup>[44]</sup> reported two related green TADF emitters **mCBPTRZ-1** ( $\lambda_{PL}$ : 472 nm;  $\Phi_{PL}$ : 95%;  $\tau_d$ : 5.65  $\mu\text{s}$ ;  $\Delta E_{ST}$ : 0.20 eV; 10 wt% in DPEPO) and **mCBPTRZ-2** ( $\lambda_{PL}$ : 482 nm;  $\Phi_{PL}$ : 74%;  $\tau_d$ : 3.21  $\mu\text{s}$ ;  $\Delta E_{ST}$ : 0.19 eV; 10 wt% in DPEPO), which are dimers of the TADF emitter ***o*-CzTRZ** ( $\lambda_{PL}$  455 nm;  $\Phi_{PL}$ : 16.7%;  $\tau_d$ : 3.90  $\mu\text{s}$ ;  $\Delta E_{ST}$ : 0.10 eV; 5wt % in mCP). The  $\Phi_{PL}$  of these two compounds are significantly enhanced compared to that of the parent emitter ***o*-CzTRZ** (16.7%). Dimerization is expected to enhance the overlap between frontier molecular orbitals, which leads to an increase of the oscillator strength, thus resulting in higher  $\Phi_{PL}$ , but at the expense of a red-shifted emission along with a larger  $\Delta E_{ST}$ . The OLED using **mCBPTRZ-1** ( $\lambda_{EL}$ : 503 nm; CIE = (0.23, 0.52) emits in the green with an  $EQE_{max}$  of 20.8% while the device using

**mCBPTRZ-2** ( $\lambda_{\text{EL}}$ : 521 nm; CIE = (0.31, 0.59)) that contains the more electron-rich 3,6-di-*tert*-butylcarbazole shows a red-shifted emission and a significantly lower EQE<sub>max</sub> of 9.3% despite of the high  $\Phi_{\text{PL}}$  of the emitter. The low device efficiency was rationalized by the authors as due to the weak exciton conversion efficiency of 41% in the **mCBPTRZ-2**-based device compared to 90% in the **mCBPTRZ-1**-based device. However, this explanation does not seem plausible given their similar molecular structures and photophysics.



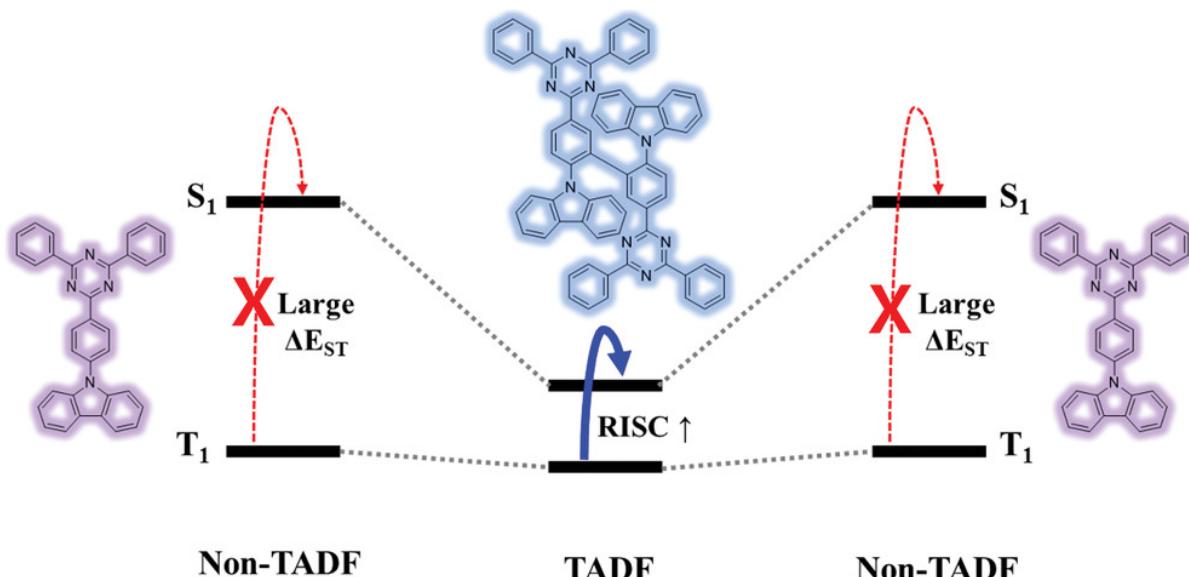
**Figure 21.** Calculated HOMO and LUMO distributions of the **p2Cz2TRZ** and **m2Cz2TRZ**.

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Lee *et al.*[45] also reported two triazine analog emitters to **2CzTPN** and **2CzIPN**, **p2Cz2TRZ** and **m2Cz2TRZ** (Figure 20), which contain *ortho*-disposed carbazole units to the bis(diphenyltriazine) moiety. According to the computed frontier molecular orbitals in Figure 21, the HOMOs of **p2Cz2TRZ** and **m2Cz2TRZ** are evenly distributed over two carbazole donors whereas the LUMOs are evenly distributed over the two TRZ acceptors. A strong overlap between the HOMO and LUMO at the central phenyl moiety contributes to the high  $\Phi_{\text{PL}}$ . Compounds **p2Cz2TRZ** ( $\lambda_{\text{PL}}$ : ~505 nm;  $\Phi_{\text{PL}}$ : 86%;  $\tau_{\text{d}}$ : 16.6  $\mu\text{s}$ ;  $\Delta E_{\text{ST}}$ : 0.18 eV; 10 wt% in DPEPO) and **m2Cz2TRZ** ( $\lambda_{\text{PL}}$ : ~510 nm;  $\Phi_{\text{PL}}$ : 96%;  $\tau_{\text{d}}$ : 12.2  $\mu\text{s}$ ;  $\Delta E_{\text{ST}}$ : 0.09 eV; 10 wt% in DPEPO) each showed green delayed fluorescence, with **m2Cz2TRZ** having a higher  $\Phi_{\text{PL}}$  and a smaller  $\Delta E_{\text{ST}}$  than **p2Cz2TRZ**. The OLED with **m2Cz2TRZ** showed a correspondingly higher EQE<sub>max</sub> of 18.5% coupled with a bluer emission [( $\lambda_{\text{EL}}$ : 493 nm; CIE = (0.20, 0.47)]

compared with the device using **p2Cz2TRZ**, which showed an EQE<sub>max</sub> of 12.5% [ $\lambda_{EL}$ : 534 nm; CIE = (0.39, 0.58)].

Kwon *et al.*<sup>[33]</sup> used a diindolocarbazole as a central donor core to connect two diphenyltriazine moieties (**Figure 20**). Like the diindolocarbazole analog **TRZ-DI** ( $\lambda_{PL}$ : 521 nm;  $\Phi_{PL}$ : 87%;  $\tau_d$ : 1.32  $\mu$ s;  $\Delta E_{ST}$ : 0.02 eV; 25 wt% in TCTA:Bepp2) shown in **Figure 8**, **DTRZ-DI** ( $\lambda_{PL}$ : 521 nm;  $\Phi_{PL}$ : 83%;  $\tau_d$ : 1.47  $\mu$ s;  $\Delta E_{ST}$ : 0.03 eV; 25 wt% in TCTA:Bepp2) showed the same green emission and similar photophysical properties. The introduction of a second TRZ acceptor did not improve the  $\Phi_{PL}$  as other studies<sup>[44, 46]</sup> had documented. The performance of the OLED based on **DTRZ-DI** is poorer with an EQE<sub>max</sub> of 26.2% [ $\lambda_{EL}$ : 526 nm; CIE = (0.32, 0.58)] in contrast to that for the **TRZ-DI** device where the EQE<sub>max</sub> is 31.4%.



**Figure 22.** Design strategy for the TADF molecule **TRZoCz**. Reproduced with permission. <sup>[46]</sup>

The dimerization strategy (**Figure 22**) could also be used to convert a non-TADF emitter into a TADF emitter, as Lee *et al.*<sup>[46]</sup> demonstrated by transforming the fluorescent emitter **Cz-TRZ** into a TADF emitters (**TRZoCz** and **TRZotCz**) by increasing the separation of the FMOs in

the dimer structures. The highly twisted geometries in **TRZoCz** and **TRZotCz** lead to markedly smaller  $\Delta E_{ST}$  (0.20 eV for **TRZoCz**, 0.06 eV for **TRZotCz**) compared to **Cz-TRZ** (0.36 eV). The  $\Phi_{PL}$  of **TRZoCz** and **TRZotCz** are 96% and 98%, respectively, for the 10 wt% films in DPEPO. The PL spectra are red-shifted compared to that of **Cz-TRZ** due to the increased conjugation within their structures, the photophysics of the two compounds are also less sensitive to the doping concentration. The  $EQE_{max}$  of the **TRZoCz** and **TRZotCz**-based OLEDs are 27.8% and 26.6%, respectively, at CIE coordinates of (0.15, 0.32) and (0.20, 0.51), respectively. It is noteworthy that the emission profiles of these two emitters are not as broad than mostly common TADF emitters ( $FWHM = 90\text{-}110$  nm), with  $FWHM$  for **TRZoCz** (67 nm) and **TRZotCz** (74 nm). This is due to the restricted conformational motion available in these compounds. The photophysics and electrochemical characteristics of the aforementioned materials are summarized in **Table 9**. The representative device performance is summarized in **Table 10**.

**Table 9.** Summary of photophysical and electrochemical properties

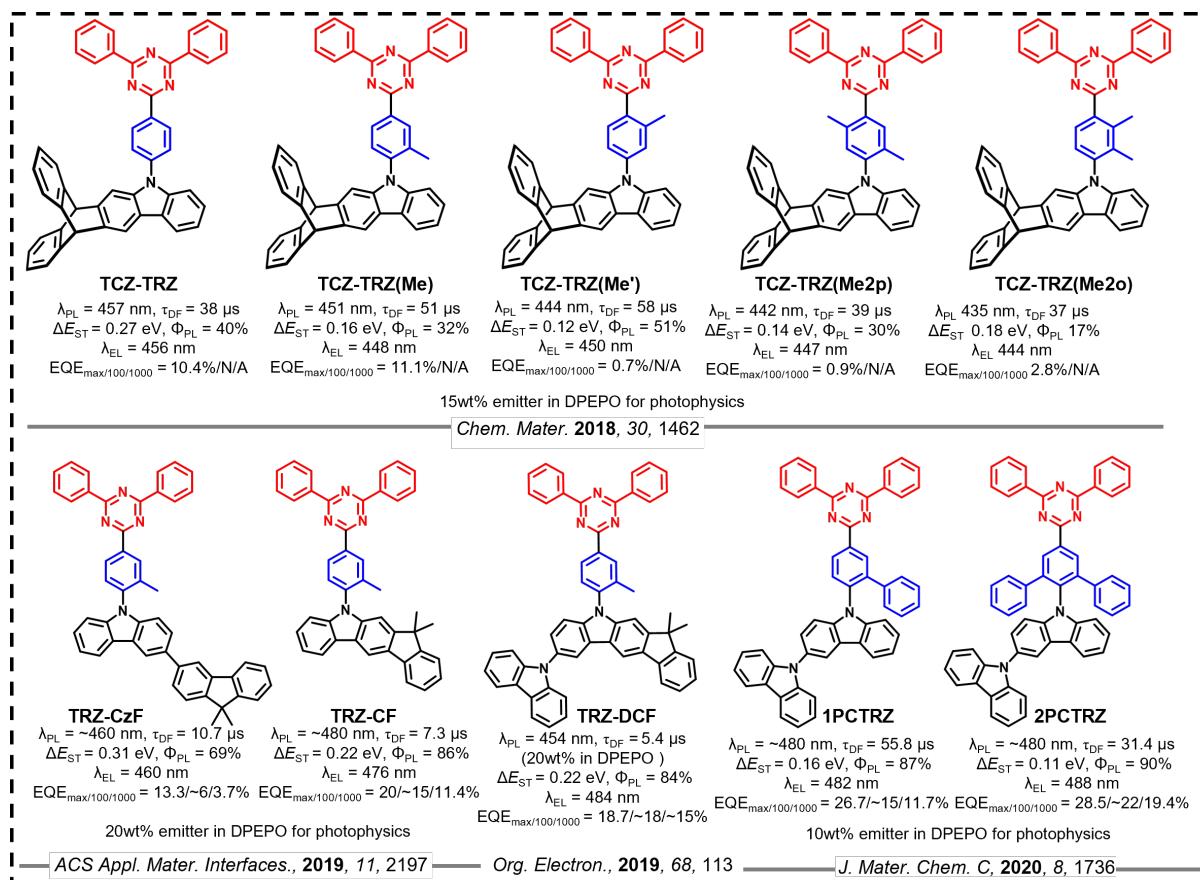
Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
mCBPTrz-1	~-/-(THF)	472/95/5.65(10 wt% in DPEPO )	0.20	-6.11	-3.32	[44]
mCBPTrz-2	~-/-(THF)	482/74/3.21(10 wt% in DPEPO )	0.19	-6.01	-3.22	[44]
p2Cz2Trz	~/-/-	~505/86.3/16.6(10 wt% in DPEPO)	0.18	-5.96	-4.16	[45]
m2Cz2Trz	~/-/-	~510/95.6/12.2(10 wt% in DPEPO)	0.09	-5.92	-3.33	[45]
DTRZ-Di	506/-/2.48(PhMe)	521/83/1.47(25wt% in TCTA:Bepp2 )	0.03	-5.68	-2.98	[33]
TRZoCz	~450/-/(PhMe)	~96/20(10% in DPEPO )	0.20	-6.03	-3.30	[46]
TRZotCz	~480/-/(PhMe)	~98/25 (10% in DPEPO )	0.06	-5.9.	-3.17	[46]

**Table 10.** Summary of device structures and performance

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a</sup> ) [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref.
mCBPTrz-1	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:mCBP Trz-1/TSPO1/TPBi/LiF/Al	503	(0.23, 0.52)	-	20.8/-	~19.0/~11.0	[44]
mCBPTrz-2	ITO/PEDOT:PSS/TAPC/mCP/DPEPO: mCBPTrz-2/TSPO1/TPBi/LiF/Al	521	(0.31, 0.59)	-	9.3/-	~9.0/~7.0	[44]
p2Cz2Trz	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30% p2Cz2Trz/TSPO1/TPBi/LiF/Al	534	(0.39,0.58)	-	12.5/-	~12.0/~8.0	[45]
m2Cz2Trz	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:30% m2Cz2Trz)/TSPO1/TPBi/LiF/Al	493	(0.20, 0.47)	-	18.5/-	~18.0/~16.0	[45]
DTRZ-Di	ITO/HATCN/NPB/TAPC/TCTA:Bepp2:25% TADF emitter/TmPyPB/ LiF/A	526	(0.32,0.58)	-	26.2/80.2/73.9	~25.0/~22.0	[33]
TRZoCz	ITO/(PEDOT:PSS)/(TAPC/mCP/DPEPO: 20% emitter/TSPO1/ TPBi/ LiF/A	478- 484	(0.15, 0.32)	-	27.8/-	~25.0/~14.0	[46]

TRZotCz	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20% emitter /TSPoI/ TPBi/ LiF/A	497– 501	(0.20, 0.51), -	26.6/-/-	~26.0/~18.0	[46]
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## 7. Modification to the Phenylene Bridge



**Figure 23.** Molecular structures and properties of TRZ based TADF emitters with modification to the phenylene bridge.

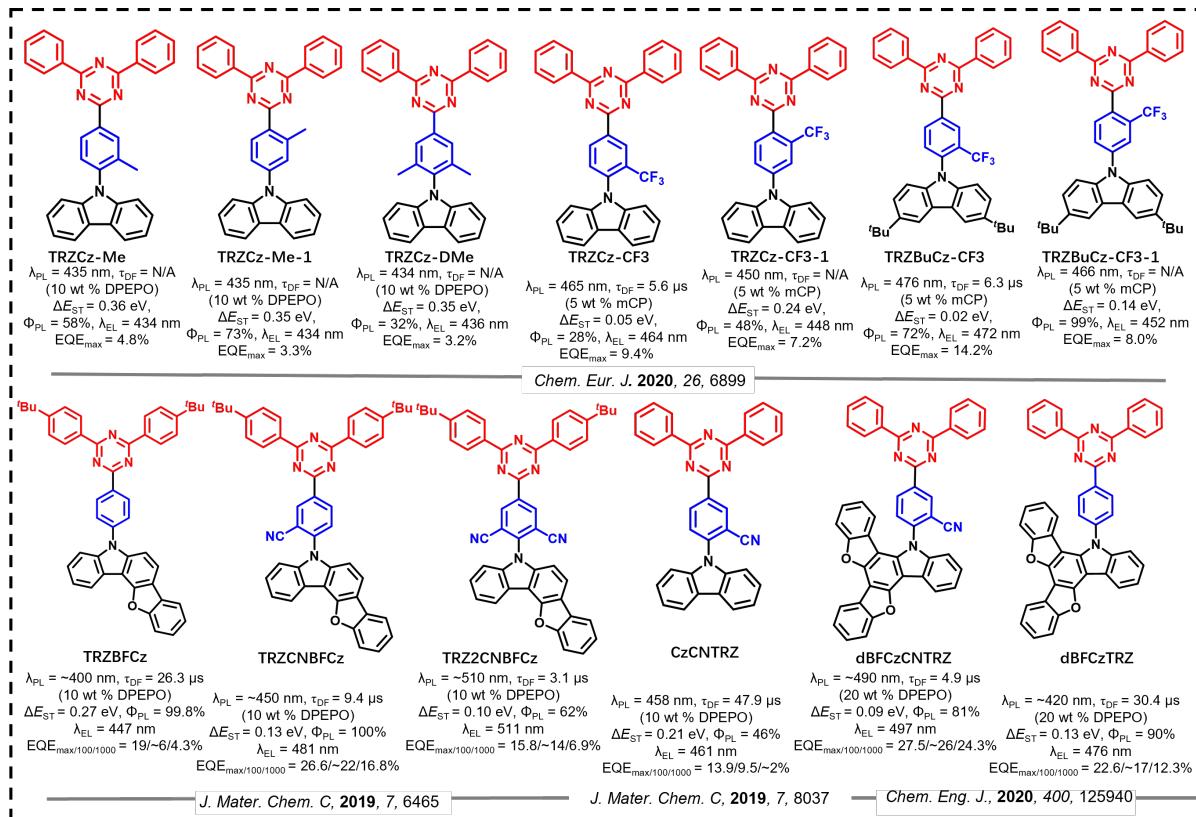
The degree of orbital overlap between the HOMO and LUMO is controlled not only by the choice of donor and acceptor but also by the bridging unit that mediates the conjugation between the two. For example, the introduction of a methyl group or phenyl group onto the phenylene bridge affects one of the torsion angles, leading to a more twisted conformation and thus a smaller  $\Delta E_{ST}$ , while the inclusion of more strongly electron-withdrawing groups such as nitrile or trifluoromethyl substituents also contributes to strengthening the electron-acceptor and localizing the electron density of the LUMO.

Buchwald *et al.*<sup>[47]</sup> reported a series of TADF emitters (**Figure 23**) that incorporate a triptycene-fused carbazole donor *para*-linked to a diphenyltriazine acceptor via a arylene bridge. The dihedral angle between donor and the bridging arylene was manipulated by controlling the position and number of methyl substituents. Compared with **TCZTRZ** ( $\lambda_{PL}$ : 457 nm;  $\Phi_{PL}$ : 40%;  $\tau_d$ : 38  $\mu$ s;  $\Delta E_{ST}$ : 0.27 eV; 15 wt% in DPEPO), the introduction of a methyl group on the arylene bridge at a position adjacent to TRZ in **TCZTRZ(Me')** ( $\lambda_{PL}$ : 444 nm;  $\Phi_{PL}$ : 51%;  $\tau_d$ : 58  $\mu$ s;  $\Delta E_{ST}$ : 0.12 eV; 15 wt% in DPEPO) results in a higher  $\Phi_{PL}$  along with a reduced  $\Delta E_{ST}$  due to the increased angle (25.1°) between the arylene plane and the TRZ plane despite the identical angle (51.5°) between the plane of triptycene-fused carbazole and the arylene plane. Placement of the methyl group close to the donor elicits a similar outcome. Emitter **TCZTRZ(Me)** ( $\lambda_{PL}$ : 451 nm;  $\Phi_{PL}$ : 32%;  $\tau_d$ : 51  $\mu$ s; 15 wt% in DPEPO;  $\Delta E_{ST}$ : 0.16 eV) also possesses a twisted structure with the torsion angle increased from 51.5° in **TCZTRZ** to 69.5° in **TCZTRZ(Me)**. The  $\Delta E_{ST}$  of **TCZTRZ(Me)** is smaller as well at 0.16 eV; however, at a cost of a lower  $\Phi_{PL}$  of 32%. The addition of a second methyl group on the arylene bridge in **TCZTRZ(Me2p)** ( $\lambda_{PL}$ : 442 nm;  $\Phi_{PL}$ : 30%;  $\tau_d$ : 39  $\mu$ s (15 wt% in DPEPO);  $\Delta E_{ST}$ : 0.14 eV) and **TCZTRZ(Me2o)** ( $\lambda_{PL}$ : 435 nm;  $\Phi_{PL}$ : 17%;  $\tau_d$ : 37  $\mu$ s (15 wt% in DPEPO);  $\Delta E_{ST}$ : 0.18 eV) results in compounds showing a blue-shifted emission, shorter  $\tau_d$  and lower  $\Phi_{PL}$  than **TCZTRZ(Me)**. The OLEDs employing **TCZTRZ** [ $\lambda_{EL}$ : 456 nm; CIE = (0.16, 0.14)] and **TCZTRZ(Me)** [ $\lambda_{EL}$ : 448 nm; CIE = (0.17, 0.18)] showed EQE<sub>max</sub> of 10.4 and 11.1%, respectively, while the devices using **TCZTRZ(Me')**, **TCZTRZ(Me2p)** and **TCZTRZ(Me2o)** as emitters showed much lower EQE<sub>max</sub> values of 0.7%, 0.9% and 2.8%, respectively. The author attributed the inferior performance of those devices to the weak absorption, low  $\Phi_{PLS}$ , and rapid degradation of the emitters under electrical excitation.

Liao *et al.*<sup>[48]</sup> reported two blue TADF materials **TRZ-CF** ( $\lambda_{PL}$ : ~480 nm;  $\Phi_{PL}$ : 85.9%;  $\tau_d$ : 7.34  $\mu$ s;  $\Delta E_{ST}$ : 0.22 eV; 20 wt% in DPEPO) and **TRZ-CzF** ( $\lambda_{PL}$ : ~460 nm;  $\Phi_{PL}$ : 68.6%;  $\tau_d$ : 10.68  $\mu$ s;  $\Delta E_{ST}$ : 0.31 eV; 20 wt% in DPEPO) that likewise incorporate a methyl group on the arylene bridge at a position *ortho* to the donor (**Figure 23**), resembling the structure of **TCZTRZ(Me)**. However, the increased oscillator strength caused by the enhanced conjugation of the donors results in a higher  $\Phi_{PL}$  in these two compounds compared to **TCZTRZ(Me)**, but at a cost of larger  $\Delta E_{ST}$ , indicating that the methyl group is not sufficiently bulky to induce a highly twisted structure that can result in the required separation of the FMOs. Despite the red-shifted PL spectra, the OLEDs still exhibited blue emission with  $\lambda_{EL}$  at 476 and 460 nm and  $EQE_{max}$  of 20% and 13.3% for the devices with **TRZ-CF** and **TRZ-CzF**, respectively. Strengthening the donor by incorporating a distal carbazole served to red-shift the emission in **TRZ-DCF** ( $\Phi_{PL}$ : 84%;  $\tau_d$ : 5.43  $\mu$ s; 20 wt% in DPEPO) while the other photophysical properties remain unchanged.<sup>[49]</sup> Compared to the OLED with **TRZ-CF** [ $\lambda_{EL}$ : 476 nm; CIE = (0.17, 0.27)], a slightly lower  $EQE_{max}$  of 18.7% was obtained for the sky-blue device with **TRZ-DCF** [ $\lambda_{EL}$ : 484 nm; CIE = (0.20, 0.33)].

The use of a phenyl substituent on the arylene bridge was explored by Lee *et al.*<sup>[50]</sup> to modulate the conformation of the emitter. A cross-comparison of the photophysical properties of **1PCTRZ** ( $\Phi_{PL}$ : 87%;  $\tau_d$ : 55.8  $\mu$ s;  $\Delta E_{ST}$ : 0.16 eV; 10 wt% in DPEPO) and **2PCTRZ** ( $\Phi_{PL}$ : 90%;  $\tau_d$ : 31.4  $\mu$ s;  $\Delta E_{ST}$ : 0.11 eV; 10 wt% in DPEPO) reveals how *ortho*-disubstitution leads to a more twisted conformation (dihedral angle for **1PCTRZ** of 60.2° is smaller than that of **2PCTRZ** of 68.2°), which translates into a smaller  $\Delta E_{ST}$  and a shorter  $\tau_d$ , but without negatively impacting  $\Phi_{PL}$ . However, the donor group adopts a less twisted conformation than those of **TCZTRZ(Me)**, **TCZTRZ(Me2p)** and **TCZTRZ(Me2o)**, all of which contain methyl substituents. This suggests that the phenyl group is effectively less bulky than the methyl substituent. High

$\text{EQE}_{\max}$  values were obtained for the OLEDs with **1PCTRZ** [ $\text{EQE}_{\max}=26.7\%$ ;  $\lambda_{\text{EL}}$ : 476 nm; CIE = (0.17, 0.27)] and **2PCTRZ** [ $\text{EQE}_{\max}=28.5\%$ ;  $\lambda_{\text{EL}}$ : 488 nm; CIE = (0.18, 0.35)].



**Figure 24.** Molecular structures and properties of TRZ based TADF emitters with modification to the phenylene bridge.

Isosteric trifluoromethyl groups have also been introduced onto the arylene bridge in a series of blue TADF emitters: **TRZCz-Me**, **TRZCz-Me-1**, **TRZCz-DMe**, **TRZCz-CF3**, **TRZCz-CF3-1**, **TRZBuCz-CF3**, and **TRZBuCz-CF3-1**, as shown in **Figure 24**.<sup>[51]</sup> Only **TRZCz-CF3** ( $\lambda_{\text{PL}}$ : 465 nm;  $\Phi_{\text{PL}}$ : 27.8%;  $\tau_{\text{d}}$ : 5.62  $\mu$ s in 5 wt% in mCP;  $\Delta E_{\text{ST}}$ : 0.05 eV) and **TRZBuCz-CF3** ( $\lambda_{\text{PL}}$ : 476 nm;  $\Phi_{\text{PL}}$ : 72%;  $\tau_{\text{d}}$ : 6.3  $\mu$ s in 5 wt% in mCP;  $\Delta E_{\text{ST}}$ : 0.02 eV), with the trifluoromethyl group located adjacent to the carbazole donor, exhibit TADF. By comparison, **TRZCz-Me** ( $\lambda_{\text{PL}}$ : 435 nm;  $\Phi_{\text{PL}}$ : 57.5%; in 10 wt% in mCP;  $\Delta E_{\text{ST}}$ : 0.36 eV), and **TRZCz-DMe** ( $\lambda_{\text{PL}}$ : 434 nm;  $\Phi_{\text{PL}}$ : 31.8%; in 10 wt% in mCP;  $\Delta E_{\text{ST}}$ : 0.35 eV) with methyl groups similarly *ortho*-disposed to

carbazole are not TADF emitters, while similar analogs **TRZ-CF** and **TRZ-CzF** with different donor structures to the above work demonstrated TADF. The  $\Phi_{PL}$  was not completely dependent on the position of substituent groups on the bridge, for example, as evidenced by the large deviation between **TRZCz-CF3-1** ( $\Phi_{PL}$  48%) and **TRZBuCz-CF3-1** ( $\Phi_{PL}$  99%), the latter of which also contains a more strongly electron-donating 3,6-ditertbutylcarbazole donor. The blue OLEDs fabricated with **TRZCz-CF3** and **TRZBuCz-CF3** achieved  $EQE_{max}$  of 9.4% and 14.2% with CIE coordinates of (0.19, 0.23) and (0.18, 0.29), respectively.

Another strong electron-withdrawing group, CN, was used in lieu of the trifluoromethyl groups in a series of emitters reported by Hong *et al* (**Figure 24**).<sup>[52]</sup> Compared to **TRZCz-CF3**, **CzCNTRZ** ( $\lambda_{PL}$ : 458 nm;  $\Phi_{PL}$ : 46%;  $\tau_d$ : 47.9  $\mu$ s;  $\Delta E_{ST}$ : 0.21 eV; 10 wt% in DPEPO) shows a similar emission maximum and improved  $\Phi_{PL}$  but at the cost of a larger  $\Delta E_{ST}$  of 0.21 eV. The similar emission was attributed to the identical electron-withdrawing ability of the two electron-withdrawing groups; however, the larger  $\Delta E_{ST}$  is indicative of a smaller steric bulk from the CN substituent. Thanks to the higher  $\Phi_{PL}$ , the  $EQE_{max}$  of the **CzCNTRZ**-based device reached 13.9%. Similar to the PL spectrum, the **CzCNTRZ** device showed blue ( $\lambda_{EL}$  = 461 nm) emission with CIE coordinates of (0.15, 0.16), which resembles the emission profile of the **TRZCz-CF3** device ( $\lambda_{EL}$  = 464 nm).

Hong *et al.*<sup>[53]</sup> also examined the effect of incorporation of a CN unit acting as a secondary electron acceptor on the TADF properties (**Figure 24**). The addition of the nitrile intensified the CT character of the emitters due to the enhanced acceptor strength, evidenced by the red-shifted emission spectrum. Compared to **TRZBFCz** ( $\Phi_{PL}$ : 99.8%;  $\tau_d$ : 26.3  $\mu$ s in 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.27 eV), the  $\Delta E_{ST}$  is reduced yet the  $\Phi_{PL}$  is maintained at 100% in

**TRZCNBFCz** ( $\Phi_{PL}$ : 100%;  $\tau_d$ : 9.4  $\mu$ s in 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.13 eV), enabling a high  $EQE_{max}$  of 26.6% in the sky-blue TADF OLED ( $\lambda_{EL}$ : 481 nm). However, the addition of two CN substituents in **TRZ2CNBFCz** resulted in both a lower  $\Phi_{PL}$  of 62% and a smaller  $\Delta E_{ST}$  of 0.1 eV ( $\Phi_{PL}$ : 62%;  $\tau_d$ : 31  $\mu$ s in 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.1 eV), producing a green device ( $\lambda_{EL}$ : 511 nm) with a lower  $EQE_{max}$  of 15.8%. The OLED with the control compound **TRZBFCz** that does not have any CN groups on the arylene bridge showed an  $EQE_{max}$  of 19% and a  $\lambda_{EL}$  of 447 nm with color coordinates of (0.15, 0.09).

Lee *et al.*<sup>[54]</sup> reported **dBFCzTRZ** ( $\lambda_{PL}$ : 420 nm;  $\Phi_{PL}$ : 90%;  $\tau_d$ : 30.4  $\mu$ s;  $\Delta E_{ST}$ : 0.13 eV; 20 wt% in DPEPO) and **dBFCzCNTRZ** ( $\lambda_{PL}$ : 490 nm;  $\Phi_{PL}$ : 81%;  $\tau_d$ : 4.9  $\mu$ s;  $\Delta E_{ST}$ : 0.09 eV; 20 wt% in DPEPO) that each employ a more conjugated donor and where the only difference between the two comes from the inclusion of the CN group on the phenylene bridge in **dBFCzCNTRZ** (**Figure 24**). Calculations showed that the HOMOs of both **dBFCzTRZ** (-5.25 eV) and **dBFCzCNTRZ** (-5.40 eV) are mainly localized on the donor moiety while the LUMOs of **dBFCzTRZ** (-2.78 eV) and **dBFCzCNTRZ** (-3.06 eV) are distributed on both the arylene linker and the triazine acceptor. The CN unit in **dBFCzCNTRZ** promotes a stabilization of both the HOMO and LUMO levels, but the LUMO more pronouncedly. The dihedral angles between the donor moiety and the arylene bridge in **dBFCzCNTRZ** and **dBFCzTRZ** are similar at 75.1 and 71.1°, respectively, indicating the negligible steric influence from the nitrile. The S<sub>1</sub>/T<sub>1</sub> energies of **dBFCzCNTRZ** (2.97/2.88 eV) are both more stabilized than those of **dBFCzTRZ** (3.15/3.02 eV). Despite the similar  $\Delta E_{ST}$ , the  $\tau_d$  for **dBFCzCNTRZ** is much faster at 4.9  $\mu$ s compared to 30.4  $\mu$ s for **dBFCzTRZ**. There is a small sacrifice in  $\Phi_{PL}$  of 81% for **dBFCzCNTRZ** compared to 90% for **dBFCzTRZ**. The OLEDs based on **dBFCzCNTRZ** and **dBFCzTRZ** demonstrated high  $EQE_{max}$  of 27.5% and 22.6%, respectively. **dBFCzCNTRZ** with the lower  $\Phi_{PL}$ , however, produced a higher efficiency device which may be attributed to

the better electron mobility resulting from the presence of the nitrile group. The operational lifetime ( $LT_{80}$ ), defined as the lifetime of the device to reach 80% of its initial luminance, of the **dBFCzCNTRZ** device [ $\lambda_{EL}$ : 497 nm;  $LT_{80}$ : ~36 h; CIE = (0.22, 0.47)] was over three times longer than that of the **dBFCzTRZ** device [ $\lambda_{EL}$ : 476 nm;  $LT_{80}$ : ~11 h; CIE = (0.16, 0.27)]. The high  $k_{RISC}$  of the **dBFCzCNTRZ** emitter contributed to the enhanced device lifetime by suppressing triplet exciton triggered degradation mechanisms. The photophysics and electrochemical characteristics of the aforementioned materials are summarized in **Table 11** and representative device performance is summarized in **Table 12**.

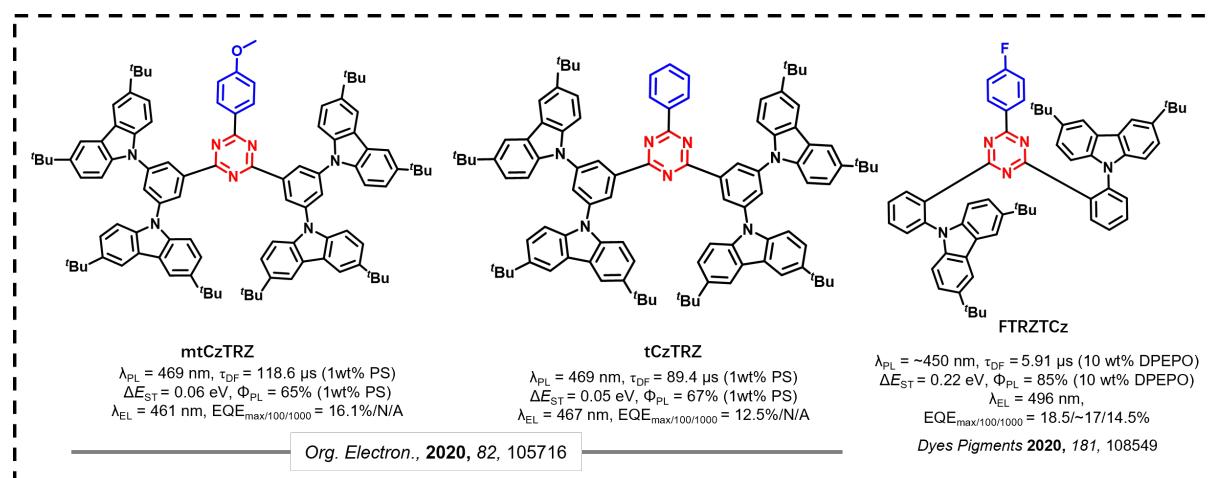
**Table 11.** Summary of photophysical and electrochemical properties

Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
<b>TCZ-TRZ</b>	432/77/-(PhMe)	457/40/38 (15wt% in DPEPO)	0.27	-5.64	-2.57	[47]
<b>TCZ-TRZ(Me)</b>	431/60/-(PhMe)	451/32/51 (15wt% in DPEPO)	0.16	-5.64	-2.54	[47]
<b>TCZ-TRZ(Me')</b>	429/80/-(PhMe)	444/51/58 (15wt% in DPEPO)	0.12	-5.60	-2.52	[47]
<b>TCZ-TRZ(Me2p)</b>	427/46/-(PhMe)	442/30/39 (15wt% in DPEPO)	0.14	-5.62	-2.46	[47]
<b>TCZ-TRZ(Me2o)</b>	427/47/-(PhMe)	435/17/37 (15wt% in DPEPO)	0.18	-5.59	-2.33	[47]
<b>TRZ-CF</b>	452/~/-(PhMe)	~480/85.9/7.34 (20 wt % in DPEPO)	0.22	-5.34	-2.37	[48]
<b>TRZ-CzF</b>	445/~/-(PhMe)	~460/68.6/10.68 (20 wt % in DPEPO)	0.31	-5.47	-2.42	[48]
<b>TRZ-DCF</b>	454/~/-(PhMe)	~/84/5.43 (20 wt % in DPEPO)	0.22	-5.36	-2.34	[49]
<b>1PCTrz</b>	~/~/--	~/87/55.8 (10 wt% in DPEPO)	0.16	-5.49	-2.51	[50]
<b>2PCTrz</b>	~/~/--	~/90/31.4 (10 wt% in DPEPO)	0.11	-5.50	-2.45	[50]
<b>TrzCz-Me</b>	414/~/-(PhMe)	435/57.5/-(10 wt% in DPEPO)	0.36	-5.66	-2.92	[51]
<b>TrzCz-Me-1</b>	414/~/-(PhMe)	435/73/-(10 wt% in DPEPO)	0.35	-5.64	-2.94	[51]
<b>TrzCz-Dme</b>	414/~/-(PhMe)	434/31.8/-(10 wt% in DPEPO)	0.35	-5.66	-2.90	[51]
<b>TrzCz-CF3</b>	457/~/-(PhMe)	465/27.8/5.62(5 wt% in mCP)	0.05	-5.76	-3.10	[51]
<b>TrzCz-CF3-1</b>	443/~/-(PhMe)	450/48.1/-(5 wt% in mCP)	0.24	-5.73	-3.05	[51]
<b>TrzBuCz-CF3</b>	476/~/-(PhMe)	476/72/6.3 (5 wt% in mCP)	0.02	-5.61	-3.12	[51]
<b>TrzBuCz-CF3-1</b>	466/~/-(PhMe)	466/99/-(5 wt% in mCP)	0.14	-5.60	-3.06	[51]
<b>TrzBFCz</b>	~400/~/-(THF)	~/99.8/26.3(10 wt% in DPEPO)	0.27	-6.04	-3.21	[53]
<b>TrzCNBFCz</b>	~450/~/-(THF)	~/100/9.4(10 wt% in DPEPO)	0.13	-6.16	-3.41	[53]
<b>Trz2CNBFCz</b>	~510/~/-(THF)	~/62/3.1(10 wt% in DPEPO)	0.10	-6.37	-4.08	[53]
<b>CzCNTrz</b>	~/~/--	458/46/47.9(10 wt% in DPEPO)	0.21	-6.27	-3.24	[52]
<b>dBFCzCNTrz</b>	~/~/--	~/0.81/4.9 (20 wt% in DPEPO)	0.09	-6.17	-3.06	[54]
<b>dBFCzTrz</b>	~/~/--	~/0.9/30.4 (20 wt% in DPEPO)	0.13	-6.17	-2.78	[54]

**Table 12.** Summary of device structures and performance

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a)</sup> [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref.
<b>TCZ-TRZ</b>	ITO/HAT-CN/NPB/mCP/DPEPO:15 wt%emitter/PPF/PPF:Liq/Liq/Al	456	(0.159,0.142)	3.8	10.4/-/11.6	N/A	[47]
<b>TCZ-TRZ(Me)</b>	ITO/HAT-CN/NPB/mCP/DPEPO:15 wt%emitter/PPF/PPF:Liq/Liq/Al	448	(0.170,0.179)	4	11.1/-/12.8	N/A	[47]
<b>TCZ-TRZ(Me')</b>	ITO/HAT-CN/NPB/mCP/DPEPO:15 wt%emitter/PPF/PPF:Liq/Liq/Al	450	(0.203,0.229)	4.4	0.7/-/1.0	N/A	[47]
<b>TCZ-TRZ(Me2p)</b>	ITO/HAT-CN/NPB/mCP/DPEPO:15 wt%emitter/PPF/PPF:Liq/Liq/Al	447	(0.197,0.214)	5	0.9/-/1.0	N/A	[47]
<b>TCZ-TRZ(Me2o)</b>	ITO/HAT-CN/NPB/mCP/DPEPO:15 wt%emitter/PPF/PPF:Liq/Liq/Al	444	(0.228,0.289)	4.6	2.8/-/2.7	N/A	[47]
<b>TRZ-CF</b>	ITO/HATCN/TAPC/TCTA/mCP/DPEPO: 20 wt %TRZ-CF/DPEPO/TmPyPB/Liq/Al	476	(0.17, 0.27)	2.9	20/28/36	~15/11.4	[48]
<b>TRZ-CzF</b>	ITO/HATCN/TAPC/TCTA/mCP/DPEPO: 20 wt % TRZ-CzF/DPEPO/TmPyPB/Liq/Al	460	(0.15, 0.16)	3.5	13.3/10.7/16.5	~6.0/3.7	[48]
<b>TRZ-DCF</b>	ITO/HAT-CN/TAPC/mCP/DPEPO: TRZ-DCF 20%/DPEPO/TmPyPB/Liq/Al	484	(0.20, 0.33)	-	18.7/30.5/39	~18.0/~15	[49]
<b>1PCTrz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20 % /TSP01/ TPBi/LiF/A	482	(0.17,0.30)	-	26.7/-	~15.0/11.7	[50]
<b>2PCTrz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20 % /TSP01/ TPBi/LiF/A	488	(0.18, 0.35)	-	28.5/-	~22.0/19.4	[50]
<b>TrzCz-Me</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	434	(0.16,0.08)	4.3	4.8/2.51/3.44	N/A	[51]
<b>TrzCz-Me-1</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	434	(0.16,0.08)	4.8	3.3/1.65/2.52	N/A	[51]
<b>TrzCz-Dme</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	436	(0.16,0.09)	4.3	3.19/1.85/2.48	N/A	[51]
<b>TrzCz-CF3</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	464	(0.19,0.23)	3.9	9.4/12.83/15.92	N/A	[51]
<b>TrzCz-CF3-1</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	448	(0.17,0.15)	3.9	7.17/12.72/15.8	N/A	[51]
<b>TrzBuCz-CF3</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	472	(0.18,0.29)	4.1	14.22/19.5/26.7	N/A	[51]
<b>TrzBuCz-CF3-1</b>	ITO/PEDOT:PSS/NPB/mCP/TCTA/DPEPO:10% emi/TmPyPB/ LiF/Al	452	(0.17,0.16)	3.9	7.97/14.39/17.88	N/A	[51]
<b>TrzBFCz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:10 %TrzBFCz/TSP01/TPBi/LiF/Al	447	(0.15, 0.09)	8.1	19/-	~6.0/4.3	[53]
<b>TrzCNBFCz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20 %TrzBFCz/TSP01/TPBi/LiF/Al	481	(0.17, 0.31)	6.5	26.6/-	~22.0/16.8	[53]
<b>Trz2CNBFCz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:10 %TrzBFCz/TSP01/TPBi/LiF/Al	511	(0.27, 0.52)	9	15.8/-	~14/6.9	[53]
<b>CzCNTrz</b>	ITO/PEDOT:PSS/TAPC/mCP/EML (25 nm:10 wt%)/TSP01/TPBi/LiF/Al	461	(0.15, 0.16)	-	13.9/18.3/18.3	9.5/~2.0	[52]
<b>dBFCzCNTrz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:emitters/TSP01/TPBi/LiF/Al	497	(0.22, 0.47)	5.2	27.5/-	~26.0/24.3	[54]
<b>dBFCzTrz</b>	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:emitters/TSP01/TPBi/LiF/Al	476	(0.16, 0.27)	5.6	22.6/-	~17.0/12.3	[54]

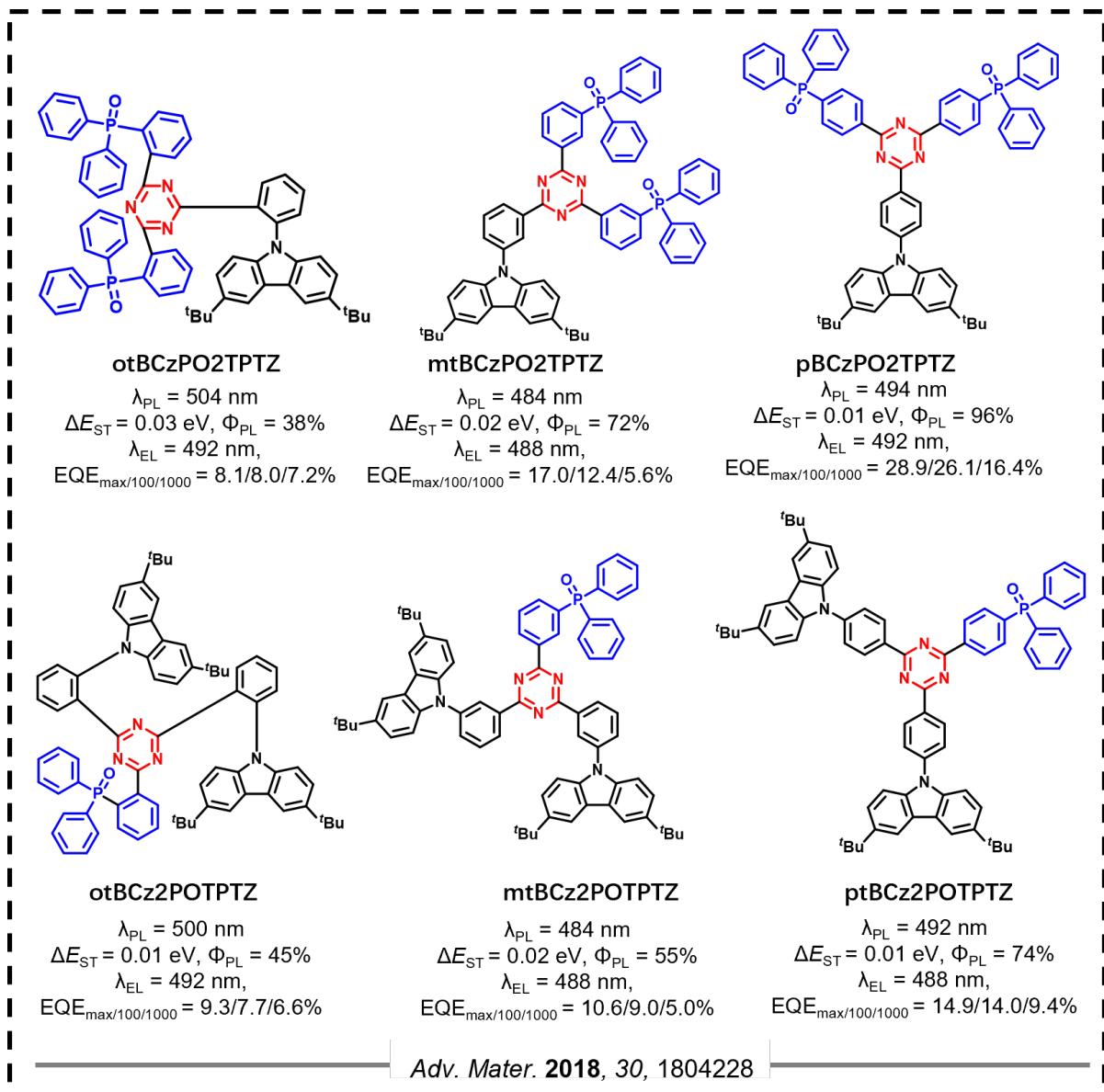
## 8. Functionalization of the triazine



**Figure 25.** Molecular structures and properties of TADF emitters based on functionalized TRZ groups.

Lee *et al.*<sup>[55]</sup> developed a blue TADF emitter, **mtCzTRZ** ( $\lambda_{PL}$ : 469 nm;  $\Phi_{PL}$ : 65%;  $\tau_d$ : 118.6  $\mu$ s in 1 wt% in PS;  $\Delta E_{ST}$ : 0.06 eV), by attaching a *para*-methoxy group to one of the phenyl rings of the TRZ acceptor, which was designed to blue shift the emission compared to **tCzTRZ** by weakening its electron-withdrawing capacity (**Figure 25**). The control compound, **tCzTRZ** ( $\lambda_{PL}$ : 469 nm;  $\Phi_{PL}$ : 67%;  $\tau_d$ : 89.4  $\mu$ s in 1 wt% in PS;  $\Delta E_{ST}$ : 0.05 eV) showed identical PL and similar photophysical properties to that of **mtCzTRZ**, indicating that there is negligible influence of the photophysical properties due to the presence of the methoxy group on the TRZ acceptor. However, the modification resulted in an improvement in the device performance. The solution-processed OLED with **mtCzTRZ** dispersed into the (5-(*tert*-butyl)-2-(4-(*tert*-butyl)phenoxy)phenyl)diphenylphosphine oxide (POBBPE)<sup>[56]</sup> host as the emitting layer, a more soluble analog of DPEPO, showed an  $EQE_{max}$  of 16.1% at  $\lambda_{EL}$  of 461 nm and CIE coordinates of (0.16, 0.20). By contrast, the device with **tCzTRZ** only afforded an  $EQE_{max}$  of 12.5% at similar though slightly red-shifted color coordinates [ $\lambda_{EL}$ : 467 nm; CIE = (0.16, 0.24)].

**FTRZTCz** ( $\Phi_{PL}$ : 85%;  $\tau_d$ : 5.91 $\mu$ s 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.22 eV), also developed by Lee *et al.*<sup>[57]</sup>, is an emitter containing a triazine acceptor with an inductively electron-withdrawing *para*-fluoro group onto one of the phenyl rings of the TRZ (**Figure 25**). 3,6-di-*tert*-butylcarbazole groups connected at the *ortho* positions of the other two phenyl rings complete the donor-acceptor structure. Their placement was to prevent intermolecular interactions, a similar strategy to that used in **o-CzTRZ** (**Figure 3**). **FTRZTCz** showed a PL spectrum in toluene with the maximum at 450 nm, which though not directly comparable is similar to the PL maximum of **o-CzTRZ** ( $\lambda_{PL}$  455 nm, 5 wt% in mCP) in mCP matrix. The missing information related to the compound without the fluoro group makes it difficult to conclude how much the fluoro substituent contributes to the shift of the PL given the similar donor-*ortho*-acceptor skeleton. However, it is still possible to infer the influence of the fluoro group by comparing the properties of the compound with the previously reported molecule **o-CzTRZ**, which only has one *ortho*-connected carbazole. The red-shifted EL for the device with **FTRZTCz** [ $\lambda_{EL}$ : 496 nm, CIE coordinates = (0.20, 0.45)] compared to that with **o-CzTRZ** [ $\lambda_{EL}$ : 470 nm, CIE = (0.15, 0.22)] reveals the presence of a stronger electron-accepting triazine derivative in **FTRZTCz** compared to **o-CzTRZ**.



**Figure 26.** Molecular structures and properties of TADF emitters based on diphenylphosphine oxide (PO) functionalized triazine.

Xu *et al.*<sup>[58]</sup> employed diphenylphosphine oxide (PO) as secondary acceptors within a series of D–A–A type TADF emitters with the collective name of **xtBCznPO3-nTPTZ** (*x* = *o*, *m*, and *p*, corresponding to placement of the diphenylphosphine oxide at the *ortho*-, *meta*-, and *para*-positions; and *n* = 1 and 2, corresponding to the number of diphenylphosphine oxide acceptors), in which tBCz is 3,6-di-*tert*-butylcarbazole, and TPTZ is triphenyltriazine (aka TRZ) as shown in **Figure 26**. The incorporation of the electron-withdrawing PO group serves to stabilize the

excited-states and to enhance the CT character, leading to a greater localization of the frontier molecular orbital (FMO) distributions. By virtue of this design strategy, **ptBCzPO2TPTZ** ( $\lambda_{PL}$ : 494 nm;  $\Phi_{PL}$ : 96%; in 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.01 eV) as the best example among this series, has almost unity  $\Phi_{PL}$  (96%) and very efficient RISC efficiency,  $\Phi_{RISC}$ , of 98% and  $k_{RISC}$  of  $5.42 \times 10^4$  s. These properties established the basis for the high-performance OLED, which showed an  $EQE_{max}$  of 28.9% at  $\lambda_{EL}$  of 492 nm, corresponding to CIE coordinates of (0.18, 0.42). It is worth noting that the analog **p-CzTRZ** (Figure 3) without the PO groups doesn't show any TADF properties due to its too large  $\Delta E_{ST}$  of 0.36 eV in 10 wt% in DPEPO. The other derivatives in this work showed moderate  $\Phi_{PLS}$  ranging from 38% to 74%. The photophysics and electrochemical data of the aforementioned materials are summarized in **Table 13** and representative device performance is summarized in **Table 14**.

**Table 13.** Summary of photophysical and electrochemical properties

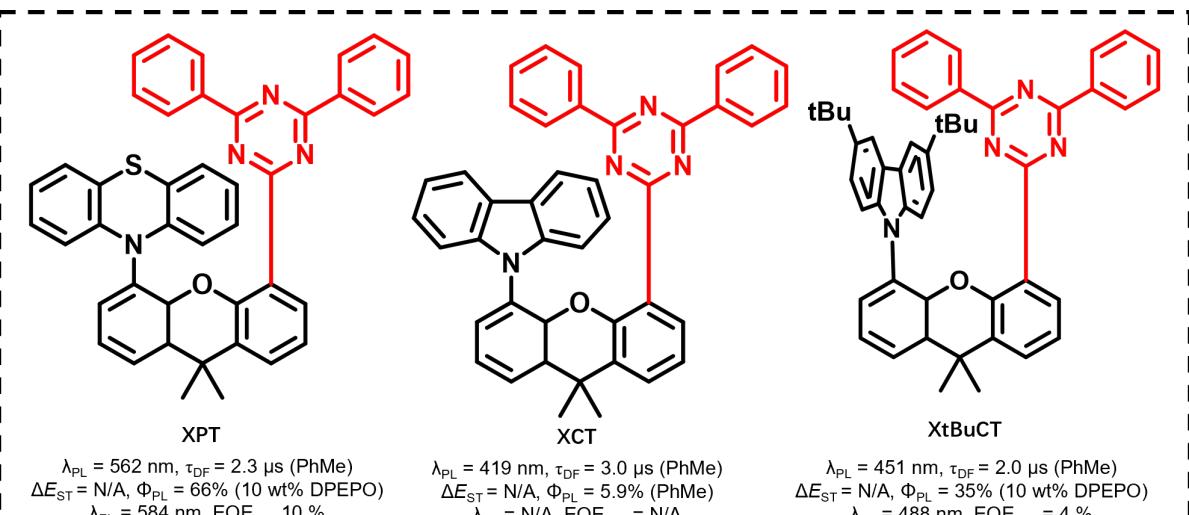
Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/ $\mu$ s]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
mtCzTrz	469/-/(PhMe)	-/65.1/118.6(1 wt% in PS)	0.06	-6.08	-3.41	[55]
tCzTrz	469/-/(PhMe)	-/66.6/89.4(1 wt% in PS)	0.05	-6.09	-3.43	[55]
ptBCzPO2TPTZ	506/-/(CHCl3)	494/96/- (10 wt% in DPEPO)	0.01	-6.05	-3.60	[58]
otBCzPO2TPTZ	518/-/(CHCl3)	504/38/- (10 wt% in DPEPO)	0.03	-6.04	-3.48	[58]
mtBCzPO2TPTZ	512/-/(CHCl3)	484/72/- (10 wt% in DPEPO)	0.02	-6.08	-3.60	[58]
ptBCzPO2TPTZ	506/-/(CHCl3)	494/96/- (10 wt% in DPEPO)	0.01	-6.05	-3.60	[58]
otBCz2POTPTZ	514/-/(CHCl3)	50045/- (10 wt% in DPEPO)	0.01	-6.05	-3.53	[58]
mtBCz2POTPTZ	514/-/(CHCl3)	48455/- (10 wt% in DPEPO)	0.02	-6.08	-3.58	[58]
ptBCz2POTPTZ	496/-/(CHCl3)	49274/- (10 wt% in DPEPO)	0.01	-6.04	-3.50	[58]
FTrzTCz	~450/-/(PhMe)	-/85/5.91(10% in DPEPO)	0.22	-6.13	-3.42	[57]

**Table 14.** Summary of device structures and performance

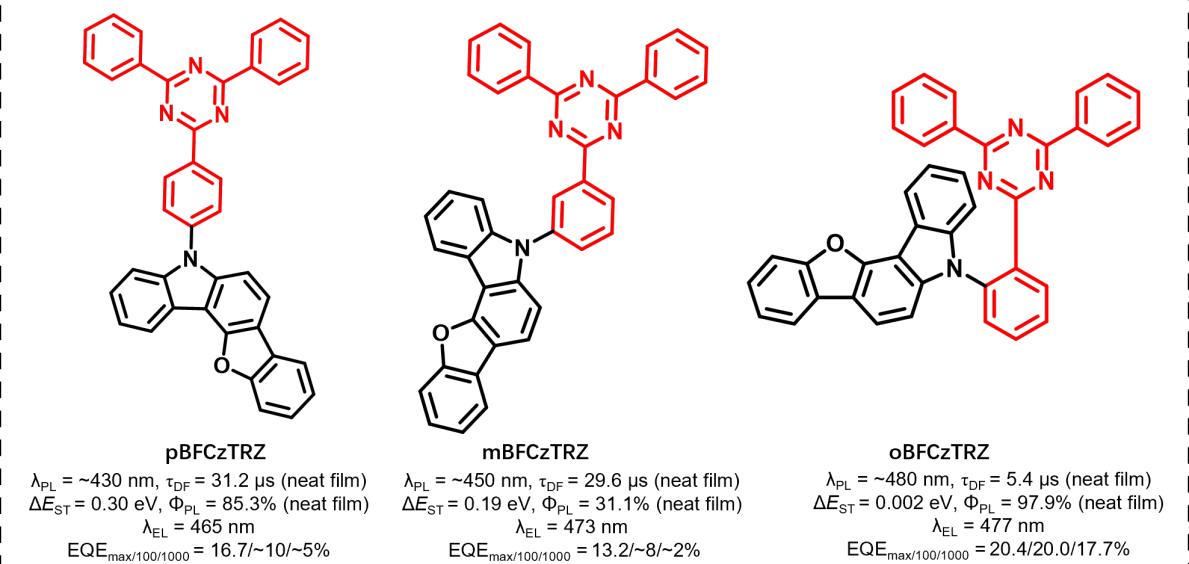
Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a</sup> ) [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref.
mtCzTrz	ITO/PEDOT:PSS/PVK/POBBPE:20 wt% mtCzTrz //TSPO1/LiF/Al	461	0.16, 0.20	7.6	16.1/9.2/25	N/A	[55]
tCzTrz	ITO/PEDOT:PSS/PVK/POBBPE:20 wt% tCzTrz //TSPO1/LiF/Al	467	0.16, 0.24	8	12.5/7.9/21.4	N/A	[55]
ptBCzPO2TPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	~488	(0.18, 0.42)	-	28.9/79.3/73.2	26.1/16.4	[58]
otBCzPO2TPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	492	(0.21, 0.46)	3.6	8.1/15.8/21.9	8.0/7.2	[58]

mtBCzPO <sub>2</sub> TPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	mtBCzPO <sub>2</sub> TPTZ:	488	(0.18,0.36)	3.2	17/34.6/39.7	12.4/5.6	[58]
ptBCzPO <sub>2</sub> TPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	ptBCzPO <sub>2</sub> TPTZ:	492	(0.18,0.42)	2.9	28.9/79.3/73.2	26.1/16.4	[58]
otBCz <sub>2</sub> POTPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	otBCz <sub>2</sub> POTPTZ:	492	(0.20,0.44)	2.9	9.3/26.5/24.5	7.7/6.6	[58]
mtBCz <sub>2</sub> POTPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	mtBCz <sub>2</sub> POTPTZ:	488	(0.18,0.37)	3	10.6/22.7/24.9	9.0/5.0	[58]
ptBCz <sub>2</sub> POTPTZ	ITO/MoO <sub>3</sub> /mCP/80 wt% DPEPO/pTPOTZ/LiF/Al	ptBCz <sub>2</sub> POTPTZ:	488	(0.17,0.32)	2.9	14.9/31.7/31.9	14.0/9.4	[58]
FTrzTCz	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20%FTrzT Cz/TSPO1/TPBi/LiF/Al		496	(0.20, 0.45)	-	18.5/-/-	~17.0/14.5	[57]

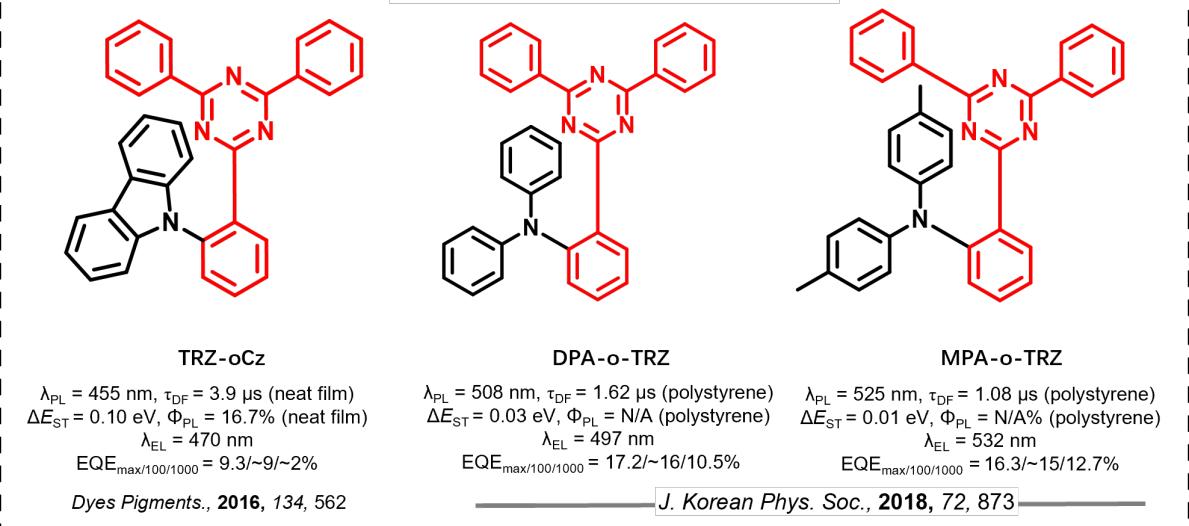
## 9. Intramolecular through space charge transfer (TSCT) of triazine based TADF emitters



J. Am. Chem. Soc. 2017, 139, 4894



ACS Appl. Mater. Interfaces 2016, 8, 23190

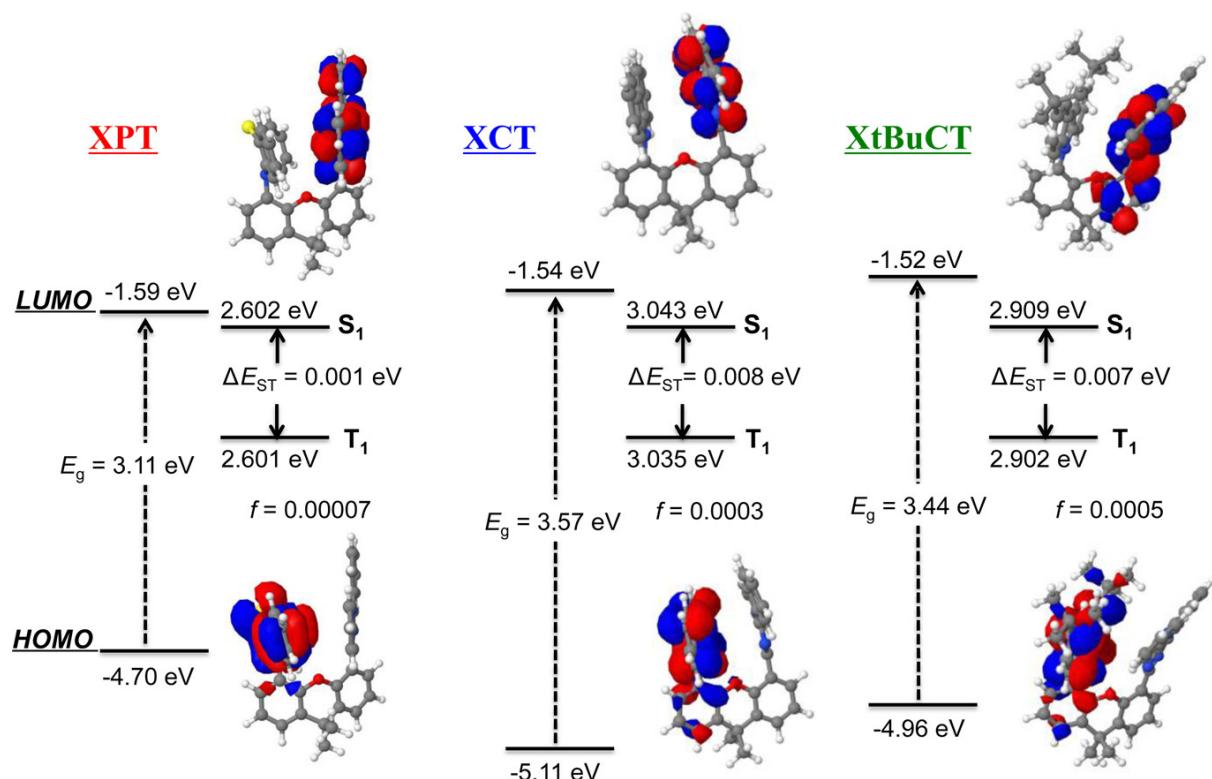


Dyes Pigments., 2016, 134, 562

J. Korean Phys. Soc., 2018, 72, 873

**Figure 27.** Molecular structures and properties of TADF emitters based on TSCT mechanism.

The prior examples have implicated an  $S_1$  state that has CT character and that the charge transfer process is mediated by a  $\pi$ -conjugated bridging moiety. This type of CT is termed through-bond charge transfer (TBCT).<sup>[59]</sup> There exists a second CT process that is mediated by  $\pi$ -stacking of the donor and acceptor units, termed through-space charge transfer (TSCT). Distinct from most of the TBCT emitters that feature strong electron coupling through covalent bonds, TSCT emitters possess weaker electronic interaction between donors and acceptors due to the relatively smaller overlap of the frontier molecular orbitals in these systems. The degree of electronic coupling is controlled by the distance and the relative angles between the donor and acceptor groups. For example, an *ortho*-substitution pattern between the donor and acceptor moieties leads to sterically congested, co-facially aligned donor and acceptor groups in the molecule, inducing either a  $\pi \cdots \pi^*$  type (as in the **B-oTC** emitter) or an  $n \cdots \pi^*$  type (as in **TRZ-oCz**) interaction. The conformationally restricted structures result in limited change in the geometry of the emitter in the excited states and small reorganization energies, reflected in the narrower emission profiles of these compounds.



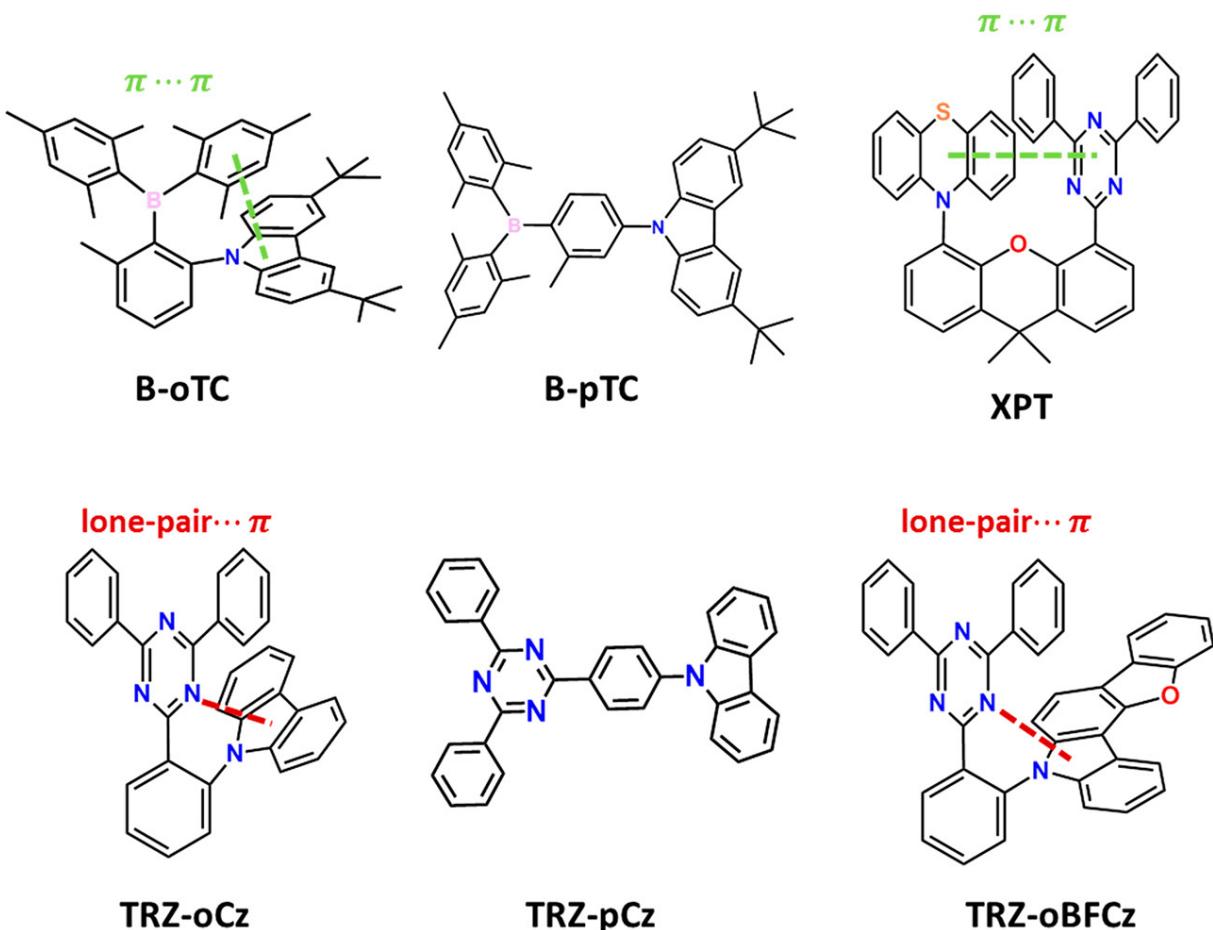
**Figure 28.** HOMO and LUMO orbital distributions and calculated bandgaps, singlet ( $S_1$ ), triplet ( $T_1$ ) energy levels, and oscillator strengths ( $f$ ) for **XPT**, **XCT**, and **XtBuCT** based on TD-DFT at the B3LYP functional and 6-31G\* basis set. Reproduced with permission.<sup>[60]</sup> Copyright © 2017, American Chemical Society.

Swager *et al.*<sup>[60]</sup> reported a series of TADF emitters possessing a TSCT state. As shown in **Figure 27**, the design of **XPT**, **XCT** and **XtBuCT** is based on a xanthene scaffold with a donor and an acceptor co-facially aligned at a well-defined distance. These compounds possess very small  $\Delta E_{ST}$  values as evidenced by DFT calculation (**Figure 28**), and the close alignment between the donor and acceptor groups restricts motion in the solid state, resulting in aggregation induced delayed fluorescence (AIDF) as exemplified by the increase in  $\Phi_{PL}$  from dilute toluene to 10 wt% DPEPO films of **XPT** and **XtBuCT** [**XPT** ( $\Phi_{PL}$ : 7.7% in toluene,  $\Phi_{PL}$ : 66% in 10 wt% in DPEPO), and **XtBuCT** ( $\Phi_{PL}$ : 6.0% in toluene,  $\Phi_{PL}$ : 35% in 10 wt% in DPEPO)]. The strength of the electron donor directly impacted the corresponding emission color of the emitters as evidenced by the PL maximum shifting from 419 nm in **XCT** to 451 nm in **XPT** and 562 nm in **XtBuCT**. The OLED with **XPT** showed a 10% EQE<sub>max</sub> with  $\lambda_{EL}$  at 586 nm while **XtBuCT**-based device exhibited a 4% EQE<sub>max</sub> (limited by its lower  $\Phi_{PL}$ ) with  $\lambda_{EL}$  at 488 nm.

Lee *et al.*<sup>[61]</sup> found that *ortho*-linked donor and acceptor compounds showed superior TADF properties to those where the donor was either *meta*- or *para*-disposed. The authors compared the optoelectronic properties of **oBFCzTRZ**, **mBFCzTRZ** and **pBFCzTRZ**, which are based on *ortho*-, *meta*-, and *para*-linked diphenyltriazine and benzofurocarbazole groups (**Figure 27**). It is noteworthy that the  $\Delta E_{ST}$  of **oBFCzTRZ** is only 0.002 eV, while those of **mBFCzTRZ** and **pBFCzTRZ** are 0.191 and 0.302 eV, respectively, revealing the much weaker electronic

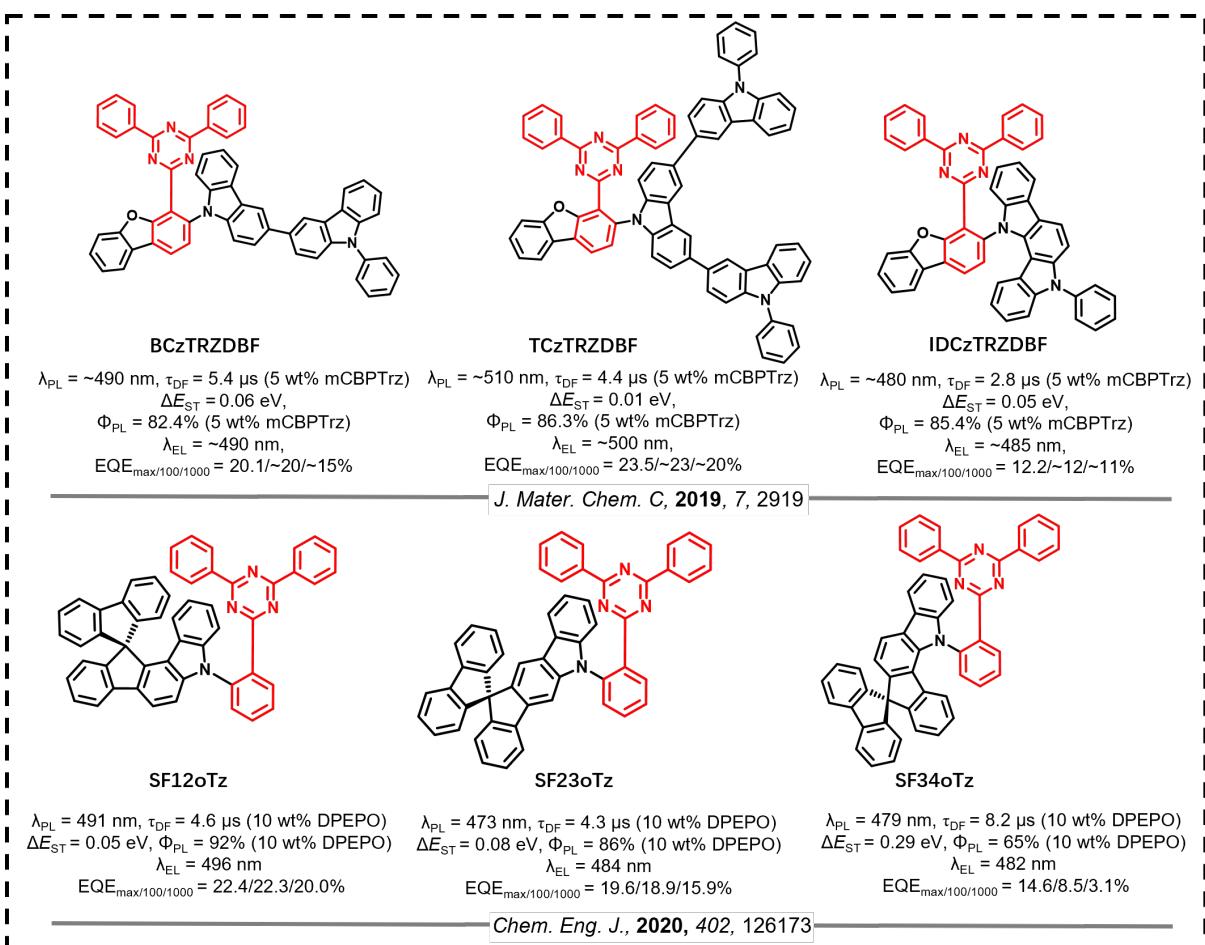
coupling within the former. Moreover, **oBFCzTRZ** ( $\Phi_{PL}$ : 97.9%;  $\tau_d$ : 5.4  $\mu$ s in 10 wt% in DPEPO) also possessed a shorter delayed lifetime, and higher  $\Phi_{PL}$  than **mBFCzTRZ** ( $\Phi_{PL}$ : 31.1%;  $\tau_d$ : 29.6  $\mu$ s in 10 wt% in DPEPO) and **pBFCzTRZ** ( $\Phi_{PL}$ : 85.3%;  $\tau_d$ : 31.2  $\mu$ s in 10 wt% in DPEPO), reflective of the more efficient RISC process in the former. The higher  $\Phi_{PL}$  is likely due to the suppression of non-radiative decay due to the more conformationally restricted structure in **oBFCzTRZ**. Blue-emitting devices with **oBFCzTRZ** ( $\lambda_{EL}$  = 477 nm) > **mBFCzTRZ** ( $\lambda_{EL}$  = 473 nm) > **pBFCzTRZ** ( $\lambda_{EL}$  = 465 nm) showed EQE<sub>max</sub> of 20.4% [ $\lambda_{EL}$ : 477 nm; CIE = (0.18, 0.31)], 13.2% [ $\lambda_{EL}$ : 473 nm; CIE = (0.17, 0.25)], and 16.7% [ $\lambda_{EL}$ : 465 nm; CIE = (0.15, 0.18)], respectively. Importantly, **oBFCzTRZ** based device showed the least amount of efficiency roll-off.

Expanding on the *ortho*-disposed donor-acceptor platform, Lee *et al.*<sup>[62]</sup> reported green TADF emitters **DPA-o-TRZ** and **MPA-o-TRZ** (Figure 27), which incorporated a diphenylamine group whose electron donating ability is stronger than benzofluorene as in **oBFCzTRZ**. Thus, both **DPA-o-TRZ** and **MPA-o-TRZ** exhibited red-shifted PL with  $\lambda_{PL}$  at around 500 nm and 530 nm, respectively. Similar to the other TSCT emitters, **DPA-o-TRZ** and **MPA-o-TRZ** also have very small  $\Delta E_{ST}$  values of 0.03 eV and 0.01 eV, respectively, and comparatively short delayed lifetimes of 1.62  $\mu$ s and 1.08  $\mu$ s, respectively. The OLEDs based on **DPA-o-TRZ** [ $\lambda_{EL}$ : 497 nm; CIE = (0.21, 0.45)] and **MPA-o-TRZ** [ $\lambda_{EL}$ : 532 nm; CIE = (0.35, 0.58)] showed EQE<sub>max</sub> of 17.2% and 16.3%.



**Figure 29.** Molecular structures of the TSCT molecules based on intramolecular  $\pi \cdots \pi$  n  $\cdots \pi$  noncovalent interactions. Reproduced with permission.<sup>[63]</sup> Copyright © 2019, American Chemical Society.

Using DFT calculations, Bredas *et al.*<sup>[63]</sup> investigated the origins of the TSCT in a number of TADF emitters (**Figure 29**). Two pathways for TSCT were identified: intramolecular  $\pi \cdots \pi$  and n  $\cdots \pi$  noncovalent interactions. The mechanism of the  $\pi \cdots \pi$  interaction is the one that is operational for **XPT**, **XCT** and **XtBuCT**.<sup>[60]</sup> However, in the cases of **TRZ-oCz**<sup>[15]</sup> and **TRZ-oBFCz**, the computations indicated a very close packing between one of the nitrogen atoms of the triazine ring with the carbazole plane, which the authors asserted triggered noncovalent interactions between the lone-pair electrons of this triazine nitrogen atom and the carbazole  $\pi$  electrons. The n  $\rightarrow$   $\pi^*$  interactions were identified as critical for enhancing the singlet–triplet spin–orbit coupling, and as a result, greatly facilitating the RISC process.



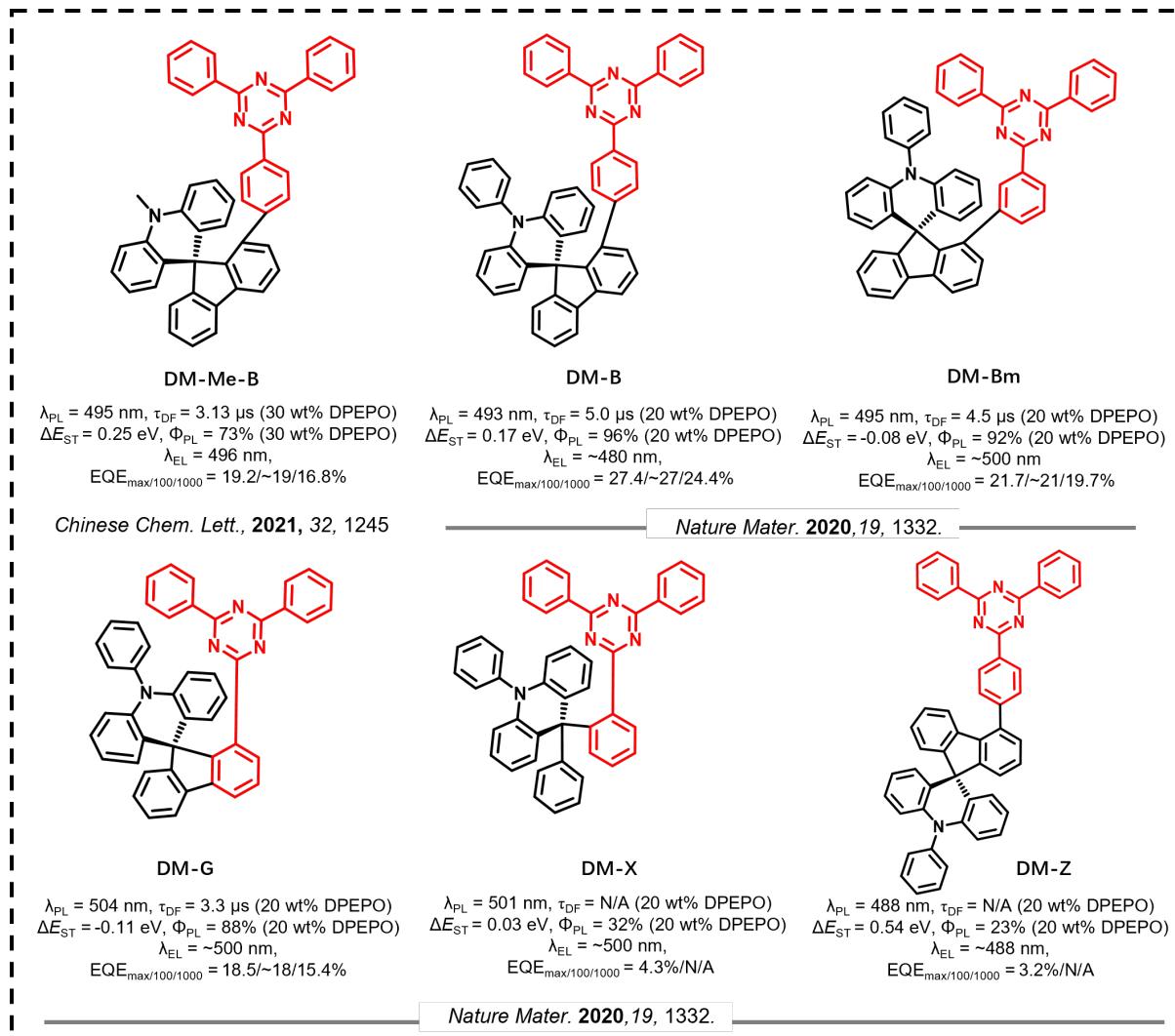
**Figure 30.** Molecular structures and properties of TADF emitters based on TSCT mechanism.

Further examples based on an *ortho*-connection strategy include the incorporation of a dibenzofuran as the bridge in **BCzTRZDBF**, **TCzTRZDBF** and **IDCzTRZDBF** (**Figure 30**) where substitution next to the oxygen of the dibenzofuran induced a large conformational distortion.<sup>[64]</sup> This, coupled with the *ortho*-connection between the donor and acceptor, leads to shortened delayed lifetimes and small  $\Delta E_{ST}$ , both of which were ascribed to the TSCT character of the emissive excited state. **BCzTRZDBF**, **TCzTRZDBF** and **IDCzTRZDBF** possess small  $\Delta E_{ST}$  of less than 0.1 eV yet with high  $\Phi_{PL}$  of 82.4, 86.3 and 85.4%, respectively, as 5 wt% in mCBPTRZ films. The best device was obtained with **TCzTRZDBF** as the emitter and showed an  $EQE_{max}$  of 23.5% with CIE coordinates of (0.27, 0.57) and FWHM of 80 nm. Like

**TCzTRZDBF**, another analogue **BCzTRZDBF**, with a less bulky dendritic donor, also exhibited a similar EQE<sub>max</sub> of 20.1% and comparable CIE coordinates of (0.24, 0.52) and FWHM of 84 nm. However, the device with **IDCzTRZDBF** showed a much worse performance with an EQE<sub>max</sub> of only 12.2%, a CIE coordinates of (0.22, 0.48) and FWHM of 81 nm. Based on the fact that **IDCzTRZDBF** has a  $\Phi_{PL}$  of 85.4%, the unexpected lower efficiency may be attributed to the unoptimized device configuration for **IDCzTRZDBFb** which has the largest HOMO-LUMO gap and the deepest HOMO level among the three compounds.

TADF molecules based on *ortho*-disposed donors and acceptors were shown by Bredas and co-workers to contain emissive excited states that comprise both TBCT and TSCT character. Yang *et al.*<sup>[65]</sup> reported a series of highly twisted emitters, **SF34oTz**, **SF23oTz** and **SF12oTz**, containing indolin-fused spirobifluorene donors (**Figure 30**). According to the DFT computations, the ratios of TBCT/TSCT were initially calculated by integrating the transition density localized on/not on the phenylene bridge. Among them, **SF34oTz** ( $\Phi_{PL}$ : 65% in 10 wt% DPEPO) showed a dominant TSCT (96.8% as suggested by TD-DFT calculations) and a relative larger  $\Delta E_{ST}$  of 0.29 eV, whereas **SF23oTz** ( $\Phi_{PL}$ : 86% in 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.08 eV) and **SF12oTz** ( $\Phi_{PL}$ : 92% in 10 wt% in DPEPO;  $\Delta E_{ST}$ : 0.05 eV) exhibit higher TBCT contributions, with the ratio increasing to 21% and 32%, respectively. The higher TBCT contribution in **SF12oTz** leads to much higher ratio of delayed fluorescence (79.1% vs. 39.2%) compared with **SF23oTz**, which the authors asserted was due to the more effective channel of TBCT over TSCT to realize charge transfer from donors to the acceptor. However, the largest  $\Delta E_{ST}$  of 0.29 eV for **SF34oTz** in the series does not align with the expectation that a compound with a greater TSCT contribution should have the smallest  $\Delta E_{ST}$  in theory. Among the three emitters, the OLED with **SF12oTz** performed the best, showing an EQE<sub>max</sub> of 22.4% [ $\lambda_{EL}$ : 496

nm; CIE = (0.23, 0.47)]. The devices with **SF23oTz** [ $\lambda_{EL}$ : 484 nm; CIE = (0.19, 0.35)] and **SF34oTz** [ $\lambda_{EL}$ : 482 nm; CIE = (0.18, 0.37)] showed EQE<sub>max</sub> of 19.6% and 14.6%, respectively. The trend in device efficiencies align with that of the  $\Phi_{PL}$ .

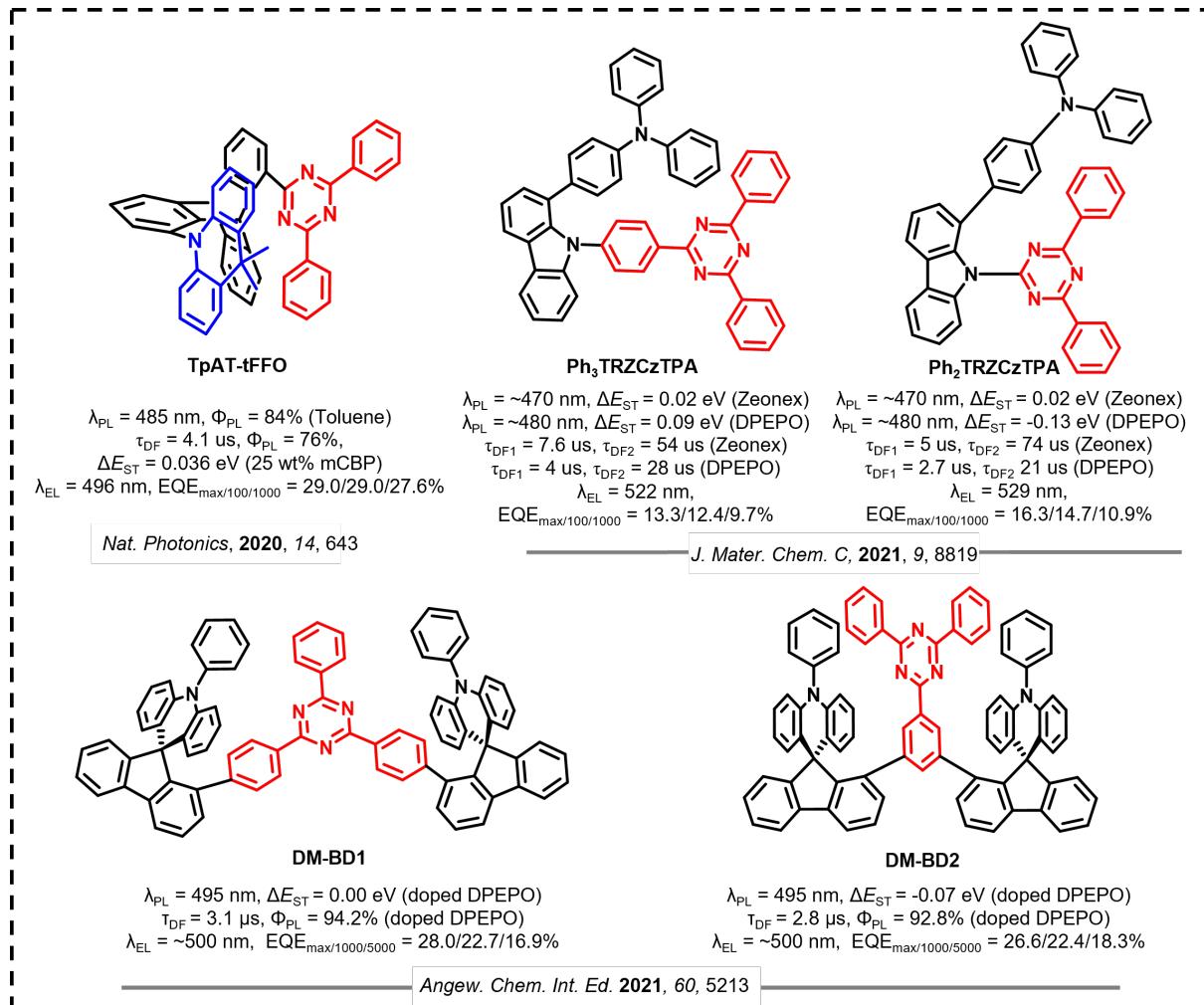


**Figure 31.** Molecular structures and properties of TADF emitters based on TSCT mechanism.

Many OLEDs that contain TADF compounds that emit from an excited state containing some TSCT character still suffer from low device performance.<sup>[60]</sup> Recently, Kaji *et al.*<sup>[66]</sup> and Liao *et al.*<sup>[67]</sup> elucidated the importance of the relative orientation of donor and acceptor groups on the electronic coupling between the two in TSCT compounds that influences also the magnitude of the non-radiative decay. Similar to the other TSCT molecules discussed in this section, **DM-B** ( $\Phi_{PL}$ : 96% in 20 wt% in DPEPO;  $\Delta E_{ST}$ : 0.17 eV) and **DM-Bm** ( $\Phi_{PL}$ : 92% in 20 wt% in

DPEPO;  $\Delta E_{ST}$ : -0.08 eV) were designed by connecting the donor and acceptor units via a rigid fluorene linker thereby confining them into a closely packed coplanar configuration (**Figure 30**). **DM-G** ( $\Phi_{PL}$ : 88% in 20 wt% in DPEPO;  $\Delta E_{ST}$ : -0.11 eV) features a diphenyltriazine acceptor instead of triphenyltriazine in **DM-B** and **DM-Bm**. By contrast, the authors also developed a more flexible control compound **DM-X** ( $\Phi_{PL}$ : 32% in 20 wt% in DPEPO;  $\Delta E_{ST}$ : 0.03 eV), which differs from **DM-G** as it contains less rotationally constrained 1,2-phenylene bridge instead of the rigid fluorene linker. Another control compound **DM-Z** ( $\Phi_{PL}$ : 23% in 20 wt% in DPEPO;  $\Delta E_{ST}$ : 0.54 eV) has also been designed to analyse the effect of increasing the D/A separation by placing the donor and acceptor units on the opposite sides of the rigid fluorene linker. Despite their similar absorption and emission spectra, the  $\Phi_{PL}$  values of the rigid **DM-B**, **DM-Bm**, **DM-G** are remarkably higher than those of the flexible control compounds, which further validated the design strategy on the suppression of intramolecular vibrations and rotations. The  $\Delta E_{ST}$  values are 0.17, -0.08, -0.11, 0.03 and 0.54 eV for **DM-B**, **DM-Bm**, **DM-G**, **DM-X** and **DM-Z**, respectively. The negative  $\Delta E_{ST}$  of **DM-Bm** and **DM-G** were attributed by the authors to different molecular geometries in their respective fully relaxed singlet and triplet states. Another analogue **DM-Me-B** ( $\Phi_{PL}$ : 73% in 30 wt% in DPEPO;  $\Delta E_{ST}$ : 0.25 eV) was also developed by replacing the phenyl group onto the donor unit of **DM-B** with a methyl group.<sup>[68]</sup> In contrast to **DM-B** ( $\Phi_{PL}$ : 96%), the slight structural change contributes to a reduction of the  $\Phi_{PL}$  to 73% although **DM-Me-B** still maintains similar TADF properties. OLEDs based on **DM-B** (50 wt% in DPEPO) achieved an  $EQE_{max}$  of 27.4%, with only minor efficiency roll-off at a luminance of 1,000 cd m<sup>-2</sup> where the  $EQE_{1000}$  = 24.4%. The  $EQE_{max}$  of the devices based on **DM-Bm** (30 wt% in DPEPO) and **DM-G** (20 wt% in DPEPO) are 21.7% and 18.5%, respectively, which are proportional to the  $\Phi_{PL}$  of these emitters. In contrast, the devices based on the more flexible and less electronically coupled **DM-X** (20 wt% in DPEPO)

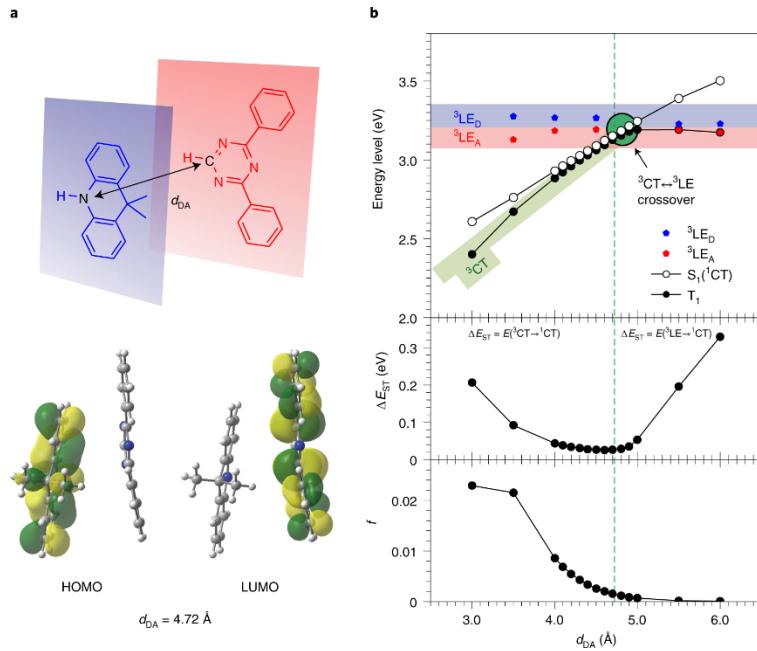
and **DM-Z** (30 wt% in DPEPO) exhibit considerably lower maximum efficiencies of 4.3% and 3.2%, respectively.



**Figure 32.** Molecular structures and properties of TADF emitters based on TSCT triazine motifs.

Kaji *et al.* [66] reported a TSCT compound **TpAT-tFFO** (Figure 32) based on the design strategy of controlling the distance and relative orientation between the adjacent tilted donor and acceptor moieties attached via the triptycene (Tp) scaffold. According to the DFT calculations, there is an increase in the energy levels of  $^1\text{CT}$  and  $^3\text{CT}$  states with increasing distance between the donor and acceptor (Figure 33b top). By contrast, the  $^3\text{LE}$  state only

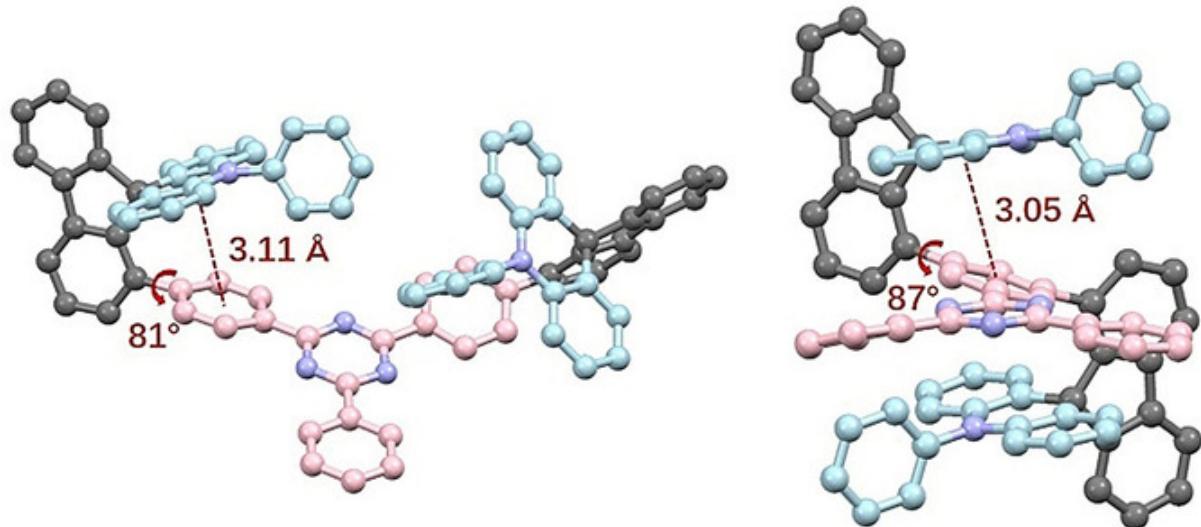
shows a weak dependence on the donor-acceptor distance. RISC is mediated by the intervening  $^3\text{LE}$  state between the  $^1\text{CT}$  and  $^3\text{CT}$  states and the high  $k_{\text{RISC}}$  of  $1.2 \times 10^7 \text{ s}^{-1}$  is ensured by the energy level matching of the three states. Importantly, the authors showed by DFT calculations that tilted face-to-face orientation between the donor and acceptor greatly enhances the SOC between the  $^1\text{CT}$  and  $^3\text{LE}$  states, which is not the case for the co-planar conformation. **TpAT-tFO** exhibited strong sky-blue emission with a  $\lambda_{\text{PL}}$  of 485 nm. The  $\Phi_{\text{PL}}$  in toluene markedly increased from 1.8% to 84% after 30 min of Ar bubbling, while the  $\Phi_{\text{PL}}$  of the film 25 wt% doped into the mCBP host and for the neat film were determined to be 76% and 71%, respectively. A device using an optimized doping concentration of 25 wt% showed an  $\text{EQE}_{\text{max}}$  of 19.2%. A gentle efficiency roll off was obtained with EQEs of 19.1% and 18.1% at 100  $\text{cd m}^{-2}$  and 1,000  $\text{cd m}^{-2}$ , respectively; moreover, EQE of 14.4% and 11.6% were retained even at high luminance of 10,000  $\text{cd m}^{-2}$  and 20,000  $\text{cd m}^{-2}$ .



**Figure 33.** a) Structures of the donor (blue) and acceptor (red) segments (top). The donor and acceptor have a tilted ( $\sim 10^\circ$ ) face-to-face alignment and are spatially separated at a certain distance,  $d_{\text{DA}}$ . HOMO and LUMO with  $d_{\text{DA}}$  of  $4.72 \text{ \AA}$  (bottom). b) DFT-calculated energy levels of  $S_1$ ,  $T_1$ ,  $^3\text{LE}_D$  and  $^3\text{LE}_A$  (top),  $\Delta E_{\text{ST}}$  (middle) and  $f$  (bottom) as a function of  $d_{\text{DA}}$ ,

where  ${}^3\text{LE}_{\text{D}}$  and  ${}^3\text{LE}_{\text{A}}$  denote the  ${}^3\text{LE}$  states on the donor and acceptor segments, respectively. The vertical green dashed line indicates  $d_{\text{DA}} = 4.72 \text{ \AA}$ , which corresponds to  $d_{\text{DA}}$  of a DFT-optimized structure for TpAT-tFFO at the ground state. The blue, red and green shaded regions highlight the energy levels of the  ${}^3\text{LE}_{\text{D}}$ ,  ${}^3\text{LE}_{\text{A}}$  and  ${}^3\text{CT}$  states, respectively. Reproduced with permission.<sup>[66]</sup> Copyright © 2020, Nature Publishing Group.

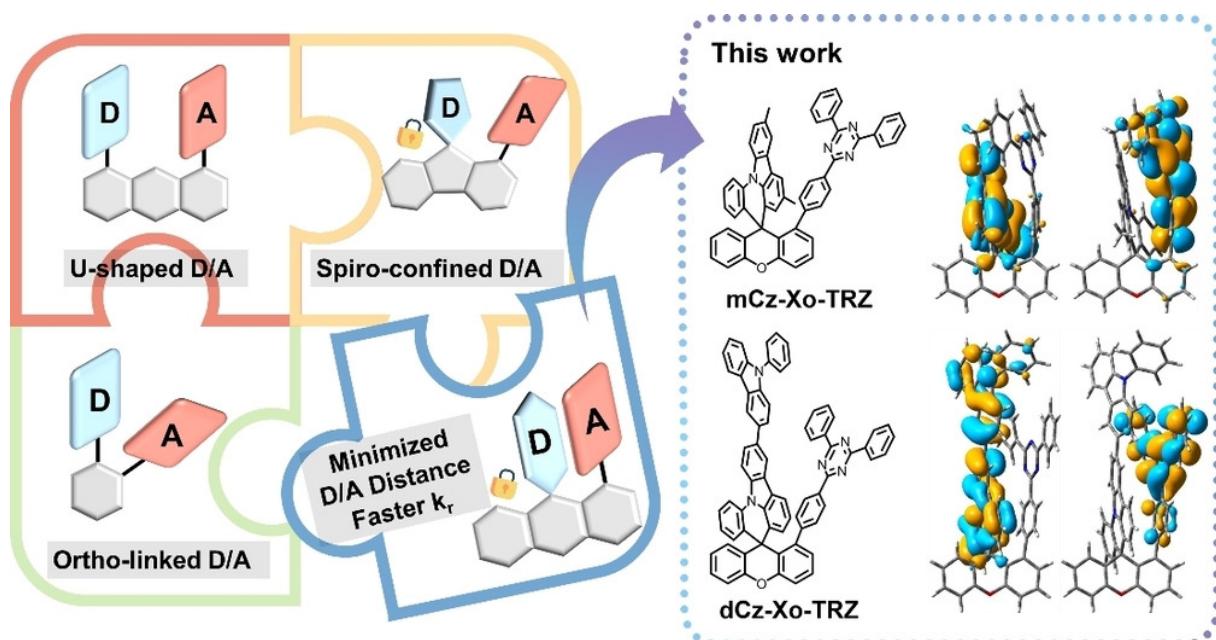
Monkman *et al.*<sup>[69]</sup> developed two unsymmetric donor–donor’–acceptor (D–D’–A) type emitters, **Ph<sub>3</sub>TRZCzTPA** and **Ph<sub>2</sub>TRZCzTPA** (Figure 32), where the co-facial overlap between D and A is controlled by the introduction of a weak (rigid) carbazole donor bridge (D’). According to the crystallographic data, the pendant (spacer) aryl rings are highly twisted from the carbazole bridge (D’) due to congested steric interactions. The short distances between the donor and acceptor enable the through space interaction. By comparing with the emission at 492 nm from the exciplex between TPA and TRZ in toluene, the identical emission observed from **Ph<sub>3</sub>TRZCzTPA** and **Ph<sub>2</sub>TRZCzTPA** indicates that the CT emission in both comes from a through-space intramolecular TRZ–TPA CT excited state. No contribution from a through-bond, TRZ – weak D’ CT pair was observed. Both **Ph<sub>3</sub>TRZCzTPA** and **Ph<sub>2</sub>TRZCzTPA** have the same small  $\Delta E_{\text{ST}}$  of 20 meV in Zeonex matrix. However, **Ph<sub>2</sub>TRZCzTPA** has a faster  $k_{\text{RISC}}$  than **Ph<sub>3</sub>TRZCzTPA**, which is ascribed to the more optimal tilted co-facial orientation of D and A forming the through-space CT state, which the authors contend is critical in controlling SOC and thus the RISC rate, as proposed by Kaji *et al.*<sup>[66]</sup> The EQE<sub>max</sub> values of the devices based on **Ph<sub>3</sub>TRZCzTPA** (20 wt% in 26DCzPPy) and **Ph<sub>2</sub>TRZCzTPA** (12 wt% in CBP) are 13.3% and 16.3%, respectively, which correlate with the relative RISC rates of the materials.



**Figure 34.** Single-crystal structures (hydrogen atoms are omitted) of **DM-BD1** (left) and **DM-BD2** (right). Reproduced with permission.<sup>[70]</sup>

The  $\pi$ -stacked scaffold has also been expanded to spatially confine donor/acceptor/donor (D/A/D) motifs.<sup>[70]</sup> Similar to the monomer analogues of **DM-B** and **DM-Bm**,<sup>[67]</sup> **DM-BD1** and **DM-BD2** also rely on spatially confined donor-acceptor interaction but with two donors within one molecule in a sandwich-like structure (**Figure 32**). The single crystal structures of **DM-BD1** and **DM-BD2** are shown in **Figure 34**. The additional donor is asserted to be helpful in realizing a more parallel D/A alignment, which results in a small  $\Delta E_{\text{ST}}$  (0.00 eV for **DM-BD1** and -0.07 eV for **DM-BD2**) and fast  $k_{\text{RISC}}$  ( $2.9 \times 10^5 \text{ s}^{-1}$  for **DM-BD1** and  $3.1 \times 10^5 \text{ s}^{-1}$  for **DM-BD2**). The torsion angles between the donor and acceptor moieties were determined to be  $30^\circ$  for **DM-BD1** and  $25^\circ$  for **DM-BD2** from the crystal structures, where the not quite co-facial orientation has been demonstrated by Kaji *et al.*<sup>[66]</sup> to be critical to turn on spin-orbit coupling and facilitate RISC. Interestingly, both compounds showed broad, structureless CT emission that are almost identical to those in solution, with a  $\lambda_{\text{PL}}$  of 495 nm. The  $\Phi_{\text{PL}}$  of **DM-BD1** (10 wt% doped in DPEPO matrix) and **DM-BD2** (10 wt% doped in DPEPO matrix) were determined to be 94.2% and 92.8%, respectively. The near unity  $\Phi_{\text{PL}}$  provides evidence of the

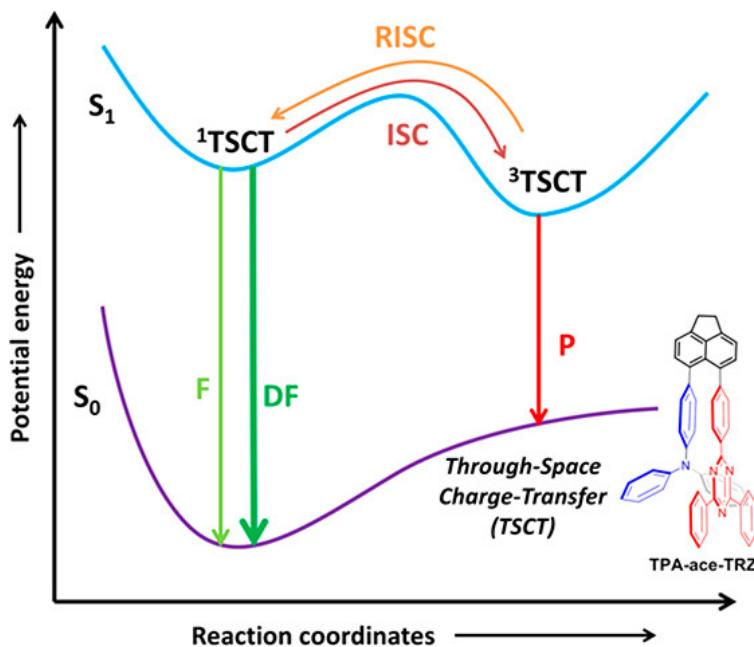
effectiveness of this strategy to suppress nonradiative loss mechanisms. The best OLEDs contained a 30 wt% emitter and showed  $\text{EQE}_{\max}$  of 28.0% and 26.6%, respectively for devices with **DM-BD1** and **DM-BD2**, corresponding to CIE coordinates of (0.21, 0.47) and (0.20, 0.46), respectively.



**Figure 35.** TSCT-TADF molecules with different types of donor–acceptor alignments and chemical structures of **mCz-Xo-TRZ** and **dCz-Xo-TRZ** with their HOMO/LUMO distribution. Reproduced with permission.<sup>[71]</sup>

To enhance the D-A electronic interactions by adjusting the D-A distance, Zhang *et al.*<sup>[71]</sup> employed a xanthene bridge in two TSCT TADF emitters, **mCz-Xo-TRZ** and **dCz-Xo-TRZ** (Figure 35), that possess space-confined face-to-face D–A alignment and minimized D–A distance down to 2.7–2.8 Å, which is shorter than twice the van der Waals radius of a carbon atom (1.7 Å). As a result, the greatly strengthened electronic interaction between D and A lead to fast  $k_r$  of 9.9 and  $8.7 \times 10^6 \text{ s}^{-1}$  for **mCz-Xo-TRZ** and **dCz-Xo-TRZ**, respectively. The stronger donor in **dCz-Xo-TRZ** versus **mCz-Xo-TRZ** leads to a red-shift of the PL spectrum in toluene

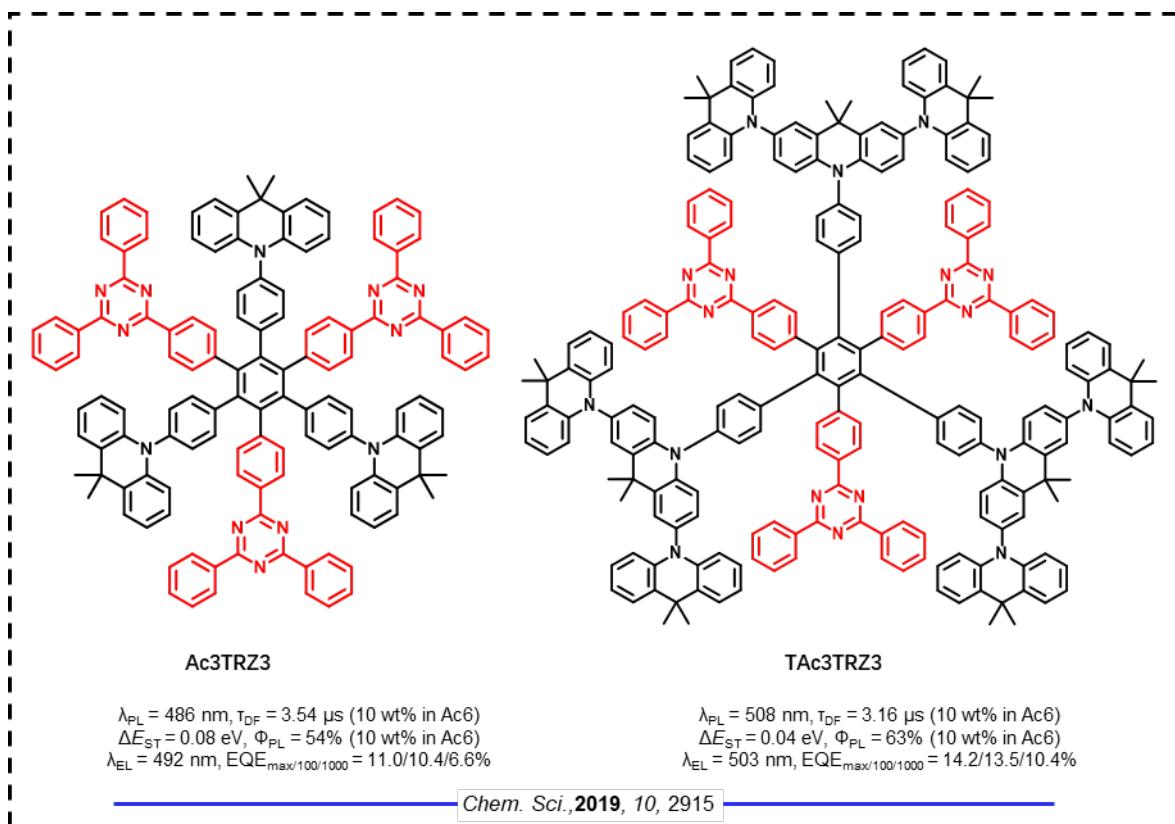
(from 454 nm to 461 nm) and the 30 wt% doped film in bis(diphenylphosphinyl)-dibenzofuran (PPF) matrix (from 469 nm to 482 nm). The  $\Phi_{PL}$  values of 90% and 92% for **mCz-Xo-TRZ** and **dCz-Xo-TRZ**, respectively, are high; however, the  $k_{RISC}$  values are  $3.0 \times 10^5 \text{ s}^{-1}$  and  $3.3 \times 10^5 \text{ s}^{-1}$ , respectively, which are significantly slower than that of **TpAT-tFFO** ( $1.2 \times 10^7 \text{ s}^{-1}$ ). The blue OLEDs showed broad emission with  $\lambda_{EL}$  of 477 and 464 nm for devices with **dCz-Xo-TRZ** and **mCz-Xo-TRZ**, corresponding to the CIE coordinates of (0.16, 0.29) and (0.15, 0.20), respectively. The  $\text{EQE}_{\max}$  of 27.8% for the device with **dCz-Xo-TRZ** and 21.0% for the device with **mCz-Xo-TRZ** showed only modest efficiency roll-off where the EQE values remained at 24.0 % and 17.1 % at a luminance of  $1000 \text{ cd m}^{-2}$ .



**Figure 36.** Molecular structure of **TPA-ace-TRZ** and schematic diagram of the potential energy involved in the  $S_0$  and  $S_1$  states. Reproduced with permission.<sup>[72]</sup> Copyright © 2021, American Chemical Society.

Zysman-Colman and co-workers recently reported a systematic study that documented explicitly in the PL spectrum emission from a TSCT in the compound **TPA-ace-TRZ**<sup>[72]</sup> (**Figure 36**). This work provides a direct evidence that the TSCT plays a major role in the

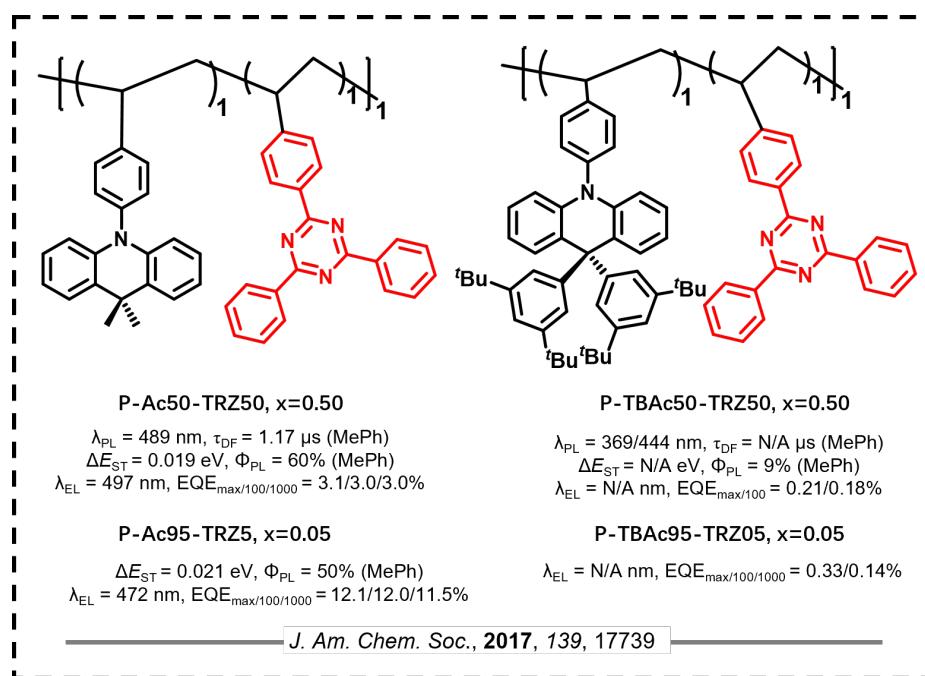
communication between the donor and acceptor. The photophysical studies in toluene of **TPA-ace-TRZ** show two characteristic lifetimes corresponding to the fast-decaying through-bond CT (TBCT) state ( $\tau_{PL} = 9.6$  ns) and longer-lived TSCT state ( $\tau_{PL} = 51$  ns). The existance of the two different CT states was ascribed to rapid decay from the initially populated TBCT state (with moderate D–A dihedral angles of  $48^\circ$  between the TPA and ace units and  $57^\circ$  between the ace and TRZ units) to the more stable TSCT state. The  $\Phi_{PL}$  of **TPA-ace-TRZ** was measured to be only 17%, which is due to the weak electronic communication between the donor and acceptor through the ace bridge (the ace unit is orthogonal and electronically decoupled from both D and A groups). It was found the the lowest-energy triplet state resides on the ace bridge, which also leads to a large  $\Delta E_{ST}$  of 0.48 eV (determined in 1 wt% ZEONEX film). The weak electronic communication makes it difficult for the triplet harvesting through the spin-vibronic coupling mechanism because the potentially mediating local triplet state resides on the ace bridge, which is orthogonal to both D and A and thus cannot efficiently couple to the TSCT state. Therefore, even though **TPA-ace-TRZ** possesses a strong TSCT state, it can hardly produce TADF because of the lack of coupling to a mediating LE triplet state. This study reveals the intimate interplay that the bridging ace group has on mediating both the TBCT state and the TSCT state.



**Figure 37.** Molecular structures and properties of TADF dendrimer emitters based on TSCT hexaarylbenzene motifs.

Wang *et al.*<sup>[73]</sup> incorporated TSCT states in the design of two star-shaped TADF emitters, containing DMAC (**Ac3TRZ3**) or a dendritic teracridan (**TAc3TRZ3**) as donors and TRZ as acceptors about a hexaphenylbenzene scaffold (**Figure 37**). Because of the steric hindrance inherent in the structure both molecules adopt a propeller-shaped conformation in which the peripheral aromatic units are almost perpendicular to the central phenyl ring. As a result of the strong TSCT character coupled with the weak TBCT character due to the physical separation of donors and acceptors, both molecules show small  $\Delta E_{ST}$  values of 0.04 eV for **TAc3TRZ3** and -0.08 eV for **Ac3TRZ3**. The negative  $\Delta E_{ST}$  is likely due to different molecular geometries in the fully relaxed singlet and triplet states as also identified in **DM-Bm** and **DM-G**. Compared to **Ac3TRZ3** [ $\Phi_{PL}$ : 54% 10 wt% in Ac6 (control compound contains six acridan donors)], **TAc3TRZ3**, containing the stronger dendritic donor, shows a higher  $\Phi_{PL}$  of 63% (10 wt% in

Ac6). The authors ascribed this higher photoluminescence quantum yield to the more efficient charge transfer in **TAc3TRZ3** mediated by the spatial  $\pi$ - $\pi$  interactions. The use of the stronger dendritic donors also leads to a red-shift of the emission spectrum from  $\lambda_{PL}$  of 486 nm for **Ac3TRZ3** to  $\lambda_{PL}$  of 508 nm for **TAc3TRZ3**. The non-doped OLEDs showed EQE<sub>max</sub> of 3.5% and 3.1% for the devices with **Ac3TRZ3** [ $\lambda_{EL}$ : 492 nm; CIE = (0.22, 0.42)] and **TAc3TRZ3** [ $\lambda_{EL}$ : 503 nm; CIE = (0.25, 0.47)], respectively. However, the doped devices showed improved EQE<sub>max</sub> of 11.0 % and 14.2 % for the devices with **Ac3TRZ3** [ $\lambda_{EL}$ : 520 nm; CIE = (0.30, 0.54)] and **TAc3TRZ3** [ $\lambda_{EL}$ : 538 nm; CIE = (0.22, 0.48)], respectively.



**Figure 38.** Molecular structures and properties of TADF polymer emitters with nonconjugated backbone and TSCT effect.

Wang *et al.*<sup>[74]</sup> also prepared a series of blue TADF polymers that exploit only TSCT to electronically couple donor and acceptor moieties (**Figure 38**). The polymers use a nonconjugated polyethylene backbone, 9,9-dimethyl-10-phenyl-acridan (**Ac**) or 9,9-bis(1,3-di-*tert*-butylphenyl)-10-phenyl-acridan (**TBAc**) as the donor pendant and TRZ as the acceptor

pendant. In this design, D and A units are not directly electronically decoupled, but are spatially close to each other, allowing through-space, rather than the through-bond charge transfer processes to occur. The TSCT character observed from the Ac-based polymer results in both a small  $\Delta E_{ST}$  of 0.019 eV and moderate  $\Phi_{PL}$  of 60% in the neat film. In comparison, the **TBAc**-based polymers only exhibited fast prompt fluorescence emission without TSCT contribution because the steric 1,3-dibutylphenyl groups separate the electron-donating acridan unit from the electron-accepting triazine unit, which weakens the TSCT transition. The device with a polymer consisting of 95 mol% Ac and 5 mol% TRZ content demonstrated the best device performance with CIE coordinates of (0.18, 0.27) and an EQE<sub>max</sub> of 12.1%. The optoelectronic properties of the aforementioned materials are summarized in **Table 15** and the device performance metrics are summarized in **Table 16**.

**Table 15.** Summary of photophysical and electrochemical properties

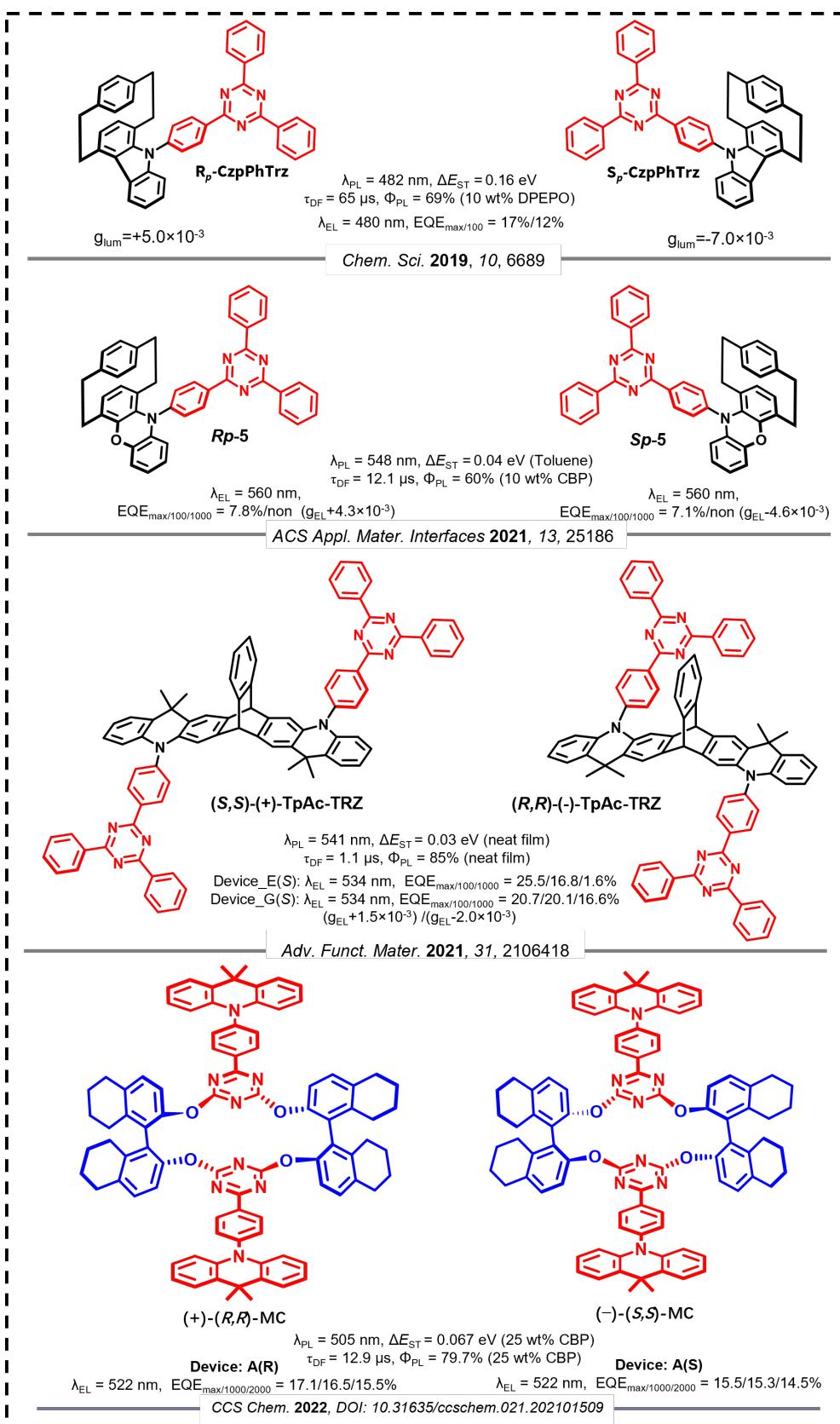
Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
XPT	562/7.7/2.3(PhMe)	-/66/-(10 wt% in DPEPO)	-	-4.99	-2.22	[60]
XCT	419/5.9/3.0(PhMe)	-/-/-	-	-5.58	-2.39	[60]
XtBuCT	451/6/2.0(PhMe)	-35/-(10 wt% in DPEPO)	-	-5.49	-2.33	[60]
oBFCzTRZ	~480/-/(PhMe)	-/97.9/5.4 (in neat film)	0.002	-6.10	-3.40	[61]
mBFCzTRZ	~450/-/(PhMe)	-/31.1/29.6 (in neat film)	0.19	-6.10	-3.30	[61]
pBFCzTRZ	~430/-/(PhMe)	-/85.3/31.2 (in neat film)	0.30	-6.10	-3.20	[61]
oBFCzTRZ	-/-/-	-/100/-(8 wt% in DPEPO)	-			[61]
DPA- <i>o</i> -TRZ	-/-/-	-/-/1.62(mCP)	0.03	-5.84	-3.25	[62]
MPA- <i>o</i> -TRZ	-/-/-	-/-/1.08(mCP)	0.01	-5.7	-3.26	[62]
BCzTRZDBF	-/-/-	~490/83.4/5.4(5 wt% in mCBPTRZ)	0.06	-5.85	-3.34	[64]
TCzTRZDBF	-/-/-	~510/86.3/4.4(5 wt% in mCBPTRZ)	0.01	-5.87	-3.43	[64]
IDCzTRZDBF	-/-/-	~480/85.4/2.8(5 wt% in mCBPTRZ)	0.05	-5.88	-3.34	[64]
SF12oTz	-/-/-	491/92/4.6(10 wt% in DPEPO)	0.05	-5.58	-2.92	[65]
SF23oTz	-/-/-	473/86/4.3(10 wt% in DPEPO)	0.08	-5.6	-2.87	[65]
SF34oTz	-/-/-	479/65/8.2(10 wt% in DPEPO)	0.29	-5.71	-2.88	[65]
DM-Me-B	-/-/(PhMe)	495/73/3.1(30 wt% in DPEPO)	0.25	-5.29	-2.88	[68]
DM-B	-/-/(PhMe)	493/96/5.0(20 wt% in DPEPO)	0.17	-	-	[67]
DM-Bm	-/-/(PhMe)	495/92/4.5(20 wt% in DPEPO)	-0.08	-	-	[67]
DM-G	-/-/(PhMe)	504/88/3.3(20 wt% in DPEPO)	-0.11	-	-	[67]
DM-X	-/-/(PhMe)	504/32/-(20 wt% in DPEPO)	0.03	-	-	[67]
DM-Z	-/-/(PhMe)	488/23/-(20 wt% in DPEPO)	0.54	-	-	[67]
Ac3TRZ3	486/-/(PhMe)	-/54/3.5(10 wt% in Ac6)	0.08	-5.28	-2.96	[73]
TAc3TRZ3	508/-/(PhMe)	-/63/3.2(10 wt% in Ac6)	0.04	-5.10	-2.94	[73]
P-Ac50-TRZ50	489/60/1.2(PhMe)	-/-/-	0.02	-5.26	-2.68	[74]

P-TBAc50-TRZ50	444/9-(PhMe)	-/-	-	-5.30	-2.64	[74]
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**Table 16.** Summary of device structures and performances

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a)</sup> [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref.
XPT	ITO/TAPC/DPEPO: emitter/DPEPO/TmPyPb/LiF/Al	10% 584	-	~3.2	10/-/-	-/-	[60]
XtBuCT	ITO/TAPC/DPEPO: emitter/DPEPO/TmPyPb/LiF/Al	10% 488	-	~4.0	4/-/-	-/-	[60]
oBFCzTRZ	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20%oB FCzTRZ/TSPO1 /TPBi/LiF/Al	477 (0.18, 0.31)	-	20.4/-/-	20.0/17.4	[61]	
mBFCzTRZ	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20%m BFCzTRZ/TSPO1 /TPBi/LiF/Al	473 (0.17, 0.25)	-	13.2/-/-	~8.0/~2.0	[61]	
pBFCzTRZ	ITO/PEDOT:PSS/TAPC/mCP/DPEPO:20%pB FCzTRZ/TSPO1 /TPBi/LiF/Al	465 (0.15, 0.18)	-	16.7/-/-	~10.0/~5.0	[61]	
DPA-o-TRZ	ITO/PEDOT:PSS/TAPC/mCP/mCP:TSPO1: 5% emitter/TSPO1/LiF/Al	497 (0.21, 0.45)	~4.2	17.2/34.5/45	~16.0/10.5	[62]	
MPA-o-TRZ	ITO/PEDOT:PSS/TAPC/mCP/mCP:TSPO1: 5% emitter/TSPO1/LiF/Al	532 (0.35, 0.58)	~4.2	16.3/41.9/54.7	~15.0/12.7	[62]	
BCzTRZDBF	ITO/DNTPD/BPBPA/PCzAC/mCBPTRZ:5 wt%BCzTRZDBF/DBFTRZ/ZADN/LiF/Al	~490	(0.24, 0.52)	-	20.1/35.1/59.6	~20.0/~15.0	[64]
TCzTRZDBF	ITO/DNTPD/BPBPA/PCzAC/mCBPTRZ:5 wt% TCzTRZDBF /DBFTRZ/ZADN/LiF/Al	~500 (0.27, 0.57)	-	23.5/44.7/74.8	~23.0/~20.0	[64]	
IDCzTRZDBF	ITO/DNTPD/BPBPA/PCzAC/mCBPTRZ:5 wt% IDCzTRZDBF /DBFTRZ/ZADN/LiF/Al	~485 (0.22, 0.48)	-	12.2/19.3/33.6	~12.0/~11.0	[64]	
SF12oTz	ITO/MoO <sub>3</sub> /TAPC/mCP/DPEPO: SF12oTz/DPEPO/TmPyPB /LiF/Al	20% 496	(0.23, 0.47)	2.9	22.4/59.1/60.4	22.3/20.0	[65]
SF23oTz	ITO/MoO <sub>3</sub> /TAPC/mCP/DPEPO: SF23oTz/DPEPO/TmPyPB /LiF/Al	20% 484	(0.19, 0.35)	3.2	19.6/45.5/46.9	18.9/15.9	[65]
SF34oTz	ITO/MoO <sub>3</sub> /TAPC/mCP/DPEPO: SF34oTz/DPEPO/TmPyPB /LiF/Al	20% 482	0.18, 0.37	3.2	14.6/31/31.4	8.5/3.1	[65]
DM-Me-B	ITO/HAT- CN/TAPC/TCTA/mCP/DPEPO:30% emitters/DPEPO /TmPyPB/Liq/Al	500 (0.26, 0.48)	-	19.2/-/-	-	[68]	
DM-B	ITO/HAT- CN/TAPC/TCTA/mCP/DPEPO:50%emitters/D PEPO /TmPyPB/Liq/Al	~488 (0.20, 0.44)	2.8	27.4/68.1/-	~27.0/24.4	[67]	
DM-Bm	ITO/HAT- CN/TAPC/TCTA/mCP/DPEPO:30%emitters/D PEPO /TmPyPB/Liq/Al	~500 (0.22, 0.48)	2.6	21.7/62.7/-	~21.0/19.7	[67]	
DM-G	ITO/HAT- CN/TAPC/TCTA/mCP/DPEPO:20%emitters/D PEPO /TmPyPB/Liq/Al	~500 (0.24, 0.50)	3	18.5/47.5/-	~18.0/15.4	[67]	
DM-X	ITO/HAT- CN/TAPC/TCTA/mCP/DPEPO:20%emitters/D PEPO /TmPyPB/Liq/Al	~500 (0.23, 0.45)	3.4	4.3/10.2/-	-/-	[67]	
DM-Z	ITO/HAT- CN/TAPC/TCTA/mCP/DPEPO:30%emitters/D PEPO /TmPyPB/Liq/Al	~488 (0.19, 0.35)	3.9	3.2/5.1/-	-/-	[67]	
Ac3TRZ3	ITO/PEDOT:PSS/Ac6: wt%Ac3TRZ3/TSPO1/TmPyPB/LiF/Al	10 492 (0.22, 0.42)	2.9	11/-/30.3	10.4/6.6	[73]	
TAc3TRZ3	ITO/PEDOT:PSS/Ac6: TAc3TRZ3/TSPO1/TmPyPB/LiF/Al	10 503 (0.25, 0.47)	2.9	14.2/-/40.6	13.5/10.4	[73]	
P-Ac50-TRZ50	ITO/PEDOT:PSS/EML/TSPO1/TmPyPB/LiF/ Al	497 0.222, 0.428	3.4	3.1/-/8.5	3.0/3.0	[74]	
P-Ac95-TRZ05	ITO/PEDOT:PSS/polymer /TSPO1/TmPyPB/LiF/Al	472 0.176, 0.269	3.2	12.1/-/24.8	12.0/11.5	[74]	
P-TBAc50-TRZ50	ITO/PEDOT:PSS/polymer /TSPO1/TmPyPB/LiF/Al	453 0.235, 0.243	3.8	0.21/-/0.6	0.18/-	[74]	
P-TBAc95-TRZ05	ITO/PEDOT:PSS/polymer /TSPO1/TmPyPB/LiF/Al	445 0.207, 0.196	5.8	0.33/-/0.51	0.14/-	[74]	

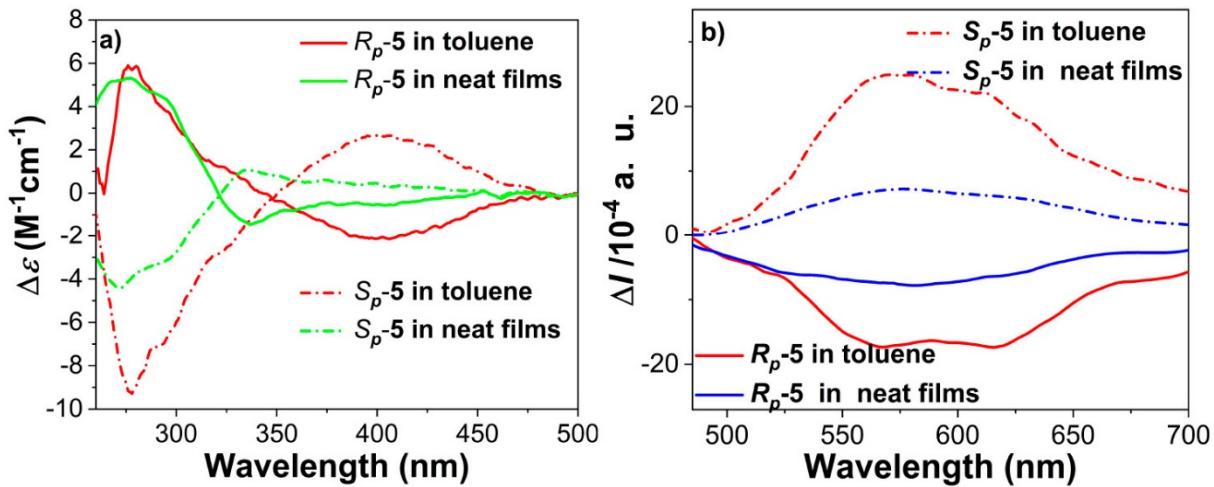
## 10. Chiral TADF emitters containing triazine



**Figure 39.** Molecular structures and properties of chiral TADF emitters based on triazine.

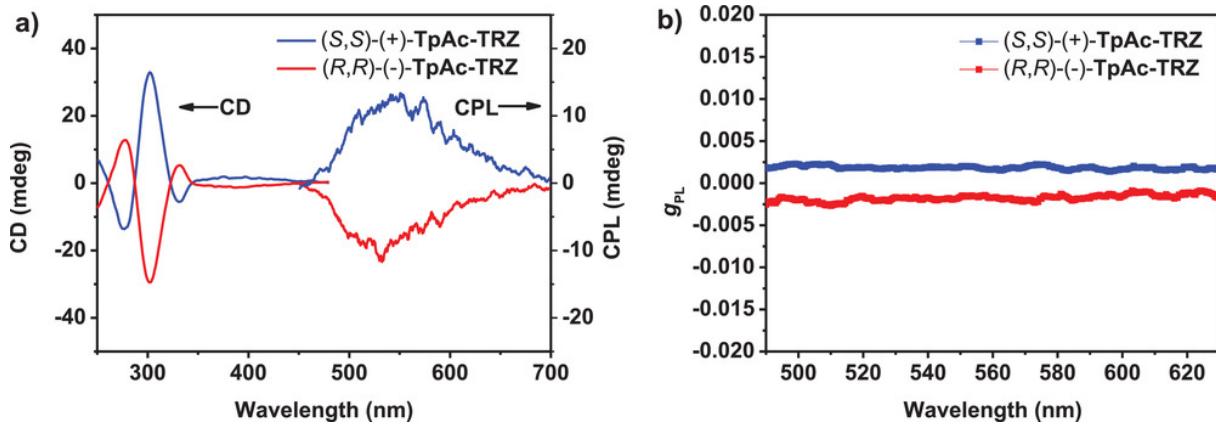
Circularly polarized luminescence (CPL) materials have attracted great attention due to their potential applications in optical data storage,<sup>[75]</sup> chirality sensing,<sup>[76]</sup> organic electronic devices<sup>[77]</sup> and bio imaging.<sup>[78]</sup> In 1997, Meijer *et al.*<sup>[79]</sup> developed the first example of a circularly polarized OLED (CP-OLED). Since then, many chiral emitters have been explored, including chiral polymers,<sup>[80]</sup> chiral phosphorescent complexes<sup>[81]</sup> and chiral TADF small molecules.<sup>[82]</sup> Of the latter, there are a few examples of chiral TADF emitters incorporating a triazine acceptor.

In 2019, Zysman-Colman and co-workers<sup>[14]</sup> introduced the carbazolophane (Czp) donor unit (indolo[2.2]paracyclophane) for the design of chiral TADF emitters, **R<sub>p</sub>-CzpPhTrz** and **S<sub>p</sub>-CzpPhTrz** (**Figure 39**). The bulky carbazolophane donor unit increased the torsion between the donor and the phenylene bridge compared to the control compound **CzPhTrz** (same structure as **CzTRZ** discussed in section 2). As discussed above, **CzPhTrz** was known to be a non-TADF compound; however, the larger torsion between the donor and the bridge in **CzpPhTrz** leads to a decreased  $\Delta E_{ST}$  of 0.16 eV, which contributes to the turn on of the TADF. **rac-CzpPhTrz** is a sky-blue emitter with  $\lambda_{PL}$  of 482 nm, and high  $\Phi_{PL}$  of 69% in 10 wt% doped **DPEPO** films. The chiroptical properties of the enantiomers **R<sub>p</sub>-CzpPhTrz** and **S<sub>p</sub>-CzpPhTrz** reveal mirror image circular dichroism (CD) and circularly polarized luminescence (CPL), with  $g_{lum}$  values of  $5 \times 10^{-3}$  /  $-7 \times 10^{-3}$ , respectively. Sky blue-emitting OLEDs were fabricated with an EQE<sub>max</sub> of 17% and associated CIE coordinates of (0.17, 0.25).



**Figure 40.** a) CD and b) CPL spectra of  $S_p\text{-}5/R_p\text{-}5$ . Reproduced with permission.<sup>[83]</sup>

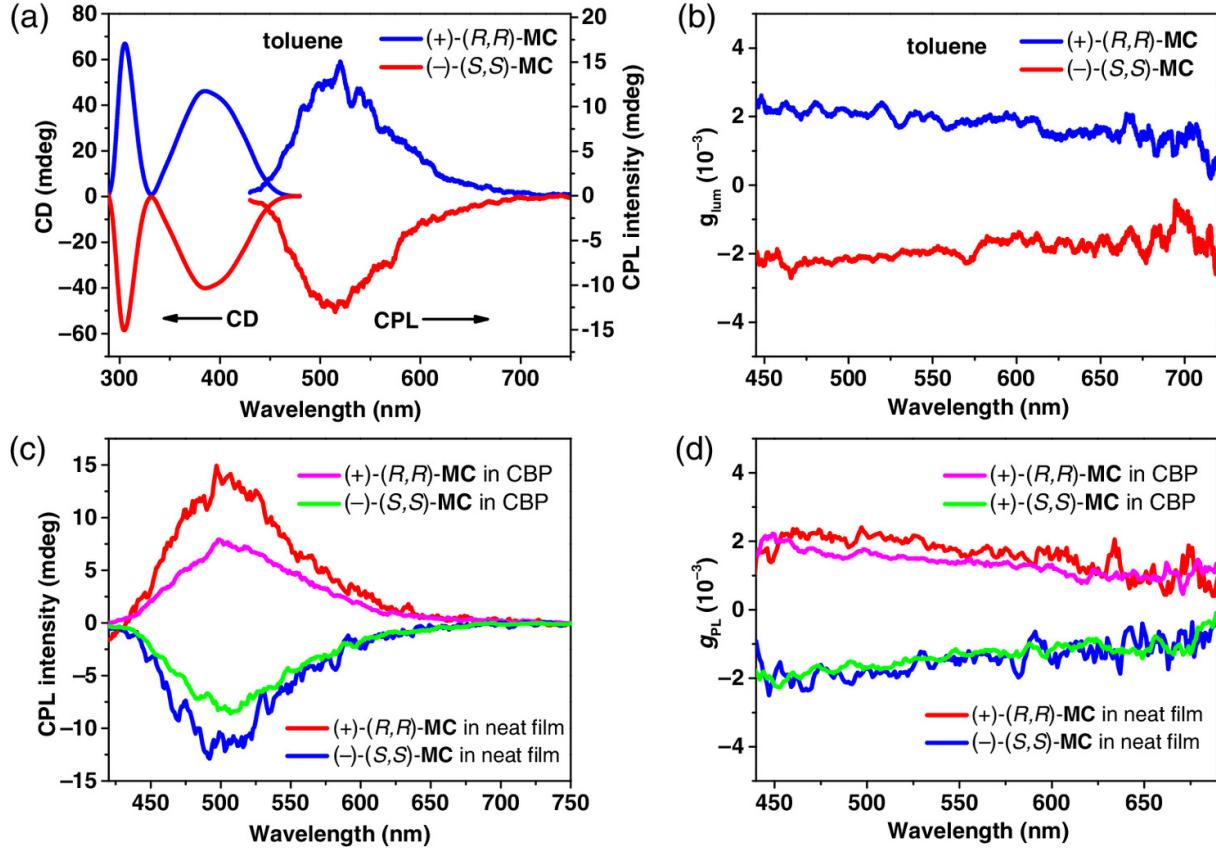
Zheng *et al.*<sup>[83]</sup> reported a pair of chiral TADF emitters (**Figure 39**),  $R_p\text{-}5$  and  $S_p\text{-}5$ , by combining a TRZ acceptor with a chiral donor group (PXZp). PXZp is a phenoxazine derivative containing an annulated [2.2]paracyclophane (pCp) skeleton which has ever been demonstrated as an efficient unit enabling a blue chiral TADF emitters by Zysman-Colman and co-workers in 2019.<sup>[14]</sup> Similar to their analogue **PXZ-TRZ** ( $\lambda_{\text{PL}}$ : 540 nm;  $\Phi_{\text{PL}}$ : 66%;  $\tau_{\text{d}}$ : 1.1  $\mu\text{s}$  in 8 wt% mCPCN;  $\Delta E_{\text{ST}}$ : 0.017 eV),  $R_p\text{-}5$  and  $S_p\text{-}5$  exhibited broad PL with  $\lambda_{\text{PL}}$  of 548 nm (in toluene), 539 nm (in neat film), and 527 nm (in 10 wt% CBP film). A moderate  $\Phi_{\text{PL}}$  of 60% as well as a small  $\Delta E_{\text{ST}}$  of 0.03 (in 10 wt% CBP film) resemble the photophysical properties of **PXZ-TRZ**, indicating the weak influence from the pCp on the TADF properties. The dissymmetry values,  $g_{\text{lum}}$ , of  $R_p\text{-}5$  and  $S_p\text{-}5$  (**Figure 40**) are  $-2.4 \times 10^{-3}/+3.3 \times 10^{-3}$  and  $-2.3 \times 10^{-3}/+2.7 \times 10^{-3}$  in toluene ( $1 \times 10^{-5}$  mol L $^{-1}$ ) and neat films, respectively, which are typical values from organic emitters. Solution-processed CP-OLEDs showed EQE<sub>max</sub> of 7.8%/7.1% and  $g_{\text{EL}}$  of  $+1.4 \times 10^{-3}/-2.0 \times 10^{-3}$ . Although the non-doped devices based on  $S_p\text{-}5/R_p\text{-}5$  possessed lower EQE<sub>max</sub> of 2.5%/1.8%; however, the  $g_{\text{EL}}$  improved modestly to  $+4.3 \times 10^{-3}/-4.6 \times 10^{-3}$ .



**Figure 41.** a) CD and CPL spectra of (S,S)-/(R,R)-TpAc-TRZ in neat film states. b)  $g_{PL}$  versus wavelength curves of (S,S)-/(R,R)-TpAc-TRZ in neat film states. Reproduced with permission.<sup>[84]</sup>

A pair of triptycene-based enantiomers, (S,S)-/(R,R)-TpAc-TRZ,<sup>[84]</sup> were developed containing a triptycene-fused diacridine chiral donor (S,S)-/(R,R)-TpAc (**Figure 39**). The compounds can be recognized as the combination of two well-known **DMAC-TRZ** TADF emitters through a nonconjugate 3D triptycene bridge. Therefore, the excellent TADF properties of **DMAC-TRZ** are still maintained for the enantiomers, especially the suppression of concentration quenching in the neat films. The (S,S)-TpAc-TRZ neat film ( $\lambda_{PL}$ : 541 nm;  $\Phi_{PL}$ : 85%;  $\tau_d$ : 1.1  $\mu$ s;  $\Delta E_{ST}$ : 0.03 eV;) showed photophysical properties reminiscent to those of the **DMAC-TRZ** in neat film ( $\lambda_{PL}$ : 500 nm;  $\Phi_{PL}$ : 83%;  $\tau_d$ : 3.6  $\mu$ s;  $\Delta E_{ST}$ : 0.05 eV;), except for the 41 nm red-shift in the PL. The circular dichroism (CD) spectra of (S,S)-(+)-TpAc-TRZ and (R,R)-(−)-TpAc-TRZ exhibited clear mirror-image relationship with alternating positive and negative Cotton effects (**Figure 41**). The  $g_{PL}$  values are  $+1.9 \times 10^{-3}$  for (S,S)-(+)-TpAc-TRZ and  $-1.8 \times 10^{-3}$  for (R,R)-(−)-TpAc-TRZ in neat films. Due to the high molecular weight of the emitters, only solution-processed OLEDs were fabricated, with an  $EQE_{max}$  of 25.5% ( $\lambda_{EL}$ : 534 nm). However, a significant efficiency roll-off was observed and the EQE decreased to 1.6% at a luminance of 1000 cd m<sup>-2</sup>. Similar to  $g_{PL}$ , the  $g_{EL}$  values reached to  $+1.5 \times 10^{-3}$  for

the **(S,S)-(+)-TpAc-TRZ-** based device and  $-2.0 \times 10^{-3}$  for **(R,R)-(-)-TpAc-TRZ**-based device.



**Figure 42.** (a) CD and CPL spectra and (b)  $g_{PL}$  versus wavelength curves of **(+)-(R,R)/(-)-(S,S)-MC** in toluene ( $c = 1 \times 10^{-4}$  M). (c) CPL spectra and (d)  $g_{PL}$  versus wavelength curves of **(+)-(R,R)/(-)-(S,S)-MC** in neat film, and doped film (25 wt % macrocyclic enantiomers: CBP).

Chen *et al.*<sup>[85]</sup> reported a pair of macrocyclic TADF enantiomers **(+)-(R,R)-MC** and **(-)-(S,S)-MC** linking chiral octahydro-binaphthol units with a DMAC-TRZ derivative (Figure 39). The four-unit non-conjugated macrocycle consisting of two donors and two acceptors shows TADF. Similar to **DMAC-TRZ**, the enantiomers **(+)-(R,R)-MC** and **(-)-(S,S)-MC** showed a small  $\Delta E_{ST}$  of 0.067 eV, a short  $\tau_d$  of 1.5  $\mu$ s, suppressed concentration quenching in neat films and  $\Phi_{PL}$  of 79.7% (25 wt% **(+)-(R,R)-MC**: CBP). The  $|g_{PL}|$  of  $2.2 \times 10^{-3}$  in solution (Figure 42)

and mirror image CD spectra are typical for chiral TADF emitters. The solution-processed CP-OLEDs (25 wt% (+)-(R,R)-MC/ (-)-(S,S)-MC: CBP) displayed EQE<sub>max</sub> of 17.1%/15.5% and g<sub>EL</sub> of  $1.5 \times 10^{-3}$  /  $-1.7 \times 10^{-3}$ .

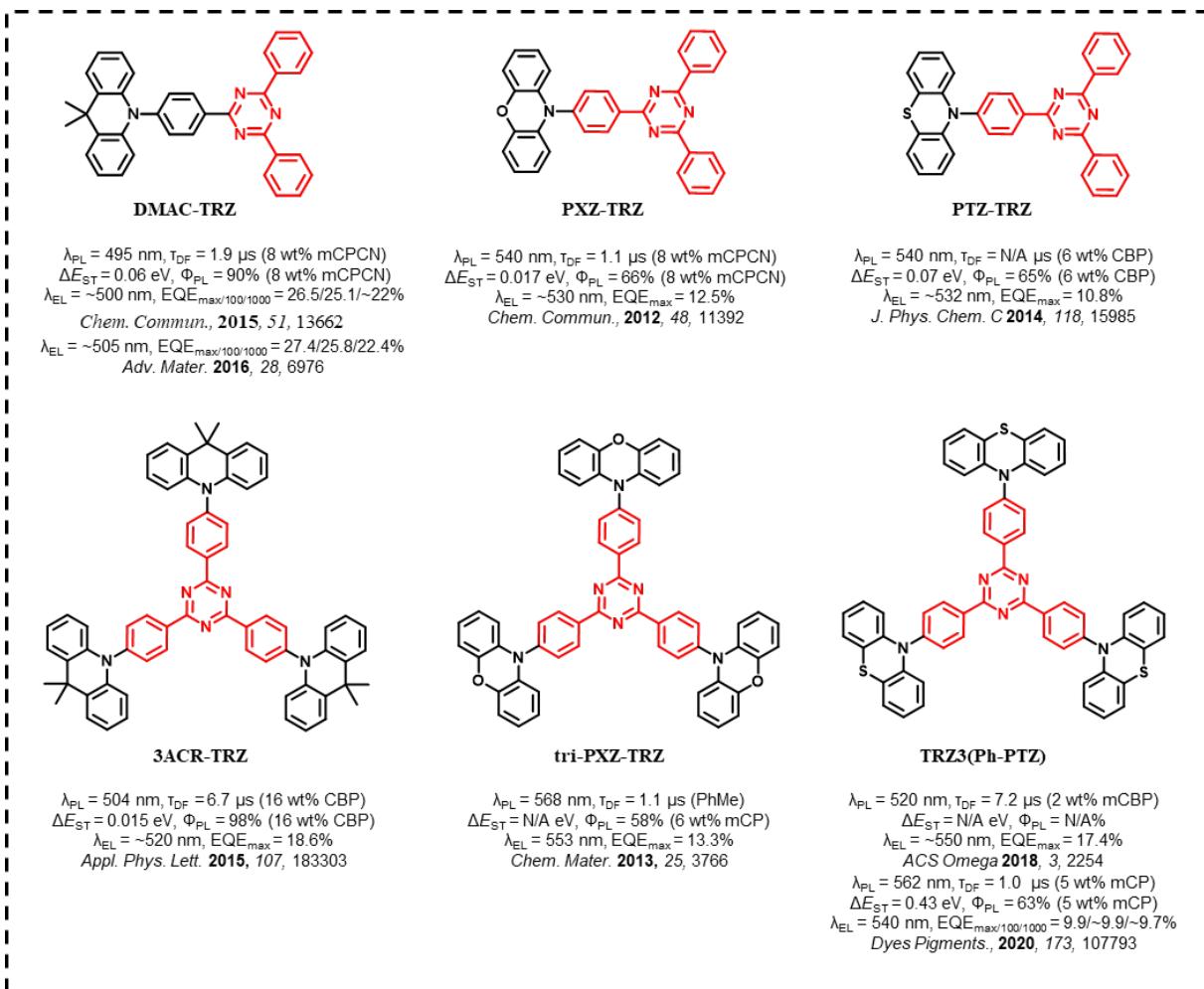
**Table 17.** Summary of photophysical and electrochemical properties

Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
CzPPhTrz	470/70/- (toluene)	482/69/65 (10 wt% DPEPO)	0.16	-5.80	-3.23	[14]
5	548/-/- (toluene)	527/60/12.1 (10 wt% CBP)	0.03	-4.11	-2.70	[83]
(S,S)-(+)-TpAc-TRZ	525/-/-	541/85/1.1 (neat film)	0.03	-5.19	-2.55	[84]
(-)-(S,S)-MC	512/-/-	505/80/12.9 (25 wt% CBP)	2.69	-5.38	-2.72	[85]

**Table 18.** Summary of device structures and performance

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a)</sup> [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref.	
R <sub>p</sub> -5	ITO/PEDOT: PSS/10 wt % R <sub>p</sub> -5:CBP/TPBI/ Ca/Ag	560	(0.42, 0.54)	4.2	7.8/-/23.8	NA	[83]	
S <sub>p</sub> -5	ITO/PEDOT: PSS/10 wt % S <sub>p</sub> -5:CBP/TPBI/ Ca/Ag	560	(0.43, 0.54)	4.2	7.1/-/21.6	NA	[83]	
(S,S)-(+)-TpAc-TRZ	ITO/PEDOT:PSS/(S,S)-(+)-TpAc-TRZ/ TPBi/LiF/Al	534	NA	2.5	25.5/95.9/88.6	16.8/1.6	[84]	
(+)-(R,R)-MC	ITO/PEDOT:PSS/CBP:25 (R,R)-MC/TPBi/LiF/Al	wt % (+)-	522	(0.28, 0.54)	3.8	17.1/37.0/53.7	NA/16.5	[85]
(-)-(S,S)-MC	ITO/PEDOT:PSS/CBP:25 (-)-(S,S)-MC/TPBi/LiF/Al	wt % (-)-(S,S)-	522	(0.28, 0.54)	3.5	15.5/34.9/48.8	NA/15.3	[85]

## 11. TADF emitters based on triazine and other donors



**Figure 43.** Molecular structures and properties of TADF emitters based on triazine and non-carbazole donors.

In addition to carbazole, many alternative *N*-heterocycle donors have also been investigated as building blocks in triazine-based TADF emitters. 9,9-dimethyl-9,10-dihydroacridine (DMAC), 10*H*-phenoxazine (PXZ), 10*H*-phenothiazine (PTZ) are the most popular six-membered *N*-heterocyclic donors used in the design of TADF emitters. The sequence of electron-donating strength typically follows PTZ>PXZ>DMAC>carbazole, with carbazole being the weakest donor. Unlike carbazole, the use of these larger *N*-heterocyclic donors generally produces a highly twisted conformation between the donor and the bridging units when they are *N*-bound because of the increased steric hindrance. As a result, the highly twisted conformation leads to a greater separation of the FMOs, which generally results in a small  $\Delta E_{ST}$ .

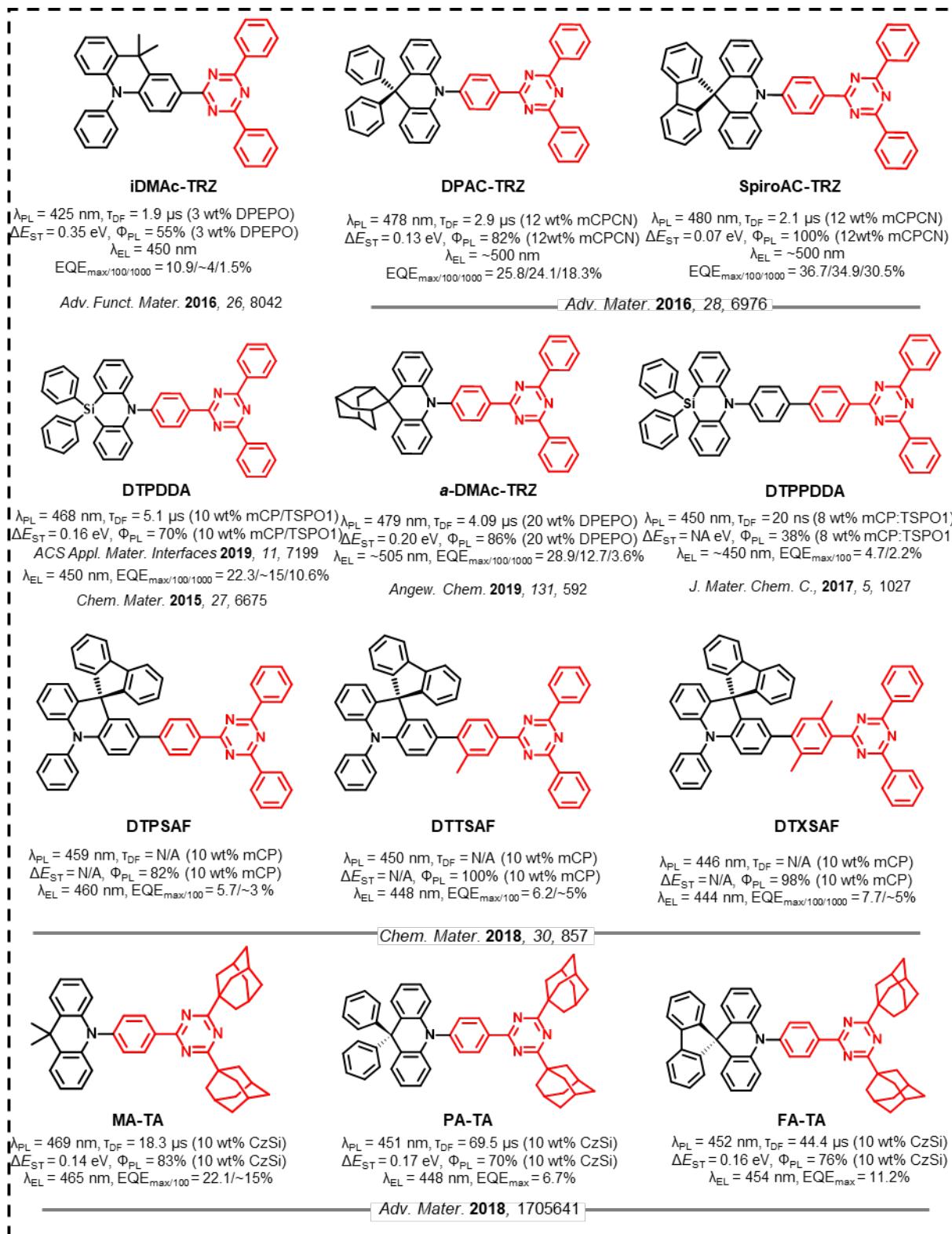
Wu *et al.*<sup>[86]</sup> reported what is now considered one of the most widely studied TADF emitters, **DMAC-TRZ** ( $\lambda_{PL}$ : 495 nm;  $\Phi_{PL}$ : 90%;  $\tau_d$ : 1.9  $\mu$ s;  $\Delta E_{ST}$ : 0.05 eV; in 8 wt% mCPCN) as shown in **Figure 43**. **DMAC-TRZ** shows high  $\Phi_{PL}$  (90%) in an 8 wt% mCPCN doped film, which is not much reduced in the neat film (83%). The  $EQE_{max}$  of doped and non-doped device based on **DMAC-TRZ** were reported to be 26.5% and 20%, respectively, reflecting in part the differences in  $\Phi_{PL}$ . Due to its good solubility, a solution-processed non-doped OLED showed only a slightly lower  $EQE_{max}$  of 17.6%. Hu *et al.*<sup>[87]</sup> observed that photoexciting CT states can lead to a magneto-PL signal in the SOC regime, but not found when photoexciting LE states. This is the first experimental evidence that SOC is produced in CT states. Furthermore, they found that the **DMAC-TRZ**-based OLEDs demonstrated magneto-EL in the high field regime (>10 mT).<sup>[87]</sup> This high-field magneto-EL signal provides direct evidence to indicate that the SOC is indeed enhanced, in the absence of heavy elements, towards developing spin-dependent TADF in OLEDs. Compared with **DMAC-TRZ** ( $\lambda_{PL}$ : 495 nm;  $\Phi_{PL}$ : 90%;  $\tau_d$ : 1.9  $\mu$ s;  $\Delta E_{ST}$ : 0.05 eV; in 8 wt% mCPCN), the emission spectrum of **PXZ-TRZ** ( $\lambda_{PL}$ : 540 nm;  $\Phi_{PL}$ : 66%;  $\tau_d$ : 1.1  $\mu$ s in 8 wt% mCPCN;  $\Delta E_{ST}$ : 0.017 eV) is red-shifted by about 45 nm due to the enhanced electron-donating ability of the PXZ. Even though both compounds show similarly small  $\Delta E_{ST}$  and relatively short delayed fluorescence lifetimes ( $\tau_d$ : 1.1  $\mu$ s;  $\Delta E_{ST}$ : 0.017 eV; 8 wt% **PXZ-TRZ** in mCPCN), the much lower  $\Phi_{PL}$  of **PXZ-TRZ** (66% in 8 wt% mCPCN) indicates greater nonradiative decay, which may be ascribed to the energy gap law.<sup>[86]</sup> An OLED containing 6 wt% **PXZ-TRZ** doped in CBP as the emitter layer exhibited an  $EQE_{max}$  of 12.5% ( $\lambda_{EL}$  of 529 nm).<sup>[88]</sup> The authors also demonstrated that the orientation of the TDM of **PXZ-TRZ** can be modulated in mCBP by varying the temperature on the ITO glass holder during the deposition of the emitting layer. The horizontal orientation of the TDM of **PXZ-TRZ** can be enhanced by

lowering the temperature of the ITO glass from 300 K to 200 K. As a result, the EQE<sub>max</sub> is improved from 9.6% at 300 K to 11.9% at 200 K.<sup>[89]</sup>

The use of PTZ as a donor introduces additional conformational dynamics due to the existence of two ground-state conformers resulting from puckering of the PTZ ring, each with their own associated  $\Delta E_{ST}$ .<sup>[90]</sup> **PTZ-TRZ** shows a slightly red shifted emission maximum ( $\lambda_{PL}$ : 562 nm; in toluene) and similar  $\Phi_{PL}$  (65.8%; in 2 wt% mCBP) to those of its analogue **PXZ-TRZ** (545 nm in toluene,  $\Phi_{PL}$  65.7%; 6 wt% in CBP)<sup>[88]</sup> while the  $\Delta E_{ST}$  (0.07 eV) of **PTZ-TRZ** is the largest amongst the three compounds: **DMAC-TRZ**, **PXZ-TRZ** and **PTZ-TRZ**. The relatively similar electron-donating strength of the PXZ and PTZ groups in **PXZ-TRZ** and **PTZ-TRZ** translates to similar emission spectra and  $\Phi_{PL}$ . The device containing **PTZ-TRZ** exhibited an EQE<sub>max</sub> of 10.8%, which is also of similar performance to that of **PTZ-TRZ**-based OLED although the two reports<sup>[88, 90]</sup> used different hosts within the emissive layer. The EL spectrum shows two emission bands, one high-energy band at around 393 nm and a stronger broad band at 532 nm, resulting from the simultaneous emission from the two conformers.

**3ACR-TRZ**, an analogue of **DMAC-TRZ** containing 3 donor groups, reported by Wada *et al.*<sup>[91]</sup> shows somewhat similar photophysical properties to **DMAC-TRZ**, with near unity  $\Phi_{PL}$  (98%; in 16 wt% CBP) and a slightly red-shifted emission ( $\lambda_{PL}$ : 504 nm; in 16 wt% CBP). The small red-shift implies that there is only a weak influence on the stabilization of the singlet state by increasing the number of DMAC units. However, the multiple donors could further delocalize the distribution of HOMO, thus leading to a smaller  $\Delta E_{ST}$  (0.015 eV). The solution-processed device showed an EQE<sub>max</sub> of 18.6% at a emission of  $\lambda_{EL}\sim$ 520 nm. Analogue compounds containing either trisubstituted PXZ or PTZ donor groups have also been reported.<sup>[92]</sup> Similar evolution of the photophysical properties from the linear D-A compounds

to the D<sub>3</sub>-symmetric analogs was observed for **tri-PXZ-TRZ** ( $\lambda_{PL}$ : 568 nm;  $\Phi_{PL}$ : 58% in 6 wt% mCP;  $\tau_d$ : 1.10  $\mu$ s in toluene) and **TRZ3(Ph-PTZ)** ( $\lambda_{PL}$ : 575 nm in toluene;  $\tau_d$ : 7.2  $\mu$ s in 2 wt% mCP).<sup>[93]</sup> The OLEDs based on **TRZ3(Ph-PTZ)** shows yellowish-green electroluminescence with a  $\lambda_{EL} \sim 550$  nm and exhibits a much higher EQE<sub>max</sub> of 17.4% than those of the devices of the **tri-PXZ-TRZ**-based OLEDs (EQE<sub>max</sub>: 13.3%,  $\lambda_{EL}$ : 553 nm). The Zysman-Colman group reported yellow-emitting OLEDs based on **tri-PXZ-TRZ** by doping the emitter into a bespoke host **4-mCBPy**, thus demonstrating an improved device performance with an EQE<sub>max</sub> of 19.4% and a dramatically reduced efficiency roll-off (EQE = 16.0% at a luminance of 10 000 cd m<sup>-2</sup>).<sup>[94]</sup>



**Figure 44.** Molecular structures and properties of TADF emitters based on triazine and DMAC derivatives donors.

**i-DMAc-TRZ** ( $\lambda_{PL}$ : 452 nm;  $\Phi_{PL}$ : 55%;  $\tau_d$ : 1840  $\mu$ s;  $\Delta E_{ST}$ : 0.35 eV; in 3 wt% DPEPO) is a constitutional isomer of **DMAC-TRZ** but where the phenylene bridge is C-bound *para* to the nitrogen atom of the DMAC donor (**Figure 44**).<sup>[95]</sup> Distinct from the highly twisted conformation of **DMAC-TRZ**, such a structural change results in a flattened conformation, leading to enhanced conjugation between the DMAC and TRZ, and hence a much larger  $\Delta E_{ST}$  (0.35 eV) than that found for **DMAC-TRZ** (0.06 eV). Despite the emission spectrum being blue-shifted from **DMAC-TRZ** ( $\lambda_{PL}$ : 495 nm) to **i-DMAc-TRZ** ( $\lambda_{PL}$ : 452 nm), the  $\Phi_{PL}$  is reduced by almost half. The device based **i-DMAc-TRZ** (10 wt% in mCBP) presents a deep-blue emission with  $\lambda_{EL}$  at 450 nm and CIE coordinates of (0.15, 0.11) and an  $EQE_{max}$  of 10.9%.

**DTPSAF** ( $\lambda_{PL}$ : 459 nm;  $\Phi_{PL}$ : 82%; in 10 wt% mCP) is the SpiroAC analogue of **i-DMAc-TRZ**. When compared to **i-DMAc-TRZ**, **DTPSAF** has a similar  $\lambda_{PL}$  but has a higher  $\Phi_{PL}$  due to the restricted rotational flexibility conferred by the rigid bulky SpiroAC group. Addition of methyl groups on the phenylene bridge of **DTPSAF** to generate **DTTSAF** further enhances the restriction of the rotation and produces a more twisted conformation, leading to a near unity  $\Phi_{PL}$  and a 9 nm blue-shift in the emission for **DTTSAF** ( $\lambda_{PL}$ : 450 nm;  $\Phi_{PL}$ : 100%; in 10 wt% mCP), and **DTXSAF** ( $\lambda_{PL}$ : 446 nm;  $\Phi_{PL}$ : 98%; in 10 wt% mCP), which contain tolueneyl and xylenyl bridges, respectively. Distinct to **i-DMAc-TRZ**, there is no reported TADF in these analogues, which based on DFT calculations can be ascribed to their too large  $\Delta E_{ST}$  values. The EQEs of the **DTTSAF**- and **DTXSAF**-doped OLEDs (6.2 and 7.7%, respectively) were higher than that of the **DTPSAF**-doped device (5.7%) due to the higher  $\Phi_{PL}$  values of **DTTSAF** ( $\Phi_{PL}$ : 100%) and **DTXSAF** ( $\Phi_{PL}$ : 98%) over that of **DTPSAF** ( $\Phi_{PL}$ : 82%). **DTXSAF**-based OLEDs show the deepest blue emission with CIE coordinates of (0.15, 0.08) and  $\lambda_{EL}$  of 444 nm compared to those of **DTPSAF** ( $\lambda_{EL}$  of 460 nm) and **DTTSAF** ( $\lambda_{EL}$  of 448 nm).<sup>[96]</sup>

Wu *et al.*<sup>[97]</sup> reported two compounds **DPAC-TRZ** ( $\lambda_{PL}$ : 478 nm;  $\Phi_{PL}$ : 82%;  $\Delta E_{ST}$ : 0.133 eV; in 12 wt% mCPCN) and **SpiroAC-TRZ** ( $\lambda_{PL}$ : 480 nm; ( $\Phi_{PL}$ : 100%;  $\Delta E_{ST}$ : 0.072 eV; in 12 wt% mCPCN), both of which are derivatives of **DMAC-TRZ** but where the two methyl groups on the acridan are replaced by phenyl and 2,2'-biphenyl (spirobiphenyl), respectively. In contrast to **DMAC-TRZ**, the structural modifications of the donor contribute to a blue-shift of the emission. The fluorescence (phosphorescence) spectra of **SpiroAC-TRZ**, **DPAC-TRZ**, and **DMAC-TRZ** in 12 wt% mCPCN films are structureless with  $\lambda_{PL}$  of 480 nm (483 nm), 478 nm (482 nm), and 495 nm (504 nm), respectively. Both compounds possess small  $\Delta E_{ST}$  values though larger than that of **DMAC-TRZ**, thus leading to relatively short delayed lifetime (**DPAC-TRZ**:  $\tau_d$  of 2.9  $\mu$ s; **SpiroAC-TRZ**:  $\tau_d$  of 2.1  $\mu$ s). **DPAC-TRZ** has a lower  $\Phi_{PL}$  of 82% than **SpiroAC-TRZ** ( $\Phi_{PL}$ : 100%), the latter of which is higher compared with **DMAC-TRZ** ( $\lambda_{PL}$ : 495 nm;  $\Phi_{PL}$ : 90%;  $\tau_d$ : 1.9  $\mu$ s;  $\Delta E_{ST}$ : 0.05 eV; in 8 wt% mCPCN). **DPAC-TRZ** and **SpiroAC-TRZ** have TDMs that are highly horizontally oriented, with  $\theta$  of 78%, 83% for **DPAC-TRZ** and **SpiroAC-TRZ**, respectively, in 12 wt% mCPCN, resulting in  $EQE_{max}$  of 25.8% and 36.7% in the respective devices. The device with **SpiroAC-TRZ** retains a high  $EQE_{1000}$  of 30.5% at a luminance of 1000 cd m<sup>-2</sup>.

Replacement of the methyl groups on the acridan with a spiro adamantyl unit, as in **a-DMAC-TRZ** ( $\lambda_{PL}$ : 479 nm;  $\Phi_{PL}$ : 86.1%;  $\tau_d$ : 4.09  $\mu$ s in 20 wt% DPEPO;  $\Delta E_{ST}$ : 0.20 eV) results in a blue-shifted emission, but with otherwise comparable photophysical properties to that of **DMAC-TRZ**<sup>[86]</sup> ( $\lambda_{PL}$ : 495 nm;  $\Phi_{PL}$ : 90%;  $\tau_d$ : 1.9  $\mu$ s;  $\Delta E_{ST}$ : 0.05 eV; in 8 wt% mCPCN). The introduction of the rigid and bulky adamantanyl moiety not only suppresses the geometry relaxation in the excited state, but also induced the formation of quasi-axial conformer (QAC) and quasi-equatorial conformer (QEC) geometries corresponding to a shoulder emission peak at around 419 nm and a main emission peak at around 479 nm in the PL spectrum, respectively.

The  $\Delta E_{ST}$  values of two conformers were confirmed separately by different excitation wavelength. With an excitation wavelength of 360 nm, the  $S_1$  and  $T_1$  energies of QAC were calculated to be 2.97 and 2.66 eV, demonstrating a large  $\Delta E_{ST}$  of 0.31 eV. At an excitation wavelength of 420 nm, the  $S_1$  and  $T_1$  energy levels of **a-DMAC-TRZ** for QEC were calculated to be 2.79 and 2.59 eV, thus translating to a smaller  $\Delta E_{ST}$  of 0.20 eV. Owing to the effect of the rigid molecular backbone and the degenerate alignment of  $^3LE$  of QAC and  $^3CT$  of QEC for efficient dual fluorescence emission, the resulting OLEDs achieved a high EQE<sub>max</sub> of 28.9% at CIE coordinates of (0.18, 0.35).<sup>[98]</sup>

Wada *et al.*<sup>[99]</sup> demonstrated that by replacing the distal phenyl groups attached to the triazine with adamantyl substituents, the acceptor becomes weaker, leading to a blue-shifted emission compared to **DMAC-TRZ**,<sup>[86]</sup> **DPAC-TRZ**<sup>[97]</sup> and **SpiroAc-TRZ**.<sup>[97]</sup> Solution-processed devices employing the emitters **FA-TA** ( $\lambda_{PL}$ : 452 nm in toluene;  $\Phi_{PL}$ : 76%;  $\tau_d$ : 44.4  $\mu$ s in 10 wt% CzSi;  $\Delta E_{ST}$ : 0.16 eV), **MA-TA** ( $\lambda_{PL}$ : 469 nm in toluene;  $\Phi_{PL}$ : 83%;  $\tau_d$ : 18.3  $\mu$ s in 10 wt% CzSi;  $\Delta E_{ST}$ : 0.14 eV) and **PA-TA** ( $\lambda_{PL}$ : 450 nm in toluene;  $\Phi_{PL}$ : 70%;  $\tau_d$ : 69.5  $\mu$ s in 10 wt% CzSi;  $\Delta E_{ST}$ : 0.17 eV) exhibited EQE<sub>max</sub> of 11.2% with CIE coordinates of (0.15, 0.13), 22.1% with CIE coordinates of (0.15, 0.19) and 6.7% with CIE coordinates of (0.15, 0.10), respectively, shown in **Figure 44**. **MA-TA** represents one of the most efficient blue emitters for solution-processed OLEDs reported to date. However, the much poorer performance of the devices with **PA-TA** and **FA-TA** are likely in part due to the random orientation of the TDMs in the spin-coated films when compared to the highly horizontally oriented TDMs of **DPAC-TRZ** and **SpiroAc-TRZ** in evaporated 12 wt% mCPCN, although the emitters are similar in structure. The higher energy excited states of **PA-TA** and **FA-TA** also make it difficult to find suitable host materials or blocking layers in devices, which might be the reason for this poor performance.

Replacement of the spiro-carbon in **DPAC-TRZ**<sup>[97]</sup> with a silicon atom results in the emitter **DTPDDA**.<sup>[100]</sup> This compound shows a 10 nm blue-shifted emission ( $\lambda_{PL}$ : 468 nm;  $\Phi_{PL}$ : 70%;  $\tau_d$ : 5.1  $\mu$ s in 10 wt% mCP/TSPO1;  $\Delta E_{ST}$ : 0.16 eV) than **DPAC-TRZ** ( $\lambda_{PL}$ : 478 nm;  $\Phi_{PL}$ : 82%;  $\Delta E_{ST}$ : 0.13 eV; in 12 wt% mCPCN), yet both compounds have comparable  $\Delta E_{ST}$ . However, the larger silicon atom radius makes **DTPDDA** more flexible, which adversely reduces its  $\Phi_{PL}$ . A blue OLED with the CIE coordinates of (0.15, 0.20) showed an  $EQE_{max}$  of 22.3%; the EQE reduced to 10.6% at 1000 cd m<sup>-2</sup> when the emitter was doped in a mixed exciplex co-host system of mCP:TSPO1.

The addition of a second phenylene ring within the bridge in **DTPDDA** resulted in an increased spatial separation between the donor and acceptor, as well as an improved horizontal dipole ratio of the TDM. **DTPPDDA** ( $\lambda_{PL}$ : 450 nm in toluene;  $\Phi_{PL}$ : 38%;  $\tau_d$ : 20 ns in 8 wt% mCP/TSPO1;  $\Delta E_{ST}$ : 0.04 eV) showed a TDM  $\theta$  value of 0.73 compared to that of **DTPDDA** (0.66: isotropic). However, the extremely short “delayed lifetime” of 20 ns clearly indicates that **DTPPDDA** is not a TADF emitter. The biphenylene ring will adversely result in a lower local-excited triplet state energy than the energy of its lowest charge transfer triplet state ( $^3LE < ^3CT$ ) because the former is sensitive to the enhanced conjugation while the latter is predominantly determined by the charge transfer strength. As a result, the upconversion of the triplet excitons become impossible due to its large barrier from the  $^3LE$  to the  $S_1$  state. Although the  $\Delta E_{ST}$  was determined to be as small as 0.04 eV, which does not align with this analysis. The triplet energy was determined according to the delayed fluorescence which however didn’t provide the delayed time for the measurement of phosphorescence. The OLED employing **DTPPDDA** showed an  $EQE_{max}$  of 4.7% and deep blue emission at CIE coordinates of (0.15,

0.09).<sup>[101]</sup> The optoelectronic characterization of the aforementioned materials are summarized in **Table 19** and device performance metrics are summarized in **Table 20**.

**Table 19.** Summary of photophysical and electrochemical properties

Emitter	Solution $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	Solid State $\lambda_{PL}/\Phi_{PL}/\tau_d$ (medium) [nm/%/μs]	$\Delta E_{ST}$ [eV]	HOMO [eV]	LUMO [eV]	Ref.
DMAC-TRZ	-/-/-	495/90/1.9(12 wt% in mCPCN)	0.06	-5.61	-3.12	[86]
PXZ-TRZ	-/29.5/0.676	540/66/1.1(8 wt% in mCPCN)	0.02	-5.50	-3.10	[86]
PTZ-TRZ	-/-/(PhMe)	540/65/-(6 wt% in CBP)	0.07	-5.50	-3.00	[90]
3ACR-TRZ	51194-(PhMe)	504/98/6.7(16 wt% in CBP)	0.02	-	-	[91]
tri-PXZ-TRZ	568/-/1.101(PhMe)	-/58/-(6 wt% in mCP)	-	-5.70	-3.40	[92]
TRZ3(Ph-PTZ)	575/-/(PhMe)	520/-/7.2(2 wt% in mCP)	-	-	-	[93]
i-DMAc-TRZ	432/53.4/-(PhMe)	452/55/1840(3 wt% in DPEPO)	0.35	-	-	[95]
DPAC-TRZ	-/-/-()	478/82/2.9(12 wt% in mCPCN)	0.13	-5.72	-3.12	[97]
SpiroAC-TRZ	-/-/-()	480/100/2.1(12 wt% in mCPCN)	0.07	-5.70	-3.12	[97]
SpiroAC-TRZ	-/-/-()	483/98/3.9(10 wt% in mCP/TSPO1)	0.07	-	-	[97]
DTPDDA	-/-/-(-)	468/70/5.1(10 wt% in mCP/TSPO1)	0.16	-	-	[100]
DTPDDA	~475/-/(PhMe)	-/-/-	-	-5.57	-2.80	[100]
a-DMAc-TRZ	479/-/(PhMe)	-/86.1/4.09(20 wt% in DPEPO)	-	-	-	[98]
DTPPDDA	450/-/(PhMe)	-/38/20ns (8 wt% in mCP : TSPO1)	-	-6.08	-2.83	[101]
DTPSAF	-/-/(PhMe)	459/82/-(10 wt % immCP)	-	-5.51	-2.63	[96]
DTTSAF	-/-/(PhMe)	450/100/-(10 wt % in mCP)	-	-5.55	-2.56	[96]
DTXSAF	-/-/(PhMe)	446/98/-(10 wt % in mCP)	-	-5.52	-2.48	[96]
MA-TA	469/-/(PhMe)	-/83/18.3(10 wt% in CzSi)	0.14	-5.86	-2.96	[99]
FA-TA	452/-/(PhMe)	-/76/44.4(10 wt% in CzSi)	0.16	-5.91	-2.96	[99]
PA-TA	451/-/(PhMe)	-/70/69.5(10 wt% in CzSi)	0.17	-6.00	-2.99	[99]

**Table 20.** Summary of device structures and performance

Emitter	Device Structure	EL <sub>max</sub> [nm]	CIE	V <sub>on</sub> [V]	EQE/PE/CE <sup>a)</sup> [%/lm W <sup>-1</sup> /cd A <sup>-1</sup> ]	EQE <sub>100/1000</sub> cd m <sup>-2</sup>	Ref
DMAC-TRZ	ITO/PEDOT:PSS/TAPC/mCP/mCPCN:DMAC-TRZ 8 wt%DPPS /3TPYMB/LiF/Al	~500	-	3	26.5/65.6/66.8	25.1/~22.0	[86]
DMAC-TRZ	ITO/MoO <sub>3</sub> /TAPC/mCPCN: 12 wt% DMAC-TRZ/LiF/Al	~505	-	3	27.4/80.8/77.1	25.8/22.4	[86]
PXZ-TRZ	ITO/α-NPD/6 wt% PXZ-TRZ: CBP/TPBi /LiF/Al	529	-	3.5	12.5/-	~-	[86]
PTZ-TRZ	ITO/α-NPD/2 wt% PTZ-TRZ:mCBP/TPBi/LiF/Al	~532	-	-	10.8/-	~-	[90]
3ACR-TRZ	ITO/PEDOT:PSS/16 wt% 3ACR-TRZ:CBP/BmPyPB/Liq/Al	~520	-	4.8	18.6/36.3/-	~-	[91]
tri-PXZ-TRZ	ITO/α-NPD/6 wt%:mCBP/TPBi /LiF/Al	553	-	-	13.3/-	~-	[92]
TRZ3(Ph-PTZ)	ITO /α-NPD/TCTA/mCP/5 wt% TRZ3(Ph-PTZ) : mCBP/DPEPO /TPBi/LiF /Al	~550	(0.23, 0.75)	4.2	17.4/17.4/58.6	~-	[93]
i-DMAc-TRZ	ITO/TAPC/10 wt% i-DMAcTRZ: mCBP/TmPyPB/LiF/Al	450	(0.15, 0.11)	-	10.9/-	~4.0/1.5	[95]
DPAC-TRZ	ITO/MoO <sub>3</sub> /TAPC/mCPCN:12 wt% DPAC-TRZ/LiF/Al	~500	-	-	25.8/62.7/60	24.1/18.3	[97]
SpiroAC-TRZ	ITO/MoO <sub>3</sub> /TAPC/mCPCN: 12 wt% SpiroAC-TRZ/LiF/Al	~500	-	-	36.7/98.4/94	34.9/30.5	[97]
DTPDDA	ITO/4 wt% ReO <sub>3</sub> :mCP/mCP/mCP:TSPO1:16 wt% DPTDDA(0.42:0.42:0.16 in wt%)/TSPO1/4 wt% Rb <sub>2</sub> CO <sub>3</sub> :TSPO1/Al	450	(0.149, 0.197)	3	22.3/30.4/35.6	~15.0/10.6	[100]
a-DMAc-TRZ	ITO/TAPC/ mCP/DPEPO:20 wt% a-DMAc-TRZ/ DPEPO/ TmPyPB)/ LiF/Al	~505	(0.18, 0.35)	3.3	28.9/-/63.7	12.7/3.6	[98]

DTPPDDA	ITO/4 wt% ReO <sub>3</sub> : mCP/mCP/mCP : TSPO1 : emitter/TSPO1/4 wt% Rb <sub>2</sub> CO <sub>3</sub> : TSPO1/Al	~450	(0.151, 0.087)	3.3	4.7/-/	2.2/-	[101]
DTPSAF	ITO/6 wt% ReO <sub>3</sub> :mCP/mCP/10 wt %:mCP/TSPO1/6 wt % Rb <sub>2</sub> CO <sub>3</sub> :TSPO1/Al	460	(0.143, 0.131)	4.1	5.7/6.9/	~3.0/-	[96]
DTTSAF	ITO/6 wt% ReO <sub>3</sub> :mCP/mCP/10 wt %:mCP/TSPO1/6 wt % Rb <sub>2</sub> CO <sub>3</sub> :TSPO1/Al	448	(0.147, 0.096)	4.1	6.2/4.8/	~5.0/-	[96]
DTXSAF	ITO/6 wt% ReO <sub>3</sub> :mCP/mCP/10 wt %:mCP/TSPO1/6 wt % Rb <sub>2</sub> CO <sub>3</sub> :TSPO1/Al	444	(0.149, 0.082)	4.1	7.7/5.2/	~5.0/-	[96]
MA-TA	ITO /PEDOT:PSS/PVK/10 wt% MA-TA:CzSi /TSPO1/TPBi/LiQ/Al	465	(0.15, 0.19)	5.1	22.1/-/	~15.0/-	[99]
FA-TA	ITO /PEDOT:PSS/PVK/10 wt% FA-TA:CzSi /TSPO1/TPBi/LiQ/Al	452	(0.15,0.13)	5.5	11.2/-/	~/-	[99]
PA-TA	ITO /PEDOT:PSS/PVK/10 wt% PA-TA:CzSi /TSPO1/TPBi/LiQ/Al	451	(0.15, 0.10)	5.5	6.7/-/	~/-	[99]

## Conclusions

The objective of this review is to provide a detailed overview of triazine-based TADF materials, to compare and contrast their optoelectronic properties and to report their performance as emitters in OLEDs. The emission properties of these TADF emitters were mainly modulated via the tuning of the donor strength, modifying the substituents about the triazine, varying the nature of the D-A bridge and switching CT channels (intramolecular CT vs. through-space CT). The plethora of examples contained within this review reveals the versatility of triazine as an acceptor in the design of TADF emitters. By simply varying the strength and number of donors along with the structure of the bridging aryl groups, the emission spectrum can be easily tuned from deep blue through to yellow. The planar conformation adopted by the TRZ moiety can facilitate to the formation of intermolecular exciplexes or through space charge transfer interactions, which enriches the photophysical behavior of these compounds and can contribute to enhancing  $k_{RISC}$ .

Despite the progress that has been made, the potential of triazine as a moiety in TADF emitter design still has not been fully realized. Firstly, one of the major factors affecting the device lifetime of OLEDs is the stability of the emitter. TADF compounds containing a TRZ acceptor have already demonstrated some potential for improved device lifetimes over their analogues containing acceptors like benzophenone or diphenyl sulfone. TRZ has long been recognized as a chemically stable moiety in the design of epoxy resin,<sup>[102]</sup> polymers,<sup>[103]</sup> and covalent TRZ frameworks (CTFs);<sup>[104]</sup> however, OLED stability studies based on TRZ-functionalized TADF emitters remain limited. Studies that probed the influence of intramolecular hydrogen bonding, glass transition temperatures and charge mobility on the stability of TRZ-functionalized molecules and the impact on the device lifetime would be welcome.

The orientation of the TDM of the emitters, which impacts the light out-coupling efficiency of the device, is correlated with the EQE of the OLED. Due to the rigid and planar structure of TRZ linked to extended donors, a number of TRZ-based compounds have been documented to show preferential horizontal orientation of their TDM, leading to enhanced light outcoupling efficiency in the device, with  $\text{EQE}_{\max} > 30\%$ . The parameters controlling the orientation of the TDM during thermal evaporation remain unclear<sup>[105]</sup> and this is clearly a design feature that can be exploited further.

Solution-processing techniques such as ink-jet printing are promising for producing large-area OLEDs, which remains challenging and expensive for thermal evaporation. More and more attention has been paid to the development of TRZ-containing TADF dendrimers and polymers as attractive classes of emitters suitable for solution-processed OLEDs. However, in contrast to the advances reported for small molecular weight TADF emitters that show high PLQY, horizontal orientation of their TDM, fast RISC rates, and examples of chiral analogs that show CPL, it remains challenging to develop TADF dendrimers and polymers inheriting these properties. Thus, the performance of solution-processed OLEDs containing TRZ-based macromolecules still lags behind small molecule TRZ-based evaporated OLEDs.

### Acknowledgements

D.S acknowledges support from the Royal Academy of Engineering Enterprise Fellowship (EF2122-13106). This project has received funding from the European Union's Horizon 2020 research and innovation program under the Marie Skłodowska-Curie grant agreements No. 897098 (AIE-RTP-PLED). T.W. acknowledges support from the Marie Skłodowska-Curie Individual Fellowship. C.S. acknowledges support from the China Scholarship Council (201806890001). We thank the EPSRC (EP/P010482/1) for financial support.

Received: ((will be filled in by the editorial staff))

Revised: ((will be filled in by the editorial staff))

Published online: ((will be filled in by the editorial staff))

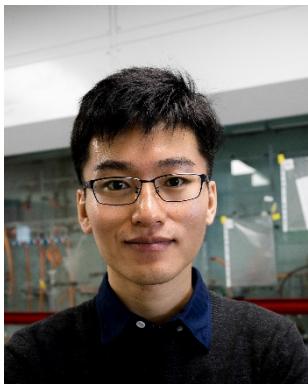
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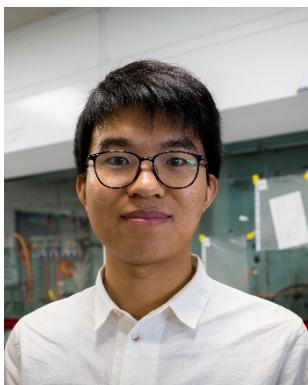
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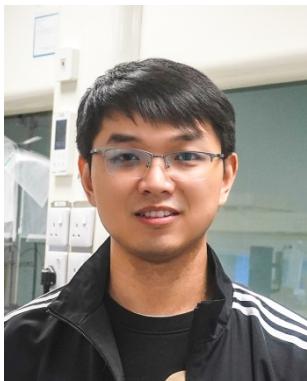
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Leverhulme Trust Senior Research Fellowship. His research program focuses on the rational design of: (I) luminophores for energy-efficient visual displays and flat panel lighting based on organic light emitting diode (OLED) and light-emitting electrochemical cell (LEEC) device architectures; (II) sensing materials employed in electrochemiluminescence; and (III) photocatalyst developing for use in organic reactions.

1,3,5-Triazine is a versatile building block in TADF emitter design for high-performance OLEDs. In this review, we document the diversity structures of triazine-based emitters, correlate these to their photophysical properties and the performance of the corresponding OLEDs.

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## 1,3,5-Triazine-Functionalized Thermally Activated Delayed Fluorescence Emitters for Organic Light-Emitting Diodes

