

1. $\Psi(x, t)$ has no direct physical meaning

☒ T ☐ F Only $\int |\Psi|^2 dr$ has a statistical interpretation

2. The kinetic energy of a wavefunction is related to its curvature.

☒ T ☐ F $\hat{E}_{\text{kin}} = \frac{\hat{p}^2}{2m} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$ has a second derivative, indicating curvature relationship.

3. According to the uncertainty principle if σ_x is very large, then the momentum is well determined.

☐ T ☒ F Common misconception; The uncertainty principle is an inequality: $\sigma_x \cdot \sigma_p \geq \frac{\hbar}{2}$.

4. The variance of a discrete variable j is given by $\sum_{j=0}^{\infty} (j - \langle j \rangle) P(j)$.

☐ T ☒ F In the equation, a power of two is missing at the end; it should be $(j - \langle j \rangle)^2$.

5. Any solution of the Schrödinger Equation can be normalized

☐ T ☒ F For example, $\Psi(x) = e^{ikx}$ is not square-integrable and cannot be normalized.

6. In the quantum mechanical description of black body radiation, every photon mode out to infinite frequency carries $k_b T$ of energy, where k_b is Boltzmann's constant and T is the absolute temperature.

☐ T ☒ F The energy of each photon mode in black body radiation depends on its frequency

7. According to Ehrenfest's theorem expectation values in quantum mechanics follow classical laws.

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8. According to the de Broglie relation, an electron cannot act simultaneously as a particle and a wave at small length scales.

☐ T ☒ F De Broglie links particle behavior to wave behavior.