Review:

Base Catalyzed 1,2-addition to carbonyl groups: H-Nu + base Nu: + H-base+

 $R_1 = H$, carbon group

Reversible if Nu is "stabilized"

e.g., ROH, R₂NH, RO, certain carbon anions

Irreversible with simple carbon anions [organometallic reagents]

R-Li, R-MgBr, R-Zn-Br

Acid catalyzed:

Equilibration is generally favorable (K > 1) for aldehydes and unfavorable (K < 1) for ketones, but important electronic and steric effects can influence K_{eq} .

Special case of nucleophile: enolate anion by proton abstraction from a ketone or aldehyde

usual site of reaction with electrophiles
$$\begin{array}{c} \text{possible site of reaction with electrophiles} \\ \text{Possible site of reaction with electrophiles$$

$$\begin{array}{c} \bigoplus_{R} R_1 + \bigoplus_{R_2} H \end{array} \longrightarrow \begin{array}{c} \bigoplus_{H} R_2 & \bigoplus_{H} R_2 & \bigoplus_{H} R_1 & \bigoplus_{H} R_2 & \bigoplus_{H} R_1 & \bigoplus_{H} R_2 & \bigoplus_{H} R_2 & \bigoplus_{H} R_1 & \bigoplus_{H} R_2 & \bigoplus_{H} R_1 & \bigoplus_{H} R_2 & \bigoplus_{H} R_2 & \bigoplus_{H} R_1 & \bigoplus_{H$$

Final Stage:

Makes the process irreversible, pulls the various equilibria steps to the right.

retro-aldol

$$R_1$$
 R_2
 R_1
 R_2
 R_1
 R_2
 R_1
 R_2
 R_2
 R_3
 R_4
 R_4
 R_5
 R_5
 R_5
 R_6
 R_7
 R_8
 R_8

Modern: Selective generation of the enolate anion, add other carbonyl component.

Classics:

Very stabilized carbanion in one component: Knoevenagel Condensation elimination of water to drive the process:

"Conjugate Addition" The Michael Reaction

Combination condensation reactions: Robinson Annulation

Sarett's Cortisone synthesis:

Note: Reactive Anions tend to react "1,2" at the carbonyl carbon and not reverse

The Benzoin reaction and Vitamin B1 [thiamine] Very old:

Role of CN: Add to C=O; favor deprotonation of -H
Easily formed nucleophile and then a good electron-withdrawing group.
[e.g., HO⁻ does not do the same job. Why?]

Nature's analog of CN: Thiamine--pKa 12.7

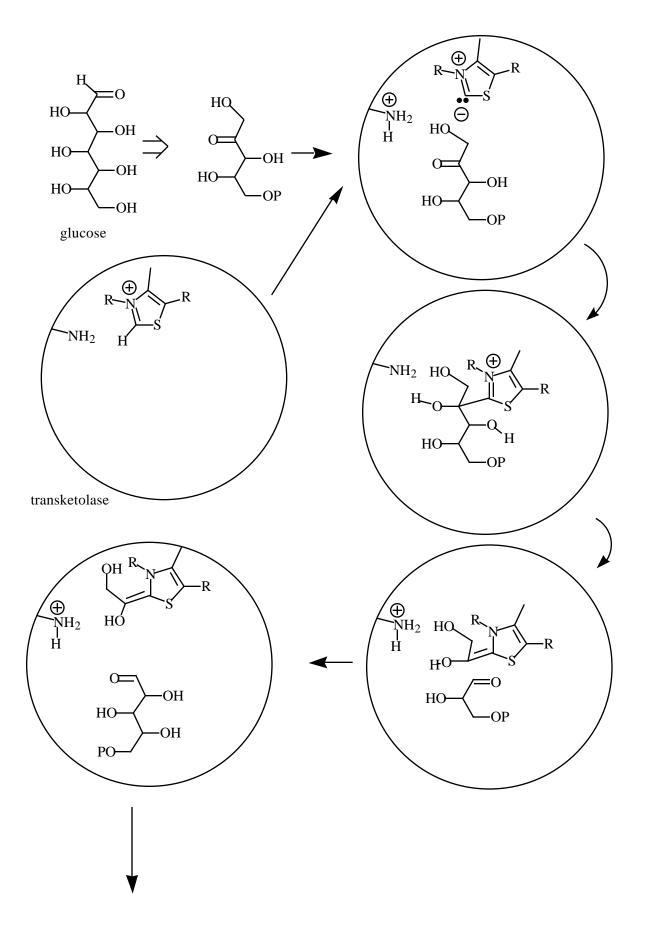
\

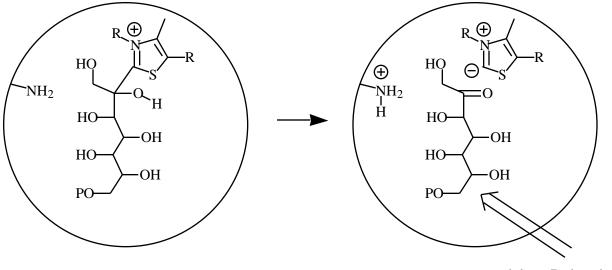
Biological CARBANION Reactions: The Pentose Phosphate Pathway

Overall: 3 Glucose 2 Fructose + glyceraldehyde + 3
$$CO_2$$

$$C_6 \qquad C_6 \qquad C_3$$

$$HO \longrightarrow OH$$





sedulose-7-phosphate

Parallel Role of Imines/immonium ions/enamines:

$$RH_{2}C-C = RH_{2}C-C = RH_{$$