

Anatomy of a Term Symbol

A term symbol is a unique “nickname” for an electronic state

Total orbital angular momentum: $\vec{L} = \sum \vec{\ell}_i$ degeneracy $g_L = 2L + 1 : M_L = -L, \dots, 0, \dots, L$

Total spin angular momentum: $\vec{S} = \sum \vec{s}_i$ degeneracy $g_S = 2S + 1 : M_S = -S, \dots, 0, \dots, S$

Total angular momentum: $\vec{J} = \vec{L} + \vec{S}$ $\vec{J} = \sum \vec{j}_i$ degeneracy $g_J = 2J + 1 : M_J = -J, \dots, 0, \dots, J$

L:	0	1	2	3	4	5
Term:	S	P	D	F	G	H

Clebsch-Gordan Series: $L = \ell_1 + \ell_2, \ell_1 + \ell_2 - 1, \dots, |\ell_1 - \ell_2|$ $M_L = \Sigma m_L$ $L = |M_L|_{\max}$

$$J = L + S, L + S - 1, \dots, |L - S|$$

Term Symbol:

$\text{multiplicity} \leftarrow 2S + 1$
 $\text{Orbital angular momentum} \rightarrow \mathbf{L}$
 $\mathbf{J} \leftarrow \text{total angular momentum}$

Example: C : p ²	$l_1 = 1, l_2 = 1 : L = 2, 1, 0$	gives D, P, and S terms
	$s_1 = \frac{1}{2}, s_2 = \frac{1}{2} : S = 1, 0$	gives triplet and singlet states

$$\begin{array}{ll} {}^1\text{D}: L = 2, S = 0: J = 2 : & {}^1\text{D}_2 \\ {}^3\text{P}: L = 1, S = 1: J = 2, 1, 0: & {}^3\text{P}_2, {}^3\text{P}_1, {}^3\text{P}_0 \\ {}^1\text{S}: L = 0, S = 0: J = 0: & {}^1\text{S}_0 \end{array}$$
$$^1S: L=0, S=0: J=0: \quad ^1S_0$$

Term Symbols for Selected Configurations*

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s^1	2S	s^2, p^6, d^{10}	1S		
$s^1 p^1$	$^{1,3}P$	$s^1 d^1$	$^{1,3}D$	$p^1 d^1$	$^{1,3}P, D, F$
p^1, p^5	2P	p^2, p^4	$^1S, D; ^3P$	p^3	$^2P, D; ^4S$
d^1, d^9	2D				
d^2, d^8	$^1S, D, G; ^3P, F$				
d^3, d^7	$^2P, D, D, F, G, H; ^4P, F$				
d^4, d^6	$^1S, S, D, D, F, G, G, I; ^3P, P, D, F, F, G, H; ^5D$				
d^5	$^2S, P, D, D, D, F, F, G, G, H, I; ^4P, D, F, G; ^6S$				

[illegible]

* J. S. Winn, *Physical Chemistry*, Harper Collins, 1995