Exam 6:

Next Monday... 9:10 am. Sleep lots!
(8/4)

Final Exam, next Friday (8/8), 8:30 am

Final Exam Review Lecture:
Tues, 8/5, 8:30 am

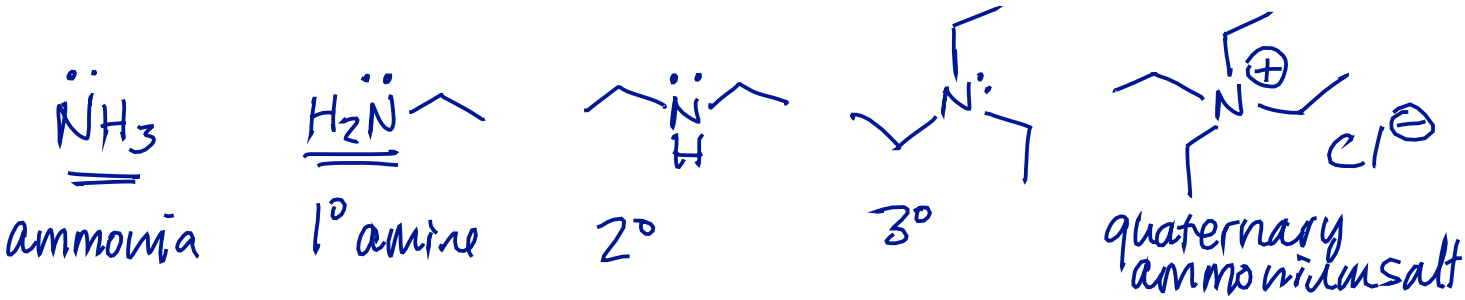
LECTURE TODAY BEGINS 9:45 am

lp on N → bases

$H^+ \leftarrow :NH_3 \rightleftharpoons NH_4^+$

An Introduction to Amines:
Basicity of Nitrogen

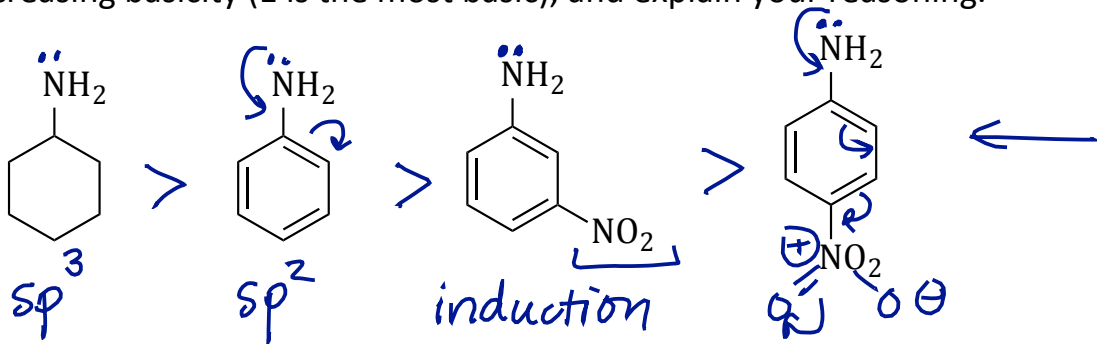
What is an **amine**? Show some examples of amines.

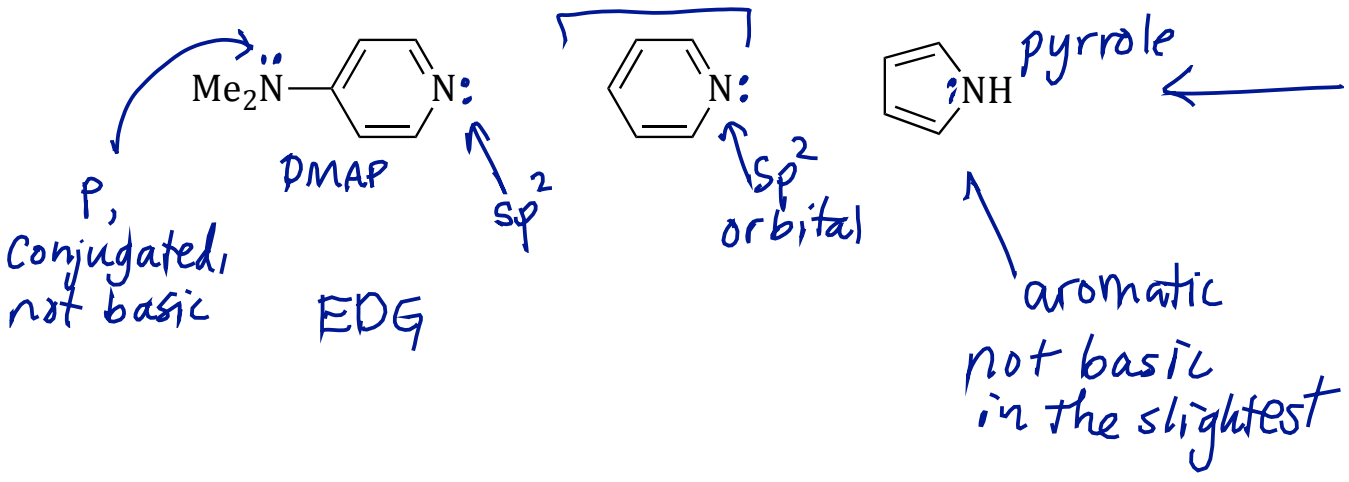


What factors determine the **basicity** of an amine? Rank the following molecules in order of decreasing basicity (1 is the most basic), and explain your reasoning.

1) conj. acid stableness

2) how high-energy is the lone pair

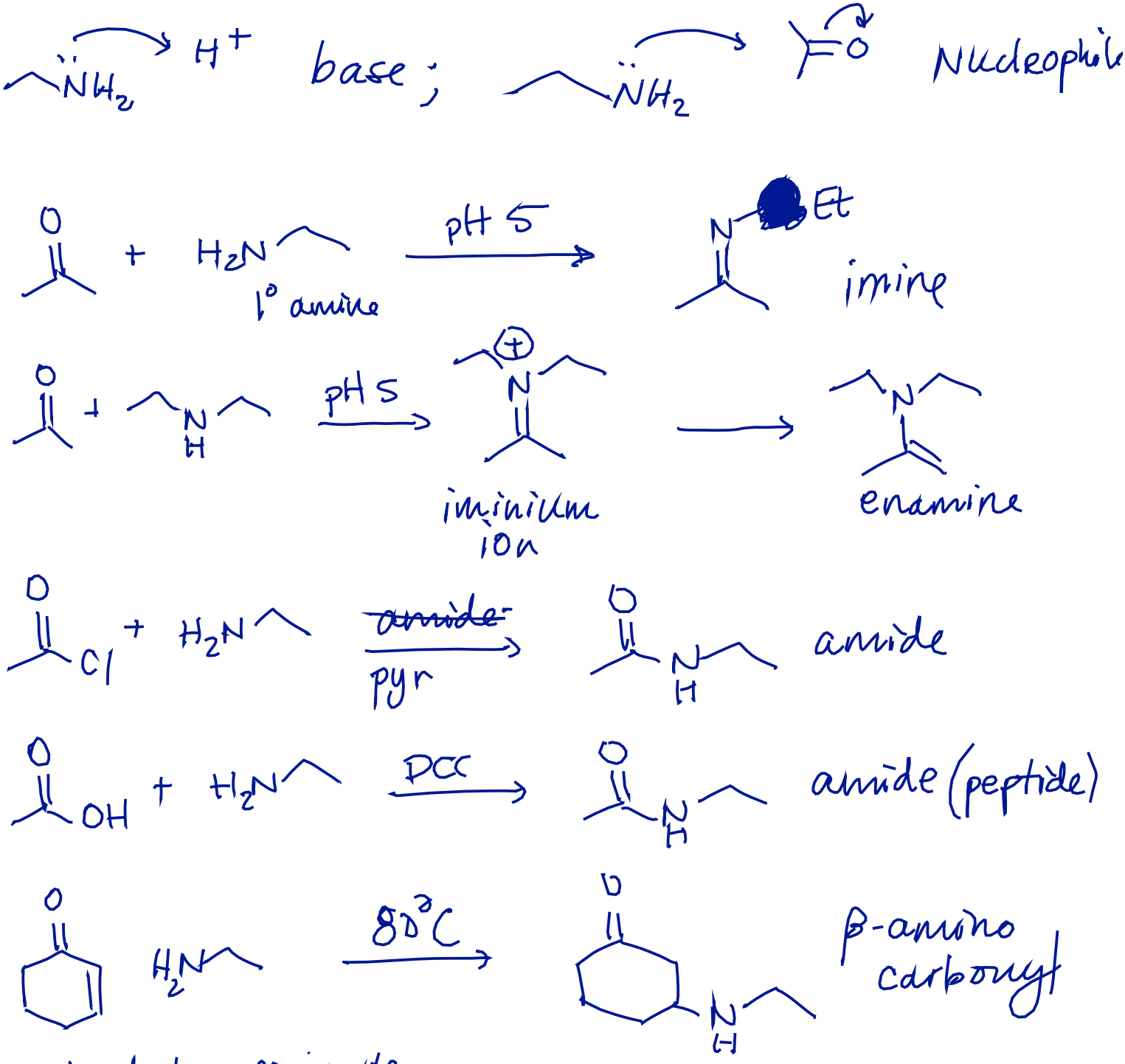




Reading: Section 23.5

Reactions of Amines with Carbonyl Compounds

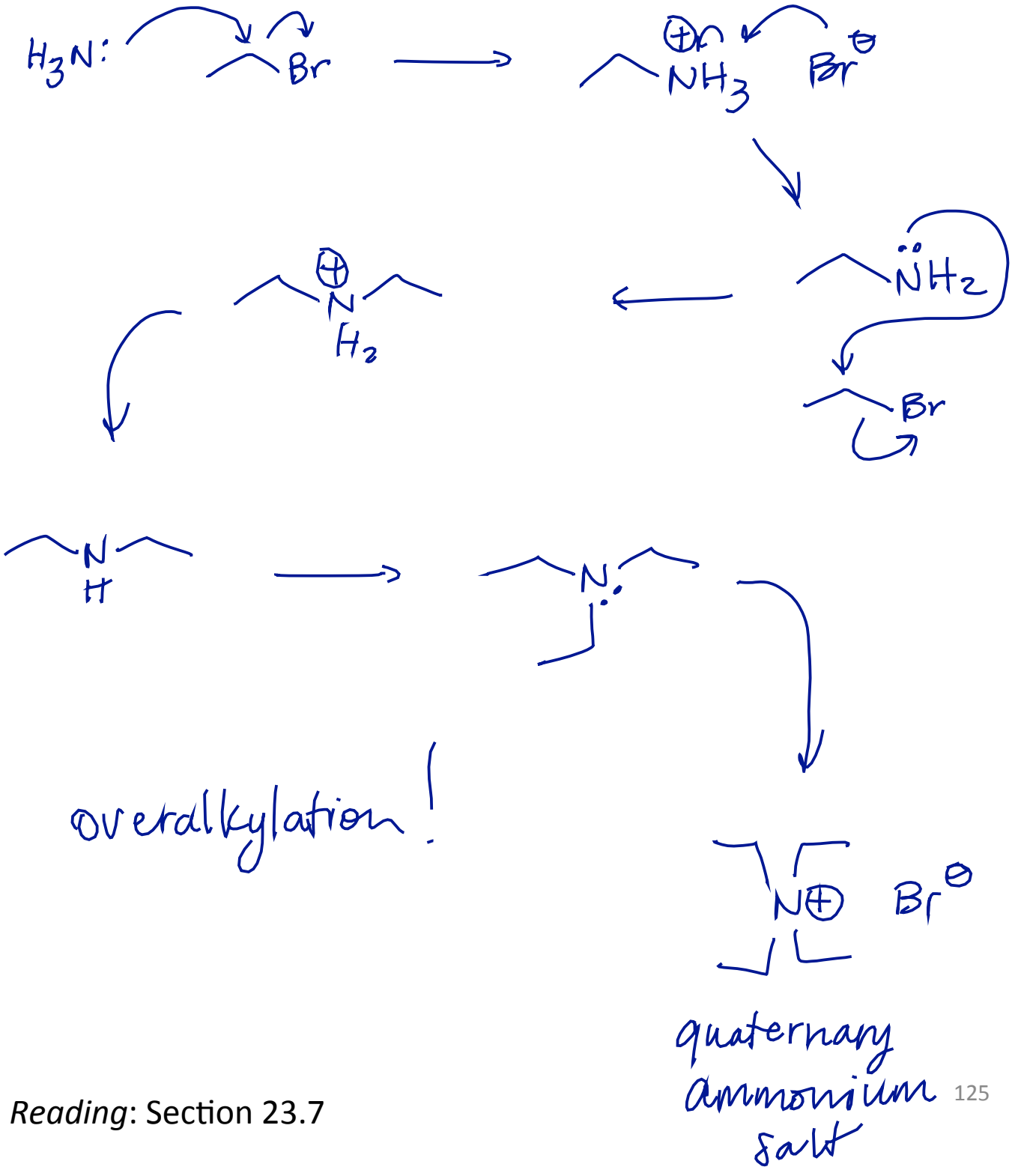
You already know some reactions of amines. What reactions do you know in which amines react with carbonyl compounds?



Reading: Section 23.7

S_N2 Reactions of Ammonia and Amines

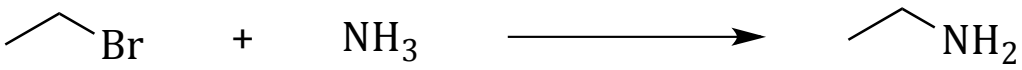
What happens if you combine ammonia with ethyl bromide? Show the mechanism and product(s) of this reaction.



Reading: Section 23.7

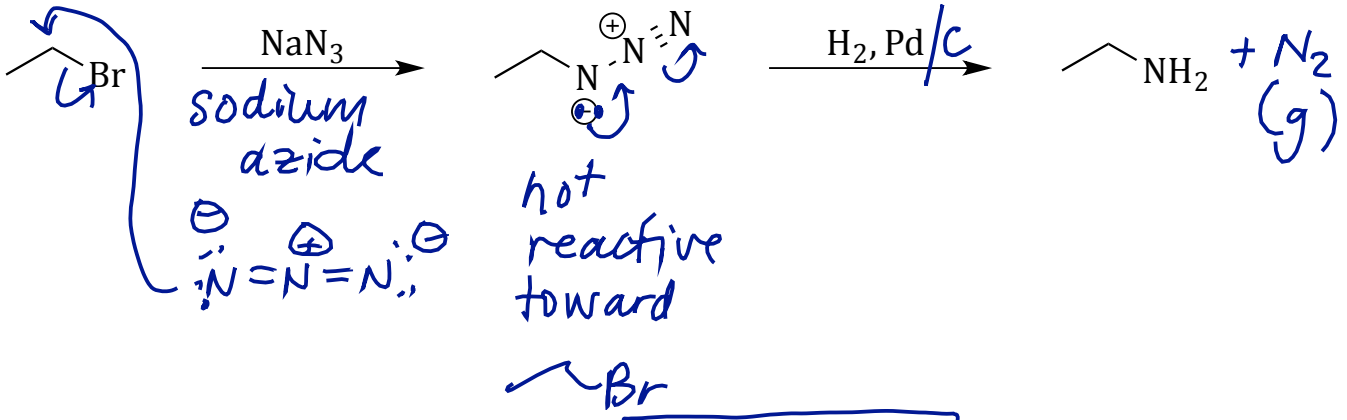
Synthesis of 1° Amines

We **cannot** use the following S_N2 reaction to synthesize a primary amine. Why?

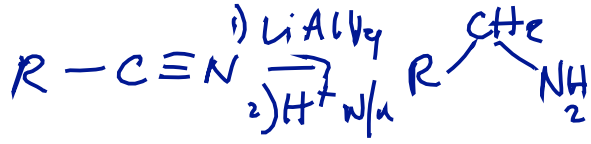


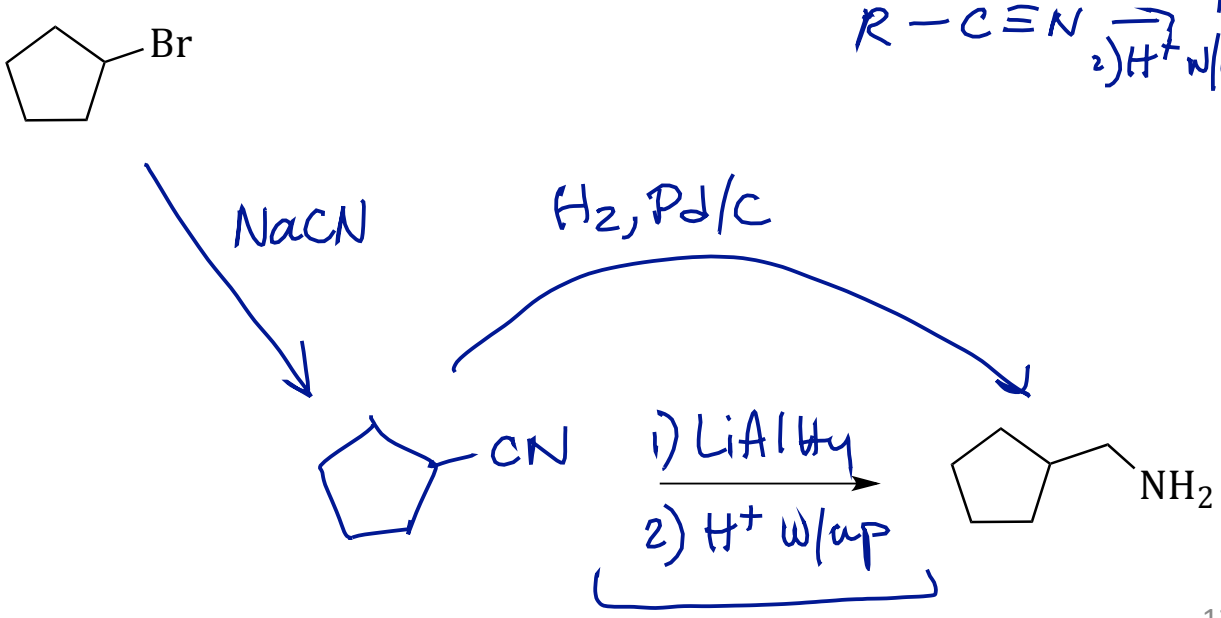
no! alkylation can't stop

The following sequence of reactions *can* be used to synthesize primary amines through an S_N2 pathway. How does this work?



Primary amines can also be synthesized by the **reduction of nitriles**. Show how this process can be used to synthesize the following amine from bromocyclopentane:



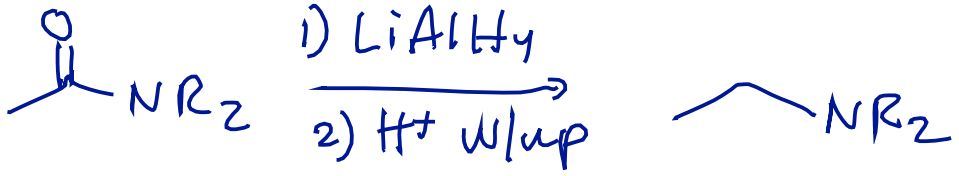


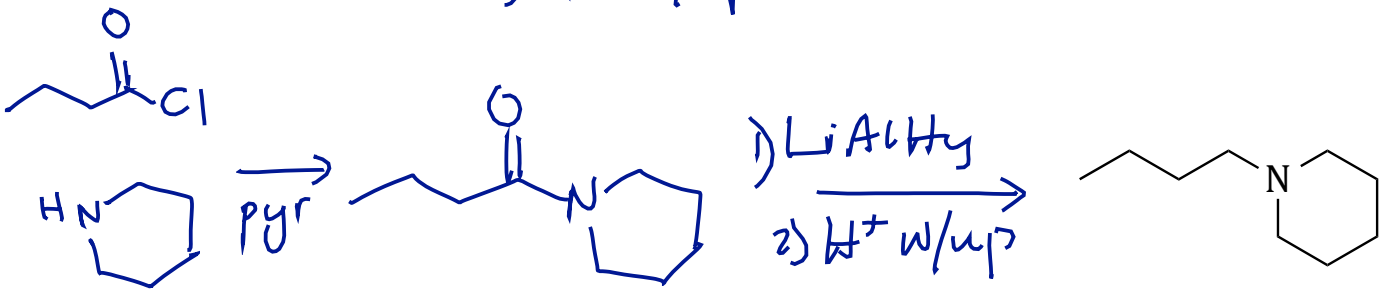
Reading: Section 23.11

Synthesis of 2° & 3° Amines

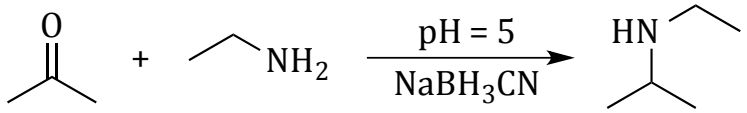
Secondary and tertiary amines also cannot be synthesized by a direct S_N2 reaction. Why?

You already know one method for synthesizing 2° or 3° amines. What is that method, and how can it be used to synthesize the following amine?

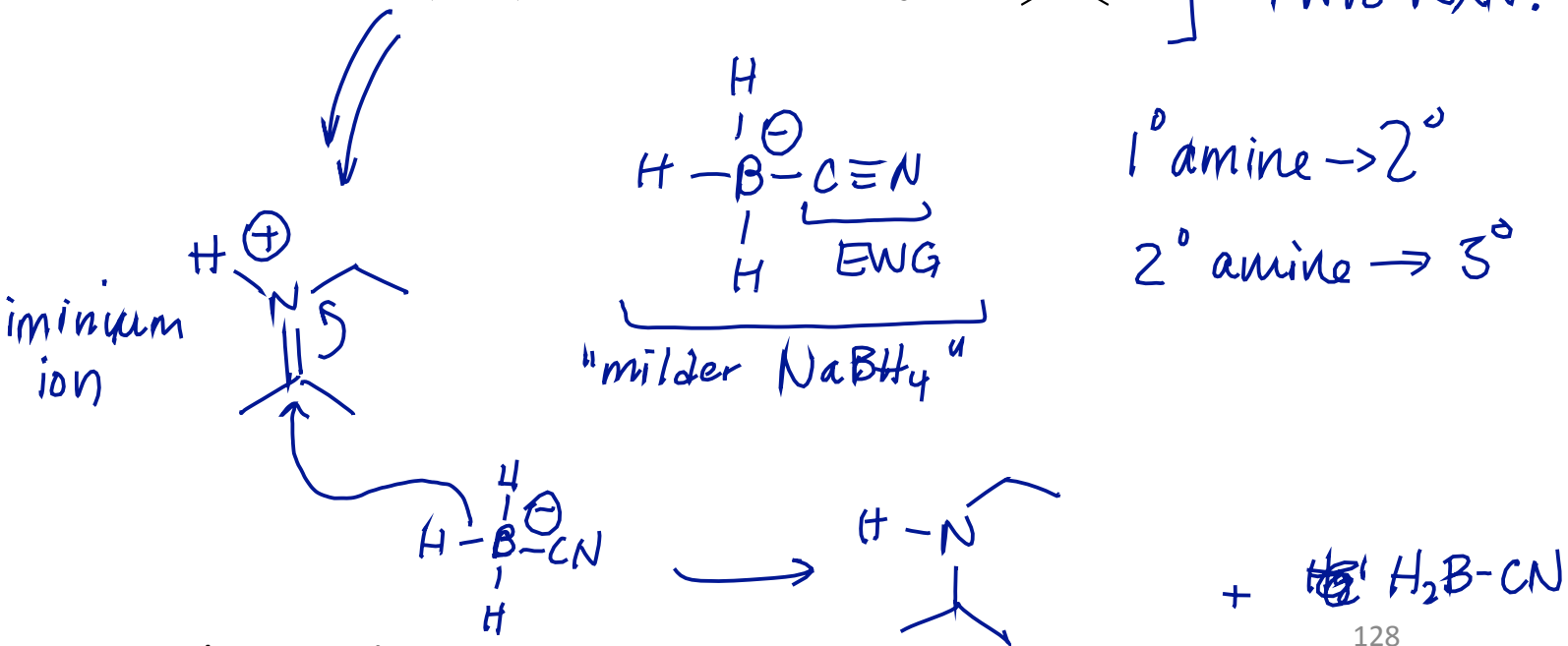




Another very useful synthesis of 2° or 3° amines is **reductive amination**. Show the mechanism of this reaction. How can it be used in synthesis?



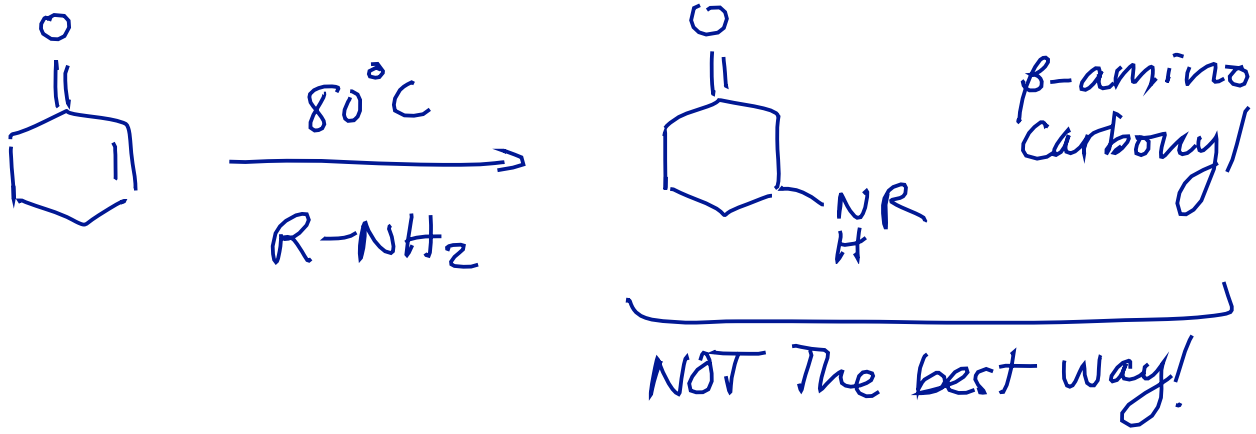
KNOW of THIS RXN!

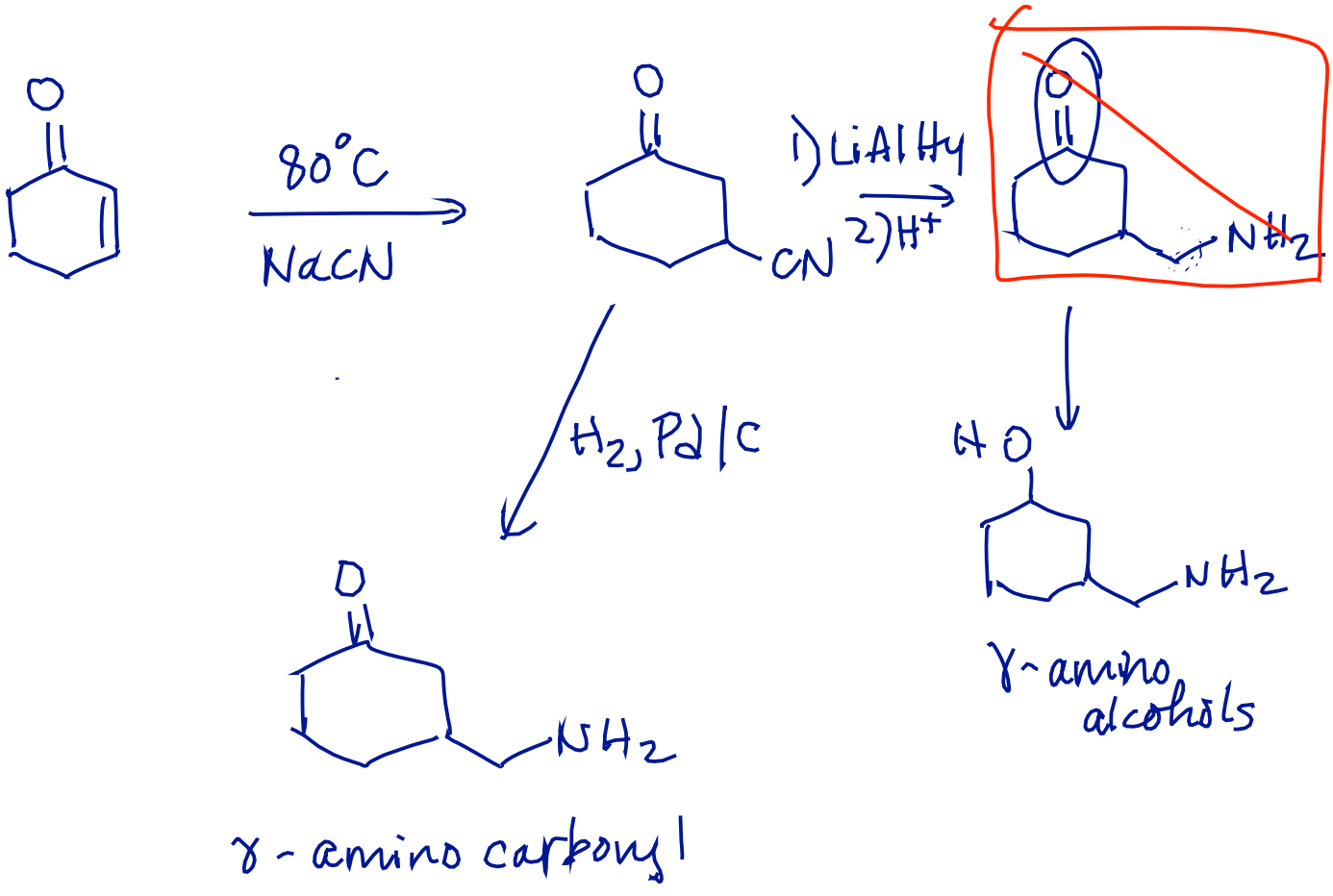


Reading: Section 23.7

Amines from Conjugate Addition

You also know two ways to synthesize amines that involve conjugate addition to an α,β-unsaturated carbonyl compound. What are those methods?



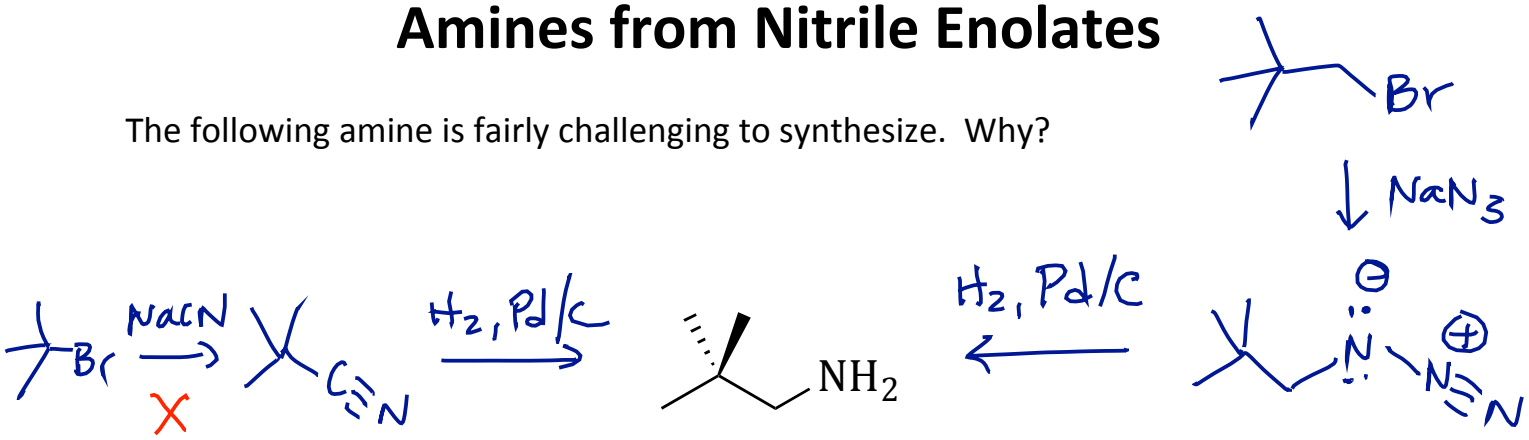


obscure – but nifty

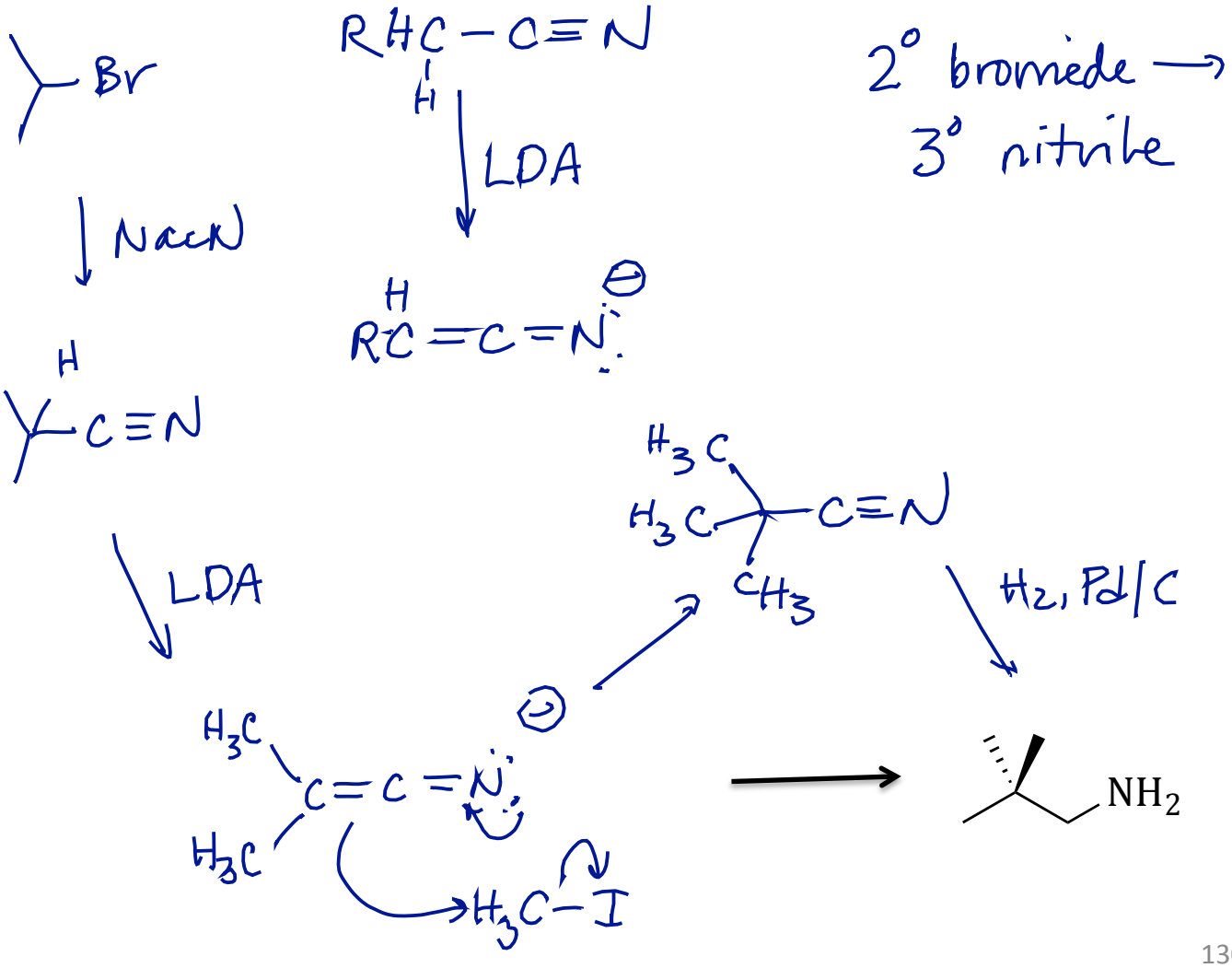
Week 6 July 28, 2014

Amines from Nitrile Enolates

The following amine is fairly challenging to synthesize. Why?

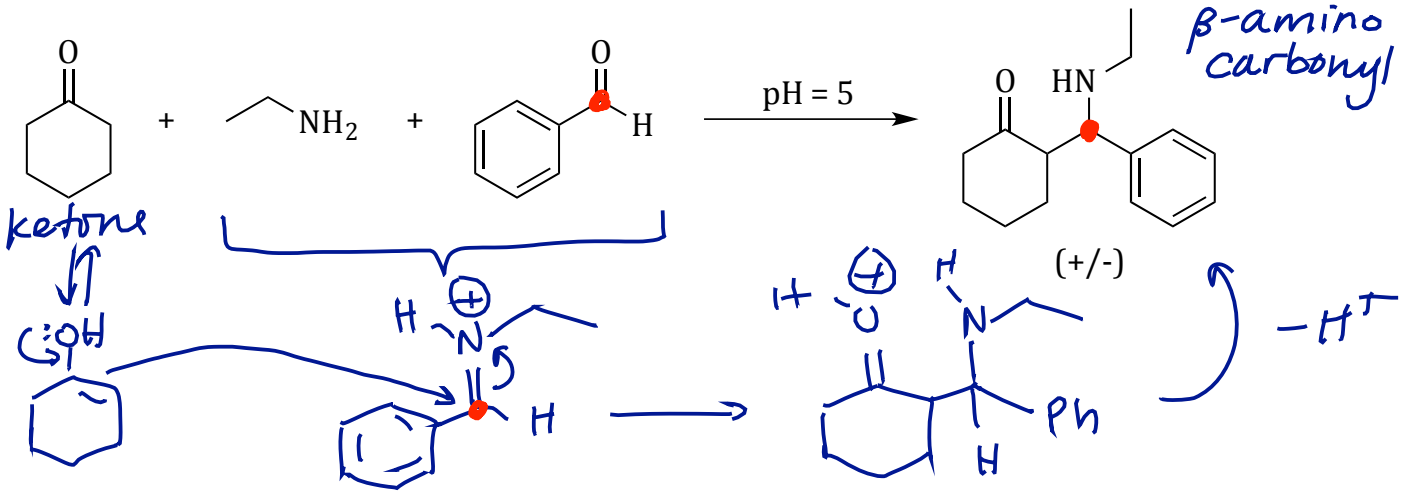


How can that amine be synthesized by a route that involves a **nitrile enolate**?



The Mannich Reaction

The reaction between an enol and an iminium ion is called the **Mannich reaction**; provide a mechanism for this transformation:



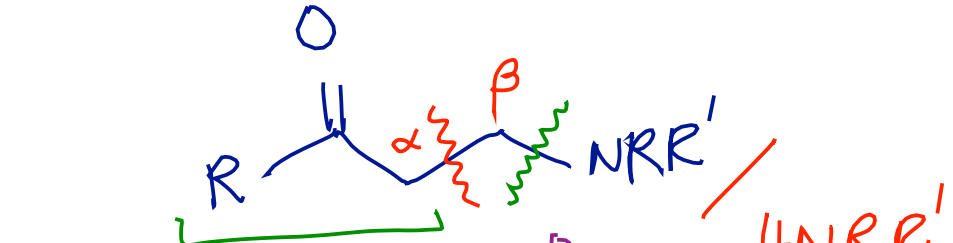
Why only one iminium ion?

1) Aldehydes are more reactive than ketones

2) Only one enolizable carbonyl

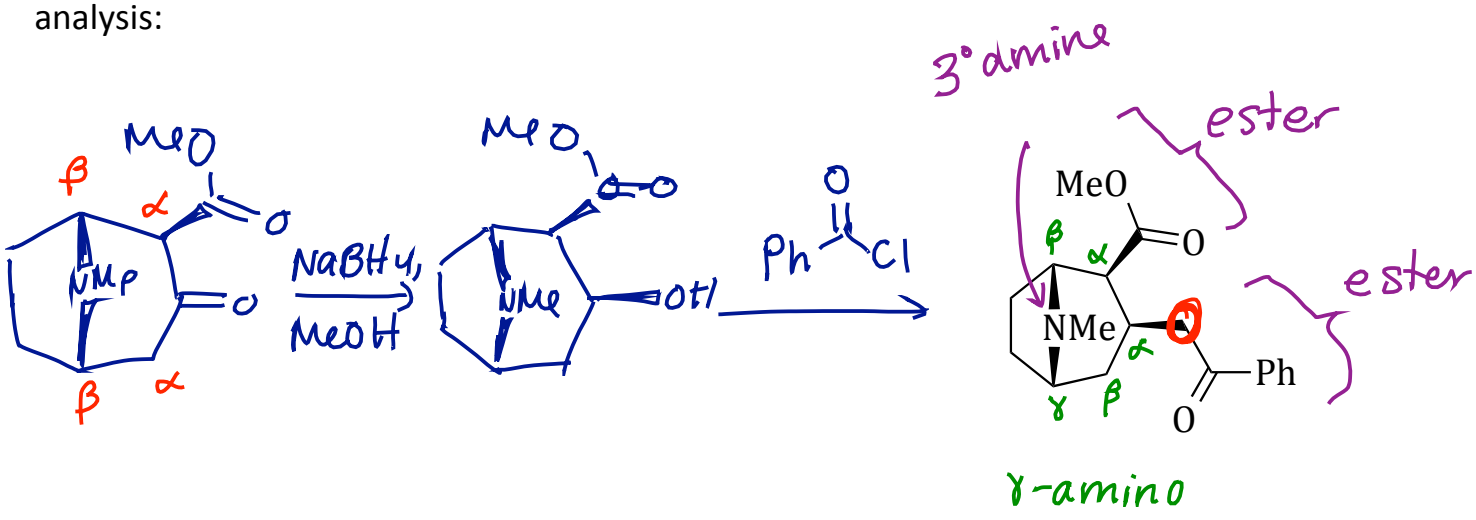
ketone + amine + aldehyde → β-amino carbonyl

How can the Mannich reaction be used in synthesis?

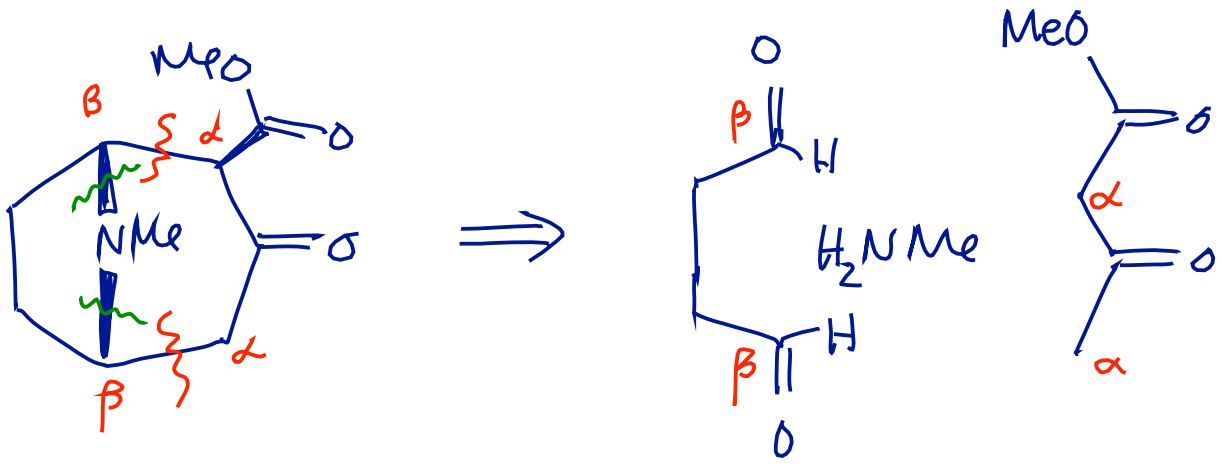


Making Cocaine:
Part 1 – The Strategy

Here is the structure of the natural product **cocaine**. Let's do some retrosynthetic analysis:



How could the Mannich reaction be used in synthesis of cocaine?



Making Cocaine:
Part 2 – The Details

Show the detailed mechanism of the "Mannich synthesis" of this precursor to cocaine: