@ On this case, all 4 isomers would be optically active since the sidesimp (SE) is optically active } Let R = de CB HBr) Let R = de CB HBr

H H RAMH IT H RAMH

Cecc Cec Cis-isotachic

H H H H H H H

H H H H H H H H

(13) On this case, only the isoteolic (3) isomer is optically active.

CH H LCH3 N 1+ H

CH3 H H CH3 H H

(1c) On this case, we have 2 different di-isotactic isomers that are optically active

Chy H H A Mile Chy H H H

CH H CH H H H H

$$\overline{M}_{n} = \sum n_{f_{i}} m_{i}$$
 $\int n_{f_{i}} = \frac{n_{i}}{\sum n_{i}}$ $\overline{M}_{k} = \sum \omega_{f_{i}} m_{i}$ $\int \omega_{f_{i}} = \frac{n_{i} m_{i}}{\sum n_{i} m_{i}}$ $m_{i} = 504$ $n_{i} = 1 \text{ mole } n_{f_{i}} = 0.5$ $m_{i} = 1 \text{ mole } n_{f_{i}} = 0.5$

$$\overline{M_n} = (0.5)(504) + (0.5)(1504)$$
 $\overline{M_n} = 100,000 \text{ g lmole}$

$$w_{f_1} = \frac{50k}{2004} = 0.25$$

$$w_{f_2} = \frac{1504}{2004} = 0.75$$

$$m_{k} = (0.25)(504) + (0.75)(1504)$$

$$= 125,000 g/mb$$

$$POI = \frac{m_{k}}{m_{n}} = 1.25$$