# A Short and Descriptive Title

#### **Your Name Here**

Sigurðsson Research Group A 6-Months Progress Report December, 2014



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## **Contents**

A١	bstract	2			
Al	bbreviations	3			
1	Introduction	4			
2 Results and Discussion					
3	Conclusions and Outlook	6			
4	Experiments	7			
	4.1 General	7			
	4.2 1,2,4,5-Tetra- <i>tert</i> butylthiobenzene (1)	8			
Bi	ibliography	9			
Li	ist of Figures	10			
Li	ist of Schemes				

## Abstract

Write your abstract here.

#### **Abbreviations**

DCM dichloromethane

DMF *N*, *N*-dimethyformamide DNA deoxyribonucleic acid

EPR electron paramagnetic resonance

ESI-MS electron-spray ionization mass-spectrometry

J coupling constant m/z mass—charge ratio

NMR nuclear magnetic resonance

ppm parts per million RNA ribonucleic acid

rt room temperature (ambient)

TEA triethylamine
THF tetrahydrofurane
TFA trifluoroacetic acid

TLC thin layer chromatography

 $\delta$  chemical shift

s singlet
d doublet
t triplet
q quartet
m multiplet

### 1 Introduction

Write your introduction here.

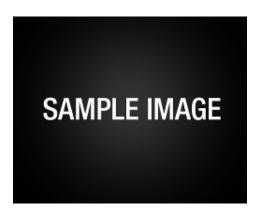


Figure 1. Caption text.

Figure 1 shows a sample image. The DNA double helix was first reported by Watson and Crick in 1953.[1] It also possible to put the citation in superscript, like this.<sup>[1]</sup>

## 2 Results and Discussion

Write your syntheses and outcomes here.

Scheme 1. A scheme with no compound numbers.

Scheme 1 shows a sample reaction scheme.

## 3 Conclusions and Outlook

Write your conclusions, and your outlook here.

#### 4 Experiments

#### 4.1 General

Chemicals were purchased primarily from Sigma-Aldrich Chemical Company and Acros, Belgium, and were used without further purification. Triethylamine was purchased anhydrous. TLC was carried out using glass plates pre-coated with silica gel (F254, Silicycle SiliPlate 60 Å). Visualisation was by UV light, and  $I_2$  staining, respectively. Silica gel was purchased from Silicycle, and used for medium pressure chromatography ("flash"-chromatography).

 $^{1}$ H and  $^{13}$ C NMR spectra were recorded at the frequencies stated, using deuterated chloroform as internal standard ( $\delta$  = 7.26 ppm for  $^{1}$ H and  $\delta$  = 77.0 ppm for  $^{13}$ C NMR). 400 MHz spectra were recorded on a Bruker Advance 400 spectrometer. All coupling constants were measured in Hertz.

All moisture sensitive reactions were carried out in flame-dried glassware using argon from standard BOC industrial cylinders, dried through an activated silica column. Diethyl ether for moisture-sensitive reactions was used freshly distilled over Na under argon atmosphere. Concentrations of *n*Bu Li in hexane were determined by titration using diphenylacetic acid.

#### 4.2 1,2,4,5-Tetra-*tert*.-butylthiobenzene (1)

To a solution of 2-methyl-2-propanethiol (94 mL, 0.84 mol) in DMF (150 mL) was added small pieces of sodium (19.37 g, 0.84 mol) at 0 °C. The mixture was allowed to reach ambient temperature and was stirred overnight. 1,2,4,5-Tetrachlorobenzene (30.3 g, 140.6 mmol) was added at once, and the resulting mixture was heated to 90 °C. As soon as the reaction mixture darkened and steam started to develop, the heating source was removed. As soon as the exothermic reaction had finished, the mixture was heated to 120 °Cfor 24 h. After being cooled to ambient temperature, the reaction mixture was poured over ice. The precipitate was removed by filtration, washed with water, and dried to give 30 g (49.5%) of the product as an off-white powder.

Notebook reference: MKI-107.

<sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>)  $\delta$  = 7.94 (2 H, s),1.36 ppm (2 H, s).

<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  = 144.78, 139.36, 48.13, 31.33 ppm.

# References

[1] J. D. Watson, F. H. Crick, "Molecular structure of nucleic acids", *Nature* **1953**, *171*, 737–738.

List	of	Fig	ures
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1	A scheme with no compound numbers	5
-	Tree with the compound numbers, the tree tree tree tree tree tree tree	•