

Chemistry 2

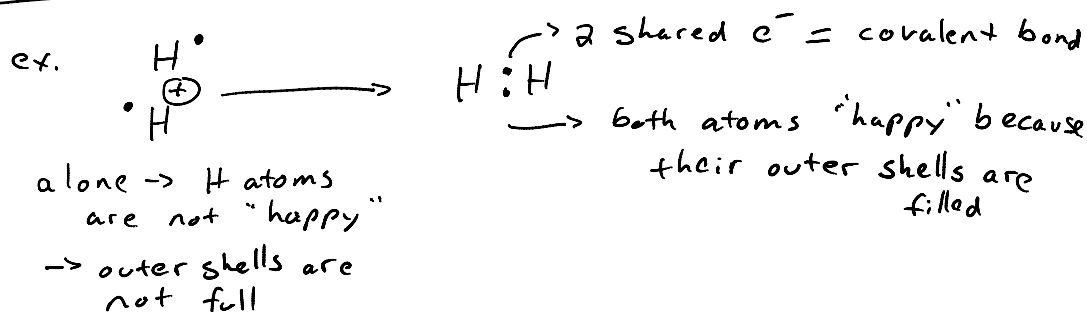
- **iClicker 9A**
- **Covalent Bonds**
 - single, double, triple
 - charge!
 - Abbreviations
- **Answer to iClicker 9B**

- **Due in Lab next week**
 - **Pre-Lab 4**
 - **LEGO Meiosis lab report**

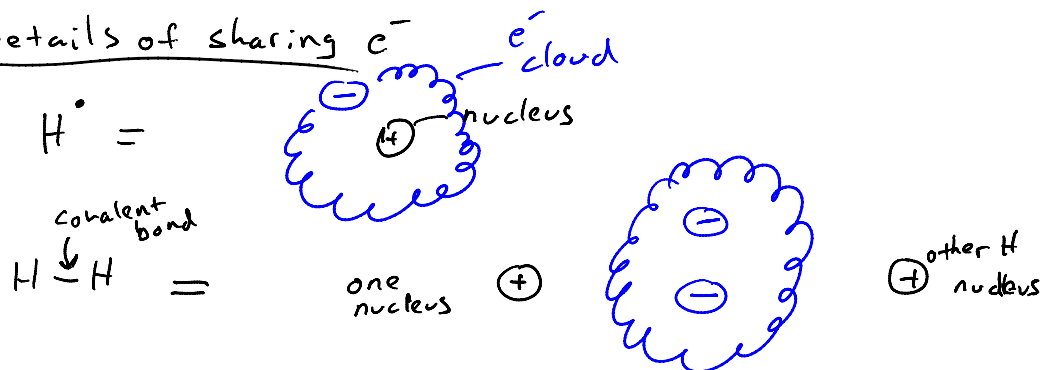
- **Register your iClicker**
-
- **Exam Information**

- **Last names starting with A through E**
 - 11th Floor of Healy
- **Last names starting with F through Z**
 - Lipke (here)

Covalent Bond 2 e⁻ are shared by 2 atoms

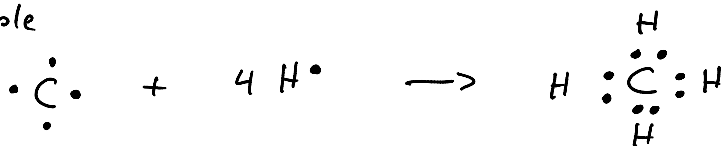


Details of sharing e⁻



- e⁻ spends most of their time between nuclei
 - (-) charged cloud in between nuclei
 - pulls (+) nuclei together → forms covalent bond
- $\text{H}-\text{H}$ or H_2

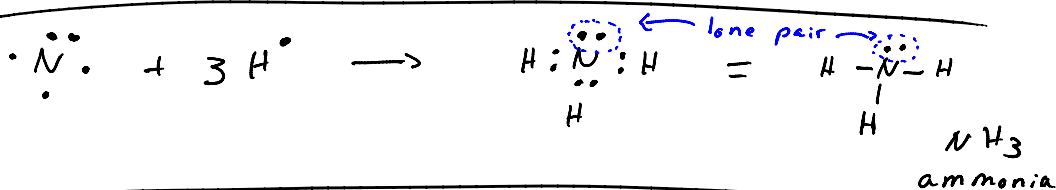
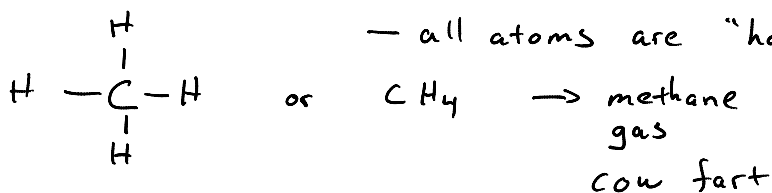
example



now, all H's have 2e⁻ in their o.s.

C has 8e⁻ in its o.s.

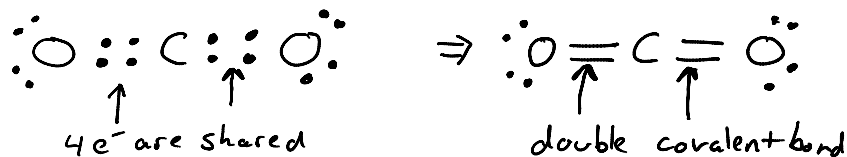
- all atoms are "happy"



et. CO_2

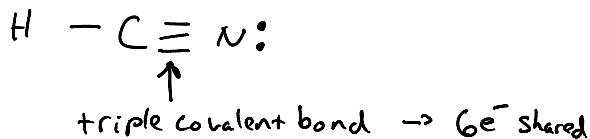
$\cdot\dot{C}\cdot$ → wants to make 4 covalent bonds

2 $\cdot\ddot{O}\cdot$ → wants to make 2 covalent bonds



double covalent bond is stronger than a single covalent bond

H C N → hydrogen cyanide
 $\downarrow \downarrow \downarrow$
 1 4 3 (# bonds)



Exceptions to # of bonds → due to interactions with water

in Bio III → oxygen & nitrogen

$H^\bullet \rightarrow H^+ + e^-$

Nitrogen



or



3 covalent bonds
 1 lone pair
 no charge

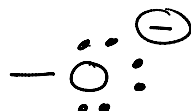
4 covalent bonds
 no lone pair
 single, positive charge

Oxygen



2 covalent bonds
 2 lone pair
 no charge

or



1 covalent bond
 3 lone pair
 single, negative charge



if e⁻ is removed from H, call this a proton

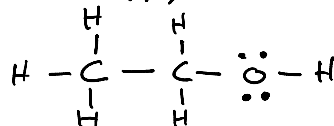
Abbreviations

① "—" → single covalent bond

- ① "—" → single covalent bond
 = → double
 ≡ → triple

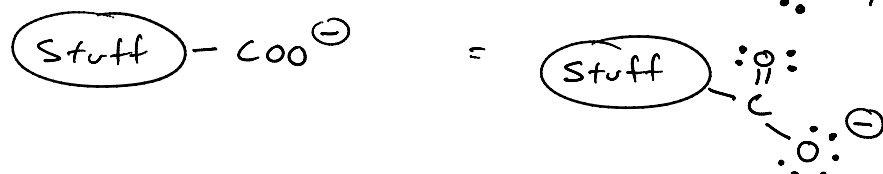
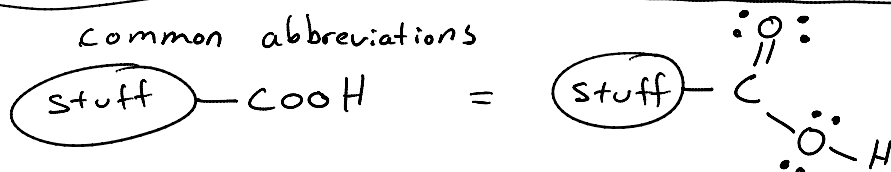
② bonds are implied where necessary to make the atoms "happy"

ex. $\text{H}_3\text{CCH}_2\text{OH}$



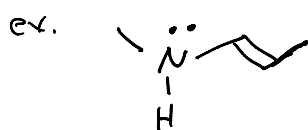
ethyl alcohol → beer & wine

common abbreviations

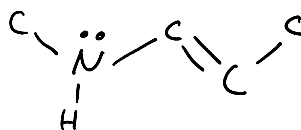


③ corners, vertices, & end of lines are C atoms

H's implied on C atoms where its necessary to make atoms "happy"



add C's



add H's

