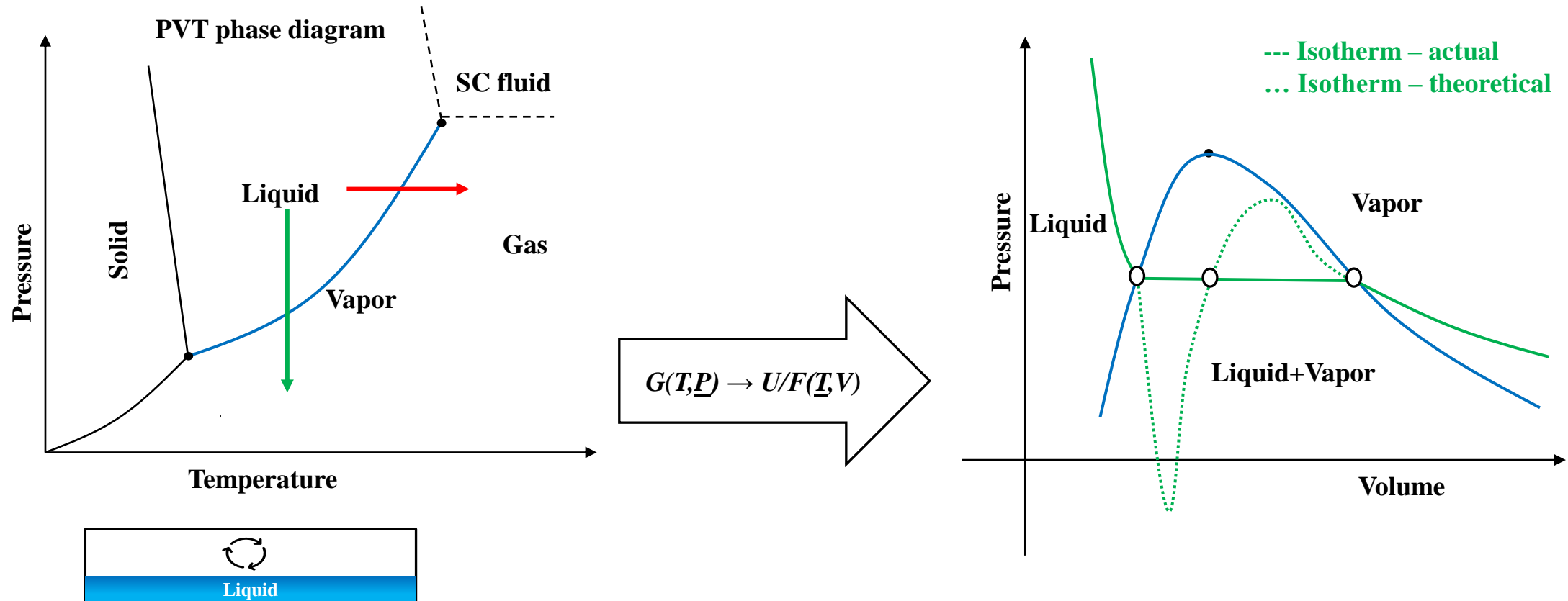


Cavitation in Nanoconfinements

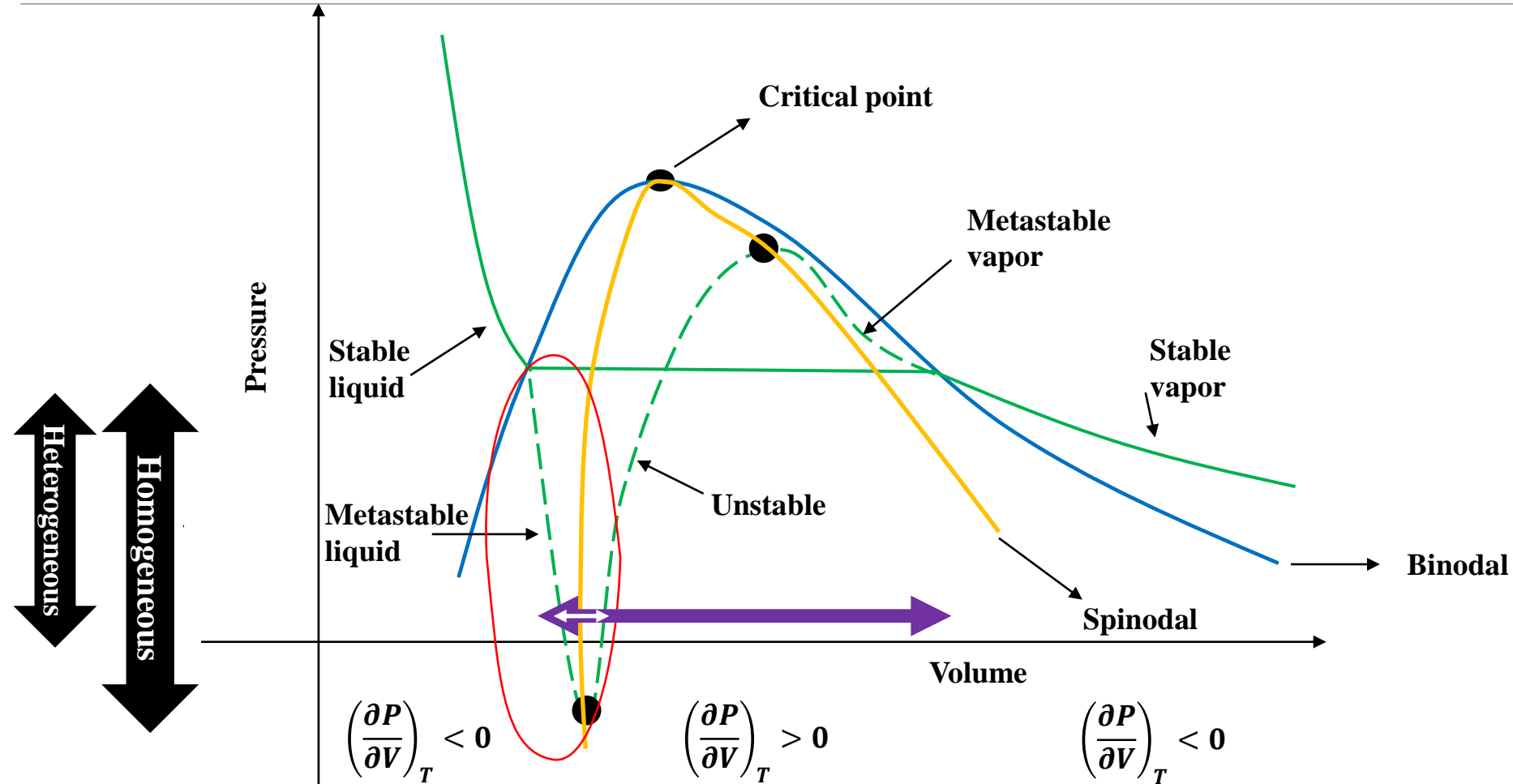
Vapor pressure, boiling and cavitation

Definition



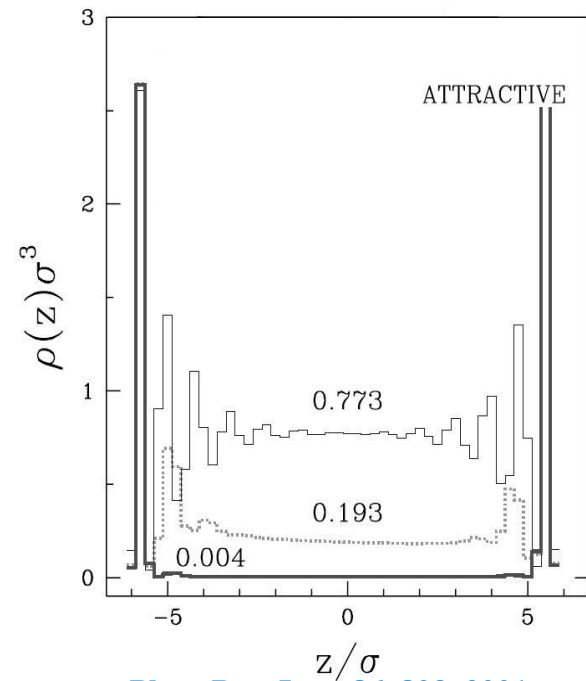
$P/T-V$ diagram

Inclusive of confinement

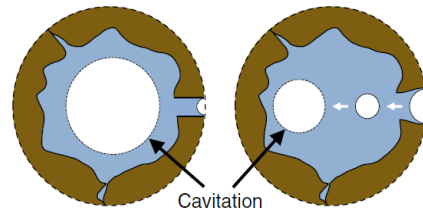


Cavitation

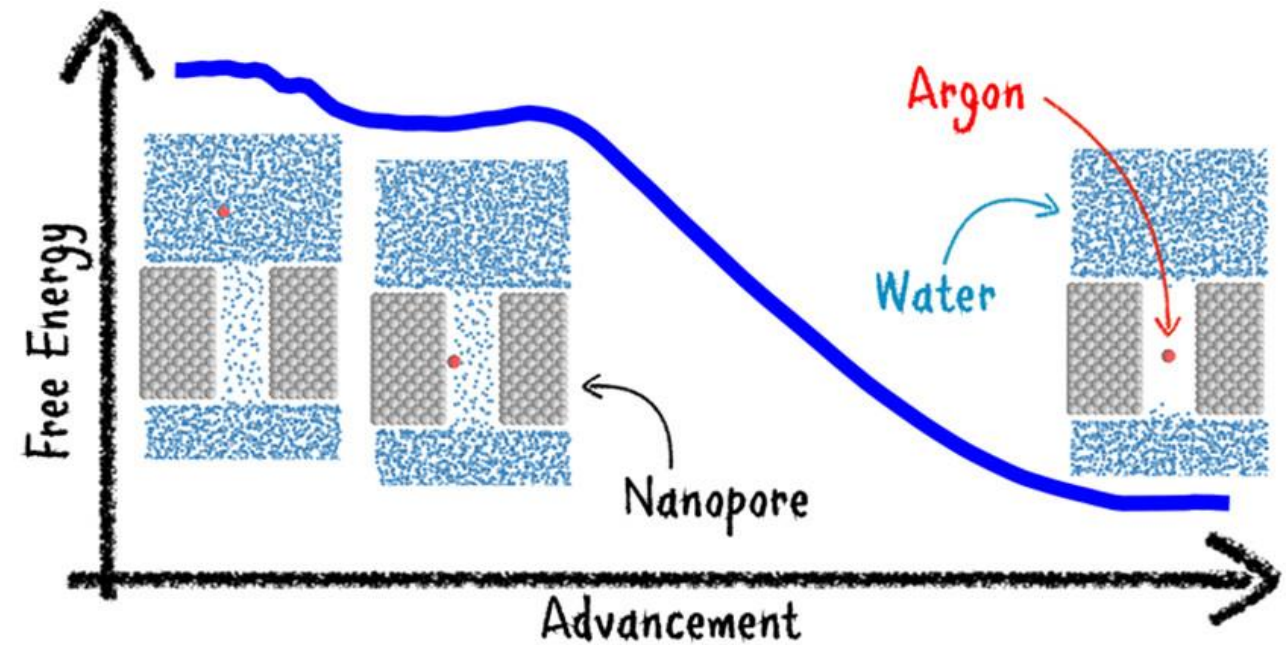
Examples



[Phys. Rev. Lett. 86, 803, 2001](#)



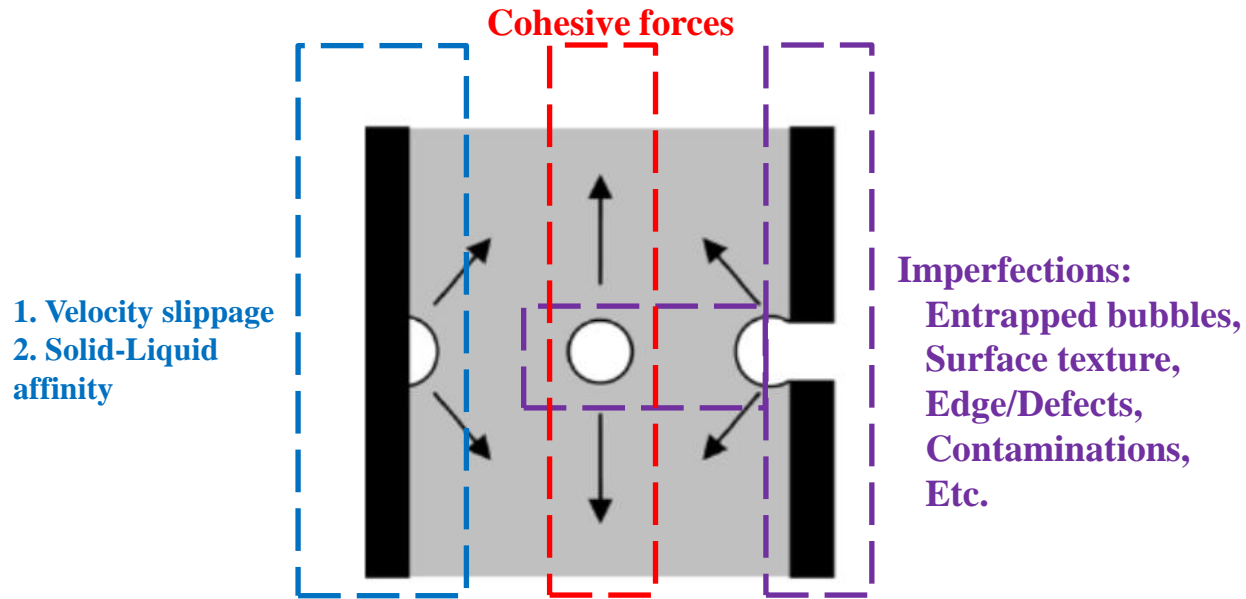
[J. Geotech. Geoenviron. Eng., 2021, 147\(8\): 04021079](#)



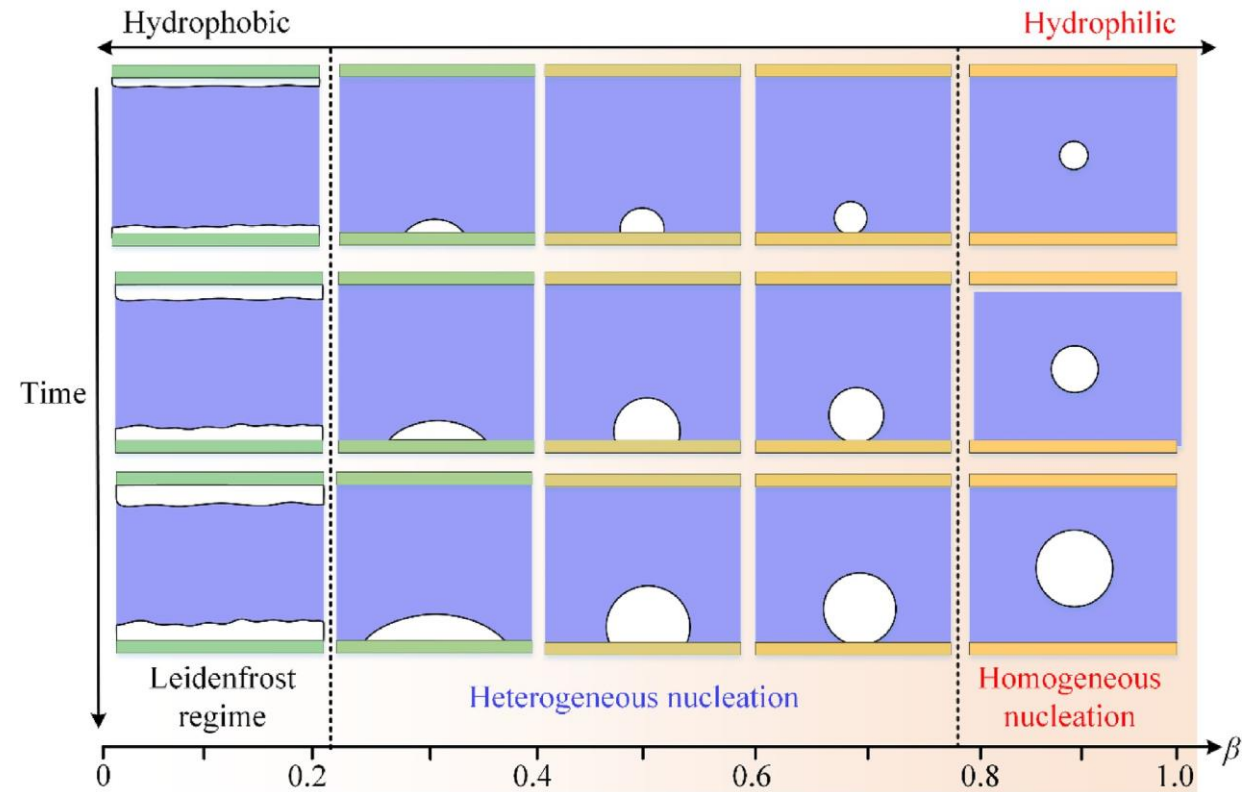
[J. Phys. Chem. Lett. 2020, 11, 21, 9171–9177](#)

Cavitation

Recapping what's known



[C. R. Physique 7 \(2006\) 1018–1026](#)



[International Journal of Thermal Sciences 145 \(2019\) 106033](#)

Experimental Specs

Observations and simulation motives

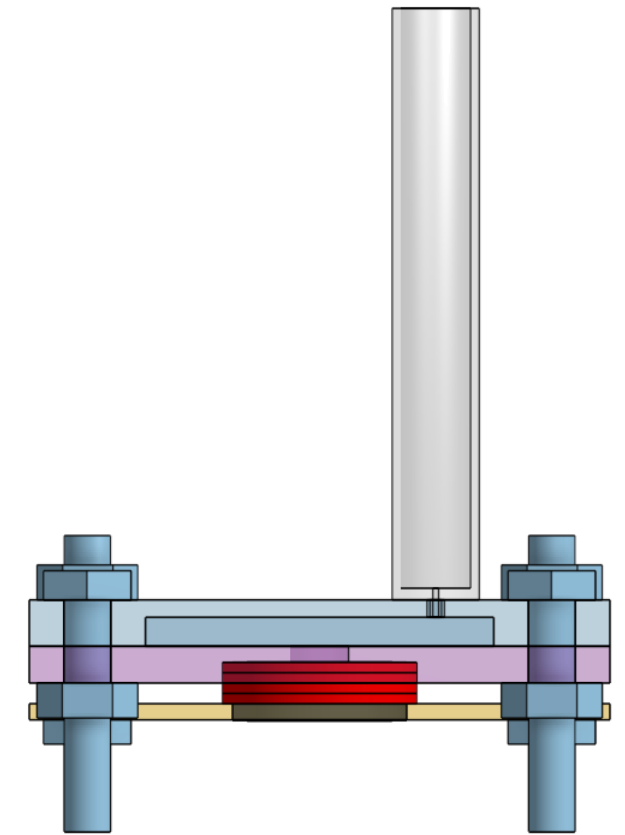
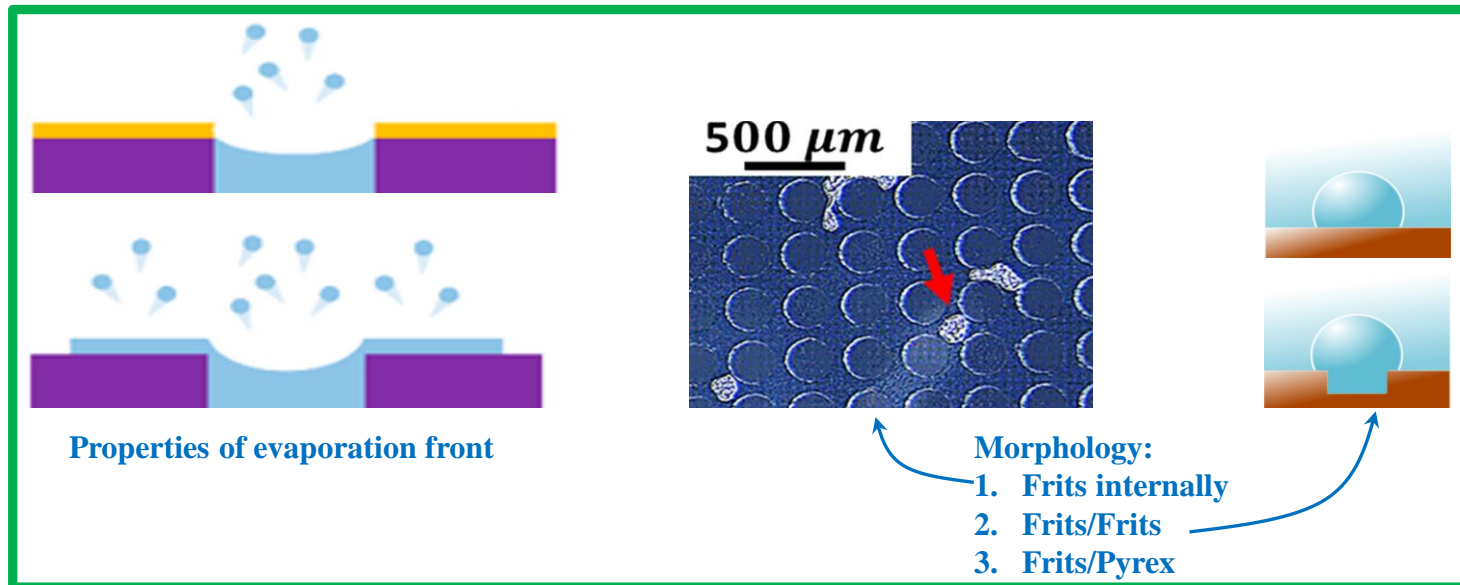
Silica frits pore diameter: **1 – 10 microns**

Silica frits thickness: **3.3 – 4 millimeter**

Hydrophilic internal surface (Silica frits and glass Pyrex)

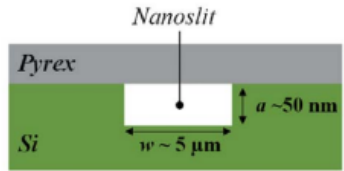
Pressure dropdown / flow induced via **continuous evaporation**

Interruptions in flow / evaporation rate / bubbles



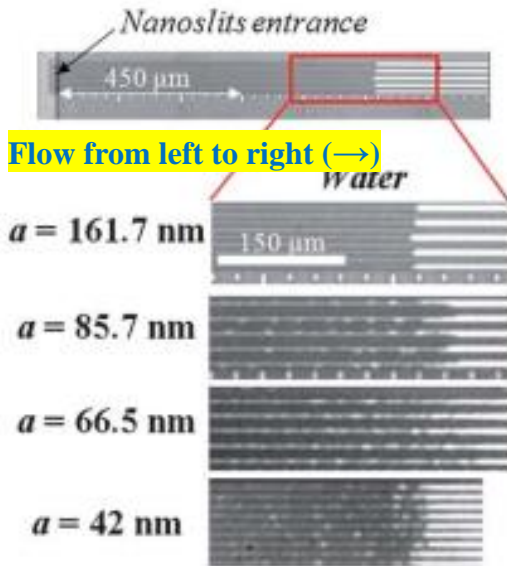
Flow Interruption

Where confinement loss effect



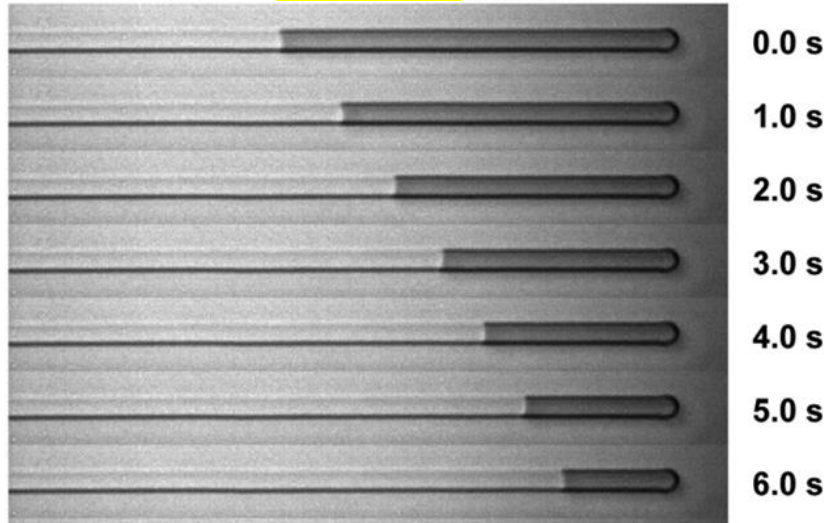
Chemically Homogeneous Surface
(Uniform Solid-Liquid Affinity)

Silicon glass = hydrophilic



[Soft Matter, 2012,8, 10738-10749](#)

Silicon nanochannels: $L=1000\mu\text{m}$, $W=3\mu\text{m}$, $H > 115\text{ nm}$

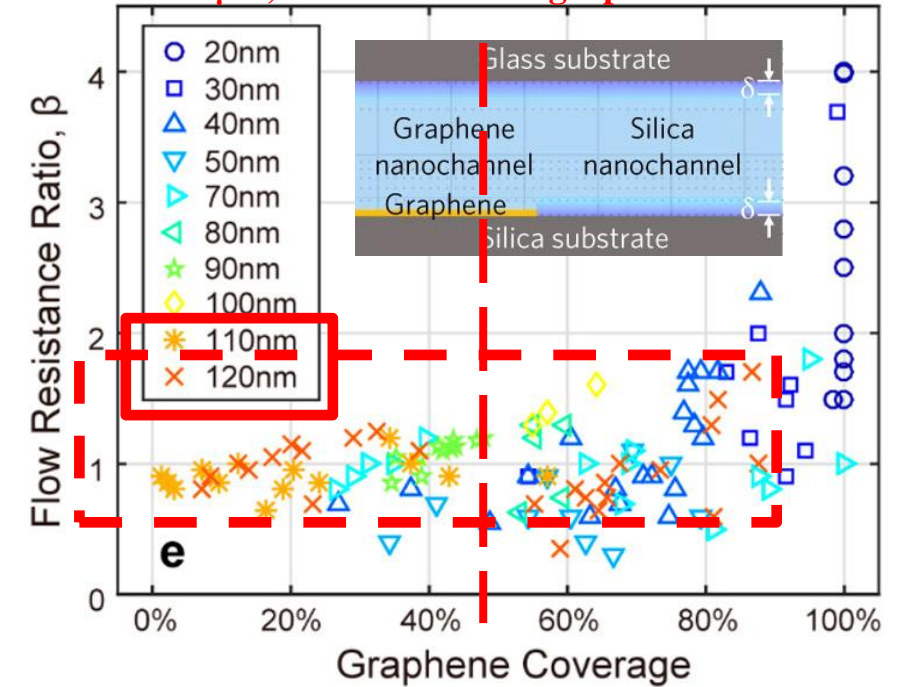


Flow from right to left (\leftarrow)

[Langmuir 2017, 33, 8395–8403](#)

Chemically Heterogeneous Surface
(Nonuniform Solid-Liquid Affinity)

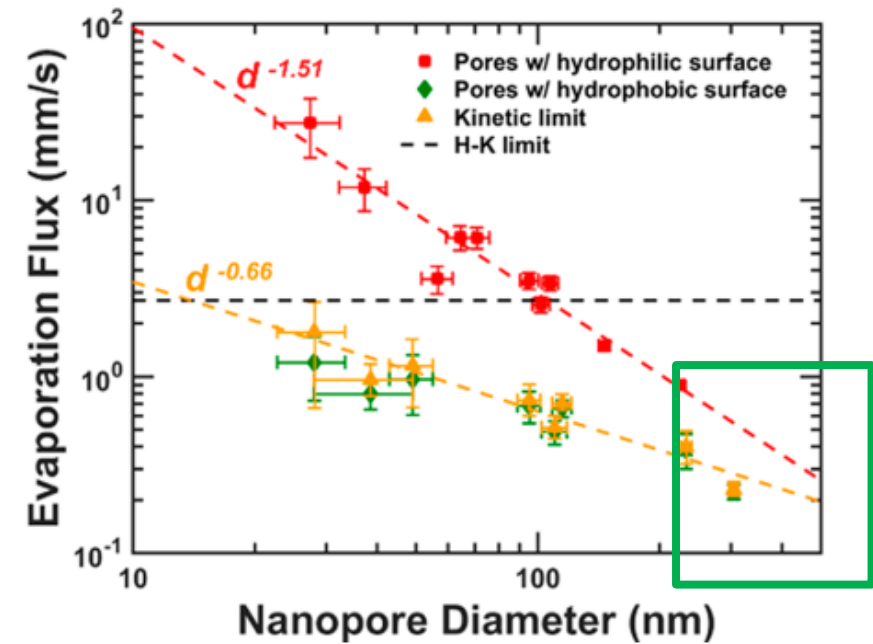
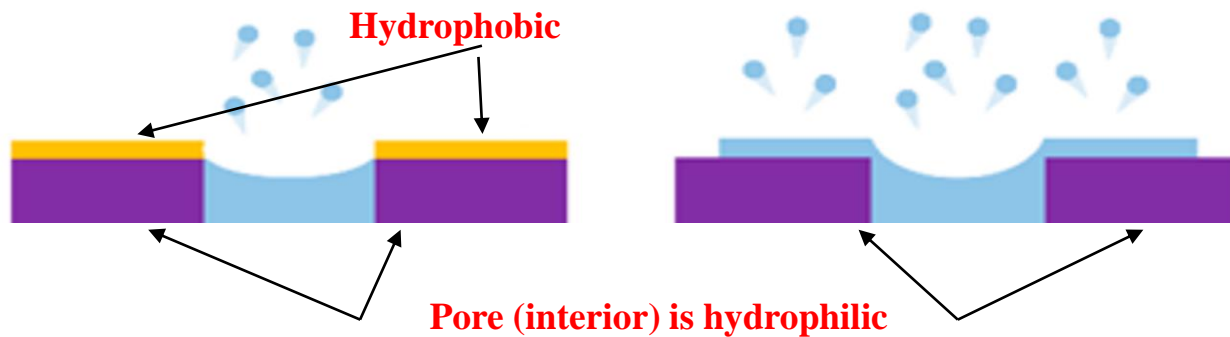
As in left; leftmost bottom% is graphene
 $L=400\mu\text{m}$, $W=H=$ labels on graph



[Nature Nanotechnology, 2018, 13, 238–245](#)

Chemical Heterogeneity

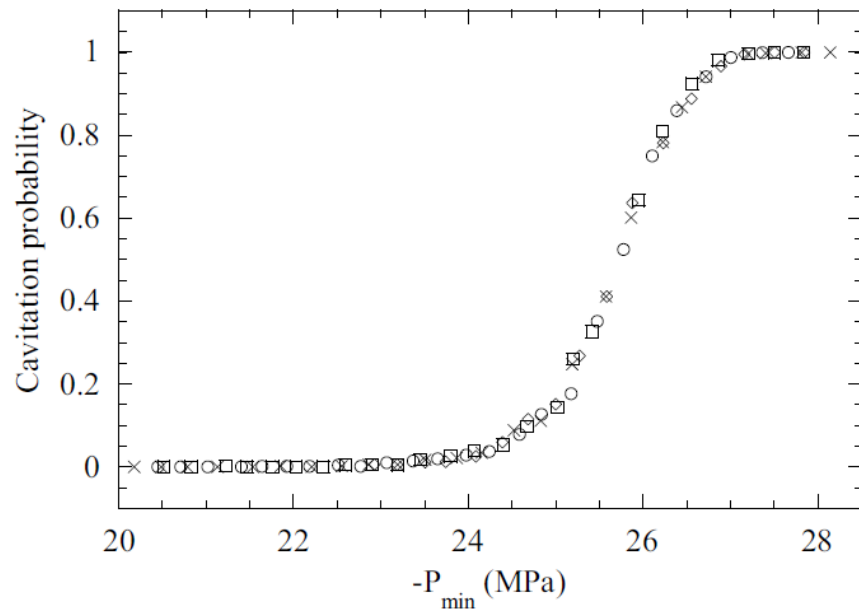
Evaporation Zone



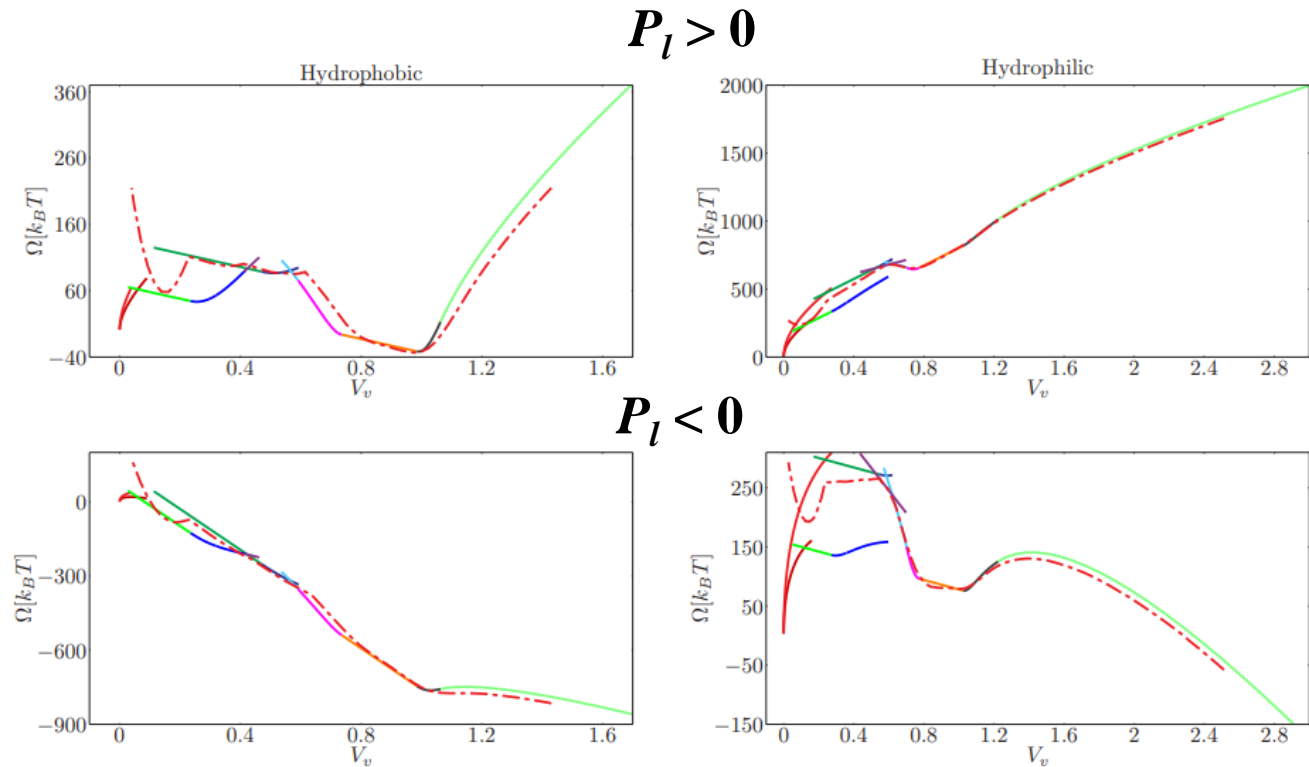
[ACS Nano 2019, 13, 3363–3372](#)

Chemical/Physical Heterogeneity

Liquid response



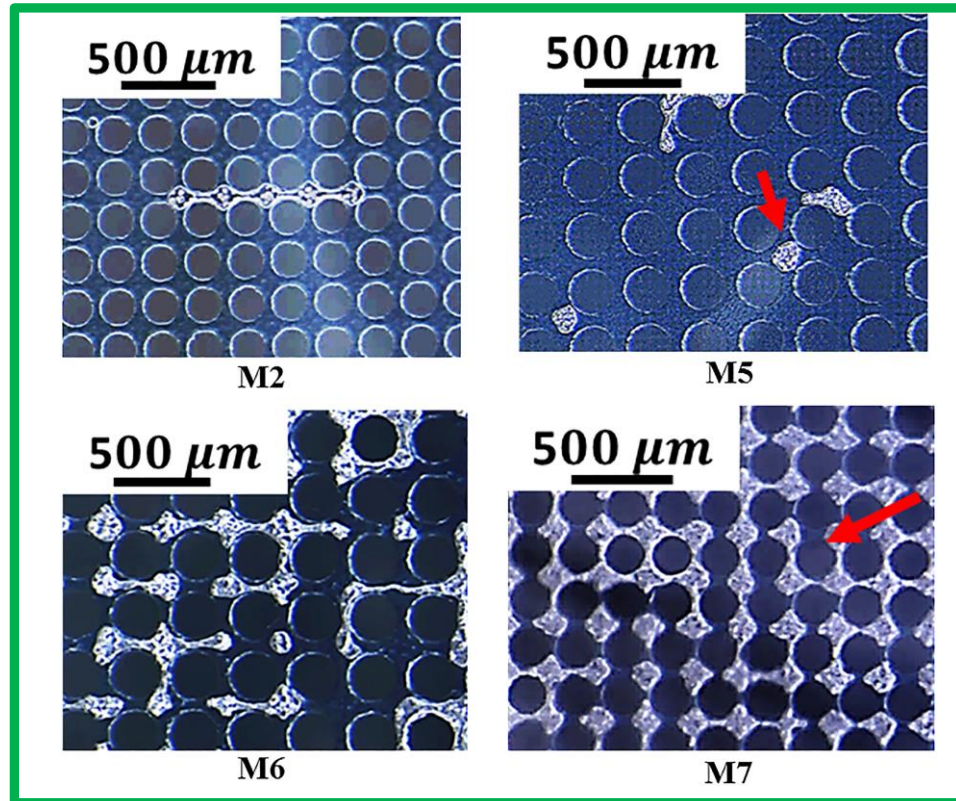
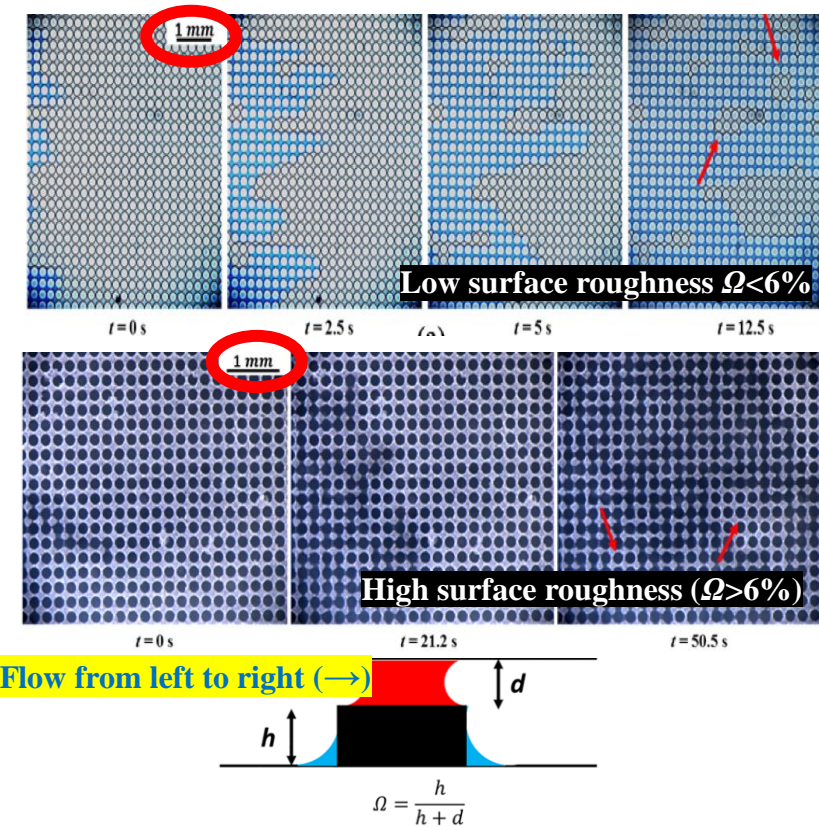
[J. Chem. Phys. 133, 134505 \(2010\)](#)



[Soft Matter, 2016,12, 3046-3055](#)

Chemical/Physical Heterogeneity

Morphology/Microstructure



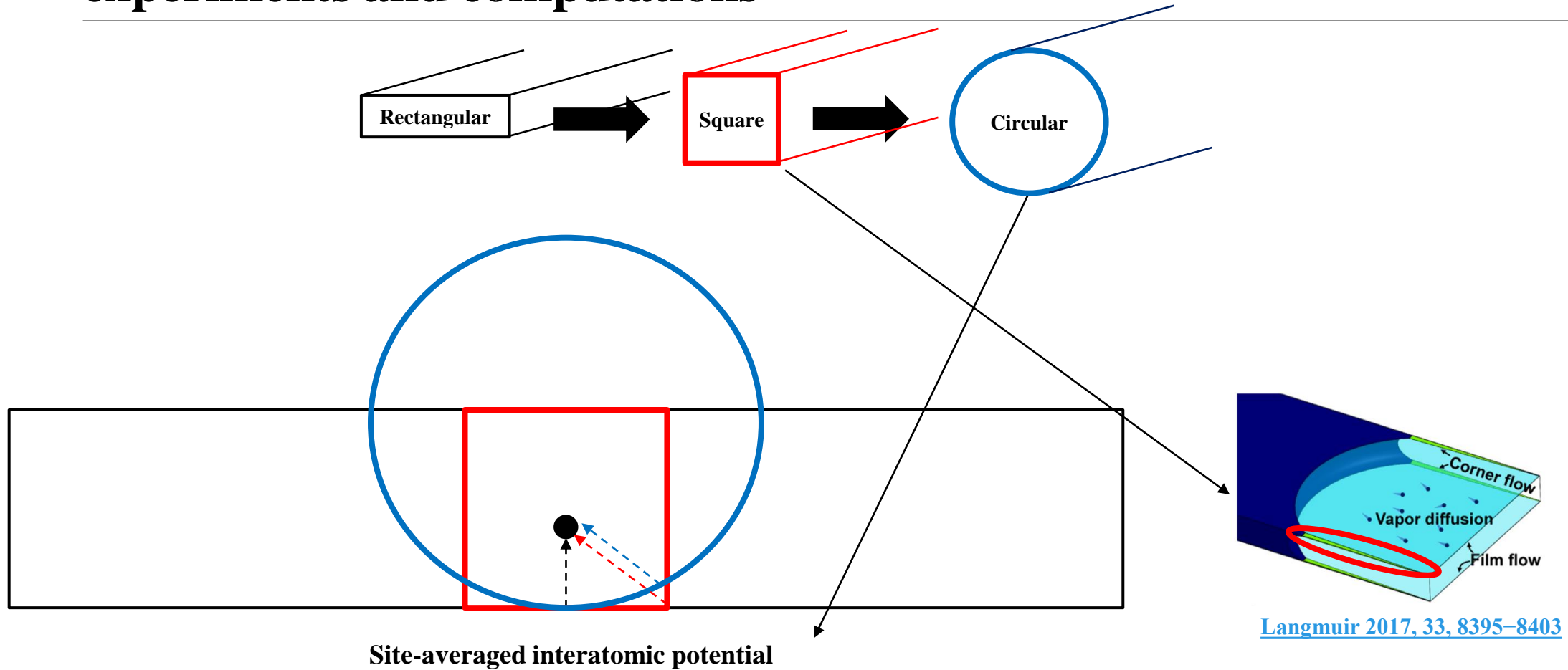
Hysteresis type	Isotherm	Configuration of the pore
H1		
H5		
H6		

[Water Resources Research, 55, 9905–9925, 2019](#)

[Langmuir 2011, 27, 7, 3511–3526](#)

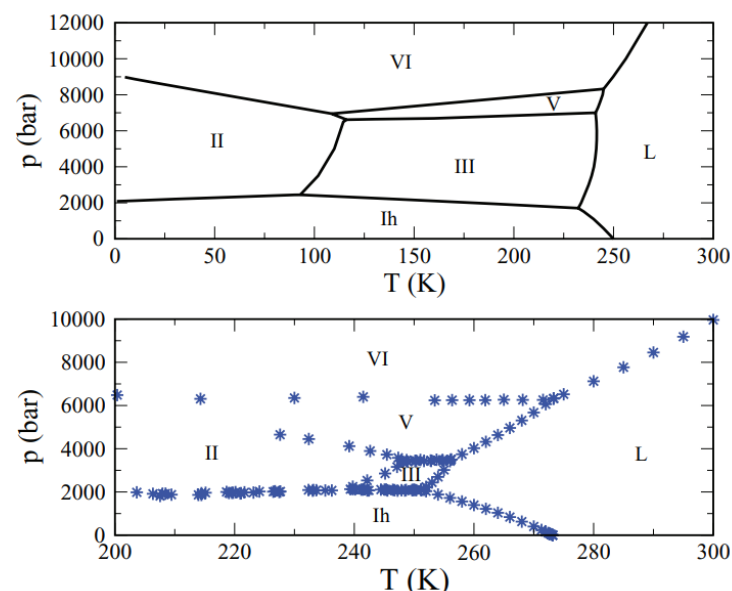
Pore Geometry Preferred

experiments and computations

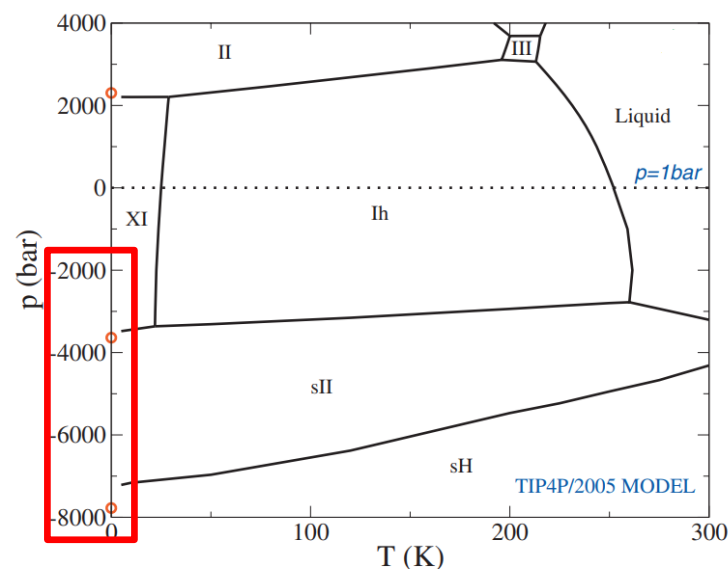


Water model

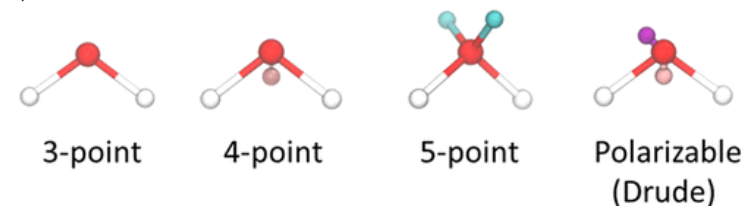
The [TIP4P/2005](#) ([NIST](#)) model (a LJ site for the O, and three charge sites).



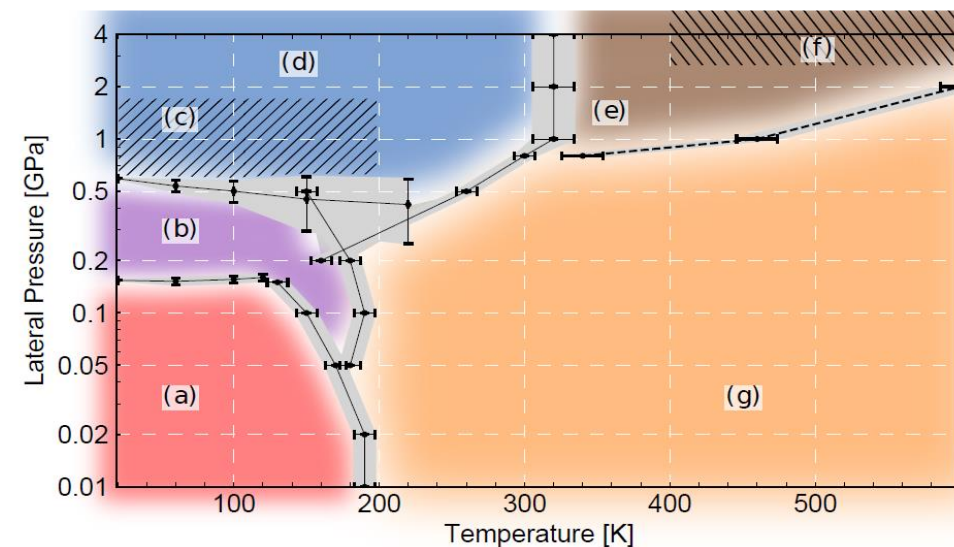
[J. Chem. Phys. 139, 154505 \(2013\).](#)



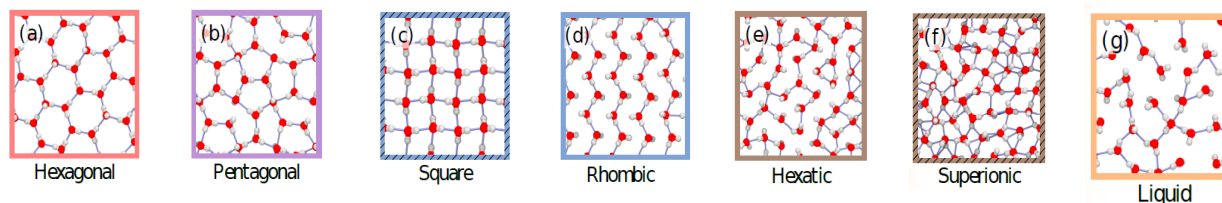
[J. Chem. Phys. 131, 034510 \(2009\).](#)



[J. Chem. Inf. Model. 2021, 61, 9, 4521–4536](#)



[arXiv:2110.14569v1 \[cond-mat.mtrl-sci\]](#)

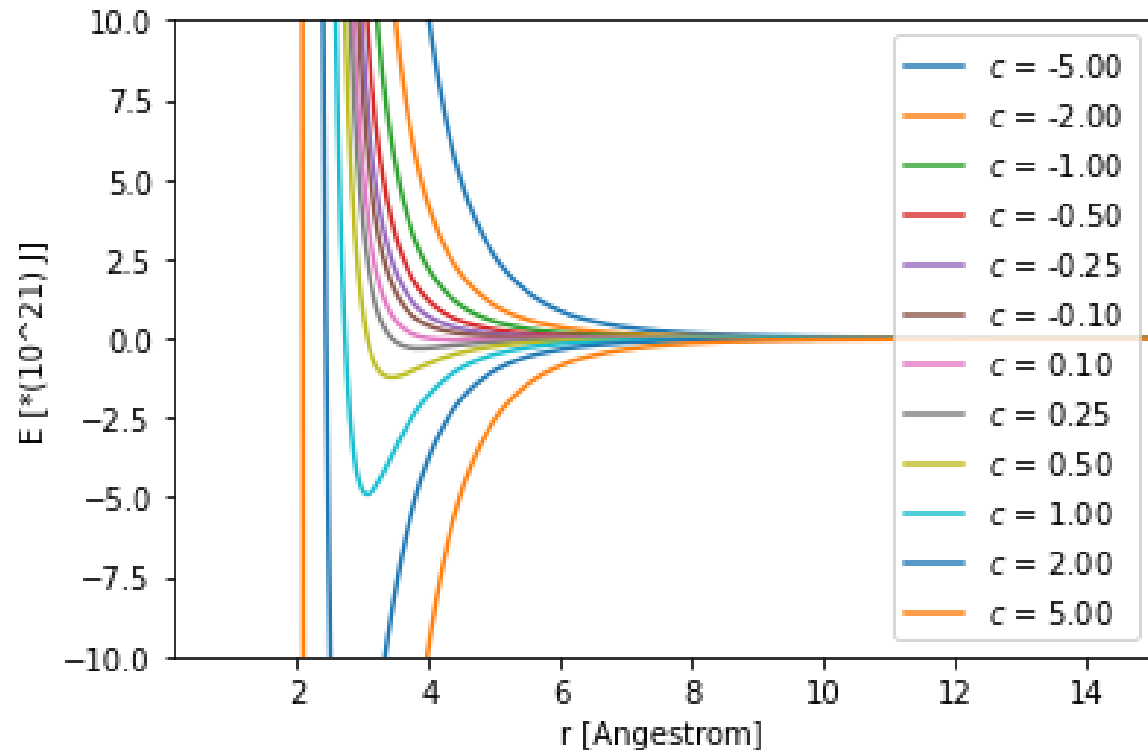
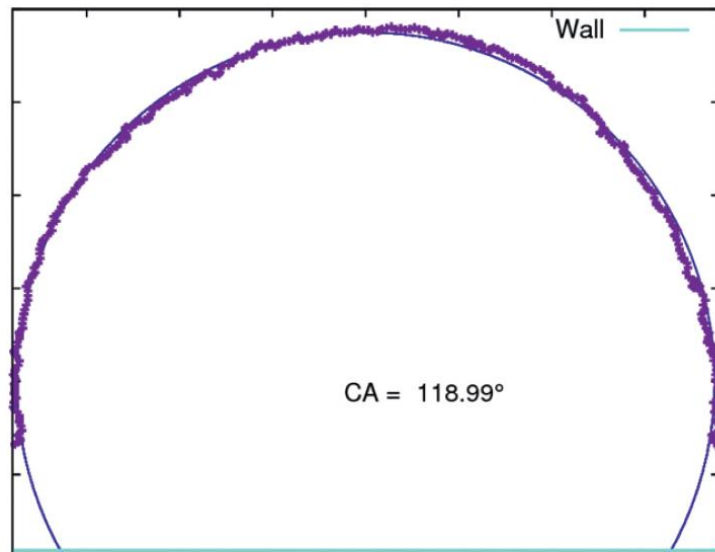


Surface model

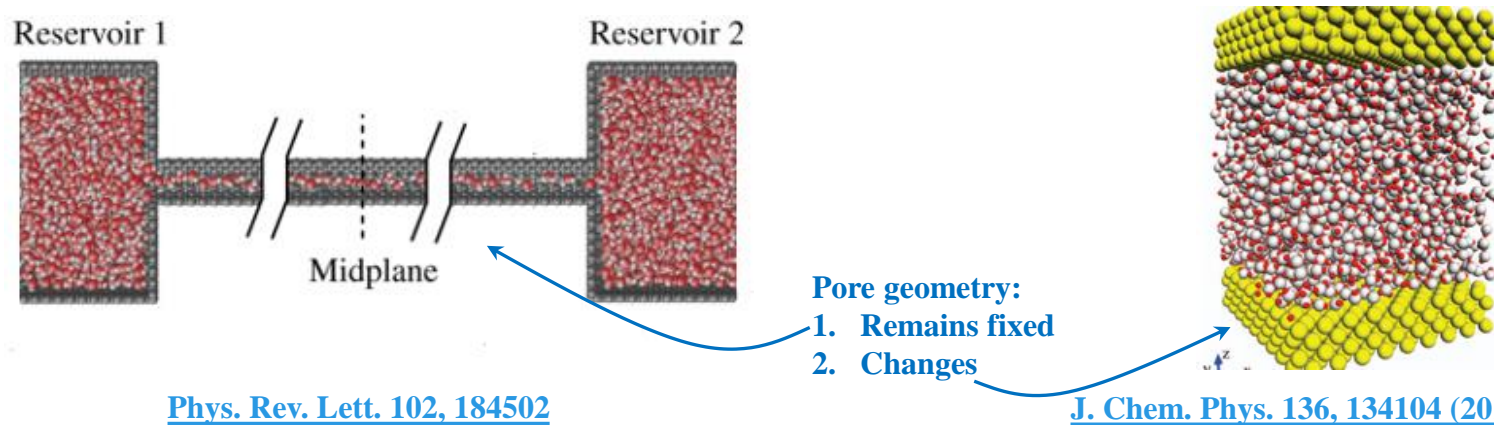
The [TIP4P/2005](#) model in (hydrophobic/hydrophilic) pore / nano-confinement

$$V(r_{ij}) = 4\epsilon \left[\left(\frac{\sigma}{r_{ij}} \right)^{12} - c \left(\frac{\sigma}{r_{ij}} \right)^6 \right]$$

[PNAS November 28, 2017 114 \(48\) E10266-E10273](#)



Inducing pressure dropdown/flow



Creating a pressure drop over pore:

1. Removing (directing to the other) particles (N) from one reservoir
2. Applying a piston on left/rightmost side of reservoirs
3. Fixing P via NPT and letting chambers V change (pseudo piston)

Creating a pressure drop:

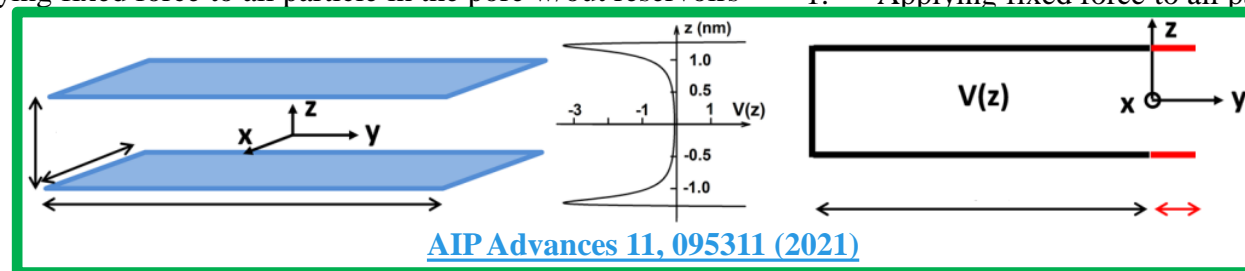
1. Moving both or one of the walls vertically
2. Removing particles (N) from one side
3. Adding an acceptable long empty part for evaporation and vapor diffusion

Creating flow (in addition to above):

1. Applying fixed force to all particle in the pore w/out reservoirs

Creating flow (in addition to above):

1. Applying fixed force to all particle



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