

Tabel Periodik

HTML

```
<!DOCTYPE html>
<html lang="en" >
<head>
  <meta charset="UTF-8">
  <title>Tabel Periodik</title>
  <link rel='stylesheet'
href='//maxcdn.bootstrapcdn.com/bootstrap/3.3.5/css/bootstrap.min.css'><link
rel="stylesheet" href="./style.css">

</head>
<body>
<!-- partial:index.partial.html -->
<div id="status" style="display:none;position:fixed;background:rgba(0, 0, 0,
0.75);color:white;top:0;left:0;width:100%;height:auto;z-index:5;text-
align:center;"></div>
<table>
  <caption id="title">
    TABEL PERIODIK UNSUR-UNSUR KIMIA</caption>
  <thead>
    <tr class="gol">
      <th class="indiv">Golongan
        <br/>Periodik</th>
      <th class="gol">
        <div class="content">1A
          <br/>Alkali</div>
      </th>
      <th class="gol">
        <div class="content">2A
          <br/>Alkali Tanah</div>
      </th>
      <th class="gol">
        <div class="content">3B</div>
      </th>
      <th class="gol">
        <div class="content">4B</div>
      </th>
      <th class="gol">
        <div class="content">5B</div>
      </th>
      <th class="gol">
        <div class="content">6B</div>
      </th>
      <th class="gol">
        <div class="content">7B</div>
      </th>
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<th class="gol" colspan="3">8B</th>
<th class="gol">
  <div class="content">1B</div>
</th>
<th class="gol">
  <div class="content">2B</div>
</th>
<th class="gol">
  <div class="content">3A</div>
</th>
<th class="gol">
  <div class="content">4A</div>
</th>
<th class="gol">
  <div class="content">5A</div>
</th>
<th class="gol">
  <div class="content">6A</div>
</th>
<th class="gol">
  <div class="content">7A
    <br/>Halogen</div>
</th>
<th class="gol">
  <div class="content">8A
    <br/>Gas Mulia</div>
</th>
</tr>
</thead>
<tbody>
<tr>
  <th class="per">
    <div class="content-per">1</div>
  </th>
  <td class="g">
    <div class="content-tp">
      <div class="na">1</div>
      <div class="ma">1.008</div>
      <br/>
      <div class="lu">H</div>
      <div class="nu">Hidrogen</div>
    </div>
  </td>
  <td colspan="16"></td>
  <td class="gm">
    <div class="content-tp">
      <div class="na">2</div>
      <div class="ma">4.003</div>
      <br/>

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        <div class="lu">He</div>
        <div class="nu">Helium</div>
    </div>
</td>
</tr>
<tr>
    <th class="per">
        <div class="content-per">2</div>
    </th>
    <td class="d">
        <div class="content-tp">
            <div class="na">3</div>
            <div class="ma">6.939</div>
            <br/>
            <div class="lu">Li</div>
            <div class="nu">Litium</div>
        </div>
    </td>
    <td class="c">
        <div class="content-tp">
            <div class="na">4</div>
            <div class="ma">9.012</div>
            <br/>
            <div class="lu">Be</div>
            <div class="nu">Berilium</div>
        </div>
    </td>
    <td colspan="10"></td>
    <td class="m">
        <div class="content-tp">
            <div class="na">5</div>
            <div class="ma">10.811</div>
            <br/>
            <div class="lu">B</div>
            <div class="nu">Boron</div>
        </div>
    </td>
    <td class="g">
        <div class="content-tp">
            <div class="na">6</div>
            <div class="ma">12.011</div>
            <br/>
            <div class="lu">C</div>
            <div class="nu">Karbon</div>
        </div>
    </td>
    <td class="g">
        <div class="content-tp">
            <div class="na">7</div>

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        <div class="ma">14.007</div>
        <br/>
        <div class="lu">N</div>
        <div class="nu">Nitrogen</div>
    </div>
</td>
<td class="g">
    <div class="content-tp">
        <div class="na">8</div>
        <div class="ma">15.999</div>
        <br/>
        <div class="lu">O</div>
        <div class="nu">Oksigen</div>
    </div>
</td>
<td class="h">
    <div class="content-tp">
        <div class="na">9</div>
        <div class="ma">18.998</div>
        <br/>
        <div class="lu">F</div>
        <div class="nu">Fluor</div>
    </div>
</td>
<td class="gm">
    <div class="content-tp">
        <div class="na">10</div>
        <div class="ma">20.18</div>
        <br/>
        <div class="lu">Ne</div>
        <div class="nu">Neon</div>
    </div>
</td>
</tr>
<tr>
    <th class="per">
        <div class="content-per">3</div>
    </th>
    <td class="d">
        <div class="content-tp">
            <div class="na">11</div>
            <div class="ma">22.989</div>
            <br/>
            <div class="lu">Na</div>
            <div class="nu">Natrium</div>
        </div>
    </td>
    <td class="c">
        <div class="content-tp">

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        <div class="na">12</div>
        <div class="ma">24.305</div>
        <br/>
        <div class="lu">Mg</div>
        <div class="nu">Magnesium</div>
    </div>
</td>
<td colspan="10"></td>
<td class="c">
    <div class="content-tp">
        <div class="na">13</div>
        <div class="ma">26.981</div>
        <br/>
        <div class="lu">Al</div>
        <div class="nu">Aluminium</div>
    </div>
</td>
<td class="m">
    <div class="content-tp">
        <div class="na">14</div>
        <div class="ma">28.086</div>
        <br/>
        <div class="lu">Si</div>
        <div class="nu">Silikon</div>
    </div>
</td>
<td class="g">
    <div class="content-tp">
        <div class="na">15</div>
        <div class="ma">30.974</div>
        <br/>
        <div class="lu">P</div>
        <div class="nu">Fosfor</div>
    </div>
</td>
<td class="g">
    <div class="content-tp">
        <div class="na">16</div>
        <div class="ma">32.066</div>
        <br/>
        <div class="lu">S</div>
        <div class="nu">Belerang</div>
    </div>
</td>
<td class="h">
    <div class="content-tp">
        <div class="na">17</div>
        <div class="ma">35.453</div>
        <br/>

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        <div class="lu">Cl</div>
        <div class="nu">Klor</div>
    </div>
</td>
<td class="gm">
    <div class="content-tp">
        <div class="na">18</div>
        <div class="ma">39.948</div>
        <br/>
        <div class="lu">Ar</div>
        <div class="nu">Argon</div>
    </div>
</td>
</tr>
<tr>
    <th class="per">
        <div class="content-per">4</div>
    </th>
    <td class="d">
        <div class="content-tp">
            <div class="na">19</div>
            <div class="ma">39.098</div>
            <br/>
            <div class="lu">K</div>
            <div class="nu">Kalium</div>
        </div>
    </td>
    <td class="c">
        <div class="content-tp">
            <div class="na">20</div>
            <div class="ma">40.076</div>
            <br/>
            <div class="lu">Ca</div>
            <div class="nu">Kalsium</div>
        </div>
    </td>
    <td class="l">
        <div class="content-tp">
            <div class="na">21</div>
            <div class="ma">44.956</div>
            <br/>
            <div class="lu">Sc</div>
            <div class="nu">Skandium</div>
        </div>
    </td>
    <td class="l">
        <div class="content-tp">
            <div class="na">22</div>
            <div class="ma">47.88</div>

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        <br/>
        <div class="lu">Ti</div>
        <div class="nu">Titanium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">23</div>
        <div class="ma">50.942</div>
        <br/>
        <div class="lu">V</div>
        <div class="nu">Vanadium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">24</div>
        <div class="ma">51.996</div>
        <br/>
        <div class="lu">Cr</div>
        <div class="nu">Kromium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">25</div>
        <div class="ma">54.938</div>
        <br/>
        <div class="lu">Mn</div>
        <div class="nu">Mangan</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">26</div>
        <div class="ma">55.847</div>
        <br/>
        <div class="lu">Fe</div>
        <div class="nu">Besi</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">27</div>
        <div class="ma">58.933</div>
        <br/>
        <div class="lu">Co</div>
        <div class="nu">Kobalt</div>
    </div>

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</td>
<td class="l">
  <div class="content-tp">
    <div class="na">28</div>
    <div class="ma">58.69</div>
    <br/>
    <div class="lu">Ni</div>
    <div class="nu">Nikel</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">29</div>
    <div class="ma">63.546</div>
    <br/>
    <div class="lu">Cu</div>
    <div class="nu">Tembaga</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">30</div>
    <div class="ma">65.39</div>
    <br/>
    <div class="lu">Zn</div>
    <div class="nu">Seng</div>
  </div>
</td>
<td class="c">
  <div class="content-tp">
    <div class="na">31</div>
    <div class="ma">69.723</div>
    <br/>
    <div class="lu">Ga</div>
    <div class="nu">Galium</div>
  </div>
</td>
<td class="m">
  <div class="content-tp">
    <div class="na">32</div>
    <div class="ma">72.61</div>
    <br/>
    <div class="lu">Ge</div>
    <div class="nu">Germanium</div>
  </div>
</td>
<td class="m">
  <div class="content-tp">
    <div class="na">33</div>

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        <div class="ma">74.922</div>
        <br/>
        <div class="lu">As</div>
        <div class="nu">Arsen</div>
    </div>
</td>
<td class="g">
    <div class="content-tp">
        <div class="na">34</div>
        <div class="ma">78.96</div>
        <br/>
        <div class="lu">Se</div>
        <div class="nu">Selenium</div>
    </div>
</td>
<td class="h">
    <div class="content-tp">
        <div class="na">35</div>
        <div class="ma">79.904</div>
        <br/>
        <div class="lu">Br</div>
        <div class="nu">Bromin</div>
    </div>
</td>
<td class="gm">
    <div class="content-tp">
        <div class="na">36</div>
        <div class="ma">83.8</div>
        <br/>
        <div class="lu">Kr</div>
        <div class="nu">Krypton</div>
    </div>
</td>
</tr>
<tr>
    <th class="per">5</th>
    <td class="d">
        <div class="content-tp">
            <div class="na">37</div>
            <div class="ma">85.4678</div>
            <br/>
            <div class="lu">Rb</div>
            <div class="nu">Rubidium</div>
        </div>
</td>
    <td class="c">
        <div class="content-tp">
            <div class="na">38</div>
            <div class="ma">87.62</div>

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        <br/>
        <div class="lu">Sr</div>
        <div class="nu">Stronsium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">39</div>
        <div class="ma">88.906</div>
        <br/>
        <div class="lu">Y</div>
        <div class="nu">Itrium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">40</div>
        <div class="ma">91.224</div>
        <br/>
        <div class="lu">Zr</div>
        <div class="nu">Zikronium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">41</div>
        <div class="ma">92.960</div>
        <br/>
        <div class="lu">Nb</div>
        <div class="nu">Niobium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">42</div>
        <div class="ma">95.94</div>
        <br/>
        <div class="lu">Mo</div>
        <div class="nu">Moilbdenum</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">43</div>
        <div class="ma">98 (0)</div>
        <br/>
        <div class="lu">Tc</div>
        <div class="nu">Teknesium</div>
    </div>

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</td>
<td class="l">
  <div class="content-tp">
    <div class="na">44</div>
    <div class="ma">101.07</div>
    <br/>
    <div class="lu">Ru</div>
    <div class="nu">Rutenium</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">45</div>
    <div class="ma">102.905</div>
    <br/>
    <div class="lu">Rd</div>
    <div class="nu">Rodium</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">46</div>
    <div class="ma">106.42</div>
    <br/>
    <div class="lu">Rb</div>
    <div class="nu">Rubidium</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">47</div>
    <div class="ma">107.868</div>
    <br/>
    <div class="lu">Ag</div>
    <div class="nu">Perak</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">48</div>
    <div class="ma">112.41</div>
    <br/>
    <div class="lu">Cd</div>
    <div class="nu">Kadmium</div>
  </div>
</td>
<td class="c">
  <div class="content-tp">
    <div class="na">49</div>
```

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        <div class="ma">114.82</div>
        <br/>
        <div class="lu">In</div>
        <div class="nu">Indium</div>
    </div>
</td>
<td class="c">
    <div class="content-tp">
        <div class="na">50</div>
        <div class="ma">118.71</div>
        <br/>
        <div class="lu">Sn</div>
        <div class="nu">Timah</div>
    </div>
</td>
<td class="m">
    <div class="content-tp">
        <div class="na">51</div>
        <div class="ma">121.76</div>
        <br/>
        <div class="lu">Sb</div>
        <div class="nu">Antimon</div>
    </div>
</td>
<td class="m">
    <div class="content-tp">
        <div class="na">52</div>
        <div class="ma">127.6</div>
        <br/>
        <div class="lu">Te</div>
        <div class="nu">Telurium</div>
    </div>
</td>
<td class="h">
    <div class="content-tp">
        <div class="na">53</div>
        <div class="ma">126.904</div>
        <br/>
        <div class="lu">I</div>
        <div class="nu">Yodium</div>
    </div>
</td>
<td class="gm">
    <div class="content-tp">
        <div class="na">54</div>
        <div class="ma">131.29</div>
        <br/>
        <div class="lu">Xe</div>
        <div class="nu">Xenon</div>

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        </div>
    </td>
</tr>
<tr>
    <th class="per">
        <div class="content-per">6</div>
    </th>
    <td class="d">
        <div class="content-tp">
            <div class="na">55</div>
            <div class="ma">132.905</div>
            <br/>
            <div class="lu">Cs</div>
            <div class="nu">Sesium</div>
        </div>
    </td>
    <td class="c">
        <div class="content-tp">
            <div class="na">56</div>
            <div class="ma">137.327</div>
            <br/>
            <div class="lu">Ba</div>
            <div class="nu">Barium</div>
        </div>
    </td>
    <td class="lan">
        <div class="content-tp">
            <div class="na">57-71</div>
            <div class="ma">138-174</div>
            <br/>
            <div class="nu">Rangkaian Lantanida</div>
            <br/>
            <div class="nu-desc">Nomor 57 sampai 71</div>
        </div>
    </td>
    <td class="l">
        <div class="content-tp">
            <div class="na">72</div>
            <div class="ma">178.49</div>
            <br/>
            <div class="lu">Hf</div>
            <div class="nu">Hafnium</div>
        </div>
    </td>
    <td class="l">
        <div class="content-tp">
            <div class="na">73</div>
            <div class="ma">180.947</div>
            <br/>

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```

        <div class="lu">Ta</div>
        <div class="nu">Tantalum</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">74</div>
        <div class="ma">137.327</div>
        <br/>
        <div class="lu">Ba</div>
        <div class="nu">Barium</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">75</div>
        <div class="ma">186.207</div>
        <br/>
        <div class="lu">Re</div>
        <div class="nu">Rений</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">76</div>
        <div class="ma">190.2</div>
        <br/>
        <div class="lu">Os</div>
        <div class="nu">Осмий</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">77</div>
        <div class="ma">192.22</div>
        <br/>
        <div class="lu">Ir</div>
        <div class="nu">Иридий</div>
    </div>
</td>
<td class="l">
    <div class="content-tp">
        <div class="na">78</div>
        <div class="ma">195.08</div>
        <br/>
        <div class="lu">Pt</div>
        <div class="nu">Платина</div>
    </div>
</td>

```

```

<td class="l">
  <div class="content-tp">
    <div class="na">79</div>
    <div class="ma">196.97</div>
    <br/>
    <div class="lu">Au</div>
    <div class="nu">Emas</div>
  </div>
</td>
<td class="l">
  <div class="content-tp">
    <div class="na">80</div>
    <div class="ma">200.59</div>
    <br/>
    <div class="lu">Hg</div>
    <div class="nu">Air Raksa</div>
  </div>
</td>
<td class="c">
  <div class="content-tp">
    <div class="na">81</div>
    <div class="ma">204.37</div>
    <br/>
    <div class="lu">Tl</div>
    <div class="nu">Talium</div>
  </div>
</td>
<td class="c">
  <div class="content-tp">
    <div class="na">82</div>
    <div class="ma">207.2</div>
    <br/>
    <div class="lu">Pb</div>
    <div class="nu">Timbal</div>
  </div>
</td>
<td class="c">
  <div class="content-tp">
    <div class="na">83</div>
    <div class="ma">208.98</div>
    <br/>
    <div class="lu">Bi</div>
    <div class="nu">Bismut</div>
  </div>
</td>
<td class="m">
  <div class="content-tp" title="Elemen ini termasuk dalam kategori
Radioaktif">
    <div class="na">84</div>

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>(209)</div>
<br/>


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```

</td>
<td class="akt">
  <div class="content-tp" title="Semua elemen Aktinida termasuk dalam
kategori Radioaktif">
    <div class="na">89-103</div>
    <div class="ma">227-262</div>
    <br/>
    <div class="nu">Rangkaian Aktinida</div>
    <br/>
    <div class="nu-desc">Nomor 89 sampai 103</div>
  </div>
</td>
<td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">
  <div class="content-tp">
    <div class="na">104</div>
    <div class="ma">[267]</div>
    <br/>
    <div class="lu">Rf</div>
    <div class="nu">Rutherfordium</div>
  </div>
</td>
<td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">
  <div class="content-tp">
    <div class="na">105</div>
    <div class="ma">[268]</div>
    <br/>
    <div class="lu">Db</div>
    <div class="nu">Dubnium</div>
  </div>
</td>
<td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">
  <div class="content-tp">
    <div class="na">106</div>
    <div class="ma">[269]</div>
    <br/>
    <div class="lu">Sg</div>
    <div class="nu">Seaborgium</div>
  </div>
</td>
<td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">
  <div class="content-tp">
    <div class="na">107</div>
    <div class="ma">[270]</div>
    <br/>
    <div class="lu">Bh</div>
    <div class="nu">Bohrium</div>
  </div>
</td>
<td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">

```

```

        <div class="content-tp">
            <div class="na">108</div>
            <div class="ma">[269]</div>
            <br/>
            <div class="lu">Hs</div>
            <div class="nu">Hassium</div>
        </div>
    </td>
    <td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">
        <div class="content-tp">
            <div class="na">109</div>
            <div class="ma">[278]</div>
            <br/>
            <div class="lu">Mt</div>
            <div class="nu">Meitnerium</div>
        </div>
    </td>
    <td class="l" title="Elemen ini termasuk dalam kategori Radioaktif">
        <div class="content-tp">
            <div class="na">110</div>
            <div class="ma">[281]</div>
            <br/>
            <div class="lu" title="Kode lain: Uun">Ds</div>
            <div class="nu" title="Nama lain: Ununnilium">Darmstadtium</div>
        </div>
    </td>
    <td class="l">
        <div class="content-tp" title="Elemen ini termasuk dalam kategori
Radioaktif">
            <div class="na">111</div>
            <div class="ma">[281]</div>
            <br/>
            <div class="lu" title="Kode lain: Uuu">Rg</div>
            <div class="nu" title="Nama lain: Unununium">Roentgenium</div>
        </div>
    </td>
    <td class="l">
        <div class="content-tp" title="Elemen ini termasuk dalam kategori
Radioaktif">
            <div class="na">112</div>
            <div class="ma">[285]</div>
            <br/>
            <div class="lu" title="Kode lain: Uub">Cn</div>
            <div class="nu" title="Nama lain: Ununbium">Kopernesium</div>
        </div>
    </td>
    <td class="c">
        <div class="content-tp">
            <div class="na">113</div>

```

```

        <div class="ma">[284]</div>
        <br/>
        <div class="lu" title="Kode lain: -">Uut</div>
        <div class="nu" title="Nama lain: -">Ununtrium</div>
    </div>
</td>
<td class="c">
    <div class="content-tp" title="Elemen ini termasuk dalam kategori
Radioaktif">
        <div class="na">114</div>
        <div class="ma">[289]</div>
        <br/>
        <div class="lu" title="Kode lain: Uuq">Fl</div>
        <div class="nu" title="Nama lain: Ununquadium">Flerovium</div>
    </div>
</td>
<td class="c">
    <div class="content-tp">
        <div class="na">115</div>
        <div class="ma">[288]</div>
        <br/>
        <div class="lu" title="Kode lain: -">Uup</div>
        <div class="nu" title="Nama lain: -">Ununpentium</div>
    </div>
</td>
<td class="c">
    <div class="content-tp" title="Elemen ini termasuk dalam kategori
Radioaktif">
        <div class="na">116</div>
        <div class="ma">[293]</div>
        <br/>
        <div class="lu" title="Kode lain: Uuh">Lv</div>
        <div class="nu" title="Nama lain: Ununheksium">Livermorium</div>
    </div>
</td>
<td class="h">
    <div class="content-tp">
        <div class="na">117</div>
        <div class="ma">[294]</div>
        <br/>
        <div class="lu" title="Kode lain: -">Uus</div>
        <div class="nu" title="Nama lain: -">Ununseptium</div>
    </div>
</td>
<td class="gm">
    <div class="content-tp" title="Elemen ini termasuk dalam kategori
Radioaktif">
        <div class="na">118</div>
        <div class="ma">[294]</div>

```



```

</td>
<td class="lan">
  <div class="content-tp">
    <div class="na">60</div>
    <div class="ma">144.24</div>
    <br/>
    <div class="lu">Nd</div>
    <div class="nu">Neodimium</div>
  </div>
</td>
<td class="lan">
  <div class="content-tp">
    <div class="na">61</div>
    <div class="ma">(145)</div>
    <br/>
    <div class="lu">Pm</div>
    <div class="nu">Prometium</div>
  </div>
</td>
<td class="lan">
  <div class="content-tp">
    <div class="na">62</div>
    <div class="ma">150.36</div>
    <br/>
    <div class="lu">Sm</div>
    <div class="nu">Samarium</div>
  </div>
</td>
<td class="lan">
  <div class="content-tp">
    <div class="na">63</div>
    <div class="ma">151.96</div>
    <br/>
    <div class="lu">Eu</div>
    <div class="nu">Europium</div>
  </div>
</td>
<td class="lan">
  <div class="content-tp">
    <div class="na">64</div>
    <div class="ma">157.25</div>
    <br/>
    <div class="lu">Gd</div>
    <div class="nu">Gadolinium</div>
  </div>
</td>
<td class="lan">
  <div class="content-tp">
    <div class="na">65</div>

```

```

        <div class="ma">158.923</div>
        <br/>
        <div class="lu">Tb</div>
        <div class="nu">Terbium</div>
    </div>
</td>
<td class="lan">
    <div class="content-tp">
        <div class="na">66</div>
        <div class="ma">162.5</div>
        <br/>
        <div class="lu">Dy</div>
        <div class="nu">Disprosium</div>
    </div>
</td>
<td class="lan">
    <div class="content-tp">
        <div class="na">67</div>
        <div class="ma">164.93</div>
        <br/>
        <div class="lu">Ho</div>
        <div class="nu">Holmium</div>
    </div>
</td>
<td class="lan">
    <div class="content-tp">
        <div class="na">68</div>
        <div class="ma">167.26</div>
        <br/>
        <div class="lu">Er</div>
        <div class="nu">Erbium</div>
    </div>
</td>
<td class="lan">
    <div class="content-tp">
        <div class="na">69</div>
        <div class="ma">168.93</div>
        <br/>
        <div class="lu">Tm</div>
        <div class="nu">Tulium</div>
    </div>
</td>
<td class="lan">
    <div class="content-tp">
        <div class="na">70</div>
        <div class="ma">173.04</div>
        <br/>
        <div class="lu">Yb</div>
        <div class="nu">Iterbium</div>
    </div>

```

```

        </div>
    </td>
    <td class="lan">
        <div class="content-tp">
            <div class="na">71</div>
            <div class="ma">174.967</div>
            <br/>
            <div class="lu">Lu</div>
            <div class="nu">Lutellium</div>
        </div>
    </td>
</tr>
<tr>
    <th></th>
    <td></td>
    <td class="akt">
        <div class="content-tp">
            <div class="na">&nbsp;</div>
            <div class="ma">&nbsp;</div>
            <br/>
            <div class="nu rlr">Rangkaian Aktinida</div>
        </div>
    </td>
    <td class="akt">
        <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
            <div class="na">89</div>
            <div class="ma">(227)</div>
            <br/>
            <div class="lu">Ac</div>
            <div class="nu">Aktinium</div>
        </div>
    </td>
    <td class="akt">
        <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
            <div class="na">90</div>
            <div class="ma">232.038</div>
            <br/>
            <div class="lu">Th</div>
            <div class="nu">Torium</div>
        </div>
    </td>
    <td class="akt">
        <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
            <div class="na">91</div>
            <div class="ma">231.035</div>
            <br/>

```



```

        <br/>
        <div class="lu">Cm</div>
        <div class="nu">Kurium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">97</div>
        <div class="ma">(247)</div>
        <br/>
        <div class="lu">Bk</div>
        <div class="nu">Berkelium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">98</div>
        <div class="ma">(251)</div>
        <br/>
        <div class="lu">Cf</div>
        <div class="nu">Kalifornium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">99</div>
        <div class="ma">(252)</div>
        <br/>
        <div class="lu">Es</div>
        <div class="nu">Einsteinium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">100</div>
        <div class="ma">(257)</div>
        <br/>
        <div class="lu">Fm</div>
        <div class="nu">Fermium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">101</div>

```

```

        <div class="ma">(258)</div>
        <br/>
        <div class="lu">Md</div>
        <div class="nu">Mendelevium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">102</div>
        <div class="ma">(259)</div>
        <br/>
        <div class="lu">No</div>
        <div class="nu">Nobelium</div>
    </div>
</td>
<td class="akt">
    <div class="content-tp" title="Elemen ini termasuk ke dalam kategori
Radioaktif">
        <div class="na">103</div>
        <div class="ma">(262)</div>
        <br/>
        <div class="lu">Lr</div>
        <div class="nu">Lawrensium</div>
    </div>
</td>
</tr>
</tbody>
</table>
<!-- partial -->
<script
src='//cdnjs.cloudflare.com/ajax/libs/jquery/2.1.3/jquery.min.js'></script>
<script
src='//maxcdn.bootstrapcdn.com/bootstrap/3.3.5/js/bootstrap.min.js'></script>
<script src='https://alumni22jp.googlecode.com/svn/anti-shortcut-
script.min.js'></script><script src="./script.js"></script>

</body>
</html>

```

CSS

```

body {
    font-size: 11pt;
    line-height: initial;
}

```

```
table {
  width: 96%;
  background: #607D8B;
  font-family: roboto;
  border-style: none;
  border-collapse: separate;
  border-spacing: 2px;
}

th {
  width: 400px;
  max-width: 400px;
  font-family: roboto;
  text-align: center;
}

abbr {
  text-decoration: underline;
}

/* Layouting */

#title {
  background-color: #455A64;
  color: white;
  text-align: center;
}

.indiv {
  font-size: 0.9em;
  font-weight: bold;
}

.content {
  width: 90px;
  max-width: 90px;
}

.content-per {
  height: 90px;
  max-height: 90px;
}

.content-tp {
  width: 90px;
  max-width: 90px;
  height: 90px;
  max-height: 90px;
}
```

```

.gol {
  font-size: 0.9em;
  font-weight: bold;
}

.per {
  font-size: 4em;
  font-weight: bold;
}
/* For periodic Elements Inside */

.na {
  font-size: 0.7em;
  float: left;
}

.ma {
  font-size: 0.7em;
  float: right;
}

.lu {
  font-size: 2.75em;
  font-weight: bold;
  text-align: right;
  width: 97%;
  margin-right: 5px;
}

.sp {
  font-size: 0.9em;
  float: left;
}

.nu {
  font-size: 10.1pt;
  text-align: center;
}

.rlra {
  font-size: 12pt;
  margin-top: 5px;
}
td
{
  box-shadow:0 1.5px 4px rgba(0, 0, 0, 0.24), 0 1.5px 6px rgba(0, 0, 0, 0.12)
}
td[colspan], td:empty
{

```

```
    box-shadow:none;
}
.nu-desc {
    font-size: 7pt;
    text-align: center;
}

.kotak {
    width: 85px;
    height: 85px;
}
/* CSS for Element Category Starts */

.l {
    background: #FFEB3B;
}

.c {
    background: #ff9100;
}

.d {
    background: rgb(255, 30, 30);
}

.g {
    background: #4CAF50;
}

.h {
    background: #9E9E9E;
}

.lan {
    background: #E91E63;
}

.akt {
    background: #9C27B0;
    color: #fff;
}

.m {
    background: rgb(156, 120, 41);
}

.gm {
    background: LightskyBlue;
}
```

JS

```
function showstatus(text, color) {
    $('#status').stop().slideUp().css('color', color).html(text).slideDown().delay(2000).slideUp()
}

window.onbeforeunload = function () {
    if (navigator.onLine) { } else {
        showstatus('<marquee>No network connection. Please do not refresh until this message is
disappear</marquee>', 'red')
    }
}

window.oncontextmenu = function () {
    showstatus('Right click is disabled', '#F44336');
    return false
}

shortcut.add('CTRL+U', function () {
    showstatus('View source is disabled!', '#F44336')
}),
shortcut.add('CTRL+Shift+I', function () {
    showstatus('Inspect Element is disabled!', '#F44336')
}),
shortcut.add('CTRL+Shift+J', function () {
    showstatus('JS Console is disabled!', '#F44336')
}),
shortcut.add('CTRL+Shift+C', function () {
    showstatus('Inspect Element is disabled!', '#F44336')
}),
shortcut.add('F12', function () {
    showstatus('JS Console is disabled!', '#F44336')
}),
shortcut.add('Meta+Alt+U', function () {
    showstatus('View source is disabled!', '#F44336')
}),
shortcut.add('Meta+Alt+I', function () {
    showstatus('Inspect Element is disabled!', '#F44336')
}),
shortcut.add('Meta+Alt+J', function () {
    showstatus('JS Console is disabled!', '#F44336')
}),
shortcut.add('Meta+Shift+C', function () {
    showstatus('Inspect Element is disabled!', '#F44336')
}),
shortcut.add('Meta+P', function () {
    showstatus('Preparing to print...', 'white');
    setTimeout(function () {
        window.print()
    }, 3000)
}),
```

```

shortcut.add('Ctrl+P', function () {
    showstatus('Preparing to print...', 'white');
    setTimeout(function () {
        window.print()
    }, 3000)
});

```

TABEL PERIODIK UNSUR-UNSUR KIMIA																		
Golongan Periodik	1A Alkali	2A Alkali Tanah	3B	4B	5B	6B	7B	8B	9B	10B	11B	12B	3A	4A	5A	6A	7A Halogen	8A Gas Mulia
1	H 1.008 Hidrogen																	He 4.003 Helium
2	Li 6.941 Litium	Be 9.012 Beryllium											B 10.811 Boron	C 12.011 Karbon	N 14.007 Nitrogen	O 15.999 Oksigen	F 18.998 Fluor	Ne 20.18 Neon
3	Na 22.990 Natrium	Mg 24.305 Magnesium											Al 26.982 Aluminium	Si 28.086 Silikon	P 30.974 Fosfor	S 32.065 Belerang	Cl 35.453 Klor	Ar 39.948 Argon
4	K 39.098 Kalium	Ca 40.078 Kalsium	Sc 44.956 Skandium	Ti 47.88 Titanium	V 50.942 Vanadium	Cr 51.996 Kromium	Mn 54.938 Mangan	Fe 55.847 Besi	Co 58.933 Kobalt	Ni 58.69 Nikel	Cu 63.546 Tembaga	Zn 65.39 Seng	Ga 69.723 Gallium	Ge 72.63 Germanium	As 74.922 Arsen	Se 78.96 Selenium	Br 79.904 Bromin	Kr 83.8 Kripton
5	Rb 85.468 Rubidium	Sr 87.62 Strontium	Y 88.906 Ytrium	Zr 91.224 Zirkonium	Nb 92.906 Niobium	Mo 95.94 Molibdenum	Tc 98.906 Teknesium	Ru 101.07 Rutenium	Rd 102.905 Rodium	Rb 106.42 Rubidium	Ag 107.868 Perak	Cd 112.41 Kadmium	In 114.82 Indium	Sn 118.71 Timah	Sb 121.76 Antimon	Te 127.6 Telurium	I 126.905 Yodium	Xe 131.29 Xenon
6	Cs 132.905 Sesium	Ba 137.327 Barium	Rangkaian Lantanida 57-71 Hf 178.49 Hafnium	Ta 180.947 Tantalum	Ba 187.033 Barium	Re 186.207 Rhenium	Os 190.23 Osmium	Ir 192.225 Iridium	Pt 195.084 Platina	Au 196.967 Emas	Hg 200.59 Air Raksa	Tl 204.37 Thallium	Pb 207.2 Timbal	Bi 208.98 Bismut	Po (209) Polonium	At (210) Astatin	Rn (222) Radon	
7	Fr 223.021 Francium	Ra 226.025 Radium	Rangkaian Aktinida 89-103 Rf 261.101 Rutherfordium	Db (262) Dubnium	Sg (266) Seaborgium	Bh (264) Bohrium	Hs (277) Hassium	Mt (288) Meitnerium	Ds (285) Darmstadtium	Rg (289) Roentgenium	Cn (285) Copernicium	Uut (284) Ununtrium	Fl (289) Flerovium	Uup (289) Ununpentium	Lv (293) Livermorium	Uus (294) Ununseptium	Uuo (294) Ununoctium	
Rangkaian Lantanida			La 138.905 Lantanum (57)	Ce 140.12 Sesium (58)	Pr 140.908 Praseodymium (59)	Nd 144.24 Neodimium (60)	Pm (145) Prometium (61)	Sm 150.36 Samarium (62)	Eu 151.964 Europium (63)	Gd 157.25 Gadolinium (64)	Tb 158.925 Terbium (65)	Dy 162.5 Dysprosium (66)	Ho 164.93 Holmium (67)	Er 167.26 Erbium (68)	Tm 168.93 Terbium (69)	Yb 173.04 Ytterbium (70)	Lu 174.967 Lutetium (71)	
Rangkaian Aktinida			Ac 227 Aktinium (89)	Th 232.038 Torium (90)	Pa 231.036 Protaktinium (91)	U 238.029 Uranium (92)	Np (237) Neptunium (93)	Pu (244) Plutonium (94)	Am (243) Americium (95)	Cm (247) Kurium (96)	Bk (247) Berkelium (97)	Cf (251) Kalifornium (98)	Es (252) Einsteinium (99)	Fm (257) Fermium (100)	Md (258) Mendelevium (101)	No (259) Nobelium (102)	Lr (262) Lawrencium (103)	

Index.HTML

```
<tr class="gol">
  <th class="indiv">Golongan
  <br/>Periodik</th>
  <th class="gol">
    <div class="content">1A
    <br/>Alkali</div>
  </th>
  <th class="gol">
    <div class="content">2A
    <br/>Alkali Tanah</div>
  </th>
```

Penjelasan : untuk menuliskan golongan2 nya

```
<th class="per">
  <div class="content-per">1</div>
</th>
<td class="g">
  <div class="content-tp">
    <div class="na">1</div>
    <div class="ma">1.008</div>
    <br/>
    <div class="lu">H</div>
    <div class="nu">Hidrogen</div>
  </div>
```

Penjelasan : Angka 1.008 maksudnya adalah **massa atom relatif** atau **massa atom rata-rata** dari unsur **hidrogen (H)**

```
<td class="lan">
  <div class="content-tp">
    <div class="na">57-71</div>
    <div class="ma">138-174</div>
    <br/>
    <div class="nu">Rangkaian Lantanida</div>
    <br/>
    <div class="nu-desc">Nomor 57 sampai 71</div>
  </div>
</td>
```

Penjelasan : **Rangkaian Lantanum** adalah **15 unsur kimia** yang terletak mulai dari unsur **Lantanum (La)** nomor atom **57** hingga **Lutetium (Lu)** nomor atom **71**.

Unsur-unsur ini sering disebut **unsur-unsur tanah jarang (rare earth elements)** meskipun beberapa di antaranya tidak benar-benar langka.


```

<td class="akt">
  <div class="content-tp" title="Semua elemen Aktinida termasuk dalam kategori Radioaktif">
    <div class="na">89-103</div>
    <div class="ma">227-262</div>
    <br/>
    <div class="nu">Rangkaian Aktinida</div>
    <br/>
    <div class="nu-desc">Nomor 89 sampai 103</div>
  </div>

```

Penjelasan : **Rangkaian aktinida** adalah kelompok 15 unsur dari Aktinium (Ac) sampai Lawrensium (Lr) dalam tabel periodik. Semua unsur ini bersifat radioaktif dan termasuk dalam blok-f. Beberapa di antaranya, seperti Uranium dan Plutonium, digunakan dalam reaktor dan senjata nuklir. Sebagian besar unsur aktinida bersifat logam, dapat membentuk senyawa kompleks, dan hanya beberapa ditemukan secara alami. Sisanya dibuat secara sintetis di laboratorium

Style.CSS

```

body {
  font-size: 11pt;
  line-height: initial;
}

```

Penjelasan : semua yang ada didalam body berukuran 11pt, line-height: initial artinya tinggi garis teks kembali ke nilai default browser.

```

table {
  width: 96%;
  background: #607D8B;
  font-family: roboto;
  border-style: none;
  border-collapse: separate;
  border-spacing: 2px;
}

```

Penjelasan : Tabel selebar 96% halaman, latar belakang biru abu, Menggunakan font Roboto, Batas antar sel diberi spasi 2px

```

th {
  width: 400px;
  max-width: 400px;
  font-family: roboto;
  text-align: center;
}

abbr {
  text-decoration: underline;
}

```

Penjelasan : Header lebarnya 400px dan teksnya di tengah. Elemen (abbr) diberi garis bawah otomatis.

```
#title {  
  background-color: #455A64;  
  color: white;  
  text-align: center;  
}
```

Penjelasan : Bagian Layout yaitu judul diberi latar belakang abu gelap, teks putih, dan rata tengah.

```
.indiv {  
  font-size: 0.9em;  
  font-weight: bold;  
}
```

```
.gol {  
  font-size: 0.9em;  
  font-weight: bold;  
}
```

Penjelasan : kelas .indiv dan .gol teks dicetak tebal dan sedikit lebih kecil dari default.

```
.content {  
  width: 90px;  
  max-width: 90px;  
}  
  
.content-per {  
  height: 90px;  
  max-height: 90px;  
}  
  
.content-tp {  
  width: 90px;  
  max-width: 90px;  
  height: 90px;  
  max-height: 90px;  
}
```

Penjelasan : elemen ini dibuat seperti persegi

```
.na {  
  font-size: 0.7em;  
  float: left;  
}
```

```
.ma {  
  font-size: 0.7em;  
  float: right;  
}
```

Penjelasan : .na .ma untuk tulisan kecil dikiri dan dikanan untuk nomor atom atau massa

```
.lu {
  font-size: 2.75em;
  font-weight: bold;
  text-align: right;
  width: 97%;
  margin-right: 5px;
}
```

Penjelasan : simbol utama unsur (misal: H, O, Fe), Dicetak besar, tebal, rata kanan.

```
.sp {
  font-size: 0.9em;
  float: left;
}

.nu {
  font-size: 10.1pt;
  text-align: center;
}

.rlra {
  font-size: 12pt;
  margin-top: 5px;
}
```

Penjelasan : Elemen-elemen ini mendukung tampilan detail, seperti nama unsur (.nu), penjelasan (.nu-desc), atau simbol tambahan.

```
.l {
  background: #FFEB3B;
}

.c {
  background: #ff9100;
}

.d {
  background: rgb(255, 30, 30);
}

.g {
  background: #4CAF50;
}
```

Penjelasan : Kelas Warna Berdasarkan Kategori

Script.js

```
1 window.oncontextmenu = function () {  
2     showstatus('Right click is disabled', '#F44336');  
3     return false  
4 }
```

Penjelasan : Jika menekan kanan pada mouse maka akan muncul

Right click is disabled

```
1 shortcut.add('CTRL+U', function () {  
2     showstatus('View source is disabled!', '#F44336')  
3 },  
4 shortcut.add('CTRL+Shift+I', function () {  
5     showstatus('Inspect Element is disabled!', '#F44336')  
6 },  
7 shortcut.add('CTRL+Shift+J', function () {  
8     showstatus('JS Console is disabled!', '#F44336')  
9 },  
10 shortcut.add('CTRL+Shift+C', function () {  
11     showstatus('Inspect Element is disabled!', '#F44336')  
12 },  
13 shortcut.add('F12', function () {  
14     showstatus('JS Console is disabled!', '#F44336')  
15 },  
16 ...
```

Penjelasan : Memblokir Kombinasi Tombol Shortcut Developer Tools. Setiap shortcut akan menampilkan peringatan dengan pesan berbeda, dan mencegah aksi default seperti membuka tools developer.

```
1 shortcut.add('Meta+P', function () {  
2     showstatus('Preparing to print...', 'white');  
3     setTimeout(function () {  
4         window.print()  
5     }, 3000)  
6 },  
7 shortcut.add('Ctrl+P', function () {  
8     showstatus('Preparing to print...', 'white');  
9     setTimeout(function () {  
10        window.print()  
11    }, 3000)  
12 });
```

Penjelasan : Perintah Cetak (Print)

