Chemistry 304B, Spring 1999

Lecture 32

Review sessions: Monday evening, 8 pm Wednesday evening, 8 pm Exam: Thursday: 7:30 pm

General:

There are 16 [cyclic] aldohexoses: can you draw them?

Do it when you are having trouble falling asleep; it is pretty tedious.

Five carbon sugars: (CH₂O)₅

8 [cyclic] stereoisomers in this series (aldehyde form)

Most important reaction of sugars: Glycoside bond formation

both anomers of the methyl glycoside glucopyranoside

$$\begin{array}{c} HO \\ HO \\ HO \\ HO \\ HO \end{array} \longrightarrow \left[\begin{array}{c} HO \\ HO \\ HO \\ HO \end{array} \right]$$

Many common sugars are disaccharides (two sugar units): C₁₂H₂₂O₁₂

sucrose, lactose, maltose

Reaction of sugars with an anomeric -OH group (hemiacetal): equilibration to the aldehyde is rapid

Polysaccharides: cellulose, starch, glycogen are common All based on glucose

Cellulose: ca 3000 units (MW 500,000) cotton, filter paper = 100% cellulose. Wood, straw = 50%

Structure of cellulose

H-bonding network:

Properties depend on how treated:

Starch:

Two components:

Glycogen: energy storage in humans, for fast release.

Nitrogen functionalized sugars: Amino sugars, modified as the amide

poly-NAG = chitin structural material, with CaCO₃, of crab and lobster shells

natural products

Daunomycin: anticancer agent

Bacterial cell walls: very different from animals--polysaccharide cross-linked with polypeptide

An enzyme cleaves the bacterial cell wall: hen eggwhite **lysozyme** (1909)

129 AA 4 disulfide bonds [handout on structure and mechanism]

Binds to a 6-unit section of the polysaccharide of the cell wall and hydrolyzes the acetal linkage

Essential reaction: