### Electrical Principles - Part 1

### Storage and transfer of electrical energy

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Slides: github.com/mariopineda/electrical-principles-slides



### **Objectives**

- Describe the major features of cells
- Know the two major types of cells

### **Battery Classification**

- ► Primary Cells: Cannot be recharged as the chemical reactions cannot be reversed.
- ► Secondary Cells: Can be recharged by passing a current through the cell in the opposite direction.

### Battery Chemistries: Lithium Polymer (LiPo)

- ▶ Type: Secondary cell
- ► Chemistry: Lithium-ion and polymer (gel) electrolyte
- Pros: High energy content, light, can be produced in any shape
- Cons: Physical damage, over charge or too high temperature can cause cells to fail catastrophically (fire, explosion)
- ▶ Voltage: 2.7-4.2V
- ► Usage: personal electronics (laptops, cell phones), RC vehicles, International Space Station (since 2017)
- ▶ Invented by Sony in 1991

# Battery Chemistries: Lithium Ion Phosphate (LiFePO4)

- ► Type:
- ► Chemistry:
- ► Pros:
- ► Cons:
- Voltage:
- ► Usage:

## Battery Chemistries: Lead-Acid Battery

- ▶ Type: Secondary cell
- Chemistry: Lead, lead dioxide, electrolyte concentrated sulfuric acid
- Pros: High energy density, many recharge cycles, cheap
- Cons: Effectiveness reduced at low temperatures, self-discharges, contains lead and concentrated sulfuric acid
- Voltage: 2V per cell
- Usage: Vehicle starter and ignition, backup power supplies (computer UPS units)
- Oldest type of rechargable

# Battery Chemistries: NiCd

- ► Type: Secondary cell
- ▶ Chemistry:
- Pros: Long Shelf life,many recharge cycles, good performance at low temperatures, can produce very large instantaneous currents (1000-8000A for a second)
- Cons: Contains Cadmium (heavy, expensive and toxic), corrosive electrolytes
- ▶ Voltage: 1.2V per cell
- Usage: Portable electronic equipment, e.g. flash lights, aircraft and satellite systems, starting large disel engines and turbines.

## Battery Chemistries: NiMH

- ► Type:
- ► Chemistry:
- ► Pros:
- ► Cons:
- ▶ Voltage:
- ► Usage:

### Fuel Cells

- ► Fuel and oxidant are continuously supplied from the outise while vaste products are removed.
- ► Pros: High-efficiency of energy conversion, no harmful waste products, no need to recharge
- ► Cons: Expensive, fuel gases must be stored in high-pressure tanks, moderate power output
- Use: The space shuttle and Apollo space program, military submarines