Table 4.1. Revised 1997 by C.G. Wohl (LBNL). Heavy element updates in May 2000 by D.E. Groom. The atomic number (top left) is the number of protons in the nucleus. The atomic mass (bottom) is weighted by isotopic abundances in the Earth's surface. Atomic masses are relative to the mass of the carbon-12 isotope, defined to be exactly 12 unified atomic mass units (u). Errors range from 1 to 9 in the last digit quoted. Relative isotopic abundances often vary considerably, both in natural and commercial samples. A number in parentheses is the mass of the longest-lived isotope of that element—no stable isotope exists. However, although Th, Pa, and U have no stable isotopes, they do have characteristic terrestrial compositions, and meaningful weighted masses can be given. For elements 110–112, the atomic numbers of known isotopes are given. Adapted from the Commission of Atomic Weights and Isotopic Abundances, "Atomic Weights of the Elements 1995," Pure and Applied Chemistry 68, 2339 (1996), and G. Audi and A.H. Wapstra, "The 1993 Mass Evaluation," Nucl. Phys. A565, 1 (1993).

1																	18
IA																	VIIIA
1 H																	2 He
Hydrogen	2											13	14	15	16	17	Helium
1.00794	IIA	i					IIIA	IVA	VA	VIA	VIIA	4.002602					
3 Li	4 Be		DED	TODIC	(m) DI	БОБ	5 B	6 C	7 N	8 O	9 F	10 Ne					
Lithium	Beryllium		PER.	IODIC	TABI	E OF	THEE	LEWIE	WI.2			Boron	Carbon	Nitrogen	Oxygen	Fluorine	Neon
6.941	9.012182						10.811	12.0107	14.00674	15.9994	18.9984032	20.1797					
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 CI	18 Ar
Sodium	Magnesium	3	4	5	6	7	8	9	10	11	12	Aluminum		Phosph.	Sulfur	Chlorine	Argon
22.989770	24.3050	IIIB	IVB	VB	VIB	VIIB		VIII		IB	IIB	26.981538	28.0855	30.973761	32.066	35.4527	39.948
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
Potassium		Scandium	Titanium		Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Gallium	German.	Arsenic	Selenium	Bromine	Krypton
39.0983		44.955910		50.9415		54.938049		58.933200	58.6934	63.546	65.39	69.723	72.61	74.92160	78.96	79.904	83.80
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
Rubidium	Strontium		Zirconium	Niobium	Molybd.	Technet.		Rhodium		Silver	Cadmium	Indium	Tin	Antimony	Tellurium	Iodine	Xenon
85.4678	87.62	88.90585	91.224	92.90638		(97.907215)		102.90550	106.42	107.8682		114.818	118.710	121.760	127.60	126.90447	131.29
55 Cs	56 Ba	57–71	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir		79 Au	80 Hg	81 TI	82 Pb	83 Bi	84 Po	85 At	86 Rn
Cesium	Barium	Lantha-	Hafnium		Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Polonium	Astatine	Radon
132.90545	137.327	nides	178.49	180.9479	183.84	186.207	190.23	192.217		196.96655		204.3833	207.2	208.98038	(208.982415)	(209.987131)	(222.017570)
87 Fr	88 Ra	89–103	104 Rf	105 Db	106 Sg	107 Bh			110	111	112						
Francium	Radium		Rutherford.		Seaborg.	Bohrium	Hassium	Meitner.									
(223.019731)	(226.025402)		(261.1089)	(262.1144)	(263.1186)	(262.1231)	(265.1306)	(266.1378)	(269, 273)	(272)	(277)						

Lanthanide series

> Actinide series

La 58

Ce 59

Pr 60

Nd 61

8	Lanthan. Cerium 138.9055 140.116		 Praseodym. 140.90765		Veodym. 144.24	Prometh. (144.912745)				Europium 151.964		Gadolin. 157.25		Terbium 158.92534		Dyspros. 162.50		Holmium 164.93032		Erbium 167.26		Thulium 168.93421		Ytterbium 173.04		Lutetium 174.967		
e 8		Ac nium 27747)	90 Thori 232.0	 91 Pa Protactin. 231.03588	U	2 U Jranium 38.0289			Pluto	nium		Am neric. 061372)	Cı		Ber	Bk kelium 070298)	Calif		Einst	tein.	100 Ferm			lelev.			Lawr	

Gd 65

Tb 66

Dy 67

Ho 68

Tm 70

Yb 71

Er 69

Eu 64

Sm 63

Pm 62