3.012 Fund of Mat Sci: Bonding – Lecture 3 (HOST IN THE MA(HINE

Image of a quantum mirage produced by a Co atom placed in the focus of a Co elliptical corral, removed for copyright reasons. Don Eigler, IBM Almaden, *Nature* (2000). See http://domino.watson.ibm.com/comm/pr.nsf/pages/rscd.quantummirage-picb.html/\$FILE/mirage2.jpg

Last time: Schrödinger equation

- 1. Time-dependent Schrödinger equation for one electron in a potential V(r,t) (a plane wave satisfies this eqn.)
- 2. For a stationary potential V(r), we introduced the method of separation of variables, and obtained a) the stationary Schrödinger equation for the spatial part $\varphi(x)$, and b) the equation for the time-dependent function f(t)
- 3. Homework: for a free particle it is easy to obtain $\varphi(x)$ and f(t), and one obtains back the equation of a plane wave
- 4. Studied a free particle in an infinite well (particle in a box)

Homework for Fri 16

- Study: 15.3 (2-,3-dim box), 16.3 (π -electrons in conjugated molecules), 16.5-6 (scanning tunnelling microscope)
- Optional read: 1986 Nobel lecture by Binnig and Rohrer (on the MIT server)

Physical Observables from Wavefunctions

• Eigenvalue equation: (the operator is obtained via the "correspondence" principle)

$$\left[-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + V(x) \right] \varphi(x) = E\varphi(x)$$

• Expectation values for the operator (energy)

$$E = \int \varphi^*(x) \left[-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x) \right] \varphi(x) dx$$

Normalization

Infinite Square Well

$$-\frac{\hbar^2}{2m}\frac{d^2\varphi(x)}{dx^2} = E\varphi(x)$$

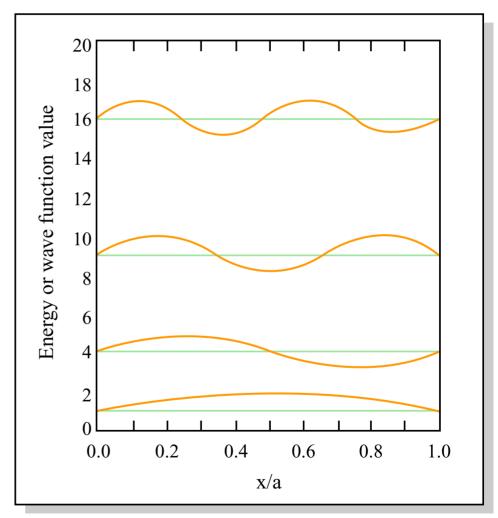


Figure by MIT OCW.

Infinite Square Well

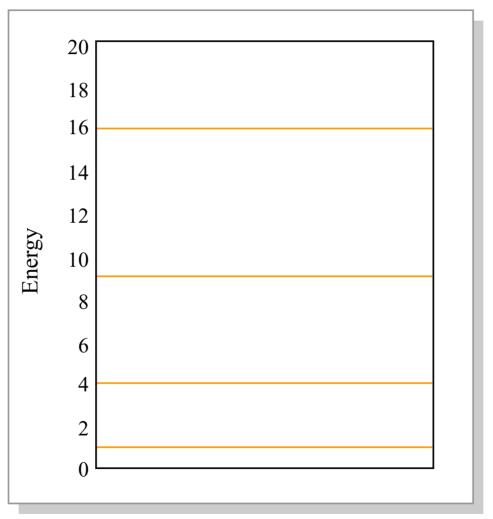


Figure by MIT OCW.

3.012 Fundamentals of Materials Science: Bonding - Nicola Marzari (MIT, Fall 2005)

Absorption Lines (atomic units)

The power of carrots

• β-carotene

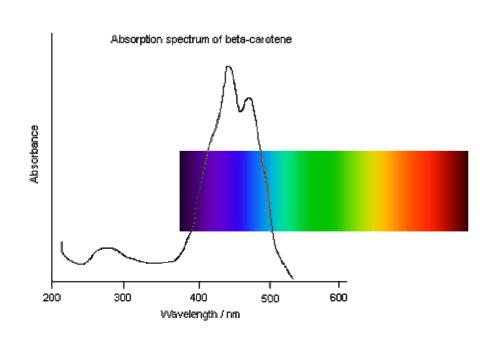




Photo courtesy of Andrew Dunn.

Particle in a 2-dim box

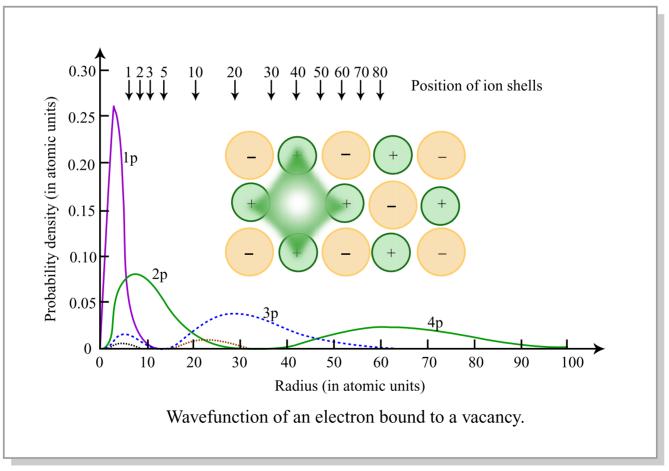
$$-\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} \right) \varphi(x, y) = E \varphi(x, y)$$

Particle in a 2-dim box

$$\varphi(x,y) = C \sin\left(\frac{l\pi x}{a}\right) \sin\left(\frac{m\pi y}{b}\right)$$

$$E = \frac{h^2}{8m} \left(\frac{l^2}{a^2} + \frac{m^2}{b^2} \right)$$

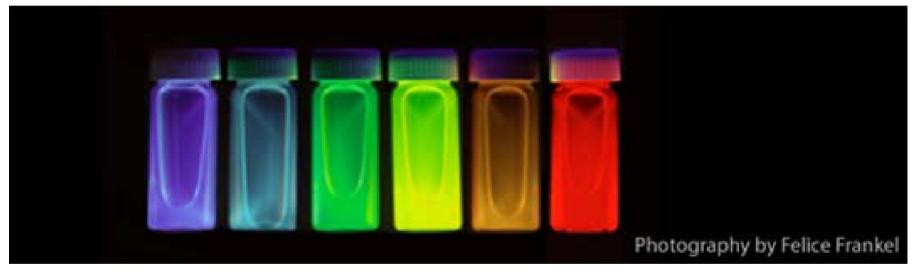
Particle in a 3-dim box: *Farbe* defect in halides (e-bound to a negative ion vacancy)



From Carl Zeiss to MIT...

Scanned image of a journal article removed for copyright reasons. See Avakian, P. and A. Smakula. "Color Centers in Cesium Halide Single Crystals." *Physical Review* 120, no. 6 (December 15, 1960).

Light absorption/emission



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