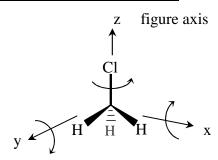
Rotational Spectroscopy- Symmetric Tops

Classical:

$$E = \frac{J_{x}^{2}}{2I_{xx}} + \frac{J_{y}^{2}}{2I_{yy}} + \frac{J_{z}^{2}}{2I_{zz}}$$



$$\begin{split} \overline{I_{yy} = I_{xx} = I_{\perp}} & \quad I_{zz} = I_{\parallel} \\ E = \frac{1}{2I_{\perp}} \left(J_x^2 + J_y^2\right) + \frac{1}{2I_{\parallel}} J_z^2 \end{split}$$

$$\overline{E = \frac{1}{2I_{\perp}}(J_{x}^{2} + J_{y}^{2}) + \frac{1}{2I_{\perp}}J_{z}^{2} + \frac{1}{2I_{\parallel}}J_{z}^{2} - \frac{1}{2I_{\perp}}J_{z}^{2}}$$

$$E = \frac{J^{2}}{2I_{\parallel}} + \left(\frac{1}{2I_{\parallel}} - \frac{1}{2I_{\perp}}\right)J_{z}^{2} \qquad J^{2} = J_{x}^{2} + J_{y}^{2} + J_{z}^{2}$$

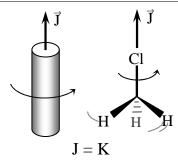
$$\overline{J^2 = \hbar^2 J(J+1)} \qquad \qquad J_z = K \ \hbar \qquad \qquad K = 0, \pm 1, \pm 2, ..., \pm J$$

$$\begin{split} \widetilde{F}_{JK} &= \widetilde{B}J(J{+}1) + (\widetilde{A} - \widetilde{B})K^2 \\ \widetilde{B} &= \frac{\hbar}{4\pi I_{\parallel}c} \qquad \qquad \widetilde{A} = \frac{\hbar}{4\pi I_{\parallel}c} \end{split}$$

dipole moment parallel to figure axis (≈ handle)

$$\Delta K = 0$$
 $\Delta J = \pm 1$

$$\begin{split} \widetilde{\nu}_{J} &= \widetilde{F}_{J',K} - \widetilde{F}_{J'-1,K} = \widetilde{B}J(J+1) - \widetilde{B}(J-1)(J-1+1) \\ &= 2\widetilde{B}J' \qquad \text{as for diatomics} \\ &J' \sim upper \ level \end{split}$$



no change in electric field as molecule rotates

