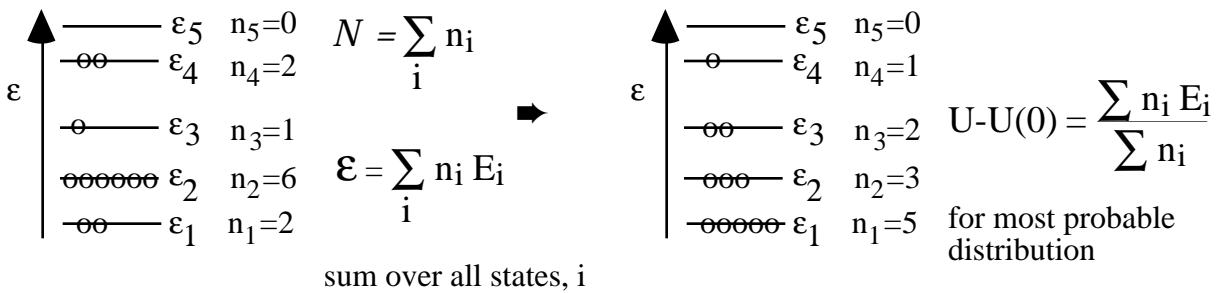


## Boltzman Distribution and the Most Probable Distribution



$$W = \frac{N!}{n_1! n_2! n_3! \dots}$$

$$\ln W = \ln N! - \sum \ln n_i!$$

$$d(\ln W) = \sum \left( \frac{\partial \ln W}{\partial n_i} \right) dn_i$$

$$\text{Constraints: } dN = dn_1 + dn_2 + dn_3 + \dots = \sum dn_i = 0$$

$$d\mathcal{E} = E_1 dn_1 + E_2 dn_2 + \dots = 0$$

$$0 = \sum \left( \frac{\partial \ln W}{\partial n_i} \right) dn_i + \alpha \sum dn_i - \beta \sum E_i dn_i \quad \alpha \text{ and } \beta \text{ undetermined multipliers}$$

$$0 = \sum \left( \left( \frac{\partial \ln W}{\partial n_i} \right) + \alpha - \beta E_i \right) dn_i \quad \text{now } n_i \text{'s are independent!}$$

$$\left( \frac{\partial \ln W}{\partial n_i} \right) + \alpha - \beta E_i = 0$$

$$\text{Sterling's Formula: } \ln x! = x \ln x - x \quad \ln W = N \ln N - N - \sum (n_j \ln n_j - n_j)$$

$$\sum n_i = N \quad \text{so} \quad \ln W = N \ln N - \sum n_j \ln n_j$$

$$\begin{aligned} \left( \frac{\partial \ln W}{\partial n_i} \right) &= - \left( n_i \frac{\partial \ln n_i}{\partial n_i} + \ln n_i \right) \\ &= - \left( n_i \frac{1}{n_i} + \ln n_i \right) \end{aligned}$$

$$\left(\frac{\partial \ln W}{\partial n_i}\right) = -(\ln n_i + 1) \cong -\ln n_i$$

$$-\ln n_i + \alpha - \beta E_i = 0$$

$$\ln n_i = \alpha - \beta E_i$$

$$n_i = e^{\alpha - \beta E_i} = e^{\alpha} e^{-\beta E_i}$$

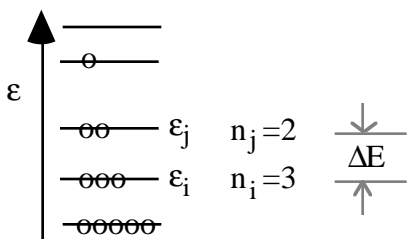
$$N = \sum_i n_i = \sum_i e^{\alpha} e^{-\beta E_i} = e^{\alpha} \sum_i e^{-\beta E_i}$$

$$e^{\alpha} = \frac{N}{\sum_i e^{-\beta E_i}}$$

$$Q = \sum_i e^{-\beta E_i} \quad Q = \text{partition function} = \text{number of accessible states}$$

$$n_i = \frac{N}{Q} e^{-E_i/kT} \quad n_i = \text{number of systems in energy state } E_i$$

$$\frac{n_i}{N} = \frac{e^{-E_i/kT}}{Q} \quad \text{probability of finding a system in energy state } E_i$$



$$\frac{n_j}{n_i} = \frac{e^{-E_j/kT}}{e^{-E_i/kT}} = e^{-(E_j - E_i)/kT}$$

$$\frac{n_j}{n_i} = e^{-\Delta E/kT}$$