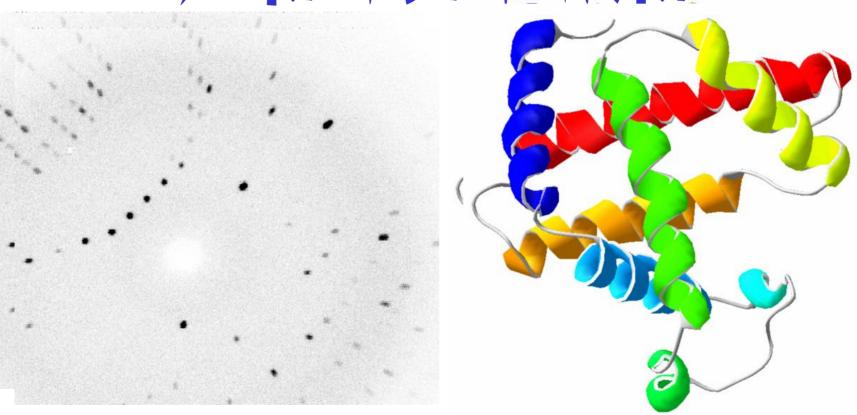
3.012 Fund of Mat Sci: Structure – Lecture 18

X-RAYS AT WORK



An X-ray diffraction image for the protein myoglobin. Source: Wikipedia.

Model of helical domains in myoglobin. Image courtesy of Magnus Manske

Homework for Wed Nov 23

- Prof Wuensch Lecture Notes
- http://capsicum.me.utexas.edu/ChE386K/ for many details (Lect 19 onwards, but note different 2π convention)
- Buy turkey

Last time:

- 1. X-rays generation: undulators and wigglers in synchrotrons, bremsstrahlung and core excitations (e.g. K_{α}) in X-ray tubes
- 2. Reciprocal lattice
- 3. Diffraction gratings Huygens construction
- 4. Laue diffraction from periodic arrays in 1-d, 2-d, 3-d

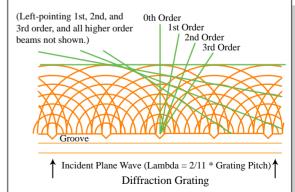


Figure by MIT OCW.

Reciprocal lattice (IV)

 $\vec{G} = h\vec{b}_1 + i\vec{b}_2 + j\vec{b}_3$ with h, i, j integers,

$$\vec{b}_1 = 2\pi \frac{\vec{a}_2 \times \vec{a}_3}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)} \quad \vec{b}_2 = 2\pi \frac{\vec{a}_3 \times \vec{a}_1}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)} \quad \vec{b}_3 = 2\pi \frac{\vec{a}_1 \times \vec{a}_2}{\vec{a}_1 \cdot (\vec{a}_2 \times \vec{a}_3)}$$

 $\vec{G} = (h, i, j)$ are the reciprocal-lattice vectors

d* is distance between two planes of Miller indices h k l

in the reciprocal lattice =
$$\frac{2\pi}{d_{hkl}}$$

$$\frac{2\pi}{d_{h$$

First and second Laue conditions

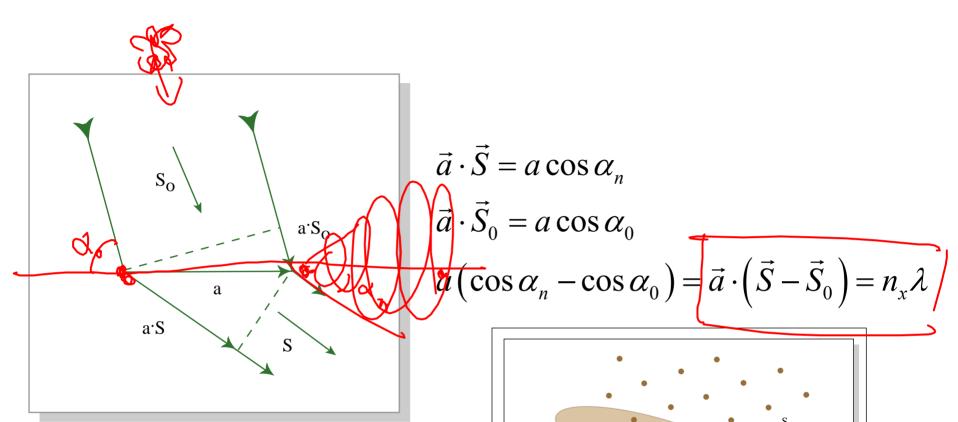


Figure by MIT OCW.

Figure by MIT OCW.

Incident Ray

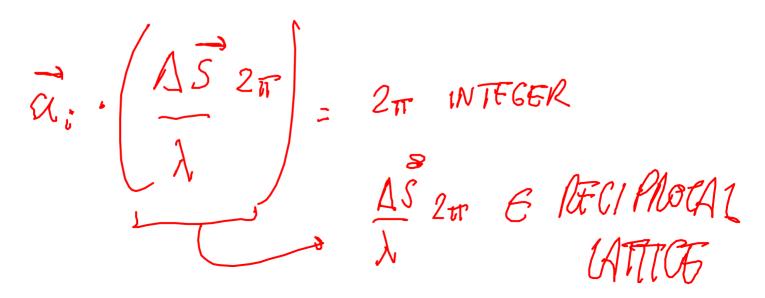
Diffracted Rays

All three Laue conditions

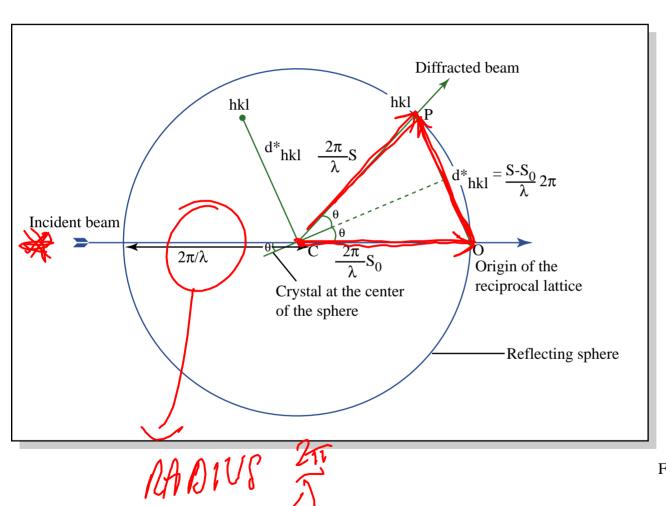
$$\vec{a}_1 \cdot (\vec{S} - \vec{S}_0) = \text{integer multiple of } \lambda$$

 $\vec{a}_2 \cdot (\vec{S} - \vec{S}_0) = \text{integer multiple of } \lambda$

$$\vec{a}_3 \cdot (\vec{S} - \vec{S}_0) = \text{integer multiple of } \lambda$$



Ewald construction



 $\frac{\Delta \vec{S}}{\lambda} = 2\pi \in RCP$ $\lambda = 4\pi T$

Figure by MIT OCW.

Laue condition needs "white" spectrum

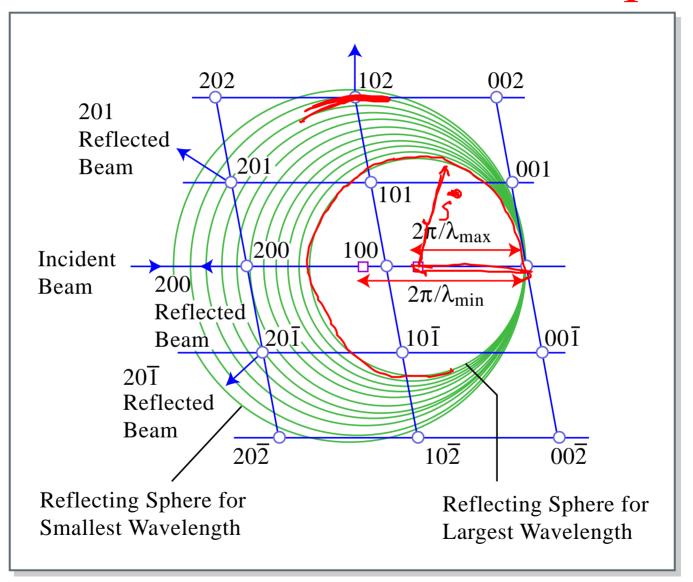


Figure by MIT OCW.

Alternate geometrical view

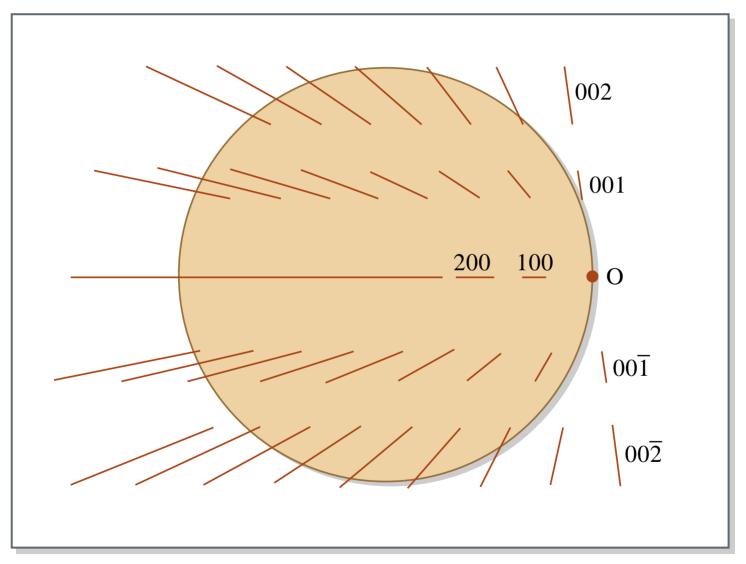


Figure by MIT OCW.

Bragg Law

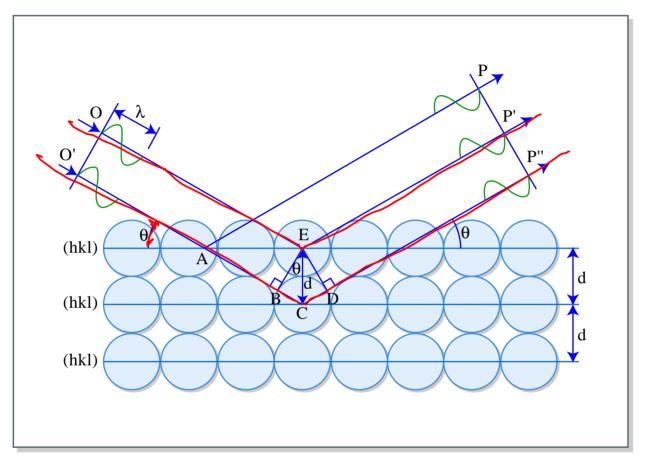
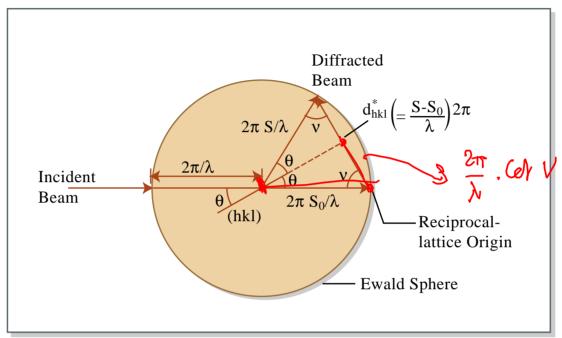
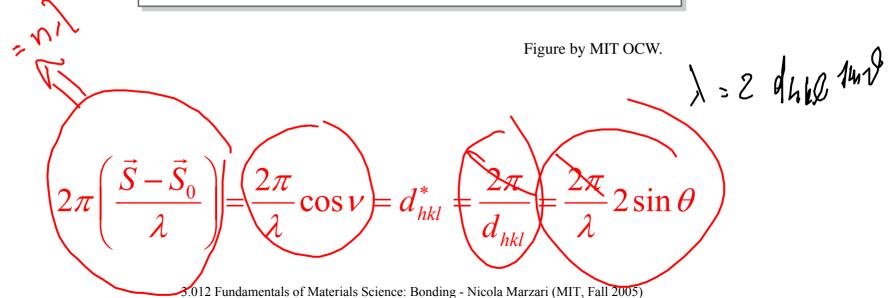


Figure by MIT OCW.

$$n\lambda = d_{hkl} 2 \sin \theta$$

Equivalence to Laue condition





Powder diffraction (I)

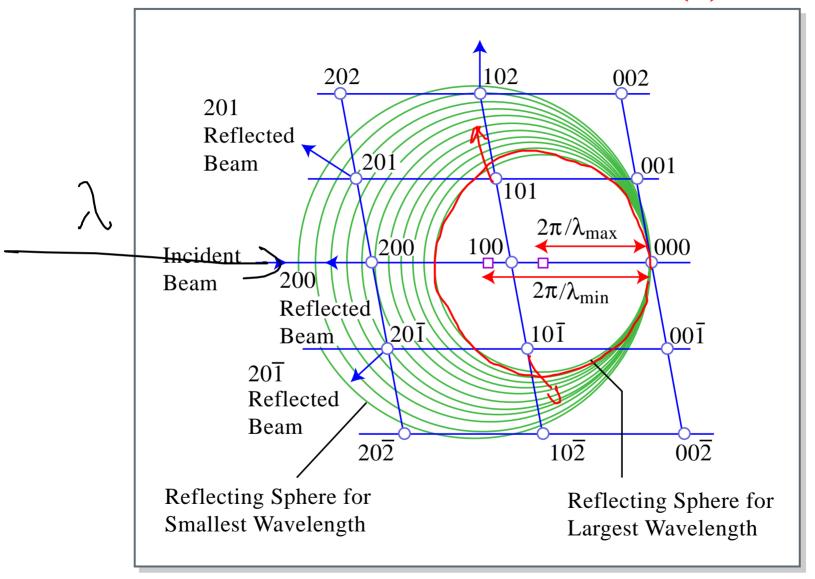


Figure by MIT OCW.

Powder diffraction (II)

Image removed for copyright reasons. Please see the diagram at http://capsicum.me.utexas.edu/ChE386K/html/powder_diffraction_3.htm.

Powder diffraction (III)

Diagrams of the Powder Method removed for copyright reasons.

See the images at http://www.matter.org.uk/diffraction/x-ray/powder_method.htm

X-ray filters

Image removed for copyright reasons.

Please see the graph at http://capsicum.me.utexas.edu/ChE386K/html/absorption_edge.htm.

Image removed for copyright reasons.

Please see the diagrams at http://capsicum.me.utexas.edu/ChE386K/html/filter.htm.

Debye-Scherrer camera

Photographs of a Debye-Scherrer camera removed for copyright reasons.

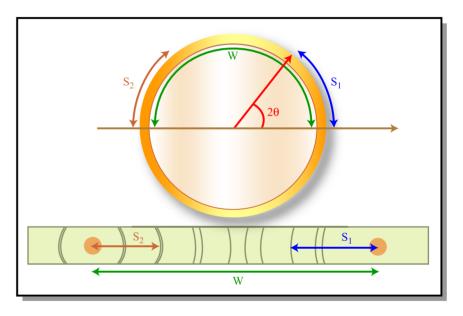


Figure by MIT OCW.

Interplanar spacings

$$\frac{1}{d^2} = \frac{h^2 + k^2 + l^2}{a^2}$$

Tetragonal:

$$\frac{1}{d^2} = \frac{h^2 + k^2}{a^2} + \frac{l^2}{c^2}$$

Hexagonal:

$$\frac{1}{d^2} = \frac{4}{3} \left(\frac{h^2 + hk + k^2}{a^2} \right) + \frac{l^2}{c^2}$$

Rhombohedral:

$$\frac{1}{d^2} = \frac{(h^2 + k^2 + l^2) \sin^2 \alpha + 2(hk + kl + hl)(\cos^2 \alpha - \cos \alpha)}{a^2(1 - 3\cos^2 \alpha + 2\cos^3 \alpha)}$$

Orthorhombic:

$$\frac{1}{d^2} = \frac{h^2}{a^2} + \frac{k^2}{b^2} + \frac{l^2}{c^2}$$

Monoclinic:
$$\frac{1}{d^2} = \frac{1}{\sin^2 \beta} \left(\frac{h^2}{a^2} + \frac{k^2 \sin^2 \beta}{b^2} + \frac{l^2}{c^2} - \frac{2hl \cos \beta}{ac} \right)$$

Triclinic:
$$\frac{1}{d^2} = \frac{1}{V^2} \left(S_{11}h^2 + S_{22}k^2 + S_{33}l^2 + 2S_{12}hk + 2S_{23}kl + 2S_{13}hl \right)$$

In the equation for triclinic crystals

$$V = abc\sqrt{1 - \cos^2\alpha - \cos^2\beta - \cos^2\gamma + 2\cos\alpha\cos\beta\cos\gamma}$$

$$S_{11} = b^2 c^2 \sin^2 \alpha,$$

$$S_{22} = a^2c^2\sin^2\beta,$$

$$S_{33} = a^2b^2\sin^2\gamma,$$

$$S_{12} = abc^2(\cos\alpha\cos\beta - \cos\gamma),$$

$$S_{23} = a^2 b c (\cos \beta \cos \gamma - \cos \alpha),$$

$$S_{13} = ab^2c(\cos\gamma\cos\alpha - \cos\beta).$$

Cubic:

$$d_{hkl}^2 = \frac{a^2}{h^2 + k^2 + l^2}$$

Debye-Scherrer camera

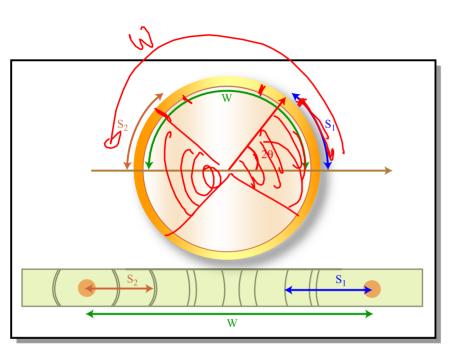


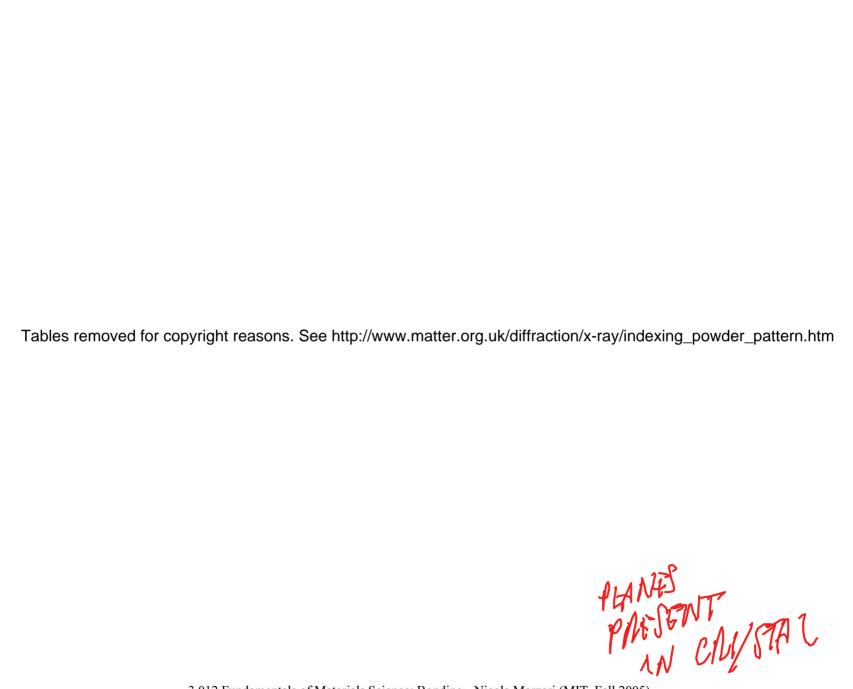
Figure by MIT OCW.

$$29 = 2\frac{S_1}{10}$$
 RADANS

$$\lambda = d_{hkl} 2 \sin \theta$$
Cubic:
$$d_{hkl}^2 = \frac{a^2}{h^2 + k^2 + l^2}$$

$$\sin \theta = \frac{\lambda}{2 d_h h_b}$$

$$\sin^2 \theta = \frac{\lambda^2}{4 q^2} \frac{h_h^2 k^2 l^2}{q^2}$$



Systematic absences

Image removed for copyright reasons.

Please see the table at http://capsicum.me.utexas.edu/ChE386K/html/systematic_absences.htm.

Effects of symmetry on diffraction

Images removed for copyright reasons.

Please see the images at http://capsicum.me.utexas.edu/ChE386K/html/diffraction_symmetry1.htm.

Structure Factor

$$\mathbf{F}(hkl) = \sum_{n=1}^{N} \int_{n} e^{2\pi i \left(hx_n + ky_n + lz_n\right)}$$

$$\text{Image removed for copyright reasons.}$$

$$\text{Please see the graph at http://capsicum.me.utexas.edu/ChE386K/html/scattering_factor_curve.htm}$$

11 F 11 = INTENSITY

Friedel's law

• The diffraction pattern is always centrosymmetric, even if the crystal is not centrosymmetric

Point symmetry + inversion = Laue

Image removed for copyright reasons.

Please see the table at http://capsicum.me.utexas.edu/ChE386K/html/diffraction_symmetry2.htm.

Back-reflection and transmission Laue

Diagrams of the Laue Method removed for copyright reasons. See the images at http://www.matter.org.uk/diffraction/x-ray/laue_method.htm.