

Statistical Mechanical Equilibrium Constants- $\text{Cl} + \text{H}_2 \rightarrow \text{HCl} + \text{H}$

Kp for atom- diatom reactions
 $\text{Cl} + \text{H}_2 \rightarrow \text{HCl} + \text{H}$
 $\text{A} + \text{BC} \rightarrow \text{AB} + \text{C}$

Mr(A) (g/mol)	34.9688		trans	rot	vib	g_{elec}	$e^{-\Delta E_o/RT}$	
Mr(B) (g/mol)	1.0078							
Mr(C) (g/mol)	1.0078		Kp=	218.3191	19.562	1	2	0.16494044
	AB=HCl	BC=H2		591.7355	1.7037	1	4	
ro (nm)	0.12746	0.07417						
symm.number	1	2						
vo (cm-1)	2989.74	4395.2	Kp=	0.34936				
Do (eV)	4.43	4.4763						
Do (kJ/mol)	427.42855	431.895806						
g (grnd.state)	1	1						
	A=Cl	C=H						
g (grnd.state)	4	2						
reduced mass	1.6266E-27	8.3676E-28						
I (kg m2)	2.6426E-47	4.6032E-48	Data from					
Be (cm-1)	10.592893	60.8130047		Karplus and Porter				Table 7.3
qr	19.5621725	1.70374742						
qv	1.00000054	1						
T (K)	298.15							
kT/hc (cm-1)	207.22							
R (J/mol/K)	8.314							
Na	6.0221E+23							
h bar (J s)	1.0546E-34							
c (m/sec)	299792500							