

“Introduction to the Stochastic  
modelling of reaction networks”

Day 1: The Gillespie algorithm  
(intro)

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# Deterministic vs. stochastic

- **Deterministic:** every set of variable states is uniquely determined by parameters in the model and by sets of previous states of these variables; therefore, a deterministic model always performs the same way for a given set of initial conditions.
- **Stochastic trajectory:** randomness is present, and every trajectory can be different from the others.
- **Stochastic:** calculate deterministically some variable of the stochastic distribution

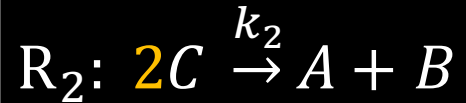
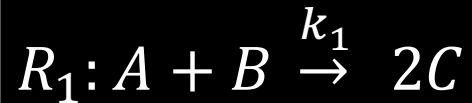
# Notation: Describing a chemical system over time

- The total volume of the system,  $V$ .
- A series of chemical species:  $S_i$  ( $i = 1, 2, \dots, N$ ),  $N$  total number of chemical species.
- The quantities of those elements at time  $t$ :  $X_i$  ( $i = 1, 2, \dots, N$ )
- The possible reactions:  $R_\mu$  ( $\mu = 1, 2, \dots, M$ )  $M$ , total number of reactions.
- Parameter for the evolution of the system:
  - $k_\mu$  (deterministic)
  - $c_\mu$  (stochastic)

# The deterministic approach to chemical reactions

- Back to high-school: A simple equilibrium

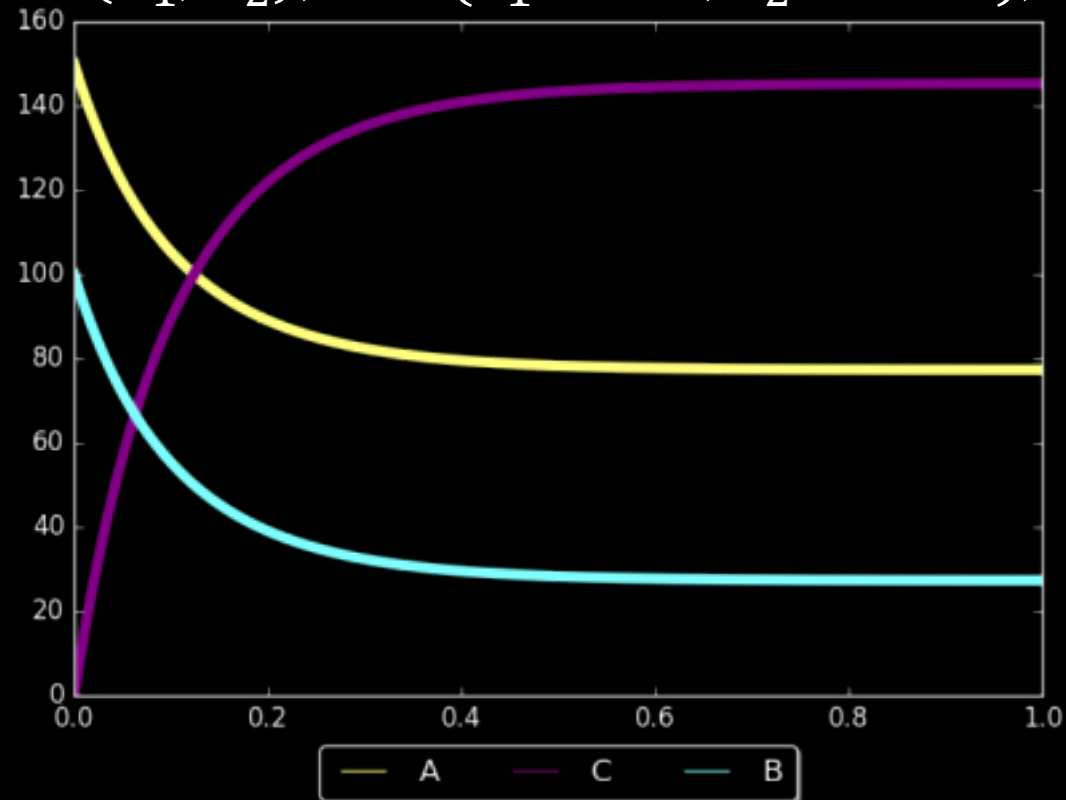
$S_i = (A, B, C); X_i = (150, 100, 0); R = (R_1, R_2); k = (k_1 = 0.5, k_2 = 0.05); V=1$



$$v_1 = \frac{dA}{dt} = \frac{dB}{dt} = -k_1[A][B]$$

$$v_2 = \frac{1}{2} \frac{dC}{dt} = -k_2[C]^2$$

*Only in the equilibrium:  $v_1 = v_2$*

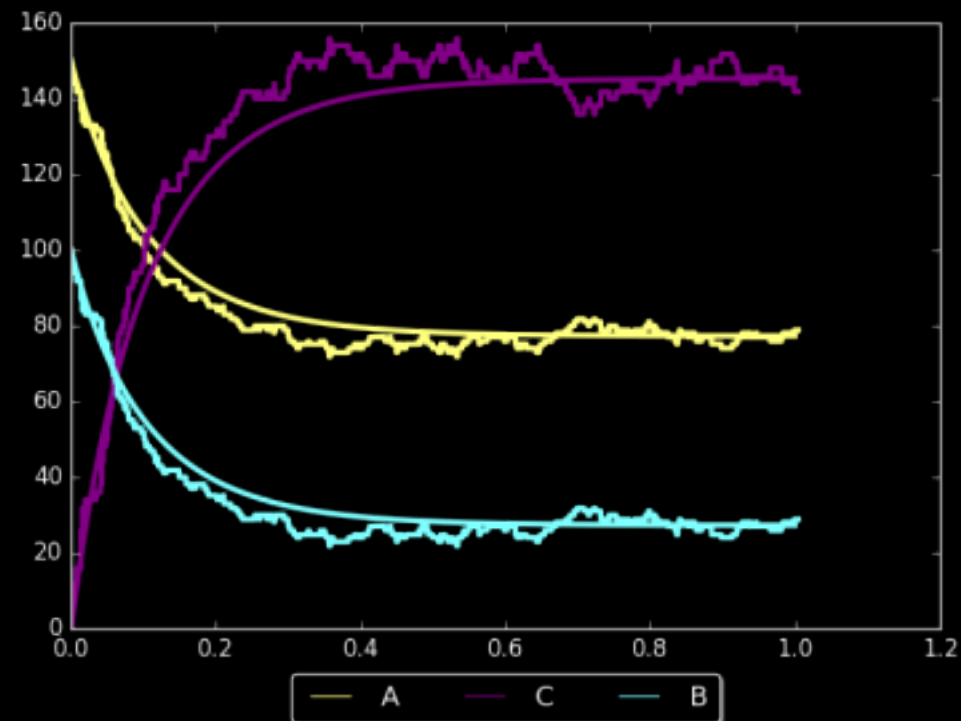


# A stochastic approach to the chemical system

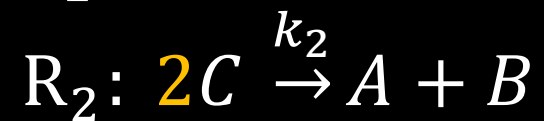
- Time evolution of a chemical system is not continuous, chemical species quantities can only change by discrete integer amounts.
- It is not deterministic if we take into account only the quantities of the chemical species. It is impossible to predict the exact molecular population levels at a future time.
- A better approach is to consider the process as a random walk

# Stochastic model assumptions

- The volume  $V$  is fixed
- The molecules are in thermal equilibrium, collisions occur in a random manner. The mixture is spatially homogeneous.
- A reaction happens when two interacting molecules collide in an "appropriate way".



Possible  
combinations



$$\frac{dP_{R_1}}{dt} = \frac{a_1}{dt} = c_1 n_A n_B$$

$$\frac{dP_{R_2}}{dt} = \frac{a_2}{dt} = \frac{c_2 n_C (n_C - 1)}{2}$$

$$v_1 = -k_1 \frac{n_A n_B}{V^2}$$

$$v_2 = -k_2 \frac{n_C^2}{V^2}$$

$$c_1 = \frac{k_1}{V^2}$$

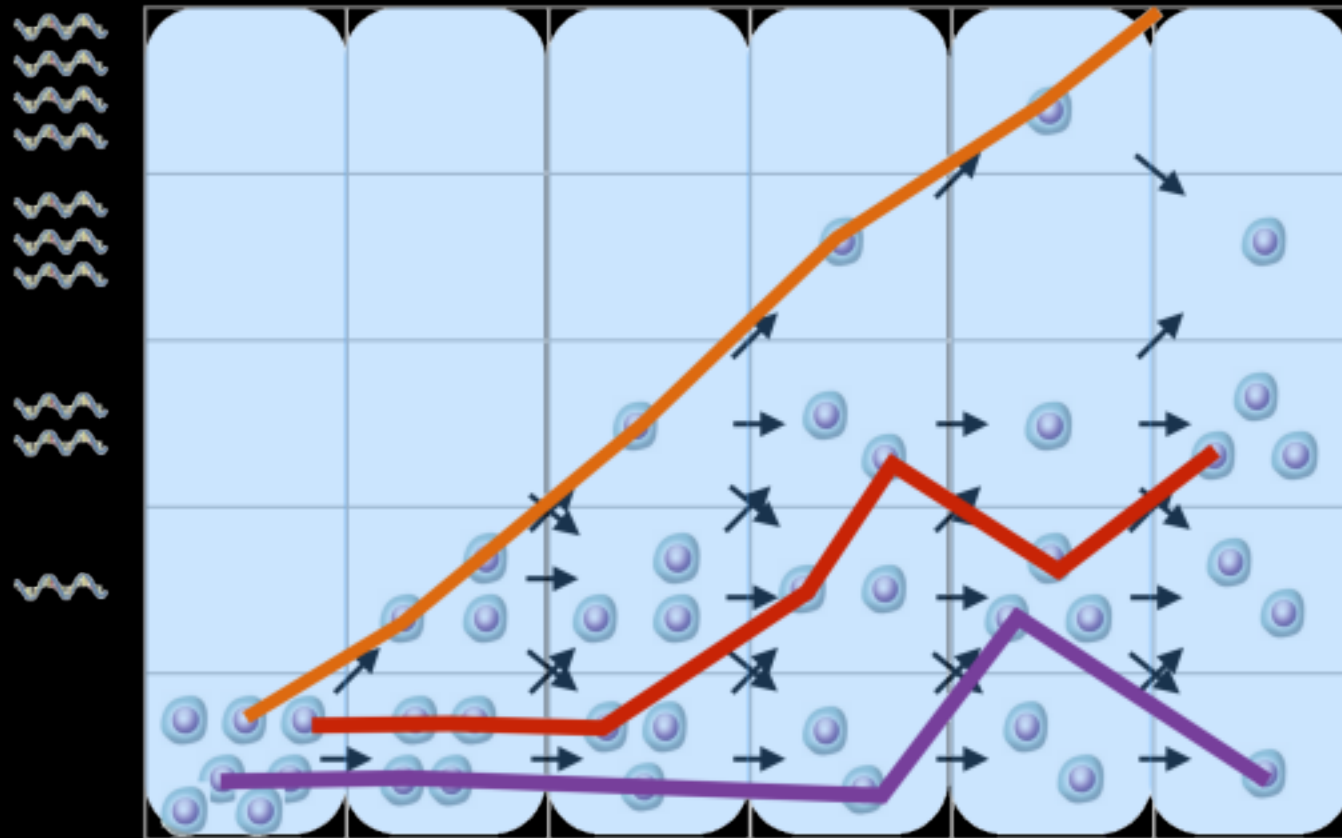
$$c_2 = 2 \frac{k_2}{V^2}$$

Units?

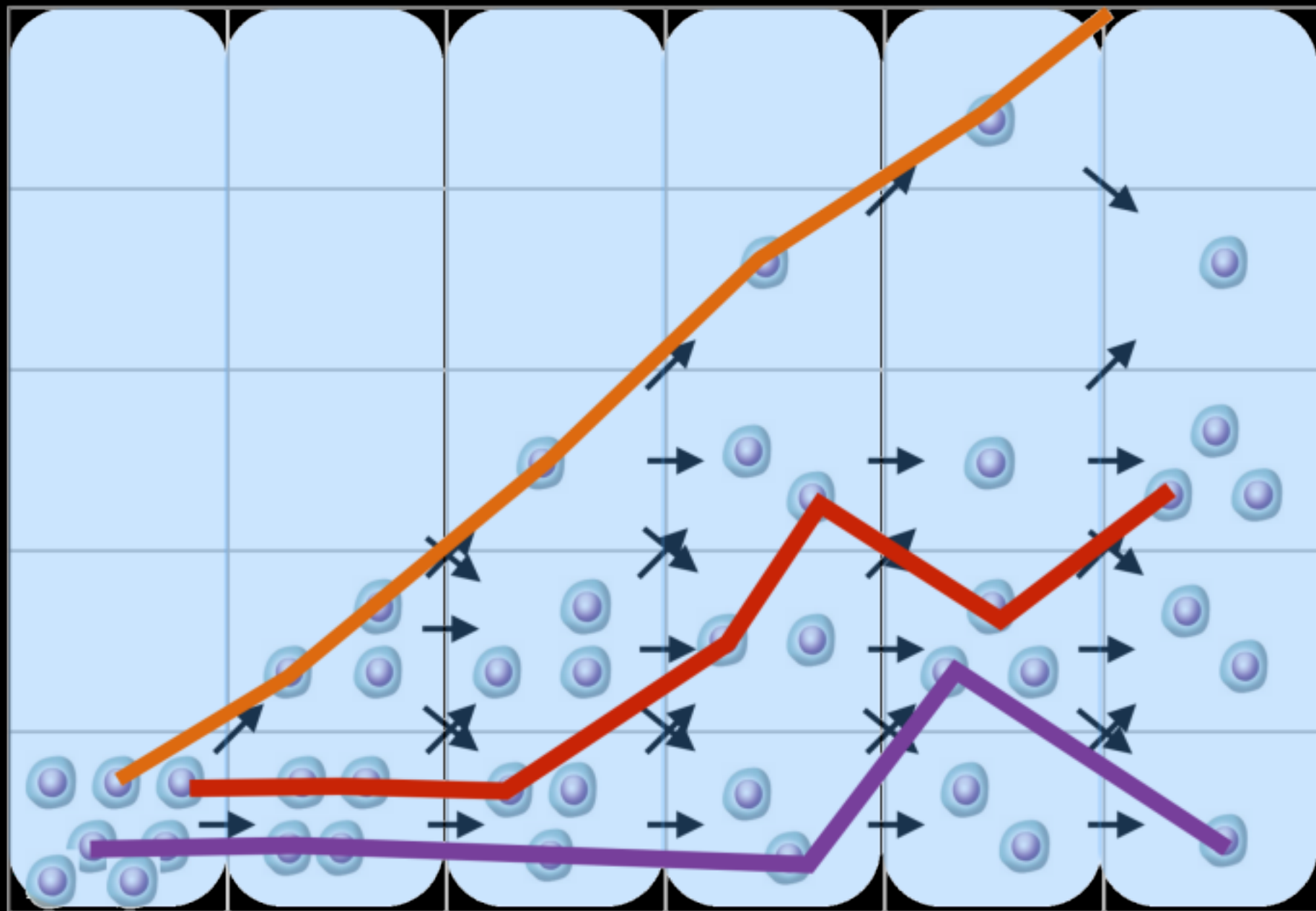
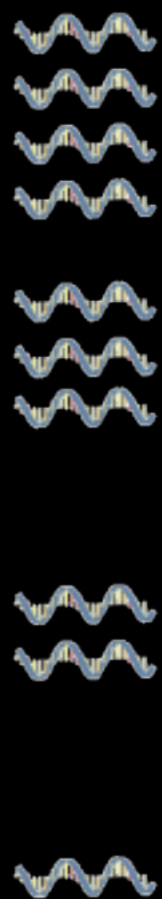
Volume effect?

# Master equation vs. stochastic simulation algorithm

- Master equation: Probability distribution over time
- Stochastic simulation: Calculates a particular trajectory







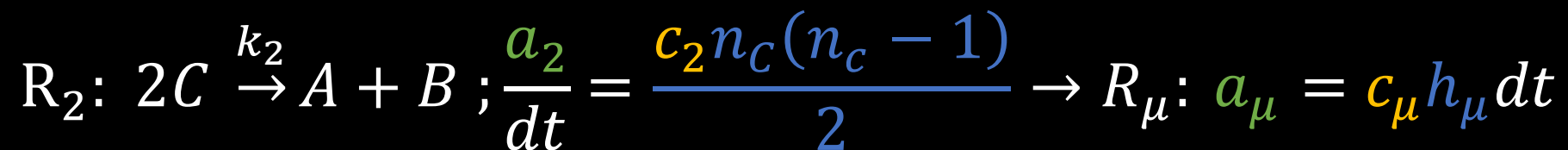
# The Gillespie algorithm

- Basically, we can simulate the evolution of a chemical system by repeatedly asking this two following questions:
  - *Which reaction will happen next?*
  - *When will it happen?*
- How to find the response to them?
  - This is the main goal of this session



# The Gillespie algorithm: First, Refreshment

- The total volume of the system,  $V$ .
- A series of chemical species:  $S_i$  ( $i = 1, 2, \dots, N$ ),  $N$  total number of chemical species.
- The quantities of those elements at time  $t$ :  $X_i$  ( $i = 1, 2, \dots, N$ )
- The possible reactions:  $R_\mu$  ( $\mu = 1, 2, \dots, M$ )  $M$ , total number of reactions.
- Parameter for the evolution of the system:  $c_\mu$  (stochastic)



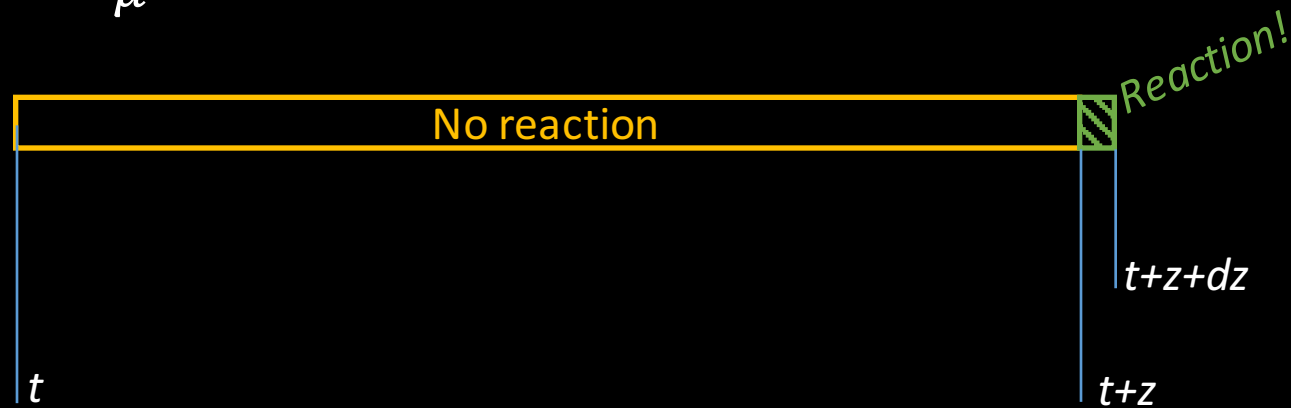
# The Gillespie algorithm:

$$P(z, \mu)dz = P_0(z)a_\mu dz$$

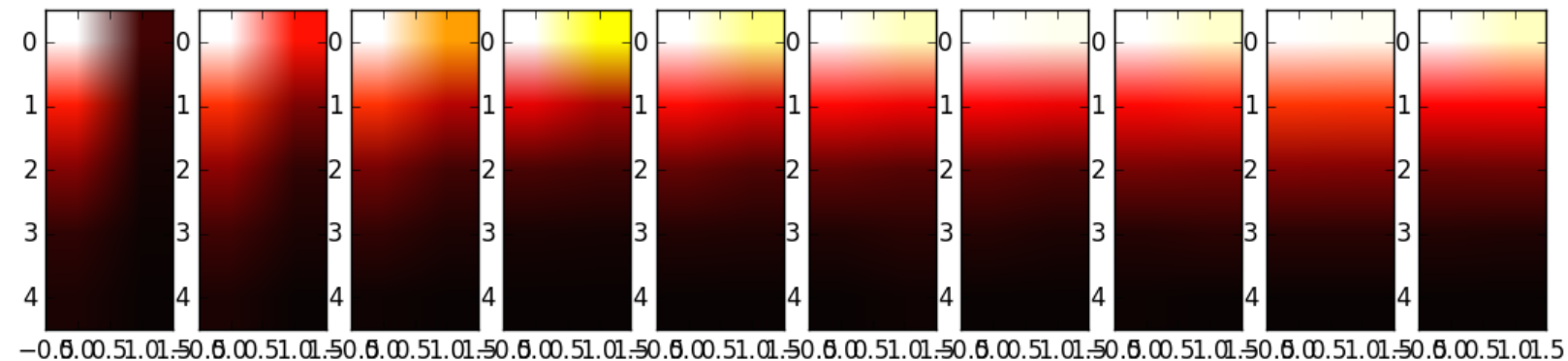
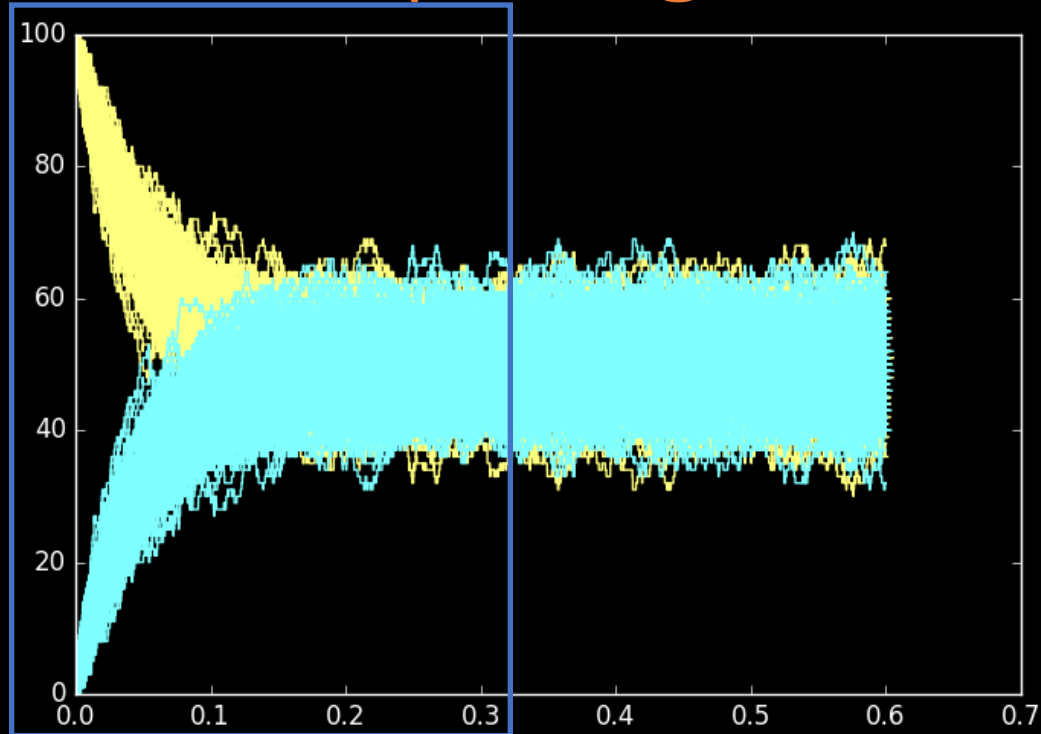
$P(z, \mu)dz$  : probability that no reaction happens in the interval  $(t, t+z)$ , and THE NEXT reaction happens in the interval  $(t+z, t+z+dz)$  and it's  $R_\mu$ .

$P_0(z)$  : probability that no reaction happens in interval  $(t, t+z)$

$a_\mu dz$  : probability that THE NEXT reaction happens in the interval  $(t+z, t+z+dz)$  and it's  $R_\mu$ .



# The Gillespie algorithm: what is $P(z, \mu)dz$ ?



# The Gillespie algorithm

Probability  
of no reaction

$$P(z, \mu)dz = P_0(z)a_\mu dz$$

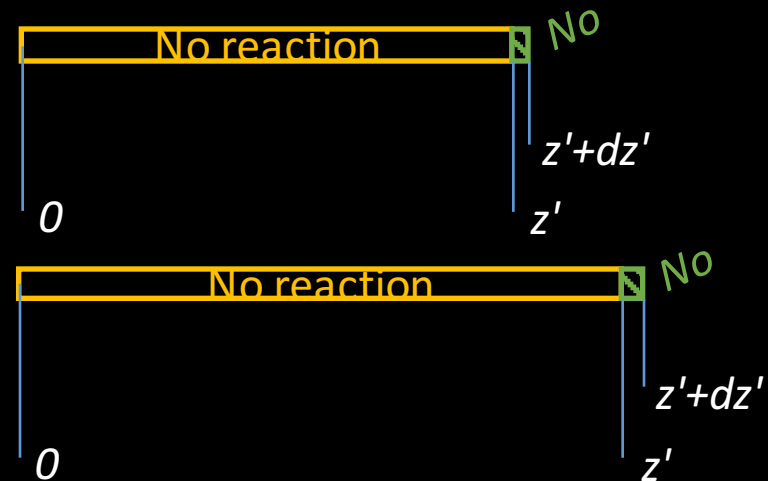
$$P_0(z' + dz') = P_0(z') \left[ 1 - \sum_{m=1}^M a_m dz' \right] = P_0(z') - P_0(z')a_0 dz$$

$$P_0(z' + dz') - P_0(z') = dP_0(z') = -P_0(z')a_0 dz$$

$$\frac{dP_0(z')}{P_0(z')} = -a_0 dz' \rightarrow \ln(P_0(z')) = a_0 z' \rightarrow P_0(z) = e^{-a_0 z}$$

Substitute in the first equation:  $P(z, \mu)dz = e^{-a_0 z} a_\mu dz$

$$P(z, \mu) = e^{-a_0 z} a_\mu$$



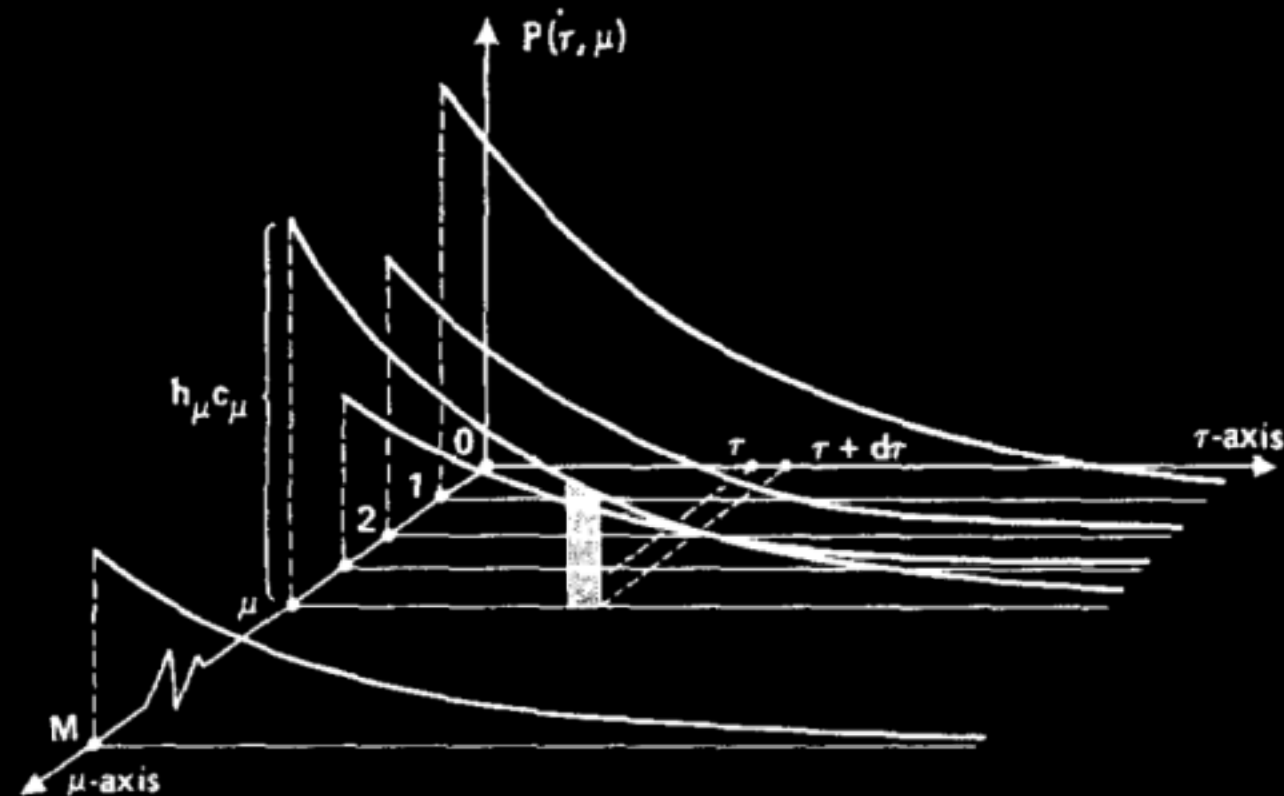
# The Gillespie algorithm

$P(z, \mu) = e^{-a_0 z} a_\mu$  is a probability density function

What does it describe?

Why is it an exponential?

Is it useful for our purpose?



# The Gillespie algorithm

Another way to think about it...  $P(z, \mu) = a_\mu e^{-a_0 z}$

$$P(z, \mu) = P_{any}(z)P(\mu|z)$$

Probability of any reaction to happen during  $z$  \* that reaction being  $\mu$

$$P_{any}(z) = \sum_{\mu=1}^M a_\mu e^{-a_0 z} = a_0 e^{-a_0 z}$$

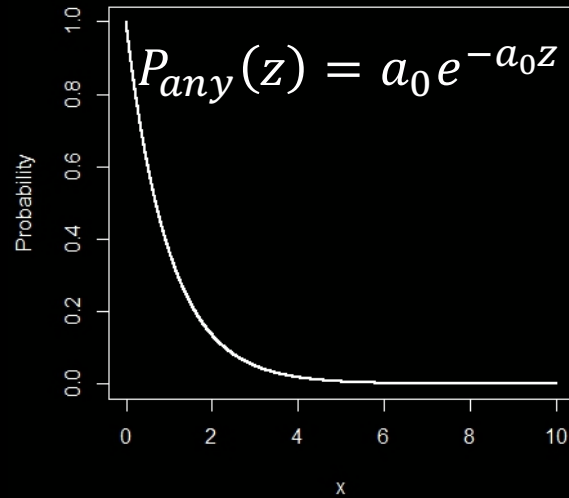
$$P(\mu|z) = \frac{a_\mu}{a_0}$$

Two independent functions describe the probabilities of our questions  
How do these pdfs look? What information can we extract from them?  
How can we use pairs of  $(z, \mu)$  that satisfy those pdfs?

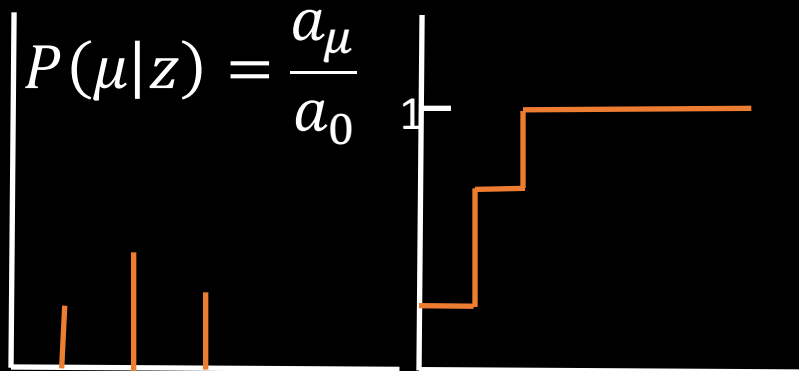
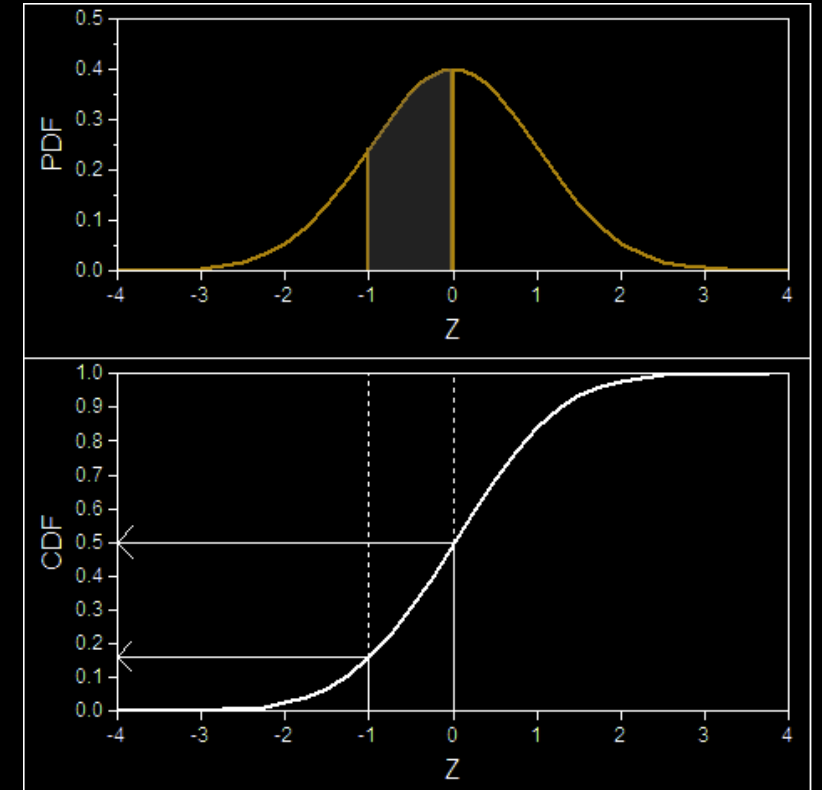
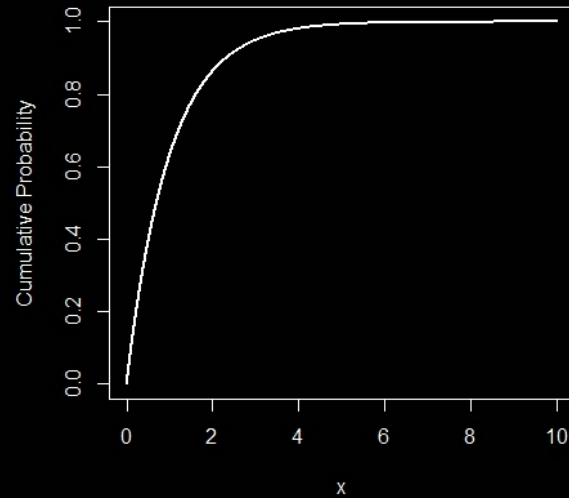


# How to sample random values from a PDF

PDF for Exponential Distribution (rate = 1)



CDF for Exponential Distribution (rate = 1)



Sampling doing  $F(x) = \text{random number}$

# How to sample values from a PDF

$$F(z) = \int_0^z P_{any}(z) dz = \int_0^z a_0 e^{-a_0 z} dz = 1 - e^{-a_0 z}$$

$$F(z) = 1 - e^{-a_0 z} \rightarrow \ln(1 - F(z)) = -a_0 z$$

$$z = \frac{1}{A} \ln\left(\frac{1}{1 - F(z)}\right)$$

$$z(r) = \frac{-1}{A} \ln(r)$$

Answer 1: When will the next reaction occur?

# How to sample values from a PDF

$$F(z) = \int_0^z P_{any}(z) dz = \int_0^z a_0 e^{-a_0 z} dz = 1 - e^{-a_0 z}$$

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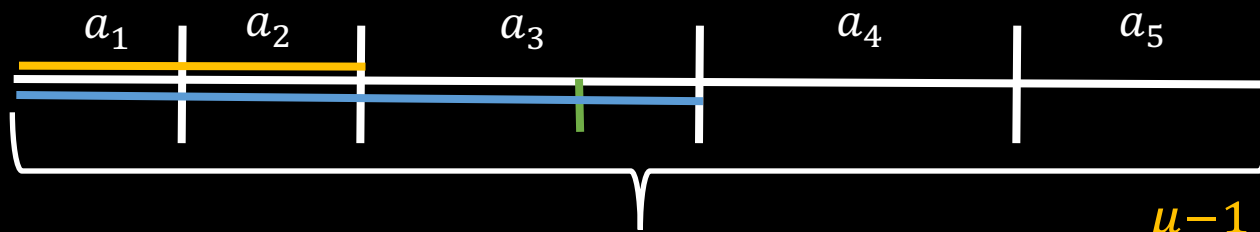
Answer 1: When will the next reaction occur?



# How to sample values from a PDF

$$P(\mu|z) = \frac{a_\mu}{a_0}$$

- How to think about it in this discrete case?



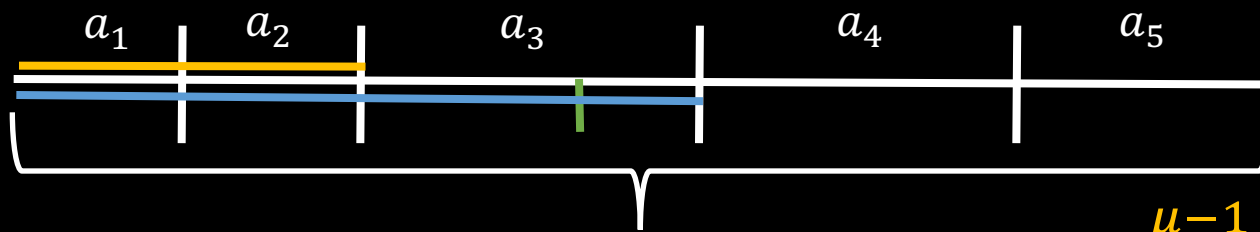
$\mu$  is the integer for which:  $\sum_{m=1}^{\mu-1} a_m < a_0 r \leq \sum_{m=1}^{\mu} a_m$

Answer 2: What reaction will it be?

# How to sample values from a PDF

$$P(\mu|z) = \frac{a_\mu}{a_0}$$

- How to think about it in this discrete case?



$\mu$  is the integer for which:  $\sum_{m=1}^{\mu-1} a_m < a_0 r \leq \sum_{m=1}^{\mu} a_m$

Answer 2: What reaction will it be?



# Possible projects

- Try it on your own data
- Build master equation
- Think of an example and compare stochastic and deterministic results
- How does  $\tau$  change with time in different situations?
- Suggestions?