# Chapter 25

## Alcohols, Phenols and Ethers

1. The major product obtained on interaction of phenol with sodium hydroxide and carbon dioxide is:

#### [AIEEE-2009]

- (1) Salicylaldehyde
- (2) Salicylic acid
- (3) Phthalic acid
- (4) Benzoic acid
- From amongst the following alcohols the one that would react fastest with conc. HCl and anhydrous ZnCl<sub>2</sub>, is [AIEEE-2010]
  - (1) 1-Butanol
- (2) 2-Butanol
- (3) 2-Methylpropan-2-ol (4) 2-Methylpropanol
- The correct order of acid strength of the following compounds is [AIEEE-2011]
  - A. Phenol
  - B. p-Cresol
  - C. m-Nitrophenol
  - D. p-Nitrophenol
  - (1) A > B > D > C
- (2) C > B > A > D
- (3) D > C > A > B
- (4) B > D > A > C
- Consider the following reaction

$$C_2H_5OH + H_2SO_4 \rightarrow Product$$

Among the following, which one cannot be formed as a product under any conditions? [AIEEE-2011]

- (1) Diethyl ether
- (2) Ethyl-hydrogen sulphate
- (3) Ethylene
- (4) Acetylene
- Arrange the following compounds in order of decreasing acidity [JEE (Main)-2013]

- $(1) | | > | \lor > | > | | |$
- (2) | > | > | | > | | > | | | > | | | | |
- (3) ||| > | > || > |V
- (4) |V > || > | > ||

An unknown alcohol is treated with the "Lucas reagent" to determine whether the alcohol is primary, secondary or tertiary. Which alcohol reacts fastest and by what mechanism?

[JEE (Main)-2013]

- (1) Secondary alcohol by S<sub>N</sub>1
- (2) Tertiary alcohol by S<sub>N</sub>1
- (3) Secondary alcohol by S<sub>N</sub>2
- (4) Tertiary alcohol by S<sub>N</sub>2
- Sodium phenoxide when heated with CO2 under pressure at 125°C yields a product which on acetylation produces C.

ONa + 
$$CO_2 \xrightarrow{125^{\circ}} B \xrightarrow{H^{+}} Ac_2O C$$

The major product C would be [JEE (Main)-2014]

8. The correct sequence of reagents for the following conversion will be [JEE (Main)-2017]

- (1) CH<sub>3</sub>MgBr, [Ag(NH<sub>3</sub>)<sub>2</sub>]+OH-, H+/CH<sub>3</sub>OH
- (2) [Ag(NH<sub>3</sub>)<sub>2</sub>]<sup>+</sup>OH<sup>-</sup>, CH<sub>3</sub>MgBr, H<sup>+</sup>/CH<sub>3</sub>OH
- (3)  $[Ag(NH_3)_2]^+OH^-$ ,  $H^+/CH_3OH$ ,  $CH_3MgBr$
- (4) CH<sub>3</sub>MgBr, H<sup>+</sup>/CH<sub>3</sub>OH, [Ag(NH<sub>3</sub>)<sub>2</sub>]<sup>+</sup>OH<sup>-</sup>

Phenol on treatment with CO<sub>2</sub> in the presence of NaOH followed by acidification produces compound X as the major product. X on treatment with (CH<sub>3</sub>CO)<sub>2</sub>O in the presence of catalytic amount of H<sub>2</sub>SO<sub>4</sub> produces [JEE (Main)-2018]

(1) 
$$CH_3$$
 (2)  $CH_3$  (2)  $CO_2H$  (3)  $CO_2H$  (4)  $CO_2H$   $CO_2H$   $CO_2H$   $CO_2H$ 

The major product formed in the following reaction is

[JEE (Main)-2018]

11. The major product of the following reaction is

#### [JEE (Main)-2019]

Br 
$$\xrightarrow{(1) \text{ KOH (aqueous)}}$$
 $(2) \text{ CrO}_3/\text{H}^+$ 
 $(3) \text{ H}_2\text{SO}_4/\Delta$ 

(1)
Br  $(2)$ 
 $(3)$ 
 $(4)$ 
 $(4)$ 

 The products formed in the reaction of cumene with O<sub>2</sub> followed by treatment with dil. HCl are

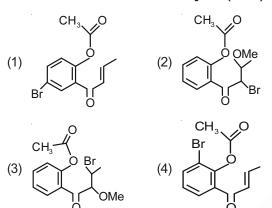
[JEE (Main)-2019]

13. Which is the most suitable reagent for the following transformation? [JEE (Main)-2019]

- (1) I<sub>2</sub>/NaOH
- (2) Alkaline KMnO<sub>4</sub>
- (3) Tollen's reagent
- (4) CrO<sub>2</sub>Cl<sub>2</sub>/CS<sub>2</sub>
- 14. The major product of the following reaction is

The major product obtained in the following conversion is

[JEE (Main)-2019]



The major product obtained in the following reaction is

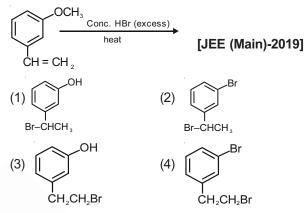
[JEE (Main)-2019]

(1) 
$$CH_3$$
 (2)  $CH_3$   $CH_3$  (3)  $CH_3$   $CH$ 

17. CH<sub>3</sub>CH<sub>2</sub> — C — CH<sub>3</sub> cannot be prepared by | Ph

[JEE (Main)-2019]

- (1) PhCOCH<sub>2</sub>CH<sub>3</sub> + CH<sub>3</sub>MgX
- (2) CH<sub>3</sub>CH<sub>2</sub>COCH<sub>3</sub>+ PhMgX
- (3) HCHO+PhCH(CH<sub>3</sub>)CH<sub>2</sub>MgX
- (4) PhCOCH<sub>3</sub>+ CH<sub>3</sub>CH<sub>2</sub>MgX
- 18. The major product of the following reaction is



 The organic compound that gives following qualitative analysis is

Inference

Insoluble

Test

(a) Dil. HCI

(b) NaOH solution Soluble
(c) Br<sub>2</sub>/water Decolourization

[JEE (Main)-2019]

(1) OH
(2) OH
(3) NH<sub>2</sub>
(4) NH<sub>2</sub>

20. p-Hydroxybenzophenone upon reaction with bromine in carbon tetrachloride gives

[JEE (Main)-2019]

21. The major product of the following reaction is :

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[JEE (Main)-2019]

22. The major products of the following reaction are:

[JEE (Main)-2019]

23. What will be the major product when m-cresol is reacted with propargyl bromide (HC  $\equiv$  C-CH<sub>2</sub>Br) in presence of K<sub>2</sub>CO<sub>3</sub> in acetone?

[JEE (Main)-2019]

24. Consider the following reactions:

$$A \xrightarrow{Ag_2O} ppt$$

$$A \xrightarrow{B} \xrightarrow{ABH_4} C \xrightarrow{ZnCl_2} \xrightarrow{Turbidity} within 5 minutes$$

'A' is

[JEE (Main)-2019]

(1) 
$$CH_3 - C \equiv CH$$

(2) 
$$CH_3 - C \equiv C - CH_3$$

(3) 
$$CH_2 = CH_2$$

25. 1-methylethylene oxide when treated with an excess of HBr produces: [JEE (Main)-2020]

$$(1) = \begin{matrix} Br \\ CH_3 \end{matrix} \qquad (2)$$

$$(2) \longrightarrow_{Br}^{Br}$$

$$(3) \quad \text{Br} \quad \text{CH}_3 \qquad \qquad (4) \quad \text{B}$$

26. In the following reaction sequence, structures of A and B, respectively will be

(1) 
$$CH_2Br$$
  $CH_2Br$ 

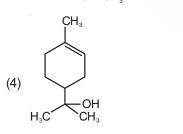
(2) 
$$\begin{array}{c} Br \\ OH \\ CH_2Br \end{array}$$

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27. The major product of the following reaction is

[JEE (Main)-2020]



28. Among the compounds A and B with molecular formula C<sub>9</sub>H<sub>18</sub>O<sub>3</sub>, A is having higher boiling point the B. The possible structures of A and B are

[JEE (Main)-2020]

29. The major product [B] in the following sequence of reactions is

$$CH_3 - C = CH - CH_2CH_3$$

$$CH(CH_3)_2$$

$$(i) B_2H_6$$

$$(ii) H_2O_2, OH$$

$$(ii) H_2O_2 + OH$$

$$(iii) H_2O_2 + OH$$

$$(iii) H_2O_3 + OH$$

$$(iii) H_2O_3 + OH$$

$$(iii) H_2O_3 + OH$$

(1) 
$$CH_3 - C - CH_2CH_2CH_3$$
 $C$ 
 $C$ 
 $C$ 
 $C$ 

(2) 
$$CH_2 = C - CH_2CH_2CH_3$$
  
 $CH(CH_3)_2$ 

(3) 
$$CH_3 - C = CH - CH_2CH_3$$
  
 $CH(CH_3)_2$ 

(4) 
$$CH_3 - CH - CH = CH - CH_3$$
  
 $CH(CH_3)_2$ 

30. Preparation of Bakelite proceeds via reactions

### [JEE (Main)-2020]

- (1) Electrophilic substitution and dehydration
- (2) Electrophilic addition and dehydration
- (3) Nucleophilic addition and dehydration
- (4) Condensation and elimination
- 31. The major aromatic product C in the following reaction sequence will be

$$\frac{\text{HBr}(\text{excess}),}{\Delta} A \xrightarrow{\text{(i) KOH (A lc.)}} B$$

[JEE (Main)-2020]

 Two compounds A and B with same molecular formula (C<sub>3</sub>H<sub>6</sub>O) undergo Grignard's reaction with methylmagnesium bromide to give products C and D. Products C and D show following chemical tests.

Test	С	D
Ceric		-
ammonium	Positive	Positive
nitrate Test		
Lucas Test	Turbidity	Turbidity obtained immediately
	after five minutes	
Iodoform Test	Positive	Negative

C and D respectively are

[JEE (Main)-2020]

(1) 
$$C = H_3C - CH_2 - CH - CH_3$$
;

$$\begin{array}{c} \operatorname{CH_3} \\ \operatorname{D} = \operatorname{H_3C} - \operatorname{C} - \operatorname{OH} \\ \operatorname{CH_3} \end{array}$$

(2) 
$$C = H_3C - CH_2 - CH_2 - CH_2 - OH$$
;

$$\label{eq:defD} \begin{split} \mathsf{D} = \mathsf{H}_3 \mathsf{C} - \mathsf{CH}_2 - \mathsf{CH} - \mathsf{CH}_3 \\ \mathsf{OH} \end{split}$$

(3) 
$$C = H_3C - CH_2 - CH_2 - CH_2 - OH$$
;

$$D = H_3C - C - OH$$
  
 $CH_3$ 

$$D = H_3C - CH_2 - CH - CH_3$$
 OH

33. An organic compound 'A' (C<sub>9</sub>H<sub>10</sub>O) when treated with conc. HI undergoes cleavage to yield compounds 'B' and 'C'. 'B' gives yellow precipitate with AgNO<sub>3</sub> where as 'C' tautomerizes to 'D'. 'D' gives positive iodoform test. 'A' could be

[JEE (Main)-2020]

(2) 
$$H_3C - CH = CH_2$$

(3) 
$$\sqrt{\phantom{a}}$$
 -CH<sub>2</sub> - O - CH = CH<sub>2</sub>

34. The major product of the following reaction is:

$$\begin{array}{c|c} \text{OH} & \text{conc. HNO}_3 + \text{conc.} \\ \hline & \text{H}_2\text{SO}_4 \\ \hline & \text{NO}_2 \\ \end{array}$$

(3) 
$$H_3C$$
  $OH$   $NO_2$   $NO_2$   $NO_2$   $NO_2$ 

35. An organic compound [A], molecular formula C<sub>10</sub>H<sub>20</sub>O<sub>2</sub> was hydrolyzed with dilute sulphuric acid to give a carboxylic acid [B] and an alcohol [C]. Oxidation of [C] with CrO<sub>3</sub> – H<sub>2</sub>SO<sub>4</sub> produced [B]. Which of the following structures are not possible for [A]? [JEE (Main)-2020]

(1) 
$$CH_3 - CH_2 - CH - OCOCH_2CH - CH_2CH_3$$
  $CH_3$   $CH_3$ 

(2) 
$$CH_3 - CH_2 - CH - COOCH_2 - CH - CH_2CH_3$$
  
 $CH_3$ 

- (3)  $(CH_3)_3C COOCH_2C(CH_3)_3$
- (4) CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>COOCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>CH<sub>3</sub>
- 36. Consider the following reaction:

The product 'P' giv

The product 'P' gives positive ceric ammonium nitrate test. This is because of the presence of which of these –OH group(s)? [JEE (Main)-2020]

- (1) (d) only
- (2) (c) and (d)
- (3) (b) only
- (4) (b) and (d)
- 37. When neopentyl alcohol is heated with an acid, it slowly converted into an 85 : 15 mixture of alkenes A and B, respectively. What are these alkenes?

[JEE (Main)-2020]

(4) 
$$H_3C$$
  $CH_3$   $H_3C$   $CH_3$  and  $CH_2$ 

38. The major product [B] in the following reactions is

$$CH_{3}$$

$$CH_{3}-CH_{2}-CH-CH_{2}-OCH_{2}-CH_{3}$$

$$\xrightarrow{\text{HI}} \text{Heat} \rightarrow [A] \text{ alcohol} \xrightarrow{\text{H}_{2}SO_{4}} \rightarrow [B]$$

[JEE (Main)-2020]

(1) 
$$CH_3 - CH_2 - CH = CH - CH_3$$

(2) 
$$CH_3 - CH_2 - C = CH_2$$

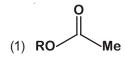
$$\begin{array}{c}
\mathsf{CH}_{3} \\
\mathsf{I}
\end{array}$$
(3)  $\mathsf{CH}_{3} - \mathsf{CH} = \mathsf{C} - \mathsf{CH}_{3}$ 

(4) 
$$CH_2 = CH_2$$

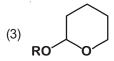
39. The major product [C] of the following reaction sequence will be

$$CH_2 = CH - CHO \xrightarrow{(i) NaBH_4} [A] \xrightarrow{Anhy.} [B] \xrightarrow{DBr} [C]$$

40. Which of the following derivatives of alcohols is unstable in an aqueous base? [JEE (Main)-2020]



(2) RO-CMe<sub>3</sub>



41. The major product of the following reaction is

[JEE (Main)-2020]

(1) 
$$CHCH_3$$
  $CH_2CH_3$  (2)  $CH_2CH_3$  (3)  $CH_2CH_3$  (4)  $CH_2CH_3$ 

42. A solution of phenol in chloroform when treated with aqueous NaOH gives compound P as a major product. The mass percentage of carbon in P is \_\_\_\_\_. (to the nearest integer)

(Atomic mass : C = 12; H = 1; O = 16)

