Molecular Structure Formulas

$$\begin{split} & \frac{\text{Notethal as Structure Formulas}}{g_s(\alpha,r)} = \frac{2\alpha}{1}^{3/4} e^{-\alpha r^2} & g_s(\alpha,r) = \left(\frac{128\alpha^{\circ}}{\pi^3}\right)^{1/4} x \ e^{-\alpha r^2} & \phi_{\mu} = \sum_{i=1}^{n} d_{mi} \ g_i(\alpha,r) \\ & \Psi_a = 1/\sqrt{2} \left| \begin{array}{c} \Psi_{1s}(1)\alpha(1) & \Psi_{1s}(1)\beta(1) \\ \Psi_{1s}(2)\alpha(2) & \Psi_{1s}(2)\beta(2) \end{array} \right| & \frac{-h^2}{2m} \nabla^2 \Psi + \frac{e^2}{4\pi\epsilon_0} \left(-\frac{1}{r_{1A}} - \frac{1}{r_{1B}} + \frac{1}{R}\right) \Psi = E \Psi \\ & c_A(H_{AA} - E \ S_{AA}) + c_B(H_{AB} - E \ S_{AB}) = 0 & E_+ = \frac{H_{AA} + H_{AB}}{1 + S} \\ c_A(H_{AB} - E \ S_{AB}) + c_B(H_{BB} - E \ S_{BB}) = 0 & E_- = \frac{H_{AA} - H_{AB}}{1 - S} \\ & \Psi_+ = \frac{1}{\sqrt{2 + 2} \ S} \left(\Psi_A + \Psi_B\right) & \Psi_- = \frac{1}{\sqrt{2 - 2} \ S} \left(\Psi_A - \Psi_B\right) \\ & E_+ = H_{AA} - \frac{(H_{AB} - H_{AA} \ S)^2}{H_{BB} - H_{AA}} & E_- = H_{BB} + \frac{(H_{AB} - H_{BB} \ S)^2}{H_{BB} - H_{AA}} \\ & d_a = \sum_{j \text{ on a } i = 1}^n n_i c_{ij}^2 & P_{ab} = \sum_{j \text{ on a } k \text{ on b } i = 1}^m n_i 2c_{ij} \ c_{ik} \ S_{jk} \\ & \tilde{\mu} = -\sum_{i=1}^n e \ \hat{\vec{r}}_i & <\mu_x > = -\sum_{i=1}^n e \int \Psi_{MO}^* \ x_i \ \Psi_{MO} \ d\tau & 1 \ D = 3.336x10^{-30} \ C \ m \\ & (\mathcal{X}_A - \mathcal{X}_B)^2 = D_o(A - B) - [D_o(A - A) \ D_o(B - B)]^{1/2} \\ & x_A = \underbrace{I_{1A} + (-EA_A)}_2 & \chi_A = 0.336(x_A - 0.615) \\ & D_o = D_e - 1/2 \ hv_o & V = 1/2 \ k(r - r_o)^2 & k = \left(\frac{\partial^2 V}{\partial r^2}\right) & \text{symmetric} \quad a & 1 \\ & \text{antisymmetric} \quad b & 2 \\ & \Psi_1 = \frac{1}{\sqrt{2}} (\ s + p \) & \Psi_2 = \frac{1}{\sqrt{2}} (\ s + p \) & \Psi_2 = \frac{1}{\sqrt{2}} (\ s + p \) + p_z) \\ & \Psi_2 = \frac{1}{\sqrt{3}} \ s - \frac{1}{\sqrt{6}} p_x + \frac{1}{\sqrt{2}} p_y & \Psi_2 = 1/2 \ (\ s + p_x + p_y + p_z) \\ & \Psi_3 = \frac{1}{\sqrt{3}} \ s - \frac{1}{\sqrt{6}} p_x - \frac{1}{\sqrt{2}} p_y & \Psi_3 = 1/2 \ (\ s + p_x - p_y - p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y - p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x - p_y + p_z) \\ & \Psi_4 = \frac{1}{2} (\ s + p_x$$