

Polarization Functions- Extended Basis Sets

Minimum Basis Sets : Just valence orbitals: STO-3G, 3-21G, 6-31G

For H just 1s

For Li-Ne just 2s, 2p_x, 2p_y, 2p_z

For Na-Ar just 3s, 3p_x, 3p_y, 3p_z (note no d's for P, S, Cl)

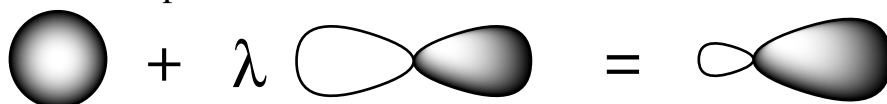
Polarization Functions: For H add p functions

For Li-Ca add d functions (also MNDO/d)

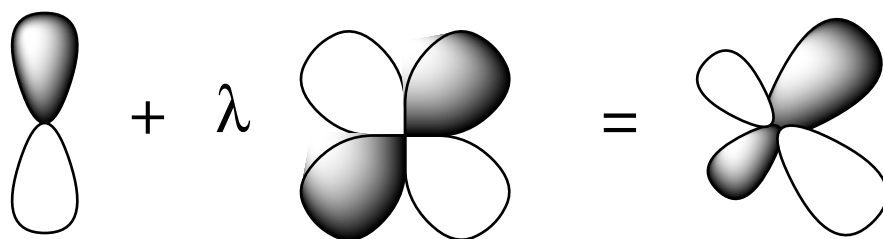
For transition metals add f functions

Polarization functions give more room for the electrons to get away from each other to minimize electron-electron repulsion.

For H add p functions



For Li-Ca add d functions



H₂ at 6-311G** :

$$\Psi_{\text{MO}} = 0.186 \text{1S}_A(\text{inner}) + 0.288 \text{1S}_A(\text{middle}) + 0.133 \text{1S}_A(\text{outer}) + 0.023 \text{2P}_{\text{ZA}} \\ + 0.186 \text{1S}_B(\text{inner}) + 0.288 \text{1S}_B(\text{middle}) + 0.133 \text{1S}_B(\text{outer}) - 0.023 \text{2P}_{\text{ZB}}$$

3-21G(*) Good general purpose level for medium to large systems

For Li-F no polarization functions added

For Na-Ca add d functions

3-21G*, 6-31G*

For Li-F add d functions

For Na-Ca add d functions

3-21G**, 6-31G**

For H add p functions

For Li-F add d functions

For Na-Ca add d functions

6-311G**

Needed to get the most benefit out of MP2 calculations
Only useful for small molecules