



Writing Expectation History

Mathematics DT Science PE Languages

Art Geography Differentiation

Progression National Curriculum

Languages English Writing Progression
Differentiation Science Art
Mathematics Expectation

Progression in the new National Curriculum

| o.m. | | | | . • | | | |
|---|--|--|---|---|--|---------------|-----------------|
| Everyday materials | | Yes | Yes | - | - | - | - |
| Year 1 | Year 2 | | | | | | |
| Distinguish between an object and the material from w made. Identify and name a variety of everyday materials, inclusion wood, plastic, glass, metal, water, and rock. Describe the simple physical properties of a variety of ematerials. Compare and group together a variety of everyday materials of their simple physical properties. | uding mo and Find everyday mo stre | iterials, incl d cardboa d out how | uding woo rd for partion the shapes | d, metal, p cular uses. of solid ob | y of a varie plastic, glas pjects made ashing, bene | s, brick, roo | ck, paper ne |

Y1

Y2

Y3

Y5

Y6

| Unit: | Y1 | Y2 | Y3 | Y4 | Y5 | Y6 |
|-------------------------------------|----|----|----|----|-----|----|
| Properties and changes of materials | - | - | - | - | Yes | _ |

Year 5

Unit:

- Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to magnets.
- Know that some materials will dissolve in liquid to form a solution, and describe how to recover a substance from a solution.
- Use knowledge of solids, liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating.
- Give reasons, based on evidence from comparative and fair tests, for the particular uses of everyday materials, including metals, wood and plastic.
- Demonstrate that dissolving, mixing and changes of state are reversible changes.
- Explain that some changes result in the formation of new materials, and that this kind of change is not usually reversible, including changes associated with burning and the action of acid on bicarbonate of soda.

