

The Brigade School Unit Test (2020-21)

Total points 12/15 ?

Class : 10

Subject : Chemistry

Paper 1: Objective

Marks :15

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0 of 0 points

Name : *

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10 A ▼

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TBSG ▼

I. Choose the Correct Answer :(10 X 1= 10)

12 of 15 points



✓ 1. If the molecular formula of an organic compound is $C_{10}H_{18}$ it is : * 1/1

- ☐ Alkene
- ☐ Alkane
- ☒ Alkyne
- ☐ Not a hydrocarbon



Feedback

Alkyne general formula C_nH_{2n-2}
 $C_{10}H_{(2 \times 10) - 2}$
 $C_{10}H_{(20 - 2)}$
 $C_{10}H_{18}$

✓ 2. The number of electron/ electrons present in the valence shell of a halogen is * 1/1

- ☐ 1
- ☐ 3
- ☐ 5
- ☒ 7



Feedback

As they are present in 17th group of P.T.



✗ 3. Among the elements of period 2, the element which has high electron affinity is : * 0/1

- ☐ Lithium
- ☐ Carbon
- ☒ Chlorine
- ☐ Fluorine

✗

Correct answer

- ☒ Fluorine

✓ 4. The functional group present in acetic acid is : * 1/1

- ☐ Ketonic
- ☐ Hydroxyl
- ☐ Aldehydic
- ☒ Carboxylic acid

✓

Feedback

Good



✗ 5. An alkali metal which burns with golden yellow flame : *

0/1

- ☐ Sodium
- ☐ Potassium
- ☒ Magnesium
- ☐ Calcium

✗

Correct answer

- ☒ Sodium

✓ 6. The IUPAC name of CH_3CHO is *

1/1

- ☐ Methanal
- ☒ Ethanal
- ☐ Ethanol
- ☐ Acetaldehyde

✓

Feedback

Good



✓ 7. The organic compound prepared by the reaction $\text{C}_2\text{H}_5\text{Br} + \text{KOH}$ (alcoholic soln.) \rightarrow * 1/1

- ☒ Ethene
- ☐ Ethane
- ☐ Ethyne
- ☐ Methane



Feedback

Good

✓ 8. A reaction in which hydrogen of an alkane is replaced by another element like chlorine. * 1/1

- ☐ Addition reaction
- ☒ Substitution reaction
- ☐ Pyrolysis
- ☐ Decarboxylation



Feedback

Good



✗ 9. $C_2H_2 + H_2 \rightarrow A + B$. Identify A and B the unsaturated and saturated products formed in the above reaction are * 0/1

- ☒ Methane and ethene
- ☐ Propene and methane
- ☐ Ethene and ethane
- ☐ Ethyne and ethane

✗

Correct answer

- ☒ Ethene and ethane

✓ 10. Tendency of an atom to attract electrons to its self when combined in a compound * 1/1

- ☐ Electron affinity
- ☒ Electronegativity
- ☐ Ionisation potential
- ☐ Ionisation energy

✓

Feedback

Good



11. Match the following : (5 X 1 =5) *

	Tendency to lose electrons	Catenation	C_nH_{2n-2}	C_nH_{2n}	Tendency to gain electrons	Score	
Self linkage property	<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	1/1	✓
Alkenes	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	1/1	✓
Metallic character	<input checked="" type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	1/1	✓
Non- metallic character	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input checked="" type="radio"/>	1/1	✓
Alkynes	<input type="radio"/>	<input type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input type="radio"/>	1/1	✓

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